

A Highly Stereocontrolled Asymmetric Synthesis of the Taxol C-13 side chain; (4S, 5R)-2,4-Diphenyloxazoline-5-carboxylic acid.

Kee-Young Lee, Yong-Hyun Kim, Min-Sung Park, and Won-Hun Ham*

College of Pharmacy, SungKyunKwan University, Suwon 440-746, Korea

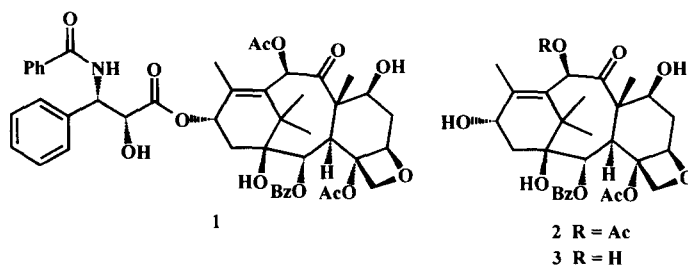
Received 2 July 1998; accepted 28 August 1998

Abstract

A stereoselective synthesis of (4S, 5R)-2,4-diphenyloxazoline-5-carboxylic acid, a precursor of the Taxol C-13 side chain was achieved using palladium-catalyzed oxazoline formation reaction from commercially available amino acid. © 1998 Elsevier Science Ltd. All rights reserved.

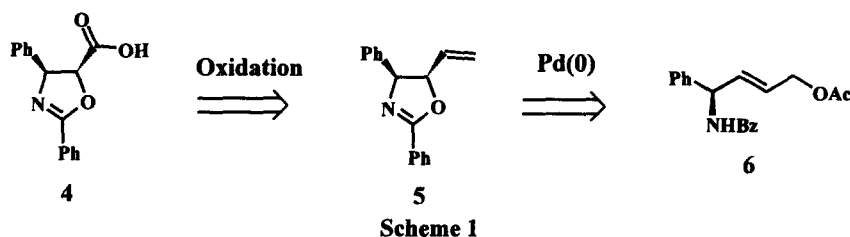
Keywords: oxazoline; palladium and compounds; asymmetric synthesis; taxoid.

Paclitaxel (Taxol[®], **1**) isolated from the bark of the Pacific Yew *Taxus brevifolia*, is currently considered the most important anticancer agent and has been approved for treatment of metastatic ovarian and breast cancer[1-2]. The scarcity has led to the search for semisynthetic routes to taxol using other plant-derived products isolable in useful quantities.

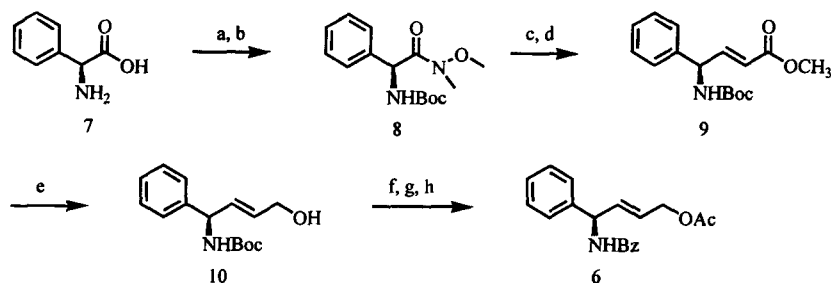


Thus, a viable approach for the preparation of **1** is to utilize more accessible baccatin III(**2**) or 10-deacetylbaccatin III(**3**) as precursors via a semisynthetic route. Indeed, the successful applications of such semisynthetic strategies to the synthesis of **1** have been

achieved[3-7], and these have prompted research into the development of Taxol side chain synthesis that is practical and adaptable to relatively large-scale production. As a part of program directed towards asymmetric synthesis of (4S, 5R)-2,4-diphenyloxazoline-5-carboxylic acid(**4**) of biological importance, we undertook a synthesis of the title amino acid in enantiopure form and report herein the details of the study(**scheme1**). The key step in our approach involves a highly enantioselective oxazoline formation catalyzed by palladium. The simplicity and low cost of starting material, and the ease of chromatographic separation may render this process the most practical route to enantiomerically pure oxazoline derivatives developed to date.



The requisite cyclization precursor (**6**) is straightforwardly prepared in high yield from the commercially available L-phenylglycine by 8 step sequences(overall 45%) as shown in **scheme 2**. L-phenylglycine (**7**) was treated with di-*tert*-butyl-dicarbonate to afford the corresponding N-Boc-phenylglycine, which was reacted with N,O-dimethylhydroxylamine hydrochloride in the presence of DCC and HOBT in CH₂Cl₂. The resulting amide (**8**) was reacted with lithium aluminium hydride at -40 °C to give the aldehyde[8], followed by a Wittig olefination using trimethylphosphonoacetate then led to the *E*-unsaturated ester (**9**). Thus, reduction of *E*-unsaturated ester, using DIBAL at -78 °C produced the corresponding allyl alcohol (**10**). Deprotection of the Boc group using aqueous hydrochloric acid followed by benzoylation of the resulting amino group formed the N-benzoyl alcohol. Acetylation of the hydroxy group yielded the key precursor (**6**) of cyclization.



Scheme 2. Reagents and conditions : a) Boc₂O, NaOH, *t*-BuOH, H₂O; b) HNCH₃(OCH₃)-HCl, DCC, HOBT, Et₃N, CH₂Cl₂, 81%(for 2steps); c) LAH, (C₂H₅)₂O, -40 °C; d) (CH₃O)₂POCH₂CO₂CH₃, LiCl, diisopropylethylamine, CH₃CN, 80%(for 2steps); e) DIBAL, BF₃ OEt₂, CH₂Cl₂, -78 °C, 89%; f) 3N-HCl in EtOAc; g) BzCl, Et₃N, THF, 0 °C; h) Ac₂O, pyridine, CH₂Cl₂, 78%(for 3steps).

Although cyclization of enamides or enimidates by activation of the olefin with electrophiles has been well documented[9-11], there are few examples of palladium-catalyzed oxazoline formation reaction[12-13]. Conditions for the cyclization of oxazoline precursor (**6**) were intensively explored by examination of the effects of bases, ligands, and solvents as summarized in **Table 1**. CH₃CN or CH₂Cl₂ was an equally effective solvent, but DMF and THF gave a mixture of oxazoline (**5**) and elimination product (**11**). Pd(PPh₃)₄, K₂CO₃, and CH₃CN turned out to be the best condition for the highest conversion yield. Especially, cyclization of **6** with K₂CO₃ in the presence of 5mol% Pd(PPh₃)₄ in CH₃CN(entry 9) proceeded smoothly to afford a 6:1 mixture of **5** and **11** in favor of **5** as the desired compound. In all cases, *cis*-oxazoline was not obtained.

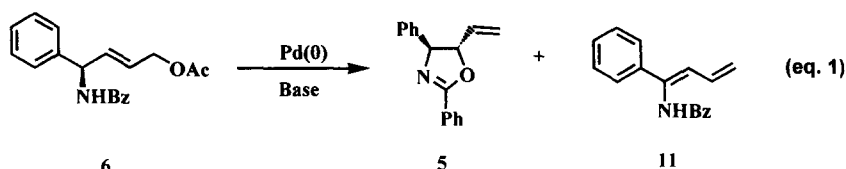


Table 1. Palladium-catalyzed oxazoline formation reaction

entry	Pd(0)	base ^a	solvent	Rx. Temp(°C)	Rx. Time(hr)	yield(%) ^b (5 : 11)
1	Pd(PPh ₃) ₄ (5mol%)	NaH	DMF	rt	8	30 : 45
2	Pd ₂ dba ₃ -CHCl ₃ (1mol%) PPh ₃ (8mol%)	NaH	DMF	rt	8	29 : 48
3	Pd ₂ dba ₃ -CHCl ₃ (1mol%) dppe(4mol%)	NaH	DMF	rt	8	0 : 90
4	Pd ₂ dba ₃ -CHCl ₃ (1mol%) PPh ₃ (8mol%)	K ₂ CO ₃	CH ₃ CN	reflux	8	63 : 10
5	Pd ₂ dba ₃ -CHCl ₃ (1mol%) PPh ₃ (8mol%)	K ₂ CO ₃	CH ₂ Cl ₂	reflux	8	19 : 4
6	Pd(PPh ₃) ₄ (5mol%)	K ₂ CO ₃	DMF	60	24	40 : 38
7	Pd(PPh ₃) ₄ (5mol%)	Hünig base	DMF	60	24	20 : 19
8	Pd(PPh ₃) ₄ (5mol%)	K ₂ CO ₃	CH ₂ Cl ₂	reflux	24	70 : 25
9	Pd(PPh ₃) ₄ (5mol%)	K ₂ CO ₃	CH ₃ CN	reflux	24	78 : 13
10	Pd(PPh ₃) ₄ (5mol%)	K ₂ CO ₃	THF	rt	4	49 : 38

a. Bases were used as follows: NaH(1equiv.), K₂CO₃(2equiv.), Hünig base(2equiv.).

b. Yields refer to isolated and chromatographically pure products.

The high stereoselectivity in cyclization of **6** may be arise due to the differences of steric interactions between the bulky phenyl group and hydrogen of π -allyl palladium complex in the transition state **A** and **B**. Consequently, cyclization proceeds through the more favored transition state **A** as shown in **Figure 1**.

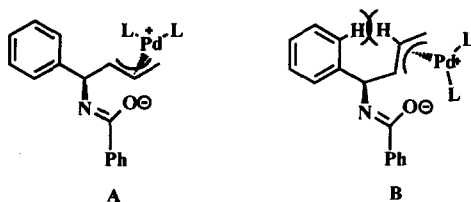
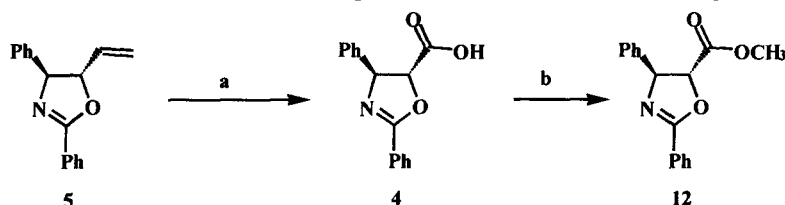


Figure 1.

Oxidative degradation of the vinyl group of oxazoline yielded the corresponding (4S, 5R)-2,4-diphenyloxazoline-5-carboxylic acid(**4**), which was then converted to its methyl ester (**12**)(**scheme 3**). From ^1H NMR studies, the observed coupling constant($J_{4,5} = 6.5\text{Hz}$) between the two protons in the oxazoline ring indicated a *trans*-relationship. Optical rotation of **12** $\{[\alpha]_{\text{D}} = +12.9$ ($c = 1$, CHCl_3) : lit. $[\alpha]_{\text{D}} = +13$ ($c = 1$, CHCl_3)[14] was in good agreement with the reported value, which also conclusively proved the stereochemical assignment.



Scheme 3. Reagents and conditions : a) RuCl_3 , NaIO_4 , NaHCO_3 , $\text{H}_2\text{O}/\text{CCl}_4/\text{CH}_3\text{CN}$, 2days, 78%; b) CH_2N_2 , $(\text{C}_2\text{H}_5)_2\text{O}$, 100%.

In summary, a new synthesis of (4S, 5R)-2,4-diphenyloxazoline-5-carboxylic acid(**4**) was accomplished by using palladium-catalyzed oxazoline ring formation reaction.

Acknowledgement : We thank Dong Kook pharmaceutical Co. LTD for financial support.

References and notes:

- [1] Nicolaou KC, Dai, W-M, Guy RK. *Angew. Chem.* 1994; 106: 38-67.
- [2] Nicolaou KC, Dai, W-M, Guy RK. *Angew. Chem., Int. Ed. Engl.* 1994; 33: 15-44.
- [3] Gennari C, Carcano M, Donghi M, Mongelli N, Vanotti E, Vulpetti A. *J. Org. Chem.* 1997; 62: 4746-4755 and references cited therein.
- [4] Brown S, Jordan AM, Lawrence NJ, Pritchard RG, McGown AT. *Tetrahedron Lett.* 1998; 39: 3559-3562.
- [5] Kobayashi S, Ishitani H, Ueno M. *J. Am. Chem. Soc.* 1998; 120: 431-432.
- [6] Lee D, Kim M-J. *Tetrahedron Lett.* 1998; 39: 2163-2166.
- [7] Cardillo G, Gentilucci L, Tolomelli A, Tomasini C. *J. Org. Chem.* 1998; 63: 2351-2353.
- [8] Goel OP, Krolls U, Kesten S. *Org. Synth., Coll. Vol. VIII*, 68-71.
- [9] Abd El Samii ZKM, Al Ashmawy MI, Mellor JM. *J. Chem. Soc., Perkin Trans. I* 1988; 2517-2522.
- [10] Engman L. *J. Org. Chem.* 1991; 56: 3425-3430.
- [11] Bongini A, Cardillo G, Orena M, Sandri S, Tomasini C. *J. Org. Chem.* 1986; 51: 4905-4910.
- [12] Cook GR, Shanker PS. *Tetrahedron Lett.* 1998; 39: 3405- 3408.
- [13] Hansel J-G, O'Hogan S, Lensky S, Ritter AR, Miller MJ. *Tetrahedron Lett.* 1995; 36: 2913-2916.
- [14] Gou D-M, Liu Y-C, Chen C-S. *J. Org. Chem.* 1993; 58: 1287-1289.