

Full Paper

Design and Synthesis of New 1,3-Benzodiazinan-4-one Derivatives as Selective Cyclooxygenase (COX-2) Inhibitors

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A new group of regioisomeric 2,3-diaryl-1,3-benzodiazinan-4-ones, possessing a methyl sulfonyl pharmacophore, were synthesized and their biological activities were tested for cyclooxygenase-2 (COX-2) inhibitory activity. *In vitro* COX-1/COX-2 inhibition studies identified 3-(*p*-fluorophenyl)-2-(4-methylsulfonylphenyl)-1,3-benzodiazinan-4-one (**2b**) as a potent and highly selective ($IC_{50} = 0.07 \mu\text{M}$; selectivity index = 572.8) COX-2 inhibitor.

Keywords: 1,3-Benzodiazinan-4-ones / COX-2 inhibition / Docking studies / SAR

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Introduction

The cyclooxygenase (COX) enzymes catalyze a key step in the conversion of arachidonate to PGH₂, the immediate substrate for a series of cell specific prostaglandin and thromboxane synthases. At least two COX isoforms have been identified thus far: the constitutively expressed COX-1 and COX-2, which is inducible under pathologic conditions and of which increased concentrations have been observed in inflamed and tumorous tissues [1, 2]. In addition to COX-1 and COX-2, a third isoform of COX enzyme (COX-3) has been also identified. It is considered to play a key role in the biosynthesis of prostanoids known to be important mediators in pain and fever [3]. It is known that selective COX-2 inhibitors can provide anti-inflammatory agents devoid of the undesirable gastro-intestinal side effects [4]. In addition to the role of COX-2 in inflammatory disorders such as rheumatoid arthritis and osteoarthritis, it is also implicated in cancer and angiogenesis. In this regard, several epidemiologic studies have reported that inhibitors of COX-2 enzyme reduce the risk of several cancer types such as colon,

breast, lung and prostate cancers [5–8] and neurodegenerative diseases such as Parkinson [9] and Alzheimer's [10] disease. Selective COX-2 inhibitors frequently belong to a class of tricycles that possess vicinal diaryl moieties attached to a central ring scaffold in conjunction with a COX-2 pharmacophore such as a *para*-SO₂NH₂, a *para*-SO₂Me, substituent on one of the phenyl rings [11]. Celecoxib and rofecoxib are two typical selective COX-2 inhibitors in this class (COXIBs). However, the market withdrawal of some COXIBs such as rofecoxib due to their increasing the risk of heart attack and cardiovascular side effects [12] encourages the researchers to explore new selective COX-2 inhibitors to evaluate their effects and may improve the safety profiles. For this reason novel scaffolds with high selectivity for COX-2 inhibition need to be found and evaluated for their biological activities. In this regard, we recently reported several investigations describing the design, synthesis, and COX inhibitory activities of new compounds possessing a bicyclic 1,3-benzthiazinan (**A**) [13] or quinoline (**B**) [14, 15] structural template (Fig. 1). Our results showed that both 1,3-benzthiazinan and quinoline rings are very suitable scaffolds for COX-2 inhibitory activity and some of these synthesized compounds exhibited high potency and selectivity on COX-2 inhibition even more than the reference drug celecoxib. As part of our research program, aimed at discovering new selective COX-2 inhibitors, we focused our attention on the synthesis, COX inhibitory activity and some molecular modeling studies of

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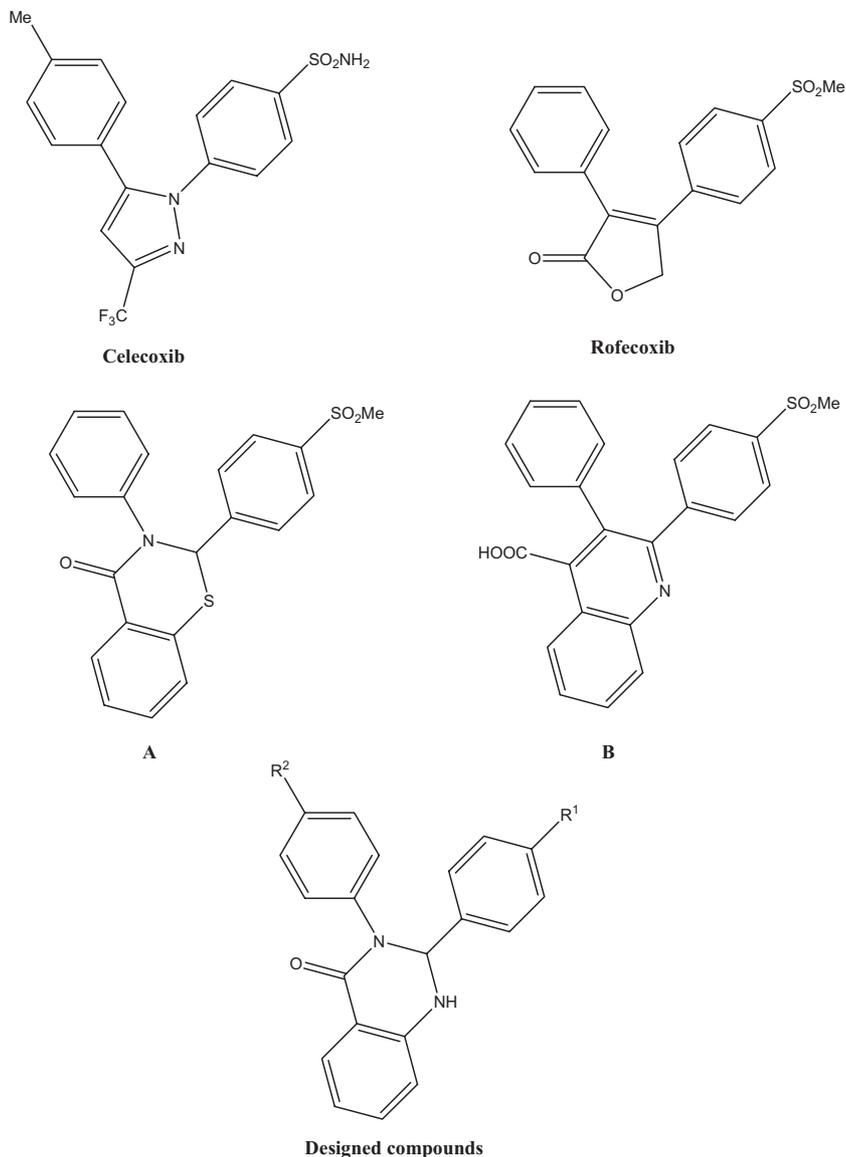


Figure 1. Some representative examples of COXIBs (celecoxib and rofecoxib), 1,3-benzthiazinane-4-one (**A**), quinoline-4-carboxylic acid (**B**) lead compounds and our 1,3-benzdiazinan-4-one scaffolds.

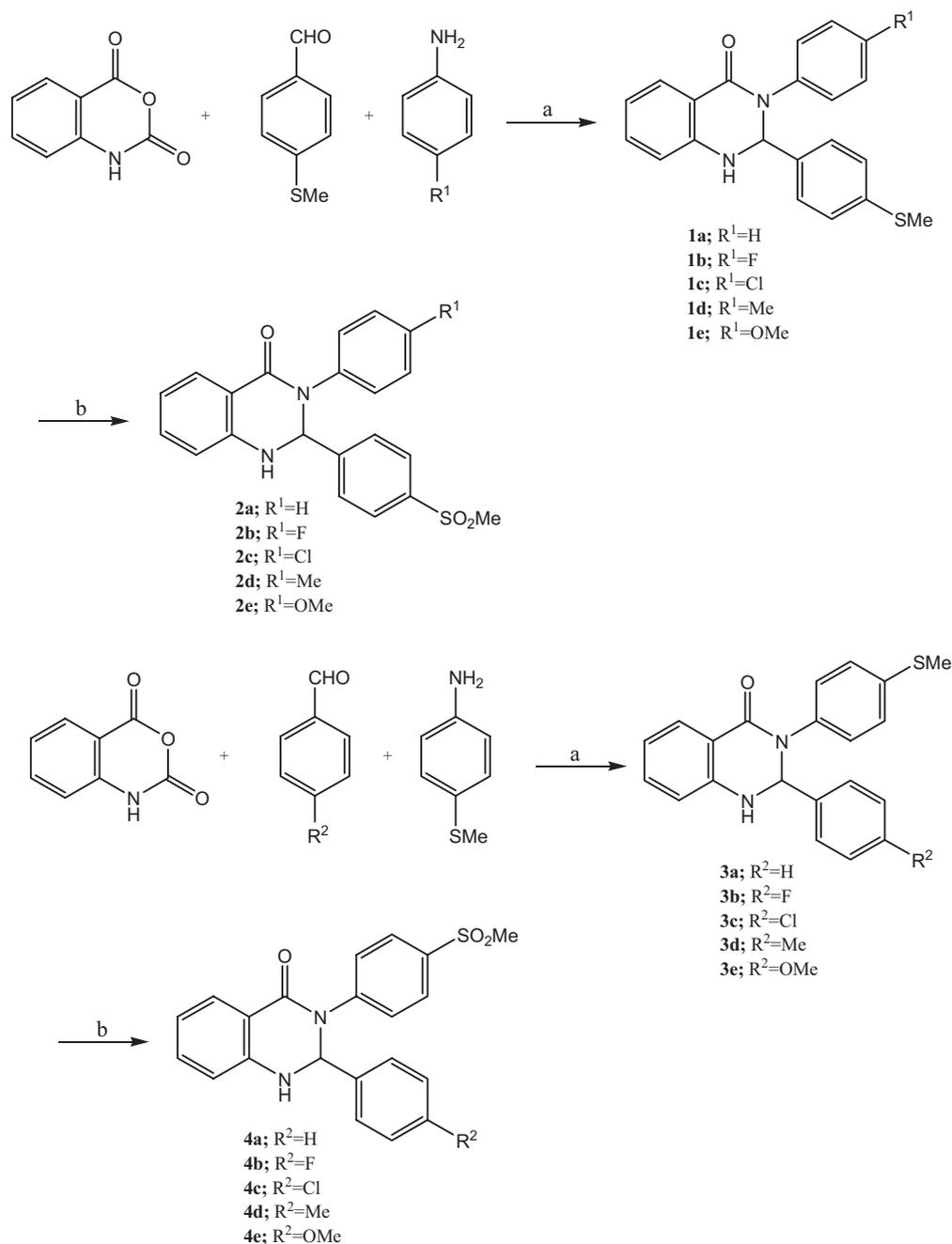
a new group of regioisomeric 2,3-diaryl-1,3-benzdiazinan-4-ones. The rationale for the design of these compounds was based on hybrid structure of 2,3-diaryl-1,3-benzthiazinane-4-ones and 2,3-diaryl-quinolines in our previously reported COX-2 inhibitors.

Results and discussion

Chemistry

The two sets of 1,3-benzdiazinan-4-one regioisomers in which the 4-methylsulfonylphenyl substituent is attached to C-2 (**2a–e**) or to N-3 (**4a–e**), were synthesized via the route outlined in Scheme 1. Accordingly, a one-pot three-component cyclocondensation of a mixture of isatoic

anhydride, aldehyde and primary amine in ethanol using alum ($KAl(SO_4)_2 \cdot 12H_2O$) as catalyst was used to prepare 1,3-benzdiazinan-4-ones (**1** or **3**; yield: 14–30%) [16]. Presumably, the reaction proceeds through condensation of isatoic anhydride with primary amine followed by decarboxylation to yield the corresponding 2-aminobenzamide. Then, condensation of the aromatic aldehyde with the amino group of 2-aminobenzamide gives the imine intermediate which undergoes cyclization to afford the 1,3-benzdiazinan-4-one product [16]. The low yield of the product may be explained by the possible formation of byproducts such as the reaction between the amine and aldehyde or low activity of starting materials such as the aldehyde having electron donating group. Oxidation of cyclic intermediate using 30% H_2O_2 in



Scheme 1. Reagents and conditions: (a) Alum, EtOH, reflux, 1 h; (b) H₂O₂ 30%, WO₃, 25°C, 4–5 h.

hydromethanol in the presence of a trace amount of WO₃ gave the target products (**2** or **4**, yield: 22–36%) [17].

Biological evaluation

A group of 1,3-benzodiazinan-4-one derivatives having a MeSO₂ group at the *para*-position of the C-2 phenyl ring containing a variety of substituents (H, F, Cl, Me, OMe) at the *para*-position of the N-3 phenyl ring (**2a–e**), and the

corresponding regioisomers (**4a–e**), were synthesized to investigate the effect of these substituents on COX-2 selectivity and potency. The ability of the 1,3-benzodiazinan-4-ones to inhibit the COX-1 and COX-2 isozymes was determined using chemiluminescent enzyme assays (see enzyme inhibition data in Table 1) according to our previously reported method [18]. In vitro COX-1/COX-2 inhibition studies showed that all compounds were selective inhibitors of the COX-2 isozyme with

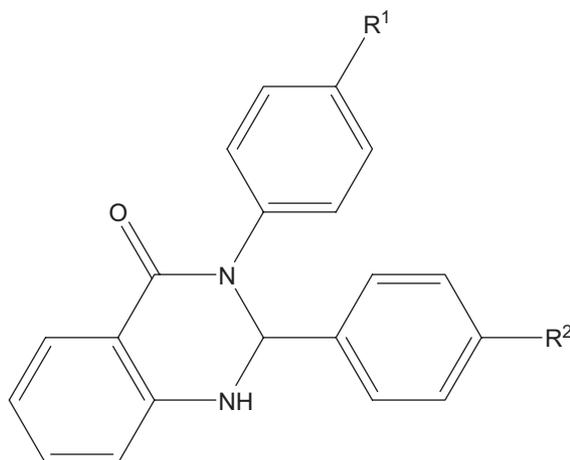
IC₅₀ values in the highly potent 0.07 to 0.22 μM range, and COX-2 selectivity indexes (S.I.) in the 95.0 to 572.8 range. SAR data (IC₅₀ μM values) acquired by determination of the *in vitro* ability of the title compounds to inhibit the COX-1 and COX-2 isozymes showed that the position of the COX-2 SO₂Me pharmacophore and the nature of the *para*-substituents on the C-2 or N-3 phenyl ring were either individual, or collective, determinants of COX-2 inhibitory potency and selectivity. *In vitro* COX-1/COX-2 inhibition studies showed that compounds having a MeSO₂ group at the *para*-position of the C-2 phenyl ring (**2a–e**) were more selective COX-2 inhibitors compared to their corresponding regioisomers (**4a–e**). These data showed that the nature and size of substituent attached to C-2 or N-3 of the 1,3-benzodiazinan-4-one ring influenced both selectivity and potency for COX-2 inhibitory activity. Our results indicated that the introduction of suitable substituents such as F (**2b**) and OMe (**2e**) at the *para*-position of N-3 phenyl ring increased both selectivity and potency for COX-2

inhibitory activity. The similar results were obtained for their corresponding regioisomers (**4b** and **4e**). However, compounds having Cl substituent (**2c** and **4c**) showed less selectivity and potency for COX-2 isozyme. According to these results, 3-(4-fluorophenyl)-2-(4-methylsulfonylphenyl)-1,3-benzodiazinan-4-one **2b** was the most potent (IC₅₀ = 0.07 μM), and selective (S.I. = 572.8), COX-2 inhibitor among the synthesized compounds. It was more selective than celecoxib (IC₅₀ = 0.06 μM; S.I. = 405) in terms of COX-2 inhibitory activity but showed less potency. These data suggest that the compound **2b** should inhibit the biosynthesis of prostaglandins via the cyclooxygenase pathway at sites of inflammation and be devoid of ulcerogenicity due to the absence of COX-1 inhibition.

Molecular modeling studies

The orientation of the highly potent and selective COX-2 inhibitor, 3-(4-fluorophenyl)-2-(4-methylsulfonylphenyl)-1,3-

Table 1. *In vitro* COX-1 and COX-2 enzyme inhibition assay data **3–9**.



Compound	R ¹	R ²	IC ₅₀ (μM) ^{a)}		COX-2 S.I. ^{b)}
			COX-1	COX-2	
2a	H	SO ₂ Me	99.8	0.35	285.1
2b	F	SO ₂ Me	40.1	0.07	572.8
2c	Cl	SO ₂ Me	17.7	0.17	104.1
2d	Me	SO ₂ Me	37.2	0.10	372.0
2e	OMe	SO ₂ Me	33.4	0.07	477.1
4a	SO ₂ Me	H	24.1	0.12	200.8
4b	SO ₂ Me	F	31.8	0.09	353.3
4c	SO ₂ Me	Cl	20.9	0.22	95.0
4d	SO ₂ Me	Me	24.1	0.11	219.1
4e	SO ₂ Me	OMe	29.7	0.11	270.0
Celecoxib			24.30	0.06	405

^{a)} Values are means of two determinations acquired using an ovine COX-1/COX-2 assay kit and the deviation from the mean is <10% of the mean value.

^{b)} *In vitro* COX-2 selectivity.

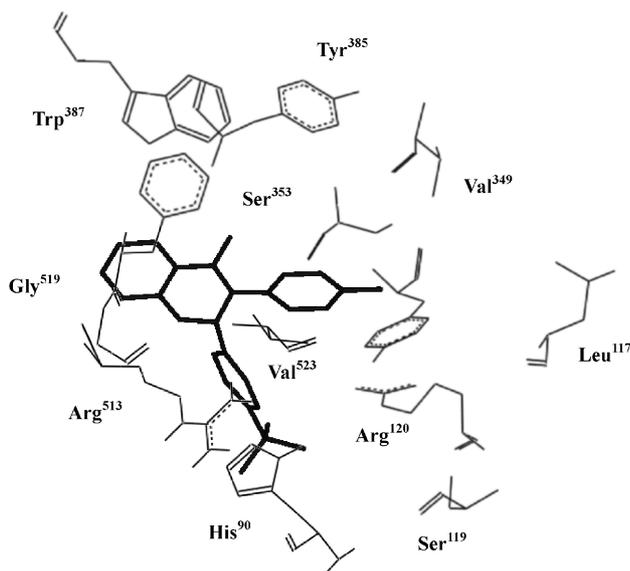


Figure 2. Compound **2b** 3-(*p*-fluorophenyl)-2-(4-methylsulfonylphenyl)-1,3-benzodiazinan-4-one docked in the active site of murine COX-2 isozyme..

benzodiazinan-4-one **2b** in the COX-2 active site was examined by a docking experiment (Fig. 2) [19, 20]. This molecular modeling shows that it binds in the primary binding site such that the C-2 *para*-SO₂Me substituent inserts into the secondary pocket present in the COX-2 isozyme. One of the O-atoms of *p*-SO₂Me forms a hydrogen bonding interaction with amino group of Arg⁵¹³ (distance = 2.9 Å) whereas the other O-atom is close to NH of His⁹⁰ amino acid (distance = 3.4 Å). Also, the NH group of benzodiazinan central ring is very close to carbonyl group of Gly⁵¹⁹ (distance = 2 Å) and can form hydrogen binding. In addition, the C=O of benzodiazinan-4-one ring is almost close to (distance = 5.4 Å) NH of Tyr³⁸⁵. It was interesting to note that the *para*-fluoro substituent of N-3 phenyl ring was forming a hydrogen bond with hydroxyl group (OH) of Tyr³⁵⁵ (distance = 4.8 Å). These observations together with experimental results provide a good explanation for the potent and selective inhibitory activity of **2b**.

Conclusions

The results of this investigation indicate that (i) the 1,3-benzodiazinan-4-one moiety is a suitable scaffold (template) to design COX-2 inhibitors, (ii) in this class of compounds COX-1/2 inhibition is sensitive to the position of the COX-2 SO₂Me pharmacophore and the nature of the C-2 or N-3 phenyl substituents, and (iii) 3-(*p*-fluorophenyl)-2-(4-methylsulfonylphenyl)-1,3-benzodiazinan-4-one (**2b**) is not only a potent, but also a highly selective, COX-2 inhibitor.

Experimental

General

All chemicals and solvents used in this study were purchased from Merck AG and Aldrich Chemical. Melting points were determined with a Thomas-Hoover capillary apparatus. Infrared spectra were acquired using a Perkin Elmer Model 1420 spectrometer. A Bruker FT-500 MHz instrument (Bruker Biosciences, USA) was used to acquire ¹HNMR spectra with TMS as internal standard. Chloroform-D and DMSO-D₆ were used as solvents. Coupling constant (*J*) values are estimated in Hertz (Hz) and spin multiples are given as s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet), and br (broad). The mass spectral measurements were performed on an 6410 Agilent LCMS triple quadrupole mass spectrometer (LCMS) with an electrospray ionization (ESI) interface. Microanalyses, determined for C and H, were within ± 0.4% of theoretical values.

General procedure for preparation of 2,3-diaryl-1,3-benzodiazinan-4-one (**1a–e** and **3a–e**)

Alum (0.2 g) was added to a mixture of isatoic anhydride (1 mmol), aldehyde (1 mmol) and primary amine (1.1 mmol) in ethanol. The mixture was heated at reflux for 1 h. After completion of the reaction, the solid 1,3-benzodiazinan (**1** or **3**) was separated, washed with water and recrystallized from ethanol (yields: 14–30%). The physical and spectral data for **1a–e** and **3a–e** are listed below.

2-(4-Methylthiophenyl)-3-phenyl-1,3-benzodiazinan-4-one (**1a**)

Yield, 20%; white crystalline powder, mp 202–203°C; IR (KBr): ν cm⁻¹ 3307 (NH), 1660 (C=O); LCMS (ESI): 347.1 (M+1)⁺; ¹HNMR (CDCl₃) δ ppm 2.48 (s, 3H, SMe), 6.09 (s, 1H, CH), 6.67 (d, 1H, benzodiazinan H₈, *J* = 7.9 Hz), 6.94 (m, 1H, benzodiazinan H₆), 7.16 (d, 2H, 4-methylthiophenyl H₃ and H₅, *J* = 8.3 Hz), 7.22 (m, 1H, benzodiazinan H₇), 7.23 (d, 2H, 4-methylthiophenyl H₂ & H₆, *J* = 8.3 Hz), 7.30–7.36 (m, 5H, phenyl), 8.06 (d, 1H, benzodiazinan H₅, *J* = 7.7 Hz); Anal. Calcd. for C₂₁H₁₈N₂OS: C, 72.80; H, 5.24; N, 8.09. Found: C, 72.99; H, 5.62; N, 8.31.

3-(4-Fluorophenyl)-2-(4-methylthiophenyl)-1,3-benzodiazinan-4-one (**1b**)

Yield, 22%; white crystalline powder, mp 242–243°C; IR (KBr): ν cm⁻¹ 3312 (NH), 1642 (C=O); LCMS (ESI): 365.1 (M+1)⁺; ¹HNMR (CDCl₃) δ ppm 2.49 (s, 3H, SMe), 6.07 (s, 1H, CH), 6.69 (d, 1H, benzodiazinan H₈, *J* = 7.9 Hz), 6.96 (m, 1H, benzodiazinan H₆), 7.02 (t, 2H, 4-fluorophenyl H₃ & H₅, *J* = 8.3 Hz), 7.15 (d, 2H, 4-methylthiophenyl H₃ & H₅, *J* = 8.3 Hz), 7.18 (d, 4-methylthiophenyl H₂ & H₆, *J* = 8.3 Hz), 7.28 (m, 1H, benzodiazinan H₇), 7.42 (d, 2H, fluorophenyl H₂ & H₆, *J* = 8.7 Hz), 8.05 (d, 1H, benzodiazinan H₅, *J* = 7.7 Hz); Anal. Calcd. for C₂₁H₁₇FN₂OS: C, 69.21; H, 4.70; N, 7.69. Found: C, 69.39; H, 4.42; N, 7.88.

3-(4-Chlorophenyl)-2-(4-methylthiophenyl)-1,3-benzodiazinan-4-one (**1c**)

Yield, 20%; white crystalline powder, mp 230–232°C; IR (KBr): ν cm⁻¹ 3314 (NH), 1640 (C=O); LCMS (ESI): 381.1 (M+1)⁺; ¹HNMR (CDCl₃) δ ppm 2.34 (s, 3H, SMe), 5.96 (s, 1H, CH), 6.03 (s, 1H, NH),

6.61 (d, 1H, benzodiazinan H₈, *J* = 8.0 Hz), 6.72 (m, 1H, benzodiazinan H₆), 7.02 (d, 2H, 4-chlorophenyl H₂ & H₆, *J* = 8.3 Hz), 7.08 (d, 2H, 4-methylthiophenyl H₃ & H₅, *J* = 8.3 Hz), 7.15 (d, 4-methyl thiophenyl H₂ and H₆, *J* = 8.3 Hz), 7.19 (d, 2H, 4-chlorophenyl H₃ and H₅, *J* = 8.7 Hz), 7.21 (m, 1H, benzodiazinan H₇), 7.83 (d, 1H, benzodiazinan H₅, *J* = 7.7 Hz); Anal. Calcd. for C₂₁H₁₇ClN₂OS: C, 66.22; H, 4.50; N, 7.35. Found: C, 66.49; H, 4.22; N, 7.68.

3-(4-Methylphenyl)-2-(4-methylthiophenyl)-1,3-benzodiazinan-4-one (**1d**)

Yield, 20%; white crystalline powder, mp 217–218°C; IR (KBr): ν cm⁻¹ 3298 (NH), 1633 (C=O); LCMS (ESI): 361.1 (M+1)⁺; ¹HNMR (CDCl₃) δ ppm 2.27 (s, 3H, Me), 2.41 (s, 3H, SMe), 6.19 (s, 1H, CH), 6.72 (d, 2H, 4-methylphenyl H₃ & H₅, *J* = 7.8 Hz), 7.14–7.16 (m, 3H, 4-methylphenyl H₂ and H₆ & benzodiazinan H₈), 7.18 (d, 2H, 4-methylthiophenyl H₃ & H₅, *J* = 8.4 Hz), 7.25 (m, 1H, benzodiazinan H₆), 7.30 (d, 4-methylthiophenyl H₂ & H₆, *J* = 8.4 Hz), 7.55 (m, 1H, benzodiazinan H₇), 7.70 (d, 1H, benzodiazinan H₅, *J* = 7.7 Hz); Anal. Calcd. for C₂₂H₂₀N₂O₂S: C, 73.30; H, 5.59; N, 7.77. Found: C, 72.99; H, 5.32; N, 7.90.

3-(4-Methoxyphenyl)-2-(4-methylthiophenyl)-1,3-benzodiazinan-4-one (**1e**)

Yield, 30%; white crystalline powder, mp 248–250°C; IR (KBr): ν cm⁻¹ 3295 (NH), 1632 (C=O); LCMS (ESI): 377.1 (M+1)⁺; ¹HNMR (CDCl₃) δ ppm 2.36 (s, 3H, SMe), 3.72 (s, 3H, OMe), 6.16 (s, 1H, CH), 6.74 (d, 2H, 4-methoxyphenyl H₃ & H₅, *J* = 7.9 Hz), 6.88 (d, 2H, 4-methylthiophenyl H₃ & H₅, *J* = 8.4 Hz), 7.17 (d, 2H, 4-methoxyphenyl H₂ & H₆, *J* = 7.9 Hz), 7.25 (m, 1H, benzodiazinan H₈), 7.25 (m, 1H, benzodiazinan H₆), 7.29 (d, 4-methylthiophenyl H₂ & H₆, *J* = 8.4 Hz), 7.50 (m, 1H, benzodiazinan H₇), 7.71 (d, 1H, benzodiazinan H₅, *J* = 7.7 Hz); Anal. Calcd. for C₂₂H₂₀N₂O₂S: C, 70.19; H, 5.35; N, 7.44. Found: C, 70.39; H, 5.21; N, 7.53.

3-(4-Methylthiophenyl)-2-phenyl-1,3-benzodiazinan-4-one (**3a**)

Yield, 24%; white crystalline powder, mp 193.5–194.5°C; IR (KBr): ν cm⁻¹ 3299 (NH), 1638 (C=O); LCMS (ESI): 347.9 (M+1)⁺, 369.8 (M+23)⁺; ¹HNMR (CDCl₃) δ ppm 2.47 (s, 3H, SMe), 4.78 (s, 1H, NH), 6.12 (s, 1H, CH), 6.67 (d, 1H, benzodiazinan H₈, *J* = 8 Hz), 6.93 (m, 1H, benzodiazinan H₆), 7.14 (d, 2H, methylthiophenyl H₃ & H₅, *J* = 8.5 Hz), 7.2 (d, 2H, methylthiophenyl H₂ & H₆, *J* = 8.5 Hz), 7.31–7.40 (m, 5H, phenyl H₂ & H₆ & 1H, benzodiazinan H₇), 8.06 (d, 1H, benzodiazinan H₅, *J* = 7.7 Hz); Anal. Calcd. for C₂₁H₁₈N₂O₂S: C, 72.80; H, 5.24; N, 8.09. Found: C, 73.03; H, 5.42; N, 8.38.

2-(4-Fluorophenyl)-3-(4-methylthiophenyl)-1,3-benzodiazinan-4-one (**3b**)

Yield, 14%; white crystalline powder, mp 191–192°C; IR (KBr): ν cm⁻¹ 3296 (NH), 1638 (C=O); LCMS (ESI): 365.8 (M+1)⁺, 387.7 (M+23)⁺; ¹HNMR (CDCl₃) δ 2.47 (s, 3H, SMe), 6.12 (s, 1H, CH), 6.69 (d, 1H, benzodiazinan H₈, *J* = 8 Hz), 6.95 (m, 1H, benzodiazinan H₆), 6.99 (m, 2H, 4-fluorophenyl H₂ & H₆), 7.1 (d, 2H, methylthiophenyl H₃ & H₅, *J* = 8.6 Hz), 7.2 (d, 2H, methylthiophenyl H₂ & H₆, *J* = 8.6 Hz), 7.35–7.39 (m, 3H, 4-fluorophenyl H₃ & H₅ & 1H, benzodiazinan H₅); Anal. Calcd. for C₂₁H₁₇FN₂O₂S: C, 69.21; H, 4.70; N, 7.69. Found: C, 69.43; H, 4.99; N, 7.96.

2-(4-Chlorophenyl)-3-(4-methylthiophenyl)-1,3-benzodiazinan-4-one (**3c**)

Yield, 15%; white crystalline powder, mp 238.5–240°C; IR (KBr): ν cm⁻¹ 3328 (NH), 1645 (C=O); LCMS (ESI): 403.6 (M+23)⁺; ¹HNMR (CDCl₃) δ ppm 2.49 (s, 3H, SMe), 6.11 (s, 1H, CH), 6.69 (d, 1H, benzodiazinan H₈, *J* = 7.9 Hz), 6.97 (m, 1H, benzodiazinan H₆), 7.13 (d, 2H, methylthiophenyl H₃ & H₅, *J* = 8.6 Hz), 7.21 (d, 2H, methylthiophenyl H₂ & H₆, *J* = 8.6 Hz), 7.29 (d, 2H, 4-chlorophenyl H₂ & H₆, *J* = 8.2 Hz), 7.32 (d, 2H, 4-chlorophenyl H₃ & H₅, *J* = 8.2 Hz), 7.36 (m, 1H, benzodiazinan H₇), 8.06 (d, 1H, benzodiazinan H₅, *J* = 7.8 Hz); Anal. Calcd. for C₂₁H₁₇ClN₂O₂S: C, 66.22; H, 4.50; N, 7.35. Found: C, 65.95; H, 4.75; N, 7.61.

2-(4-Methylphenyl)-3-(4-methylthiophenyl)-1,3-benzodiazinan-4-one (**3d**)

Yield, 20%; white crystalline powder, mp 217–218°C; IR (KBr): ν cm⁻¹ 3298 (NH), 1633 (C=O); LCMS (ESI): 361.1 (M+1)⁺; ¹HNMR (CDCl₃) δ ppm 2.27 (s, 3H, Me), 2.41 (s, 3H, SMe), 6.19 (s, 1H, CH), 6.72 (d, 2H, 4-methylphenyl H₃ & H₅, *J* = 7.8 Hz), 7.14–7.16 (m, 3H, 4-methylphenyl H₂ and H₆ & benzodiazinan H₈), 7.18 (d, 2H, 4-methylthiophenyl H₃ & H₅, *J* = 8.4 Hz), 7.25 (m, 1H, benzodiazinan H₆), 7.30 (d, 4-methylthiophenyl H₂ & H₆, *J* = 8.4 Hz), 7.55 (m, 1H, benzodiazinan H₇), 7.70 (d, 1H, benzodiazinan H₅, *J* = 7.7 Hz); Anal. Calcd. for C₂₂H₂₀N₂O₂S: C, 73.30; H, 5.59; N, 7.77. Found: C, 73.59; H, 5.88; N, 7.60.

2-(4-Methoxyphenyl)-3-(4-methylthiophenyl)-1,3-benzodiazinan-4-one (**3e**)

Yield, 15%; white crystalline powder, mp 199–201°C; IR (KBr): ν cm⁻¹ 3307 (NH), 1638 (C=O); LCMS (ESI): 377.4 (M+1)⁺, 399.3 (M+23)⁺; ¹HNMR (DMSO-d₆) δ ppm 2.46 (s, 3H, SMe), 3.68 (s, 3H, OMe), 6.19 (d, 1H, CH, *J* = 2.1 Hz), 6.58 (m, 1H, benzodiazinan H₈), 6.71 (m, 1H, benzodiazinan H₆), 6.74 (d, 2H, 4-methoxyphenyl H₃ & H₅, *J* = 7.9 Hz), 6.85 (d, 2H, methylthiophenyl H₃ & H₅, *J* = 8.2 Hz), 7.24 (d, 2H, 4-methoxyphenyl H₂ & H₆, *J* = 7.9 Hz), 7.29 (d, 2H, methylthiophenyl H₂ & H₆, *J* = 8.2 Hz), 7.51 (d, 1H, NH, *J* = 2.1 Hz) 7.59 (m, 1H, benzodiazinan H₇), 7.71 (d, 1H, benzodiazinan H₅, *J* = 7.7 Hz); Anal. Calcd. for C₂₂H₂₀N₂O₂S: C, 70.19; H, 5.35; N, 7.44. Found: C, 70.41; H, 5.66; N, 7.71.

General procedure for preparation of aryl-4-methylsulfonylphenyl-1,3-benzodiazinan-4-one (**2a–e** and **4a–e**)

0.5 g of **1** or **3** was dissolved in 10 mL methanol and 5 mL 30% hydrogen peroxide solution and 100 mg WO₃ as catalyst were added. The mixture was stirred at room temperature for 5–6 hours, after evaporation of methanol, the mixture was extracted with ethyl acetate and dried with anhydrous sodium sulfate and then evaporated, the product was recrystallized in ethanol (yields: 22–36%). The physical and spectral data for **2a–e** and **4a–e** are listed below.

2-(4-Methylsulfonylphenyl)-3-phenyl-1,3-benzodiazinan-4-one (**2a**)

Yield, 30%; yellow crystalline powder, mp 175–176°C; IR (KBr): ν cm⁻¹ 3298 (NH), 1645 (C=O), 1320, 1165 (SO₂); LCMS (ESI): 379.1 (M+1)⁺; ¹HNMR (CDCl₃) δ ppm 3.03 (s, 3H, SO₂CH₃),

6.24 (s, 1H, CH), 6.73 (d, 1H, benzodiazinan H₈, *J* = 8.4 Hz), 6.98 (t, 1H, benzodiazinan H₆, *J* = 7.5 Hz), 7.25 (d, 2H, phenyl H₃ & H₅, *J* = 7.9 Hz), 7.28 (m, 1H, benzodiazinan H₇), 7.35–7.40 (m, 3H, phenyl H₃–H₅), 7.62 (d, 2H, 4-methylsulfonylphenyl H₂ & H₆, *J* = 8.0 Hz), 7.89 (d, 2H, 4-methylsulfonylphenyl H₃ & H₅, *J* = 8.0 Hz), 8.07 (d, 1H, benzodiazinan H₅, *J* = 8.0 Hz); Anal. Calcd. for C₂₁H₁₈N₂O₃S: C, 66.65; H, 4.79; N, 7.40. Found: C, 66.34; H, 4.62; N, 7.29.

3-(4-Fluorophenyl)-2-(4-methylsulfonylphenyl)-1,3-benzodiazinan-4-one (2b)

Yield, 35%; cream crystalline powder, mp 215–217°C; IR (KBr): ν cm⁻¹ 3400 (NH), 1610 (C=O), 1300, 1150 (SO₂); LCMS (ESI): 397.1 (M+1)⁺; ¹HNMR (CDCl₃) δ ppm 3.04 (s, 3H, SO₂CH₃), 4.9 (s, 1H, NH), 6.20 (s, 1H, CH), 6.72 (d, 1H, benzodiazinan H₈, *J* = 8.0 Hz), 6.98 (t, 1H, benzodiazinan H₆, *J* = 7.5 Hz), 7.04 (t, 2H, 4-fluorophenyl H₃ & H₅, *J* = 8.4 Hz), 7.21 (dd, 2H, 4-fluorophenyl H₂ & H₆), 7.40 (m, 1H, benzodiazinan H₇), 7.61 (d, 2H, 4-methylsulfonylphenyl H₂ & H₆, *J* = 8.2 Hz), 7.89 (d, 2H, 4-methylsulfonylphenyl H₃ & H₅, *J* = 8.2 Hz), 8.07 (d, 1H, benzodiazinan H₅, *J* = 6.7 Hz); Anal. Calcd. for C₂₁H₁₇FN₂O₃S: C, 63.62; H, 4.32; N, 7.07. Found: C, 63.48; H, 4.12; N, 7.22.

3-(4-Chlorophenyl)-2-(4-methylsulfonylphenyl)-1,3-benzodiazinan-4-one (2c)

Yield, 31%; cream crystalline powder, mp 191–193°C; IR (KBr): ν cm⁻¹ 3315 (NH), 1648 (C=O), 1322, 1159 (SO₂); LCMS (ESI): 413.1 (M+1)⁺; ¹HNMR (CDCl₃) δ ppm 3.20 (s, 3H, SO₂CH₃), 6.39 (s, 1H, CH), 6.72 (d, 1H, benzodiazinan H₈, *J* = 8.1 Hz), 6.78 (t, 1H, benzodiazinan H₆, *J* = 8.3 Hz), 7.12 (d, 2H, 4-chlorophenyl H₂ & H₆, *J* = 8.1 Hz), 7.25 (d, 2H, 4-chlorophenyl H₃ & H₅, *J* = 8.1 Hz), 7.30 (m, 1H, benzodiazinan H₇), 7.56 (d, 2H, 4-methylsulfonylphenyl H₂ & H₆, *J* = 8.7 Hz), 7.80 (d, 1H, benzodiazinan H₅, *J* = 6.7 Hz), 7.87 (d, 2H, 4-methylsulfonylphenyl H₃ & H₅, *J* = 8.7 Hz); Anal. Calcd. for C₂₁H₁₇ClN₂O₃S: C, 61.09; H, 4.15; N, 6.78. Found: C, 59.87; H, 4.36; N, 6.88.

3-(4-Methylphenyl)-2-(4-methylsulfonylphenyl)-1,3-benzodiazinan-4-one (2d)

Yield, 31%; cream crystalline powder, mp 181–182°C; IR (KBr): ν cm⁻¹ 3312 (NH), 1632 (C=O), 1315, 1151 (SO₂); LCMS (ESI): 393.1 (M+1)⁺; ¹HNMR (CDCl₃) δ ppm 2.33 (s, 3H, CH₃), 3.04 (s, 3H, SO₂CH₃), 6.38 (s, 1H, CH), 6.69 (d, 1H, benzodiazinan H₈, *J* = 8.1 Hz), 6.95 (t, 1H, benzodiazinan H₆, *J* = 7.5 Hz), 7.12 (d, 2H, 4-methylphenyl H₃ & H₅, *J* = 8.4 Hz), 7.16 (d, 2H, 4-methylphenyl H₂ & H₆, *J* = 8.4 Hz), 7.36 (m, 1H, benzodiazinan H₇), 7.59 (d, 2H, 4-methylsulfonylphenyl H₂ & H₆, *J* = 8.3 Hz), 7.87 (d, 2H, 4-methylsulfonylphenyl H₃ & H₅, *J* = 8.7 Hz), 8.04 (d, 1H, benzodiazinan H₅, *J* = 6.6 Hz); Anal. Calcd. for C₂₂H₂₀N₂O₃S: C, 67.33; H, 5.14; N, 7.14. Found: C, 67.30; H, 5.26; N, 7.22.

3-(4-Methoxyphenyl)-2-(4-methylsulfonylphenyl)-1,3-benzodiazinan-4-one (2e)

Yield, 30%; cream crystalline powder, mp 218–220°C; IR (KBr): ν cm⁻¹ 3298 (NH), 1628 (C=O), 1320, 1152 (SO₂); LCMS (ESI): 409.1 (M+1)⁺; ¹HNMR (CDCl₃) δ ppm 3.05 (s, 3H, SO₂CH₃), 3.66 (s, 3H, OCH₃), 6.18 (s, 1H, CH), 6.86 (d, 2H, 4-methoxyphenyl H₃ & H₅, *J* = 8.9 Hz), 6.99 (d, 1H, benzodiazinan H₈, *J* = 8.1 Hz), 7.15 (d, 2H, 4-methoxyphenyl H₂ & H₆, *J* = 8.9 Hz), 7.36 (t, 1H, benzodiazinan H₆, *J* = 7.3 Hz), 7.57 (m, 1H, benzodiazinan H₇), 7.60

(d, 2H, 4-methylsulfonylphenyl H₃ & H₅, *J* = 8.3 Hz), 7.89 (d, 2H, 4-methylsulfonylphenyl H₂ & H₆, *J* = 8.7 Hz), 8.07 (d, 1H, benzodiazinan H₅, *J* = 6.1 Hz); Anal. Calcd. for C₂₂H₂₀N₂O₄S: C, 64.59; H, 4.93; N, 6.86. Found: C, 64.77; H, 4.74; N, 6.95.

3-(4-Methylsulfonylphenyl)-2-phenyl-1,3-benzodiazinan-4-one (4a)

Yield, 34%; yellow crystalline powder, mp 208–209°C; IR (KBr): ν cm⁻¹ 3327 (NH), 1647 (C=O), 1326, 1162 (SO₂); LCMS (ESI): 401.7 (M+23)⁺; ¹HNMR (CDCl₃) δ ppm 3.21 (s, 3H, SO₂CH₃), 6.26 (s, 1H, CH), 6.37 (m, 1H, benzodiazinan H₆), 6.80 (d, 1H, benzodiazinan H₈, *J* = 8.1 Hz), 7.29–7.32 (m, 3H, phenyl H₃–H₅), 7.35 (m, 1H, benzodiazinan H₇), 7.39 (d, 2H, phenyl H₂ & H₆, *J* = 7.6 Hz), 7.58 (d, 2H, 4-methylsulfonylphenyl H₂ & H₆, *J* = 8.7 Hz), 7.83 (d, 1H, benzodiazinan H₅, *J* = 7.8 Hz), 7.87 (d, 1H, NH, *J* = 2.9 Hz), 7.90 (d, 2H, 4-methylsulfonylphenyl H₃ & H₅, *J* = 8.7 Hz); Anal. Calcd. for C₂₁H₁₈N₂O₃S: C, 66.65; H, 4.79; N, 7.40. Found: C, 66.29; H, 4.51; N, 7.32.

2-(4-Fluorophenyl)-3-(4-methylsulfonylphenyl)-1,3-benzodiazinan-4-one (4b)

Yield, 27%; yellow crystalline powder, mp 170–171°C; IR (KBr): ν cm⁻¹ 3369 (NH), 1666 (C=O), 1298, 1140 (SO₂); LCMS (ESI): 419.3 (M+23)⁺; ¹HNMR (CDCl₃) δ ppm 3.21 (s, 3H, SO₂CH₃), 6.48 (s, 1H, CH), 6.73 (m, 1H, benzodiazinan H₆), 6.8 (d, 1H, benzodiazinan H₈, *J* = 8.1 Hz), 7.16 (m, 2H, 4-fluorophenyl H₃ & H₅), 7.31 (m, 1H, benzodiazinan H₇), 7.42 (dd, 2H, 4-fluorophenyl H₂ & H₆), 7.56 (d, 2H, 4-methylsulfonylphenyl H₂ & H₆, *J* = 8.7 Hz), 7.74 (d, 1H, benzodiazinan H₅, *J* = 7.8 Hz), 7.84 (d, 1H, NH, *J* = 2.9 Hz), 7.89 (d, 2H, 4-methylsulfonylphenyl H₃ & H₅, *J* = 8.7 Hz); Anal. Calcd. for C₂₁H₁₇FN₂O₃S: C, 63.62; H, 4.32; N, 7.07. Found: C, 63.29; H, 4.22; N, 7.12.

2-(4-Chlorophenyl)-3-(4-methylsulfonylphenyl)-1,3-benzodiazinan-4-one (4c)

Yield, 22%; yellow crystalline powder, mp 220–221°C; IR (KBr): ν cm⁻¹ 3338 (NH), 1650 (C=O), 1294, 1147 (SO₂); LCMS (ESI): 413.1 (M+1)⁺, 435.1 (M+23)⁺; ¹HNMR (CDCl₃) δ ppm 3.21 (s, 3H, SO₂Me), 6.48 (d, 1H, CH, *J* = 3.1 Hz), 6.74 (m, 1H, benzodiazinan H₆), 6.8 (d, 1H, benzodiazinan H₈, *J* = 8.3 Hz), 7.31 (m, 1H, benzodiazinan H₇), 7.4 (m, 4H, 4-chlorophenyl H₂–H₆), 7.57 (d, 2H, 4-methyl sulfonylphenyl H₂ & H₆, *J* = 8.7 Hz), 7.74 (d, 1H, benzodiazinan H₅, *J* = 7.8 Hz), 7.88 (d, 1H, NH, *J* = 3.1 Hz), 7.9 (d, 2H, 4-methylsulfonylphenyl H₃ & H₅, *J* = 8.7 Hz); Anal. Calcd. for C₂₁H₁₇ClN₂O₃S: C, 61.09; H, 4.15; N, 6.78. Found: C, 61.29; H, 4.29; N, 6.84.

2-(4-Methylphenyl)-3-(4-methylsulfonylphenyl)-1,3-benzodiazinan-4-one (4d)

Yield, 24%; yellow crystalline powder, mp 177–179°C; IR (KBr): ν cm⁻¹ 3038 (NH), 1670 (C=O), 1326, 1158 (SO₂); LCMS (ESI): 415.6 (M+23)⁺; ¹HNMR (CDCl₃) δ ppm 2.22 (s, 3H, Me), 3.2 (s, 3H, SO₂Me), 6.39 (d, 1H, CH, *J* = 2.9 Hz), 6.72 (m, 1H, benzodiazinan H₆), 6.78 (d, 1H, benzodiazinan H₈, *J* = 8.1 Hz), 7.12 (d, 2H, 4-methylphenyl H₃ & H₅, *J* = 8.0 Hz), 7.26 (d, 2H, 4-methylphenyl H₂ & H₆, *J* = 8.0 Hz), 7.99 (m, 1H, benzodiazinan H₇), 7.56 (d, 2H, 4-methylsulfonylphenyl H₂ & H₆, *J* = 8.7 Hz), 7.73 (d, 1H, benzodiazinan H₅, *J* = 7.7 Hz), 7.8 (d, 1H, NH, *J* = 2.9 Hz), 7.88 (d, 2H, 4-methylsulfonyl phenyl H₃ & H₅,

$J = 8.7$ Hz); Anal. Calcd. for $C_{22}H_{20}N_2O_3S$: C, 67.33; H, 5.14; N, 7.14. Found: C, 67.19; H, 5.31; N, 7.18.

2-(4-Methoxyphenyl)-3-(4-methylsulfonylphenyl)-1,3-benzodiazinan-4-one (4e)

Yield, 29%; yellow crystalline powder, mp 175–176°C; IR (KBr): ν cm^{-1} 3313 (NH), 1632 (C=O), 1323, 1152 (SO₂); LCMS (ESI): 409.6 (M+1)⁺, 431.5 (M+23)⁺; ¹HNMR (CDCl₃) δ ppm 3.2 (s, 3H, SO₂Me), 3.68 (s, 3H, OMe), 6.38 (d, 1H, CH, $J = 2.7$ Hz), 6.72 (m, 1H, benzodiazinan H₆), 6.78 (d, 1H, benzodiazinan H₈, $J = 8.1$ Hz), 6.87 (d, 2H, 4-methoxyphenyl H₃ & H₅, $J = 8.7$ Hz) 7.29 (d, 2H, 4-methoxyphenyl H₂ & H₆, $J = 8.6$ Hz), 7.3 (m, 1H, benzodiazinan H₇), 7.55 (d, 2H, 4-methylsulfonyl phenyl H₂ & H₆, $J = 8.7$ Hz), 7.73 (d, 1H, benzodiazinan H₅, $J = 7.7$ Hz), 7.76 (d, 1H, NH, $J = 2.7$ Hz), 7.88 (d, 2H, 4-methylsulfonylphenyl H₃ & H₅, $J = 8.7$ Hz); Anal. Calcd. for $C_{22}H_{20}N_2O_4S$: C, 64.59; H, 4.93; N, 6.86. Found: C, 64.39; H, 4.71; N, 6.88.

In vitro cyclooxygenase (COX) inhibition assays

The assay was performed using an enzyme chemiluminescent kit (Cayman Chemical, MI, USA) according to our previously reported method [17]. The Cayman chemical chemiluminescent COX (ovine) inhibitor screening assay utilizes the heme-catalyzed hydroperoxidase activity of ovine cyclooxygenases to generate luminescence in the presence of a cyclic naphthalene hydrazide and the substrate arachidonic acid. Arachidonate-induced luminescence was shown to be an index of real-time catalytic activity and demonstrated the turnover inactivation of the enzyme. Inhibition of COX activity, measured by luminescence, by a variety of selective and nonselective inhibitors showed potencies similar to those observed with other *in vitro* and whole cell methods.

Molecular modeling (docking) studies

Docking studies were performed using Autodock software Version 3.0. The coordinates of the X-ray crystal structure of the selective COX-2 inhibitor SC-558 bound to the murine COX-2 enzyme was obtained from the RCSB Protein Data Bank (1cx2) and hydrogens were added. The ligand molecules were constructed using the Builder module and were energy minimized for 1000 iterations reaching a convergence of 0.01 kcal/mol Å. The energy minimized ligands were superimposed on SC-558 in the PDB file 1cx2 after which SC-558 was deleted. The purpose of docking is to search for favorable binding configuration between the small flexible ligands and the rigid protein. Protein residues with atoms greater than 7.5 Å from the docking box were removed for efficiency. These docked structures were very similar to the minimized structures obtained initially. The quality of the docked structures was evaluated by measuring the intermolecular energy of the ligand-enzyme assembly [19, 20].

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