



A modular olefination reaction between aldehydes and diborylsilylmethide lithium salts†

Oriol Salvado and Elena Fernández *

Cite this: *Chem. Commun.*, 2021, 57, 6300

Received 8th April 2021,
Accepted 24th May 2021

DOI: 10.1039/d1cc01882e

rsc.li/chemcomm

We describe the preparation of densely functionalised 1,1-silylborylated trisubstituted alkenes, via a boron-Wittig reaction, between LiC(Bpin)₂(SiMe₃) and aliphatic or aromatic aldehydes. The condensation of diborylsilylmethide lithium salts with α,β -unsaturated aldehydes provides a direct pathway to synthesize 1,1-silylborylated conjugated dienes and diynes.

The condensation reaction between carbonyl substrates and α -boryl carbanions, appointed as a boron-Wittig reaction, is considered of great synthetic utility for the preparation of valuable alkenes.¹ The renaissance of the boron-Wittig reaction in 2010, has firmly consolidated the benefits of this type of condensation reaction using stable pinacolboryl (Bpin) moieties in the α -boryl carbanions, but also controlling the stereoselectivity of the alkenes formed through the *syn*-B–O elimination (Scheme 1a)² depending on the substrates^{3a} or additives^{3b} (Scheme 1b).

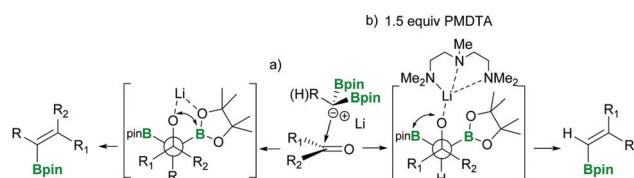
The synthesis of trisubstituted vinyl boronates, via boron-Wittig reaction between geminal bis(boronates) and linear or α -branched aldehydes, was addressed by Morken and co-workers, with relative control over the selectivity (Scheme 2a).⁴ As a trend, they observed that the *E* isomer can be favoured when either bulky diborylated reagents or large aldehyde substituents (R') are employed. In contrast, when small aldehydes (*i.e.*, linear alkyl) and small diborylated species are employed, the *Z* isomer seems to be favoured. However, aromatic aldehydes were not studied in that condensation reaction.

Attending to the challenging boron-Wittig reaction when aldehydes are involved, we became interested to launch a systematic study on the olefination with both aromatic and aliphatic aldehydes in the presence of the reagent HC(Bpin)₂(SiMe₃) (1) with the aim to get insights about the stereoselectivity on the formation of densely functionalised 1,1-silylborylated trisubstituted alkenes (Scheme 2b).

The introduction of silyl groups in the polyborylated reagent can be used as an alternative strategy to control the stereoselectivity on the boron Wittig reactions, as well as a direct methodology to introduce an extra functionality, however to the best of our knowledge only two approaches have been reported to date.^{5,6}

The reagent HC(Bpin)₂(SiMe₃) (1) was deprotonated in the presence of LiTMP, to form a diborylsilyl stabilised carbanion at 0 °C. As a model substrate, benzaldehyde was added to the *in situ* prepared LiC(Bpin)₂(SiMe₃) in THF, at 0 °C. The reaction was warmed to room temperature and stirred for 16 h to accomplish the formation of the 1,1-silylborylated trisubstituted alkene 2, proving that B–O elimination is favoured *versus* Si–O Peterson elimination.⁷ The selectivity observed shows a preference for the 2-*E* stereoisomer with the Bpin moiety *syn* with respect to the aryl group, suggesting a favoured intermediate **A** *versus* **A'** (Scheme 3a). This is in contrast to the trend observed by Morken and co-workers, on the boron-Wittig olefination between bis(pinacolboryl)methane and aldehydes, in the presence of LiTMP, to furnish *trans*-vinylboronate esters, through a plausible favoured **B'** intermediate (Scheme 3b).⁴ Despite the fact that the *E*:*Z* ratio for 2 is modest, it is noticeable to mention that the 2-*E* isomer has never been isolated before. In fact, previous synthetic attempts, based on the hydroboration of 1-phenyl-2-trimethylsilylacetylene, provided only the 2-*Z* isomer as an expected B–H *syn* addition to the triple bond.⁸

We next explored the condensation of picolinaldehyde with 1/LiTMP and the observed *E*:*Z* ratio on the trisubstituted alkene 3 increased slightly in THF or CPME as solvent (Scheme 4a), being the first synthetic approach towards the 3-*E* since the 3-*Z* was prepared

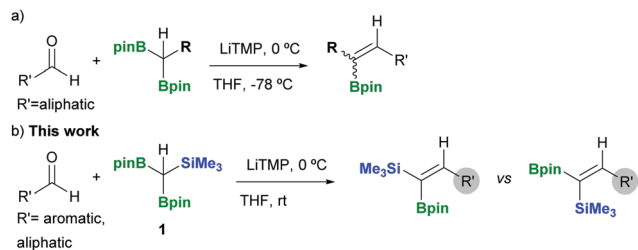


Scheme 1 Boron-Wittig reaction of ketones for tri- and tetrasubstituted alkenes.

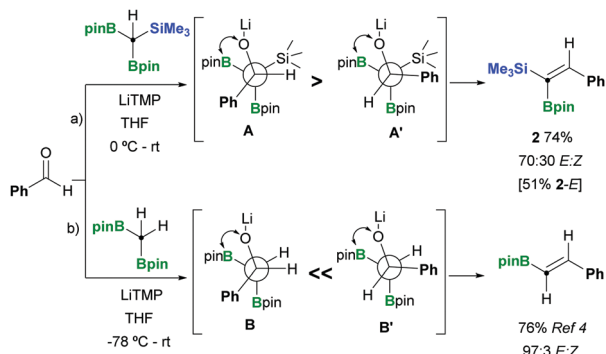
Dept. Química Física i Inorgànica, University Rovira i Virgili, Tarragona, Spain.
E-mail: mariaelena.fernandez@urv.cat

† Electronic supplementary information (ESI) available. CCDC 2060900. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/d1cc01882e

Communication

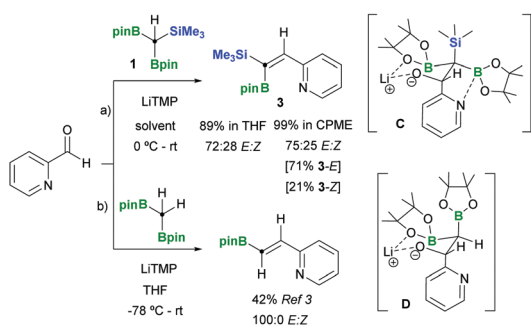


Scheme 2 Boron-Wittig reaction of aldehydes with (a) $\text{HC}(\text{Bpin})_2(\text{R})$ and (b) $\text{HC}(\text{Bpin})_2(\text{SiMe}_3)$ for trisubstituted alkene synthesis.

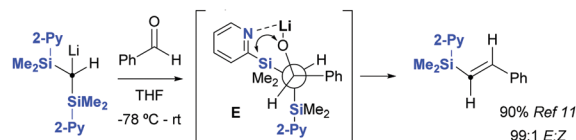


Scheme 3 Boron-Wittig reaction between benzaldehyde and (a) $\text{LiC}(\text{Bpin})_2(\text{SiMe}_3)$ and (b) $\text{LiCH}(\text{Bpin})_2$ for comparison.

from catalytic hydroboration strategies.^{8c} The stereocontrol with $\text{LiC}(\text{Bpin})_2(\text{SiMe}_3)$ contrasts with the one observed using $\text{LiCH}(\text{Bpin})_2$, since the *trans*-vinylboronate ester is generated exclusively in 42% yield (Scheme 4b).³ Considering the plausible chair-like intermediate, an intramolecular interaction between the N and the Bpin moiety, forming a five member ring in C, might explain the observed enriched selectivity on 3-*E*.⁹ In fact, product 3-*Z* shows a characteristic ¹¹B NMR signal at 32.5 ppm, whereas for 3-*E* the signal appears at 27.6 ppm as a consequence of the subtle N-B intramolecular interaction. Scheme 4b shows that in the absence of chelation control, the intermediate D might predict the *trans*-vinylboronate ester formation.¹⁰ For comparison, Yoshida and co-workers found that the reagent $\text{LiCH}(\text{SiMe}_2(2\text{-Py}))_2$ reacts with aldehydes to give the corresponding *trans*-vinylsilanes presumably *via* stereodetermining intermediate E where the intramolecular chelation of the pyridyl group with Li^+ locks the conformation



Scheme 4 Boron-Wittig reaction between picolinealdehyde and (a) $\text{LiC}(\text{Bpin})_2(\text{SiMe}_3)$ and (b) $\text{LiCH}(\text{Bpin})_2$ for comparison.

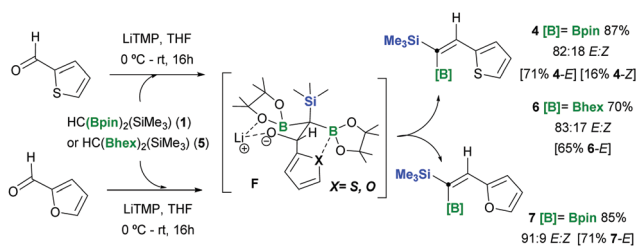


Scheme 5 Boron-Wittig reaction between benzaldehyde and $\text{LiCH}(\text{SiMe}_2)(2\text{-Py})_2$.

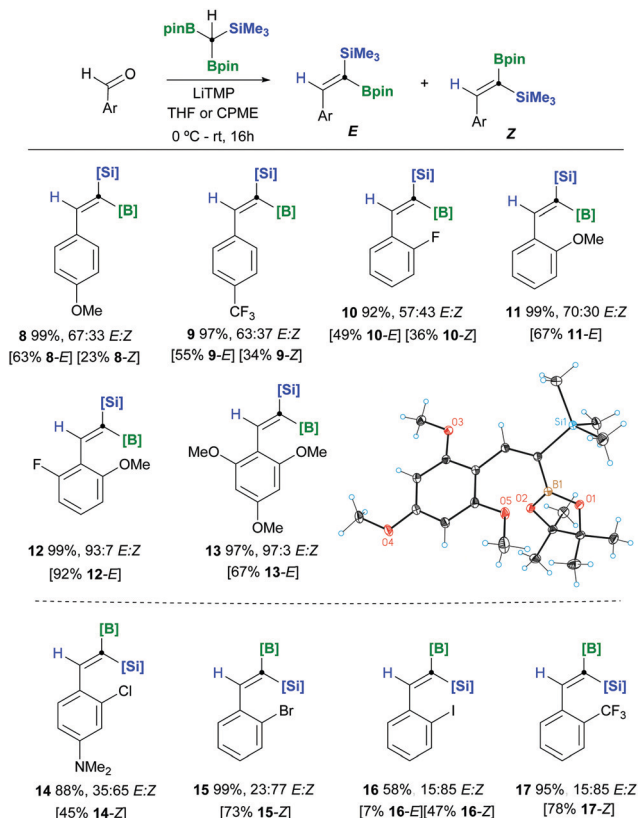
(Scheme 5).¹¹ That is an interesting point since the alternative reaction between benzaldehyde and the reagent $\text{LiCH}(\text{SiMe}_3)_2$ produces the vinylsilanes only in *E:Z* ratio = 58:42.^{12,13}

Next, we observed that the boron-Wittig reaction of thiophene-2-carbaldehyde and $1/\text{LiTMP}$ produces the corresponding trisubstituted alkene 4 in an improved *E:Z* ratio = 82:12 (Scheme 6). Interestingly, 4-*E* is synthesised for the first time in this work whereas 4-*Z* had been prepared through Lewis acid $\text{HB}(\text{C}_6\text{F}_5)_2$ -catalysed hydroboration of trimethyl(thiophen-2-ylethynyl)silane in 50% yield.¹⁴ Alternatively, we studied the condensation of thiophene-2-carbaldehyde with $\text{LiC}(\text{Bhex})_2(\text{SiMe}_3)$ (Bhex = hexylene glycolato boryl), generated *in situ* from $\text{HC}(\text{Bhex})_2(\text{SiMe}_3)$ (5) and LiTMP . The product 6 was obtained with an *E:Z* ratio 83:17, although the conversion was reduced to 70% probably due to the steric hindrance associated to the Bhex moiety (Scheme 6). The ¹¹B NMR signal at 28.9 ppm for 6-*E* might correlate with a subtle S-B intramolecular interaction. Furan-2-carbaldehyde reacted with $1/\text{LiTMP}$ providing a similar reaction outcome towards the trisubstituted alkene 7 with a notably increased *E:Z* ratio of 91:1, using both THF or CPME as a solvent (Scheme 6). The intermediate F has been suggested to justify the enhanced stereoselectivity observed in this reaction, through a plausible intramolecular X-B interaction (X = O, S). Compound 7 has been prepared for the first time in this work.

Next, we performed several experiments to demonstrate that steric and electronic modifications on aryl aldehyde substrates contributed not only to obtain the exclusive formation of stereoisomer *E*, but also to reverse the stereocontrol toward formation of stereoisomer *Z*, depending on the substituents present in the aryl group. Electron donating and electron withdrawing groups in a *para* position do not change the reaction outcome with a modest preference for the *E*-8 and *E*-9 stereoisomers (Scheme 7). Both are synthesised for the first time in this work, since only stereoisomers *Z*-8 and *Z*-9 are known.^{8c,15} However, OMe substituents in an *ortho* position seem to have a major influence on the *E*-stereocontrol for *E*-11 and *E*-12 in comparison to the influence of F towards *E*-10



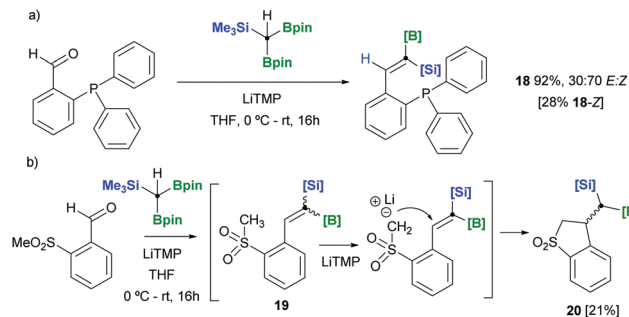
Scheme 6 Boron-Wittig reaction between thiophene-2-carbaldehyde or furan-2-carbaldehyde with $\text{LiC}(\text{Bpin})_2(\text{SiMe}_3)$ or $\text{LiC}(\text{Bhex})_2(\text{SiMe}_3)$.



Scheme 7 Substrate scope for stereoselective boron-Wittig reaction with $\text{HC(Bpin)}_2(\text{SiMe}_3)$ (**1**) and LiTMP.

(Scheme 7). However, higher stereoselectivity is achieved when substrate 2,4,6-(OMe)₃-C₆H₂ is transformed into product **13** in 97% yield as *E:Z* ratio 97:3 (Scheme 7). X-Ray diffraction of *E*-**13** shows a short B₁-O₅ length distance (2.76 Å) compared to B₁-O₃ (4.81 Å), forcing a torsion angle C-C=C-C about 12.02°. It is suggested an attractive interaction between B₁-O₅ since the covalent bond lengths are B₁-O₁ 1.3796(8) and B₁-O₂ 1.3794(8). None of these polyfunctionalised products with *ortho* substituents have been prepared before. Interestingly, when the *ortho* substituents are Cl, Br, I or CF₃, we noticed that the *Z* stereoisomers became preferred in products **14**–**17**.¹⁶ The reversed trend might be correlated with steric effects of the silyl group that controls the B–O elimination towards the preferred products **16** and **17** as *E:Z* 15:85 ratio (Scheme 7).

Similarly, sterically hindered *ortho* substituents are also compatible with the boron-Wittig reaction, as it has been seen in the synthesis of triaryl phosphine **18** in high yield and preference on the *Z* stereoisomer (Scheme 8a). Surprisingly, when 2-(methylsulfonyl)benzaldehyde reacted with reagent **1**/LiTMP, the expected trisubstituted alkene **19** was scarcely isolated (17% **19-E**), and instead the cyclic system **20** was detected and isolated (Scheme 8b). Its formation might be postulated throughout the formation of (phenylsulfonyl)methylene lithium intermediate (*via* deprotonation of the aryl sulfonyl group with the excess of LiTMP)¹⁷ which interacts intramolecularly with the alkene to form the cyclic saturated compound **20**. This type of benzothiophene 1,1-dioxide has

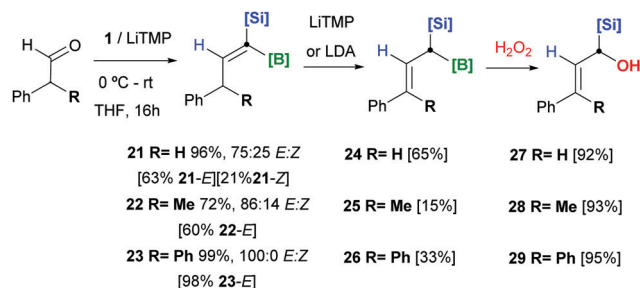


Scheme 8 Boron-Wittig reaction between sterically hindered *ortho* substituted aryl aldehydes and $\text{HC(Bpin)}_2(\text{SiMe}_3)$ (**1**)/LiTMP.

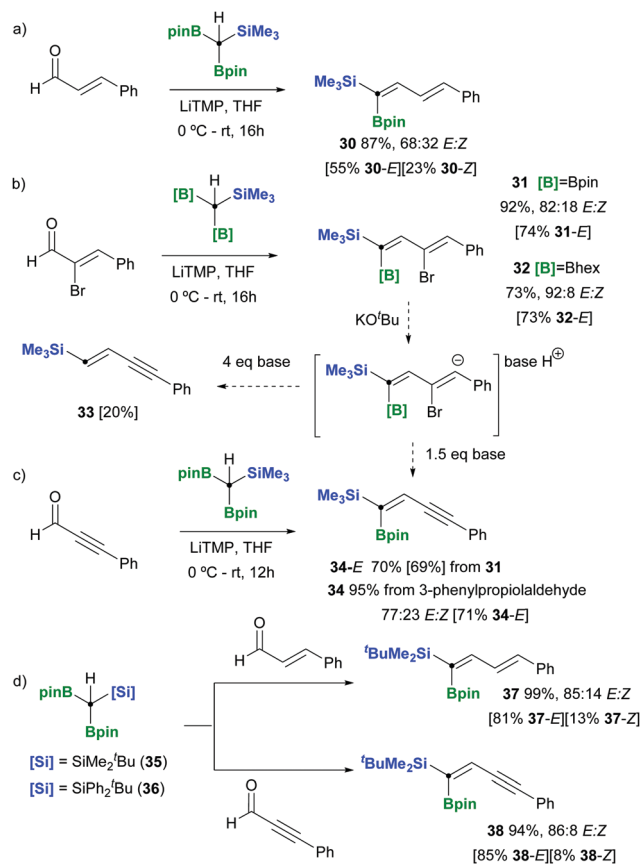
been prepared through catalysed hydrogenation of the corresponding cyclic unsaturated substrates.¹⁸

The olefination of aliphatic aldehydes with **1**/LiTMP has demonstrated that steric factors might also influence the stereoselective reaction outcome. Whereas the boron-Wittig of 2-phenylacetaldehyde generates the corresponding trisubstituted olefin **21** in 75:25 *E:Z* ratio, the olefination of substrate 2-phenylpropanal increases the ratio up to 86:14 *E:Z* in **22**. Interestingly, the most sterically hindered substrate, 2,2-diphenylacetaldehyde, is transformed exclusively towards the *E*-**23** stereoisomer (Scheme 9). All these products could be converted into valuable allylic 1,1-borylsilylalkanes by proton abstraction/isomerization in the presence of LiTMP or LDA. The allylic compounds **24**–**26** could be isolated in moderate yield with exclusive *E* stereoselectivity and efficiently oxidised towards α-(hydroxyallyl)silanes **27**–**29** (Scheme 9).

Finally, we explored the olefination of α,β-unsaturated aldehydes with $\text{HC(Bpin)}_2(\text{SiMe}_3)$ /LiTMP, resulting in a chemo-selective preference on the nucleophilic attack of diborylsilylmethide lithium salt to the aldehyde functionality *versus* the conjugated β position. Scheme 10 shows that cinnamaldehyde condenses with $\text{LiC(Bpin)}_2(\text{SiMe}_3)$ to give the trisubstituted conjugated dienyl compound **30** in a 68:32 *E:Z* ratio, however the analogue substrate 2-bromo-3-phenylacrylaldehyde conducted the boron-Wittig reaction with $\text{LiC(Bpin)}_2(\text{SiMe}_3)$ towards the preferred formation of **31** in a 82:18 *E:Z* ratio. Even higher stereoselectivity could be achieved using reagent $\text{LiC(Bhex)}_2(\text{SiMe}_3)$, generating product *E*-**32** in a 92:8 *E:Z* ratio (Scheme 10). Interestingly, the addition of 1.5 equiv. of KO^tBu



Scheme 9 Selective trend in olefination of aliphatic aldehydes with $\text{HC(Bpin)}_2(\text{SiMe}_3)$ /LiTMP and subsequent proton abstraction/isomerization and eventual oxidation.



Scheme 10 Chemo and stereoselective boron-Wittig reaction with α,β -unsaturated aldehydes.

to **31** contributed to the HBr elimination and subsequent diyne *E*-**34** formation with exclusive stereoselectivity (Scheme 10). However, the addition of an excess of base resulted in a complete proto-deborylation process with the concomitant formation of (*E*)-trimethyl(4-phenylbut-1-en-3-yn-1-yl)silane (**33**). To have a complete picture of the olefination of α,β -unsaturated aldehydes, we conducted the boron-Wittig condensation between 3-phenylpropionaldehyde and reagent **1**. As Scheme 10 shows, chemoselective formation of product **34** could be attained although with lower stereoselectivity, in comparison to *E*-**34** formed from **31**. We have investigated the substituent effect of the silane group on *E/Z* selectivities. The reagents $\text{LiC}(\text{Bin})_2[\text{Si}]$ (**35**, ($[\text{Si}] = \text{SiMe}_2^t\text{Bu}$)) **36**, ($[\text{Si}] = \text{SiPh}_2^t\text{Bu}$) have been synthesized¹⁹ and the stereoselectivity on the boron-Witting indicates that the stereoselectivity on the *E* stereoisomer is favoured when the more sterically hindered substituents in **35** are involved (Scheme 10, products **37** and **38**), although with reagent **36** the conversion was very low, probably due to the highly congested diborylmethylsilane.

In summary, we have conducted the olefination reaction between aromatic or aliphatic aldehydes and $\text{LiC}[\text{B}]_2[\text{Si}]$ ($[\text{B}] = \text{Bpin}$ or Bhex , $[\text{Si}] = \text{SiMe}_3$, SiMe_2^tBu , SiPh_2^tBu) with a special focus on the challenging stereocontrol on the 1,1-silylborylated trisubstituted

alkene formation. We have found that picolinaldehyde, thiophene-2-carbaldehyde and furan-2-carbaldehyde could be involved in stereodetermining intermediates *via* intramolecular interaction of N, S or O with B. Also a divergent stereocontrol has been observed when OMe *ortho* substituents in the aromatic aldehydes favour the *E*-borylsilylalkane formation whereas halide *ortho* substituents stabilise the *Z*-borylsilylalkanes. The condensation of α,β -unsaturated substrates with $\text{LiC}(\text{Bpin})_2[\text{Si}]$ allows access to 1,1-silylborylated conjugated dienes and diynes, with relative high stereoselectivity.

We thank Ministerio de Economía y Competitividad y por el Fondo Europeo de Desarrollo Regional FEDER through project PID2019-109674GB-I00. We thank AllyChem for the gift of diboron reagents.

Conflicts of interest

There are no conflicts to declare.

Notes and references

- 1 A. B. Cuenca and E. Fernández, *Chem. Soc. Rev.*, 2021, **50**, 72.
- 2 K. Endo, M. Hirokami and T. Shibata, *J. Org. Chem.*, 2010, **75**, 3469.
- 3 (a) S. Namirembe, C. Gao, R. P. Wexler and J. P. Morken, *Org. Lett.*, 2019, **21**, 4392; (b) M. Kovalenko, D. V. Yarmoliuk, D. Serhiichuk, D. Chernenko, V. Smyrnov, A. Breslavskiy, O. V. Hryshchuk, I. Kleban, Y. Rassukana, A. V. Tymsunyk, A. A. Tolmachev, Y. O. Kuchkovska and O. O. Grygorenko, *Eur. J. Org. Chem.*, 2019, 5624.
- 4 J. R. Coombs, L. Zhang and J. P. Morken, *Org. Lett.*, 2015, **17**, 1708.
- 5 K. Endo, A. Sakamoto, T. Ohkubo and T. Shibata, *Chem. Lett.*, 2011, **40**, 1440.
- 6 E. La Cascia, A. B. Cuenca and E. Fernández, *Chem. – Eur. J.*, 2016, **22**, 18737.
- 7 D. J. Peterson, *J. Org. Chem.*, 1968, **33**, 780.
- 8 (a) K. Semba, T. Fujihara, J. Terao and Y. Tsuji, *Chem. – Eur. J.*, 2012, **18**, 4179; (b) Y. D. Bidal, F. Lazreg and C. S. J. Cazin, *ACS Catal.*, 2014, **4**, 1564; (c) L. Ferrand, Y. Lyu, A. Rivera-Hernandez, B. J. Fallon, M. Amatore, C. Aubert and M. Petit, *Synthesis*, 2017, 3895.
- 9 (a) H. E. Zimmerman and M. D. Traxler, *J. Am. Chem. Soc.*, 1957, **79**, 1920; (b) D. Bergelson, L. Barsukor and M. M. Shemyaker, *Tetrahedron*, 1967, **26**, 2709; (c) D. A. Evans, J. V. Nelson and T. R. Taber, *Top. Stereochem.*, 1982, **13**, 1.
- 10 A. R. Bassindale, R. J. Ellis, J. C.-Y. Lau and P. G. Taylor, *J. Chem. Soc., Chem. Commun.*, 1986, 98.
- 11 K. Itami, T. Nokami and J. Yoshida, *Org. Lett.*, 2000, **2**, 1299.
- 12 B.-T. Gröbel and D. Seebach, *Angew. Chem., Int. Ed. Engl.*, 1974, **13**, 83.
- 13 P. K. Hudrik, E. L. O. Agwaramgbo and A. M. Hudrik, *J. Org. Chem.*, 1989, **54**, 5613.
- 14 M. Fleige, J. Möbus, T. Vom Stein, F. Glorius and D. W. Stephan, *Chem. Commun.*, 2016, **52**, 10830.
- 15 Y. M. Chae, J. S. Bae, J. H. Moon, J. Y. Lee and J. Yun, *Adv. Synth. Catal.*, 2014, **356**, 843.
- 16 (a) H.-Y. Jung and J. Yun, *Org. Lett.*, 2012, **14**, 2606; (b) Y. Gu, Y. Duan, Y. Shen and R. Martin, *Angew. Chem., Int. Ed.*, 2020, **59**, 2061.
- 17 J. J. Eisch, S. K. Dua and M. Behrooz, *J. Org. Chem.*, 1985, **50**, 3674.
- 18 (a) G. Liu, K. Tian, C. Li, C. You, X. Tan, H. Zhang, X. Zhang and X.-Q. Dong, *Org. Lett.*, 2021, **23**, 668; (b) A. Tpsatti and A. Pfaltz, *Angew. Chem., Int. Ed.*, 2017, **56**, 4579.
- 19 J. Kim and S. H. Cho, *ACS Catal.*, 2019, **9**, 230.