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Scope and standards – updated October 2017

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We welcome research that shows new or significantly improved protocols or methodologies in total synthesis, synthetic methodology or physical and theoretical organic chemistry as well as research that shows a significant advance in the organic chemistry or molecular design aspects of chemical biology, catalysis, supramolecular and macromolecular chemistry, theoretical chemistry, mechanism-oriented physical organic chemistry, medicinal chemistry or natural products.

More details about key areas of our scope are below. In all cases authors should include in their article clear rationale for why their research has been carried out.

Organic synthesis: We welcome important research in all areas of organic synthesis, including studies on small organic molecules and biomolecules, and studies that report purely synthetic work without biological data. Total or multistep syntheses should report new or improved strategies or methods, or a more efficient route to the target compound. Methodology studies should show a significant improvement on known methods. Research that extends known methodology to a different class of compounds is generally not suitable, unless that class is significantly different in scope to previously reported methodology.

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Further Respond to Reviewers

(OB-COM-03-2018-000699)

Reviewer 3

The revised manuscript has addressed the main questions properly. It is recommended for publication after a few minor changes.

We are grateful to the reviewer for positive comments and kind suggestions on our revised manuscript. Corresponding changes have been made according to the reviewer's suggestions as bellow:

1, Remove the citation (reference 10) in the sentence "...we would like to report our recent study on the iron-catalysed intermolecular vicinal aminoazidation of alkenes10 (Scheme 1c)." No reference should be cited before "(Scheme 1c)"

We appreciate the reviewer's suggestion, and have remove the cited reference in this sentence.

2, Change the citation 10 to 10a in the sentence "...On the contrary, no matter how much the reaction time was, the yields of 2a remained poor in absence of ligand when changing the iron catalyst to CuCl10 (entry 12),.."

We appreciate the reviewer's suggestion, and have changed the citation 10 to 10a in this sentence.

3, The following paragraph should be revised for clarity. The information is somehow confusing and doesn't fit well the cited references.

"In 2014, Zhang and Studer reported a novel copper-catalysed aminoazidation of styrenes (Scheme 1b),10a which has drawn our attention since organic azides could serve as versatile intermediates to easily transform to other nitrogen-containing functional groups including amines,10, 15 albeit their protocol could only be effective on styrenes and could not react in absence of a ligand. Zhang and Studer also demonstrated that both benzenesulfonyl and azide two different protecting groups of amines, which can be deprotected under different conditions sequentially,10a allowing maximal synthetic flexibility and providing an opportunity to prepare vicinal diamines with two miscellaneously functionalized amino groups. "

Consider the revision as below:

In 2014, Zhang and Studer reported a novel copper-catalysed aminoazidation of styrenes (Scheme 1b),10a which has drawn our attention since organic azides could serve as versatile intermediates to easily transform to other nitrogen-containing functional groups including amines.10, 15 They also demonstrated that both benzenesulfonyl and azide two different protecting groups of amines can be deprotected under different conditions sequentially.10a

Although it offers maximal synthetic flexibility and an opportunity to prepare vicinal diamines with two miscellaneous functionalized amino groups, this copper-catalysed protocol was only be effective on styrenes and could not react in absence of a ligand.

We appreciate the reviewer's suggestion, and have changed this paragraph as the reviewer suggested, highlighted in yellow background.

4, remove the yellow highlight in the manuscript for the accepted version.

We appreciate the reviewer's kind suggestion. Beside the revised manuscript with yellow highlight to point out the changes we have made, we also provide a DOC file and a PDF file without the yellow highlight for the accepted version.

In summary, we would like to gratefully thank all the editors and reviewers again for the above positive comments and constructive suggestions on our manuscript.

Best regards,

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NFSI-participated Intermolecular Aminoazidation of Alkene through Iron Catalysis†

Received 00th January 20xx,
Accepted 00th January 20xxBowen Lei,^a Xiaojiao Wang,^a Lifang Ma,^a Yan Li,^a and Ziyuan Li^{*a}

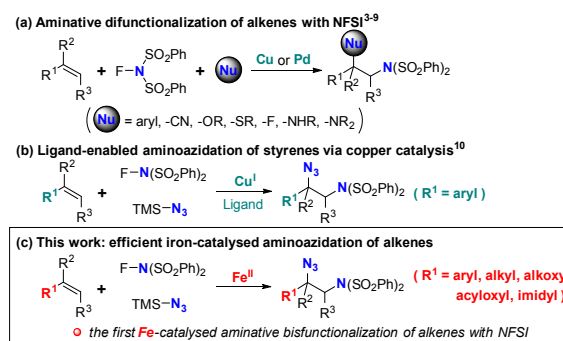
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An iron-catalysed intermolecular vicinal aminoazidation of alkenes, using *N*-fluorobenzenesulfonimide (NFSI) and trimethylsilyl azide (TMSN₃) as the imidating and azidating reagents respectively, is described, which could potentially provide a valuable route toward diverse vicinal diamine derivatives of great significance in medicinal chemistry and organic synthesis. Such iron-catalysed aminative bisfunctionalization of alkene with NFSI has not yet been reported before. Comparing to previously employed copper or palladium catalysts, the iron catalyst, FeCl₂, was demonstrated to be a good alternative for its comparable efficiency and broad alkene scope. Preliminary mechanistic study suggested that this iron-catalysed reaction is realized through radical process.

Vicinal diamine is a ubiquitous scaffold of great significance, which widely resides in miscellaneous ligands, organocatalysts, synthetic building blocks, and biologically active natural or artificial molecules.¹ To acquire such significant moiety, many efforts have been spent on transition metal-catalysed 1,2-diamination of alkenes, but the catalysts used in these reactions were generally limited to copper and palladium.² Recently, NFSI (*N*-fluorobenzenesulfonimide), which could easily generate an electrophilic disulfonimidyl radical, has been broadly used as an oxidative imidating reagent in transition metal-catalysed or metal-free amination reactions through cascade radical processes. Since Michael group and Liu group disclosed two pioneering works on NFSI-involved diamination^{3a} and aminofluorination^{3b} of alkene, NFSI has been widely employed as a powerful nitrogen donor in vicinal aminative difunctionalization of alkenes (Scheme 1a)⁴⁻¹⁰, such as aminoarylation⁴, aminocyanation⁵, aminooxygenation⁶, aminothiation⁷, aminofluorination⁸, diamination^{5a, 9}, and aminoazidation^{10a}. Beside aminative difunctionalization of alkene, Zhang group reported the first two examples on Pd-catalysed

oxidative benzylic^{11a} and aromatic^{11b} C-H aminations with NFSI, followed by the widely applications of this imidating reagent in C-H amination of arenes¹², alkanes¹³ and alkenes¹⁴.



Scheme 1 Transition-metal catalyzed aminative functionalization with NFSI

Among these bisfunctionalization of alkenes with NFSI, the diamination and aminoazidation could provide a potential approach toward the above-mentioned significant vicinal diamine moiety. However, all these reactions were Cu-^{5a, 9a-b, 10a} or Pd-^{9c-f} catalysed, and require assistance of a ligand^{5a, 9a-e-f} or an additive^{9b-d} to realize the reaction or to guarantee gratifying yields. More importantly, most of these reactions were only effective on active alkenes like styrenes^{5a, 9a-b} or allylic ethers^{9c}. Michael group reported a series of significant works on 1,2-diamination of inactive alkenes with NFSI,^{9d-f} but one of the two amination steps was proceeded intramolecularly, affording aminated *N*-heterocycles only, which limited the diversity of vicinal diamine products. Up to now, no NFSI-participated intermolecular diamination of inactive alkenes has been reported.

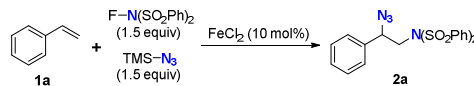
In 2014, Zhang and Studer reported a novel copper-catalysed aminoazidation of styrenes (Scheme 1b),^{10a} which has drawn our attention since organic azides could serve as versatile intermediates to easily transform to other nitrogen-containing functional groups including amines.^{10, 15} They also demonstrated that both benzenesulfonyl and azide group, two different protecting groups of amines, can be deprotected under different conditions sequentially.^{10a} Although it offers

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† Electronic Supplementary Information (ESI) available: Detailed experimental procedures, ¹H and ¹³C NMR data and spectra. See DOI: 10.1039/x0xx00000x

maximal synthetic flexibility and an opportunity to prepare vicinal diamines with two miscellaneously functionalized amino groups, this copper-catalysed protocol was only effective on styrenes and could not react in absence of a ligand. Though limitations were observed, these previous reports have enlightened us to develop new transition metal catalysts with comparable efficiency but broader alkene scope for the NFSI-participated aminative bisfunctionalization. Herein, as an advancement of our previous studies on C-H functionalization¹⁶ including the first iron-catalysed C-H imidation with NFSI^{16a}, as well as other works on transition metal-catalysed annulations with simple nitrogen donors including azide,¹⁷ we would like to report our recent study on the iron-catalysed intermolecular vicinal aminoazidation of alkenes (Scheme 1c). To our knowledge, the iron-catalysed aminative bisfunctionalization of alkenes with NFSI has not been disclosed previously.

Table 1 Optimization of the reaction conditions^a

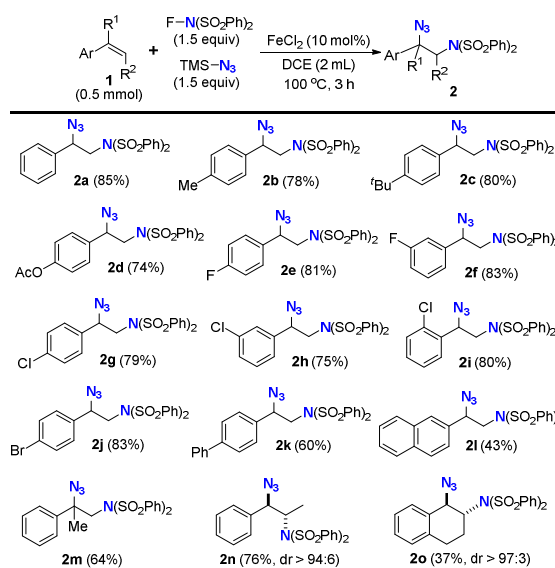


Entry	Solvent (2 ml)	Temperature (°C)	Time (h)	Yield (%) ^b
1	DCE	70	15	44
2	DMF	70	15	trace
3	DME	70	15	0
4	CH ₃ CN	70	15	35
5	Dioxane	70	15	0
6	Toluene	70	15	trace
7	DCE	100	15	73
8 ^c	DCE	100	15	84
9 ^{c, d}	DCE	100	15	78
10 ^c	DCE	100	5	(87)
11 ^c	DCE	100	3	85 (90)
12 ^{c, e}	DCE	100	3	(14)
13 ^{c, f}	DCE	100	3	0

^aReaction conditions: styrene **1a** (0.5 mmol), NFSI (0.75 mmol), TMSN₃ (0.75 mmol), FeCl₂ (0.05 mmol) in solvent (2 mL) in air (1 atm). ^bIsolated yields of **2a**. The yields in the parentheses are determined by NMR using CH₂Br₂ as an internal standard. ^cConducted under N₂ (1 atm). ^dWith NFSI (1 mmol) and TMSN₃ (1 mmol). ^eCatalyzed by CuCl (0.05 mmol) instead of FeCl₂. ^fIn the absence of FeCl₂.

Our investigation was commenced with the screening of reaction conditions for this novel iron-catalysed aminoazidation using styrene **1a** as the alkene substrate, and the results are summarized in Table 1. Initial trial with 10 mol% of FeCl₂ at 70 °C in air afforded corresponding aminoazidated product **2a** in 44% yield after 15 hours (entry 1). Several commonly used solvents were screened (entries 2-6), and most gave no or only trace product **2a** under the same conditions, except for acetonitrile with 35% yield (entry 4), indicating that DCE is the best solvent. The yield of **2a** was considerably elevated when reaction temperature was increased to 100 °C (entry 7), and could be further promoted to 84% when this aminoazidation was conducted under N₂ (1 atm), suggesting molecular dioxygen is detrimental to this reaction (entry 8). Kinetic experiments were then conducted,

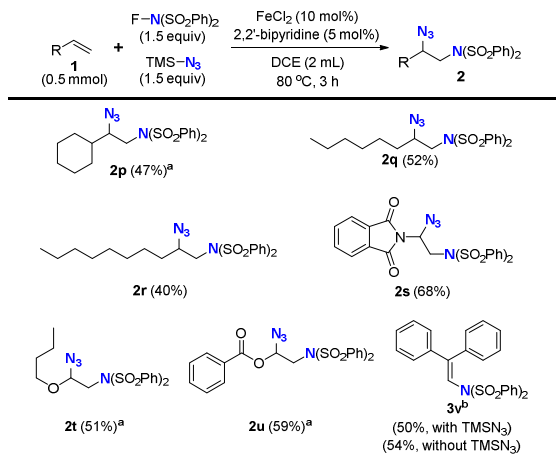
in order to investigate whether the relatively long reaction time is necessary.¹⁸ The results showed that the reaction catalysed by FeCl₂ was fast, providing **2a** in 90% NMR yield after 3 hours (entry 11), and no ligand or additive is required, indicating that elevating temperature solely is sufficient to promote the efficiency of this iron-catalysed aminoazidation. Prolonging the reaction time could not further promote the yield (entry 10). On the contrary, no matter how much the reaction time was, the yields of **2a** remained poor in absence of ligand when changing the iron catalyst to CuCl^{10a} (entry 12), suggesting that, for the copper-catalysed reaction, increasing temperature is less helpful than the employment of an appropriate ligand. In addition, no aminoazidated product was generated when the reaction was conducted without FeCl₂ (entry 13). Therefore, the reaction conditions in entry 11 were selected as the optimized conditions for subsequent exploration of alkene scope.



Scheme 2 Fe-catalysed aminoazidation of styrenes with NFSI. Reaction conditions: see entry 11, Table 1. Isolated yields.

With the optimized conditions established, a variety of styrene derivatives were investigated for this aminoazidation (Scheme 2, **2a-m**). Generally, all styrenes underwent this Fe-catalysed ligand-free aminoazidation smoothly, affording corresponding products **2b-j** in good yields, and no remarkable differences in yields were observed between electron-rich methyl-, *tert*-butyl- or acetoxyl-substituted products (**2b-d**) and electron-deficient fluoro-, chloro- or bromo-substituted styrenes (**2e-j**). Moreover, steric hindrance on the benzene ring seemed to share little impact on the yield, since the yields of *para*-substituted **2g** and *ortho*-substituted **2i** are comparable. However, conversion of 4-vinylbiphenyl or 2-vinylnaphthalene with larger conjugated system was poorer, and the yields of their aminoazidated products (**2k-l**) were moderate. In addition, a 1,1-disubstituted styrene **1m** also gave corresponding product **2m** in a slightly dropped yield. When internal alkenes, β -methylstyrene (**1n**) or 1,2-dihydronaphthalene (**1o**), were conducted under the

optimized conditions, the yield of aminoazidated product **2n** was well maintained, while the yield of **2o** generated from more rigid substrate **1o** declined remarkably. The diastereoselectivity of **2n** and **2o** could be readily determined by ^1H -NMR spectra of the crude products,¹⁰ and both showed excellent *trans*-selectivity (94:6 and 97:3, respectively), which is comparable to the Cu-catalysed aminoazidation reported by Zhang and Studer.¹⁰

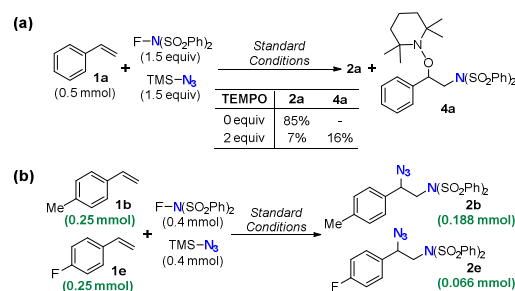


Scheme 3 Fe-catalysed aminoazidation of alkenes beside styrenes. Reaction conditions: alkene **1** (0.5 mmol), NFSI (0.75 mmol), TMSN₃ (0.75 mmol), FeCl₂ (0.05 mmol) and 2,2'-bipyridine (5 mol%) in DCE (2 mL) under N₂ (1 atm) at 80 °C. Isolated yields. ^aReaction time was prolonged to 6 h. ^bAt 100 °C, without 2,2'-bipyridine.

Subsequently, the substrate scope of this efficient Fe-catalysed aminoazidation was expanded to other types of alkene beside styrene (Scheme 3, **2p-u**). Assisted by 5 mol% of 2,2'-bipyridine, inactivated alkenes such as 1-octene (**1q**), 1-decene (**1r**) and vinylcyclohexane (**1p**) could provide aminoazidated products **2p-r** in moderate yields at lower temperature, while such aliphatic alkenes did not react under Cu-catalysed conditions.¹⁰ Heteroatom-substituted alkenes including *N*-vinylphthalimide (**1s**), *n*-butyl vinyl ether (**1t**) and vinyl benzoate (**1u**) could also undergo this aminoazidation, affording corresponding products **2s-u** in moderate to good yields. These results suggested that the alkene scope of this Fe-catalysed aminoazidation is broader than previous copper catalysis.¹⁰ Interestingly, the desired aminoazidated product could not be obtained when this reaction was conducted on 1,1-diphenylethylene (**1v**). Instead, an oxidative C-H aminated product **3v** was generated with or without TMSN₃, indicating that 1,1-diaryl alkene might undergo hydrogen elimination more easily than azidation after the addition of disulfonimidyl group.

To gain some insights into mechanistic picture preliminarily, some control experiments were then performed.¹⁸ First, the addition of TEMPO (2,2,6,6-tetramethylpiperidine-1-oxyl) severely suppressed this aminoazidation. Meanwhile, an amino-oxygenated product **4a** was obtained in 16% yield (Scheme 4a), suggesting that a benzyl radical might probably be formed after the attack of the disulfonimidyl radical generated from NFSI, coinciding with copper-catalysed aminofluorination of styrenes with NFSI studied by Zhang.^{8a}

Then a competition experiment was performed to elucidate the electronic preference on the alkene substrate. An equimolecular mixture of styrenes **1b** and **1e** was subjected to insufficient loadings of NFSI and TMSN₃, and the ratio between electron-rich aminoazidated products **2b** and electron-deficient product **2e** was approximately 2.85 : 1 (Scheme 4b), indicating that this Fe-catalysed aminoazidation could be categorized as an electrophilic radical addition on alkenes, which is consistent with previously reported copper- or palladium-catalysed aminative bisfunctionalization of alkene with NFSI.



Scheme 4 Control experiments.

Conclusions

To summarize, we have developed an intermolecular vicinal aminoazidation of alkene with NFSI and TMSN₃ through iron catalysis. By elevating the reaction temperature, this iron-catalysed aminoazidation on styrene could be achieved in absence of ligand, and the alkene scope of this reaction is broader than previous copper catalysis, suggesting that iron catalysis might probably be a promising alternative for diverse aminative bisfunctionalization of alkenes in addition of copper- or palladium-catalysis. Fully revealing the precise mechanistic picture, as well as further development of other iron-catalysed aminative bisfunctionalization of alkene with NFSI, are undergoing.

Acknowledgements

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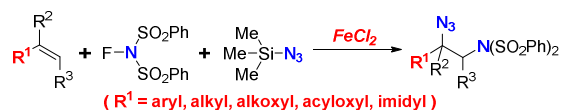
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- 18 Please see the Electronic Supplementary Information for the detailed results.



The iron-catalysed intermolecular vicinal aminoazidation of alkene with NFSI is reported, with broader alkene scope comparing to previously reported aminoazidation.

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NFSI-participated Intermolecular Aminoazidation of Alkene through Iron Catalysis

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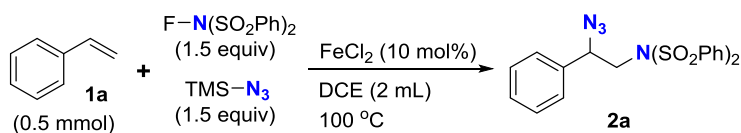
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General Remarks

All commercially available compounds were purchased from Sigma-Aldrich, Alfa-Aesar, Acros, J&K Chemicals, Adamas-beta, Accela ChemBio and Aladdin Chemicals. FeCl_2 was purchased from Alfa-Aesar (99.99% purity, ultra dry, CAS No. 7758-94-3). TMSN_3 was purchased from TCI (>95%, CAS No. 4648-54-8). *N*-Fluorobenzenesulfonimide (NFSI) (98% purity, CAS No. 133745-75-2) and 2,2'-bipyridine (98% purity, CAS No. 366-18-7) were purchased from Accela ChemBio. Unless otherwise noted, materials obtained from commercial suppliers were used without further purification. Products were purified by flash chromatography on silica gel using petroleum ether, ethyl acetate and dichloromethane as the eluents. ^1H -NMR spectra were recorded on Bruker AVANCE III-400 spectrometers. Chemical shifts (in ppm) were referenced with TMS in CDCl_3 (0 ppm). ^{13}C -NMR spectra were obtained by using the same NMR spectrometers and were calibrated with CDCl_3 ($\delta = 77.00$ ppm). High resolution mass spectra were obtained from an Agilent 6520B Q-TOF mass spectrometer with electron spray ionization (ESI) as the ion source.

Kinetic Studies



To a reaction tube charged with FeCl_2 (6.3 mg, 0.05 mmol) and NFSI (236.5 mg, 0.75 mmol) was added a solution of styrene (**1a**, 57.5 μL , 0.5 mmol), TMSN_3 (98.6 μL , 0.75 mmol) in DCE (2 mL) via a syringe under N_2 (1 atm). The reaction mixture was stirred at 100 $^\circ\text{C}$ for indicated time in Table S1. After rapidly cooled by ice, the mixture was diluted with ethyl acetate, filtered through a celite pad, and concentrated *in vacuo*. The residue was analyzed with ^1H -NMR to determine the yields of **2a** through iron-catalysis, using CH_2Br_2 as an internal standard. Similarly, the yields of **2a** via copper-catalysis, using the same copper catalyst employed by Zhang and Studer,^[1] were determined in the same way. Then the yields of **2a** via iron- or copper-catalysis in different times were summarized in Table S1 and Figure S1, accordingly.

Table S1. Detailed Results of Kinetic Studies.

Time (min)	20	40	60	80	100	120	180	300	420
FeCl_2	39.8	53.0	64.1	72.7	78.0	82.5	89.8	87.2	85.6
CuCl	9.7	13.7	15.4	16.3	15.8	14.6	13.6	11.8	9.6

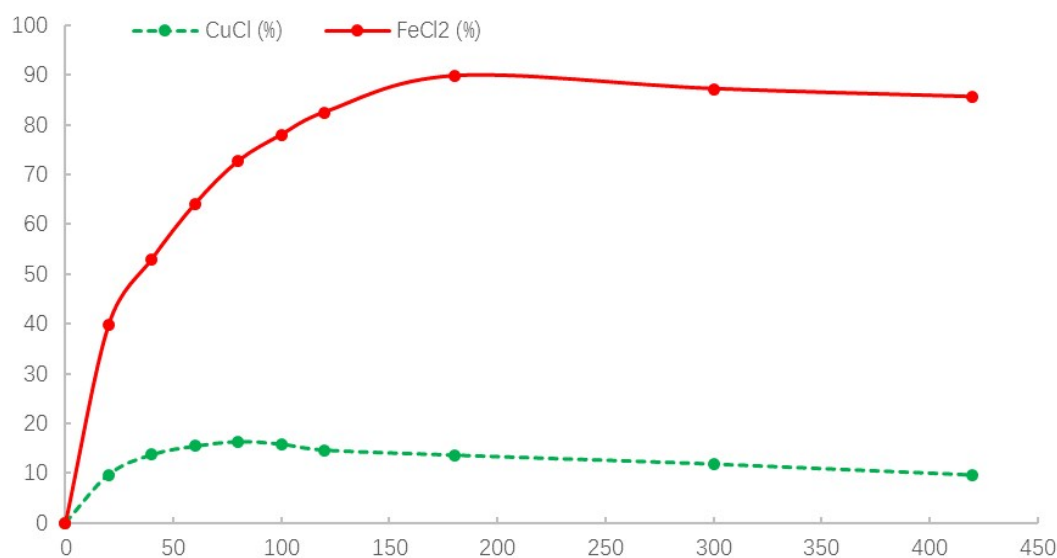
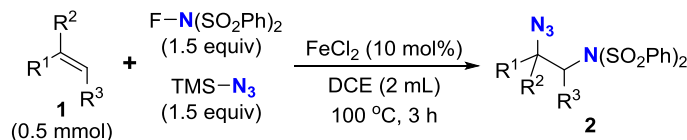


Figure S1. Detailed Results of Kinetic Studies.

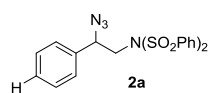
From the data in Table S1 and Figure S1, this aminoazidation catalyzed by FeCl_2 is remarkably faster than catalyzed by CuCl , which was previously employed by Zhang and Studer,^[1] while the yields of **2a** remained very low, no matter how much the reaction time was, when the reaction was catalyzed by CuCl . These results demonstrated that the iron catalyst, which could achieve this reaction rapidly without any ligand or additive and provide the product in excellent yield after 3 hours, is a competitive alternative to previously used copper catalyst, which could not effectively realize this reaction in absence of certain ligands.^[1]

Experimental Procedure and Characterization Data



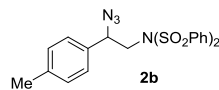
Typical Procedure: To a reaction tube charged with FeCl_2 (6.3 mg, 0.05 mmol) and NFSI (236.5 mg, 0.75 mmol) was added a solution of alkene (**1a-u**, 0.5 mmol), TMSN_3 (98.6 μL , 0.75 mmol) in DCE (2 mL) via a syringe under N_2 (1 atm). The reaction mixture was stirred at 100 $^\circ\text{C}$ for 3 hours. After rapidly cooling by ice, the mixture was diluted with ethyl acetate, filtered through a celite pad, and concentrated *in vacuo* to give dark residue, which was then purified by flash chromatography using petroleum ether and ethyl acetate as the eluent on silical gel to afford aminoazidated product **2a-u**.

N-(2-Azido-2-phenylethyl)-*N*-(phenylsulfonyl)benzenesulfonamide^[1] (**2a**):



The reaction of 0.5 mmol of styrene (**1a**) with NFSI and TMSN_3 afforded 187.2 mg of **2a** (85%) as colorless oil, after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, v/v) as the eluent. ^1H NMR (CDCl_3 , 400 MHz): δ = 8.07-8.05 (m, 4H), 7.67-7.63 (m, 2H), 7.56-7.52 (m, 4H), 7.40-7.35 (m, 5H), 5.01 (dd, J = 9.6 Hz, 4.2 Hz, 1H), 4.06 (dd, J = 15.6 Hz, 9.6 Hz, 1H), 3.72 (dd, J = 15.6 Hz, 4.2 Hz, 1H) ppm; ^{13}C NMR (CDCl_3 , 100 MHz): δ = 139.22, 136.44, 134.02, 129.12, 129.02, 128.98, 128.57, 127.19, 65.57, 53.18 ppm. HRMS m/z (ESI) calcd for $[\text{C}_{20}\text{H}_{18}\text{N}_4\text{O}_4\text{S}_2+\text{Na}]^+$ 465.0662, found 465.0669.

N-(2-Azido-2-(*p*-tolyl)ethyl)-*N*-(phenylsulfonyl)benzenesulfonamide^[1] (**2b**):

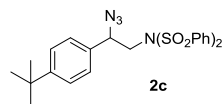


The reaction of 0.5 mmol of *p*-methylstyrene (**1b**) with NFSI and TMSN_3 afforded 178.1 mg of **2b** (78%) as white solid (m.p. 133.4-134.8 $^\circ\text{C}$), after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, v/v) as the eluent. ^1H NMR (CDCl_3 , 400 MHz): δ = 8.06-8.04 (m, 4H), 7.66-7.62 (m, 2H), 7.55-7.52 (m, 4H), 7.25 (d, J = 8.1 Hz, 2H), 7.20 (d, J = 8.1 Hz, 2H), 4.98 (dd, J = 9.6 Hz, 4.3 Hz, 1H), 4.05 (dd, J = 15.6 Hz, 9.6 Hz, 1H), 3.70 (dd, J = 15.6 Hz, 4.3 Hz, 1H), 2.36 (s, 3H) ppm; ^{13}C NMR (CDCl_3 , 100 MHz): δ = 139.20, 138.95, 133.98, 133.32, 129.75, 128.94, 128.56, 127.15, 65.32, 53.09, 21.16 ppm. HRMS m/z (ESI) calcd for $[\text{C}_{21}\text{H}_{20}\text{N}_4\text{O}_4\text{S}_2+\text{Na}]^+$ 479.0818,

found 479.0827.

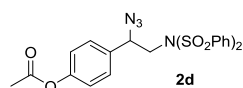
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***N*-(2-Azido-2-(4-(*tert*-butyl)phenyl)ethyl)-*N*-(phenylsulfonyl)benzenesulfonamide^[1] (**2c**):**



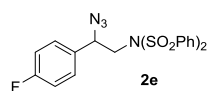
The reaction of 0.5 mmol of *p*-*tert*-butylstyrene (**1c**) with NFSI and TMSN₃ afforded 199.4 mg of **2c** (80%) as light yellow oil, after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, *v/v*) as the eluent. ¹H NMR (CDCl₃, 400 MHz): δ = 8.08-8.06 (m, 4H), 7.65-7.62 (m, 2H), 7.55-7.51 (m, 4H), 7.42 (d, *J* = 8.3 Hz, 2H), 7.29 (d, *J* = 8.3 Hz, 2H), 4.98 (dd, *J* = 9.8 Hz, 3.9 Hz, 1H), 4.08 (dd, *J* = 15.6 Hz, 9.8 Hz, 1H), 3.71 (dd, *J* = 15.6 Hz, 3.9 Hz, 1H), 1.32 (s, 9H) ppm; ¹³C NMR (CDCl₃, 100 MHz): δ = 152.08, 139.25, 133.97, 133.36, 128.93, 128.54, 126.84, 125.96, 65.30, 53.11, 34.61, 31.21 ppm. HRMS *m/z* (ESI) calcd for [C₂₄H₂₆N₄O₄S₂+Na]⁺ 521.1293, found 521.1290; calcd for [C₂₄H₂₆N₄O₄S₂+K]⁺ 537.1027, found 537.1030.

4-(1-Azido-2-(*N*-(phenylsulfonyl)phenylsulfonamido)ethyl)phenyl acetate^[1] (2d**):**



The reaction of 0.5 mmol of 4-vinylphenyl acetate (**1d**) with NFSI and TMSN₃ afforded 184.3 mg of **2d** (74%) as colorless oil, after flash chromatography on silica gel using petroleum ether and ethyl acetate (12:1 to 8:1, *v/v*) as the eluent. ¹H NMR (CDCl₃, 400 MHz): δ = 8.06-8.04 (m, 4H), 7.66-7.62 (m, 2H), 7.56-7.52 (m, 4H), 7.37 (d, *J* = 8.6 Hz, 2H), 7.12 (d, *J* = 8.6 Hz, 2H), 5.02 (dd, *J* = 9.5 Hz, 4.2 Hz, 1H), 4.03 (dd, *J* = 15.6 Hz, 9.5 Hz, 1H), 3.72 (dd, *J* = 15.6 Hz, 4.2 Hz, 1H), 2.29 (s, 3H) ppm; ¹³C NMR (CDCl₃, 100 MHz): δ = 169.12, 150.97, 139.02, 134.00, 133.91, 128.95, 128.47, 128.25, 122.26, 64.99, 53.02, 20.97 ppm. HRMS *m/z* (ESI) calcd for [C₂₂H₂₀N₄O₆S₂+Na]⁺ 523.0716, found 523.0720.

***N*-(2-Azido-2-(4-fluorophenyl)ethyl)-*N*-(phenylsulfonyl)benzenesulfonamide^[1] (**2e**):**

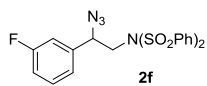


The reaction of 0.5 mmol of *p*-fluorostyrene (**1e**) with NFSI and TMSN₃ afforded 185.3 mg of **2e** (81%) as white solid (m.p. 104.7-105.9 °C), after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, *v/v*) as the eluent. ¹H NMR (CDCl₃, 400 MHz): δ = 8.06-8.03 (m, 4H), 7.67-7.63 (m, 2H), 7.56-7.52 (m, 4H), 7.34-7.31 (m, 2H), 7.09-7.04 (m, 2H), 5.01 (dd, *J* = 9.3 Hz, 4.5 Hz, 1H), 4.02 (dd, *J* = 15.6 Hz, 9.3 Hz, 1H), 3.71 (dd, *J* = 15.6 Hz, 4.5 Hz, 1H) ppm; ¹³C NMR (CDCl₃, 100 MHz): δ = 162.91 (d, *J* = 247.0 Hz), 139.12, 134.06, 132.27 (d, *J* = 3.0 Hz), 129.06, 128.99, 128.51, 116.09 (d, *J* = 21.4 Hz), 64.86, 53.16 ppm. HRMS *m/z* (ESI) calcd for [C₂₀H₁₇FN₄O₄S₂+Na]⁺

483.0567, found 483.0569.

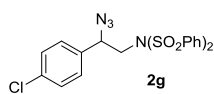
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***N*-(2-Azido-2-(3-fluorophenyl)ethyl)-*N*-(phenylsulfonyl)benzenesulfonamide^[1] (**2f**):**



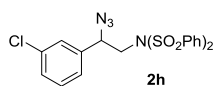
The reaction of 0.5 mmol of *m*-fluorostyrene (**1e**) with NFSI and TMSN₃ afforded 190.2 mg of **2f** (83%) as colorless oil, after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1 to 12:1, *v/v*) as the eluent. ¹H NMR (CDCl₃, 400 MHz): δ = 8.07-8.05 (m, 4H), 7.66-7.63 (m, 2H), 7.56-7.52 (m, 4H), 7.38-7.32 (m, 1H), 7.14-7.13 (m, 1H), 7.07-7.02 (m, 2H), 5.01 (dd, *J* = 9.5 Hz, 4.2 Hz, 1H), 4.04 (dd, *J* = 15.6 Hz, 9.5 Hz, 1H), 3.71 (dd, *J* = 15.6 Hz, 4.2 Hz, 1H) ppm; ¹³C NMR (CDCl₃, 100 MHz): δ = 162.92 (d, *J* = 246.5 Hz), 139.08, 138.95 (d, *J* = 7.1 Hz), 134.07, 130.74 (d, *J* = 8.1 Hz), 129.00, 128.48, 122.74 (d, *J* = 2.9 Hz), 115.95 (d, *J* = 21.0 Hz), 114.13 (d, *J* = 22.1 Hz), 65.02 (d, *J* = 1.2 Hz), 53.16 ppm. HRMS *m/z* (ESI) calcd for [C₂₀H₁₇FN₄O₄S₂+Na]⁺ 483.0567, found 483.0570.

***N*-(2-Azido-2-(4-chlorophenyl)ethyl)-*N*-(phenylsulfonyl)benzenesulfonamide^[1] (**2g**):**



The reaction of 0.5 mmol of *p*-chlorostyrene (**1g**) with NFSI and TMSN₃ afforded 188.0 mg of **2g** (79%) as white solid (m.p. 144.8-146.0 °C), after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, *v/v*) as the eluent. ¹H NMR (CDCl₃, 400 MHz): δ = 8.04-8.01 (m, 4H), 7.66-7.62 (m, 2H), 7.55-7.51 (m, 4H), 7.34 (d, *J* = 8.5 Hz, 2H), 7.28 (d, *J* = 8.5 Hz, 2H), 5.01 (dd, *J* = 9.1 Hz, 4.8 Hz, 1H), 3.99 (dd, *J* = 15.6 Hz, 9.1 Hz, 1H), 3.73 (dd, *J* = 15.6 Hz, 4.8 Hz, 1H) ppm; ¹³C NMR (CDCl₃, 100 MHz): δ = 138.99, 134.89, 134.86, 134.04, 129.26, 128.97, 128.58, 128.45, 64.83, 52.98 ppm. HRMS *m/z* (ESI) calcd for [C₂₀H₁₇ClN₄O₄S₂+Na]⁺ 499.0277, found 499.0278; calcd for [C₂₀H₁₇ClN₄O₄S₂+K]⁺ 515.0011, found 515.0017.

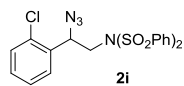
***N*-(2-Azido-2-(3-chlorophenyl)ethyl)-*N*-(phenylsulfonyl)benzenesulfonamide (**2h**):**



The reaction of 0.5 mmol of *m*-chlorostyrene (**1h**) with NFSI and TMSN₃ afforded 177.9 mg of **2h** (75%) as colorless oil, after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, *v/v*) as the eluent. ¹H NMR (CDCl₃, 400 MHz): δ = 8.07-8.05 (m, 4H), 7.67-7.63 (m, 2H), 7.56-7.52 (m, 4H), 7.33-7.31 (m, 3H), 7.25-7.23 (m, 1H), 4.99 (dd, *J* = 9.4 Hz, 4.3 Hz, 1H), 4.03 (dd, *J* = 15.6 Hz, 9.4 Hz, 1H), 3.71 (dd, *J* = 15.6 Hz, 4.3 Hz, 1H) ppm; ¹³C NMR (CDCl₃, 100 MHz): δ = 139.06, 138.51, 134.96, 134.08, 130.39, 129.16, 129.00, 128.48, 127.29, 125.25, 64.99, 53.14 ppm. HRMS

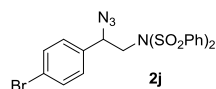
***m/z* (ESI)** calcd for $[C_{20}H_{17}ClN_4O_4S_2+Na]^+$ 499.0272, found 499.0278.

***N*-(2-Azido-2-(2-chlorophenyl)ethyl)-*N*-(phenylsulfonyl)benzenesulfonamide (**2i**):**



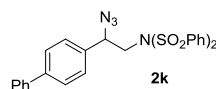
The reaction of 0.5 mmol of *o*-chlorostyrene (**1i**) with NFSI and TMSN₃ afforded 189.4 mg of **2i** (80%) as white solid (m.p. 133.1-134.5 °C), after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, *v/v*) as the eluent. **¹H NMR (CDCl₃, 400 MHz):** δ = 8.15-8.14 (m, 4H), 7.69-7.65 (m, 2H), 7.59-7.55 (m, 4H), 7.52-7.49 (m, 1H), 7.40-7.27 (m, 3H), 5.51 (dd, *J* = 10.5 Hz, 3.8 Hz, 1H), 4.05 (dd, *J* = 15.6 Hz, 10.5 Hz, 1H), 3.69 (dd, *J* = 15.6 Hz, 3.8 Hz, 1H) ppm; **¹³C NMR (CDCl₃, 100 MHz):** δ = 139.17, 134.54, 134.06, 132.92, 130.03, 129.96, 129.02, 128.65, 128.33, 127.70, 61.82, 51.57 ppm. **HRMS *m/z* (ESI)** calcd for $[C_{20}H_{17}ClN_4O_4S_2+Na]^+$ 499.0272, found 499.0276.

***N*-(2-Azido-2-(4-bromophenyl)ethyl)-*N*-(phenylsulfonyl)benzenesulfonamide^[1] (**2j**):**

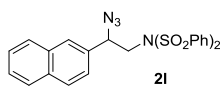


The reaction of 0.5 mmol of *p*-bromostyrene (**1j**) with NFSI and TMSN₃ afforded 214.5 mg of **2j** (83%) as white solid (m.p. 122.1-123.7 °C), after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, *v/v*) as the eluent. **¹H NMR (CDCl₃, 400 MHz):** δ = 8.03-8.01 (m, 4H), 7.67-7.63 (m, 2H), 7.56-7.49 (m, 6H), 7.22 (d, *J* = 8.4 Hz, 2H), 5.00 (dd, *J* = 9.0 Hz, 4.8 Hz, 1H), 3.98 (dd, *J* = 15.6 Hz, 9.0 Hz, 1H), 3.73 (dd, *J* = 15.6 Hz, 4.8 Hz, 1H) ppm; **¹³C NMR (CDCl₃, 100 MHz):** δ = 139.01, 135.46, 134.08, 132.27, 129.01, 128.92, 128.51, 123.10, 64.95, 52.93 ppm. **HRMS *m/z* (ESI)** calcd for $[C_{20}H_{17}BrN_4O_4S_2+Na]^+$ 542.9767, found 542.9776.

***N*-(2-([1,1'-Biphenyl]-4-yl)-2-azidoethyl)-*N*-(phenylsulfonyl)benzenesulfonamide^[1] (**2k**):**



The reaction of 0.5 mmol of 4-vinyl-1,1'-biphenyl (**1k**) with NFSI and TMSN₃ afforded 153.9 mg of **2k** (60%) as light yellow solid (m.p. 132.0-133.7 °C), after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, *v/v*) as the eluent. **¹H NMR (CDCl₃, 400 MHz):** δ = 8.07-8.05 (m, 4H), 7.65-7.58 (m, 6H), 7.54-7.50 (m, 4H), 7.47-7.35 (m, 5H), 5.07 (dd, *J* = 9.4 Hz, 4.5 Hz, 1H), 4.08 (dd, *J* = 15.6 Hz, 9.4 Hz, 1H), 3.79 (dd, *J* = 15.6 Hz, 4.5 Hz, 1H) ppm; **¹³C NMR (CDCl₃, 100 MHz):** δ = 141.95, 140.15, 139.17, 135.31, 134.01, 128.97, 128.87, 128.56, 127.75, 127.71, 127.67, 127.03, 65.32, 53.06 ppm. **HRMS *m/z* (ESI)** calcd for $[C_{26}H_{22}N_4O_4S_2+Na]^+$ 541.0975, found 541.0977.

***N*-(2-Azido-2-(naphthalen-2-yl)ethyl)-*N*-(phenylsulfonyl)benzenesulfonamide^[1] (**2l**):**View Article Online
DOI: 10.1039/C8OB00699G

The reaction of 0.5 mmol of 2-vinylnaphthalene (**1l**) with NFSI and TMSN₃ afforded

105.1 mg of **2l** (43%) as light yellow solid (m.p. 125.2-126.8 °C), after flash

chromatography on silica gel using petroleum ether and ethyl acetate (15:1, v/v) as the eluent. ¹H NMR

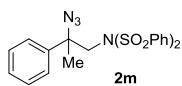
(CDCl₃, 400 MHz): δ = 8.01-7.99 (m, 4H), 7.90-7.79 (m, 4H), 7.60-7.42 (m, 9H), 5.21 (dd, *J* = 8.8 Hz, 5.0 Hz,

1H), 4.10 (dd, *J* = 15.6 Hz, 8.8 Hz, 1H), 3.88 (dd, *J* = 15.6 Hz, 5.0 Hz, 1H) ppm; ¹³C NMR (CDCl₃, 100 MHz):

δ = 139.03, 133.96, 133.65, 133.42, 133.11, 129.18, 128.88, 128.51, 128.09, 127.74, 127.12, 126.73, 126.66,

124.22, 65.73, 52.94 ppm. HRMS *m/z* (ESI) calcd for [C₂₄H₂₀N₄O₄S₂+Na]⁺ 515.0818, found 515.0825;

calcd for [C₂₄H₂₀N₄O₄S₂+K]⁺ 531.0563, found 531.0566.

***N*-(2-Azido-2-phenylpropyl)-*N*-(phenylsulfonyl)benzenesulfonamide^[1] (**2m**):**

The reaction of 0.5 mmol of prop-1-en-2-ylbenzene (**1m**) with NFSI and TMSN₃

afforded 144.8 mg of **2m** (64%) as colorless oil, after flash chromatography on silica gel

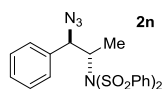
using petroleum ether and ethyl acetate (15:1, v/v) as the eluent. ¹H NMR (CDCl₃, 400 MHz): δ =

8.11-8.09 (m, 4H), 7.65-7.61 (m, 2H), 7.57-7.53 (m, 4H), 7.44-7.30 (m, 5H), 4.17 (d, *J* = 15.8 Hz, 1H), 3.99 (d, *J*

= 15.8 Hz, 1H), 1.67 (s, 3H) ppm; ¹³C NMR (CDCl₃, 100 MHz): δ = 141.43, 140.58, 133.73, 128.90, 128.85,

128.43, 128.27, 125.82, 66.92, 58.85, 21.54 ppm. HRMS *m/z* (ESI) calcd for [C₂₁H₂₀N₄O₄S₂+Na]⁺ 479.0818,

found 479.0823.

***N*-(1-Azido-1-phenylpropan-2-yl)-*N*-(phenylsulfonyl)benzenesulfonamide^[1] (**2n**):**

The reaction of 0.5 mmol of (*E*)-prop-1-en-1-ylbenzene (**1n**) with NFSI and TMSN₃

afforded 173.6 mg of **2n** (76%, dr > 94:6) as colorless oil, after flash chromatography on

silica gel using petroleum ether and ethyl acetate (15:1, v/v) as the eluent. ¹H NMR (CDCl₃, 400 MHz): δ

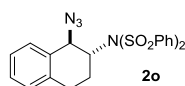
= 7.64-7.61 (m, 4H), 7.57-7.53 (m, 2H), 7.40-7.24 (m, 9H), 5.21 (d, *J* = 9.7 Hz, 1H), 4.37 (dq, *J* = 9.7 Hz, 6.7 Hz,

1H), 1.51 (d, *J* = 6.8, 3H) ppm; ¹³C NMR (CDCl₃, 100 MHz): δ = 135.87, 133.69, 129.13, 128.81, 128.73,

128.67, 128.41, 127.56, 69.71, 61.48, 17.94 ppm. HRMS *m/z* (ESI) calcd for [C₂₁H₂₀N₄O₄S₂+Na]⁺ 479.0818,

found 479.0818; calcd for [C₂₁H₂₀N₄O₄S₂+K]⁺ 495.0558, found 495.0559.

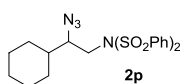
***N*-(1-Azido-1,2,3,4-tetrahydronaphthalen-2-yl)-*N*-(phenylsulfonyl)benzenesulfonamide^[1] (**2o**):**



The reaction of 0.5 mmol of 1,2-dihydronaphthalene (**1o**) with NFSI and TMSN₃ afforded 85.6 mg of **2o** (37%, dr > 97:3) as light yellow oil, after flash chromatography

on silica gel using petroleum ether and ethyl acetate (15:1, v/v) as the eluent. ¹H NMR (CDCl₃, 400 MHz): δ = 8.14-8.12 (m, 4H), 7.70-7.66 (m, 2H), 7.60-7.56 (m, 4H), 7.41 (d, *J* = 7.5 Hz, 1H), 7.27-7.20 (m, 2H), 7.08 (d, *J* = 7.2 Hz, 1H), 5.38 (d, *J* = 10.0 Hz, 1H), 4.36-4.29 (m, 1H), 2.82-2.78 (m, 2H), 2.63-2.52 (m, 1H), 1.89-1.83 (m, 1H) ppm; ¹³C NMR (CDCl₃, 100 MHz): δ = 135.66, 134.08, 134.00, 129.02, 128.69, 128.56, 128.04, 127.94, 126.83, 64.97, 62.78, 29.99, 28.44 ppm. HRMS *m/z* (ESI) calcd for [C₂₂H₂₀N₄O₄S₂+Na]⁺ 491.0818, found 491.0826.

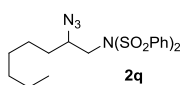
N-(2-Azido-2-cyclohexylethyl)-*N*-(phenylsulfonyl)benzenesulfonamide (**2p**):



In presence of 2,2'-bipyridine (3.9 mg, 0.025 mmol), the reaction of 0.5 mmol of vinyl cyclohexane (**1p**) with NFSI and TMSN₃ at 80°C for 6 hours afforded 104.2 mg of **2p**

(47%) as colorless oil, after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, v/v) as the eluent. ¹H NMR (CDCl₃, 400 MHz): δ = 8.12-8.09 (m, 4H), 7.68-7.65 (m, 2H), 7.59-7.55 (m, 4H), 3.89 (dd, *J* = 15.8 Hz, 10.5 Hz, 1H), 3.65-3.61 (m, 2H), 1.77-1.75 (m, 3H), 1.67-1.64 (m, 2H), 1.51-1.43 (m, 1H), 1.26-1.12 (m, 5H) ppm; ¹³C NMR (CDCl₃, 100 MHz): δ = 139.33, 133.97, 129.04, 128.40, 67.49, 50.80, 41.10, 29.80, 28.03, 26.00, 25.90, 25.73 ppm. HRMS *m/z* (ESI) calcd for [C₂₀H₂₄N₄O₄S₂+K]⁺ 487.0871, found 487.0880.

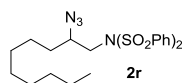
N-(2-Azido-octyl)-*N*-(phenylsulfonyl)benzenesulfonamide (**2q**):



In presence of 2,2'-bipyridine (3.9 mg, 0.025 mmol), the reaction of 0.5 mmol of 1-octene (**1q**) with NFSI and TMSN₃ at 80°C afforded 116.0 mg of **2q** (52%) as colorless

oil, after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, v/v) as the eluent. ¹H NMR (CDCl₃, 400 MHz): δ = 8.10-8.08 (m, 4H), 7.68-7.64 (m, 2H), 7.58-7.55 (m, 4H), 3.86 (dd, *J* = 15.1 Hz, 9.2 Hz, 1H), 3.75-3.69 (m, 1H), 3.57 (dd, *J* = 15.1 Hz, 3.9 Hz, 1H), 1.50-1.42 (m, 2H), 1.32-1.23 (m, 8H), 0.89 (t, *J* = 6.9 Hz, 3H) ppm; ¹³C NMR (CDCl₃, 100 MHz): δ = 139.40, 134.01, 129.06, 128.41, 62.00, 52.15, 32.29, 31.52, 28.86, 25.80, 22.47, 13.99 ppm. HRMS *m/z* (ESI) calcd for [C₂₀H₂₆N₄O₄S₂+Na]⁺ 473.1288, found 473.1298.

N-(2-Azido-decyl)-*N*-(phenylsulfonyl)benzenesulfonamide (**2r**):

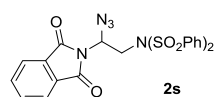


In presence of 2,2'-bipyridine (3.9 mg, 0.025 mmol), the reaction of 0.5 mmol of

1-decene (**1r**) with NFSI and TMSN₃ at 80°C afforded 94.3 mg of **2r** (40%) as colorless oil, after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, v/v) as the eluent.

¹H NMR (CDCl₃, 400 MHz): δ = 8.10-8.08 (m, 4H), 7.68-7.64 (m, 2H), 7.58-7.54 (m, 4H), 3.86 (dd, *J* = 15.1 Hz, 9.3 Hz, 1H), 3.75-3.69 (m, 1H), 3.56 (dd, *J* = 15.1 Hz, 3.9 Hz, 1H), 1.50-1.43 (m, 2H), 1.32-1.26 (m, 12H), 0.89 (t, *J* = 6.9 Hz, 3H) ppm; **¹³C NMR (CDCl₃, 100 MHz):** δ = 139.31, 133.99, 129.02, 128.35, 61.96, 52.09, 32.23, 31.71, 29.25, 29.15, 29.07, 25.79, 22.56, 14.03 ppm. **HRMS *m/z* (ESI)** calcd for [C₂₂H₃₀N₄O₄S₂+Na]⁺ 501.1606, found 501.1607; calcd for [C₂₂H₃₀N₄O₄S₂+Na]⁺ 501.1601, found 501.1607.

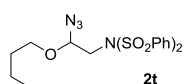
***N*-(2-Azido-2-(1,3-dioxoisindolin-2-yl)ethyl)-*N*-(phenylsulfonyl)benzenesulfonamide (2s):**



In presence of 2,2'-bipyridine (3.9 mg, 0.025 mmol), the reaction of 0.5 mmol of *N*-vinylphthalimide (**1s**) with NFSI and TMSN₃ at 80°C afforded 172.5 mg of **2s** (68%)

as white solid (m.p. 177.3-178.6 °C), after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1 to 3:1, v/v) as the eluent. **¹H NMR (CDCl₃, 400 MHz):** δ = 8.05-8.03 (m, 4H), 7.90-7.88 (m, 2H), 7.77-7.75 (m, 2H), 7.67-7.63 (m, 2H), 7.57-7.53 (m, 4H), 6.04 (dd, *J* = 7.3 Hz, 4.9 Hz, 1H), 4.42-4.30 (m, 2H) ppm; **¹³C NMR (CDCl₃, 100 MHz):** δ = 166.97, 138.34, 134.60, 134.29, 131.35, 129.14, 128.62, 123.91, 65.30, 47.32 ppm. **HRMS *m/z* (ESI)** calcd for [C₂₂H₁₇N₅O₆S₂+K]⁺ 550.0252, found 550.0258.

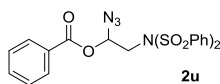
***N*-(2-Azido-2-butoxyethyl)-*N*-(phenylsulfonyl)benzenesulfonamide (2t):**



In presence of 2,2'-bipyridine (3.9 mg, 0.025 mmol), the reaction of 0.5 mmol of 1-(vinylloxy)butane (**1t**) with NFSI and TMSN₃ at 80°C for 6 hours afforded 111.2 mg of

2t (51%) as colorless oil, after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, v/v) as the eluent. **¹H NMR (CDCl₃, 400 MHz):** δ = 8.11-8.09 (m, 4H), 7.69-7.65 (m, 2H), 7.59-7.55 (m, 4H), 4.81 (t, *J* = 6.0 Hz, 1H), 3.93 (dd, *J* = 15.7 Hz, 6.1 Hz, 1H), 3.80 (dd, *J* = 15.7 Hz, 5.8 Hz, 1H), 3.67 (dt, *J* = 9.5 Hz, 6.9 Hz, 1H), 3.40 (dt, *J* = 9.5 Hz, 6.6 Hz, 1H), 1.50-1.43 (m, 2H), 1.34-1.26 (m, 2H), 0.90 (t, *J* = 7.3 Hz, 3H) ppm; **¹³C NMR (CDCl₃, 100 MHz):** δ = 139.17, 134.03, 128.97, 128.55, 90.33, 69.82, 50.54, 31.19, 19.00, 13.73 ppm. **HRMS *m/z* (ESI)** calcd for [C₁₈H₂₂N₄O₅S₂+Na]⁺ 461.0924, found 461.0930.

1-Azido-2-(*N*-(phenylsulfonyl)phenylsulfonamido)ethyl benzoate (2u):



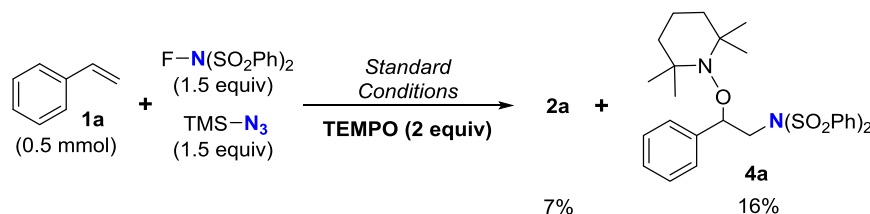
In presence of 2,2'-bipyridine (3.9 mg, 0.025 mmol), the reaction of 0.5 mmol of vinyl benzoate (**1u**) with NFSI and TMSN₃ at 80°C for 6 hours afforded 142.3 mg of

2u (59%) as colorless oil, after flash chromatography on silica gel using petroleum ether and ethyl acetate (15:1, v/v) as the eluent. **¹H NMR (CDCl₃, 400 MHz):** δ = 8.06-8.02 (m, 6H), 7.62-7.58 (m, 3H), 7.50-7.43 (m, 6H), 6.42 (t, *J* = 6.3 Hz, 1H), 3.80 (dq, *J* = 15.5 Hz, 6.3 Hz, 2H) ppm; **¹³C NMR (CDCl₃, 100 MHz):** δ = 165.66, 139.11, 134.14, 133.87, 130.09, 129.13, 128.52, 128.35, 128.24, 83.39, 49.56 ppm. **HRMS *m/z* (ESI)** calcd for [C₂₁H₁₈N₄O₆S₂+K]⁺ 525.0299, found 525.0309.

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DOI: 10.1039/C8OB00699G

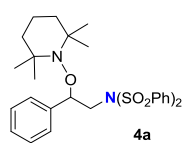
Control Experiments

1) TEMPO-suppressed Control Experiment



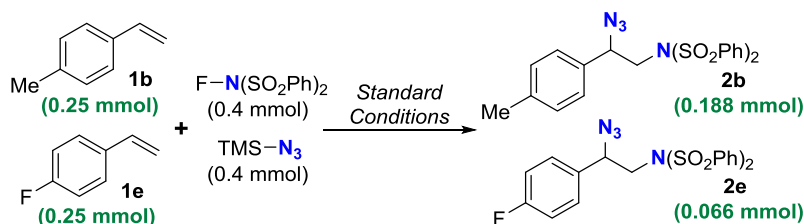
Typical Procedure: To a reaction tube charged with FeCl_2 (6.3 mg, 0.05 mmol), NFSI (236.5 mg, 0.75 mmol) and TEMPO (156.3 mg, 1 mmol) was added a solution of styrene (**1a**, 57.5 μL , 0.5 mmol), TMSN_3 (98.6 μL , 0.75 mmol) in DCE (2 mL) via a syringe under N_2 (1 atm). The reaction mixture was stirred at 100 $^\circ\text{C}$ for 3 hours. After rapidly cooling by ice, the mixture was diluted with ethyl acetate, filtered through a celite pad, and concentrated *in vacuo* to give dark residue, which was then purified by flash chromatography using petroleum ether and ethyl acetate as the eluent on silica gel to afford aminoazidated product **2a** (15.1 mg, 7%) and TEMPO-captured product **4a** (43.4 mg, 16%).

***N*-(2-Phenyl-2-((2,2,6,6-tetramethylpiperidin-1-yl)oxy)ethyl)-*N*-(phenylsulfonyl)benzenesulfonamide^[2] (**4a**):**



The reaction of 0.5 mmol of styrene (**1a**) under standard conditions, adding 1 mmol of TEMPO afforded 43.4 mg of **4a** (16%) as white solid (m.p. 132.3-133.7 $^\circ\text{C}$), after flash chromatography on silica gel using petroleum ether and ethyl acetate (20:1 to 15:1, v/v) as the eluent. **^1H NMR (CDCl_3 , 400 MHz):** δ = 7.60-7.56 (m, 6H), 7.43-7.39 (m, 4H), 7.36-7.30 (m, 5H), 5.23 (dd, J = 11.0 Hz, 4.7 Hz, 1H), 4.39 (dd, J = 14.9 Hz, 11.0 Hz, 1H), 4.03 (dd, J = 14.9 Hz, 4.7 Hz, 1H), 1.51-1.39 (m, 5H), 1.27-1.24 (m, 4H), 1.11-0.97 (m, 6H), 0.88-0.70 (m, 3H) ppm; **^{13}C NMR (CDCl_3 , 100 MHz):** δ = 139.90, 138.49, 133.58, 129.28, 128.81, 128.62, 128.12, 127.83, 84.10, 59.95, 50.09, 40.50, 34.90, 34.23, 20.22, 17.11 ppm. **HRMS m/z (ESI)** calcd for $[\text{C}_{29}\text{H}_{36}\text{N}_2\text{O}_5\text{S}_2+\text{H}]^+$ 557.2138, found 557.2144.

2) Competition Experiment between Electron-rich/-deficient Styrenes



To a reaction tube charged with FeCl_2 (6.3 mg, 0.05 mmol) and NFSI (126.1 mg, 0.4 mmol) was added a solution of *p*-methylstyrene (**1b**, 32.9 μL , 0.25 mmol), *p*-fluorostyrene (**1e**, 29.8 μL , 0.25 mmol), TMSN_3 (52.6 μL , 0.4 mmol) in DCE (2 mL) via a syringe under N_2 (1 atm). The reaction mixture was stirred at 100 $^\circ\text{C}$ for 3 hours. After rapidly cooling by ice, the mixture was diluted with ethyl acetate, filtered through a celite pad, and concentrated *in vacuo* to give dark residue, which was then purified by flash chromatography using petroleum ether and ethyl acetate as the eluent (15:1, v/v) on silical gel to afford 116.0 mg of the combined methyl- and fluoro-aminoazidated product **2b** and **2e**. The average ratio of **2b/2e** was determined by ^1H -NMR as 1 : 0.35, as shown in Figure S2. Herein, this competition reaction afforded 0.188 mmol of **2b** and 0.066 mmol of **2e**.

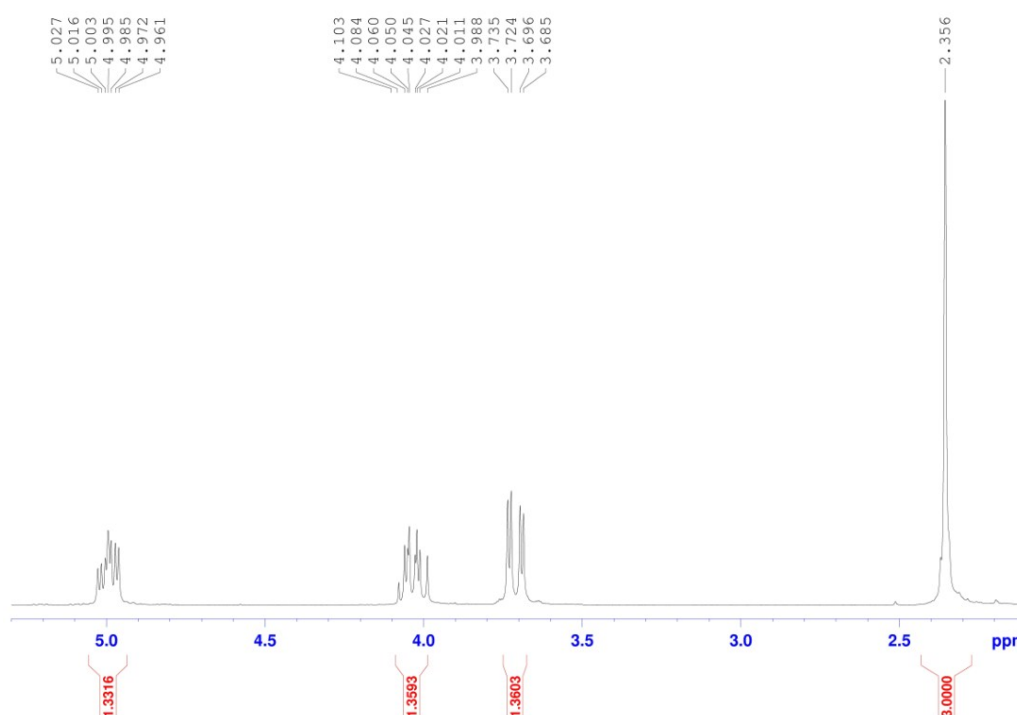
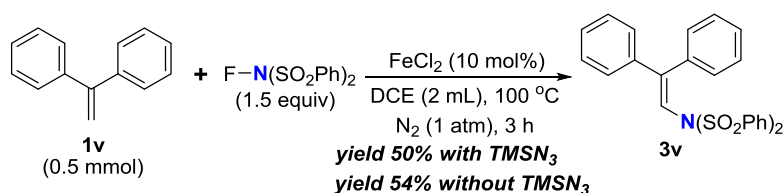


Figure S2. ^1H -NMR of the competition experiment between styrenes **1b** and **1e**.

3) C-H Amination of 1,1-disubstituted Alkenes with NFSI



Typical Procedure: To a reaction tube charged with FeCl_2 (6.3 mg, 0.05 mmol) and NFSI (236.5 mg, 0.75 mmol) was added a solution of 1,1-diphenylethylene (**1v**, 88.3 μL , 0.5 mmol) and TMSN_3 (98.6 μL , 0.75 mmol, or 0 μL , 0 mmol) in DCE (2 mL) via a syringe under N_2 (1 atm). The reaction mixture was stirred at

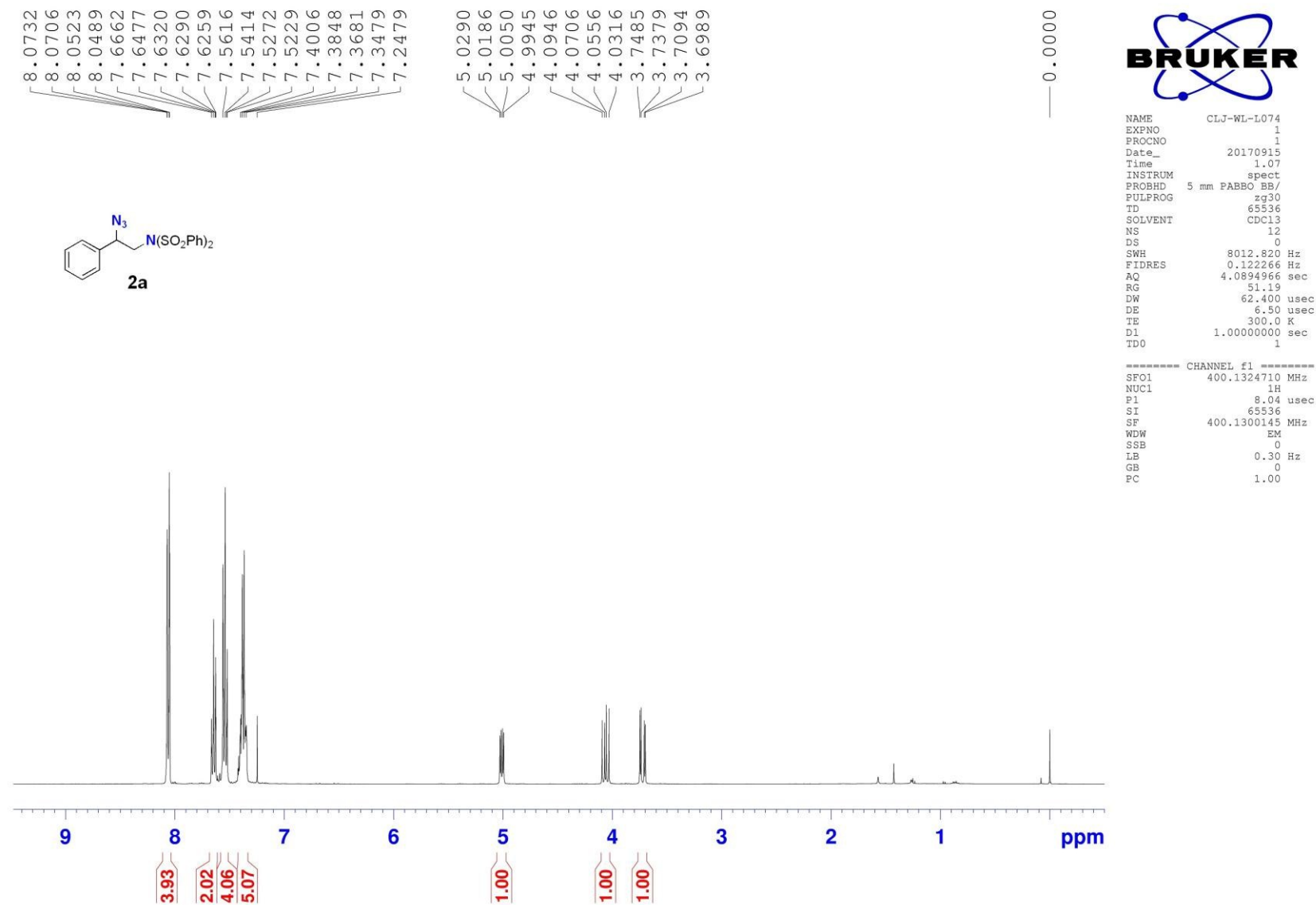
100 °C for 3 hours. After rapidly cooling by ice, the mixture was diluted with ethyl acetate, filtered through a celite pad, and concentrated *in vacuo* to give dark residue, which was then purified by flash chromatography using petroleum ether and ethyl acetate (15:1, *v/v*) as the eluent on silical gel to afford C-H aminated product **3v** (117.8 mg, 50% with TMSN₃, or 128.2 mg, 50% without TMSN₃) as white solid (m.p. 170.5-171.9 °C).

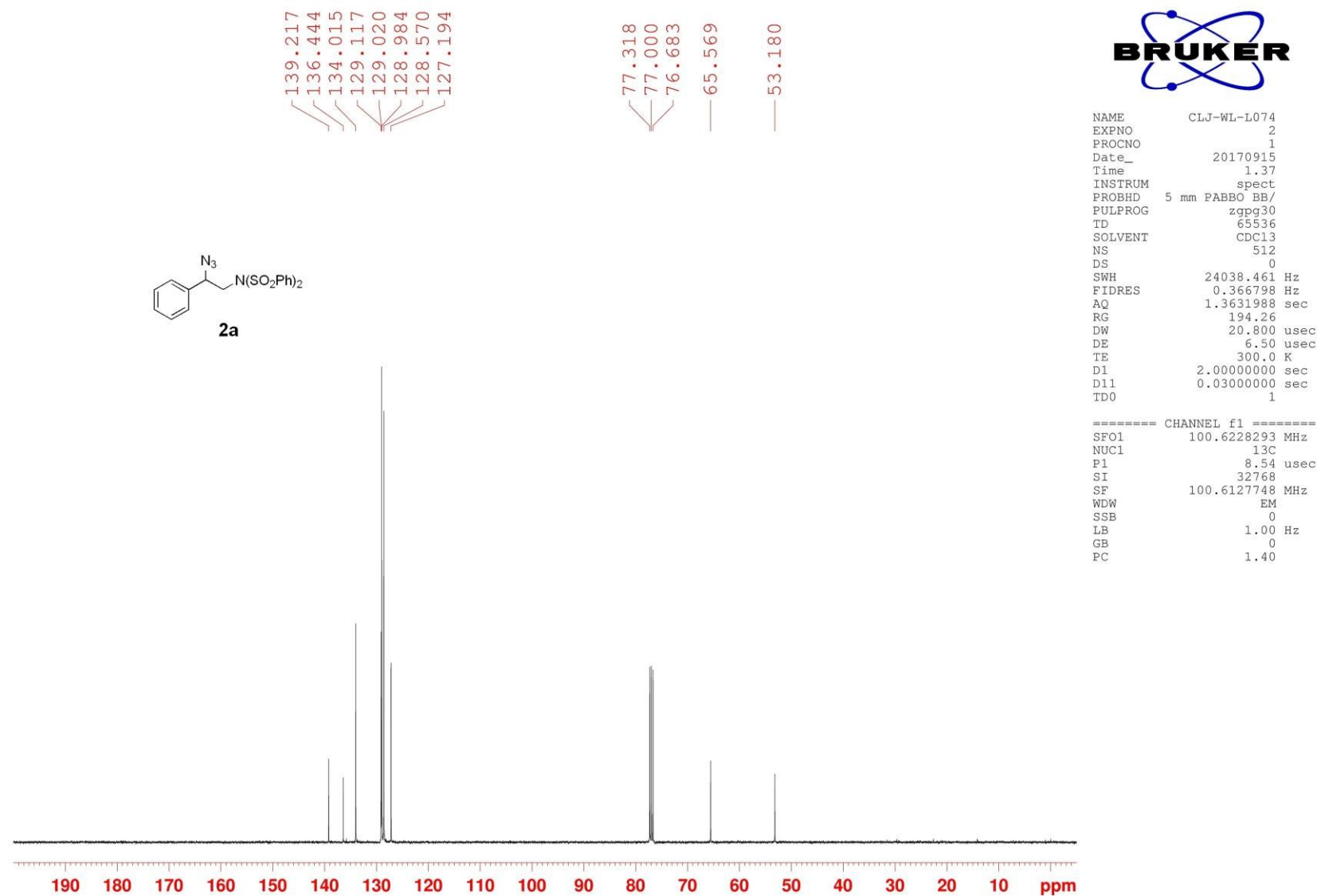
***N*-(2,2-Diphenylvinyl)-*N*-(phenylsulfonyl)benzenesulfonamide^[3] (**3v**):**

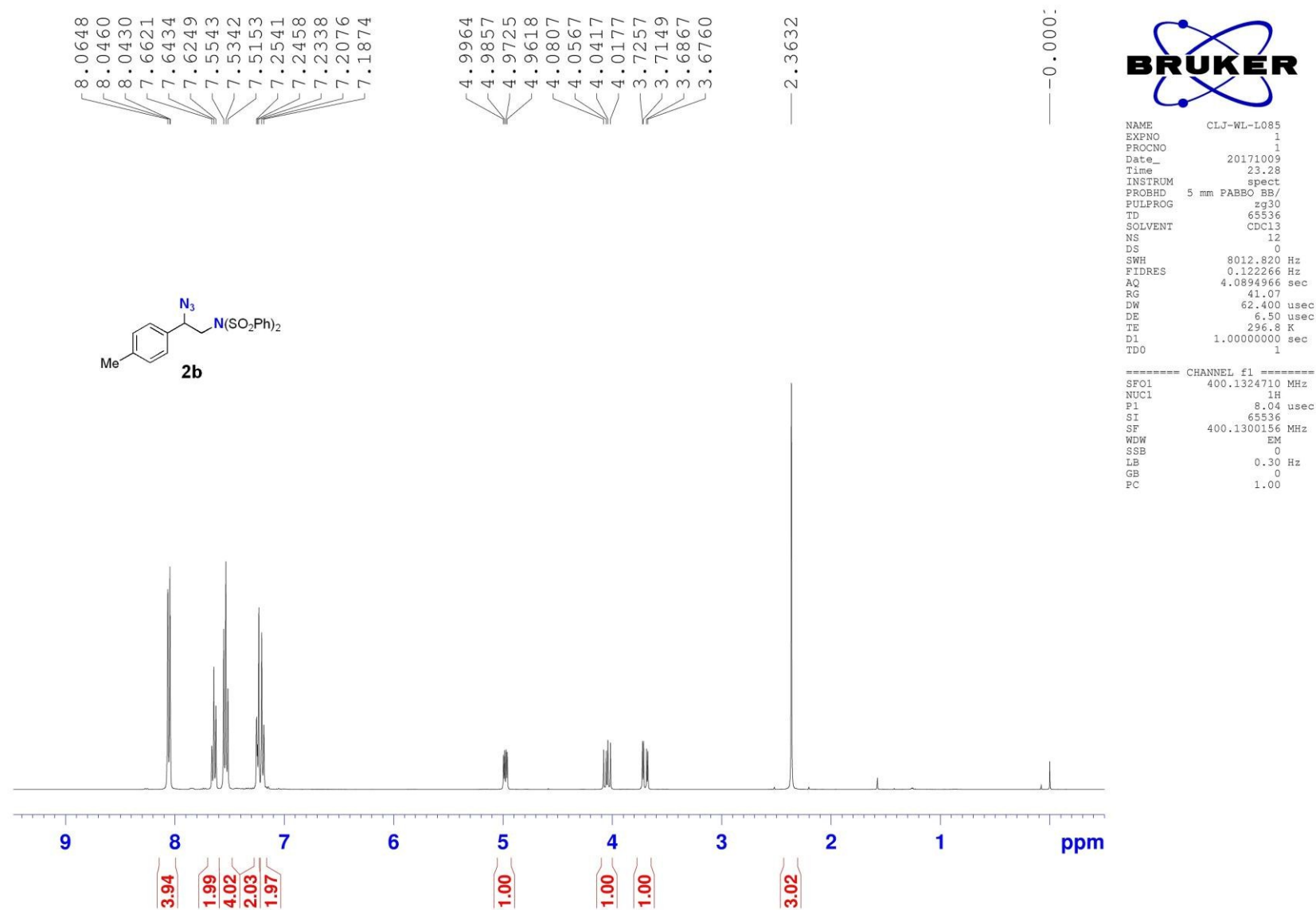
¹H NMR (CDCl₃, 400 MHz): δ = 7.71-7.69 (m, 4H), 7.58-7.55 (m, 2H), 7.41-7.36 (m, 4H), 7.34-7.20 (m, 10H), 6.13 (s, 1H) ppm; **¹³C NMR (CDCl₃, 100 MHz):** δ = 152.25, 139.82, 138.63, 136.73, 133.78, 129.87, 129.07, 128.76, 128.69, 128.64, 128.35, 128.23, 128.13, 116.26 ppm. **HRMS *m/z* (ESI)** calcd for [C₂₆H₂₁NO₄S₂+Na]⁺ 498.0810, found 498.0818.

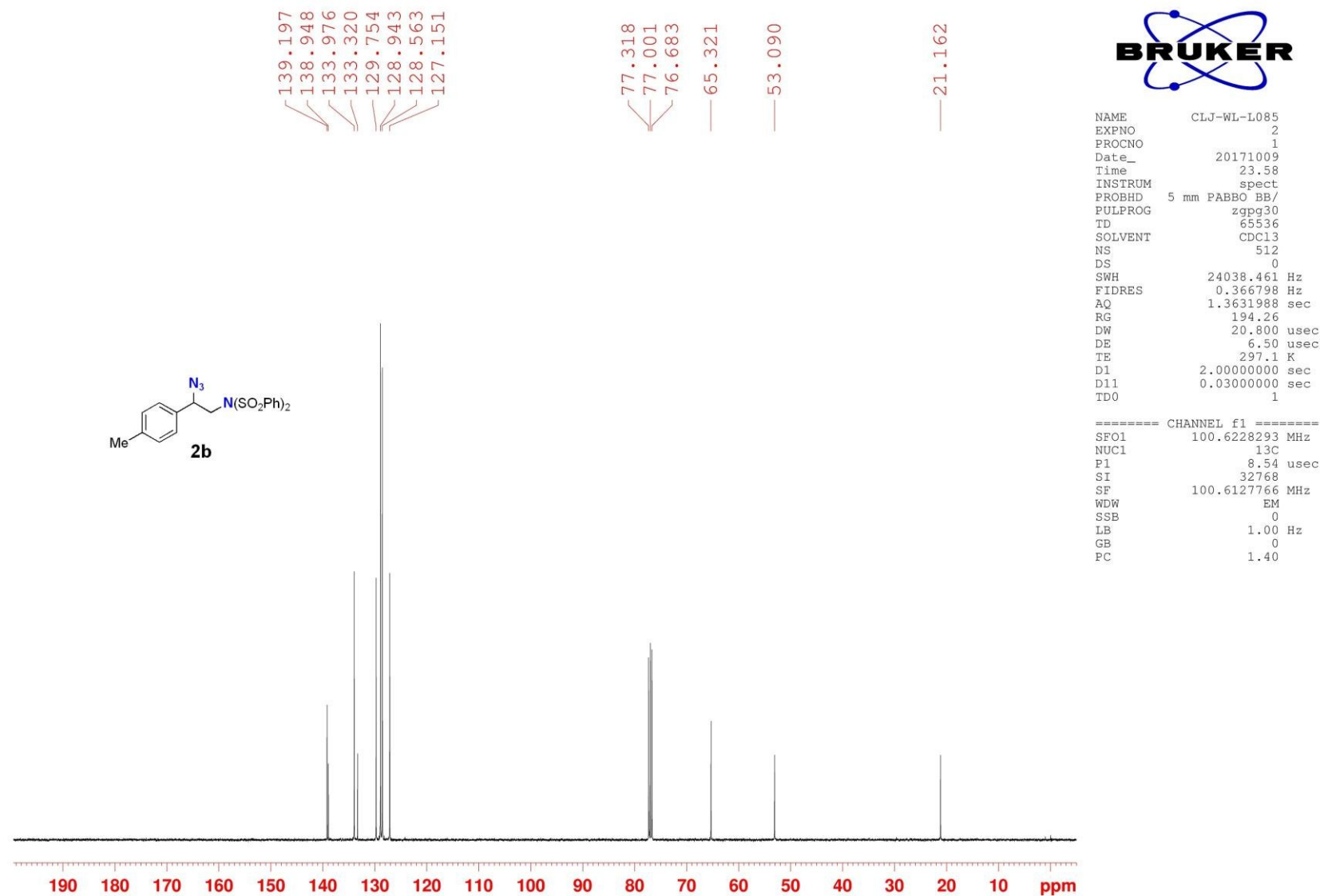
References

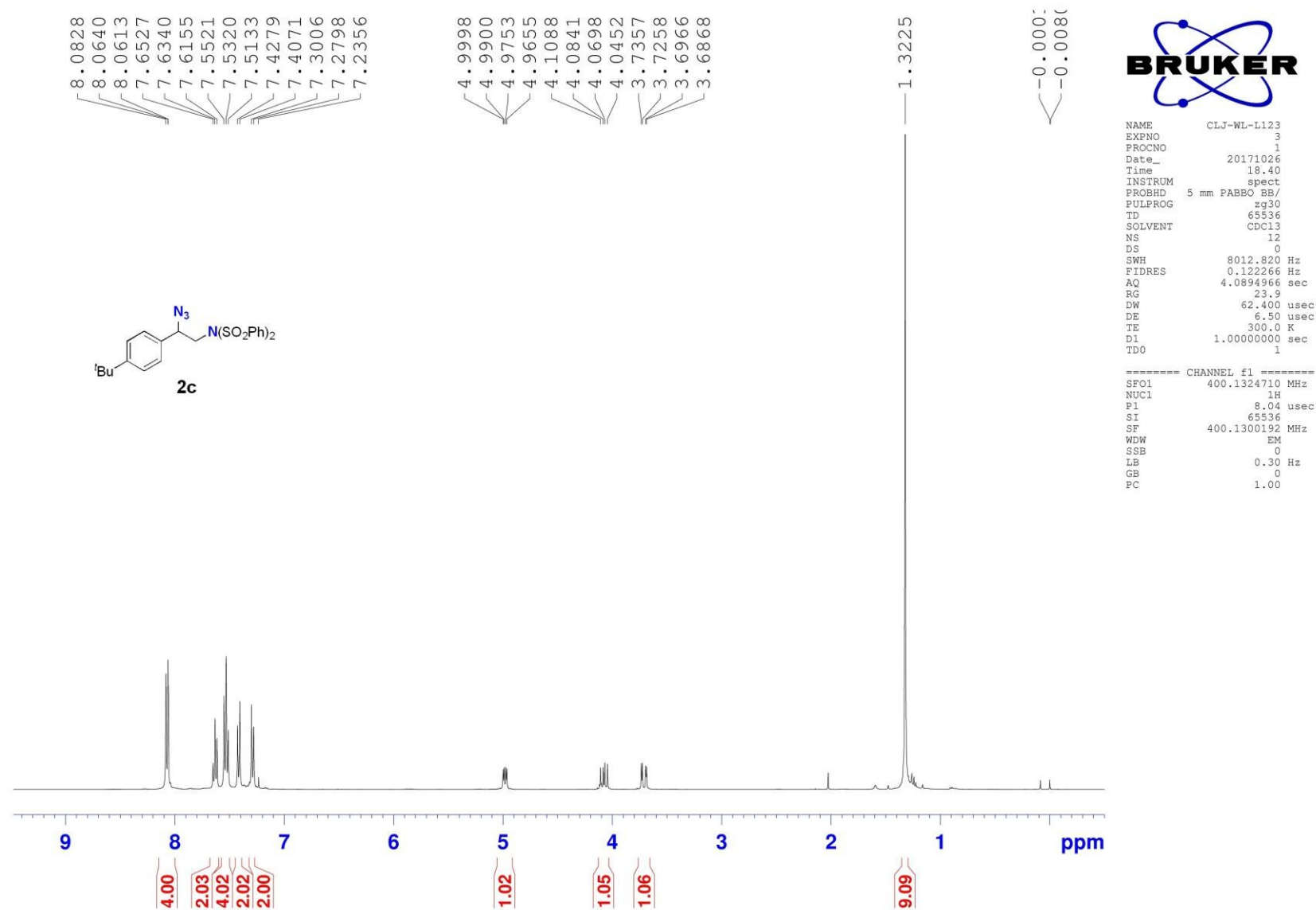
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- [2] Zhang, H.; Song, Y.; Zhao, J.; Zhang, J.; Zhang, Q. *Angew. Chem. Int. Ed.* **2014**, *53*, 11079-11083.
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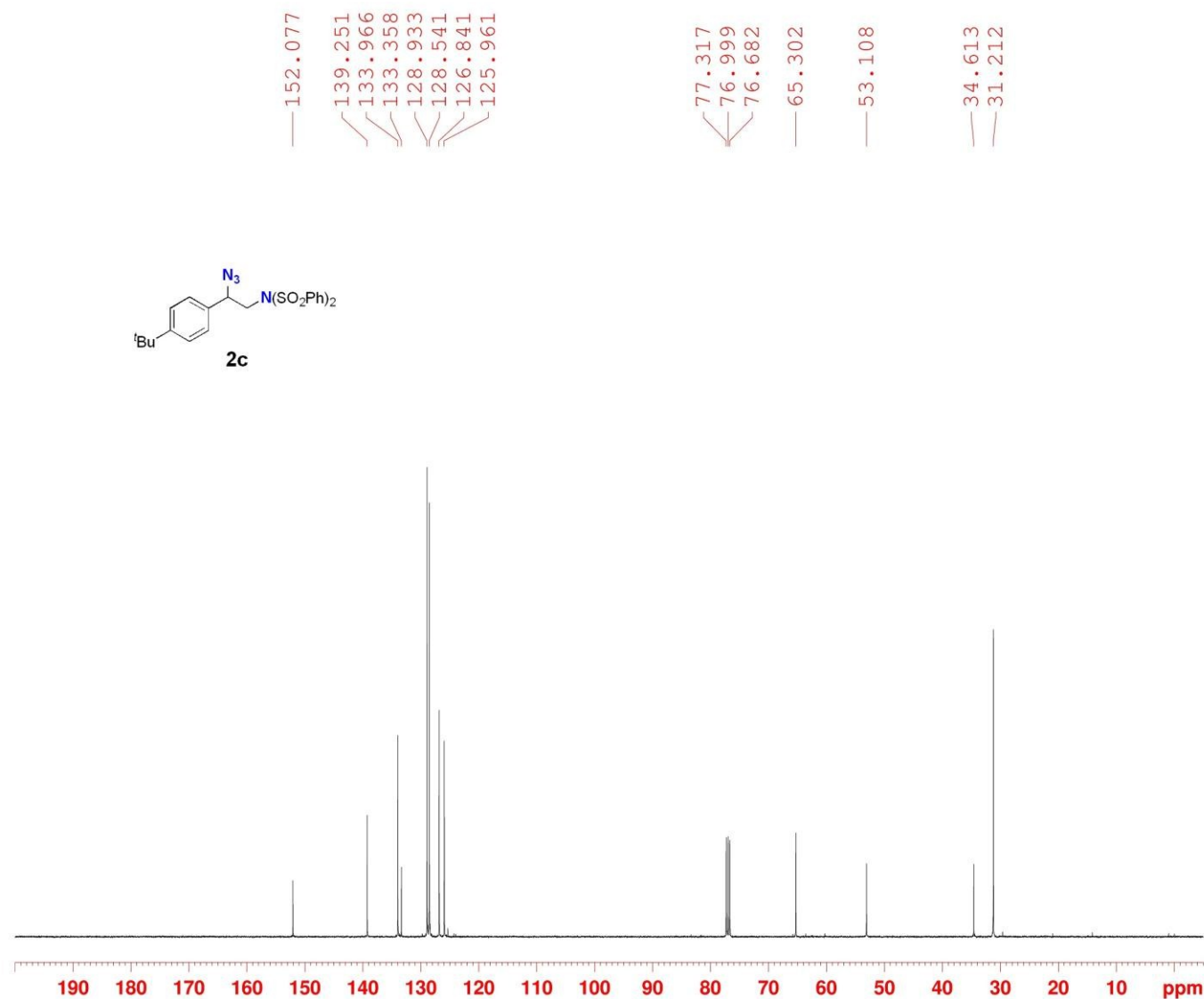






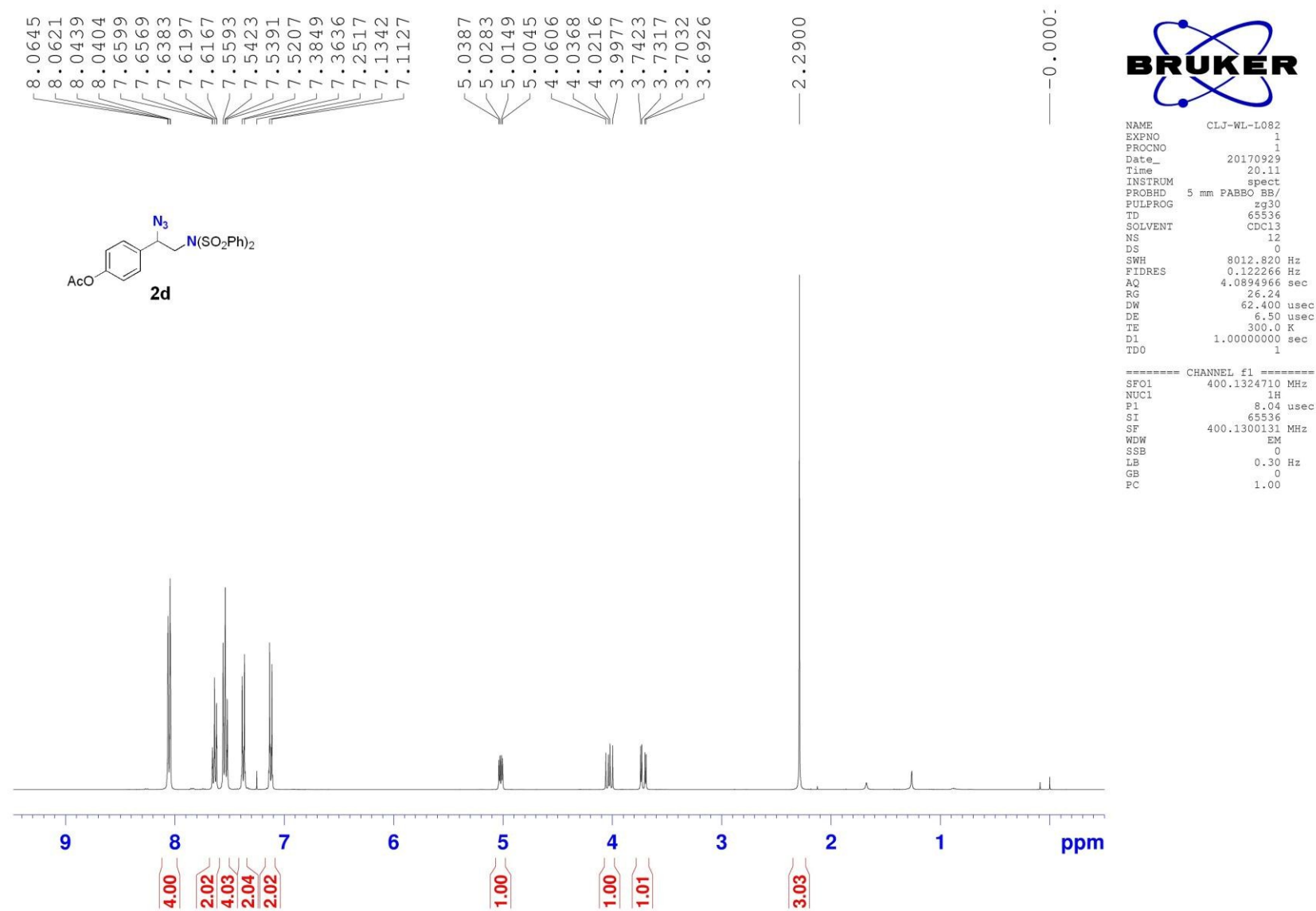


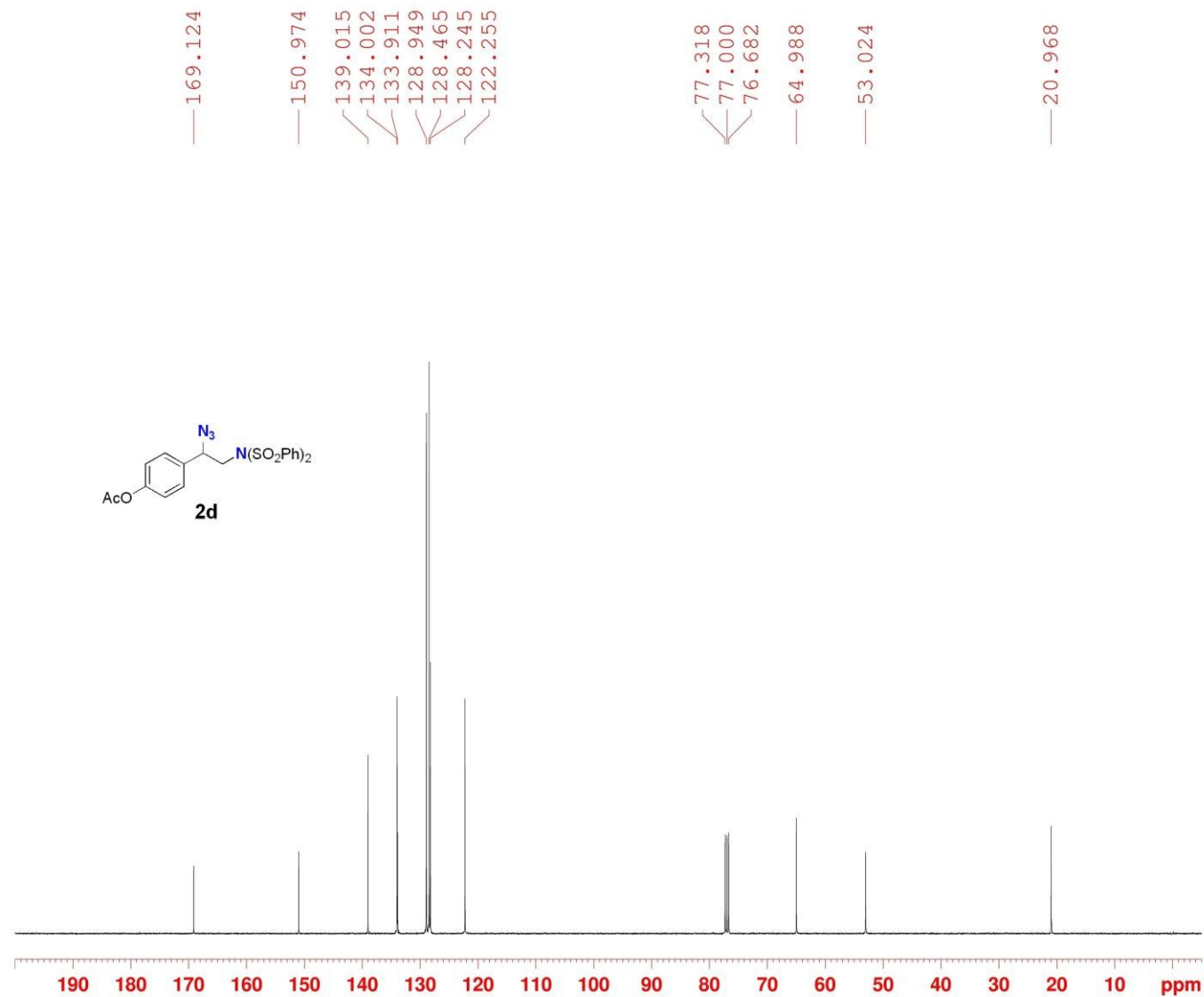




NAME CLJ-WL-L123
EXPNO 4
PROCNO 1
Date_ 20171026
Time 19.11
INSTRUM spect
PROBHD 5 mm PABBO BB/
PULPROG zgpg30
TD 65536
SOLVENT CDCl3
NS 512
DS 0
SWH 24038.461 Hz
FIDRES 0.366798 Hz
AQ 1.3631988 sec
RG 194.26
DW 20.800 usec
DE 6.50 usec
TE 300.0 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1

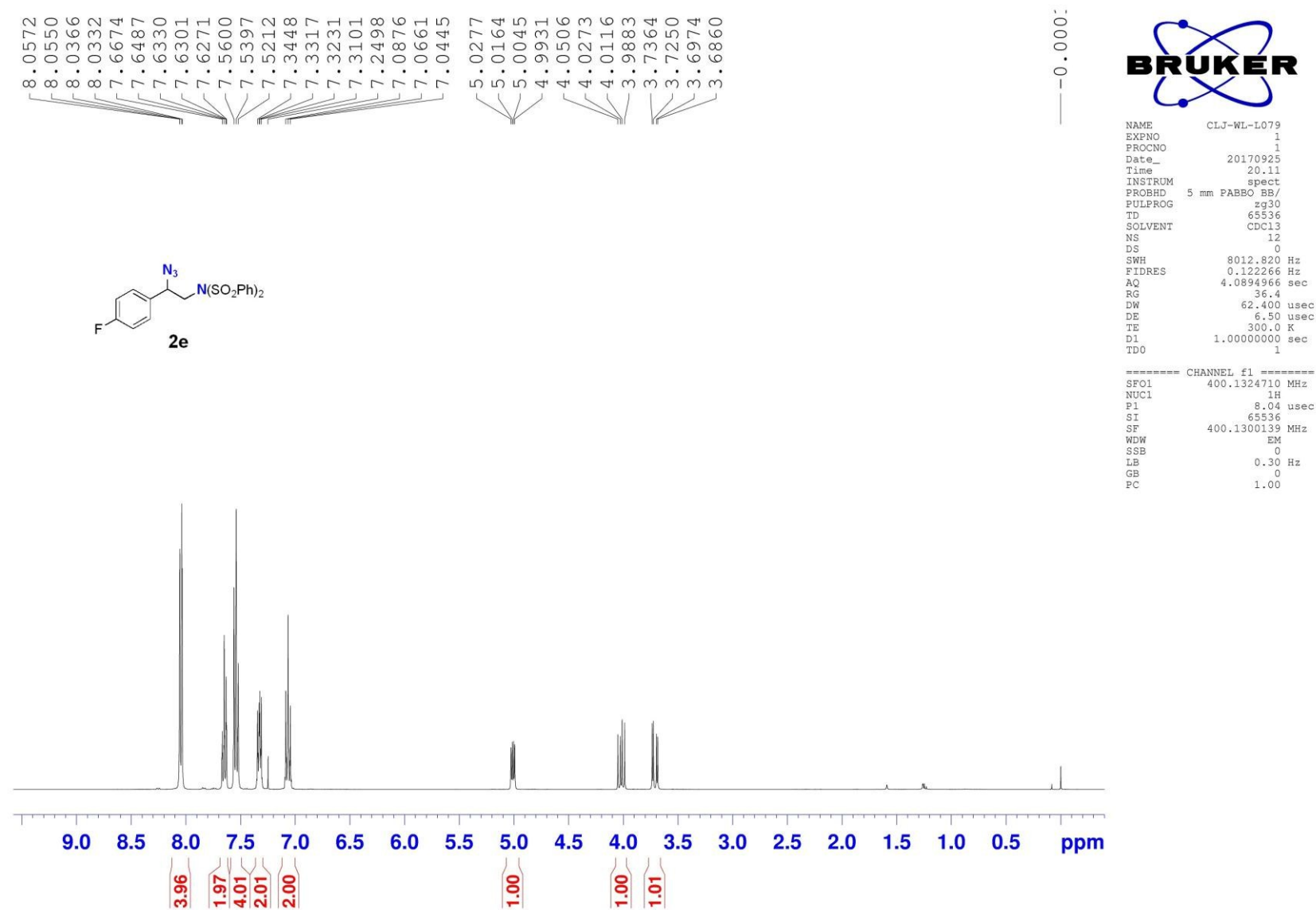
===== CHANNEL f1 =====
SFO1 100.6228293 MHz
NUC1 13C
P1 8.54 usec
SI 32768
SF 100.6127793 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40

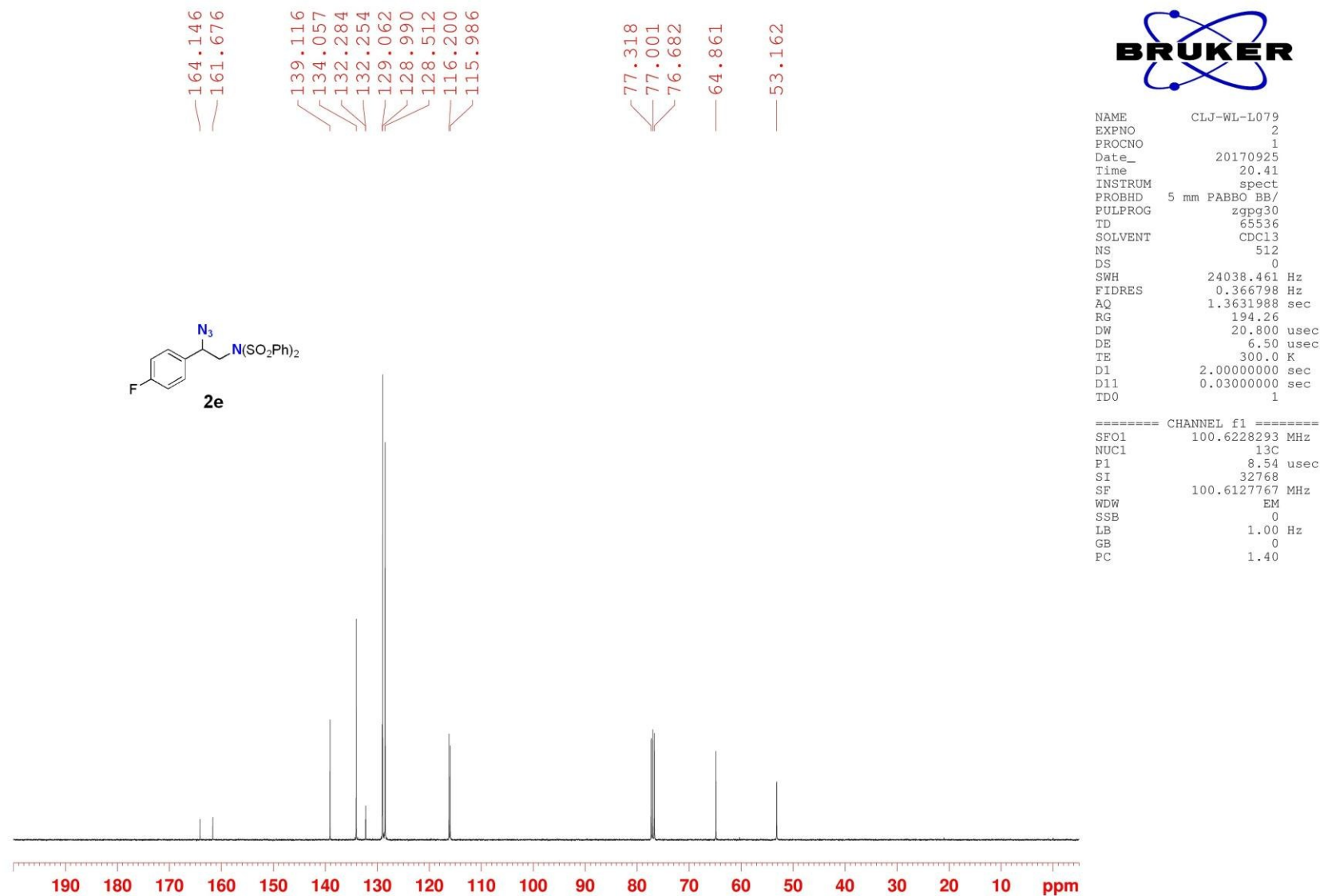


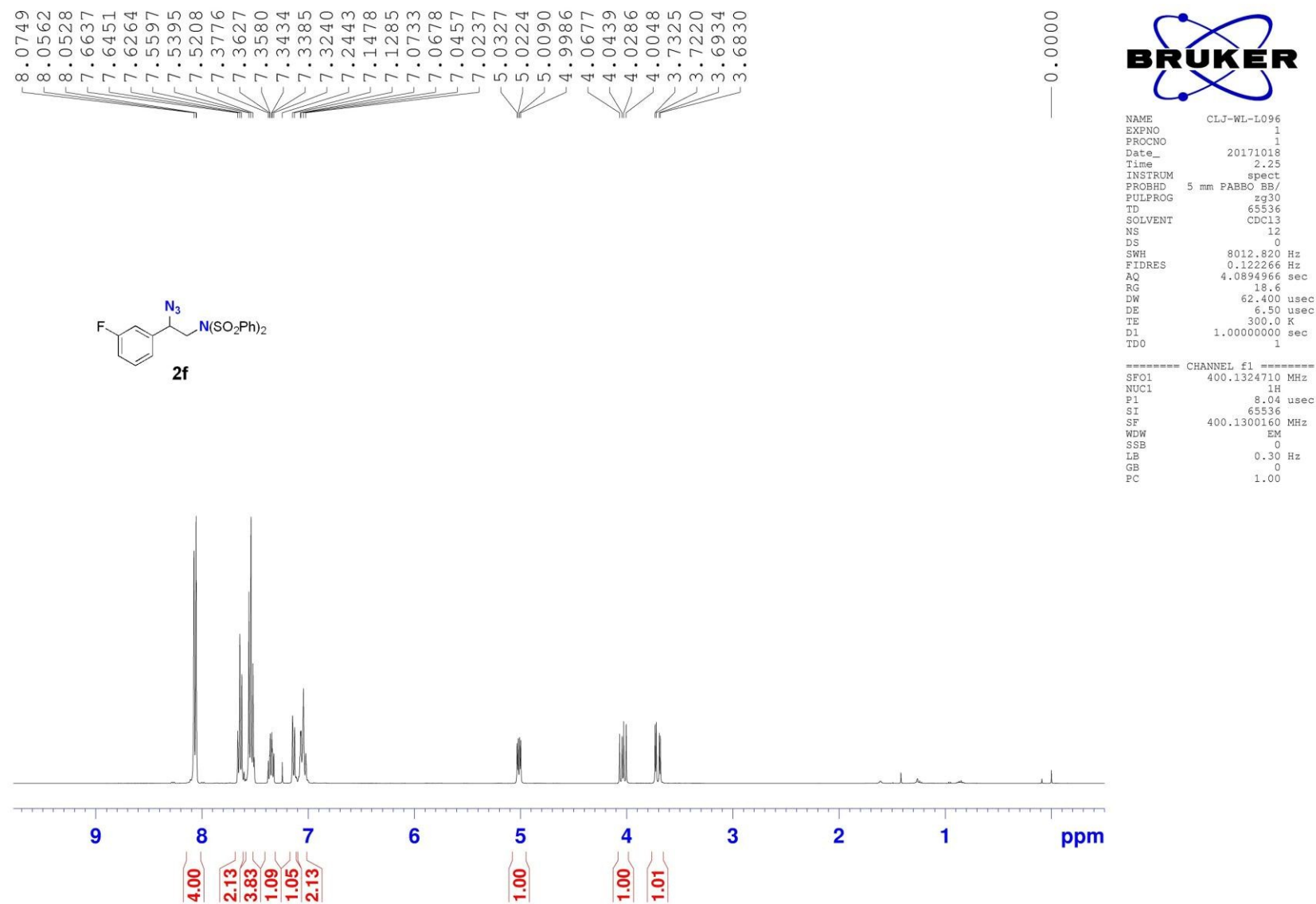


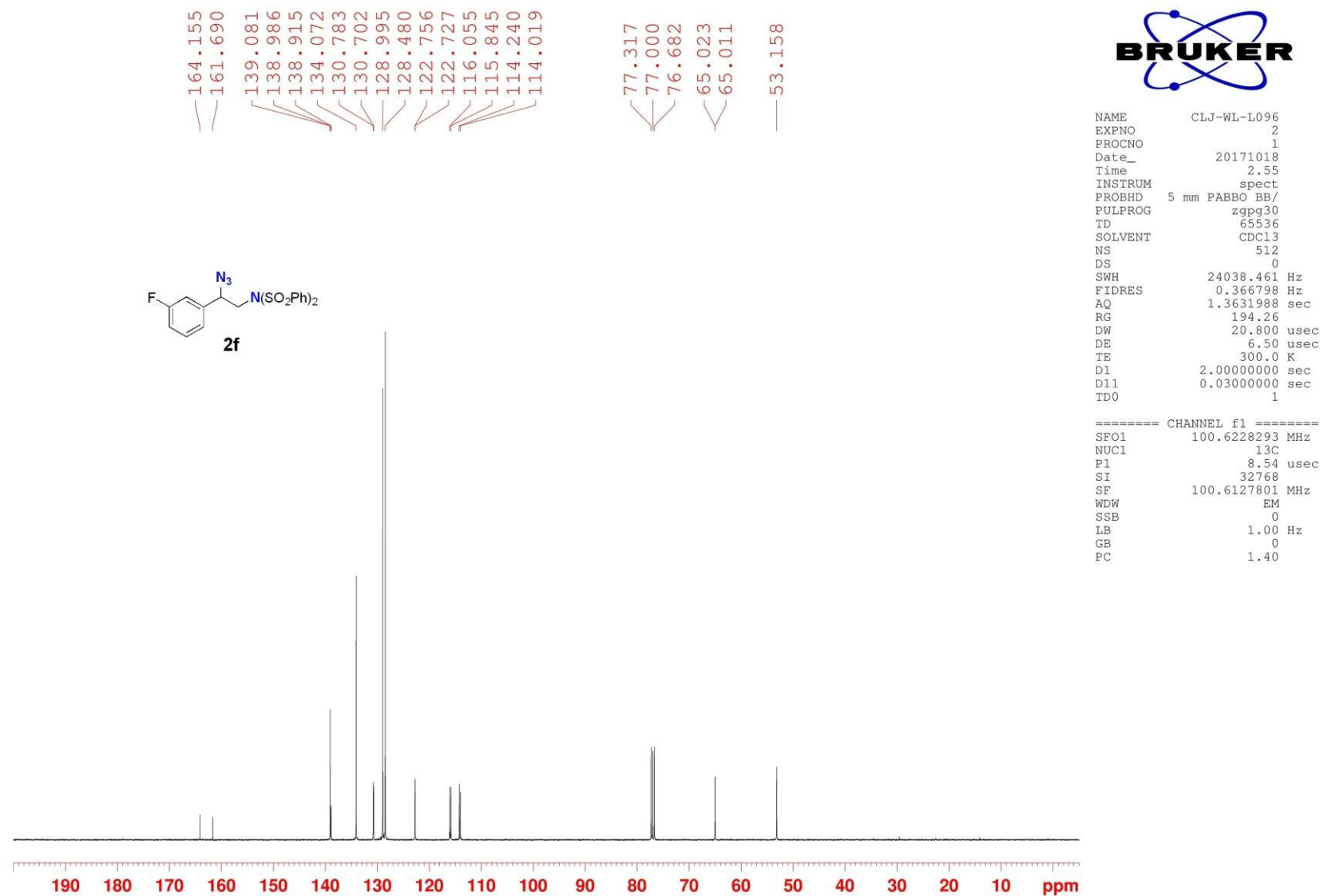
NAME CLJ-WL-L082
 EXPNO 2
 PROCNO 1
 Date_ 20170929
 Time 20.41
 INSTRUM spect
 PROBHD 5 mm PABBO BB/
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 512
 DS 0
 SWH 24038.461 Hz
 FIDRES 0.366798 Hz
 AQ 1.3631988 sec
 RG 194.26
 DW 20.800 usec
 DE 6.50 usec
 TE 300.0 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 1

===== CHANNEL f1 =====
 SFO1 100.6228293 MHz
 NUC1 13C
 P1 8.54 usec
 SI 32768
 SF 100.6127839 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40

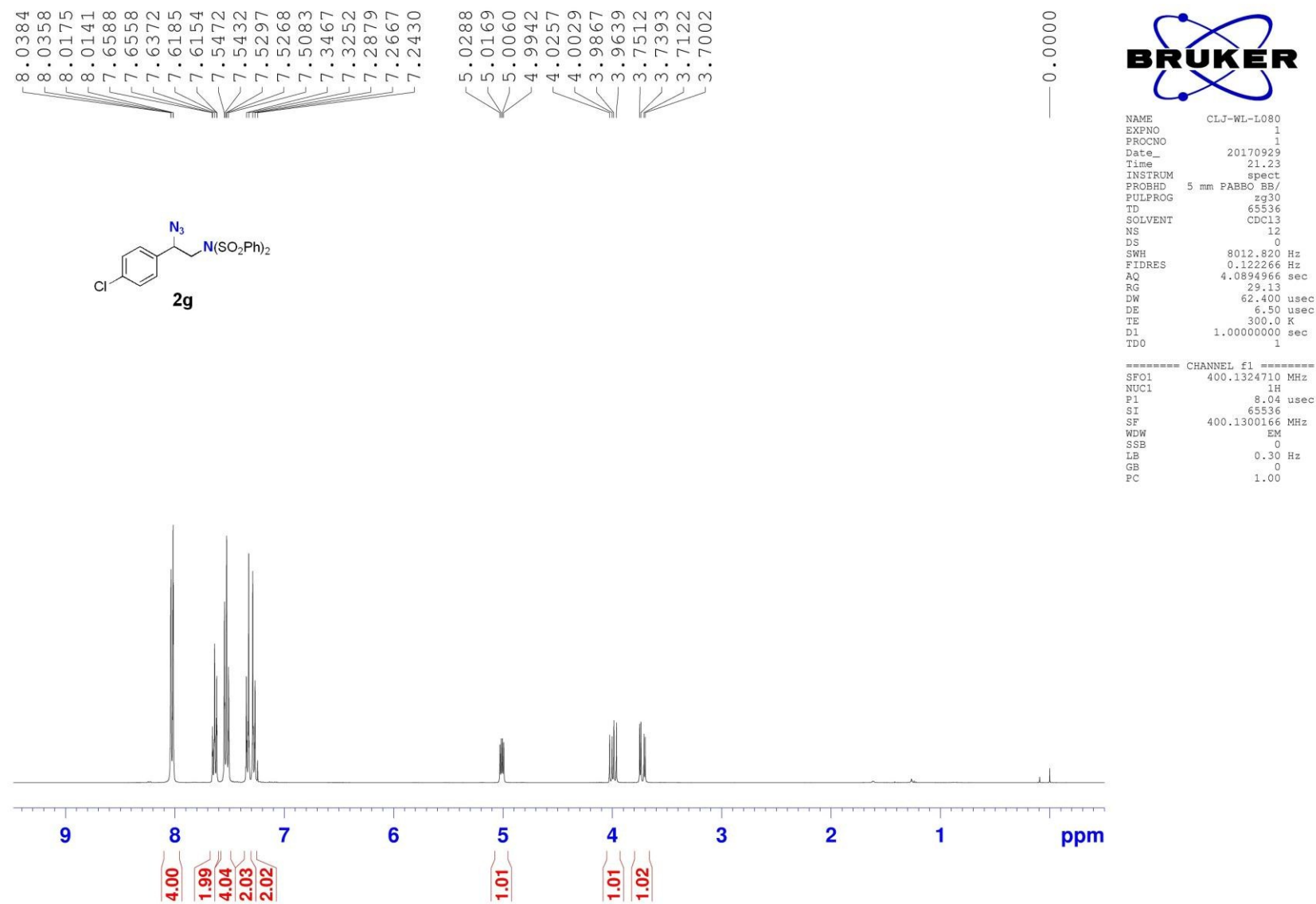


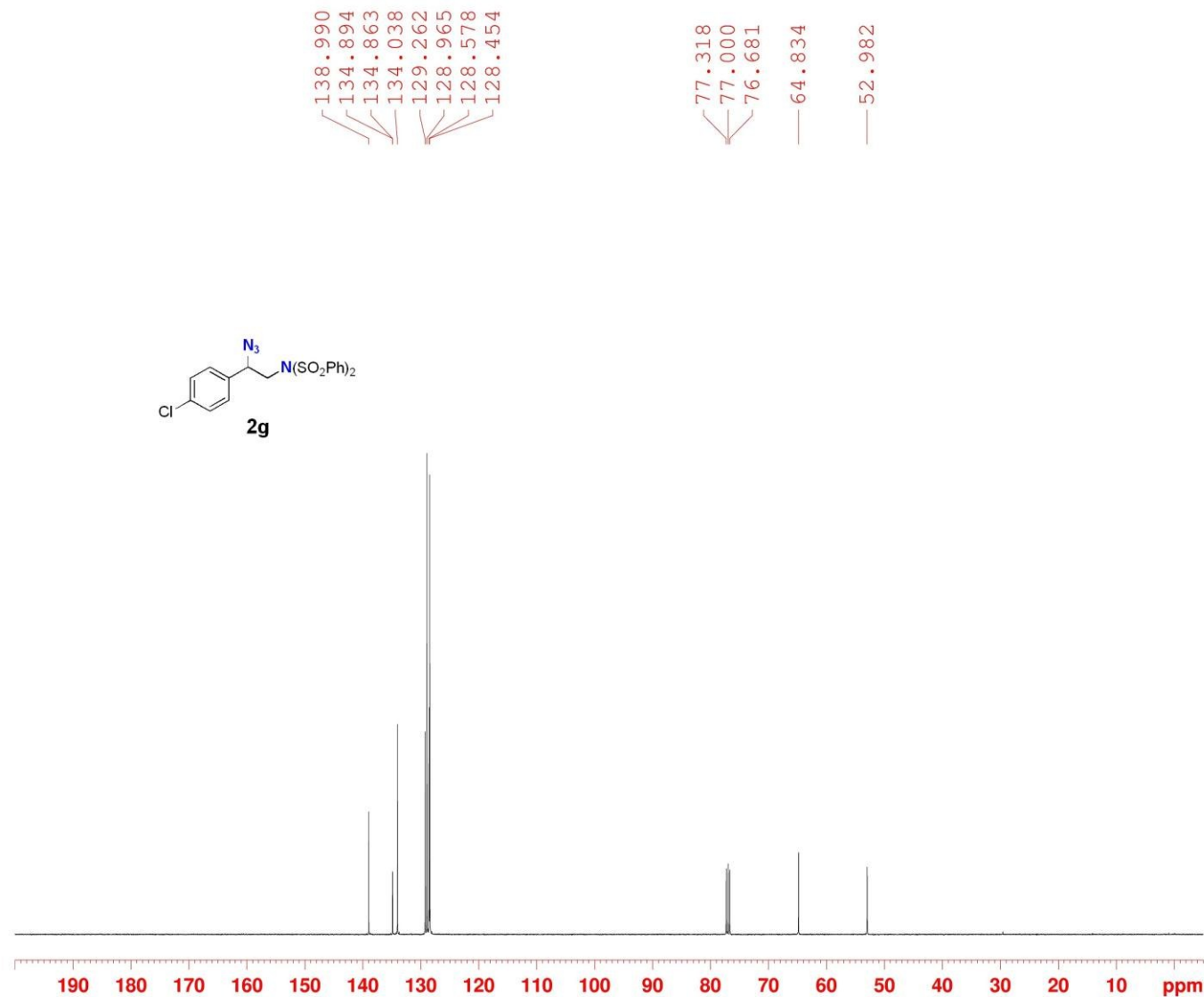






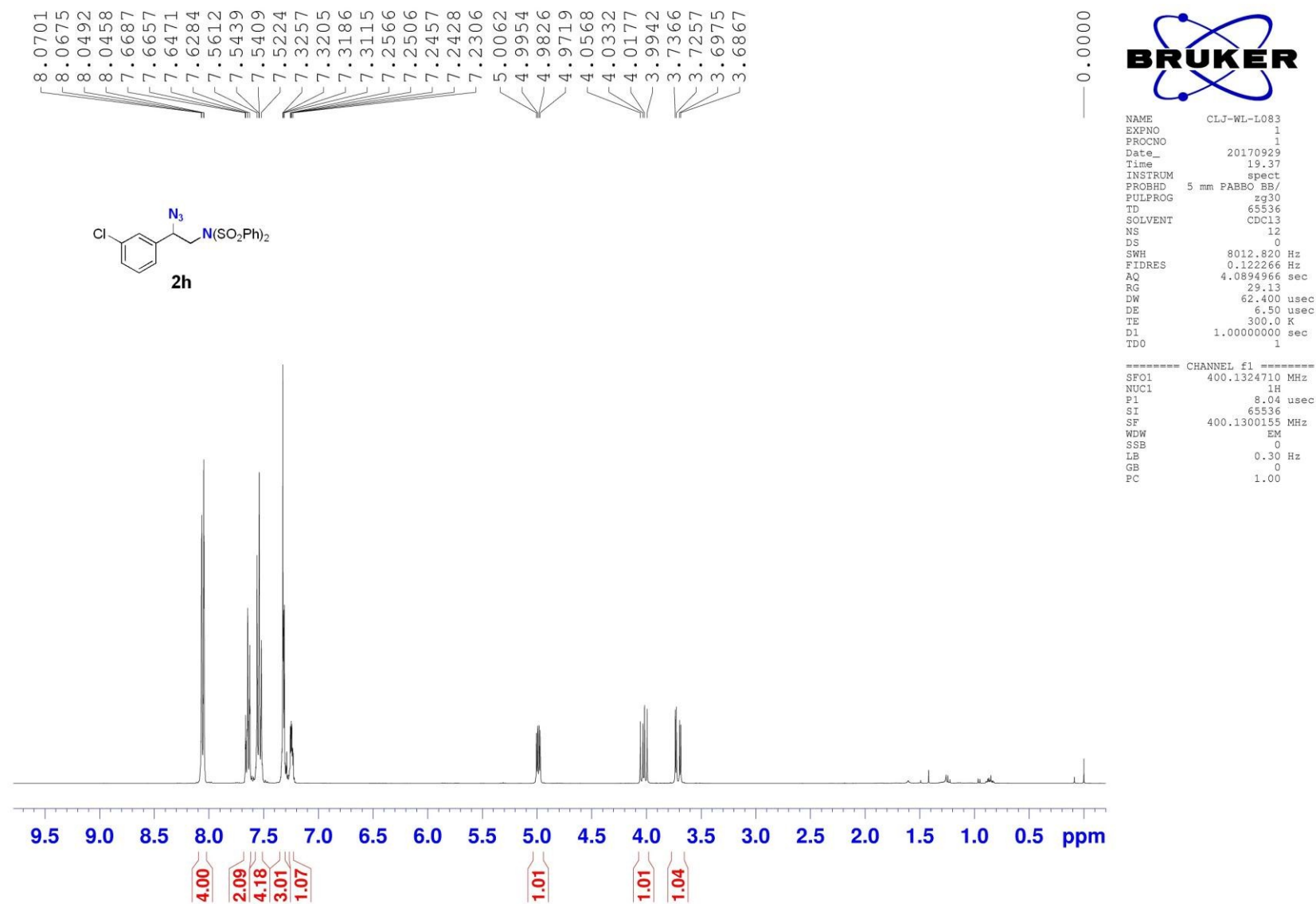
NAME CLJ-WL-L096
 EXPNO 2
 PROCNO 1
 Date_ 20171018
 Time 2.55
 INSTRUM spect
 PROBHD 5 mm PABBO BB/
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 512
 DS 0
 SWH 24038.461 Hz
 FIDRES 0.366798 Hz
 AQ 1.3631988 sec
 RG 194.26
 DW 20.800 usec
 DE 6.50 usec
 TE 300.0 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 1

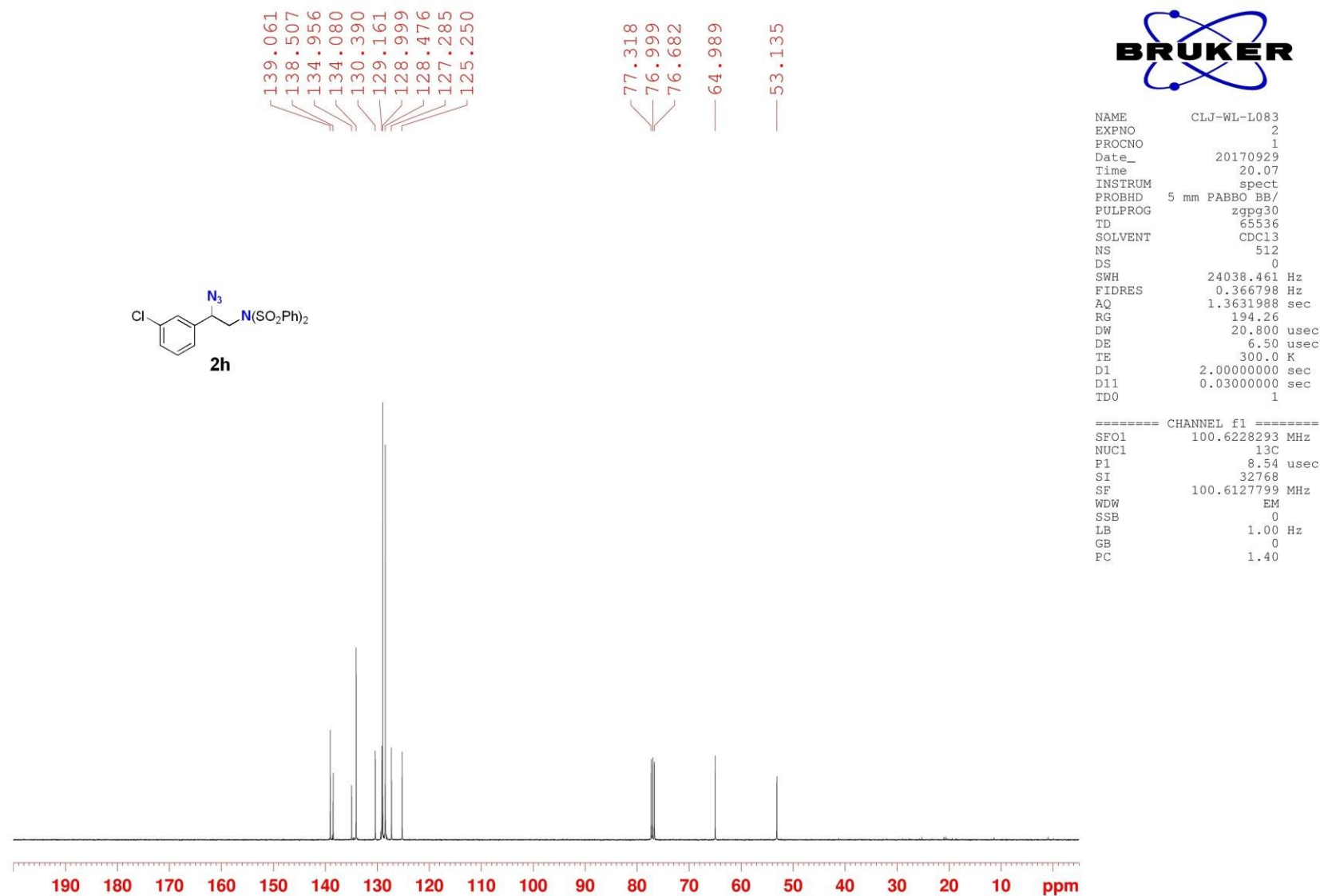


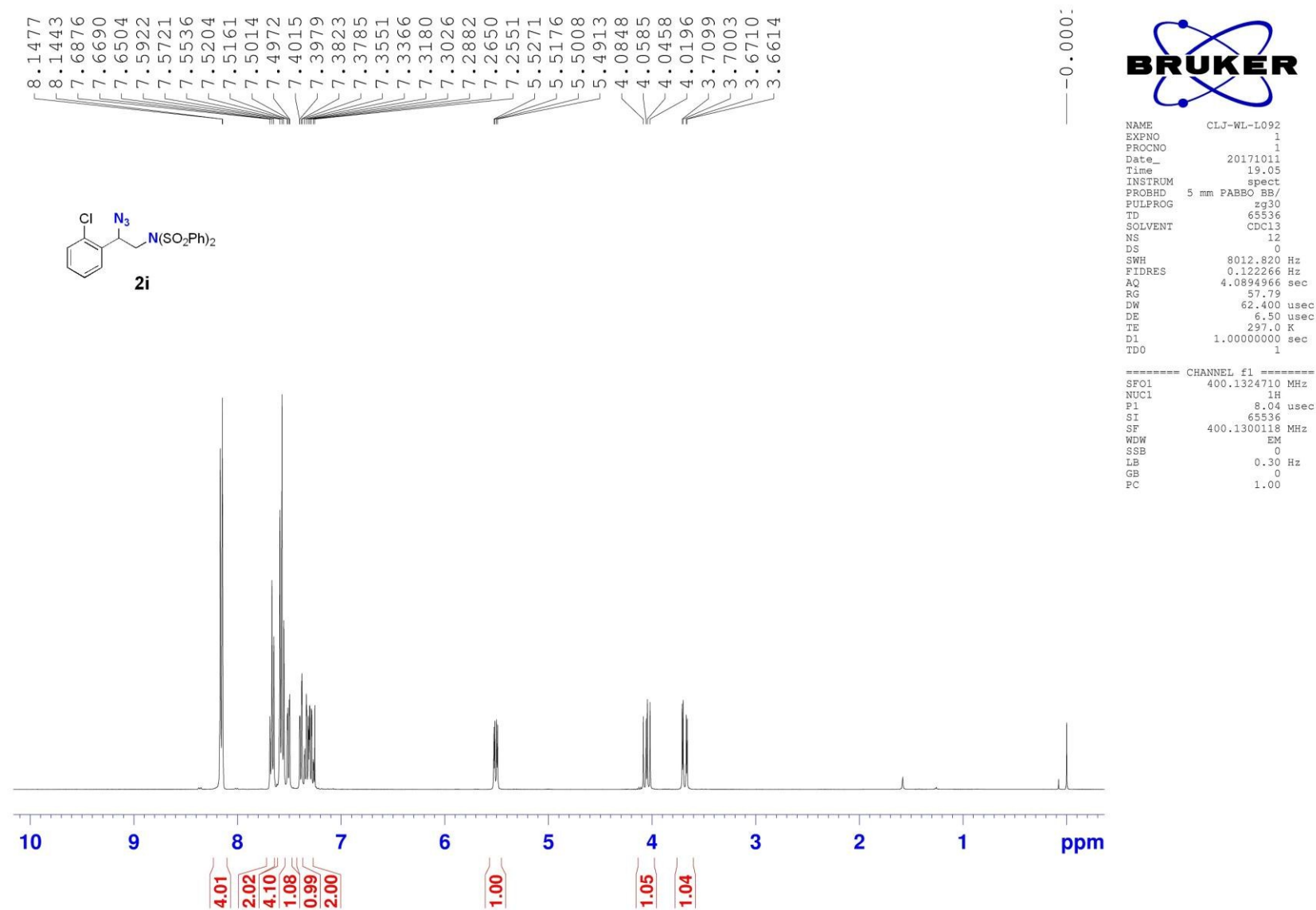


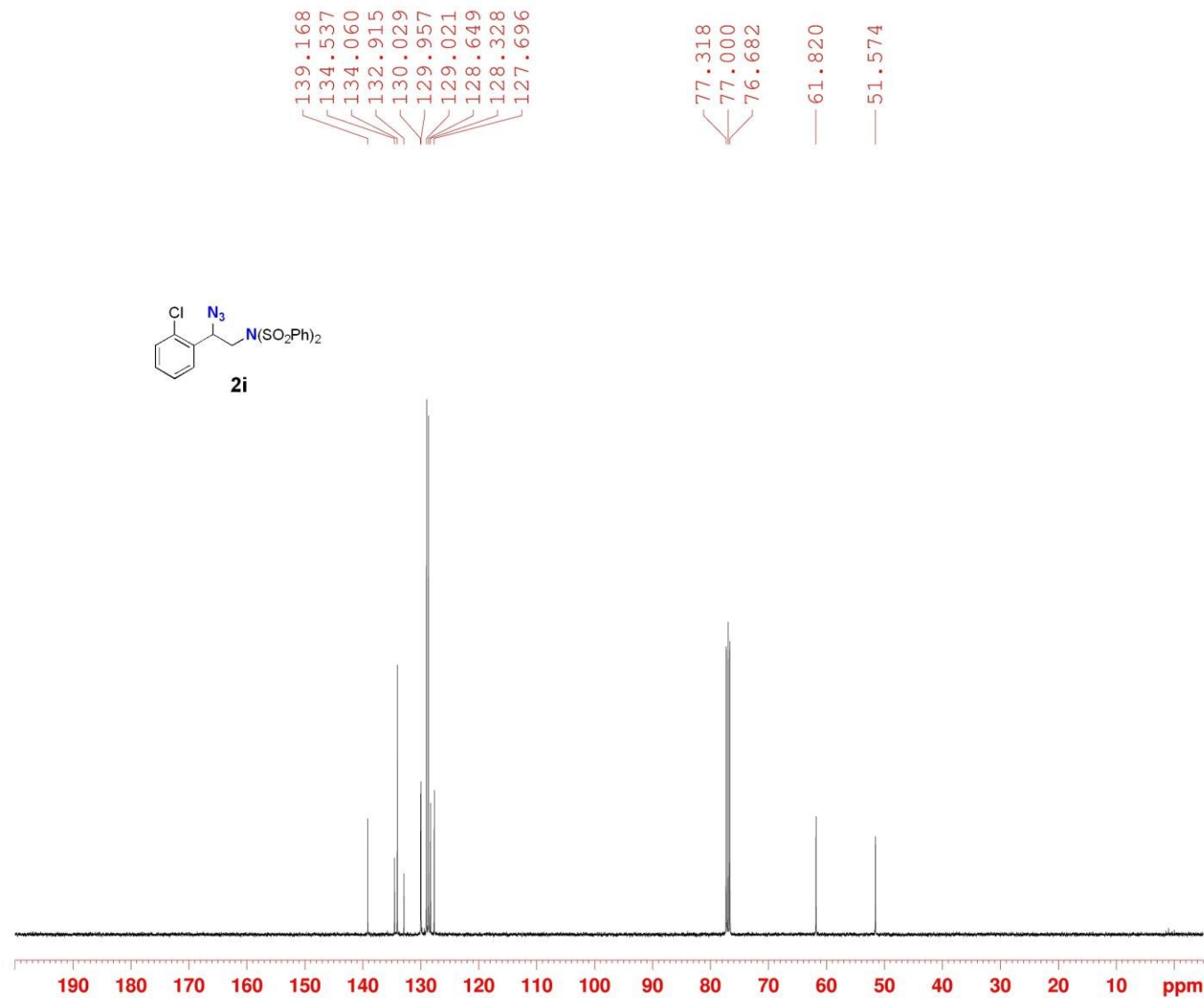
NAME CLJ-WL-L080
 EXPNO 2
 PROCNO 1
 Date_ 20170929
 Time 21.54
 INSTRUM spect
 PROBHD 5 mm PABBO BB/
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 512
 DS 0
 SWH 24038.461 Hz
 FIDRES 0.366798 Hz
 AQ 1.3631988 sec
 RG 194.26
 DW 20.800 usec
 DE 6.50 usec
 TE 300.0 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 1

===== CHANNEL f1 =====
 SFO1 100.6228293 MHz
 NUC1 13C
 P1 8.54 usec
 SI 32768
 SF 100.6127820 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40



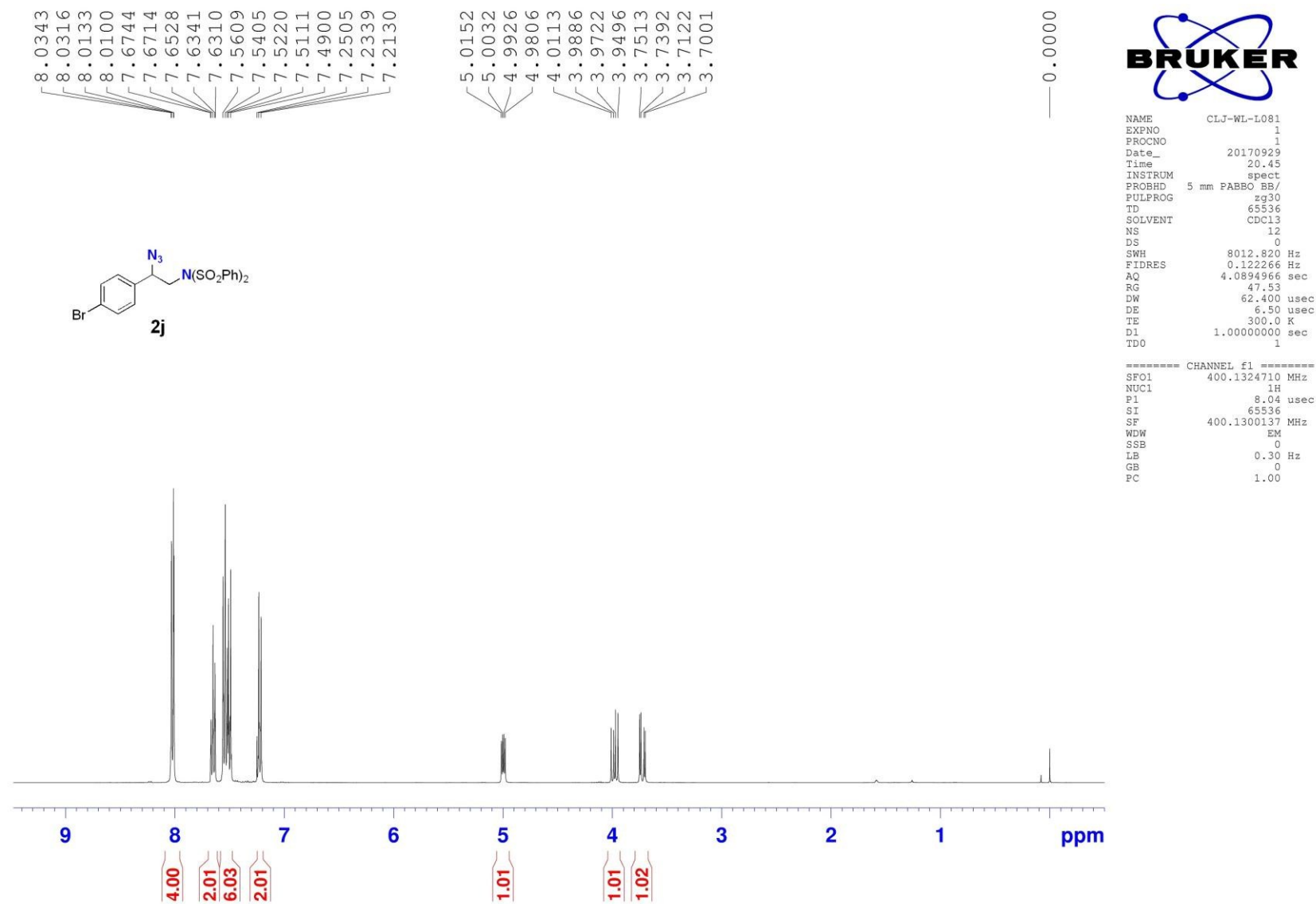


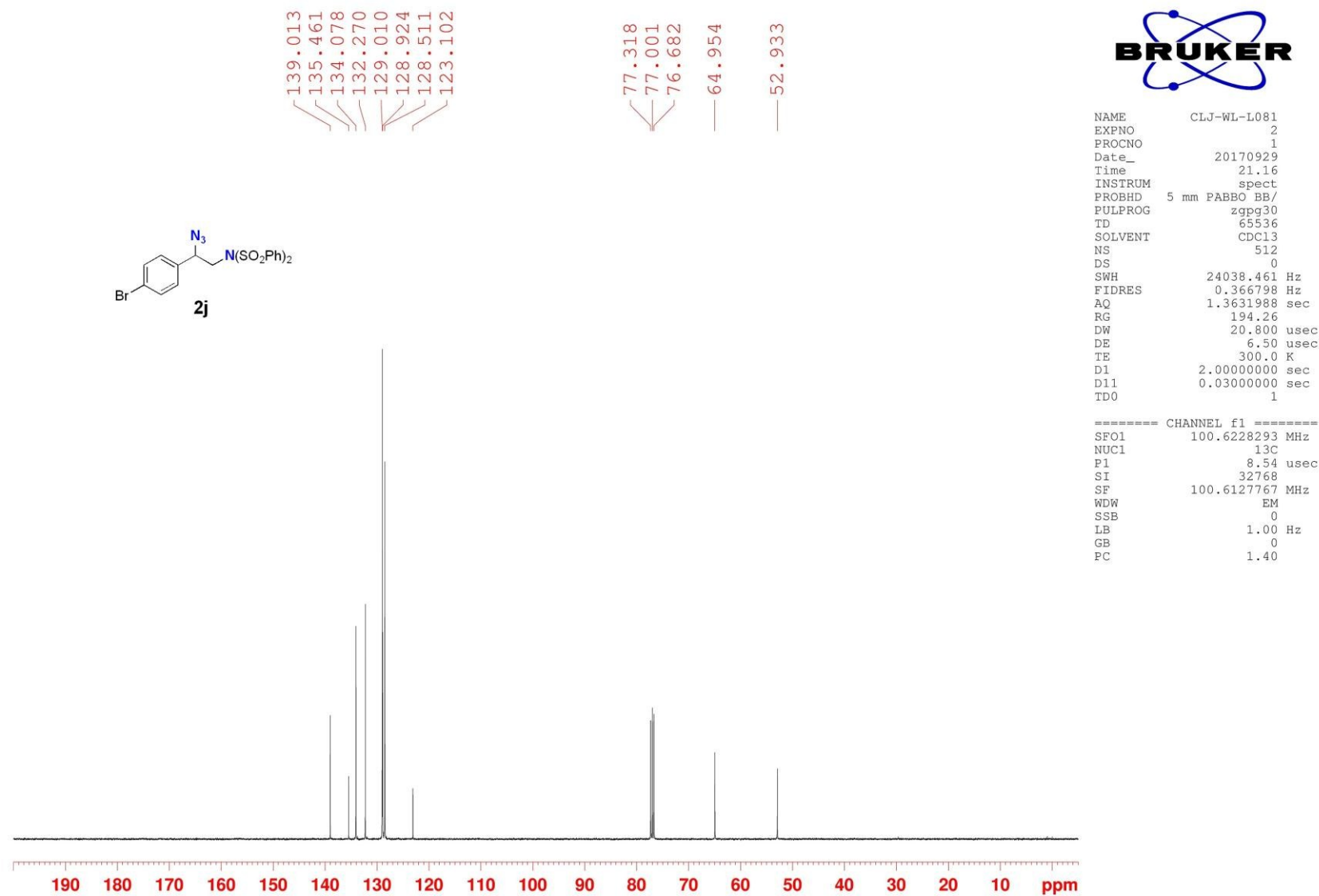


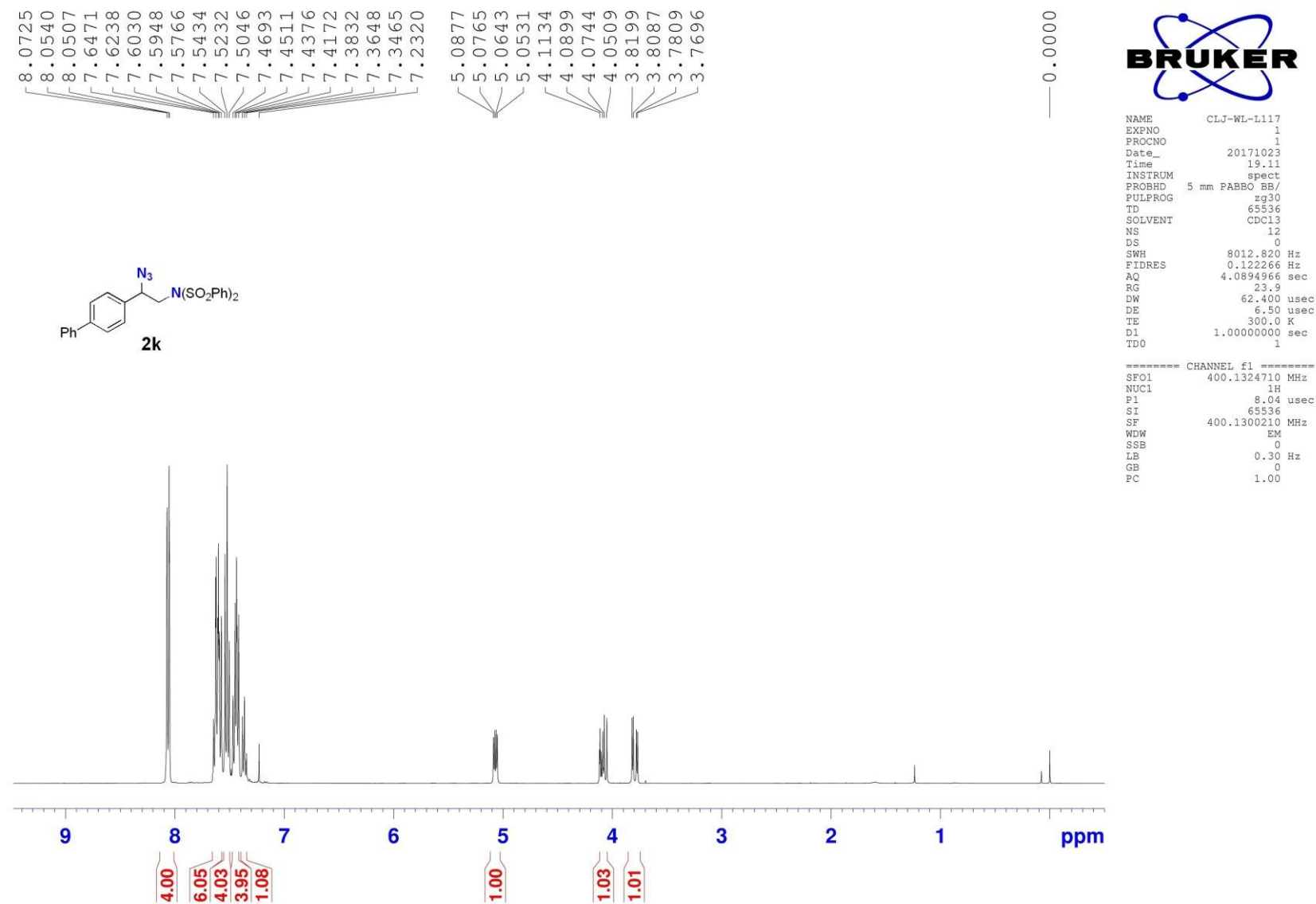


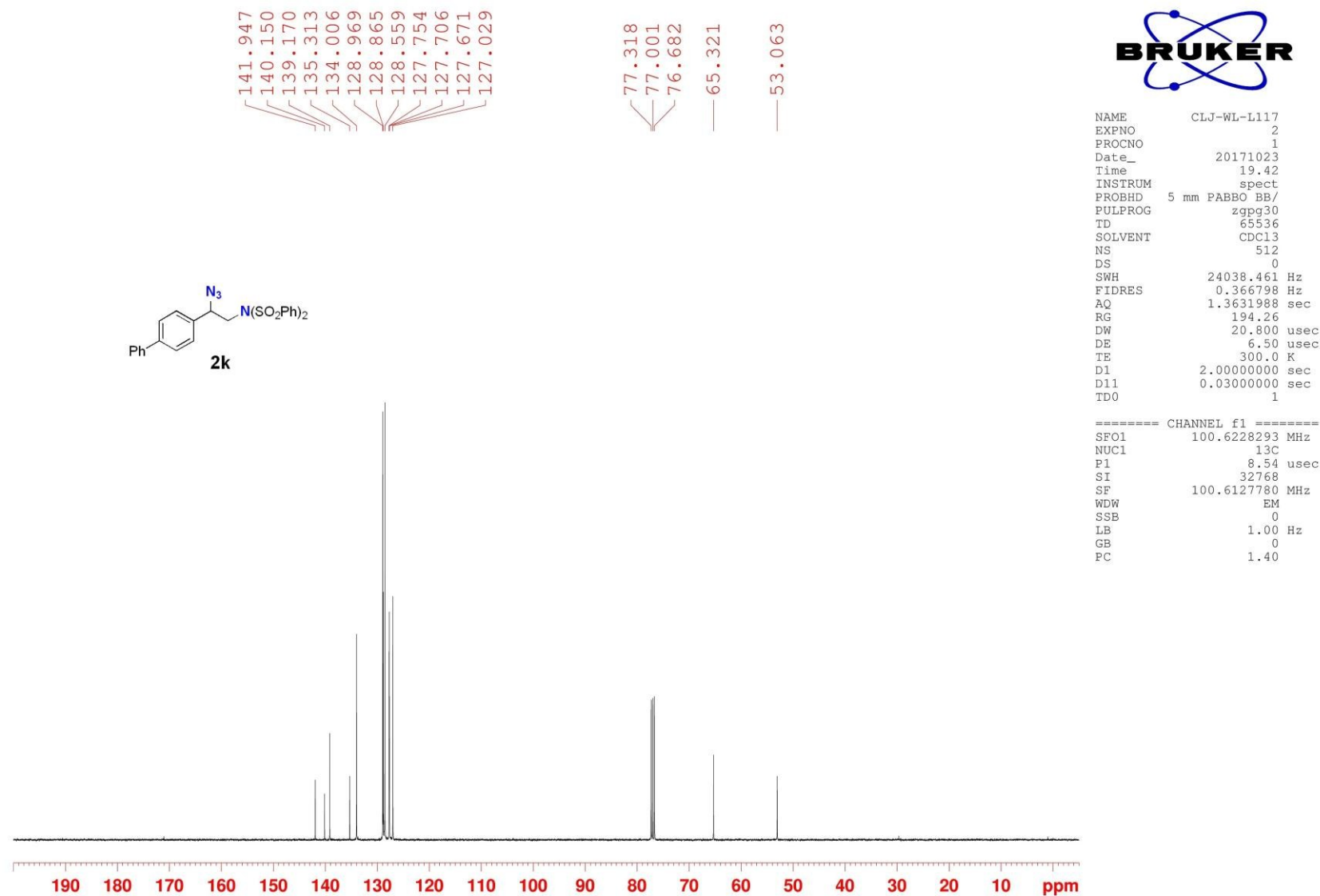
NAME CLJ-WL-L092
 EXPNO 2
 PROCNO 1
 Date_ 20171011
 Time 19.36
 INSTRUM spect
 PROBHD 5 mm PABBO BB/
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 512
 DS 0
 SWH 24038.461 Hz
 FIDRES 0.366798 Hz
 AQ 1.3631988 sec
 RG 194.26
 DW 20.800 usec
 DE 6.50 usec
 TE 297.3 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 1

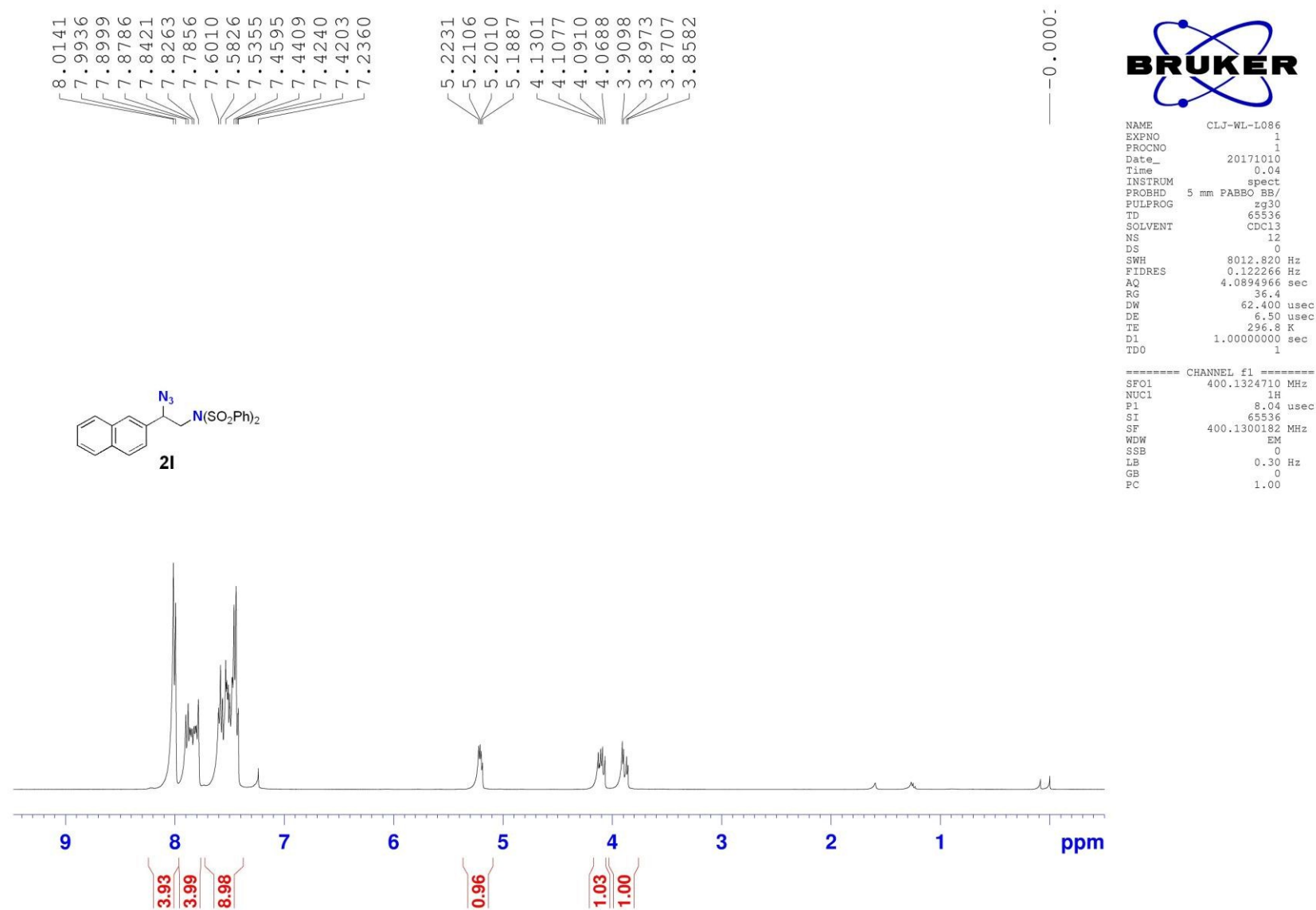
===== CHANNEL f1 =====
 SFO1 100.6228293 MHz
 NUC1 13C
 P1 8.54 usec
 SI 32768
 SF 100.6127746 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40

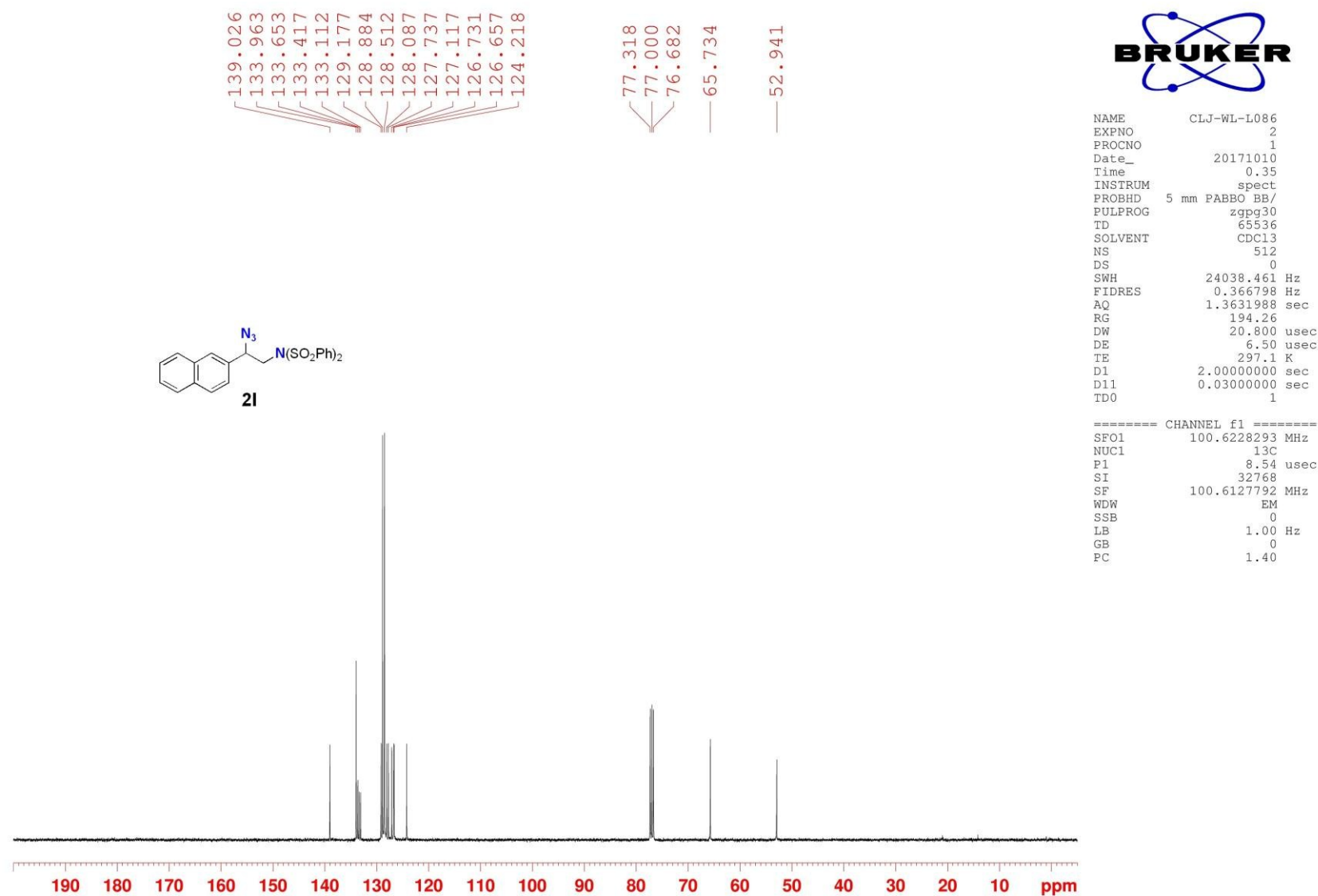


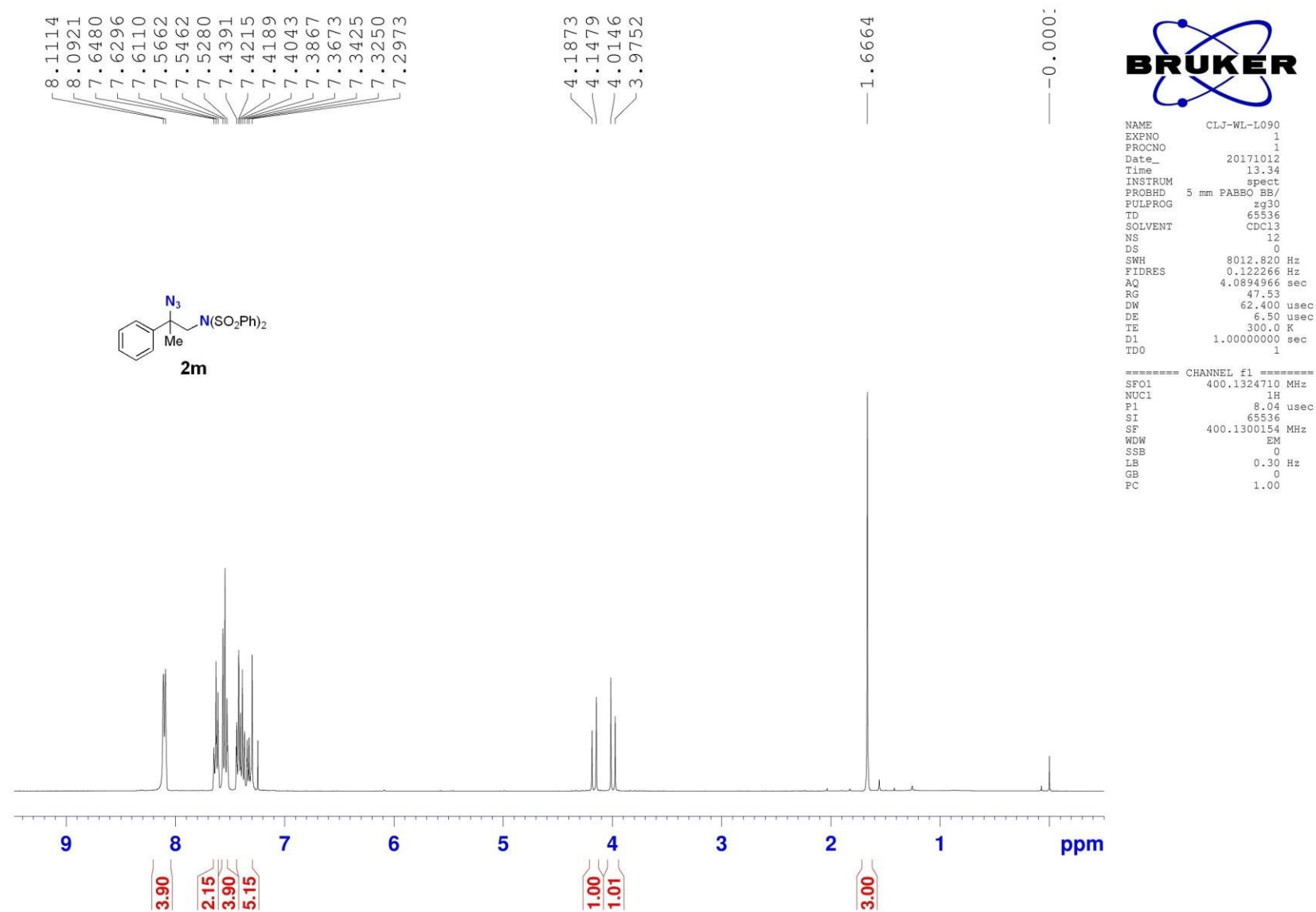


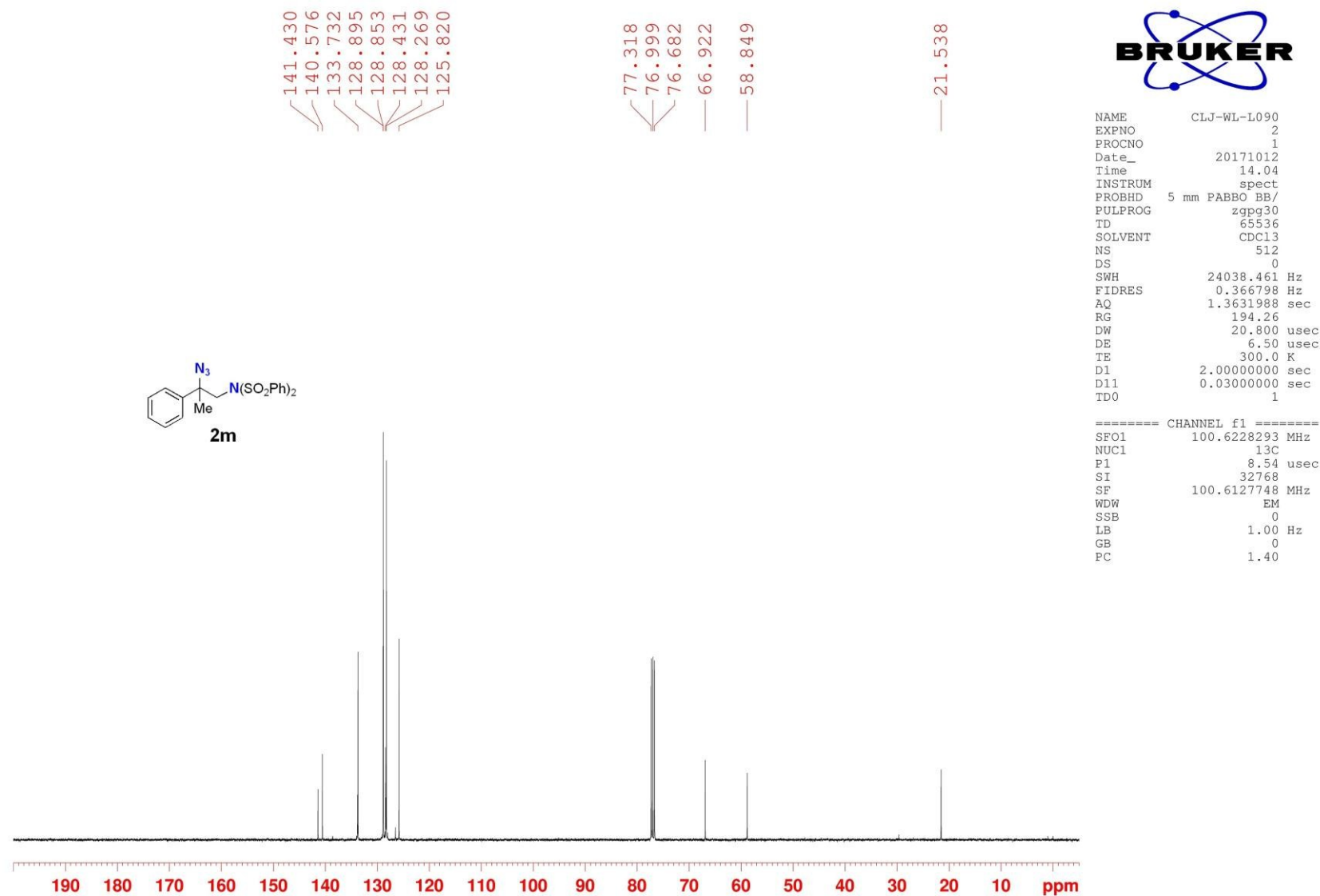


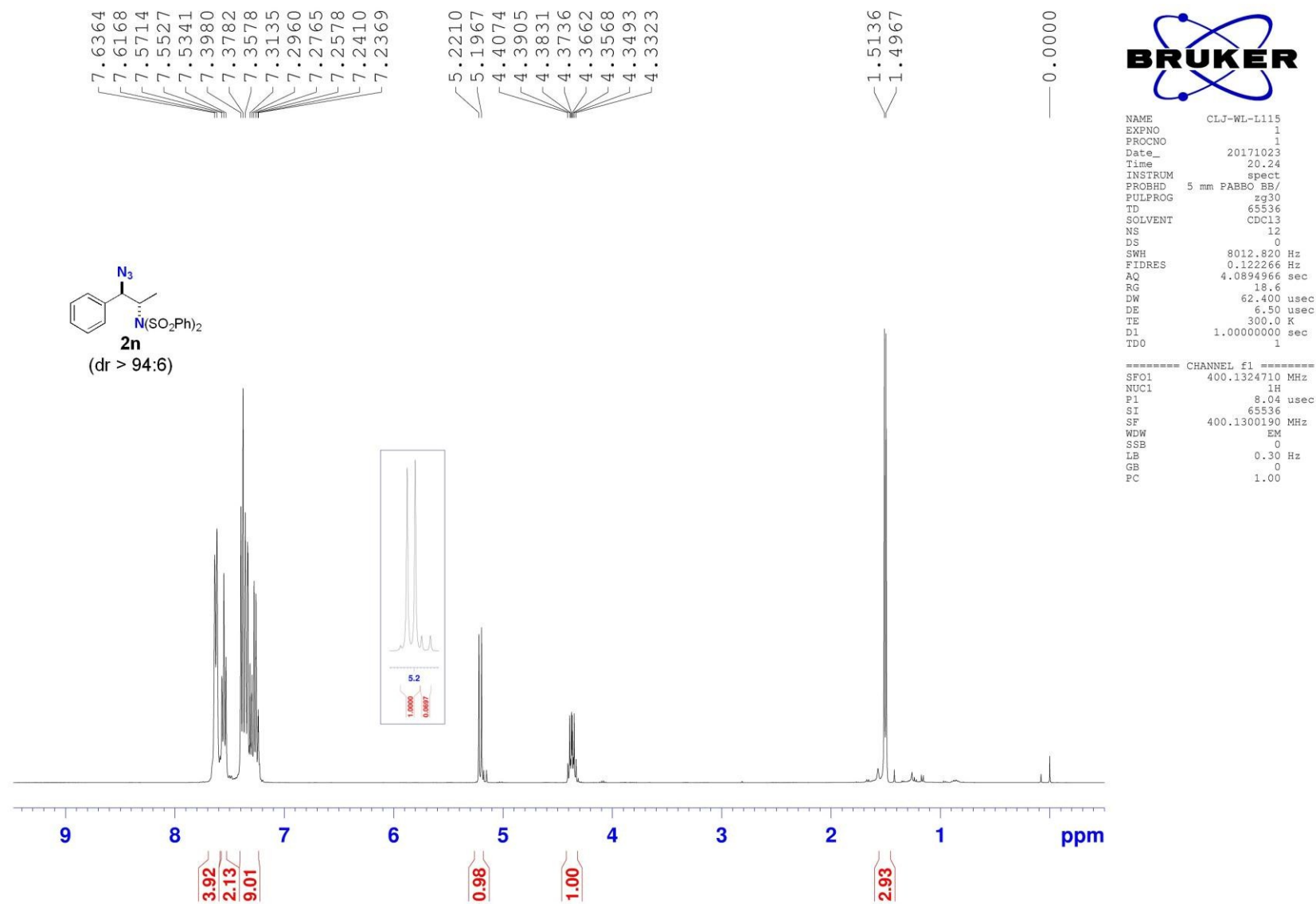


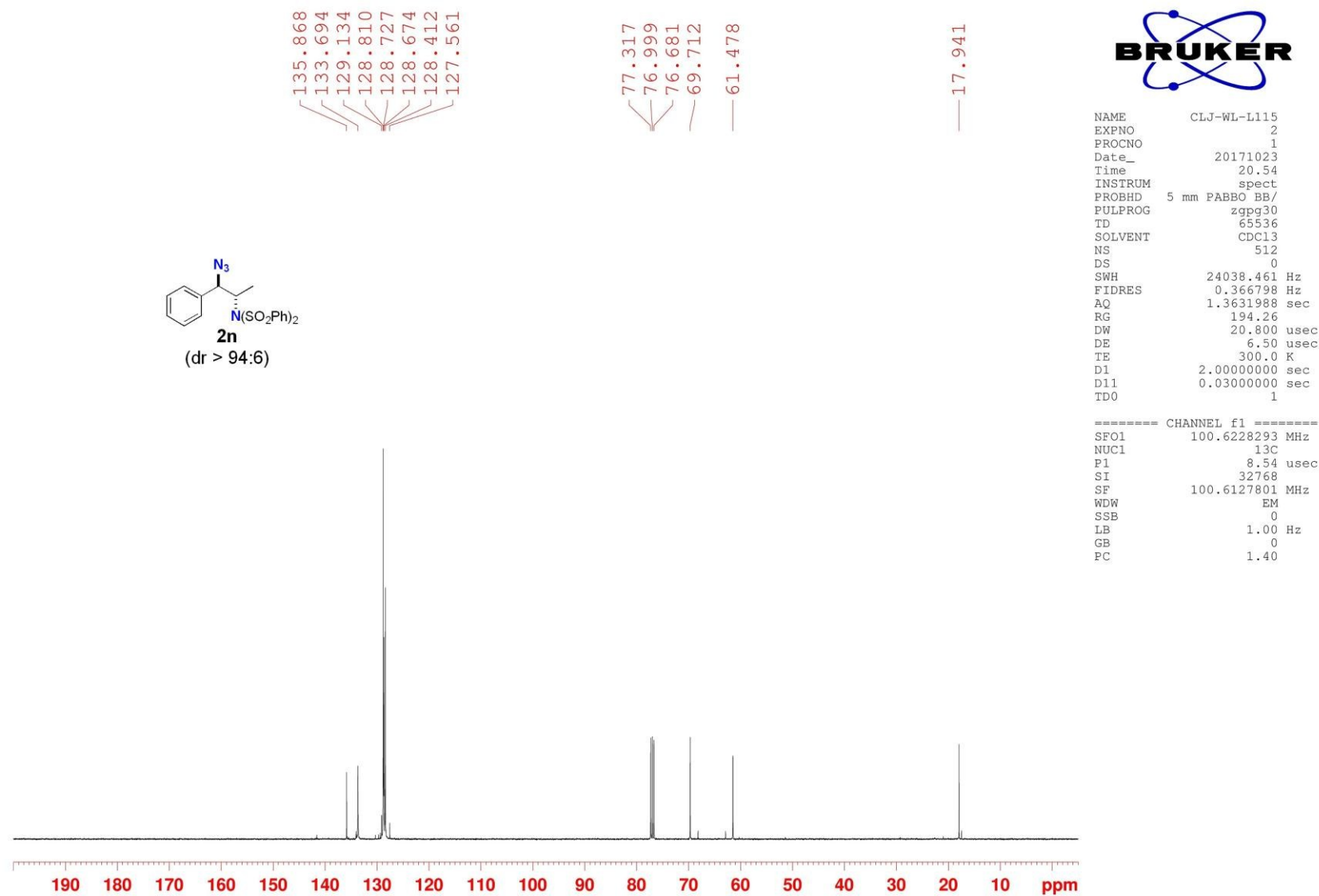


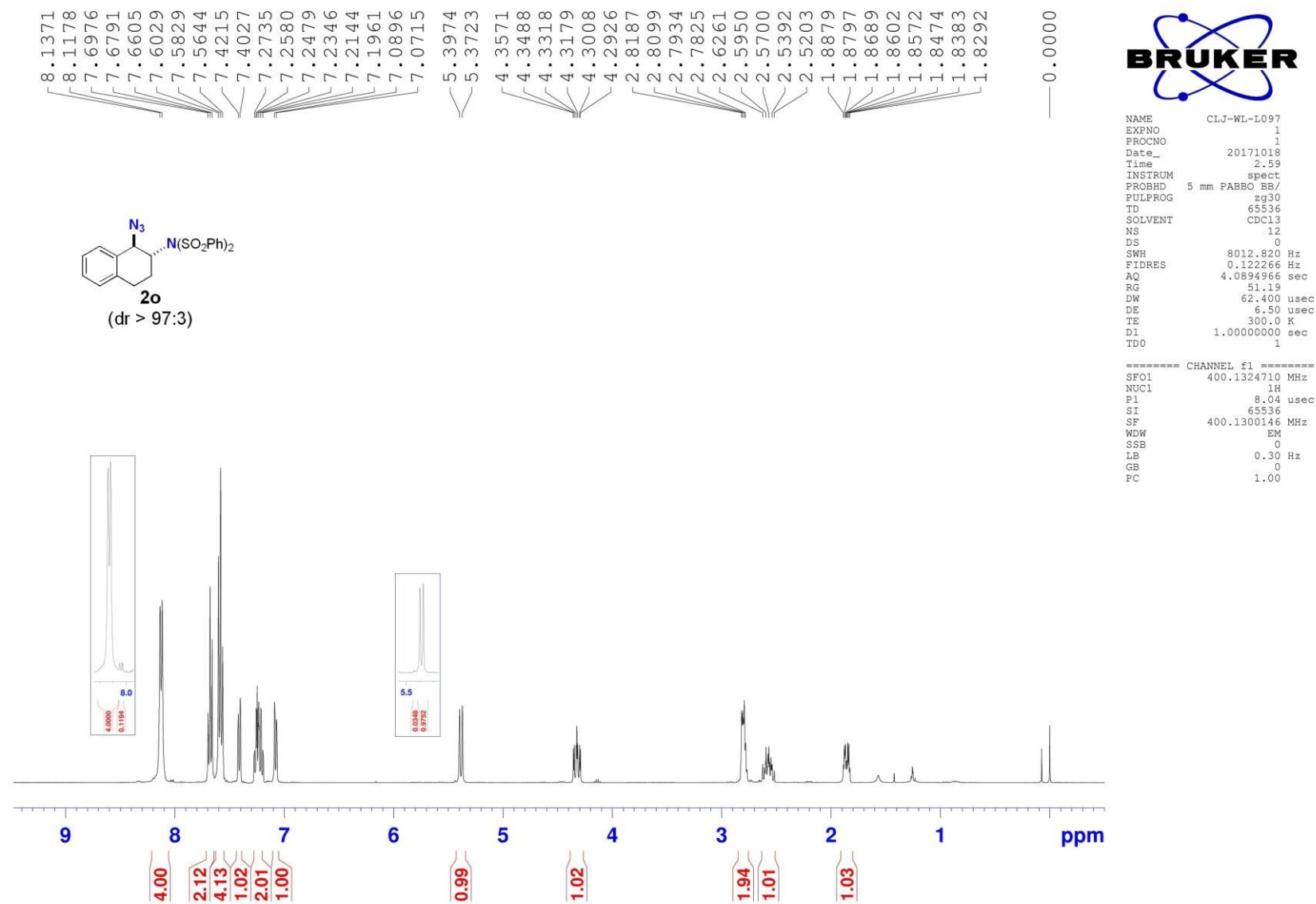


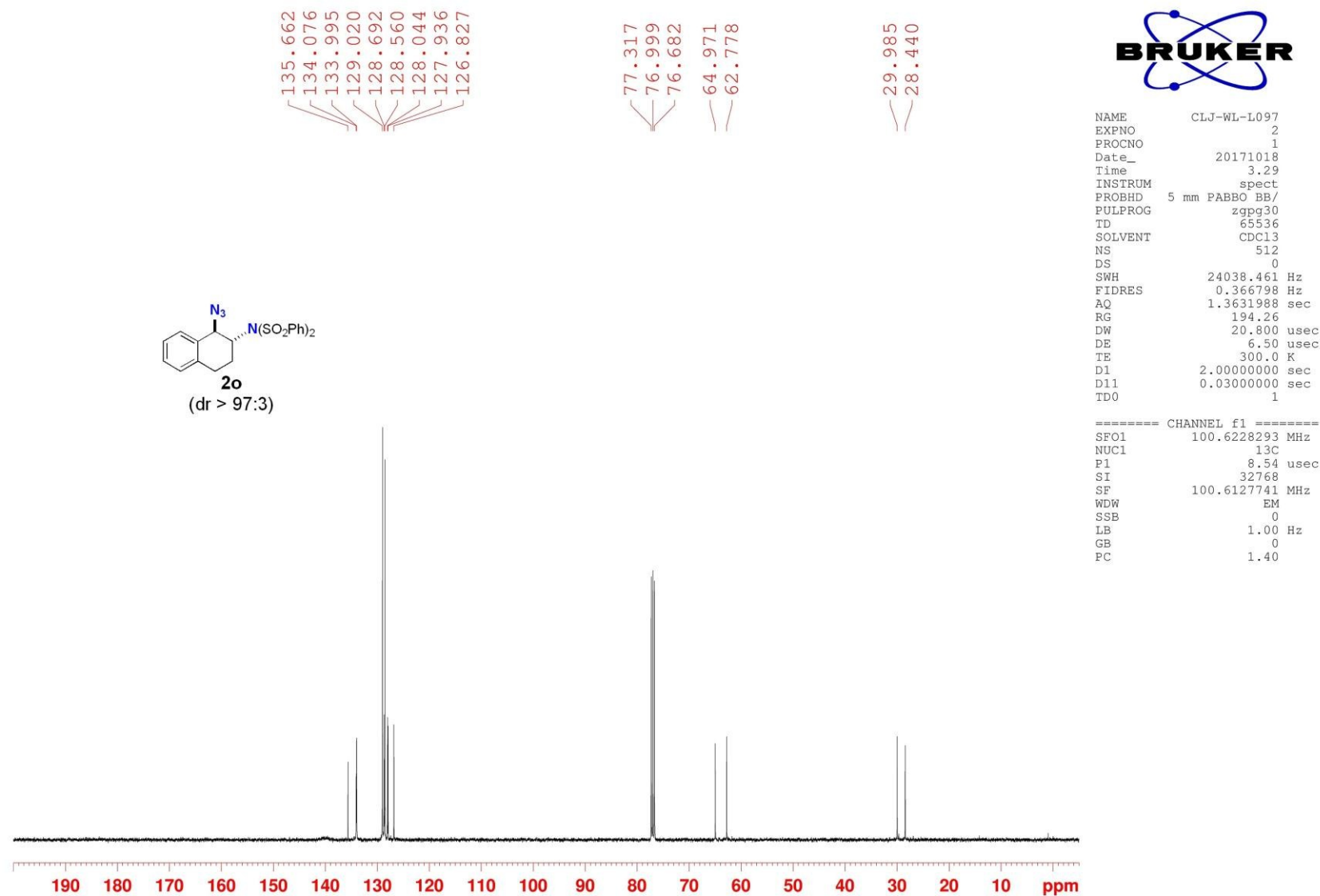


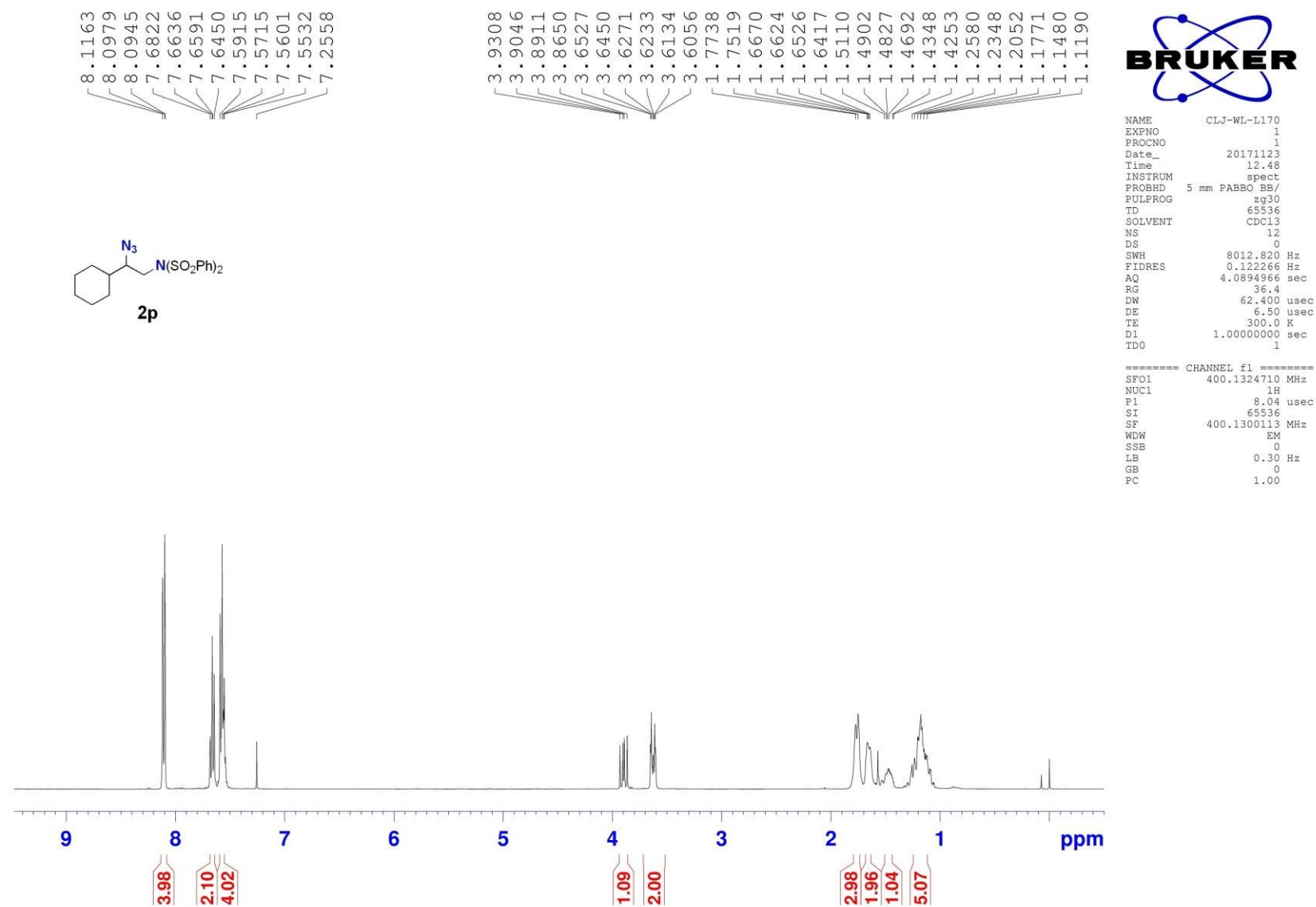


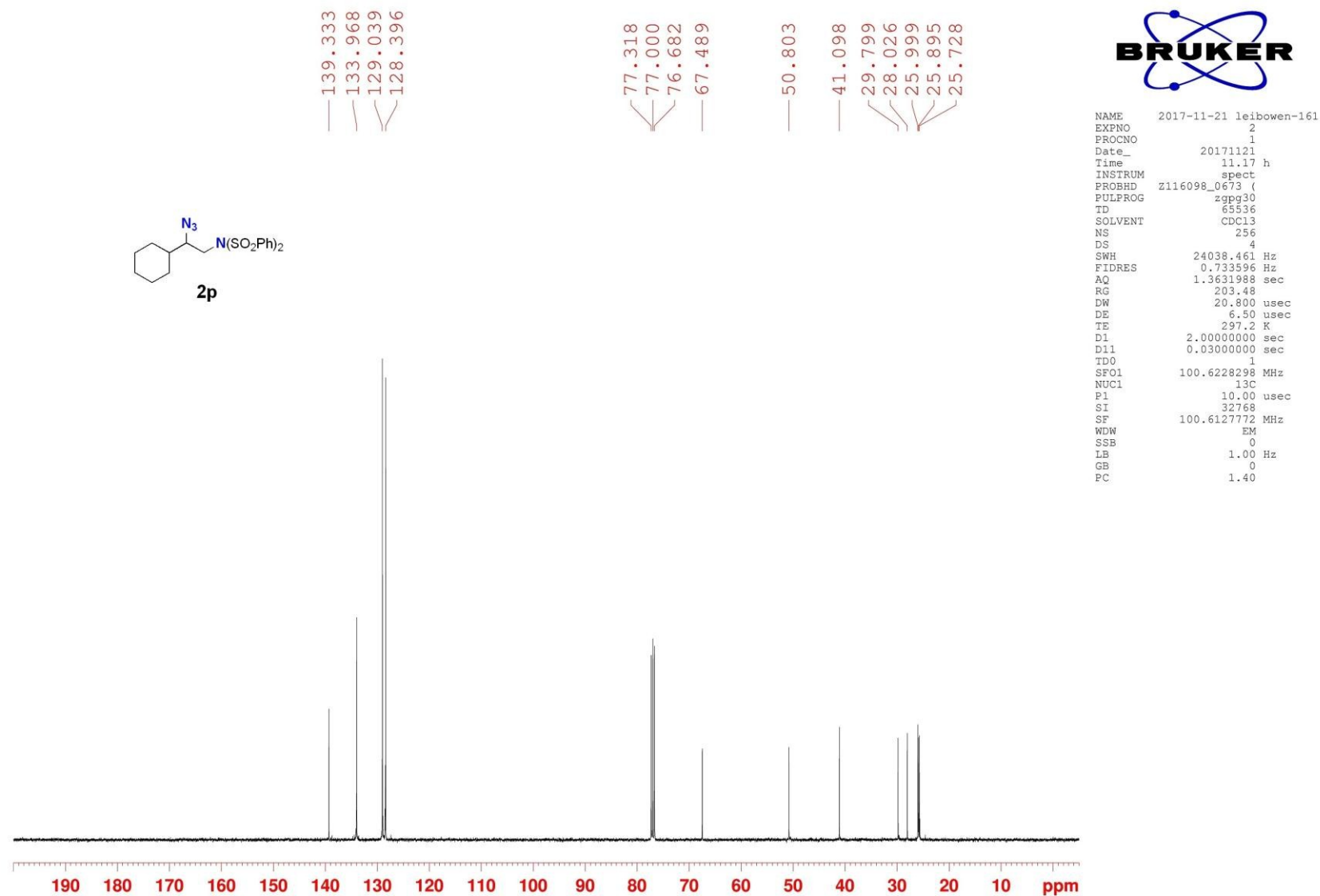


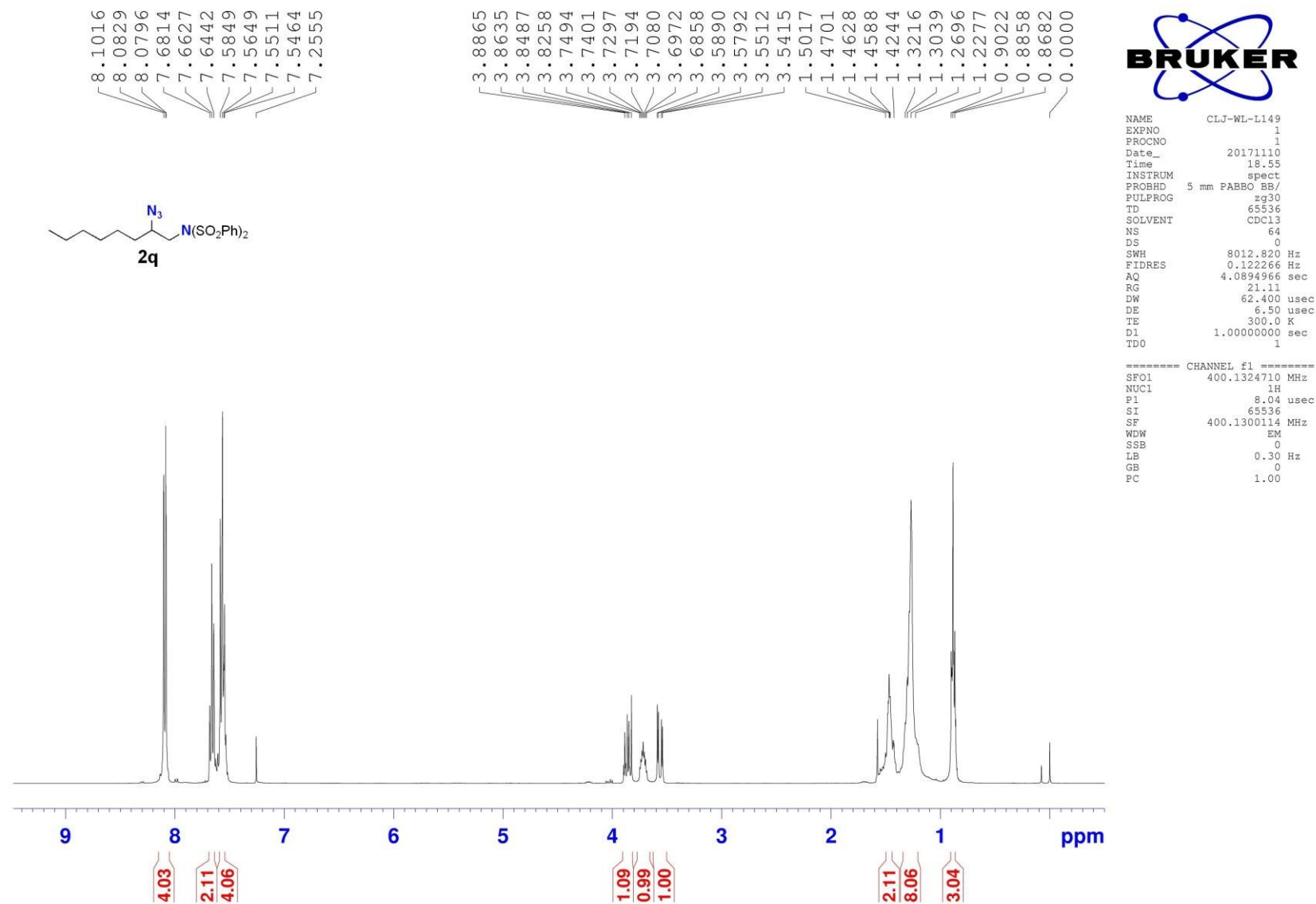


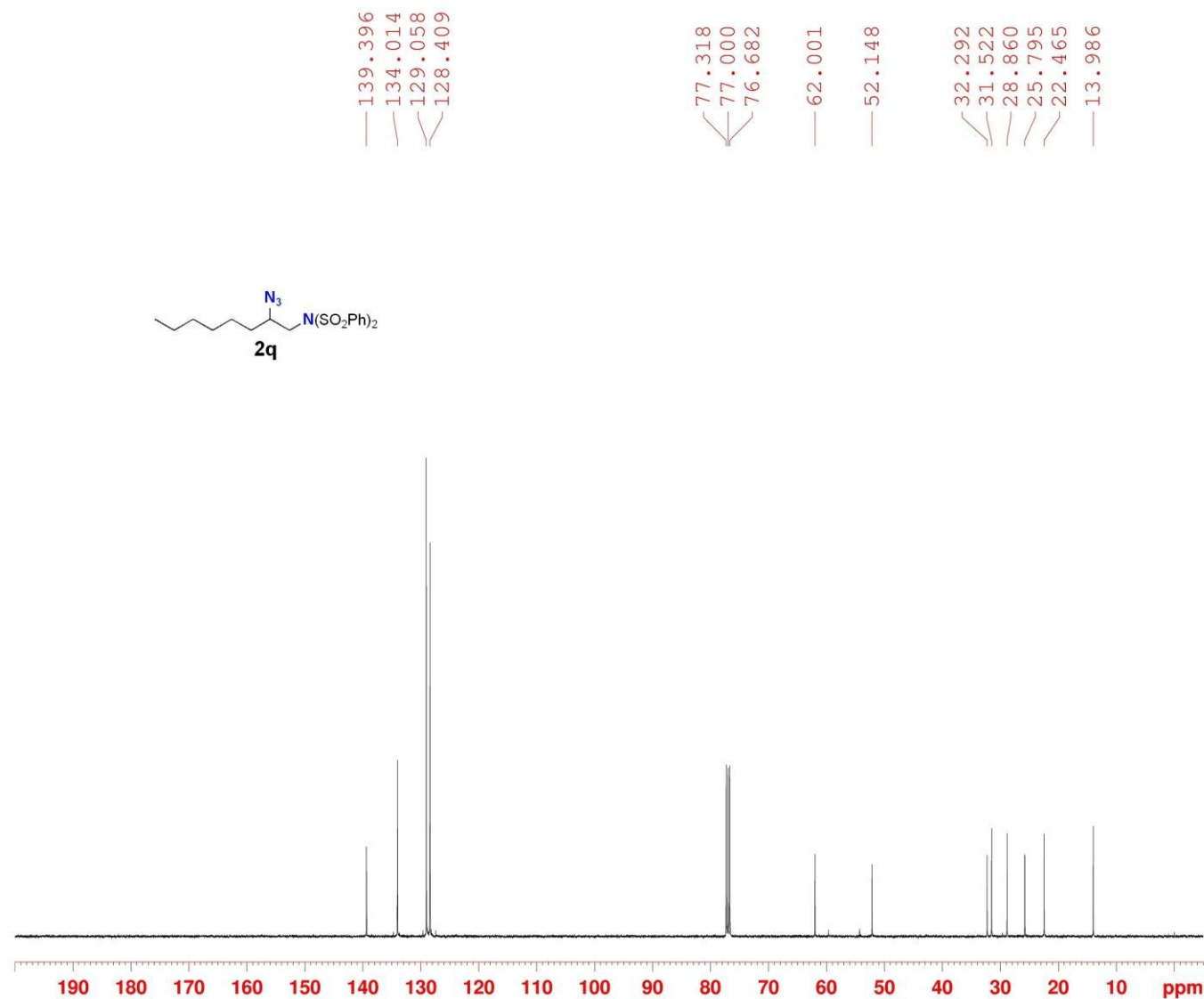






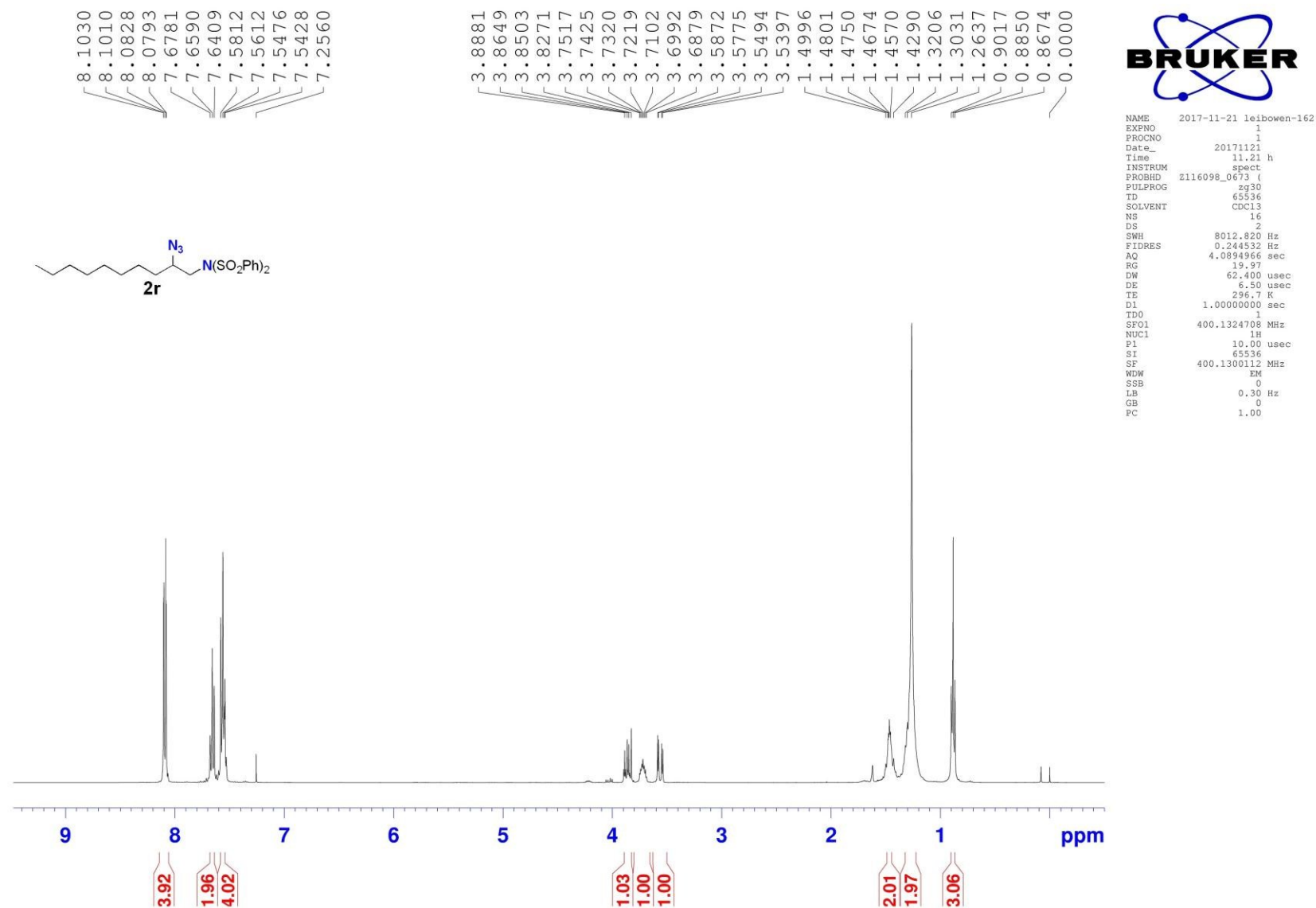




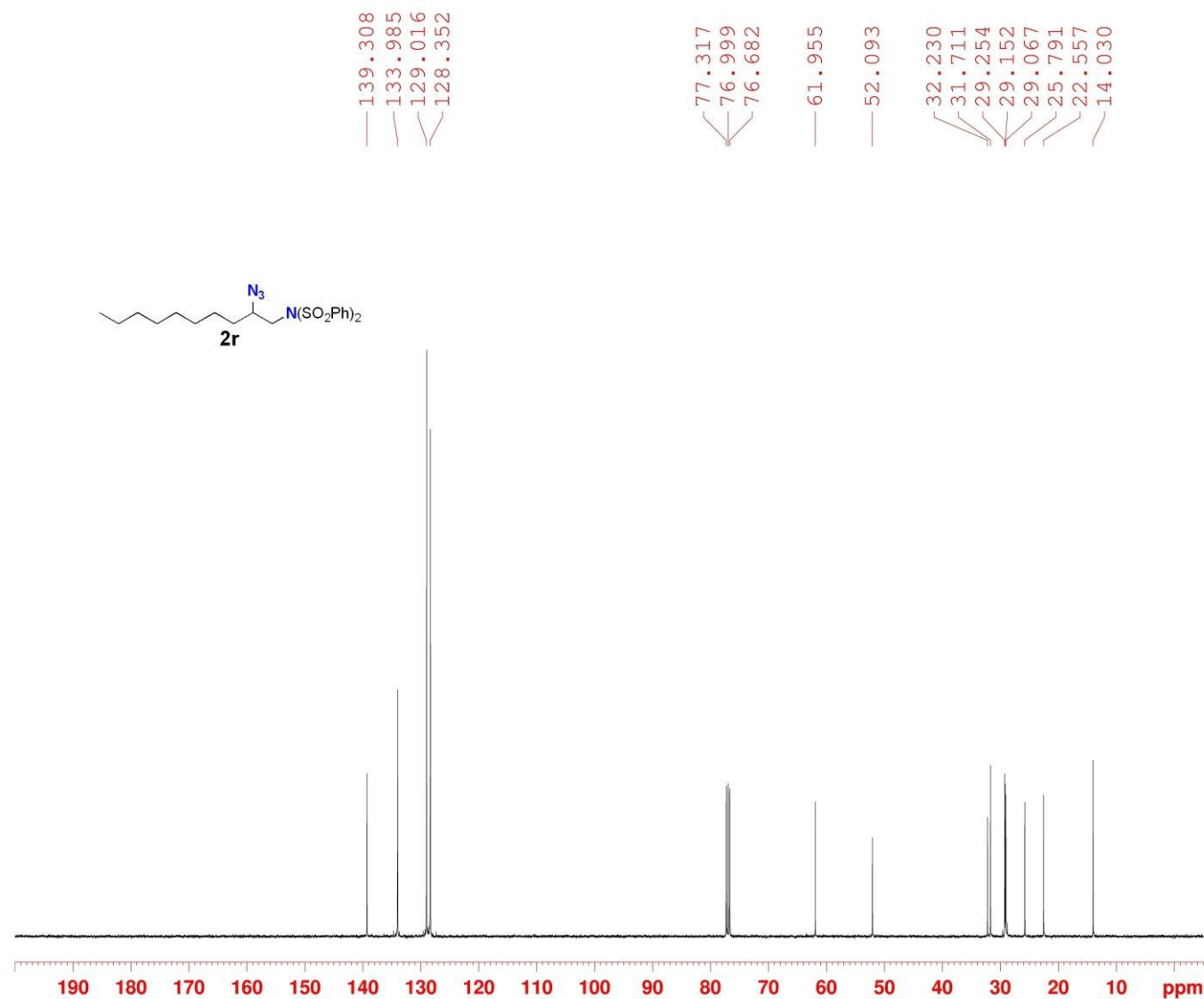


NAME CLJ-WL-L134
EXPNO 2
PROCNO 1
Date_ 20171106
Time 3.20
INSTRUM spect
PROBHD 5 mm PABBO BB/
PULPROG zgpgg30
TD 65536
SOLVENT CDCl3
NS 512
DS 0
SWH 24038.461 Hz
FIDRES 0.366798 Hz
AQ 1.3631988 sec
RG 194.26
DW 20.800 usec
DE 6.50 usec
TE 299.1 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1

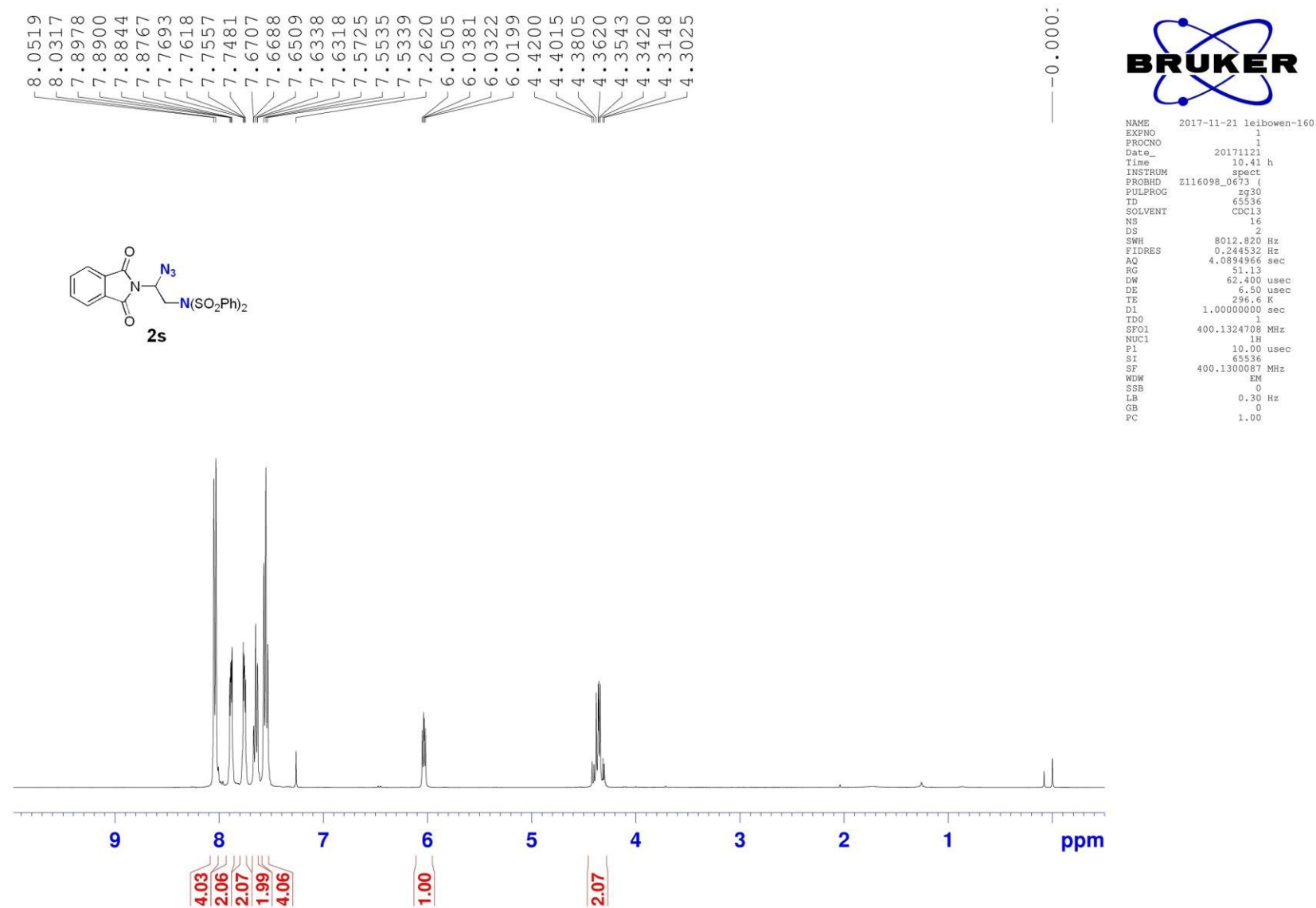
===== CHANNEL f1 =====
SFO1 100.6228293 MHz
NUC1 13C
P1 8.54 usec
SI 32768
SF 100.6127742 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40

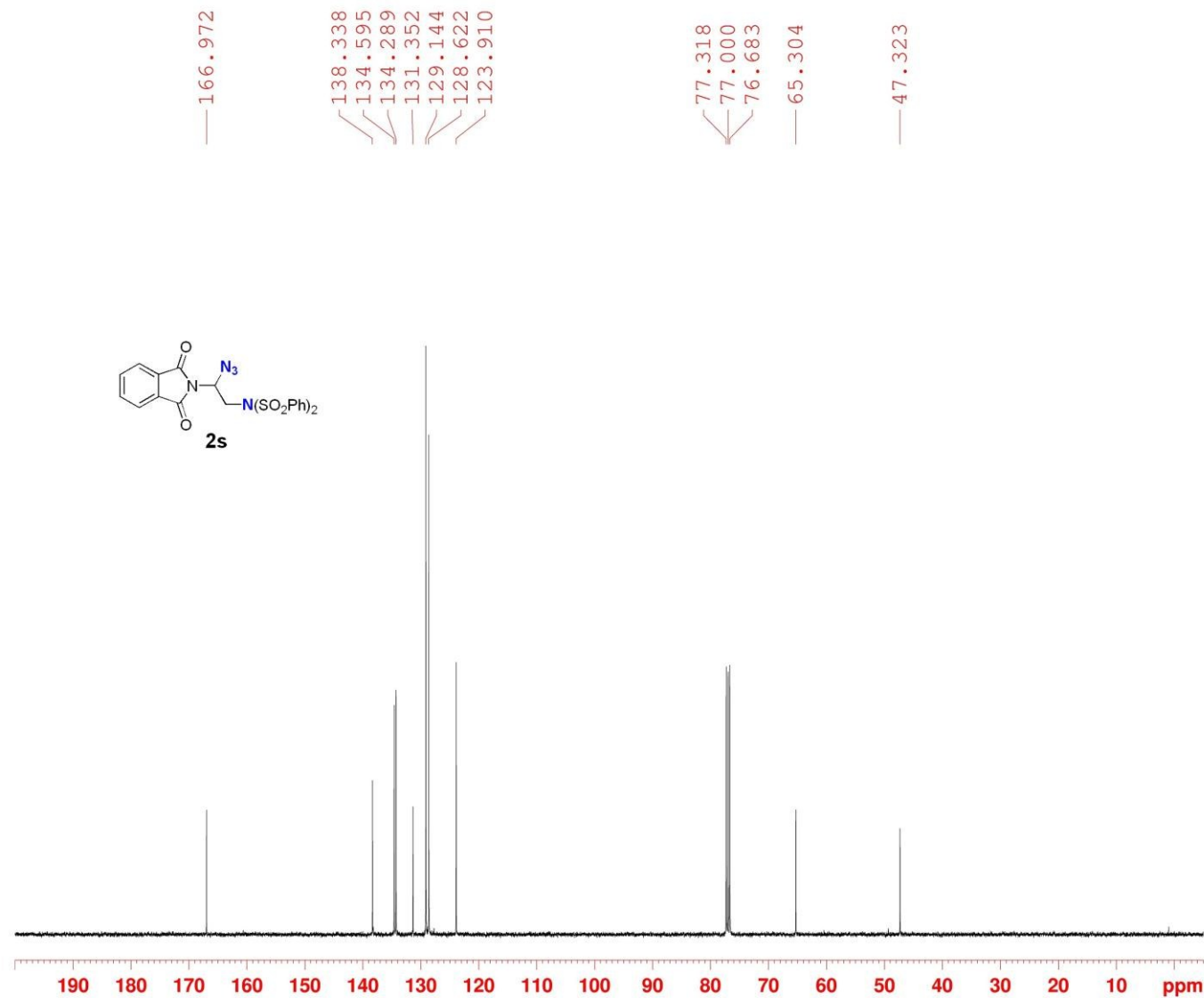


S51



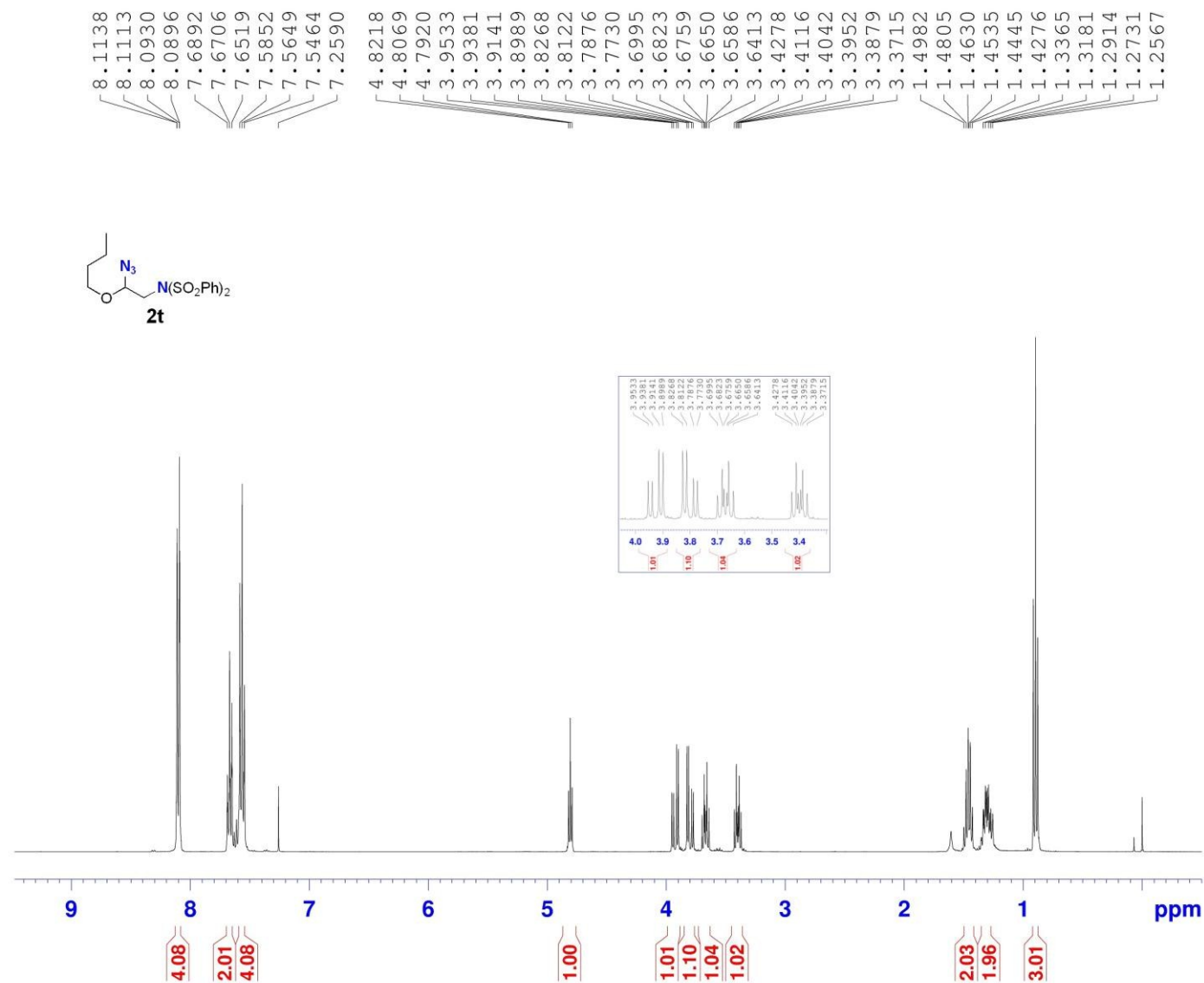
NAME 2017-11-21 leibowen-162
 EXPNO 2
 PROCNO 1
 Date_ 20171121
 Time 11.37 h
 INSTRUM spect
 PROBHD z116098_0673 (zpgg30)
 PULPROG zgpg30
 TD 65536
 SOLVENT CDC13
 NS 256
 DS 4
 SWH 24038.461 Hz
 FIDRES 0.733596 Hz
 AQ 1.3631988 sec
 RG 203.48
 DW 20.800 usec
 DE 6.50 usec
 TE 297.2 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 1
 SFO1 100.6228298 MHz
 NUC1 13C
 P1 10.00 usec
 SI 32768
 SF 100.6127798 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40



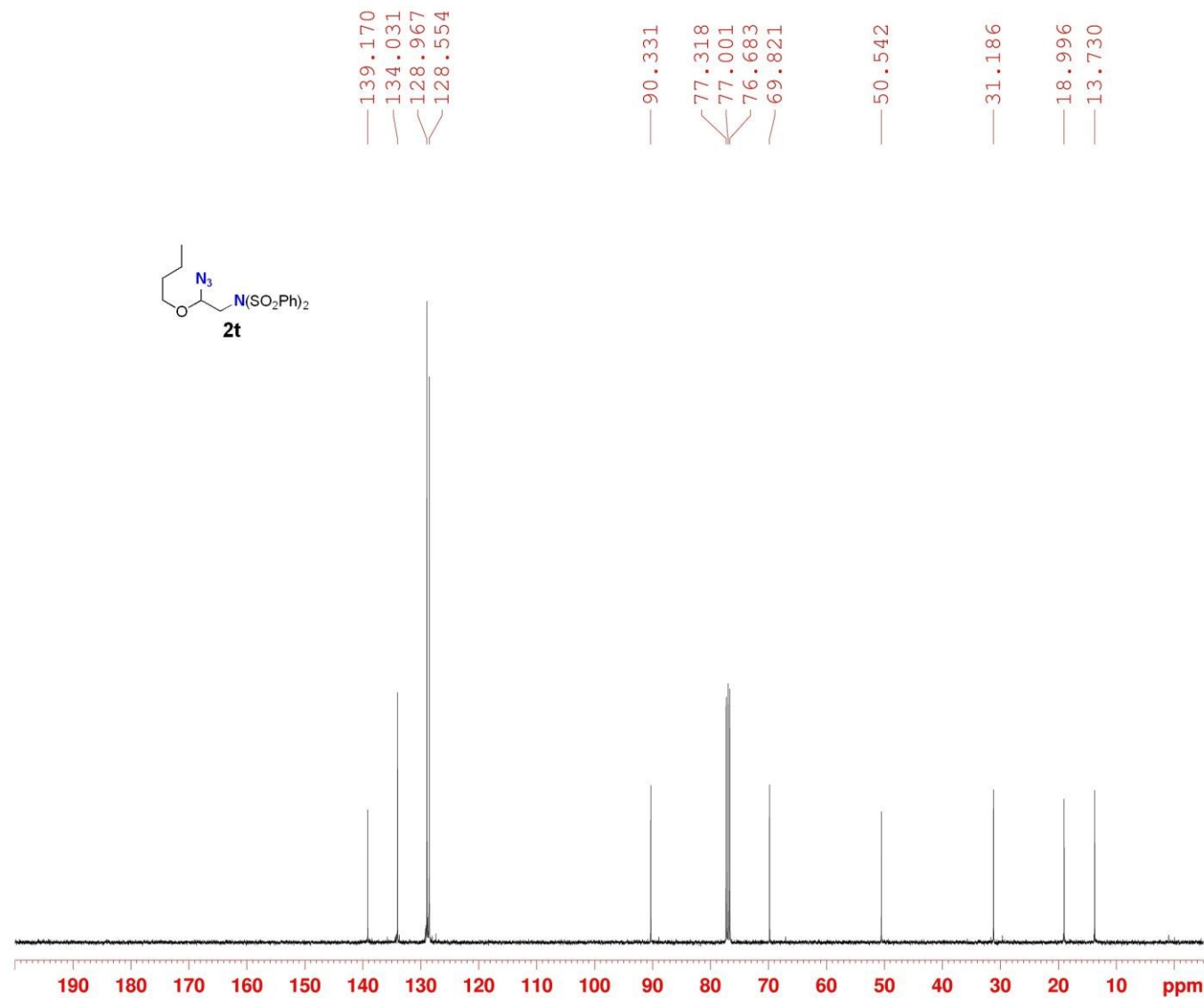


```

NAME      2017-11-21 leibowen-160
EXPNO     2
PROCNO    1
Date_     20171121
Time      10.57 h
INSTRUM    spect
PROBHD     Z116098_0673 (
PULPROG    zgpg30
TD         65536
SOLVENT    CDC13
NS         256
DS         4
SWH        24038.461 Hz
FIDRES     0.733596 Hz
AQ         1.3631988 sec
RG         203.48
DW         20.800 usec
DE         6.50 usec
TE         297.1 K
D1         2.00000000 sec
D11        0.03000000 sec
TD0        1
SF01       100.6228298 MHz
NUC1       13C
P1         10.00 usec
SI         32768
SF         100.6127776 MHz
WDW        EM
SSB        0
LB         1.00 Hz
GB         0
PC         1.40
    
```

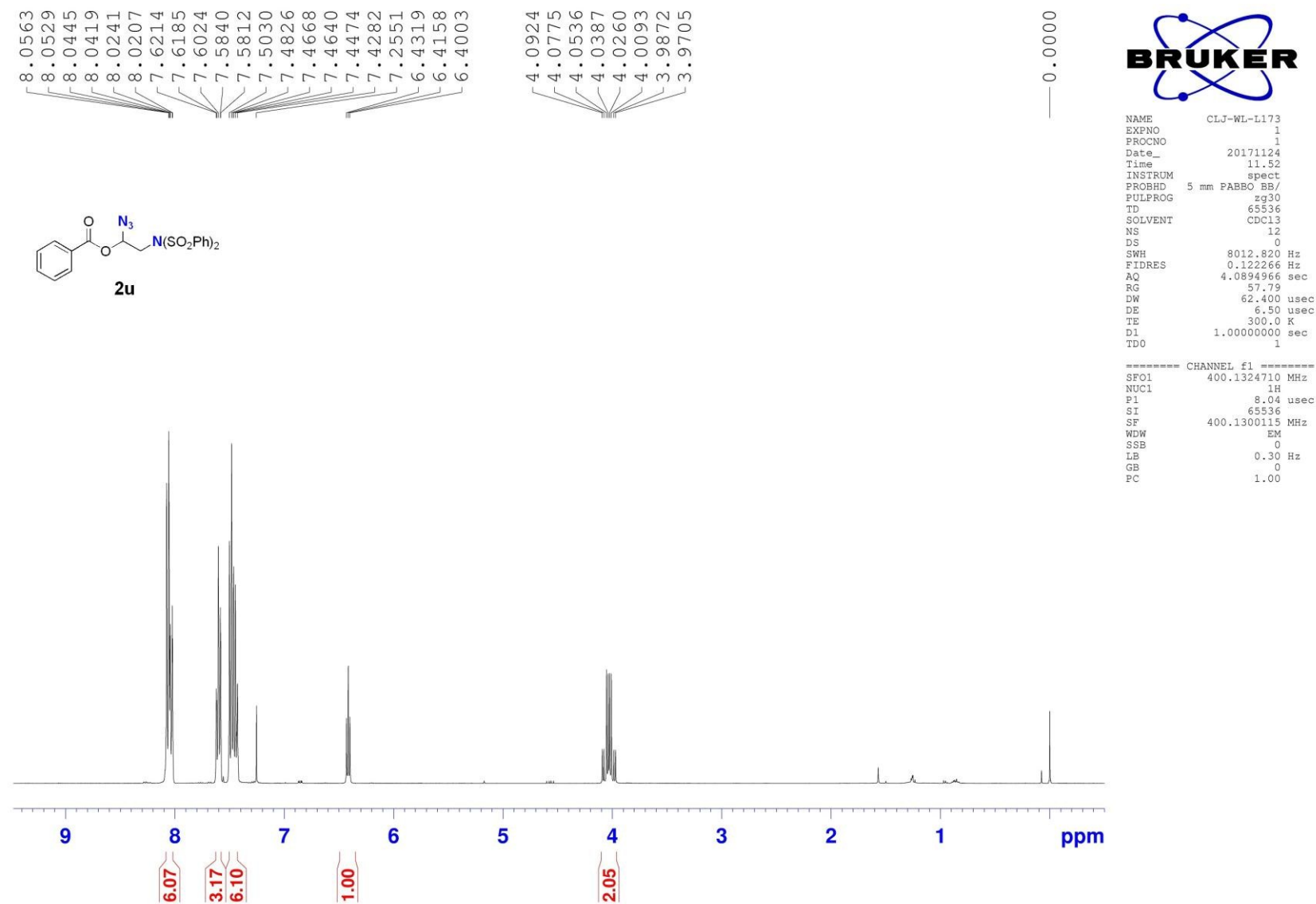


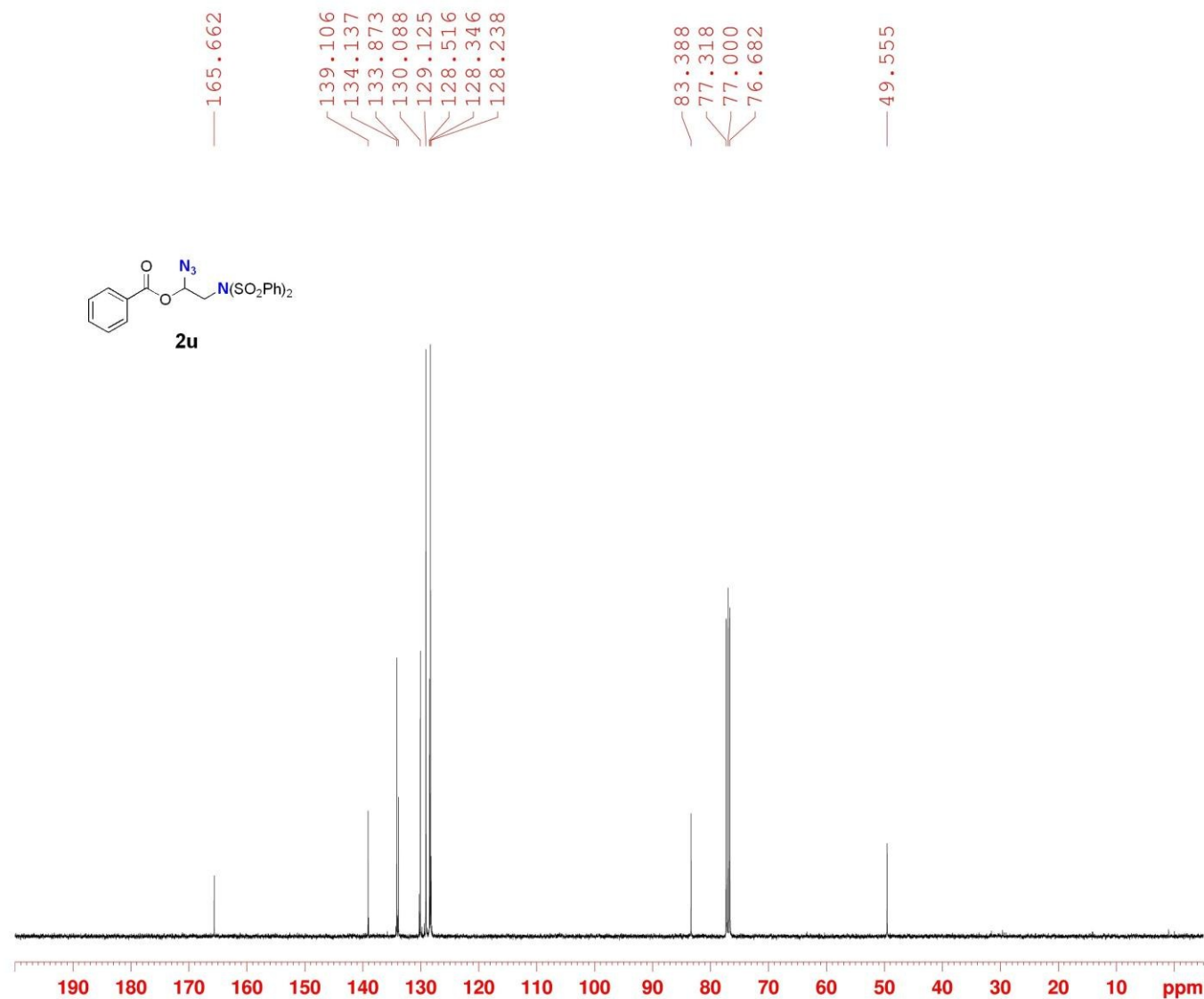
NAME 2017-11-21 leibowen-159
 EXPNO 1
 PROCNO 1
 Date_ 20171121
 Time 10.20 h
 INSTRUM spect
 PROBHD Z116098_0673 (
 PULPROG zg30
 TD 65536
 SOLVENT CDCl3
 NS 16
 DS 2
 SWH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 4.0894966 sec
 RG 31.12
 DW 62.400 usec
 DE 6.50 usec
 TE 296.4 K
 D1 1.00000000 sec
 TD0 1
 SFO1 400.1324708 MHz
 NUC1 1H
 P1 10.00 usec
 SI 65536
 SF 400.1300101 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00



```

NAME      2017-11-21 leibowen-159
EXPNO     2
PROCNO    1
Date_     20171121
Time      10.36 h
INSTRUM   spect
PROBHD    Z116098_0673 (
PULPROG   zgpg30
TD         65536
SOLVENT   CDCl3
NS         256
DS         4
SWH        24038.461 Hz
FIDRES     0.733596 Hz
AQ         1.3631988 sec
RG         203.48
DW         20.800 usec
DE         6.50 usec
TE         297.0 K
D1         2.00000000 sec
D11        0.03000000 sec
TD0        1
SF01       100.6228298 MHz
NUC1       13C
P1         10.00 usec
SI         32768
SF         100.6127763 MHz
WDW        EM
SSB        0
LB         1.00 Hz
GB         0
PC         1.40
    
```





NAME CLJ-WL-L173
EXPNO 2
PROCNO 1
Date_ 20171124
Time 12.22
INSTRUM spect
PROBHD 5 mm PABBO BB/
PULPROG zgpg30
TD 65536
SOLVENT CDC13
NS 512
DS 0
SWH 24038.461 Hz
FIDRES 0.366798 Hz
AQ 1.3631988 sec
RG 194.26
DW 20.800 usec
DE 6.50 usec
TE 300.0 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1

===== CHANNEL f1 =====
SFO1 100.6228293 MHz
NUC1 13C
P1 8.54 usec
SI 32768
SF 100.6127731 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40

