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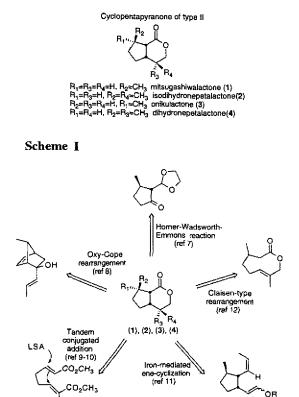
Formal Synthesis of (±)-Mitsugashiwalactone and (±)-Isodihydronepetalactone from Norborn-5-en-2-one Involving Shapiro Reaction

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A formal synthesis of (\pm) -mitsugashiwalactone (1) and (\pm) -isodihydro nepetalactone (2) was accomplished. Baeyer-Villiger lactonization of ketone 9 followed by acidic treatment led to the rearranged lactone 8, which underwent a series of functional group transformations to give cyclopentanone derivatives 19 and 20. Shapiro reaction on 21 and 22 in the presence of excess dry ice gave lactones 5 and 6. Lactones 5 and 6 previously have been converted to mitsugashiwalactone (1) and isodihydronepetalactone (2), respectively.

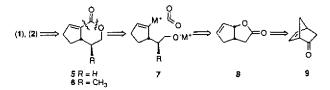
INTRODUCTION

The iridoids comprise a large family of monoterpenoid characterized by a cyclopentane ring *cis*-fused to a dihydropyran or δ -lactone.¹⁻³ These iridoid monoterpenoids have contiguous chiral centers on their molecular system and pronounced physiological activities.⁴⁻⁶ Strategies adopted for the synthesis of cyclopentapyranones of type II,^{3,7} such as mitsugashiwalactone (1), isodihydronepetalactone (2), onikulactone (3), and dihydronepetalactone (4) have attracted the attention of synthetic chemists and generally fallen into one of the five categories⁷⁻¹² (as shown in Scheme I).



Sakan and his co-workers isolated mitsugashiwalactone (1) and isodihydronepetalactone (2) from *Boschniakia rossica* Hult.¹³ and *Actinidia polygama* Miq.,^{4,6} respectively. The biological activities toward *Felidae* animals and insecticidal properties are the focal points of the synthetic approaches.^{7,9-12,14-24} Our plan for a general entry to the cyclopentapyranones of type II represented a different approach for constructing the fused bicyclic framework. We now wish to describe the application of Shapiro reaction²⁵ to the synthesis of two simple iridoid monoterpene lactones. The precursor **5** and **6** of mitsugashiwalactone (1) and isodihydronepetalactone (2) are obtained from norborn-5-en-2one (9)²⁶⁻²⁸ via a series of functional group transformations (Scheme II).

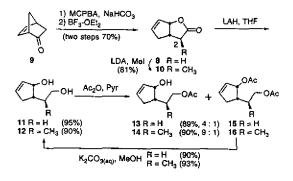




RESULTS AND DISCUSSION

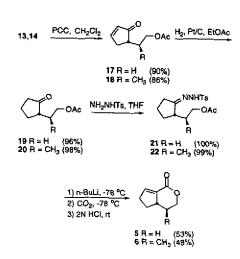
As shown in Scheme III and IV, our synthesis started with lactone 8, which was easily available by Baeyer-Villiger oxidation of norborn-5-en-2-one (9) followed by Lewis acid (boron trifluoride etherate) mediated allylic rearrangement in nearly 70% overall yield. The stereochemistry is evidently dictated by the preference for maintaining a *cis* ring junction at the bicyclo[3.3.0]octane ring system. Methylation of lactone 8 by using lithium diisopropylamide as base established the configuration at C-2 in lactone 10. Reduction of lactones 8 and 10 with lithium aluminum hydride, followed by acetylation of the resulting diols 11 and 12, gave the monoacetylated alcohols 13 and 14 as major products. The diacetylation products 15 and 16 were easily transformed back to diols 11 and 12.

Scheme III



With allylic alcohols 13 and 14 in hand, the alcohols were oxidized with pyridinium chlorochromate to give the corresponding enones 17 and 18. Hydrogenation of enones 17 and 18 with platinium on activated carbon in ethyl acetate gave ketones 19 and 20, respectively. Condensation of 19 and 20 with 4-toluenesulfonylhydrazine afforded hydrazones 21 and 22 in nearly quantitative yields. Finally, conversion of 21 and 22 to cyclopentapyranone $5^{14-16,18}$ and $6^{9-10,23}$ was accomplished *via* Shapiro reaction. The lactone 5 was identical with the authentic sample by Professor Ogasawara²⁹ and the ¹H NMR spectrum of lactone 6 is identical to those reported by Professor Uyehara and Yamamoto.³⁰ The above procedures constitute a new approach to the formal synthesis of mitsugashiwalactone (1) and isodihydronepetalactone (2).

Scheme IV



EXPERIMENTAL SECTION

General

Tetrahydrofuran (THF) was distilled prior to use from a deep-blue solution of sodium-benzophenone ketyl. All other reagents and solvents were obtained from commercial sources and used without further purification. Reactions were routinely carried out under an atmosphere of dry nitrogen with magnetic stirring. The extracted solution of products were dried with anhydrous MgSO₄ before concentration *in vacuo*. Crude products were purified by preparative TLC or column chromatography on silica gel. All reported temperatures are uncorrected.

(3aS*,6aS*)-3,3a,4,6a-Tetrahydro-2H-cyclopenta[b]furan-2-one (8)

A solution of norborn-5-en-2-one (9) (10.0 g, 92.6 mmol) in methylene chloride (50 mL) was added dropwise at 0 °C to a solution of m-chloroperoxybenzoic acid (16.8 g, 97.3 mmol) and sodium bicarbonate (30 g) in methylene chloride (200 mL), and the resulting mixture was stirred at this temperature for 6 h. The precipitate was filtered off, and the filtrate was washed with aqueous saturated sodium bicarbonate solution $(3 \times 50 \text{ mL})$. The organic layer was evaporated to ca. 20 mL. The resulting solution was treated with boron trifluoride etherate (1 mL) and stirred at 0 °C for 20 min, then saturated aqueous sodium bicarbonate solution (10 mL) was added. The organic layer was separated and the aqueous phase was extracted with ethyl acetate (3×20) mL). The combined organic extracts were washed with brine, dried (MgSO₄), and concentrated to give crude lactone 8. Chromatography on silica gel (hexane:ethyl acetate = 10:1) afforded 8.0 g (70%) of lactone 8 as a colorless oil: IR (CHCl₃) 1768 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 2.28-2.36 (m, 2H), 2.73-2.89 (m, 2H), 3.10-3.20 (m, 1H), 5.53 (d, J = 7.5 Hz, 1H), 5.86-5.90 (m, 1H), 6.08-6.11 (m, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 34.9, 35.9, 39.4, 89.5, 128.8, 136.8, 177.0; MS (30 eV) m/z 124 (M⁺, 15), 79 (100); HRMS calcd C7H8O2 124.0524, found 124.0515.

(3S*,3aS*,6aS*)-3-Methyl-3,3a,4,6a-tetrahydro-2H-cyclopenta[b]furan-2-one (10)

To a solution of lithium diisopropylamide, prepared from 4.4 mL (31.5 mmol) of diisopropylamine in 100 mL of freshly distilled tetrahydrofuran and 18.1 mL (29.0 mmol) of *n*-butyllithium (1.6 M in hexane) at -78 °C, was added a solution of 3.0 g (24.2 mmol) of lactone 8 in 10 mL of tetrahydrofuran. After stirring this mixture for an additional 30 min at -78 °C, 1.6 mL (25.5 mmol) methyl iodide was added. The reaction mixture was stirred at -78 °C for 6 h, warmed to 25 °C over 1 h. The reaction was quenched with water, and the solvent was removed under reduced pressure. To residue was added 10 mL of water and was extracted with ethyl acetate (3 × 30 mL). The combined organic extracts were washed with brine, dried (MgSO₄), filtered and concentrated to afford crude 10. Chromatography on silica gel (hexane:ethyl acetate = 10:1) afforded 2.7 g (81%) of 10 as a colorless oil: IR (CHCl₃) 1760 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.32 (d, *J* = 7.5 Hz, 3H), 2.28-2.40 (m, 2H), 2.66-2.70 (m, 2H), 5.45-5.47 (m, 1H), 5.85-5.88 (m, 1H), 6.04-6.06 (m, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 15.7, 38.0, 42.0, 44.0, 87.2, 129.3, 135.9, 179.6; MS (30 eV) *m/z* 138 (M⁺, 1), 79 (100); HRMS calcd C₈H₁₀O₂ 138.0681, found 138.0688.

$2-[(1S^*,2S^*)-2-Hydroxy-3-cyclopentenyl]ethyl Acetate (13) and <math>2-[(1S^*,2S^*)-2-Methylcarbonyloxy-3-cyclopentenyl]ethyl Acetate (15)$

To a stirred cold (0 °C) slurry of lithium aluminum hydride (0.6 g, 16.0 mmol) in tetrahydrofuran (50 mL) was added a solution of lactone 8 (2.0 g, 16.1 mmol) in tetrahydrofuran (10 mL) via syringe. The mixture was refluxed for 4 h, quenched with saturated aqueous ammonium chloride solution (5 mL) under cooling, and concentrated. The residue was diluted with water (15 mL) and extracted with ethyl acetate $(3 \times 30 \text{ mL})$. The organic layer was washed with brine and water, dried (MgSO₄), filtered and concentrated to afford crude 11 (1.96 g) as a colorless oil. Without purification, the acetic anhydride (2.0 g, 19.6 mmol) was added dropwise to a cold (0 °C), magnetically stirred solution of diol 11 (1.9 g, 14.8 mmol) in pyridine (10 mL). After being allowed to warm to 25 °C, the mixture was stirred for 2 h. Then water (10 mL) was added to the mixture and extracted with ethyl acetate $(3 \times 30 \text{ mL})$. The combined organic extracts were washed with water and brine, dried (MgSO₄), filtered and concentrated to produce crude products. Chromatography on silica gel (hexane:ethyl acetate = 3:1) afforded monoacetate 13 (1.8 g, 71%) and diacetate 15 (0.6 g, 18%) as a colorless oil:

13: IR (CHCl₃) 1729, 3450 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 2.07 (s, 3H), 1.74-2.19 (m, 5H), 2.37-2.50 (m, 1H), 4.10-4.25 (m, 2H), 4.59-4.63 (m, 1H), 5.90-5.94 (m, 1H), 6.02-6.04 (m, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 21.0, 28.0, 36.7, 39.6, 64.1, 76.3, 132.9, 135.6, 171.1; MS (30 eV) *m*/z 170 (M⁺, 1), 110 (100); HRMS calcd C₉H₁₄O₃ 170.0943, found 170.0937.

15: IR (CHCl₃) 1729 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.68-1.78 (m, 1H), 1.87-1.99 (m, 1H), 2.04 (s,

3H), 2.05 (s, 3H), 2.16-2.24 (m, 1H), 2.33-2.45 (m, 1H), 2.46-2.53 (m, 1H), 4.11 (dd, J = 6.6, 6.6 Hz, 2H), 5.59-5.62 (m, 1H), 5.86-5.90 (m, 1H), 6.11-6.14 (m, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 21.0, 21.1, 28.1, 37.1, 38.2, 63.5, 79.2, 129.5, 137.5, 170.7, 171.0.

(2S*)-2-[(1S*,2S*)-2-Hydroxy-3-cyclopentenyl]propyl Acetate (14) and (2S*)-2-[(1S*,2S*)-2-Methylcarbonyloxy-3-cyclopentenyl]propyl Acetate (16)

The title compound was prepared with the same procedures as for 13 to 15.

14: IR (CHCl₃) 1728, 3458 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.03 (d, J = 6.9 Hz, 3H), 1.75-1.86 (m, 1H), 1.99 (br s, 1H), 2.08 (s, 3H), 2.04-2.39 (m, 3H), 3.99 (dd, J =10.8, 6.6 Hz, 1H), 4.23 (dd, J = 10.8, 4.5 Hz, 1H), 4.60-4.62 (m, 1H), 5.94-5.96 (m, 1H), 6.01-6.05 (m, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 16.3, 21.0, 32.1, 35.0, 46.5, 69.2, 75.1, 133.0, 135.7, 171.3; MS (30 eV) *m*/z 184 (M⁺, 2), 107 (84), 43 (100); HRMS calcd C₁₀H₁₆O₃ 184.1100, found 184.1095.

16: IR (CHCl₃) 1730 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.02 (d, J = 6.3 Hz, 3H), 2.03 (s, 3H), 2.05 (s, 3H), 2.08-2.18 (m, 2H), 2.23-2.32 (m, 1H), 2.36-2.45 (m, 1H), 3.86 (dd, J = 10.8, 6.3 Hz, 1H), 4.10 (dd, J = 10.8, 4.5 Hz, 1H), 5.50-5.53 (m, 1H), 5.94-5.98 (m, 1H), 6.11-6.15 (m, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 16.0, 20.9, 21.2, 31.9, 35.3, 44.4, 68.4, 78.2, 130.0, 137.5, 170.7, 171.0.

Conversion of 15 and 16 to 11 and 12.

To a solution of diacetylated product 15 (100 mg) in methanol (10 mL) was added saturated aqueous potassium carbonate solution (3 mL). After 2 h of stirring at room temperature, most of the methanol was evaporated under reduced pressure. The residue was dissolved in water and extracted with ethyl acetate (3×20 mL). The combined organic extracts were washed with water and brine, dried (MgSO₄), filtered and concentrated to produce crude product 11. Chromatography on silica gel (hexane:ethyl acetate = 3:1) afforded diol 11 (55 mg, 90%). 16 was transformed to 12 in 93% yield with the same reaction conditions and procedure.

2-(2-Oxo-3-cyclopentenyl)ethyl Acetate (17) and (2S*)-2-[(1S*)-2-Oxo-3-cyclopentenyl]propyl Acetate (18)

The allylic alcohol 13 (1.5 g, 8.8 mmol) in methylene chloride (10 mL) was added to a mixture of pyridinium chlorochromate (2.3 g, 10.7 mmol) and Celite (5.0 g) in methylene chloride (50 mL). After being stirred at room temperature for 2 h, the mixture was diluted with ethyl acetate (20 mL) and filtered through a short silica gel column. The filtrate was washed with water (2 \times 10 mL), dried

(MgSO₄), filtered and concentrated to produce crude enone 17. Chromatography on silica gel (hexane:ethyl acetate = 4:1) afforded enone 17 (1.3 g, 90%). 14 was transformed to 18 with the same reaction conditions and procedure in 86% yield.

17: IR (CHCl₃) 1694, 1737 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.60-1.72 (m, 1H), 2.05 (s, 3H), 2.12-2.23 (m, 1H), 2.37-2.49 (m, 2H), 2.94 (ddt, J = 19.2, 6.6, 2.4 Hz, 1H), 4.20 (dd, J = 6.6, 6.6 Hz, 2H), 6.21 (ddd, J = 5.4, 4.2, 1.8 Hz, 1H), 7.71 (dd, J = 5.4, 2.4 Hz, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 20.8, 30.1, 35.5, 41.9, 62.5, 133.5, 163.2, 170.8, 211.0; MS (30 eV) *m*/z 168 (M⁺, 1), 82 (100); HRMS calcd C₉H₁₂O₃ 168.0787, found 168.0784.

18: IR (CHCl₃) 1698, 1729 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.91 (d, J = 7.2 Hz, 3H), 2.09 (s, 3H), 2.05-2.13 (m, 1H), 2.15 (dd, J = 18.9, 2.7 Hz, 1H), 2.43 (dd, J = 18.9, 6.6 Hz, 1H), 3.05-3.20 (m, 1H), 3.97 (dd, J = 11.1, 6.6 Hz, 1H), 4.07 (dd, J = 11.1, 5.7 Hz, 1H), 6.23 (dd, J = 5.7, 2.4 Hz, 1H), 7.64 (dd, J = 5.7, 2.4 Hz, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 13.2, 20.8, 35.3, 37.2, 43.6, 67.3, 134.6, 166.3, 170.8, 209.2; MS (30 eV) *m*/z 183 (M*+1, 3), 122 (92), 94 (100); HRMS calcd C₁₀H₁₅O₃ [M*+1] 183.1022, found 183.1017.

2-(2-Oxo-cyclopentyl)ethyl Acetate (19) and (2S*)-2-[(1S*)-2-Oxo-cyclopentyl]propyl Acetate (20)

A solution of enone 17 (1.1 g, 6.5 mmol) in ethyl acetate (30 mL) was stirred under 1 atm of hydrogen at room temperature with 10% platinium in charcoal (20 mg) for 2 h. The mixture was filtered, and the filtrate was evaporated to give crude ketone 19. Chromatography on silica gel (hexane:ethyl acetate = 4:1) afforded ketone 19 (1.06 g, 96%). 18 was transformed to 20 in 98% yield with the same reaction conditions and procedure.

19: IR (CHCl₃) 1730 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.50-1.65 (m, 2H), 1.72-1.88 (m, 1H), 2.04 (s, 3H), 2.00-2.19 (m, 4H), 2.24-2.39 (m, 2H), 4.16 (dd, *J* = 6.6, 6.6 Hz, 2H); ¹³C NMR (CDCl₃, 75 MHz) δ 20.7, 20.9, 28.6, 29.6, 37.7, 46.3, 62.6, 170.9, 220.0; MS (30 eV) *m*/z 170 (M⁺, 1), 110 (82), 99 (80), 84 (100); HRMS calcd C₉H₁₄O₃ 170.0943, found 170.0938.

20: IR (CHCl₃) 1729 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.82 (d, J = 6.6 Hz, 3H), 1.60-1.78 (m, 2H), 2.06 (s, 3H), 2.00-2.10 (m, 3H), 2.20-2.50 (m, 3H), 3.90-4.10 (m, 2H); ¹³C NMR (CDCl₃, 75 MHz) δ 12.7, 20.6, 20.9, 24.1, 31.5, 38.7, 50.7, 67.6, 170.9, 220.0; MS (30 eV) *m*/*z* 184 (M⁺, 18), 124 (55), 84 (51), 43 (100); HRMS calcd C₁₀H₁₆O₃ 184.1100, found 184.1103.

1,3,4,4a,5,6-Hexahydrocyclopenta[c]oxin-1-one (5) and (4S*,4aS*)-4-Methyl-1,3,4,4a,5,6-hexahydrocyclopenta-[c]oxin-1-one (6)

p-Toluenesulfonylhydrazine (165 mg, 0.88 mmol) and ketone 19 (150 mg, 0.88 mmol) in 30 mL of tetrahydrofuran were stirred at room temperature for 6 h. The mixture was concentrated in vacuo to give crude 21 (mp 98-100 °C) in excellent yield. The tosylhydrazone 21 (150 mg) was dissolved in 40 mL of anhydrous tetrahydrofuran under a nitrogen atmosphere. To the solution was added n-butyllithium in n-hexane (1.6 M, 2.0 mL, 3.2 mmol) at -78 °C and the mixture was stirred at this temperature for 1 h and the reaction mixture was warmed to 25 °C over 1 h and cooled at -78 [°]C again. To the reaction mixture was added excess of dry ice and the solution was stirred at -78 °C for 1 h. The reaction was quenched with 2 N aqueous hydrogen chloride (3 mL) at room temperature, and the mixture was stirred at room temperature for 1 h. The solvent was removed under reduced pressure to give the residues which were taken up with 10 mL of water and extracted with ethyl acetate (3×30) mL). The combined organic extracts were washed with brine, dried (MgSO₄), filtered and concentrated to afford crude 5. Chromatography on silica gel (hexane:ethyl acetate = 10:1) afforded 64 mg (53%) of 5 as a colorless oil. 20 was transformed to 6 in 48% yield with the same reaction conditions and procedure. Tosylhydrazone 22 mp 84-86 °C.

5: IR (CHCl₃) 1719 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.59-1.74 (m, 2H), 2.06-2.14 (m, 1H), 2.34-2.50 (m, 3H), 2.90-3.10 (m, 1H), 4.27-4.32 (m, 1H), 4.35-4.49 (m, 1H), 6.99-7.01 (m, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 30.8, 31.7, 32.7, 42.0, 69.7, 134.6, 145.5, 163.6; MS (30 eV) *m*/z 138 (M^{*}, 38), 110 (30), 79 (100); HRMS calcd C₈H₁₀O₂ 138.0681, found 138.0685.

6: IR (CHCl₃) 1744 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.97 (d, J = 6.3 Hz, 3H), 1.50-1.60 (m, 2H), 1.70-1.75 (m, 1H), 2.30-2.70 (m, 3H), 3.94 (dd, J = 11.7, 11.7 Hz, 1H), 4.29 (dd, J = 11.7, 4.5 Hz, 1H), 7.00 (dd, J = 2.7, 2.4 Hz, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 14.3, 31.1, 31.8, 36.2, 48.9, 75.2, 134.0, 145.7, 163.4; MS (30 eV) m/z 152 (M⁺, 40), 122 (40), 93 (85), 79 (100); HRMS calcd C₉H₁₂O₂ 152.0838, found 152.0838.

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Key Words

Iridoid; Mitsugashiwalactone; Isodihydronepetalactone; Dihydronepetalactone; Onikulactone; Shapiro reaction; Baeyer-Villiger oxidation.

REFERENCES

- Demuth, M.; Schaffner, K. Angew. Chem. Int. Ed. Engl. 1982, 21, 820.
- 2. Tietze, L. F. Angew. Chem. Int. Ed. Engl. 1983, 22, 828.
- Nangia, A.; Prasuna, G.; Rao, P. B. Tetrahedron 1997, 53, 14507.
- Sakan, T.; Isoe, S.; Hyeon, S. B.; Katsumura, R.; Maeda, T.; Wolinsky, J.; Dickerson, D.; Slabaugh, M.; Nelson, D. Tetrahedron Lett. 1965, 4097.
- Pagnoni, U. M.; Pinetti, A.; Trave, R.; Garanti, L. Aust. J. Chem. 1976, 29, 1375.
- Sakai, T.; Nakajima, K.; Sakan, T. Bull. Chem. Soc. Jpn. 1980, 53, 3683.
- 7. Nangia, A.; Prasuna, G. Tetrahedron 1996, 52, 3435.
- 8. Fleming, I.; Terrett, N. K. Tetrahedron Lett. 1984, 25, 5103.
- Uyehara, T.; Shida, N.; Yamamoto, Y. J. Org. Chem. 1992, 57, 3139.
- Uyehara, T.; Shida, N.; Yamamoto, Y. J. Chem. Soc., Chem. Commun. 1989, 113.
- Takacs, J. M.; Myoung, Y. C. Tetrahedron Lett. 1992, 33, 317.
- 12. Funk, R. L.; Abelman, M. M.; Munger, J. D. Tetrahedron 1986, 42, 2831.
- 13. Sakan, T.; Murai, F.; Isoe, S.; Hyeon, S. B.; Hayashi, Y.

J. Chem. Soc. Jpn., Pure Chem. Sect. 1969, 90, 507.

- Yamane, T.; Takabashi, M.; Ogasawara, K. Synthesis 1995, 444.
- Amri, H.; Rambaud, M.; Villieras, J. Tetrahedron 1990, 46, 3535.
- 16. Amri, H.; Villieras, J. Tetrahedron Lett. 1987, 28, 5521.
- 17. Mikami, K.; Takahashi, K.; Nakai, T. Synlett 1989, 45.
- Nugent, W. A.; Hobbs, F. W. Jr. J. Org. Chem. 1986, 51, 3376.
- 19. Ohta, H.; Kobori, T.; Fujisawa, T. J. Org. Chem. 1977, 42, 12312.
- Fujisawa, T.; Kobori, T.; Ohta, H. J. Chem. Soc., Chem. Commun. 1976, 186.
- 21. Ficini, J.; d'Angelo, J. Tetrahedron Lett. 1976, 6087.
- Kigawa, M.; Tanaka, M.; Mitsuhashi, H.; Wakamatsu, T. Heterocycles 1992, 33, 117.
- 23. Tanimor, S.; Nakayama, M. Agric. Biol. Chem. 1991, 55, 1181.
- Wolinsky, J.; Eustace, E. J. J. Org. Chem. 1972, 37, 3376.
- Kolonko, K. J.; Shapiro, R. H. J. Org. Chem. 1978, 43, 12312.
- 26. Greeen, A. E.; Drian, C. L.; Crabbe, P. J. Am. Chem. Soc. 1980, 102, 7583.
- 27. Baumgartner, H.; Marschner, C.; Pucher, R.; Griengl, H. Tetrahedron Lett. 1991, 32, 611.
- 28. Grieco, P. A.; Ohfune, Y.; Yokoyama, Y.; Owens, W. J. Am. Chem. Soc. 1979, 101, 4749.
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