

Asymmetric Total Synthesis of a Pentacyclic *Lycopodium* Alkaloid: Huperzine-Q**

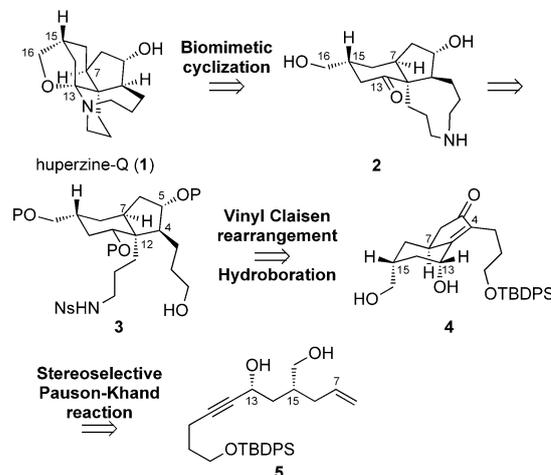
Atsushi Nakayama, Noriyuki Kogure, Mariko Kitajima, and Hiromitsu Takayama*

Lycopodium alkaloids have unique skeletal characteristics^[1] and a variety of biological activities, such as acetylcholine esterase (AChE) inhibition^[2] and neurite outgrowth promotion,^[3] which have sustained the interest of many researchers of natural product chemistry, synthetic chemistry, and medicinal chemistry.

In particular, the structural diversity of fawcettimine-type *Lycopodium* alkaloids has attracted the attention of several groups as targets for total synthesis.^[4]

Huperzine-Q (**1**; Figure 1), which was isolated from *Huperzia serrata* by Zhu and co-workers in 2002,^[5] consists of a unique pentacyclic skeleton possessing a spiroaminal moiety and six stereogenic centers, including a quaternary carbon center. Although its structure and relative stereochemistry were determined by spectroscopic and single-crystal X-ray diffraction analysis, its absolute configuration and biological activities have not been reported thus far. To develop an efficient synthetic route to **1** and to confirm its absolute configuration, we embarked on the asymmetric total synthesis of huperzine-Q (**1**).

Our synthetic plan is shown in Scheme 1. Biogenetically, **1** would be derived from the fawcettimine derivative **2** by



Scheme 1. Retrosynthetic analysis of huperzine-Q (**1**). Ns = 2-nitrobenzenesulfonyl, TBDPS = *tert*-butyldiphenylsilyl.

intramolecular spiroaminal formation between a primary alcohol at C16 and a secondary amine. We anticipated an efficient synthesis of **2** to arise from azonane ring formation by utilizing the intramolecular Mitsunobu reaction and subsequent functional group transformations of **3**, which could be a key intermediate to several fawcettimine-type *Lycopodium* alkaloids such as lycoposerramine-A^[6] (Figure 1). We envisioned that successive chiral centers (C5, C4, and C12) in **3** could be constructed from the bicyclic cyclopentenone **4** by means of a vinyl Claisen rearrangement, and the subsequent hydroboration/oxidation process. Bicyclic compound **4** was expected to be elaborated from the chiral diol **5** through the novel stereoselective Pauson–Khand reaction (PKR; Scheme 1).

Our synthesis commenced with the coupling between the acyl chloride **6** and alkyne **7** to afford ketone **8**,^[7] which was transformed into the optically active lactone **9** in a one-pot operation involving the Noyori reduction^[8] and successive treatment with PPTS. The enantiomeric excess was determined to be 83% by HPLC analysis using a chiral stationary phase^[9] (the enantiomeric excess of the product was finally raised to 99% *ee* during conversion into **15**. See below). Then, an allyl unit was introduced to the α position of the carbonyl group in **9** to furnish **10** and **11** in quantitative yields in a ratio of 2.3:1.^[10] The conversion of **10** into **11**, having the desired stereochemistry at C15, was successfully achieved by treatment with LHMDS and subsequent addition of a hindered acid (BHT), thus giving the kinetically controlled product **11** with excellent selectivity (**10/11** = 1:16.5). The reduction of lactone **11** afforded diol **5** for the PKR (Scheme 2).

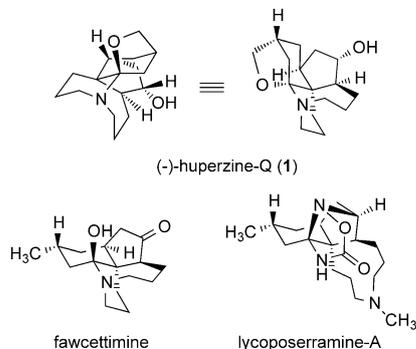


Figure 1. Structures of huperzine-Q (**1**), fawcettimine, and lycoposerramine-A.

[*] A. Nakayama, Dr. N. Kogure, Dr. M. Kitajima, Prof. Dr. H. Takayama Graduate School of Pharmaceutical Sciences, Chiba University 1-33 Yayoi-cho, Inage-ku, Chiba 263-8522 (Japan)
E-mail: htakayam@p.chiba-u.ac.jp
Homepage: <http://www.p.chiba-u.ac.jp/lab/seitai/index.html>

[**] This work was supported by a Grant-in-Aid for Scientific Research from the Japan Society for the Promotion of Science and the Takeda Science Foundation.

Supporting information for this article is available on the WWW under <http://dx.doi.org/10.1002/anie.201103550>.

Initial attempts to perform the PKR with **5** gave, however, **12**, the undesired C7 epimer as the major product. Mechanistic considerations indicated that a reaction intermediate such as **5i** would have a chairlike conformation with an equatorial side chain at C15, and therefore control the stereochemistry at C7. On this basis, we devised the silyl-tethered compound **13**, which would alter the conformation of the reaction intermediate to yield a bicyclic product having the desired C7 stereochemistry. Actually, the silyl-tethered

compound **13** gave the desired compound **14** in 57% yield under conventional PKR conditions.^[11] NOE experiments showed that C7 in **14** had the desired stereochemistry (Scheme 3). Next, we optimized the reaction conditions to develop the one-pot operation to transform **5** into **4**. First, we prepared the silyl-tethered compound **13**, to which $[\text{Co}_2(\text{CO})_8]$ was added, thus affording the alkyne cobalt complex of **13**. Then, we diluted the reaction mixture with toluene and heated it under CO atmosphere to give the bicyclic compound **14**, which in turn was directly treated with concentrated hydrochloric acid in MeOH to give the desilylated compound **4** in 92% yield from **5**. To the best of our knowledge, this is a first example of a stereoselective PKR that utilizes a seven-membered silyl-tethered compound.

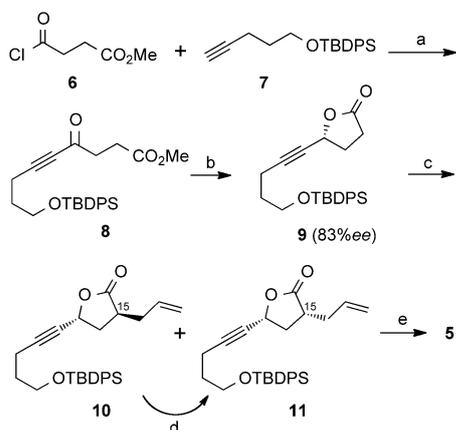
With the PKR product **4** in hand, we next focused on the construction of the quaternary carbon center C12 (Scheme 4). MOM groups were introduced to the two hydroxy groups in **4** and then the enone was reduced with the (*R*)-Me-CBS reagent^[12] to furnish allyl alcohol **15** with good stereoselectivity. The stereochemistry at C5 was determined by NOE experiments, and at this point, the enantiomeric excess of **15** was determined by HPLC analysis to be 99% ee.^[13]

After conversion of **15** into sulfoxide **16**,^[14] **16** was heated at 170°C in 1,2-dichlorobenzene to afford the desired aldehyde **17** in excellent yield. Treatment of aldehyde **17** with the Wittig reagent gave the diene compound **18**.

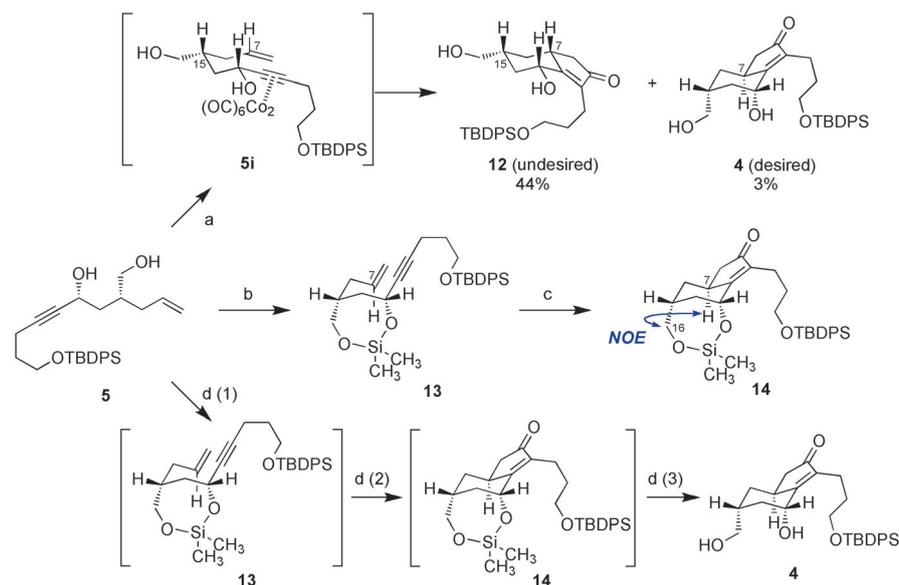
Our next task was the construction of an azonane ring using the intramolecular Mitsunobu reaction (Scheme 5). We prepared substrate **19** for the Mitsunobu reaction from diene **18** by using a sequential reaction that involved a simultaneous hydroboration/oxidation^[15] at two positions (C4–C5 and C9–C10), the introduction of a nitrogen functional group to C9, and the subsequent removal of a TBDPS group. With the Ns-

protected derivative **19** in hand, we tried to construct the azonane ring. After several screening steps, we found that the treatment of **19** with diethyl azodicarboxylate (DEAD) in the presence of PPh_3 in toluene at 70°C afforded **20** in excellent yield.^[4e,16] At this stage, the X-ray crystallographic analysis of **20** was performed, and enabled us to confirm the absolute configuration of all the chiral centers.^[17]

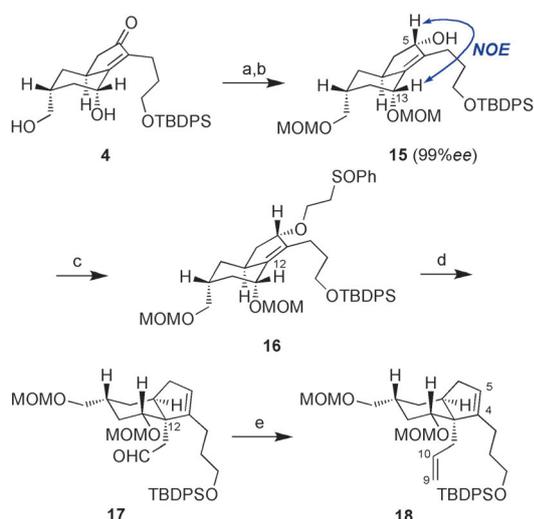
For the completion of the total synthesis of **1**, we converted **20** into the fawcettimine derivative **2** as follows (Scheme 6). The removal of the two MOM groups with trimethylsilyl bromide gave the corresponding diol **21** in quantitative yield.^[18] Then, the selective acetylation of the primary alcohol^[19] at C16 and the subsequent Dess–Martin oxidation of the secondary alcohol at C13 were carried out in a one-pot operation to afford **22**. The successive removal of the Ns group and diacetyl groups was also



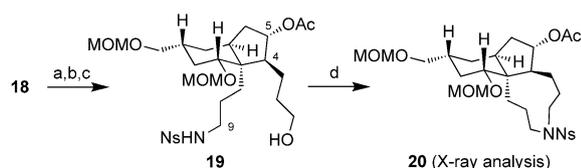
Scheme 2. Synthesis of the chiral diol **5**. Reagents and conditions: a) $n\text{BuLi}$, ZnCl_2 , THF, -78°C , 91%; b) $[\text{Ru}\{(\text{R,R})\text{-Tsdpen}\}(p\text{-cymene})]$, $i\text{PrOH}$, 28°C and then PPTS, toluene, 80°C , 68%; c) Allylbromide, LHMDS, HMPA, THF, -78°C , quant, $10/11 = 2.3:1$; d) LHMDS, THF, -78°C and then BHT, 86%, $10/11 = 1:16.5$; e) LiBH_4 , THF, RT, 95%. BHT = 2,6-di-*t*-butylated-hydroxytoluene, dpen = 1,2-diphenylethylenediamine, HMPA = hexamethylphosphoramide, LHMDS = lithium bis(trimethylsilyl)amide, PPTS = pyridinium *p*-toluenesulfonate, THF = tetrahydrofuran.



Scheme 3. Stereoselective synthesis of bicyclic compounds **12** or **14** and one-pot operation to give **4**. Reagents and conditions: a) $[\text{Co}_2(\text{CO})_8]$ toluene, RT to 100°C under CO atmosphere, 44% of **12** and 3% of **4**; b) SiMe_2Cl_2 , Et_3N , DMAP, CH_2Cl_2 , RT, 85%; c) $[\text{Co}_2(\text{CO})_8]$, toluene, RT to 100°C under CO atmosphere, 57%; d) 1. SiMe_2Cl_2 , Et_3N , DMAP, $(\text{CH}_2\text{Cl})_2$, RT, 2. $[\text{Co}_2(\text{CO})_8]$, toluene, RT to 100°C under CO atmosphere, 3. HCl, MeOH, 0°C , 92%. DMAP = *N,N*-4-dimethylaminopyridine.



Scheme 4. Synthesis of the diene **18**. Reagents and conditions: a) MOMCl, DIPEA, TBAI, CH₂Cl₂, RT, 84%; b) (*R*)-Me-CBS, BH₃·THF, THF, RT, 88% (d.r. 9.8:1); c) Phenylvinylsulfoxide, KH, NaH, THF, RT, 98%; d) NaHCO₃, 1,2-dichlorobenzene, 170 °C, 89%; e) *n*BuLi, PPh₃CH₃Br, THF, RT, 95%. (*R*)-Me-CBS = (*R*)-methyloxazaborolidine, DIPEA = diisopropylethylamine, MOM = methoxymethyl, TBAI = tetra-*n*-butylammonium iodide.

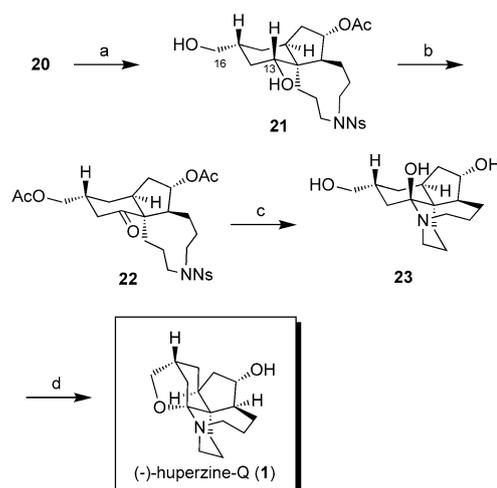


Scheme 5. Synthesis of the azonane compound **20**. Reagents and conditions: a) BH₃·SMe₂, THF, 0 °C; BH₃·THF, 0 °C; then NaBO₃·4 H₂O, RT, 67%; b) 1. MsCl, Et₃N, CH₂Cl₂, 0 °C then Ac₂O, DMAP, pyridine; 2. NH₂Ns, K₂CO₃, DMF, 80 °C, 97%; c) TBAF, AcOH, THF, RT, quant; d) DEAD, PPh₃, toluene, 70 °C, 94%. DEAD = diethyl azodicarboxylate, DMF = *N,N*-dimethylformamide, Ms = methanesulfonyl, Ns = 2-nitrobenzenesulfonyl, TBAF = tetra-*n*-butylammonium fluoride.

achieved in a one-pot operation to give a product that was proven to exist as carbinolamine form **23** by analysis of NMR spectra.

Then, we attempted to convert **23** into the spiroaminal form **1** based on a biogenetic hypothesis. After considerable efforts, we found that this spiroaminal formation occurred by treating of **23** with anhydrous (+)-camphorsulfonic acid in refluxing toluene to furnish (–)-huperzine-Q (**1**) in 86% yield. By direct comparison with natural huperzine-Q, which was isolated from *Lycopodium serratum* in our laboratory, we found that synthetic **1** was completely identical in all respects with the natural product, including the optical properties.^[20] Hence, the structure including the absolute configuration was confirmed.

In summary, the first asymmetric total synthesis of (–)-huperzine-Q (**1**) has been achieved in 19 steps and 16.4% overall yield starting from methyl-4-chloro-4-oxobutylate (**8**). The synthesis involved a novel stereoselective PKR utilizing a



Scheme 6. Completion of total synthesis of (–)-huperzine-Q (**1**). Reagents and conditions: a) TMSBr, CH₂Cl₂, 0 °C, quant; b) AcCl, 2,6-lutidine, CH₂Cl₂, –78 °C and then Dess–Martin periodinane, RT, 96%; c) PhSH, K₂CO₃, MeCN, 30 °C and then MeOH, K₂CO₃, 30 °C, 98%; d) CSA, toluene, reflux, 86%. CSA = (+)-camphorsulfonic acid, TMS = trimethylsilyl.

silyl-tethered substrate, the construction of a quaternary carbon center through a vinyl Claisen rearrangement, and a biomimetic spiroaminal formation. This strategy is an effective approach to use towards other fawcettimine-type *Lycopodium* alkaloids.

Received: May 24, 2011
 Published online: July 12, 2011

Keywords: alkaloids · natural products · Pauson–Khand reaction · rearrangement · total synthesis

- [1] For recent reviews, see: a) “*Lycopodium* Alkaloids: Isolation and Asymmetric Synthesis”: M. Kitajima, H. Takayama in *Topics in Current Chemistry* (Ed.: H.-J. Knölker), Springer, Berlin, **2011**, DOI: 10.1007/128_2011_126; b) Y. Hirasawa, J. Kobayashi, H. Morita, *Heterocycles* **2009**, *77*, 679; c) J. Kobayashi, H. Morita in *The Alkaloids*, Vol. 61 (Ed.: G. A. Cordell), Academic Press, New York, **2005**, p. 1; d) W. A. Ayer, L. S. Trifonov in *The Alkaloids*, Vol. 45 (Eds.: G. A. Cordell, A. Brossi), Academic Press, New York, **1994**, p. 233; e) X. Ma, D. R. Gang, *Nat. Prod. Rep.* **2004**, *21*, 752.
- [2] a) X. C. Tang, Y. F. Han, X. P. Chen, X. D. Zhu, *Acta Pharmacol. Sin.* **1986**, *7*, 507; b) X. C. Tang, P. De Sarno, K. Sugaya, E. Giacobini, *J. Neurosci. Res.* **1989**, *24*, 276.
- [3] a) Y. Hirasawa, J. Kobayashi, Y. Obara, N. Nakahata, N. Kawahara, Y. Goda, H. Morita, *Heterocycles* **2006**, *68*, 2357; b) Y. Hirasawa, H. Morita, J. Kobayashi, *Org. Lett.* **2004**, *6*, 3389; c) J. Kobayashi, Y. Hirasawa, N. Yoshida, H. Morita, *J. Org. Chem.* **2001**, *66*, 5901; d) J. Kobayashi, Y. Hirasawa, N. Yoshida, H. Morita, *Tetrahedron Lett.* **2000**, *41*, 9069; e) K. Ishiuchi, T. Kubota, T. Hoshino, Y. Obara, N. Nakahata, J. Kobayashi, *Bioorg. Med. Chem.* **2006**, *14*, 5995.
- [4] For recent reports on the total synthesis of fawcettimine-type *Lycopodium* alkaloids, see: a) X. Linghu, J. J. Kennedy-Smith, F. D. Toste, *Angew. Chem.* **2007**, *119*, 7815; *Angew. Chem. Int. Ed.* **2007**, *46*, 7671; b) J. A. Kozak, G. R. Dake, *Angew. Chem.*

- 2008, 120, 4289; *Angew. Chem. Int. Ed.* **2008**, 47, 4221; c) A. Nakayama, N. Kogure, M. Kitajima, H. Takayama, *Org. Lett.* **2009**, 11, 5554; d) M. E. Jung, J. J. Chang, *Org. Lett.* **2010**, 12, 2962; e) Y. Otsuka, F. Inagaki, C. Mukai, *J. Org. Chem.* **2010**, 75, 3420; f) S. M. Canham, D. J. France, L. E. Overman, *J. Am. Chem. Soc.* **2010**, 132, 7876; g) J. Ramharter, H. Weinstabl, J. Mulzer, *J. Am. Chem. Soc.* **2010**, 132, 14338; h) X. M. Zhang, Y. Q. Tu, F. M. Zhang, H. Shao, X. Meng, *Angew. Chem.* **2011**, 123, 4002; *Angew. Chem. Int. Ed.* **2011**, 50, 3916.
- [5] C. H. Tan, X. Q. Ma, G. F. Chen, D. Y. Zhu, *Helv. Chim. Acta* **2002**, 85, 1058.
- [6] H. Takayama, K. Katakawa, M. Kitajima, H. Seki, K. Yamaguchi, N. Aimi, *Org. Lett.* **2001**, 3, 4166; H. Takayama, K. Katakawa, M. Kitajima, H. Seki, K. Yamaguchi, N. Aimi, *Org. Lett.* **2002**, 4, 1243.
- [7] C. W. Huh, W. R. Roush, *Org. Lett.* **2008**, 10, 3371.
- [8] a) R. Noyori, S. Hashiguchi, *Acc. Chem. Res.* **1997**, 30, 97; b) K. Matsumura, S. Hashiguchi, T. Ikariya, R. Noyori, *J. Am. Chem. Soc.* **1997**, 119, 8738; c) K. J. Haack, S. Hashiguchi, A. Fujii, T. Ikariya, R. Noyori, *Angew. Chem.* **1997**, 109, 297; *Angew. Chem. Int. Ed. Engl.* **1997**, 36, 285.
- [9] We isolated a portion of the product after the Noyori reduction and determined the enantiomeric excess of the alcohol by HPLC analysis. For details see the Supporting Information.
- [10] B. Li, R. A. Buzon, M. J. Castaldi, *Org. Process Res. Dev.* **2001**, 5, 609.
- [11] a) I. U. Khand, G. R. Knox, P. L. Pauson, W. E. Watts, *Chem. Commun.* **1971**, 36, 36; b) N. E. Schore, M. C. Croudace, *J. Org. Chem.* **1981**, 46, 5436; c) P. Magnus, L. M. Principe, M. J. Slater, *J. Org. Chem.* **1987**, 52, 1483.
- [12] a) E. J. Corey, R. K. Bakshi, S. Shibata, *J. Am. Chem. Soc.* **1987**, 109, 5551; b) K. A. Parker, M. W. Ledebner, *J. Org. Chem.* **1996**, 61, 3214.
- [13] For details see the Supporting Information.
- [14] T. Mandai, M. Ueda, S. Hasegawa, M. Kawada, J. Tsuji, *Tetrahedron Lett.* **1990**, 31, 4041.
- [15] H. C. Brown, R. Liotta, L. Brener, *J. Am. Chem. Soc.* **1977**, 99, 3427.
- [16] a) T. Fukuyama, C. K. Jow, M. Cheung, *Tetrahedron Lett.* **1995**, 36, 6373; b) W. Kurosawa, T. Kan, T. Fukuyama, *Org. Synth.* **2002**, 79, 186; c) T. Kan, T. Fukuyama, *Chem. Commun.* **2004**, 353; d) T. Toma, Y. Kita, T. Fukuyama, *J. Am. Chem. Soc.* **2010**, 132, 10233.
- [17] CCDC 826584 (**20**) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [18] S. Hanessian, D. Delorme, Y. Dufresne, *Tetrahedron Lett.* **1984**, 25, 2515.
- [19] K. Ishihara, H. Kurihara, H. Yamamoto, *J. Org. Chem.* **1993**, 58, 3791.
- [20] For details see the Supporting Information.