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Taxamycin Studies: Synthesis of Taxoid-Calicheamicin Hybrids

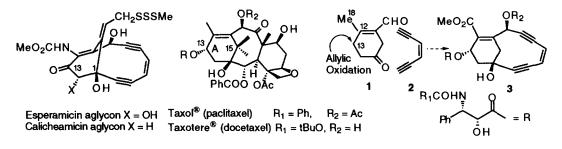
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Abstract: The synthesis of the bicyclic enediyne compounds 14-18 is described. The bicyclo[7.3.1]trideca-4,9-dien-2,6-diynes (21) are calicheamicin-taxoid mimics, that possess the Taxotere[®] side chain and an enediyne core. Cyclization of the iodo-aldehyde 13 [CrCl₂(THF)₂] afforded the bicyclic alcohol 14 as a single diastereomer (76%) and allylic oxidation (SeO₂) followed by reaction with the oxazolidine intermediate 19, resolution and hydrolysis provided the taxamycins 21. © 1998 Elsevier Science Ltd. All rights reserved.

Taxol[®] and Taxotere[®] are important drugs for the treatment of ovarian and breast cancer, while esperamicin¹ and calicheamicin² are key members of the enediyne family of anticancer antibiotics. The novel mechanism of action of these natural products have stimulated the synthesis of simpler less toxic analogues and mimics^{3,4} and delineation of their mode of action by cycloaromatization and tubulin polymerization studies.⁵



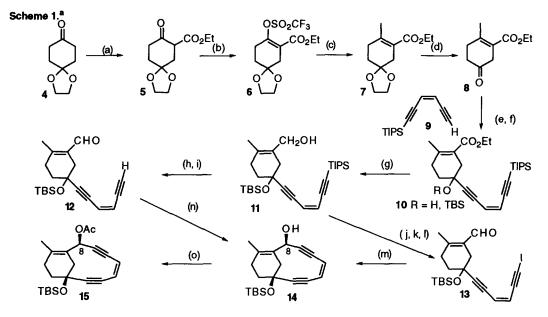
We have designed a new family of potential anti-tumor agents that combine aspects of the important pharmacophores in these families in which suitable functionality should facilitate tubulin binding and cycloaromatization would induce cell damage. For proper evaluation, the synthesis of these taxoid mimics $(taxamycins)^6$ requires the incorporation of a paclitaxel or docetaxel side chain at C₁₃ of the bicyclic enediyne framework. This was not feasible in the methyl substituted 11 and 12 membered ring systems.^{6c} In a continuation of our studies, we wish to report the synthesis of 14-18 which contain the bicyclo[7.3.1]trideca-4,9-dien-2,6-diyne core structure possessed by both esperamicin and calicheamicin and the normethyltaxamycin-10 compounds 21a-d with the docetaxel (Taxotere[®]) side chain attached.

The Taxol[®] side chain has been attached to modified nuclei with retention of respectable tubulin binding.⁷ Other enediyne hybrids incorporate the steroid nucleus (estramycins),⁸ β -lactams (lactendiynes),⁹ and enediynes linked to diethylstilbestrol.¹⁰

Our previous experience suggested that addition of a suitable enediyne building block 2 to a dicarbonyl substituted cyclohexene (1) would lead to the bicyclic system 3 in which oxygen functionality at C_{13} and at the

vinyl methyl group could be introduced by allylic oxidation to facilitate the required functionalization and facilitate cycloaromatization.

As outlined in Scheme 1, the *mono*-ethylene ketal of 1,4-cyclohexanedione (4) was treated with LDA followed by ethyl cyanoformate,¹¹ to generate the β -keto-ester 5 (60%). Exposure of 5 to KHMDS and PhN(SO₂CF₃)₂ gave the corresponding triflate 6 in quantitative yield. The vinyl methyl group was introduced with Me₂CuLi¹² to give the cyclohexene 7 and the ketal removed with 35% aqueous CF₃COOH (3 d) to provide the keto-ester 8 in 80% yield.

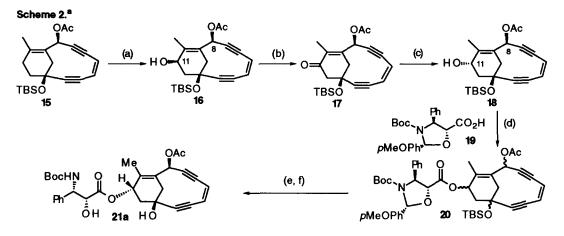


^aReagents: (a) LDA, THF, -78 °C, to -15 °C, 1 h; EtOCOCN, HMPA, -78 °C, 30 min, 60%; (b) KHMDS, THF, -78 °C, 30 min; PhN(SO₂CF₃)₂, -78 °C to 21 °C, 1 h 99%; (c) MeLi, Cul, Et₂O, 0 °C, 15 min, -78 °C to 21 °C, 99%; (d) 35% aq. CF₃CO₂H, CHCl₃, 21 °C, 3 d, 80%; (e) *t*-BuLi, CeCl₃, THF, -78 °C, 10 min, 84%; (f) TBSCI, Et₃N, CH₂Cl₂, 0 °C to 21 °C, 98%; (g) DIBAL-H, CH₂Cl₂, -78 °C to -60 °C, 2 h, 99%; (h) Dess-Martin periodinane, CH₂Cl₂, 21 °C, 1 h, 99%; (f) CSF, Ac₂O, NaHCO₃, CH₃CN, mol. sieves, 4 d, 48%; (j) TBAF, THF, -78 °C to 0 °C, 4 h, 93%; (k) l₂, morpholine, benzene, 60 °C, 2 h, 69%; (l) Dess-Martin periodinane, CH₂Cl₂, 21 °C, 1 h, 98%; (m) [CrCl₂ (THF)₂] (8 equiv.), THF, 30 min, 76%; (n) KHMDS, THF, -78 °C, 60%; (o) Ac₂O, DMAP (cat.), Py, 21 °C, 15 h, 99%.

Condensation of the cerium acetylide of enediyne 9^6 with 8 afforded the corresponding alcohol 10 (R = H) in 84% yield. The alcohol was protected, and the ester function was reduced (*i*-Bu₂AlH,-78 °C) in quantitative yield. Subsequent removal of the triisopropylsilyl group (*n*-Bu₄NF) generated the corresponding acetylene which was converted to the desired iodoacetylene 10 (R = I) with iodine and morpholine. This product was oxidized with Dess-Martin periodinane to provide iodoaldehyde 13 in an overall yield of 62% for the 4 steps.

The intramolecular coupling of the iodoacetylene proceeded readily based on our previous research.⁶ Exposure of 13 to the freshly prepared [CrCl₂(THF)₂] complex¹³ in THF afforded the bicyclic system 14 as a single diastereomer in 76% yield. Reasonable yields (60-70%) were also obtained with commercial anhydrous (toluene azeotrope) CrCl₂. In contrast to the more sterically demanding cyclizations examined earlier,⁶c

catalytic NiCl₂ was not required in these decadiynene systems. This is consistent with related 10-membered ring enediyne cyclizations,¹⁴ although the mechanistic role of Ni(II) in these reactions remains obscure, as many cases require a catalytic amount of NiCl₂.¹⁵ A 60% yield of 14 was also achieved from treatment of the aldehyde 12 with KHMDS at -78 °C, although the anhydrous fluoride-mediated cyclization procedure¹⁶ on the corresponding silylacetylene-aldehyde failed.



^aReagents: (a) SeO₂, dioxane, 70 °C, 30 min, 66%; (b) Dess-Martin periodinane, CH₂Cl₂, 21 °C, 45 min., 99%; (c) NaBH₄, MeOH, CeCl₃, 0 °C, 41%; (d) DCC, **19**, DMAP (cat.), toluene, 21 °C, 30 min, 99%, (1:1); (e) PTSA, MeOH, 21°C, 1.5 h, 94%; (f) TBAF, THF, 21 °C, 30 min, 75%.

Conversion of 14 to its acetate 15 set the stage for oxygen introduction, by treatment of 15 with SeO₂ to provide the allylic alcohol 16 (Scheme 2) as a single diastereomer in a yield of 66%. Consistent with a related example,¹⁷ this oxidation occurred from the top face as a consequence of the steric hindrance of the enediyne bridge. The diastereomeric alcohol 18 was prepared by oxidation of 16 with Dess-Martin periodinane to afford the crystalline cyclohexenone 17. X-ray analysis of 17 confirmed the structures in these series.¹⁸ Reduction of the ketone 17 with sodium borohydride in the presence of CeCl₃ resulted in the preferential formation of alcohol 18 accompanied by its diastereoisomer (60:40 ratio).

Independently the two alcohols 16 and 18 were coupled with the oxazolidine 19^{19} (a protected form of the Taxotere[®] side chain) to provide two separate diastereometric mixtures (1:1) in quantitative yield represented by the heterocyclic ester 20. Acid catalyzed ring opening afforded the four distinct esters that were resolved by chromatography (SiO₂). The individual silyl ether isometric were converted to their C₁ alcohols upon exposure to *n*-tetrabutylammonium fluoride in THF. For clarity only the taxamycin-10 21a, possessing the Taxotere[®] side chain, is illustrated. This compound's stereochemical and electronic features most closely mimic those of the natural products.

Compound **21a** displayed negligible effects on tubulin polymerization and in protecting against Ca^{2+} induced depolymerization compared to $Taxol^{\textcircled{8}}$. Cytotoxic activity with the HT-29 human cancer cell line²⁰ was also weak (70% cell death at 10^{-4} M; $Taxol^{\textcircled{8}}$ 90% at 10^{-9} M). This cytotoxic activity should increase in the presence of a cycloaromatization initiator which removes the bridgehead double bond. This double bond was inert to hydrogenation, a feature consistent with hyperstable alkenes,²¹ and activation by oxidation of the vinyl methyl group failed (SeO₂, CrO₃, PCC). To circumvent this difficulty, potential triggering mechanisms which involve intramolecular reactions with the olefin are under investigation.

In summary, these studies have established a reliable strategy for the synthesis of bicyclic enediynes and examined the biological activity of taxamycin-10 systems. This information will help delineate the structural and functional features required for future investigation.

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References and Notes

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- (1) (a) Golik, J. et al. J. Am. Chem. Soc. 1987, 109, 3461. (b) Golik, J. et al. J. Am. Chem. Soc. 1987, 109, 3462.
- (2) (a) Lee, M. D. et al. J. Am. Chem. Soc. 1987, 109, 3464. (b) Lee, M. D. et al. J. Am. Chem. Soc. 1987, 109, 3466.
- (3) (a) Enediyne Antibiotics as Antitumor Agents, Borders, D.B.; Doyle, T.W., Eds.; Marcel Dekker, Inc.: New York, 1995. (b) Nicolaou, K. C.; Smith, A.L. In Modern Acetylene Chemistry; Stang, P. J.; Diederich, F., Eds.; VCH: Weinheim, 1995; Chapt. 7, pp 203-279. (c) Lhermitte, H.; Grierson, D. S. Contemp. Org. Syn. 1996, 3, 41. (d) Lhermitte, H.; Grierson, D. S. Contemp. Org. Syn. 1996, 5, 43. (e) Danishefsky, S. J.; Shair, M. D. J. Org. Chem. 1996, 61, 16.
- (4) (a) Boa, A. N.; Jenkins, P. R.; Lawrence, N. J. Contemp. Org. Syn. 1994, 47, 47. (b) Nicolaou, K. C.; Dai, W.-M.; Guy, R. K. Angew. Chem., Int. Ed. Engl. 1994, 33, 15. (c) Kingston, D. J. I.; Molinero, A. A.; Rimoldi, J. M. Prog. Chem. Org. Nat. Prod. 1993, 61, 1. (d) Swindell, C. S. Org. Prep. Proced. Int. 1991, 23, 465.
- (5) (a) Lavelle, F.; Guéritte-Voegelein, F.; Guénard, D. Bull. Cancer 1993, 80, 326. (b) Taxane Anticancer Agents (Georg, G. I.; Chen, T. T.; Ojima, I.; Vyas D. M. Amer. Chem. Soc. Washington (1994).
- (6) (a) Lu, Y.-F.; Harwig, C. W.; Fallis, A. G. J. Org. Chem. 1993, 58, 4202. (b) Lu, Y.-F.; Harwig, C. W.; Fallis, A. G. Can. J. Chem. 1995, 73, 2253. (c) Harwig, C. W.; Py, S.; Fallis, A. G. J. Org. Chem. 1997, 62, 7902.
- (7) (a) Blechert, S.; Kleine-Klausing, A. Angew. Chem. Int. Ed. Engl. 1991, 30, 412. (b) Samaranayake, G.; Magri, N. F.; Jitrangsri, C.; Kingston, D. G. I. J. Org. Chem. 1991, 56, 5114.
- (a) Wang, J.; De Clercq, P. J. Angew. Chem. Int. Ed. Engl. 1995, 34, 1749. (b) Wang, J.; De Clercq, P. J. Tetrahedron Lett. 1995, 37, 3395.
- (9) Banfi, L.; Guanti, G. Angew. Chem. Int. Ed. Engl. 1995, 34, 2393.
- (10) Jones, G. B.; Huber, R. S.; Mathews, J. E.; Li, A. Tetrahedron Lett. 1996, 37, 3643.
- (11) Mander, L. N.; Sehti, S. P. Tetrahedron Lett. 1983, 24, 5425.
- (12) (a) McMurry, J. E.; Scott, W. J. Tetrahedron Lett. 1980, 21, 4313. (b) Taber, D.; Amedio, J. C.; Raman, K. J. Org. Chem. 1988, 53, 2984.
- (13) Larkworthy, L. F.; Nelson-Richardson, M. H. O. Chem. Ind. (London) 1974, 164.
- (14) Nishikawa, T.; Shibuya, S.; Hosokawa, S.; Isobe, M. Synlett 1994, 485.
- (15) (a) Okude, Y.; Hirano, S.; Hiyama, T.; Nozaki, H. J. Am. Chem. Soc. 1977, 99, 3179. (b) Hiyama, T.; Kimura, K.; Nozaki, H. Tetrahedron Lett. 1981, 22, 1037. (c) Takai, K.; Kimura, K.; Kuroda, T.; Hiyama, T.; Nozaki, H. Tetrahedron Lett. 1983, 24, 5281. (d) Takai, K.; Kuroda, T.; Nakatsukasa, S.; Oshima, K.; Nozaki, H. Tetrahedron Lett. 1985, 26, 5585. (e) Jin, H.; Uenishi, J.; Christ, W. J.; Kishi, Y. J. Am. Chem. Soc. 1986, 108, 5644. (f) Takai, K.; Yagashira, M.; Kuroda, T.; Oshima, K.; Uchimoto, K.; Nozaki, H. J. Am. Chem. Soc. 1986, 108, 6048. (g) Cintas, P. Synthesis 1992, 248. (h) Hodgson, D.M. J. Organomet. Chem. 1994, 476, 1. (i) Crevisy, C.; Beau, J.-M. Tetrahedron Lett. 1991, 32, 3171. (j) Nicolaou, K. C.; Liu, A.; Zeng, Z.; McComb, S. J. Am. Chem. Soc. 1992, 114, 9279. (k) Maier, M. E.; Brandstetter, T. Tetrahedron Lett. 1992, 33, 7511. (l) Meyers, A. G.; Finney, N. S. J. Am. Chem. Soc. 1992, 114, 10986. (m) Eckhardt, M.; Brückner, R.; Suffert, J. Tetrahedron Lett. 1995, 36, 5167.
- (16) (a) Wender, P. A.; Zercher, C. K.; Beckham, S.; Haubold, E.-M. J. Org. Chem. 1993, 58, 5868. (b) Nishikawa, T.; Shibuya, S.; Isobe, M. Synlett 1994, 482. (c) Wender, P. A.; Beckham, S.; Mohler, D. L. Tetrahedron Lett. 1995, 36, 209.
- (17) (a) Kadow, J. F.; Cook, D. J.; Doyle, T. W.; Langley, D. R.; Pham, K. M.; Vyas, D. M.; Wittman, M. D. Tetrahedron 1989, 30, 3499. (b) Tjepkema, M. W.; Wong, T.; Wilson, P. D.; Fallis, A. G. Can. J. Chem. 1997, 75, 1542.
- (18) X-ray data and figure will be reported in a full paper.
- (19) Didier, E.; Fouque, E.; Taillepied, I.; Commerçon, A. Tetrahedron Lett., 1994, 35, 2349.
- (20) (a) Schiff, P. B.; Fant, J.; Horwitz, S. B. Nature 1979, 277, 665. (b) Skehan, P.; Storeng, R.; Sardeiro, D.; Monks,
- A.; McMahon, J.; Vistica, D.; Warren, J. T.; Bokesch, H.; Kenny, S.; Boyd, M. R. J. Natl. Canc. Inst. 1990, 82, 1107.
- (21) Maier, W.F.; Schleyer, P. v. R. J. Am. Chem. Soc. 1981, 103, 1891.