

pubs.acs.org/OPRD

Large-Scale Synthesis of Eldecalcitol

Hyung Wook Moon, Seung Jong Lee, Seong Hu Park, Se Gyo Jung, In A Jung, Chang Hun Seol, Seung Woo Kim, Seon Mi Lee, Bogonda Gangganna, Seokhwi Park, Kee-Young Lee, Chang-Young Oh, Juyoung Song, Jaehun Jung, Ji Soo Heo, Kang Hee Lee, Hae Sol Kim, Won Taek Lee, Areum Baek, and Hyunik Shin*



ABSTRACT: Industrial-scale synthesis of eldecalcitol is described. AA highly diastereoselective epoxidation of *p*-methoxybenzyl (PMB) protected dienol at room temperature provides the key epoxide intermediate with a secondary hydroxyl group, which is alkylated with a triflate to set up all of the subunits at the C-1, C-2, and C-3 positions of the A-ring fragment. Selective protecting group manipulation followed by palladium-catalyzed cyclization then provides the A-ring synthon. The C/D-ring fragment is obtained by (1) direct C-H hydroxylation of Grundman's ketone using in situ prepared trifluoropropanone dioxirane and (2) protection. Finally, the coupling of the A-ring with the C/D-ring fragment, global deprotection, and recrystallization provide the highly crystalline eldecalcitol.

KEYWORDS: eldecalcitol, C₂-symmetric, C-H hydroxylation, Sharpless epoxidation, triflate alkylation

INTRODUCTION

Vitamin D plays a crucial role as a hormonelike molecule in the control of phosphorous and calcium trafficking. It is uniquely converted into its biologically active form, calcitriol, via photoisomerization and subsequent oxidative hydroxylation.¹ Hinted by its prominent biological activities, a useful derivative that does not interfere with the major functions of endogenous calcitriol has been intensively searched for,² which led to the commercial launch of alfacalcidol, tacalcitol, maxacalcitol, paricalcitol, and eldecalcitol.³

Among them, eldecalcitol (1) has achieved prominent commercial success for the treatment of osteoporosis in Japan and has shown superior activity in lowering the risk of fracture and in increasing the bone mineral density over alfacalcidol.^{4,5} In this context, extensive efforts have been made to devise an efficient synthesis of eldecalcitol.⁶⁻¹⁰ These could be categorized into two main strategies: (1) the late-stage photoisomerization of a fully functionalized 7-dehydrocholesterol derivative (Figure 1, Route 1) and (2) the coupling reaction of the A- and the C/D-ring precursors (Routes 2 and 3). The former strategy was developed by Chugai Pharmaceutical, commencing from lithocholic acid, and is now used in the commercial process.⁸ The final step of this route, however, requires a special photoreactor for its implementation and suffers from low chemical yield with the formation of many side products. This feature necessitates preparative HPLC purification (Route 1). The latter strategy was developed using either a coupling of the A-ring synthon via the Horner-Wittig reaction (Route 2)⁹ or the transition-metal catalyzed cyclization of the A-ring ene-yne fragment with the appropriate C/D-ring-containing synthon (Route 3).9d,10 These routes also suffered from low yields in the final cyclization and coupling reactions, respectively. Herein, we disclose a highly practical

preparation of the A and C/D-ring fragments and their subsequent manipulation toward eldecalcitol.

RESULTS AND DISCUSSION

In general, the efficient stereoselective synthesis of the A-ring synthon represented a formidable challenge. The first attempt in 1993 by Takahashi et al. derived the C-1, C-2, and C-3 chiral centers of the A-ring from the C-2, C-3, and C-4 chirality of D-mannitol.^{9a} Although this approach cleanly installed the three contiguous stereogenic centers with high fidelity, the early stage of this route was not suitable for scale-up due to low yield (30%) at the selective acetonide deprotection step. Another approach by Kubodera et al. in 1997 started from chiral tartaric acid,^{9b} which was transformed into a C_2 symmetric epoxide via a multistep process.¹¹ Although the subsequent regioselective epoxide ring opening resulted in a highly stereocontrolled installation of the C-2 substituent, this route suffered from low stereoselectivity in installing the C-1 stereocenter. Taking these considerations into account, we have devised a highly stereoselective large-scale synthesis of the A-ring fragment by exploiting the C₂-symmetric structural features of D-mannitol (Figure 2).

The synthesis of the A-ring fragment started from the known C₂-symmetric divinyl diol **3**, which was obtained from D-mannitol via three steps (Scheme S1).¹² The diol moiety of **3** was protected as a *p*-methoxybenzyl acetal group to give **4**,

Received: September 29, 2020



Article



Figure 1. Synthetic strategies toward eldecalcitol (1).



Figure 2. Retrosynthesis of A-ring fragment 2.

which was then reduced by DIBAL to afford *p*-methoxybenzy (PMB)-monoprotected intermediate 5. Having 5 in hand, we studied the diastereoselective epoxidation to install the C-3 stereogenic center of 6 (Scheme 1). Initial attempts with mCPBA or VO(acac)₂ as an oxidant gave unsatisfactory diastereomeric ratios and chemical yields (entries 1–3, Table 1). In contrast, diastereoselective Sharpless epoxidation^{11,13} at low temperature gave satisfactory diastereoselectivity, although





^aReagents and conditions: (a) *p*-anisaldehyde, *p*-TsOH, C_6H_{12} , 40 °C, 4 h; (b) DIBAL, toluene, 0 °C, 1 h, 40% (for two steps); (c) Ti(OiPr)₄, (+)-DIPT, cumene hydroperoxide, 3 Å MS, CH₂Cl₂, rt, 5 h, 85%.

Table 1. Epoxidation of 5

Ū.	OPMB <u>conditions</u> CH ₂ Cl ₂	0,, (R) : ŌH	/ ОРМВ ⁺	(S) <u>:</u> OH	OPMB
	-	•	time	ratio	vield
entry	reagents and conditions	temp.	(h)	$(6/6')^a$	$(\%)^b$
1	mCPBA	0 °C to rt	25	46.5:53.5	20
2	$VO(acac)_2$	0 °C	23	64.8:35.2	23
3	$VO(acac)_2$	rt	23	70.3:29.7	30
4	Ti(O <i>i</i> Pr) ₄ , (+)-DIPT, TBHP	−20 °C	94	98.9:1.1	59
5	Ti(O <i>i</i> Pr) ₄ , (+)-DIPT, TBHP	−10 °C	43	98.6:1.4	69
6	Ti(O <i>i</i> Pr) ₄ , (+)-DIPT, TBHP	0 °C	19	98.5:1.5	66
7	Ti(O <i>i</i> Pr) ₄ , (+)-DIPT, TBHP	rt	5	97.9:2.1	85
8	Ti(O <i>i</i> Pr) ₄ , (+)-DIPT, CMHP	rt	5	97.9:2.1	85

"Ratios were determined by ¹H NMR analysis. ^bYields refer to the yields of mixed products.

a prolonged reaction time was required (entries 4–6). Propitiously, increasing the reaction temperature resulted in a significantly faster reaction with an enhanced chemical yield and minimal loss of diastereoselectivity (entry 7, 6/6' = 97.9:2.1). Finally, we replaced TBHP with cumene hydroperoxide (CMHP), which is more viable for large-scale operation,¹⁴ which afforded good results (entry 8, 6/6' = 97.9:2.1).

After establishing the optimized reaction conditions for the preparation of epoxide 6, we recognized that it will be highly efficient if intermediate 6 could be purified without the use of column chromatography. Toward this end, we successfully purified crude intermediates 5 and 6 via thin film distillation under a high vacuum. Considering that 5 and 6 represent key early intermediates toward the synthesis of fragment A, the removal of the need for chromatographic purifications encompassing all chemical steps leading up to 6 provides significant scale-up efficiency.

Scheme 2. Synthesis of 2^a



^{*a*}Reagents and conditions: (a) Et₃N, TBDPSCl, CH₂Cl₂, rt, 15 h, 90%; (b) (TfO)₂O, DIPEA, *n*-heptane, 10 °C, 30 min, 95%; (c) **6**, NaH, CH₃CN, 0 °C, 10 min, then rt, 30 min, 85%; (d) **10**, *n*-BuLi, BF₃·OEt₂, THF, -65 °C, 1 h, then rt, 1 h; (e) TBSCl, imidazole, DMF, 60 °C, 15 h; (f) DDQ, CH₂Cl₂, H₂O, rt, 2 h, 56% (for three steps); (g) TBSCl, imidazole, DMF, 60 °C, 6 h, 90%; (h) Me₂AlCl, CH₂Cl₂, rt, 5 h, 85%; (i): (i) Red-Al, Et₂O, rt, 4 h; (ii) I₂, THF, -65 °C, 30 min, then rt, 1 h, 70%; (j) Ph(PPh₃)₄, Et₃N, CH₃CN, reflux, 1 h, 90%; (k) NCS, Me₂S, CH₂Cl₂, -20 °C, 20 min, then rt, 30 min, 90%; (l): (i) HPPh₂, *n*-BuLi, THF, -60 °C, 1 h; (ii) 5% H₂O₂, CHCl₃, rt, 10 min, 80%.

Next, a Williamson ether reaction of secondary alcohol 6 was examined (Scheme 2). Early attempts at O-alkylation with analogs of 8, which contained halide-leaving groups (-Br and -I), resulted in complex mixtures under various conditions. We then tested triflate 8, which was synthesized from 1,3propanediol in two steps. Initial attempts at alkylation using alkoxide bases or amine bases such as DIPEA did not afford the desired product. The use of NaH in tetrahydrofuran (THF), however, provided 9 in 60% yield. By careful TLC analysis, we found that triflate 8 was unstable in solvents such as THF, toluene, DME, and DCM, while good solution stability was observed in acetonitrile. With these observations in mind, coupling of 6 with 8 using NaH as a base in acetonitrile furnished 9 in 85% yield. It is important to note that excess triflate 8 should be quenched completely at the end of the reaction by the addition of MeOH. If not, residual triflate 8 slowly decomposes to generate triflic acid, which catalyzes the decomposition of the desired product 9, especially when kept as a crude concentrate.

The reaction of THP-protected propargyl alcohol 10 with 9 in the presence of boron trifluoride etherate afforded 11. The attempted deprotection of the PMB group using ceric ammonium nitrate (CAN) resulted in a cyclic side product instead of the desired diol. Due to this issue, a sequential *tert*butyldimethylsilyl (TBS) protection of 11 to 12 and deprotection of the PMB group to 13 followed by TBS protection to 14 was effected.

Next, the required orthogonal deprotection of the THP group of 14 to yield 15 proved to be nontrivial. The well-known conditions using $PPTS^{15}$ were complicated by the concomitant partial deprotection of the silyl groups (entry 1, Table 2). The use of LiCl¹⁶ at a high temperature afforded the

Table 2. THP Deprotection of 14						
TBSO``	OTHP OTBS OTBDPS 14	TBSO	OH OTBS	OTBDPS		
entry	reagents and conditions	temp.	time (h)	yield (%)		
1	PPTS (0.1 equiv), acetone, H_2O	0 °C to rt	24	5		
2	LiCl (5.0 equiv), DMSO, H ₂ O	90 °C	8	50		
3	NaCNBH ₃ (1.5 equiv), BF ₃ ·OEt ₂ (1.0 equiv), THF	rt	48	94		
4	$BF_3 \cdot OEt_2$ (2.0 equiv), SiEt ₃ H (5.0 equiv), THF	rt	24	35		
5	MgBr ₂ (3.0 equiv), THF	rt	7	0		
6	MgBr ₂ (7.0 equiv), Et ₂ O	rt	24	90		
7	MgBr ₂ (3.0 equiv), CH ₂ Cl ₂	rt	3	30		
8	Me ₂ AlCl (2.0 equiv), CH ₂ Cl ₂	0 °C to rt	5	85		

desired product in a moderate yield (entry 2). The use of NaCNBH₃/BF₃·OEt₂¹⁷ led to a high yield, although a prolonged reaction time was required (entry 3). The reaction using BF₃·OEt₂/SiEt₃H provided a low yield (entry 4). Interestingly, the use of MgBr₂¹⁸ as a Lewis acid showed a significant solvent effect: no reaction was observed in THF (entry 5); a high yield was achieved in Et₂O after a prolonged reaction time (entry 6); and a moderate yield was observed in CH₂Cl₂ (entry 7). Finally, the use of Me₂AlCl¹⁹ (entry 8) led

Organic Process Research & Development

to a good yield in a short reaction time at ambient temperature and was selected as the final method for THP deprotection.

The partial reduction of the triple bond using Red-Al and subsequent iodination reaction conducted in the same pot gave vinyl iodide 16. Its Pd(0)-catalyzed intramolecular Heck reaction afforded 6-exo cyclized (*Z*)-diene 17. Chlorination of the allylic alcohol was achieved using the Corey–Kim reagent, and subsequent reaction with lithium diphenylphosphide furnished the required phosphine oxide 2 in a good overall yield.

The C/D-ring fragment 21^{20} was prepared via the direct hydroxylation of Grundman's ketone 19, which was readily obtained from the ozonolysis of vitamin D₃ (Scheme 3).

Scheme 3. Synthesis of 21^a



^aReagents and conditions: (a) O₃, NaHCO₃, CH₂Cl₂, MeOH, -65 °C, 4 h, 85%; (b) CF₃COMe, Oxone, NaHCO₃, CH₃CN, H₂O, 0 °C, 24 h, 53%; (c) TMS-imidazole, CH₂Cl₂, rt, 1.5 h, 78%.

Although the use of isolated dioxirane in this reaction provides high yield, we deployed in situ generated methyl-(trifluoromethyl)dioxirane (TFDO), which was derived from Oxone and trifluoroacetone to mitigate the potential safety issues arising from this conversion.²¹

Finally, a Horner–Wittig reaction was required to unite the A-ring fragment 2 and the C/D-ring fragment 21. The known conditions using *n*-BuLi were first attempted, which led to olefin 22 in a low to moderate yield (Scheme 4).^{9e,f} Critically, the use of an excess amount of 2 (2.0–3.0 equiv) led to higher yields (Table 3, entries 4 and 5). With fully functionalized 22 in hand, the global deprotection of the silyl groups using *p*-

POPh₂



Table	3.	Optimi	zation	of Cou	pling	Reaction	between	2	and
21									

entry	2 (equiv)	n-BuLi (equiv)	21 (equiv)	22 yield (%)		
1	1.0	1.0	1.0	>10		
2	1.5	1.5	1.0	50		
3	1.0	1.0	2.0	>10		
4	2.0	2.0	1.0	89 ^a		
5	3.0	3.0	1.0	95		
^a 0.7 equiv of 2 was recovered.						

TsOH in MeOH/CH₂Cl₂ afforded eldecalcitol (1), which was recrystallized in acetonitrile–water to provide the highly pure and crystalline product. Our synthetic sample was in good spectroscopic agreement with the reported eldecalcitol.^{7a,9f,g} Its X-ray single-crystal structure is depicted in Figure 3.

CONCLUSIONS

In summary, a large-scale synthetic route toward eldecalcitol (1) was devised. Our synthesis features no chromatographic purification for the preparation of the key early intermediate 6, which is accessed from a highly efficient Sharpless epoxidation conducted at ambient temperature. The selective deprotection of the THP group of 14 is worthy of recognition for its efficiency. Finally, the coupling of A-ring phosphine oxide 2 with C/D-ring ketone 21 has been accomplished in high yield, and the final recrystallization of 1 afforded high-quality eldecalcitol as a crystalline solid.

EXPERIMENTAL SECTION

General Methods. Optical rotations were measured with a JASCO P-2000 polarimeter in ethanol solvent. ¹H and ¹³C nuclear magnetic resonance (NMR) spectroscopic data were recorded with a Fourier transform NMR (FT-NMR) spectrometer at 300 or 75 MHz. Chemical shift values are reported in parts per million (ppm) relative to TMS or CDCl₃ as the internal standard, and coupling constants are reported in hertz. Infrared (IR) spectra were measured with a Fourier transform IR (FT-IR) spectrometer. Mass spectroscopic data were obtained with a Jeol JMS 700 high-resolution mass spectrometer equipped with a magnetic-sector-electric-sector double-focusing analyzer. Flash chromatography was performed using the mixtures of ethyl acetate and hexane as eluents. Unless otherwise stated, all of the nonaqueous reactions were carried out under an argon atmosphere with commercial-grade reagents and solvents. THF was distilled



^aReagents and conditions: (a) *n*-BuLi, THF, -65 °C, 1 h, then -10 °C, 2 h, 89%; (b) *p*-TsOH·H₂O, CH₂Cl₂, MeOH, rt, 5 h, 80%.

pubs.acs.org/OPRD



Figure 3. X-ray single-crystal structure of eldecalcitol (1).

from sodium and benzophenone. Dichloromethane was distilled from calcium hydride.

(4R,5R)-2-(4-Methoxybenzyl)-4,5-divinyl-1,3-dioxolane (4). To a solution of 3 (3.4 kg, 30 mol) in cyclohexane (34 L) were added p-anisaldehyde (4.7 L, 39 mol) and p-TsOH (4.3 kg, 22 mol) at rt. The reaction mixture was stirred at 40 °C for 4 h, cooled to rt, and quenched with aqueous 5% NaHCO₃ solution (56 L). The organic layer was separated and washed with aqueous 10% NaHSO₃ solution (133 L) for the removal of unreacted *p*-anisaldehyde. The organic layer was separated, washed with aqueous 3% NaHCO₃ solution (35 L) and water (35 L), dried over Na₂SO₄, and concentrated in vacuo. Crude oil 4 (6.9 kg, 30 mol) was used in the next step without further purification. ¹H NMR (300 MHz, CDCl₃) δ 7.44 (2H, dd, J = 6.6, 1.8 Hz), 6.91 (2H, dd, J = 6.7, 1.9 Hz), 6.00 (1H, s), 5.94-5.85 (2H, m), 5.45-5.35 (2H, m), 5.33-5.24 (2H, m), 4.37-4.19 (2H, m), 3.81 (3H, s); ¹³C NMR (75 MHz, CDCl₃) & 160.6, 134.4, 133.7, 130.3, 128.1, 119.4, 118.6, 113.9, 103.9, 84.0, 82.3, 55.5.

(3R,4R)-4-((4-Methoxybenzyl)oxy)hexa-1,5-dien-3-ol (5). To a solution of crude 4 (6.9 kg, 30 mol) in toluene (35 L) was added 1.2 M DIBAL in toluene (30 L, 36 mol) at 0 °C. The reaction mixture was stirred at 0 °C for 1 h. MeOH (1 L) was added to the reaction mixture. The reaction mixture was cooled to 10 °C, quenched with aqueous 5% NaOH solution (48 L), and stirred for 10 min. The organic layer was separated, dried over Na₂SO₄, and concentrated in vacuo. Thin film distillation of the crude mixture at 120 °C under 0.2 Torr gave product 5 (2.8 kg, 12 mol, 40% for two steps) as colorless oil. ¹H NMR (300 MHz, CDCl₃) δ 7.25 (2H, d, I = 8.7 Hz), 6.89 (2H, dd, J = 6.7, 1.9 Hz), 5.86–5.67 (2H, m), 5.99–5.35 (2H, m), 5.32–5.28 (1H, m), 5.20 (1H, dt, J = 10.5, 1.5 Hz), 4.59 (1H, d, J = 11.1 Hz), 4.31 (1H, d, J = 11.1 Hz), 4.07– 4.01 (1H, m), 3.80 (3H, s), 3.67 (1H, t, J = 7.5 Hz), 2.76 (1H, d, I = 3.0 Hz); ¹³C NMR (75 MHz, CDCl₃) δ 159.5, 136.3, 134.9, 130.1, 129.7, 120.3, 117.1, 114.0, 83.8, 74.8, 70.4, 55.4.

(15,2R)-2-((4-Methoxybenzyl)oxy)-1-((R)-oxiran-2-yl)but-3-en-1-ol (6). To a solution of activated 3 Å molecular sieves (0.7 kg) in CH₂Cl₂ (8.5 L) were added Ti(OiPr)₄ (2.12 L, 7.16 mol) and (+)-diisopropyl tartrate (1.76 L, 8.35 mol) at rt. The reaction mixture was stirred at rt for 20 min. A dilute solution of compound 5 (1.40 kg, 5.98 mol) in CH₂Cl₂ (5.5 L) was added to the reaction mixture. The reaction mixture was stirred at rt for 20 min and cooled to 15 °C. Cumene hydroperoxide (80%, 1.14 L, 6.14 mol) was added dropwise to the reaction mixture using a dropping funnel. The reaction mixture was stirred at rt for 5 h and filtered through a pad of Celite. A solution of FeSO4·7H2O (1.99 kg) and citric acid (0.69 kg) in water (28 L) was added to the resulting solution. The organic layer was separated, filtered through a pad of Celite, and washed with a solution of NaOH (1.0 kg) in aqueous 20% NaCl solution (13 L). The organic layer was separated, washed with 1 N HCl solution (4.3 L), aqueous 4.6% NaHCO₃ solution (13 L), and water (13 L), dried over Na₂SO₄, and concentrated in vacuo. Thin-film distillation of the crude mixture at 80 °C under 0.2 Torr and further purification by thin film distillation at 140 °C under 0.2 Torr gave product 6 (1.27 kg, 5.07 mol, 85%) as yellow oil. ¹H NMR (300 MHz, CDCl₃) δ 7.25 (2H, d, J = 8.4 Hz), 6.89 (2H, d, J = 8.7 Hz), 5.92-5.80 (1H, m), 5.42 (1H, s), 5.40-5.35 (1H, m), 4.62 (1H, d, J = 11.4 Hz), 4.35 (1H, d, J = 11.4 Hz), 3.94-3.87 (1H, m), 3.81 (3H, s), 3.48 (1H, t, J = 5.1Hz), 3.06–3.01 (1H, m), 2.78 (2H, d, J = 3.3 Hz), 2.46 (1H, brs); ¹³C NMR (75 MHz, CDCl₃) δ 159.5, 134.7, 130.0, 129.7, 120.1, 114.0, 81.4, 73.3, 70.3, 55.4, 51.9, 45.0.

(3R,4S)-1-(4-Methoxyphenyl)-11,11-dimethyl-4-((R)-oxiran-2-yl)-10,10-diphenyl-3-vinyl-2,5,9-trioxa-10-siladodecane (9). To a solution of 60% dispersion of NaH in mineral oil (0.28 kg, 7.0 mol) in CH₃CN (8.5 L) was added a dilute solution of compound 6 (1.27 kg, 5.07 mol) in CH₃CN (3.5 L) slowly at -5 °C. To this reaction mixture was added dropwise a solution of triflate 8 (3.40 kg, 7.60 mol) in CH_3CN (3 L) using a dropping funnel. The reaction mixture was stirred at 0 °C for 10 min, stirred at rt for 30 min, and quenched with MeOH (0.41 L, 10.1 mol). To the resulting solution were added water (12.5 L) and AcOEt (12.5 L) and stirred for 10 min. The organic layer was separated, washed with water (12.5 L), dried over Na_2SO_4 , and concentrated in vacuo. Purification by silica gel column chromatography (hexanes/AcOEt = 7:1) gave product 9 (2.36 kg, 4.32 mol, 85%) as yellow oil. ¹H NMR (300 MHz, CDCl₃) δ 7.67–7.61 (4H, m), 7.44-7.33 (6H, m), 7.25-7.22 (2H, m), 6.86-6.83 (2H, m), 5.91-5.81 (1H, m), 5.31-5.24 (2H, m), 4.60 (1H, d, J = 12.0 Hz), 4.34 (1H, d, J = 11.7 Hz), 3.92 (1H, q, J = 3.9Hz), 3.78 (3H, s), 3.76-3.68 (3H, m), 3.64-3.56 (1H, m), 3.22 (1H, t, J = 4.5 Hz), 3.10-3.06 (1H, m), 2.76-2.72 (2H, m)m), 1.83-1.74 (2H, m), 1.03 (9H, s); ¹³C NMR (75 MHz, $CDCl_3$) δ 159.3, 135.7, 135.3, 134.9, 134.1, 130.5, 129.8,

Organic Process Research & Development

129.7, 129.5, 127.9, 127.8, 118.8, 113.8, 81.0, 80.5, 70.4, 68.8, 60.9, 55.4, 51.2, 45.5, 33.2, 27.0, 26.7, 19.4.

(3R,4R,5R)-4-(3-((tert-Butyldiphenylsilyl)oxy)propoxy)-3-((4-methoxybenzyl)oxy)-9-((tetrahydro-2H-pyran-2-yl)oxy)non-1-en-7-yn-5-ol (11). To a solution of THP-protected propargyl alcohol 10 (1.81 kg, 12.9 mol) in THF (16.5 L) was added 2.5 M n-BuLi in n-hexane (4.83 L, 12.1 mol) dropwise at -65 °C. The reaction mixture was stirred at -65 °C for 1 h. To the reaction mixture were added dropwise a solution of compound 9 (2.36 kg, 4.32 mol) in THF (7 L) and BF₃·OEt₂ (0.59 L, 4.75 mol) subsequently. The reaction mixture was warmed to rt, stirred at rt for 1 h, and quenched with aqueous 8.9% NH₄Cl solution (11.5 L). The organic layer was extracted with AcOEt (23 L), dried over Na₂SO₄, and concentrated in vacuo. The crude oil of 11 (2.96 kg, 4.31 mol) was used in the next step without further purification. ¹H NMR (300 MHz, CDCl₃) δ 7.66–7.63 (4H, m), 7.45–7.34 (6H, m), 7.25–7.21 (2H, m), 6.86 (2H, dt, I = 9.2, 2.5 Hz), 5.98–5.86 (1H, m), 5.35–5.29 (2H, m), 4.79 (1H, t, J = 3.3 Hz), 4.59 (1H, d, J = 11.7 Hz), 4.30 (1H, d, J = 11.7 Hz), 4.27–4.23 (2H, m), 4.11-4.07 (1H, m), 3.87-3.82 (2H, m), 3.79 (3H, s), 3.75-3.65 (4H, m), 3.55–3.46 (1H, m), 3.37–3.34 (1H, m), 2.99 (1H, d, J = 4.8 Hz), 2.56-2.40 (2H, m), 1.82-1.49 (8H, m),1.04 (9H, s); ¹³C NMR (75 MHz, CDCl₃) δ 159.5, 135.7, 135.0, 134.0, 129.9, 129.8, 129.7, 127.8, 119.1, 114.0, 96.9, 83.2, 82.1, 79.5, 78.2, 70.4, 69.6, 68.9, 62.2, 60.8, 55.4, 54.8, 33.2, 30.5, 27.0, 25.5, 24.0, 19.3, 14.3.

(5R,6S)-6-((R)-1-((4-Methoxybenzyl)oxy)allyl)-2,2,3,3,13,13-hexamethyl-12,12-diphenyl-5-(4-((tetrahydro-2H-pyran-2-yl)oxy)but-2-yn-1-yl)-4,7,11-trioxa-3,12-disilatetradecane (12). To a solution of crude compound 11 (2.96 kg, 4.31 mol) in DMF (30 L) were added imidazole (0.73 kg, 10.72 mol) and TBSCl (1.49 kg, 9.89 mol) subsequently at rt. The reaction mixture was stirred at 60 °C for 15 h, cooled to rt, and quenched with water (3 L) and aqueous 5% NH₄Cl solution (23 L). The organic layer was separated with AcOEt (30 L), washed with water (30 L), dried over Na₂SO₄, and concentrated in vacuo. The crude oil of **12** (3.45 kg, 4.31 mol) was used in the next step without further purification. ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3) \delta 7.68 - 7.64 (4\text{H}, \text{m}), 7.43 - 7.33 (6\text{H}, \text{m}),$ 7.23-7.20 (2H, m), 6.82 (2H, dt, J = 9.2, 2.5 Hz), 5.93-5.81 (1H, m), 5.31–5.28 (1H, m), 5.25 (1H, s), 4.84–4.76 (1H, m), 4.49 (1H, d, J = 11.4 Hz), 4.30–4.21 (3H, m), 3.97–3.74 (10H, m), 3.53–3.47 (1H, m), 3.38–3.34 (1H, m), 2.51–2.49 (2H, m), 1.89–1.48 (8H, m), 1.04 (9H, s), 0.86 (9H, s), 0.07 (3H, s), 0.01 (3H, s); ¹³C NMR (75 MHz, CDCl₃) δ 159.1, 136.5, 135.7, 134.2, 130.9, 129.6, 129.5, 127.7, 118.5, 113.8, 96.8, 96.7, 85.6, 84.7, 84.6, 80.8, 71.6, 70.4, 70.2, 62.1, 61.2, 60.5, 55.4, 54.8, 54.7, 33.5, 30.5, 27.0, 26.0, 25.8, 25.6, 23.6, 21.2, 19.3, 18.2, 14.3, -4.3, -4.5.

(3R,4S,5R)-5-((tert-Butyldimethylsilyl)oxy)-4-(3-((tertbutyldiphenylsilyl)oxy)propoxy)-9-((tetrahydro-2H-pyran-2yl)oxy)non-1-en-7-yn-3-ol (13). To a solution of crude compound 12 (3.45 kg, 4.31 mol) in CH₂Cl₂ (35 L) were added water (25 L) and DDQ (1.17 kg, 5.15 mol) at rt. The reaction mixture was stirred at rt for 2 h, quenched with aqueous 0.75% NaOH solution (35 L), and stirred at rt for 10 min. The organic layer was separated, and aqueous 10% NaHSO₃ solution (38 L) was added and stirred at rt for 10 min. The organic layer was separated, dried over Na₂SO₄, and concentrated in vacuo. Purification by silica gel chromatography (hexanes/AcOEt = 9:1) gave product 13 (1.64 kg, 2.41 mol, 56% for three steps) as colorless oil. ¹H NMR (300 MHz, CDCl₃) δ 7.67–7.64 (4H, m), 7.45–7.35 (6H, m), 5.99–5.88 (1H, m), 5.35 (1H, dt, *J* = 17.2, 1.7 Hz), 5.18 (1H, dt, *J* = 10.5, 1.5 Hz), 4.80 (1H, t, *J* = 3.1 Hz), 4.31–4.22 (3H, m), 3.93–3.79 (3H, m), 3.75–3.64 (3H, m), 3.54–3.47 (1H, m), 3.33 (1H, dd, *J* = 5.7, 3.9 Hz), 2.66 (1H, d, *J* = 6.3 Hz), 2.58–2.43 (2H, m), 1.87–1.65 (4H, m), 1.61–1.50 (4H, m), 1.04 (9H, s), 0.90 (9H, s), 0.11 (3H, s), 0.09 (3H, s); ¹³C NMR (75 MHz, CDCl₃) δ 138.7, 135.7, 134.0, 129.7, 127.8, 116.2, 98.7, 83.8, 83.4, 83.3, 78.2, 71.3, 70.9, 69.6, 62.2, 62.1, 60.9, 60.5, 54.6, 33.3, 30.5, 27.0, 25.9, 25.5, 24.1, 21.2, 19.3, 18.2, 14.3, -4.4, -4.6.

(5R,6R)-6-((1R)-1-((tert-Butyldimethylsilyl)oxy)-5-((tetrahydro-2H-pyran-2-yl)oxy)pent-3-yn-1-yl)-2,2,3,3,13,13-hexamethyl-12,12-diphenyl-5-vinyl-4,7,11-trioxa-3,12-disilatetradecane (14). To a solution of compound 13 (1.64 kg, 2.41 mol) in DMF (16.5 L) were added imidazole (0.263 kg, 3.86 mol) and TBSCl (0.546 kg, 3.62 mol) subsequently at rt. The reaction mixture was stirred at 60 °C for 6 h, cooled to rt, and quenched with water (1.6 L) and aqueous 5% NH₄Cl solution (13 L). The organic layer was separated with AcOEt (16.5 L), washed with water (16.5 L), dried over Na₂SO₄, and concentrated in vacuo. Purification by silica gel chromatography (hexanes/AcOEt = 10: 1) gave product 14 (1.65 kg, 2.17 mol, 90%) as colorless oil. ¹H NMR (300 MHz, CDCl₃) δ 7.68-7.64 (4H, m), 7.44-7.34 (6H, m), 5.93-5.81 (1H, m), 5.24 (1H, dt, J = 17.4, 1.5 Hz), 5.14–5.10 (1H, m), 4.79–4.78 (1H, m), 4.30–4.25 (1H, m), 4.20–4.13 (2H, m), 3.98–3.94 (1H, m), 3.86-3.67 (5H, m), 3.54-3.47 (1H, m), 3.26 (1H, dd, I = 6.6, 1.5 Hz), 2.51–2.33 (2H, m), 1.86–1.47 (8H, m), 1.04 (9H, s), 0.89 (9H, s), 0.87 (9H, s), 0.08 (3H, s), 0.06 (3H, s), 0.04 (3H, s), 0.02 (3H, s); ¹³C NMR (75 MHz, CDCl₃) δ 138.5, 135.7, 134.2, 129.6, 127.7, 116.0, 98.9, 96.8, 74.4, 72.2, 69.6, 62.1, 61.3, 60.5, 54.9, 54.8, 33.5, 30.4, 27.0, 26.1, 26.0, 25.5, 23.6, 21.2, 19.3, 18.4, 18.2, 14.4, -4.3, -4.4, -4.5.

(5R,6R,7R)-5,7-Bis((tert-butyldimethylsilyl)oxy)-6-(3-((tertbutyldiphenylsilyl)oxy)propoxy)non-8-en-2-yn-1-ol (15). To a solution of compound 14 (1.65 kg, 2.17 mol) in $\rm CH_2\rm Cl_2$ (16.5 L) was added 0.9 M Me₂AlCl in heptane (4.83 L, 4.34 mol) dropwise at 10 °C. The reaction mixture was stirred at rt for 5 h, cooled to 0 °C, and quenched with MeOH (132 mL, 3.25 mol). To the resulting solution was added aqueous 3.6% NaOH solution (16.5 L) and stirred for 10 min. The organic layer was separated, dried over Na₂SO₄, and concentrated in vacuo. Purification by silica gel column chromatography (hexanes/AcOEt = 9:1) gave product 15 (1.42 kg, 1.84 mol, 85%) as colorless oil. ¹H NMR (300 MHz, CDCl₂) δ 7.68– 7.64 (4H, m), 7.44-7.34 (6H, m), 5.93-5.81 (1H, m), 5.24 (1H, dt, J = 17.3, 1.6 Hz), 5.13 (1H, dt, J = 10.5, 1.5 Hz),4.29-4.21 (2H, m), 4.18-4.13 (1H, m), 3.98-3.93 (1H, m), 3.87–3.67 (4H, m), 3.27 (1H, dd, J = 6.6, 1.8 Hz), 2.49–2.34 (2H, m), 1.87-1.78 (2H, m), 1.37 (1H, t, J = 6.0 Hz), 1.04 (9H, s), 0.89 (9H, s), 0.88 (9H, s), 0.09 (3H, s), 0.06 (3H, s), 0.05 (3H, s), 0.02 (3H, s); 13 C NMR (75 MHz, CDCl₃) δ 138.5, 135.7, 134.2, 129.7, 127.8, 116.0, 87.0, 85.4, 79.5, 74.4, 72.1, 69.7, 61.0, 51.7, 33.5, 27.0, 26.1, 26.0, 23.5, 19.4, 18.4, 18.2, -4.3, -4.4, -4.5.

(5R,6R,7R, Z)-5,7-Bis((tert-butyldimethylsilyl)oxy)-6-(3-((tert-butyldiphenylsilyl)oxy)propoxy)-3-iodonona-2,8-dien-1-ol (16). To a solution of compound 15 (1.42 kg, 1.84 mol) in Et₂O (18.5 L) was added 60% Red-Al solution in toluene (7.39 L, 2.65 mol) dropwise at 0 °C. The reaction mixture was stirred at rt for 4 h and cooled to 0 °C, and AcOEt (0.2 L) was added at 0 °C and cooled to -65 °C. A solution of I₂ (0.937 kg, 3.69 mol) in THF (2.5 L) was added to the reaction mixture at -65 °C. The reaction mixture was stirred at -65 °C for 30 min, stirred at rt for 1 h, quenched with aqueous 10% NH₄Cl solution (4.2 L), and stirred for 10 min. The organic layer was separated, washed with aqueous 10% Na₂S₂O₃ solution (11.7 L), dried over Na₂SO₄, and concentrated in vacuo. Purification by silica gel column chromatography (hexanes/AcOEt = 9:1) gave product 16 (1.08 kg, 1.29 mol, 70%) as a colorless liquid. ¹H NMR (300 MHz, CDCl₃) δ 7.69-7.62 (4H, m), 7.45-7.34 (6H, m), 5.93-5.76 (2H, m), 5.31 (1H, dt, I = 17.3, 1.5 Hz), 5.18 (1H, dt, I = 10.5, 1.4 Hz), 4.20-4.01 (4H, m), 3.89-3.69 (4H, m), 3.28 (1H, dd, J = 7.5, 0.9 Hz), 2.74-2.67 (1H, m), 2.61-2.56 (1H, m), 1.89-1.80 (2H, m), 1.39 (1H, t, J = 6.0 Hz), 1.05 (9H, s), 0.90 (9H, s), 0.84 (9H, s), 0.09 (3H, s), 0.06 (3H, s), 0.05 (3H, s), 0.02 (3H, s); ¹³C NMR (75 MHz, CDCl₃) δ 138.0, 136.5, 135.7, 134.2, 134.1, 129.7, 127.8, 116.8, 107.9, 87.5, 74.6, 71.2, 69.9, 67.5, 61.5, 60.5, 48.7, 33.6, 27.0, 26.1, 21.2, 19.4, 18.4, 18.1, 14.4, -3.8, -3.9, -4.4, -4.5.

(Z)-2-((3R,4R,5R)-3,5-Bis((tert-butyldimethylsilyl)oxy)-4-(3-((tert-butyldiphenylsilyl)oxy)propoxy)-2methylenecyclohexylidene)ethan-1-ol (17). A solution of compound 16 (1.08 kg, 1.29 mol) in CH₃CN (32.5 L) was degassed under Ar gas for 30 min. To the reaction mixture were added Pd(PPh₃)₄ (90 g, 78 mmol) and Et₃N (19.5 mL, 1.39 mol). The reaction mixture was refluxed at 81 °C for 1 h. The reaction mixture was cooled to rt and concentrated in vacuo. Purification by silica gel column chromatography (hexanes/AcOEt = 9:1) gave product 17 (0.83 kg, 1.16 mol, 90%) as yellow oil. ¹H NMR (300 MHz, CDCl₃) δ 7.67–7.64 (4H, m), 7.44–7.33 (6H, m), 5.52 (1H, t, J = 6.7 Hz), 5.23 (1H, s), 4.84-4.82 (1H, m), 4.25-4.09 (4H, m), 3.80-3.62 (4H, m), 3.18 (1H, dd, J = 6.9, 2.1 Hz), 2.43-2.36 (1H, m), 2.26-2.17 (1H, m), 1.88-1.79 (2H, m), 1.20-1.00 (1H, m), 1.04 (9H, s), 0.89 (9H, s), 0.86 (9H, s), 0.06 (3H, s), 0.04-0.03 (9H, m); 13 C NMR (75 MHz, CDCl₃) δ 145.4, 139.2, 135.7, 134.1, 129.7, 127.7, 126.8, 84.5, 74.2, 69.1, 68.4, 61.4, 60.5, 60.0, 41.3, 33.5, 27.0, 25.9, 21.2, 19.4, 18.4, 18.3, 14.4, -4.5, -4.6, -4.7.

(((1R,2R,3R,Z)-2-(3-((tert-Butyldiphenylsilyl)oxy)propoxy)-5-(2-chloroethylidene)-4-methylenecyclohexane-1,3-diyl)bis(oxy))bis(tert-butyldimethylsilane) (18). A solution of NCS (0.318 kg, 2.38 mol) in CH_2Cl_2 (7.5 L) was cooled to 0 °C. To the reaction mixture was added Me₂S (0.19 L, 2.55 mol) at 0 °C. The reaction mixture was stirred at 0 °C for 30 min and cooled to -20 °C. A solution of compound 17 (0.83 kg, 1.16 mol) in CH_2Cl_2 (5 L) was added dropwise to the reaction mixture. The reaction mixture was stirred at -20 °C for 20 min, stirred at rt for 30 min, and quenched with water (7.5 L). The organic layer was separated, dried over Na_2SO_4 , and concentrated in vacuo. Purification by silica gel column chromatography (hexanes/AcOEt/Et₃N = 10:1:0.1) gave product 18 (0.76 kg, 1.05 mol, 90%) as yellow oil. ¹H NMR (300 MHz, CDCl₃) δ 7.67–7.63 (4H, m), 7.44–7.34 (6H, m), 5.52 (1H, t, J = 7.8 Hz), 5.29 (1H, s), 5.04–5.03 (1H, m), 4.25 (1H, d, J = 7.2 Hz), 4.19-4.09 (3H, m), 3.79-3.62 (4H, m),3.17–3.15 (1H, m), 2.39 (1H, dd, J = 13.5, 6.9 Hz), 2.23–2.17 (1H, m), 1.88-1.79 (2H, m), 1.04 (9H, s), 0.89 (9H, s), 0.85 (9H, s), 0.06 (3H, s), 0.04 (3H, s), 0.03 (3H, s), 0.02 (3H, s); ^{13}C NMR (75 MHz, CDCl₃) δ 141.5, 135.7, 134.1, 129.7, 127.8, 123.1, 69.0, 68.5, 61.4, 44.9, 41.4, 33.5, 27.0, 26.0, 25.9, 19.4, 18.4, 18.3, -4.5, -4.7.

((Z)-2-((3R,4R,5R)-3,5-Bis((tert-butyldimethylsilyl)oxy)-4-(3-((tert-butyldiphenylsilyl)oxy)propoxy)-2methylenecyclohexylidene)ethyl)diphenylphosphine oxide (2). A solution of HPPh₂ (0.27 L, 1.58 mol) in THF (9.5 L) was cooled to 0 °C. To the reaction mixture was added dropwise 2.5 M *n*-BuLi in *n*-hexane (0.63 L, 1.58 mol) at 0 °C. The reaction mixture was stirred at 0 °C for 30 min to generate LiPPh₂ solution. A solution of compound 18 (0.76 kg, 1.05 mol) in THF (9.5 L) was cooled to -60 °C. LiPPh₂ solution was added dropwise to the reaction mixture. The reaction mixture was stirred at -60 °C for 1 h, quenched with water (95 mL) at -60 °C, and evaporated. The resulting substance was diluted with CHCl₃ (13.5 L) and 5% H₂O₂ (4.5 L) was added at rt and stirred at rt for 10 min. The organic layer was separated, washed with aqueous 5% Na₂S₂O₃ solution (16.5 L), dried over Na₂SO₄, and concentrated in vacuo. Purification by silica gel column chromatography ($CH_2Cl_2/THF = 15:1$) gave product 2 (0.75 kg, 0.838 mol, 80%) as a colorless solid. ¹H NMR (300 MHz, CDCl₃) δ 7.78–7.71 (4H, m), 7.70–7.66 (4H, m), 7.59-7.37 (12H, m), 5.36 (1H, dd, J = 14.4, 7.2 Hz),5.26 (1H, s), 4.84 (1H, t, *J* = 1.8 Hz), 4.27 (1H, d, *J* = 7.5 Hz), 4.15-4.11 (1H, m), 3.79-3.64 (4H, m), 3.43-3.31 (1H, m), 3.28–3.15 (1H, m), 3.12 (1H, dd, J = 7.7, 2.0 Hz), 2.38–2.33 (1H, m), 2.23-2.15 (1H, m), 1.91-1.82 (2H, m), 1.07 (9H, s), 0.93 (9H, s), 0.83 (9H, s), 0.10 (3H, s), 0.05-0.03 (9H, m); ¹³C NMR (75 MHz, CDCl₃) δ 140.6, 135.7, 134.1, 131.9, 131.3, 131.2, 131.1, 129.7, 128.8, 128.7, 128.6, 127.7, 115.4, 69.2, 68.6, 61.5, 33.4, 27.0, 26.0, 19.3, 18.4, 18.3, -4.5, -4.6, -4.7.

(1R,3aR,7aR)-7a-Methyl-1-((R)-6-methylheptan-2-yl)octahydro-4H-inden-4-one (19). To a solution of vitamin D_3 (1.46 kg, 3.80 mol) in CH₂Cl₂ (47 L) and MeOH (15 L) was added NaHCO₃ (0.022 g, 0.26 mol) at rt. The reaction mixture was cooled to -65 °C. Ozone was passed through the solution (0.1 MPa, 2.0 mL/min) for 4 h until the reaction is complete. The reaction mixture was warmed to -30 °C, quenched with DMS (1.42 L, 19.02 mol), warmed to rt, and stirred for 18 h at rt. To the resulting solution was added water (15 L) and stirred for 30 min. The organic layer was extracted with CH_2Cl_2 (30 L), dried over Na_2SO_4 , and concentrated in vacuo. Purification by silica gel column chromatography (hexanes/AcOEt = 4:1) gave product 19 (0.91 kg, 3.44 mol, 85%) as a colorless liquid. ¹H NMR (300 MHz, CDCl₃) δ 2.47 (1H, dd, J = 11.7, 7.5 Hz), 2.29-2.25 (2H, m), 2.25-1.00(17H, m), 0.97 (3H, d, I = 6.0 Hz), 0.90 (3H, d, I = 6.6 Hz), 0.88 (3H, d, J = 6.6 Hz), 0.66 (3H, s); ¹³C NMR (75 MHz, CDCl₃) δ 212.2, 62.0, 56.7, 49.9, 41.0, 39.4, 39.0, 36.0, 35.5, 28.0, 27.5, 24.1, 23.8, 22.8, 22.5, 19.1, 18.7, 12.5.

(1R,3aR,7aR)-1-((R)-6-Hydroxy-6-methylheptan-2-yl)-7amethyloctahydro-4H-inden-4-one (**20**). To a solution of compound **19** (0.91 kg, 3.44 mol) in CH₃CN (33 L) and water (3.6 L) was added trifluoroacetone (6.1 L, 68.8 mol) at 0 °C. Oxone (12.7 kg, 20.6 mol) and NaHCO₃ (3.5 kg, 41.7 mol) were added to the reaction mixture. The reaction mixture was stirred at 0 °C for 24 h, quenched with water (9 L) and CH₂Cl₂ (9 L), stirred at rt for 1 h, and filtered. Water (18 L) was added to the crude mixture, and the mixture was stirred at rt for 30 min. The organic layer was separated, dried over Na₂SO₄, and concentrated in vacuo. Purification by silica gel chromatography (hexanes/AcOEt = 4:1) gave product **20** (0.51 kg, 1.82 mol, 53%) as colorless oil. ¹H NMR (300 MHz, CDCl₃) δ 2.45 (1H, dd, J = 11.7, 7.2 Hz), 2.28–1.00 (19H, m), 1.22 (6H, s), 0.97 (3H, d, J = 6.0 Hz), 0.64 (3H, s); ¹³C NMR (75 MHz, CDCl₃) δ 212.1, 71.0, 62.0, 56.7, 49.9, 44.3, 41.0, 39.0, 36.2, 35.5, 29.4, 29.2, 27.5, 24.1, 20.7, 19.1, 18.7, 12.5.

 $(1R, 3aR, 7aR) - 7a - Methyl - 1 - ((R) - 6 - methyl - 6 - ((trimethylsilyl)oxy)heptan-2-yl)octahydro-4H-inden-4-one (21). To a solution of compound 20 (0.51 kg, 1.82 mol) in CH₂Cl₂ (5.1 L) was added TMS-imidazole (0.76 kg, 5.42 mol) at rt. The reaction mixture was stirred at rt for 1.5 h, quenched with water (5.1 L), and stirred at rt for 10 min. The organic layer was separated, dried over Na₂SO₄, and concentrated in vacuo. Purification by silica gel chromatography (CH₂Cl₂) gave product 21 (0.50 kg, 1.42 mol, 78%) as colorless oil. ¹H NMR (300 MHz, CDCl₃) <math>\delta$ 2.55–2.41 (1H, m), 2.38–1.15 (17H, m), 1.22 (6H, s), 1.15–1.00 (1H, m), 0.98 (3H, d, J = 6.3 Hz), 0.66 (3H, s), 0.12 (9H, s); ¹³C NMR (75 MHz, CDCl₃) δ 212.2, 74.0, 62.0, 56.8, 50.0, 45.2, 41.0, 39.0, 36.3, 35.5, 30.0, 29.9, 27.5, 24.1, 20.8, 19.1, 18.7, 12.5, 2.6.

(((1R,2R,3R,Z)-2-(3-((tert-Butyldiphenylsilyl)oxy)propoxy)-5-(2-((1R,3aS,7aR,E)-7a-methyl-1-((R)-6-methyl-6-((trimethylsilyl)oxy)heptan-2-yl)octahydro-4H-inden-4ylidene)ethylidene)-4-methylenecyclohexane-1,3-diyl)bis-(oxy))bis(tert-butyldimethylsilane) (22). To a solution of compound 2 (0.146 kg, 0.17 mol) in THF (1.46 L) was added 2.5 M *n*-BuLi in *n*-hexane (0.065 L, 0.17 mol) dropwise at -65 °C. The reaction mixture was stirred at -65 °C for 1 h. A solution of compound 21 (0.029 kg, 0.083 mol) in THF (0.29 L) was added dropwise to the reaction mixture at -65 °C. The reaction mixture was warmed to -10 °C for 2 h, guenched with water (1.46 L), and warmed to rt, AcOEt (1.46 L) was added and stirred at rt for 30 min. The organic layer was separated, dried over Na2SO4, and concentrated in vacuo. Purification by silica gel chromatography (hexanes/AcOEt = 30:1) gave product 22 (0.075 kg, 0.073 mol, 89%) as colorless oil. ¹H NMR (300 MHz, CDCl₃) δ 7.71-7.59 (4H, m), 7.45-7.31 (6H, m), 6.21 (1H, d, J = 11.1 Hz), 5.99 (1H, d, J = 11.1 Hz), 5.29–5.18 (1H, m), 4.96 (1H, d, J = 2.4 Hz), 4.24–4.10 (2H, m), 4.12 (1H, q, J = 7.2 Hz), 3.83–3.62 (4H, m), 3.22 (1H, dd, J = 6.6, 2.1 Hz), 2.94-2.87 (1H, m), 2.50-2.38 (1H, m)m), 2.21 (1H, dd, J = 13.2, 3.6 Hz), 2.08–1.12 (18H, m), 1.20 (6H, s), 1.10–0.96 (10H, m), 0.93 (3H, d, J = 6.3 Hz), 0.90– 0.80 (18H, m), 0.53 (3H, s), 0.10––0.10 (21H, m); ¹³C NMR $(75 \text{ MHz}, \text{CDCl}_3) \delta$ 141.3, 135.7, 134.6, 134.2, 129.6, 127.7, 123.4, 118.0, 74.3, 69.6, 68.4, 61.4, 60.5, 56.8, 56.5, 45.9, 45.4, 40.8, 36.6, 36.3, 33.5, 30.1, 30.0, 27.9, 27.0, 26.0, 23.7, 22.3, 21.1, 19.4, 19.0, 18.4, 18.3, 14.3, 12.1, 2.8, -4.5, -4.7.

(1R,2R,3R,Z)-5-(2-((1R,3aS,7aR,E)-1-((R)-6-Hydroxy-6methylheptan-2-yl)-7a-methyloctahydro-4H-inden-4ylidene)ethylidene)-2-(3-hydroxypropoxy)-4-methylenecyclohexane-1,3-diol [Eldecalcitol] (1). To a solution of compound 22 (0.075 kg, 0.073 mol) in CH_2Cl_2 (0.56 L) and MeOH (0.56 L) was added p-TsOH·H₂O (0.014 kg, 0.073 mol) at rt. The reaction mixture was stirred at rt for 5 h, quenched with aqueous 3% NaHCO₃ solution (0.24 L), washed with aqueous 10% NaCl solution (0.24 L), dried over Na₂SO₄, and concentrated in vacuo. Purification by silica gel chromatography ($CH_2Cl_2/MeOH = 30:1$) gave eldecalcitol (1, 0.029 kg, 0.058 mol, 80%) as a white solid. Further purification by recrystallization from a mixture of CH₃CN (0.43 L) and water (0.43 L) gave 1 (0.020 kg, 0.041 mol). $[\alpha]_{20}^{D} = -39.4$ (c 0.4, EtOH); IR (neat) v max 3535, 3421, 3329, 2941, 2924, 2860, 1447, 1373, 1114, 1073, 1045 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 6.35 (1H, d, J = 11.2 Hz), 6.04 (1H, d, J =11.3 Hz), 5.50 (1H, t, J = 2.1 Hz), 5.08 (1H, t, J = 2.1 Hz),

4.33–4.29 (1H, m), 4.26–4.24 (1H, m), 3.95–3.87 (1H, m), 3.83 (2H, q, *J* = 5.3 Hz), 3.75–3.68 (1H, m), 3.27–3.23 (2H, m), 2.83–2.74 (2H, m), 2.63 (1H, d, *J* = 2.3 Hz), 2.54 (1H, dd, *J* = 14.5, 4.0 Hz), 2.41 (1H, d, *J* = 14.1 Hz), 2.02–1.93 (2H, m), 1.92–1.82 (3H, m), 1.71–1.63 (2H, m), 1.59–1.36 (8H, m), 1.29–1.20 (4H, m), 1.21 (6H, s), 1.09–1.00 (1H, m), 0.94 (3H, d, *J* = 6.3 Hz), 0.55 (3H, s), missing C-25 hydroxyl proton; ¹³C NMR (75 MHz, CDCl₃) δ 144.4, 143.1, 132.3, 125.0, 117.4, 111.9, 85.5, 71.6, 71.2, 68.4, 66.7, 61.2, 56.7, 56.5, 46.0, 44.5, 40.6, 36.5, 36.2, 32.0, 29.5, 29.3, 29.2, 27.8, 23.8, 22.5, 20.9, 19.0, 12.1; HRMS(ESI+) [M + Na]⁺ m/ z calcd for C₃₀H₅₀O₅Na⁺ 513.3550, found 513.3560.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.oprd.0c00436.

Synthesis of dienol **3** from D-mannitol (Scheme S1), experimental procedures of **23**, **24**, **3**, 7, and **8**, and copies of ¹H and ¹³C NMR spectra of all compounds (PDF)

AUTHOR INFORMATION

Corresponding Author

Hyunik Shin – R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea; orcid.org/0000-0002-0222-8418; Email: hishin@ yonsungchem.co.kr

Authors

- Hyung Wook Moon R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Seung Jong Lee R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Seong Hu Park R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Se Gyo Jung R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- In A Jung R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Chang Hun Seol R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Seung Woo Kim R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Seon Mi Lee R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Bogonda Gangganna R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Seokhwi Park R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Kee-Young Lee R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Chang-Young Oh R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Juyoung Song R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Jaehun Jung R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Ji Soo Heo R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea
- Kang Hee Lee R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea

Organic Process Research & Development

pubs.acs.org/OPRD

Hae Sol Kim – R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea

Won Taek Lee – R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea

Areum Baek – R&D Center, Yonsung Fine Chemicals Co., Ltd., Suwon-si, Gyeonggi-do 16675, Republic of Korea

Complete contact information is available at: https://pubs.acs.org/10.1021/acs.oprd.0c00436

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

This work is supported by the Advanced Technology Center (ATC) Program (No. 10048257) funded by the Ministry of Trade, Industry, and Energy of Korea. We thank Prof. Ok-Sang Jung at Pusan National University for x-ray crystallographic analysis of **1**.

REFERENCES

(1) Plum, L. A.; DeLuca, H. F. Vitamin D, disease and therapeutic opportunities. *Nat. Rev. Drug Discov.* **2010**, *9*, 941–955.

(2) Sintov, A. C.; Yarmolinsky, L.; Dahan, A.; Ben-Shabat, S. Pharmacological effects of vitamin D and its analogs: recent developments. *Drug Discovery Today* **2014**, *19*, 1769–1774.

(3) Leyssens, C.; Verlinden, L.; Verstuyf, A. The future of vitamin D analogs. *Front. Physiol.* **2014**, *5*, 122.

(4) Matsumoto, T.; Ito, M.; Hayashi, Y.; Hirota, T.; Tanigawara, Y.; Sone, T.; Fukunaga, M.; Shiraki, M.; Nakamura, T. A new active vitamin D_3 analog, eldecalcitol, prevents the risk of osteoporotic fractures – A randomized, active comparator, double-blind study. *Bone* **2011**, 49, 605–612.

(5) Sanford, M.; McCormack, P. L. Eldecalcitol: A Review of its Use in the Treatment of Osteoporosis. *Drugs* **2011**, *71*, 1755–1770.

(6) Kubodera, N. Diverse and Important Contributions by Medicinal Chemists to the Development of Pharmaceuticals: An Example of Active Vitamin D_3 Analog, Eldecalcitol. *Heterocycles* **2016**, *92*, 1013–1029.

(7) (a) Kubodera, N.; Hatakeyama, S. Synthesis of All Possible Aring Diastereomers at the 1- and 3-Positions of 1α ,25-Dihydroxy- 2β -(3-hydroxypropoxy)vitamin D₃ (ED-71) Using C₂-Symmetrical Epoxide as a Common Starting Material. Anticancer Res. **2009**, 29, 3571–3578. (b) Kubodera, N.; Hatakeyama, S. Process Development for the Practical Production of Eldecalcitol by Linear Convergent and Biomimetic Syntheses. Anticancer Res. **2012**, 32, 303–309.

(8) Miyamoto, K.; Murayama, E.; Ochi, K.; Watanabe, H.; Kubodera, N. Synthetic Studies of Vitamin D Analogues. XIV. Synthesis and Calcium Regulating Activity of Vitamin D₃ Analogues Bearing a Hydroxyalkoxy Group at the 2β -Position. *Chem. Pharm. Bull.* **1993**, *41*, 1111–1113.

(9) (a) Takahashi, T.; Nakazawa, M. Pd-Catalyzed Cyclization Approach to Chiral A-Ring Synthon for $1\alpha, 2\beta, 25$ -Trihydroxyvitamin D₃. Synlett 1993, 37-39. (b) Hatakeyama, S.; Ikeda, T.; Maeyama, J.; Esumi, T.; Iwabuchi, Y.; Irie, H.; Kawase, A.; Kubodera, N. Convergent synthesis of 1α , 25-dihydroxy- 2β -(3-hydroxypropoxy)vitamin D3 (ED-71). Bioorg. Med. Chem. Lett. 1997, 7, 2871-2874. (c) Hijikuro, I.; Doi, T.; Takahashi, T. Parallel Synthesis of Vitamin D₃ Library in the Solid-Phase. J. Am. Chem. Soc. 2001, 123, 3716-3722. (d) Maeyama, J.; Hiyamizu, H.; Takahashi, K.; Ishihara, J.; Hatakeyama, S.; Kubodera, N. Two convergent approaches to the synthesis of 1α ,25-dihydroxy- 2β -(3-hydroxypropoxy)vitamin D₃ (ED-71) by the Lythgoe and the Trost coupling reactions. Heterocycles 2006, 70, 295-307. (e) Sawada, D.; Katayama, T.; Tsukuda, Y.; Saito, N.; Saito, H.; Takagi, K.; Ochiai, E.; Ishizuka, S.; Takenouchi, K.; Kittaka, A. Synthesis of 2α - and 2β -substituted-14-epi-previtamin D₃ and their genomic activity. Tetrahedron 2010, 66, 5407-5423.

(f) Zhao, G.-D.; Liu, Z.-P. Structural revisions of the reported Aring phosphine oxide synthon for ED-71 (Eldecalcitol) and a new synthesis. *Tetrahedron* **2015**, *71*, 8033–8040. (g) In the course of the preparation of the manuscript, a similar approach has been published. See: Deng, W.; Pu, Q.; Gong, Y.; Zhou, L.; Sun, J.; Wang, C. Total synthesis of 1α ,25-dihydroxy- 2β -(3-hydroxypropoxy)vitamin D₃ (ED-71). *Tetrahedron* **2020**, *76*, 131081.

(10) Kittaka, A.; Suhara, Y.; Takayanagi, H.; Fujishima, T.; Kurihara, M.; Takayama, H. A Concise and Efficient Route to 2α -(ω -Hydroxyalkoxy)-1 α ,25-dihydroxyvitamin D₃: Remarkably High Affinity to Vitamin D Receptor. *Org. Lett.* **2000**, *2*, 2619–2622.

(11) Nicolaou, K. C.; Papahatjis, D. P.; Claremon, D. A.; Magolda, R. L.; Dolle, R. E. Total synthesis of ionophore antibiotic X-14547A. J. Org. Chem. 1985, 50, 1440–1456.

(12) (a) Schmidt, B.; Kunz, O.; Petersen, M. H. Total Syntheses of Naturally Occurring Seimatopolide A and Its Enantiomer from Chiral Pool Starting Materials Using a Bidirectional Strategy. J. Org. Chem. **2012**, 77, 10897–10906. (b) Schmidt, B.; Nave, S. Synthesis of Dihydrofurans and Dihydropyrans with Unsaturated Side Chains Based on Ring Size-Selective Ring-Closing Metathesis. Adv. Synth. Catal. **2007**, 349, 215–230. (c) Burke, S. D.; Sametz, G. M. Total Synthesis of 3-Deoxy-D-manno-2-octulosonic Acid (KDO) and 2-Deoxy- β -KDO. Org. Lett. **1999**, 1, 71–74. (d) Crombez-Robert, C.; Benazza, M.; Frechou, C.; Demailly, G. Efficient synthesis of α,ω dibromodideoxyalditols as precursors for α,ω -dithioalkylalditols. Carbohydr Res. **1997**, 303, 359–365. (e) Rama Rao, A. V.; Mysorekar, S. V.; Gurjar, M. K.; Yadav, J. S. Synthesis of (3R,4R)-1,5-hexadien-3,4-diol and its unsymmetrical derivatives: Application to (R)-(+)- α -lipoic acid. Tetrahedron Lett. **1987**, 28, 2183–2186.

(13) (a) Katsuki, T.; Sharpless, K. B. The first practical method for asymmetric epoxidation. J. Am. Chem. Soc. 1980, 102, 5974–5976. (b) Wang, C.-Y.; Hou, D.-R. Asymmetric Synthesis of (+)-Aspicilin. J. Chin. Chem. Soc. 2012, 59, 389–393. (c) Coleman, R. S.; Kong, J.-S. Stereocontrolled Synthesis of the Fully Elaborated Aziridine Core of the Azinomycins. J. Am. Chem. Soc. 1998, 120, 3538–3539.

(14) Satam, V. S.; Pedada, S. R.; Kamaraj, P.; Antao, N.; Singh, A.; Hindupur, R. M.; Pati, H. N.; Thompson, A. M.; Launay, D.; Martin, D. Development of a Scalable Process for the Synthesis of DNDI-VL-2098: A Potential Preclinical Drug Candidate for the Treatment of Visceral Leishmaniasis. *Org. Process Res. Dev.* **2017**, *21*, 52–59.

(15) Miyashita, M.; Yoshikoshi, A.; Grieco, P. A. Pyridinium *p*-toluenesulfonate. A mild and efficient catalyst for the tetrahydropyranylation of alcohols. *J. Org. Chem.* **1977**, *42*, 3772–3774.

(16) Maiti, G.; Roy, S. C. A Mild and Efficient Method for Selective Deprotection of Tetrahydropyranyl Ethers to Alcohols. *J. Org. Chem.* **1996**, *61*, 6038–6039.

(17) Srikrishna, A.; Sattigeri, J. A.; Viswajanani, R.; Yelamaggad, C. V. Simple and Regioselective Reductive Cleavage of Tetrahydropyranyl Ethers to Alcohols. J. Org. Chem. **1995**, 60, 2260.

(18) Kim, S.; Park, J. H. Selective removal of tetrahydropyranyl ethers in the presence of *tert*-butyldimethylsilyl ethers with magnesium bromide in ether. *Tetrahedron Lett.* **1987**, *28*, 439–440.

(19) Ogawa, Y.; Shibasaki, M. Selective removal of tetrahydropyranyl ethers in the presence of *t*-butyldimethylsilyl ethers. *Tetrahedron Lett.* **1984**, 25, 663-664.

(20) (a) Baggiolini, E. G.; Iacobelli, J. A.; Hennessy, B. M.; Batcho, A. D.; Sereno, J. F.; Uskoković, M. R. Stereocontrolled total synthesis of 1 α ,25-dihydroxycholecalciferol and 1 α ,25-dihydroxyergocalciferol. J. Org. Chem. **1986**, 51, 3098–3108. (b) Kiegiel, J.; Wovkulich, P. M.; Uskoković, M. R. Chemical conversion of vitamin D₃ to its 1,35dihydroxy metabolite. Tetrahedron Lett. **1991**, 32, 6057–6060. (c) Nagasawa, K.; Matsuda, N.; Noguchi, Y.; Yamanashi, M.; Zako, Y.; Shimizu, I. Stereoselective synthesis of cyclopentanones by reductive cleavage of 6-oxonorbornane-2-carboxylates and its application to the synthesis of 1 α ,25-dihydroxyvitamin D₃ CD ring. J. Org. Chem. **1993**, 58, 1483–1490. (d) Hernández-Martín, A.; González-García, T.; Lawlor, M.; Preston, L.; Gotor, V.; Fernández, S.; Ferrero, M. Synthesis of vitamin D₃ analogues with A-ring modifications to directly measure vitamin D levels in biological samples. *Bioorg. Med. Chem.* **2013**, *21*, 7779–7789. (21) Cruci, R.; D'Accolti, L.; Fusco, C. A Novel Approach to the

(21) Cruci, R.; D'Accolti, L.; Fusco, C. A Novel Approach to the Efficient Oxygenation of Hydrocarbons under Mild Conditions. Superior Oxo Transfer Selectivity Using Dioxiranes. *Acc. Chem. Res.* **2006**, *39*, 1–9.