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Selective 3- and 6-OH modification of (-)-clausenamide

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Abstract

(–)-Clausenamide is a drug candidate under Phase I clinical trial for treatment of Alzheimer's disease (AD). In order to elucidate the substituent related structure–activity relationship, six one-substituent modified (–)-clausenamide analogues were designed, and four of them, namely 3-*O*-methyl, 6-*O*-methyl, 3-des-hydroxyl and 6-des-hydroxyl analogues were prepared by selective 3- and 6-OH modification of (–)-clausenamide.

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Keywords: (-)-Clausenamide; Derivatives; Synthesis; Alzheimer's disease

Racemic clausenamide ((\pm)-1, Fig. 1) is a natural product isolated from aqueous extract of the leaves of *Clausena lansium* (*Lour*) *skeels* [1]. The racemate showed remarkable nootropic activity in a number of *in vivo* biological tests, but the *levo* rotatory enantiomer, ((–)-clausenamide 1, Fig. 1), was eventually found to be the active isomer [2,3]. Presently, this chiral compound is under clinical trial for treatment of Alzheimer's disease. In order to get the insight view of stereochemistry related structure–activity relationship (SAR), all 16 stereoisomers of (–)-1 were prepared and their biological activities evaluated in an earlier study [4].

As the molecule of (-)-clausenamide is featured not only with the four chiral centers but also with the four substituents: 3-OH, 6-OH, 4-Ph and 6-Ph, it is interesting to find how the presence of these substituents is related to its biological activity. In order to get this question answered, six one-substituent modified (-)-clasuenamide analogues (2–5, 6 and 7, Fig. 1) were designed. In this letter, we describe the preparation of compound 2–5, using (-)-clausenamide as starting material since it provides the fastest access to the target molecules in the forms of pure enantiomers.

Compound **2** and **3** are simple mono-*O*-methylated derivatives of (-)-**1**. Because the 3-OH of (-)-**1** is *trans* to both C4 and C5 substituents and less hindered, compound **2** was expected to be prepared by selective 3-*O*-methylation of the starting material. After screening of several commonly used *O*-methylation conditions including MeI/NaH in DMF, MeI/K₂CO₃ in acetone and Me₂SO₄/KOH in toluene/H₂O, reaction under the last biphasic condition was found to give the best result, and (-)-**1** was converted into compound **2** as single product in excellent yield (90%) (Scheme 1). This encouraged us to go one step further to carry out the 3-OH selective protection of (-)-**1** with TBS under classic silylation condition [5], and compound **8** was isolated in 81% yield. Subsequent 6-OH methylation of **8** followed by release of the 3-OH with TBAF afforded compound **3** in 65% yield over two steps (Scheme 1).

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Мe

2



Scheme 1. Synthesis of 2 and 3. Reagents and conditions: (a) Me₂SO₄/KOH in toluene/H₂O; (b) TBSCl/imidazole; (c) MeI/NaH; (d) TBAF.

Мe

(-)-Clausenamide 1

Ŵе

8

ОH

Мe

3

MeÕ

In our initial plan (Scheme 2), an S-methyl xanthate mediate dehydroxylation was designed as the first step to realize the transformation of 8 into compound 4. Unfortunately, treatment of 8 with CS₂/NaH and MeI failed to give 9. Therefore, an alternative plan *en route* intermediate **10** was proposed. Compound **8** was thus treated with POCl₃/Py, a commonly used dehydration reagent [6], but afforded an unexpected product 11, which was believed to form from double bond migration of the original elimination product 10 in 87% yield. To our delight, when the TBS group was removed by treatment of 11 with TBAF, the location of C=C double bond switched back to C5–C6 with the restoration of R configuration at C4 [7], and 12 was isolated in 77% yield. With intermediate 12 in hand, a variety of hydrogenation conditions were tested, and the reaction in anhydrous THF with Raney Ni as catalyst was found to give the optimal yield of 4 (74% after crystallization of the crude product from ethyl acetate).

Similar to that of 4, synthesis of 5 was also started with intermediate 8, which was first treated with benzyl bromide/ NaH to have the 6-OH protected, and then with TBAF to remove the 3-O-TBS group to give key intermediate 13 in



Scheme 2. Synthesis of 4. Reagents and conditions: (a) 1-NaH, imidazole; 2-CS2; 3-MeI; (b) POCl₃/Py; (c) TBAF; (d) 10% Pd/C, H₂.



Scheme 3. Synthesis of 5. Reagents and conditions: (a) NaH, BnBr; (b) TBAF; (c) 1—NaH, imidazole; 2—CS₂; 3—MeI; (d) ((*n*-Bu)₄N)₂S₂O₈, HCOONa; (e) T-1 Raney Ni, H₂; (f) 10% Pd/C, H₂.

63% yield over two steps (Scheme 2). Unlike compound **8**, conversion of **13** into the corresponding *S*-methyl xanthate proved to be easy and **14** was isolated in 78% yield. Treatment of *S*-methyl xanthate **14** with sodium formate in the presence of $((n-Bu)_4N)_2S_2O_8$ as radical initiator [8] afforded an inseparable 1:1 [9] mixture of dehydroxylation product **16** and elimination product **15** in a total yield of 66%. Raney Ni catalyzed hydrogenation of this mixture, during which compound **15** was saturated with the incoming hydrogen atoms attacking the less hindered α face, gave **16** in pure form in 91% yield. Deprotection of 6-OH was then furnished by palladium catalyzed hydrogenolysis of **16** to afford target molecule **5** in 96% yield (Scheme 3).

In conclusion, a small library consists of six one-substituent modified (–)-clausenamide analogues was designed as probe for structure–activity relationship elucidation, and four members possessing retained carbon skeletons were synthesized in optical active form by selective OH modifications using readily available (–)-clausenamide as starting material [10]. It was demonstrated for the first time that the steric hindrance margin between the 3- and 6-OH in the molecule of (–)-clausenamide provided a sufficiently wide open window for manipulation of OH protective groups. Des-hydroxyl modifications to the 6- or 3-OH could then be carried out by adoption of an elimination–hydrogenation sequence or an *S*-methyl xanthate mediated dehydroxylation procedure, respectively. Synthesis of the other two library members, **6** and **7**, as well as the results from biological studies on the nootropic effect of compound **2–7** will be published separately.

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- [9] The proportions of the two compounds were determined by 1H NMR of the mixture.
- [10] Analytical data for compounds 2–5. Compound 2: mp: 191.0–191.9 °C; $[\alpha]_D^{20}$ –184.3 (c 1.07, CHCl₃); ¹H NMR (300 MHz, CDCl₃): δ 2.84 (s, 3H), 3.55 (s, 3H), 3.83 (d, 1H, J = 9.0 Hz), 3.81 (dd, 1H, J = 9.0, 8.7 Hz), 4.15 (dd, 1H, J = 4.2, 8.7 Hz), 4.75 (d, 1H, J = 4.2 Hz), 6.85-7.37 (m, 10H); ¹³C NMR (100 MHz, DMSO-d_s): δ 172.7, 140.9, 135.7, 128.6, 127.6, 127.2, 126.4, 126.3, 77.4, 71.1, 65.2, 57.3, 47.3, 29.5; FT-IR (cm⁻¹): 1681.1; EI-MS (*mlz*, %): 312 (M⁺+1, 2), 204 (100), 176 (100); HRMS (M⁺+H) calcd. for C₁₉H₂₁NO₃ 312.1600, found 312.1590; *Compound 3*: mp: 157.5–58.4 °C; $[\alpha]_D^2$ –146.4 (*c* 1.11, CHCl₃); ¹H NMR (300 MHz, DMSO-*d*₆): δ 2.71 (s, 3H), 2.77 (s, 3H), 3.55 (dd, 1H, J = 11.1, 8.1 Hz), 3.77 (dd, 1H, J = 6.3, 11.1 Hz), 4.12 (d, 1H, J = 4.2 Hz), 4.25 (dd, 1H, J = 4.2, 8.1 Hz), 5.44 (d, 1H, J = 6.3 Hz), 6.68-7.38 (m, 10H); ¹³C NMR (100 MHz, DMSO-*d*₆): δ 174.1, 137.1, 136.2, 128.6 × 2, 127.9, 126.6, 82.6, 68.3, 63.7, 55.9, 49.5, 30.6; FT-IR: 1672.6; EI-MS (*m*/*z*, %): 312 (M⁺ + 1, 2), 191 (84), 162 (96), 134 (100); HRMS (M⁺+H) calcd. for C₁₉H₂₁NO₃ 312.1600, found 312.1517; *Compound* 4: mp: 143.3–144.0 °C; $[\alpha]_D^{20}$ – 267.4 (c 0.87, CHCl₃); ¹H NMR (300 MHz, CDCl₃): δ 2.44 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 1H, J = 8.1, 14.1 Hz), 2.62 (s, 3H), 2.64 (dd, 2H), 1H, J = 4.2, 14.1 Hz), 3.73 (dd, 1H, J = 8.1, 10.8 Hz), 4.04 (ddd, 1H, J = 8.1, 8.1, 4.2 Hz), 4.59 (d, 1H, J = 10.8 Hz), 6.83-7.42 (m, 10H); ¹³C NMR (100 MHz, CDCl₃): δ 174.3, 137.6, 135.9, 129.1, 128.8, 128.5, 128.1, 127.4, 126.7, 70.0, 63.0, 51.7, 37.0, 30.3; FT-IR (cm⁻¹): 1763.4, 1603.4; MS (*m*/*z*, %): 281 (M⁺, 1), 190 (100), 91 (85); HRMS (M⁺+H) calcd. for C₁₈H₂₀NO₂ 282.1488, found 282.1489; *Compound* 5: mp: 104.5–105.8 °C; [α]_D²⁰ –119.7 (c 0.8, CHCl₃); ¹H NMR (300 MHz, CDCl₃): δ 2.35 (d, 2H, J = 9.9), 2.75 (s, 3H), 3.92 (dd, 1H, J = 9.9, 7.8 Hz), 4.15 (dd, 1H, J = 5.1, 7.8, 3.9 Hz), 4.73 (d, 1H, J = 5.1 Hz), 6.91–7.40 (m, 10H); ¹³C NMR (100 MHz, CDCl3): δ 174.7, 139.7, 137.0, 128.7, 139.7, 139.7, 137.0, 128.7, 139.7, 1 128.3, 128.27, 127.5, 127.3, 73.7, 68.8, 41.9, 33.7, 30.9; FT-IR (cm⁻¹): 1671.5, 1602.6, 1484.9; FAB-MS (*m/z*, %): 282 (M⁺+1, 100); HRMS $(M^{+} + 1)$: calcd. for $C_{18}H_{20}NO_2$ 282.1494, found 282.1501.