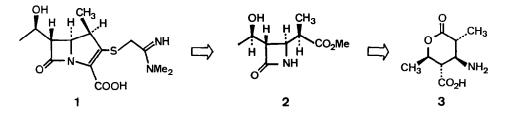
A STEREOSELECTIVE APPROACH TO 1β -METHYLCARBAPENEM ANTIBIOTIC STARTING FROM (<u>R</u>)-(-)-3-HYDROXYBUTYRIC ACID ESTER

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<u>Summary</u>; The chiral key intermediate of 1β -methylcarbapenem antibiotic is synthesized via $(2\underline{R}, 3\underline{R}, 4\underline{S}, 5\underline{R})$ -3-amino-2,5-dimethyl-5-pentanolide-4-carboxylic acid, which is prepared in a stereoselective manner from (\underline{R}) -(-)-3-hydroxy-butyrate.

Synthetic carbapenem antibiotic, 1β -methylcarbapenem (1), has attracted much attention.¹ Thienamycin and related naturally occurring carbapenem compounds, in spite of their potent, broad-spectrum antibacterial activity, suffer serious disadvantages that they are chemically unstable and metabolized by renal dehydropeptidase-I (DHP-I). In intensive efforts to improve the properties by the chemical modification of these antibiotics, Shih et al. have first synthesized 1β -methylcarbapenem compound 1,^{1a} which showed enhanced chemical stability and resistance to DHP-I. The major problem in the synthesis of this compound is a stereocontrolled construction of the four continuous chiral centers in the molecule. Although several recent publications have dealt with the synthesis of 1,^{1,2} most of the methods was not satisfactory in terms of stereoselectivity and/or practical sense. Herein we report a stereocontrolled synthesis of the key intermediate 2 of 1β -methylcarbapenem compound via the lactone 3 containing the chirality and functionality of 1.

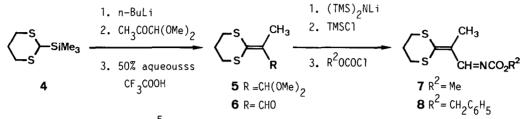


In the preceding paper, we described the reaction of the dianion of (\underline{R}) -(-)-3-hydroxybutyrate with N-acylaldimine leading to the (+)-thienamycin precursor having the three continuous chiral centers.³ The methodology was applied successfully to synthesis of **3**, in which the chirality of the starting material, (\underline{R}) -(-)-3-hydroxybutyric acid, controlled construction of the four continuous chiral centers in a highly stereoselective manner.⁴

The starting material, 2-(1,3-dithian-2-ylidene)-propanal (6, mp 80-82

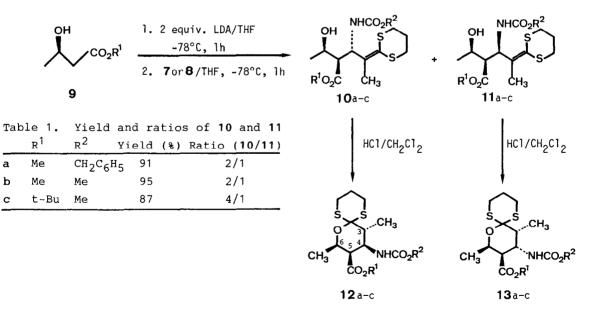
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°C), was prepared in 71% yield by condensation of 2-trimethylsilyl-1,3-dithiane (4) with pyruvaldehyde dimethylacetal followed by acid hydrolysis. The aldehyde 6 was then converted quantitatively in one-pot into the N-alkoxy-carbonylimine 7 (mp 85-88 °C) and 8 (mp 89-91 °C) by treating in turn with lithium bis(trimethylsilyl)amide (-78°C, 1h), chlorotrimethylsilane (-78°C~r.t., 1.5h) and then benzyl chloroformate and methyl chloroformate,



respectively (r.t., 10h).5

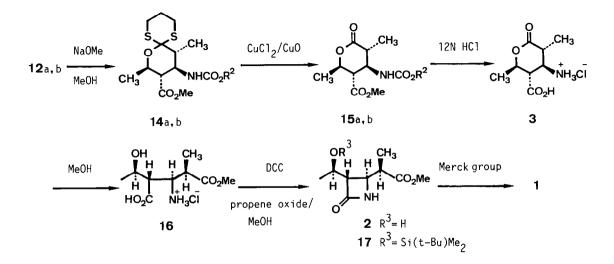
The N-alkoxycarbonylimines thus obtained were subjected to reaction with the dianions of R-(-)-3-hydroxybutyric acid esters in a similar manner to that reported³ (Table 1). The methyl ester of 9 reacted with 7 and 8 to give 2:1 mixtures of the (syn, anti)-products 10 and the (syn, syn)-products 11. Although the t-butyl ester of 9 gave a 4:1 mixture of 10c and 11c with better stereoselection,⁶ the methyl esters 10a and 10b were used in subsequent transformation because the t-butyl ester of 9 was obtained only in poor yield via titanate-mediated transesterification of the methyl ester⁷. When the compounds



10a and **10b** were treated with catalitic amounts (below 0.1 molar equivalent) of hydrogen chloride in dichloromethane at 0°C, cyclization proceeded smoothly and stereoselectively to give almost exclusively the desired 3α -methyl cyclic dithioorthoesters **12a** (oil, $[\alpha]_{D}^{29}$ +133.6° (c 0.75, MeOH)) and **12b** (mp 72-74°C,

 $[\alpha]_D^{29}$ +166° (c 1.0,MeOH)) in 87% and 86% yields, respectively. The stereochemistry of 12 was estimated on the basis of the coupling constants (for example 12b: $J_{3,4}$ =11.5 Hz, $J_{4,5}$ =4.8 Hz, $J_{5,6}$ =3.8 Hz) observed in their ¹H-NMR spectra. Cyclization of 11a and 11b under the same conditions also afforded the 3 α -methyl compounds 13a and 13b in 71% and 68% yields, respectively. It is noted that highly preferred formation of the 3 α -methyl products from both of 10 and 11 implies that this cyclization process may provide a general method for 1.4-chiral transfer.⁸

The remaining problem of the stereochemistry was resolved by heating 12a with sodium methoxide in refuxing methanol in the presence of methyl acetate, which effected inversion of the methoxycarbonyl group at C-5 to give 14a (mp 145-146°C, $[\alpha]_{29}^{29}$ +81.7° (c 0.85, MeOH), ¹H-NMR: J_{3,4}=11.1 Hz, J_{4,5}=10.8 Hz,



 $J_{5,6}$ =10.6 Hz) in 74% yield. The compound **14a** was then converted in 91% yield by treatment with cupric chloride and cupric oxide in aqueous acetone into the desired lactone **15a**⁹ having the same chirality with 1 α -methylcarbapenem **1**. Treatment of **15a** with 12N hydrochloric acid at room temperature gave the amino acid **3** as the hydrochloride quantitatively. On the other hand, an attempt to remove the methoxycarbonyl group of **15b**, prepared through the same reaction sequences from **12b**, in refluxing 12N hydrochloric acid was accompanied by epimerization of the C-2 methyl group. Finally, methanolysis of **3** followed by cyclization of **16** with dicyclohexylcarbodiimide (DCC) furnished the β -lactam 2^{10} in 75% yield. The stereochemistry of **2** was confirmed by converting into the known O-protected β -lactam **17**, the key intermediate of Merck's synthesis of $1,^1$ the physical properties of which were identical with those reported by Shih et al..^{1a}

Thus, we achieved highly stereoselective and short-step synthesis of the key intermediate of 1 via continuous construction of the four chiral centers from (\underline{R})-(-)-3-hydroxybutyrate¹¹.

References and Notes

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- 3. M. Hatanaka and H. Nitta, Tetrahedron Lett., The preceding paper.
- 4. After this work finished, an alternative approach to the β -lactam 17 starting from (<u>R</u>)-(-)-3-hydroxybutyric acid has been published: T. Iimori and M. Shibasaki, <u>Tetrahedron Lett.</u>, 27, 2149 (1986).
- The literature method was modified: R. Kupger, S. Meier and E-U. Wurthmein, <u>Synthesis</u>, 688 (1984).
- In comparison with the reaction of 9 with N-methoxycarbonyl (2-phenyl-thio)ethenylcarboxaldimine (see ref. 3), the effect of the ester moiety of 9 on stereoselection was lowered in the present case, presumably due to increasing steric bulkiness of N-methoxycarbonylaldimine 7.
- D. Seebach, E. Hungerbuhler, R. Naef, P. Schnurrenberger, B. Wiedmann, and M. Zuger, <u>Synthesis</u>, 1982, 138.
- After this paper was completed, Suzuki et al., have reported quite similar type of cyclization of ketene dithioacetal-alcohol, which proceeded via thermodynamically controlled process: K. Suzuki, T. Masuda, Y. Fukazawa, and G. Tsuchihashi, <u>Tetrahedron Lett.</u> 27, 3661 (1986).
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- 10. The compound 2: mp 102-104°C; $[\alpha]_D^{29} 44.7^\circ$ (c 0.45, CHCl₃); IR (CH₂Cl₂) 3410, 1768, 1735 cm⁻¹; ¹H-NMR (CDCl₃) δ 1.27 (3H, d, J=7.1 Hz), 1.31 (3H, d, J=6.3 Hz), 2.67 (1H, m), 2.98 (1H, dd, J=2.1, 7.0 Hz), 3.72 (3H, s), 3.77 (1H, dd, J=2.1, 7.7 Hz), 4.16 (1H, m), 6.09 (1H, broad s).
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