Design, Synthesis, and Binding Studies of New Potent Ligands of Cannabinoid Receptors

Antonella Brizzi,†,* Vittorio Brizzi,† Maria Grazia Cascio,‡ Tiziana Bisogno,‡ Rossella Sirianni,† and Vincenzo Di Marzo[‡]

Dipartimento Farmaco Chimico Tecnologico, Università degli Studi di Siena, Via A. Moro 2, 53100 Siena (Italy), and Endocannabinoid Research Group, Istituto di Chimica Biomolecolare, Consiglio Nazionale delle Ricerche, Via Campi Flegrei 34, Comprensorio Olivetti, 80078 Pozzuoli, Napoli (Italy)

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Despite their different chemical structures, Δ^9 -tetrahydrocannabinol (THC) and anandamide (AEA) have common pharmacological properties. This study was aimed at finding new cannabinoid receptor ligands that overcome the instability of AEA and its analogues. To this end we planned the synthesis of a series of compounds which retained both a rigid structure, like that of plant cannabinoids, and a flexible portion similar to that of anandamide. Binding studies on CB₁ and CB₂ receptors, anandamide membrane transporter (AMT), and fatty acid amide hydrolase (FAAH) showed that some of the newly developed compounds have high affinity and specificity for cannabinoid CB₁ and CB₂ receptors. Compound 25 is a potent CB₁ and CB₂ ligand, with affinity constants significantly lower than AEA and similar to WIN 55-212, compound **52** is a potent CB₂ ligand, although not very selective over CB₁ receptors, and compound 43 is CB₂ ligand, with at least a 26-fold selectivity over CB₁ receptors. Compound 25 behaved as a inverse agonist at CB₁ receptors as assessed in the cyclic AMP functional

Introduction

The great popularity of hashish and marihuana, both derived from the Indian hemp Cannabis Sativa L., and of their major psychoactive component, (-)-trans- Δ^9 tetrahydrocannabinol (THC, Chart 1),1 is due to their medicinal as well as their psychotomimetic effects. Cannabinoids have been shown to produce a wide range of pharmacological effects² mediated by two receptors: CB₁, very abundant in the brain³ and cloned in 1990,⁴ and CB2, expressed predominantly in the periphery,5 mainly in the spleen and in immune cells. The presence of these receptors in mammalian cells was indicative of the existence of an endogenous cannabinoid system. Several years ago the first natural ligand was isolated from porcine brain⁶ and identified as an amide related to arachidonic acid, the cis-5,8,11,14-eicosatetraenoylethanolamide, or anandamide (AEA), showing properties similar to the plant-derived agonist THC. Three years later a well-known intermediate in phosphoglyceride metabolism, 2-arachidonoylglycerol (2-AG), was found to act also as a potential ligand at cannabinoid receptors.⁷⁻⁹

The molecular basis of the biological effects of these endogenous compounds were elucidated also thanks to the understanding of their metabolism, including biochemical pathways that lead to their synthesis and inactivation. Both AEA and 2-AG are produced from the metabolism of membrane phospholipids and immediately released into extracellular space. Inactivation occurs in two steps: cellular reuptake, facilitated by a putative transporter protein known as the anandamide

Chart 1

membrane transporter (AMT)¹⁰ and enzymatic hydrolysis by a fatty acid amide hydrolase (FAAH)11 or by a monoacylglycerol lipase, in the case of 2-AG. The proteins of the endocannabinoid system, including CB₁, CB₂, AMT, and FAAH, represent excellent targets for the development of new therapeutic drugs to be employed in many pathologies, like pain, immunodepression, vascular perpherical disorders, increase or lowering of appetite, and motor disorders. The therapeutic value of these compounds as analgesic, anti-glaucoma, and antiinflammatory agents, and in the chemotherapyinduced nausea and vomiting and in appetite stimulation of patients with AIDS, is nowadays of great interest. $^{\bar{1}2}$

Since these discoveries several groups have carried out the synthesis and biological evaluation of AEA analogues, obtained by modifications of the fatty acyl chain or/and of the ethanolamide "head". 13 Unfortunately, arachidonic acid and analogous polyunsaturated fatty acids are very expensive and/or easily oxidized. AEA itself is very unstable and in the body is rapidly metabolized by FAAH and/or oxidation. To prevent in part the degradation of AEA and of its synthetic

^{*} Corresponding author. Fax: +39 577 234333; Phone: +39 577 234327; e-mail: brizzi3@unisi.it.

Università degli Studi di Siena.

[‡] Consiglio Nazionale delle Ricerche.

Figure 1.

Figure 2.

Scheme 1a

OH O(CH₂)nCOOCH₃ O(CH₂)nCONHR²

$$R^{1} \xrightarrow{i} R^{1} \xrightarrow{ii} R^{1}$$
1-14 15-56

 a Reagents: (i) Br-(CH₂)nCOOCH₃, CH₃COCH₃, K₂CO₃, KF, reflux, 48–72 h; (ii) Method A: MeOH/NaOH, reflux, 3 h; amine, HOBt, CMC, r.t., overnight; (iii) Method B: H₂NCH₂CH₂OH, 120–130 °C, 5 h.

analogues their activity at cannabinoid receptors is often assessed by conducting receptor binding assays in the presence of a serine protease inhibitor, phenylmethane-sulfonyl fluoride (PMSF).¹⁴

While AEA is a very flexible molecule, as demonstrated also by NMR studies, ¹⁵ and can assume both linear and nonlinear conformations, THC, and cannabinoid analogues, including tricyclic and bicyclic cannabinoids, are rigid compounds. Nevertheless, AEA and THC have been shown to share many, albeit not all, pharmacological properties. Bearing in mind the pharmacophore requirements ¹⁶ of both AEA and THC, necessary for the binding to cannabinoid CB₁ and CB₂ receptors, we have planned the synthesis of a series of stable compounds which consist of both a rigid aromatic portion, as in THC, and a flexible chain, as in AEA.

Olivetol, or 5-pentyl-benzene-1,3-diol (5-pentylresorcinol), is involved in the biogenesis of THC in Cannabis¹⁷ (Figure 1), and it has been selected as the basic aromatic scaffold of the rigid part of our new derivatives; other modified phenols have also been considered in this work in order to define a structure—activity relationship. As flexible chains we have chosen saturated carboxylic acids with chain lengths of 6 or 11 or 16 carbon atoms, easily converted into the final amides. Finally, we have evaluated the influence of several factors such as substitutions of the aromatic backbone, the length of the chain carrying the amidic head, and nature of amide.

As a result, different series of new compounds were designed and synthesized, whose general structure is reported in Figure 2.

Chemistry. As shown in Scheme 1, the synthesis of the new compounds starts from a substituted phenol [3-pentadecylphenol, olivetol, resorcinol (or benzene-1,3-diol), and 4-hexylresorcinol] which is reacted with a bromoalkylmethyl ester (6-bromohexanoic acid methyl ester or 11-bromoundecanoic acid methyl ester or 16-bromo-hexadecanoic acid methyl ester) in dry acetone in the presence of anhydrous K_2CO_3 and KF, giving esters 1-14. The final amides are obtained by either of two simple methods. Method A: esters were refluxed with methanolic/aqueous sodium hydroxide solution to

alkylation OH in C-3

alkylation OH in C-1

Figure 3.

give the corresponding acids, which were used as such and reacted with amines in the presence of 1-hydroxy-benzotriazole (HOBt) and 1-cyclohexyl-3-(2-morpholinoethyl)carbodiimide methyl-p-toluenesulfonate (CMC) in dichloromethane or acetonitrile. ¹⁸ Method B: esters were warmed with ethanolamine as solvent.

The O-alkylation of phenols was carried out with or without KF; as expected, in both conditions the reaction afforded all possible alkylated products when the aromatic nucleus had two free hydroxyl groups, although addition of KF improved the yields and specifically increased those of monoalkylated over dialkylated products. Moreover, O-alkylation of 4-hexylresorcinol afforded two different monoalkylated regioisomers. The reaction occurred mainly on the hydroxyl in C-1 irrespectively of the bromomethyl ester used and independently from the length of its chain; in fact the C-1 position results less hindered when the hexyl chain is present as para substituent. The structure of two regioisomers has been definitively assigned by NOESY experiments which showed a NOE effect between the OCH₂ and both the ortho aromatic hydrogens (H-2 and H-6) in the isomer with the alkylated hydroxyl in C-1, and a NOE effect between the OCH₂ and the only ortho aromatic hydrogen (H-2) in the isomer with the alkylated hydroxyl in C-3 (Figure 3).

Pharmacological Evaluation. Binding Assays. For CB₁ and CB₂ receptor binding assays, the new compounds were first subjected to a preliminary screening carried out using three concentrations (5, 10, and $25 \,\mu\text{M}$), membranes from rat brain and spleen, and [3H]-N-(piperidin-1-yl)-5-(4-chlorophenyl)-1-(2,4-dichlorophenvl)-4-methyl-1*H*-pyrazole-3-carboxamide hydrochloride (SR141716A, 2 nM) as the high affinity ligand. 19 Compounds which displaced [3H]-SR141716A by more than 50% at 10 μ M were further analyzed by using P₂ membranes from COS cells transfected with either the human CB₁ or CB₂ receptor and [³H]-(-)-cis-3-[2-hydroxy-4-(1,1-dimethylheptyl)phenyl]-trans-4-(3-hydroxypropyl)cyclohexanol ([3 H]CP-55,940) ($K_d = 690$ pM) as the high affinity ligand as described by the manufacturer (Perkin-Elmer, Italy). Displacement curves were generated by incubating drugs with 0.5 nM of [3H]CP-55,940. In all cases, K_i values were calculated by applying the Cheng-Prusoff equation to the IC₅₀ values (obtained by GraphPad) for the displacement of the bound radioligand by increasing concentrations of the test compounds.

Fatty Acid Amide Hydrolase Assays. The effect of compounds on the enzymatic hydrolysis of [14 C]-anandamide (6 μ M) was studied by using membranes prepared from rat brain incubated with increasing concentrations of compounds in 50 mM Tris-HCl, pH 9, for 30 min at 37 °C. 19 [14 C]Ethanolamine produced from [14 C]anandamide hydrolysis was measured by scintilla-

Table 1. Radioligand Binding Assays and Selectivity over Other Proteins of the Endocannabinoid System of the Synthesized Compoundsa

$$O(CH_2)nCONHR^2$$

| compd | R | R_1 | n | R_2 | CB_1 | CB_2 | FAAH | AMT |
|-----------|--------------|--|----------|---|-------------------------|----------------------------|------|-------------|
| 15 | H | $3-CH_2(CH_2)_{13}CH_3$ | 5 | CH_2CH_2OH | n.t. | n.t. | n.a. | >25 |
| 16 | H | $3-CH_2(CH_2)_{13}CH_3$ | 5 | $c.\mathrm{C_3H_5}$ | n.t. | n.t. | n.a. | n.t. |
| 17 | H | $3-CH_2(CH_2)_{13}CH_3$ | 5 | $p.OH-C_6H_4$ | n.t. | n.t. | n.a. | >25 |
| 18 | H | $3-CH_2(CH_2)_{13}CH_3$ | 10 | CH_2CH_2OH | n.t. | n.t. | n.t. | n.t. |
| 19 | H | $3-CH_2(CH_2)_{13}CH_3$ | 10 | $c.\mathrm{C_3H_5}$ | n.t. | n.t. | n.t. | n.t. |
| 20 | H | $3-CH_2(CH_2)_{13}CH_3$ | 10 | $p.OH-C_6H_4$ | n.t. | n.t. | n.t. | n.t. |
| 21 | 3-OH | 5-CH ₂ (CH ₂) ₃ CH ₃ | 5 | $\mathrm{CH_{2}CH_{2}OH}$ | n.t. | n.a | n.a. | n.t. |
| 22 | 3-OH | 5-CH ₂ (CH ₂) ₃ CH ₃ | 5 | $c.\mathrm{C_3H_5}$ | 1.3 | 0.96 | n.a | n.t. |
| 23 | 3-OH | 5-CH ₂ (CH ₂) ₃ CH ₃ | 5 | $p.OH-C_6H_4$ | n.t. | n.t. | n.a. | n.t. |
| 24 | 3-OH | $5-CH_2(CH_2)_3CH_3$ | 10 | CH ₂ CH ₂ OH | 0.8 | 0.16 | n.a | 25 |
| 25 | 3-OH | 5-CH ₂ (CH ₂) ₃ CH ₃ | 10 | $c.C_3H_5$ | 0.0052 | 0.013 | n.a | 17 |
| 26 | 3-OH | 5-CH ₂ (CH ₂) ₃ CH ₃ 5-CH ₂ (CH ₂) ₃ CH ₃ | 10 | p.OH-C ₆ H ₄ | 3 | 1.4 | n.a. | n.a. |
| 20 27 | 3-OH | | 15 | $_{ m CH_2CH_2OH}^{ m p.OH-C_6H_4}$ | 12.5 | | | |
| | | 5-CH ₂ (CH ₂) ₃ CH ₃ | 15 | | | n.a. | n.t. | n.t. |
| 28 | 3-OH | $5-\mathrm{CH}_2(\mathrm{CH}_2)_3\mathrm{CH}_3$ | 15 | $c.\mathrm{C}_{3}\mathrm{H}_{5}$ | >10 | n.a. | n.t. | n.t. |
| 29 | 3-OH | $5\text{-CH}_2(\text{CH}_2)_3\text{CH}_3$ | 15 | $p.\mathrm{OH}\text{-}\mathrm{C}_6\mathrm{H}_4$ | >10 | n.a. | n.t. | n.t. |
| 30 | 3-OH | H | 5 | $\mathrm{CH_{2}CH_{2}OH}$ | n.t. | n.t. | n.a. | n.t. |
| 31 | 3-OH | H | 5 | $c.\mathrm{C}_{3}\mathrm{H}_{5}$ | n.t. | n.t. | n.a. | n.t. |
| 32 | 3-OH | H | 5 | $p.\mathrm{OH}	ext{-}\mathrm{C}_6\mathrm{H}_4$ | n.t. | n.t. | n.t. | n.t. |
| 33 | 3-OH | H | 10 | $\mathrm{CH_{2}CH_{2}OH}$ | >10 | n.a. | n.a. | n.t. |
| 34 | 3-OH | H | 10 | $c.\mathrm{C_3H_5}$ | >10 | 5.4 | n.a. | n.t. |
| 35 | 3-OH | H | 10 | $p.OH-C_6H_4$ | n.a. | >10 | n.t. | n.t. |
| 36 | 3-OH | H | 15 | CH_2CH_2OH | >10 | n.a. | n.t. | n.t. |
| 37 | 3-OH | H | 15 | $c.\mathrm{C_3H_5}$ | 4.25 | n.a. | n.t. | n.t. |
| 38 | 3-OH | H | 15 | $p.OH-C_6H_4$ | 5 | 10 | n.t. | n.t. |
| 39 | 3-OH | 4-CH ₂ (CH ₂) ₄ CH ₃ | 5 | CH ₂ CH ₂ OH | n.t. | 6.43 | n.a. | n.t. |
| 40 | 3-OH | 4-CH ₂ (CH ₂) ₄ CH ₃ | 5 | $c.\mathrm{C_3H_5}$ | 0.18 | 0.54 | n.a. | n.t. |
| 41 | 3-OH | 4-CH ₂ (CH ₂) ₄ CH ₃ | 5 | p.OH-C ₆ H ₄ | n.t. | n.t. | n.t. | n.t. |
| 41 42 | 3-OH | 4-CH ₂ (CH ₂) ₄ CH ₃ 4-CH ₂ (CH ₂) ₄ CH ₃ | 10 | $_{ m CH_2CH_2OH}^{ m p.OH-C_6H_4}$ | >10 | 2.7 | | |
| 42 43 | 3-OH | | | | >10 | 0.35 | n.a. | n.a. >25 |
| | | $4-\mathrm{CH}_2(\mathrm{CH}_2)_4\mathrm{CH}_3$ | 10 | $c.\mathrm{C_3H_5}$ | | | n.a. | |
| 44 | 3-OH | $4\text{-CH}_2(\text{CH}_2)_4\text{CH}_3$ | 10 | $p.\mathrm{OH-C_6H_4}$ | n.a. | >10 | 50 | >25 |
| 45 | 3-OH | $4\text{-CH}_2(\text{CH}_2)_4\text{CH}_3$ | 15 | CH_2CH_2OH | n.a. | n.a. | n.t. | n.t. |
| 46 | 3-OH | $4-CH_2(CH_2)_4CH_3$ | 15 | $c.\mathrm{C}_{3}\mathrm{H}_{5}$ | n.a. | n.a. | n.t. | n.t. |
| 47 | 3-OH | $4-\mathrm{CH}_2(\mathrm{CH}_2)_4\mathrm{CH}_3$ | 15 | $p.\mathrm{OH}	ext{-}\mathrm{C}_6\mathrm{H}_4$ | >10 | n.a. | n.t. | n.t. |
| 48 | 5-OH | $2\text{-CH}_2(\text{CH}_2)_4\text{CH}_3$ | 5 | $\mathrm{CH_{2}CH_{2}OH}$ | n.t. | >10 | n.a. | n.t. |
| 49 | 5-OH | $2\text{-CH}_2(\text{CH}_2)_4\text{CH}_3$ | 5 | $c.\mathrm{C_3H_5}$ | $\mathbf{n.t.}$ | 3.57 | n.a. | n.t. |
| 50 | 5-OH | $2-CH_2(CH_2)_4CH_3$ | 5 | $p.OH-C_6H_4$ | n.t. | n.t. | n.t. | n.t. |
| 51 | 5-OH | $2-CH_2(CH_2)_4CH_3$ | 10 | $\mathrm{CH_{2}CH_{2}OH}$ | 1.13 | 0.42 | n.a. | >25 |
| 52 | 5-OH | $2\text{-CH}_2(\text{CH}_2)_4\text{CH}_3$ | 10 | $c.\mathrm{C_3H_5}$ | 0.21 | 0.03 | >25 | 13 |
| 53 | 5-OH | $2-CH_2(CH_2)_4CH_3$ | 10 | $p.OH-C_6H_4$ | 3.3 | 2.11 | 18.0 | >25 |
| 54 | 5-OH | $2 \cdot \text{CH}_2(\text{CH}_2)_4 \text{CH}_3$ $2 \cdot \text{CH}_2(\text{CH}_2)_4 \text{CH}_3$ | 15 | CH ₂ CH ₂ OH | n.a | n.a. | n.t. | n.t. |
| 55 | 5-OH | 2-CH ₂ (CH ₂) ₄ CH ₃ 2-CH ₂ (CH ₂) ₄ CH ₃ | 15 | $c.C_3H_5$ | 5.5 | 2.49 | n.t. | n.t. |
| 56 | 5-OH 5-OH | $2-\text{CH}_2(\text{CH}_2)_4\text{CH}_3$ $2-\text{CH}_2(\text{CH}_2)_4\text{CH}_3$ | 15 15 | $p.OH-C_6H_4$ | 3.5 | | n.t. | n.t. |
| anandami | | 2-С112(С112)4СП3 | 19 | $p.011-0611_4$ | $72\pm3.1~\mathrm{n~M}$ | n.a. | 11 | 11.6. |
| | | | | | | - 91 01 ~ M | | |
| WIN, 55- | 212 | | | | $21\pm1.1~\mathrm{nM}$ | $2.1 \pm 0.1 \mathrm{nM}$ | | |
| HU-210 | | | | | - | 0.15 ± 0.03 | | |

^a Data are means \pm SEM of n=3 separate experiments and are expressed as K_i (μ M), for CB₁ and CB₂ binding assays, and in IC₅₀ $(\mu \mathrm{M})$ for FAAH and AMT assays. Reference compounds were tested under the same conditions in this study. Anandamide was tested in the presence of PMSF (100 nM). n.a. = $IC_{50} > 10$ in the preliminary screening carried out with rat brain and spleen membranes; n.t. = not tested; insol. = insoluble in any of the solvents normally used in binding assays (DMSO, ethanol, or methanol). Binding affinity costants of the most potent compounds $(K_i \leq 1 \mu M)$ are highlighted in bold.

tion counting of the aqueous phase after extraction of the incubation mixture with 2 volumes of CHCl₃/CH₃-OH (2:1 by volume). In most cases, only the most potent compounds in the binding assays were subjected to this assav.

Anandamide Membrane Transporter (AMT) Assays. The effect of compounds on the uptake of anandamide by RBL-2H3 cells was studied as described previously.¹⁹ Cells were incubated with [¹⁴C]anandamide $(4 \mu M)$ for 5 min at 37 °C, in the presence or absence of varying concentrations of the inhibitors. Residual [14C]anandamide in the incubation media after extraction with CHCl₃/CH₃OH (2:1 by volume) was used as a measure of the anandamide that was taken up by cells. Data are expressed as the concentration exerting 50% inhibition of anandamide uptake (IC₅₀) calculated with Graph-Pad. In most cases, only the most potent compounds in the binding assays were subjected to this assay.

Cyclic AMP Assay. Cyclic AMP assays were performed on intact confluent N18TG2 cells plated in sixwell dishes and stimulated for 10 min at 37 °C with forskolin 1 µM in 400 µL of serum-free Dulbecco's modified Eagle's Medium containing 20 mM HEPES. 0.1 mg/mL BSA, 0.1 mM 1-methyl-3-isobutylxanthine.²⁰ Cells were treated with vehicle (methanol, 0.1%) or compounds, or WIN55,212-2 at various concentrations, or with SR141716A (100 nM). After incubation, 800 μ L of ethanol was added, cells were extracted and cyclic AMP was determined by means of a cyclic AMP assay kit (Amersham, UK), as advised by the manufacturer.

Results and Discussion

All the newly synthesized compounds that exibited $IC_{50} \leq 10 \ \mu M$ in the preliminary screening for cannabinoid receptor binding activity were evaluated in radioligand binding assays for affinity at recombinant human CB₁, CB₂ overexpressed in COS cells, or for their inhibitory actions on FAAH and AMT, and the results are summarized in Table 1. Our preliminary goal was to evaluate the hypothesis that putting together in a

Figure 4.

single structure the chemical stability of THC with the flexibility of AEA might lead to compounds active on the endocannabinoid systhem. To this end we have introduced an aliphatic chain carrying an amidic "head" in the aromatic structure of 3-pentadecylphenol, olivetol, resorcinol, and 4-hexylresorcinol, achieving five series of O-alkylated derivatives from which we could get useful information on structure—activity relationships. In fact, the binding assay results of the compounds described here allow us to establish the effect of several factors on the affinity for CB₁ and CB₂ receptors, and the most important points can be summarized as follows: (a) the presence of a phenolic hydroxyl group seems to play an essential role, since without it the 3-pentadecyphenol derivatives 15-17 are unable to bind to cannabinoid receptors with high affinity; (b) the length of aliphatic chain on the aromatic ring has a crucial influence on the affinity of our compounds, since a chain of five or six carbon atoms, such as in olivetol and 4-hexylresorcinol, is required, while a longer chain, as in 3-pentadecylphenol, or its absence, as in resorcinol, leads respectively to insoluble (18-20) or inactive (30-38) compounds; (c) regarding the alkyloxy chain carrying the amidic "head", its length also turned out to be of great importance: in fact, the short, five-carbon atom derivatives (15-17, 21-23, 30-32, 39-41, and 48-50) or the long fifteen carbon atom derivatives (27-29, 36-**38**, 45-47, and 54-56) are inactive independently of their aromatic structure. In support to this analysis one can observe that the most potent compounds are olivetol (24, 25, and 26) and 4-hexylresorcinol derivatives (51 and 52), all of them with a chain of 10 carbon atoms; one notable exception to this rule was compound 40 which was more potent than compound 43 at CB₁, and equipotent at CB2, receptors; (d) in accordance with literature data, ¹³ cyclopropylamides (25, 43, and 52) are more potent than the respective ethanolamides (24, 42, and 51), and in particular special attention should be given to compound 25 (Figure 4), which exhibits the lowest K_i values and behaves as a very potent CB₁ (5.2) nM) and CB₂ ligand (13 nM), with affinities similar to WIN55-212-2 (CB₁ 21 \pm 1.1 nM and CB₂ 2.1 \pm 0.1 nM); (e) furthermore, it appears interesting to compare two isomers of 4-hexylresorcinol: while compound 52 is a potent CB₂ ligand (30 nM), although not very selective over CB₁ receptors (210 nM), its regioisomer 43, even though less potent, resulted in a CB2 selective ligand with at least a 26-fold selectivity over CB₁ receptors (CB₂ 0.35 μ M and CB₁ > 10 μ M); (f) regarding the interaction of the most potent cannabinoid receptor ligands with FAAH and the AMT, the data obtained here show that all such compounds are essentially unable to bind to either the enzyme or the putative transporter.

To gain some insight into the possible use as a therapeutic agent of the most potent ligand described here, i.e., compound **25**, we carried out functional assays

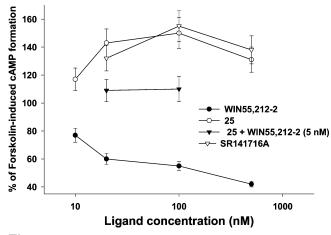


Figure 5.

for its activity at CB₁ receptors. We assessed the effect of this compound on forskolin-induced adenylate cyclase activity in mouse neuroblastoma N18TG2 cells, which constitutively and selectively express cannabinoid CB₁ receptors. As shown in Figure 5, similar to the wellknown CB₁ inverse agonist/antagonist, SR141716A, compound 25 significantly stimulated forskolin-induced cyclic AMP formation (half-maximal effect observed at a concentration of 13 nM similar to the K_i for this compound at CB₁ receptors). As expected, WIN55,212-2 behaved as an agonist since it inhibited forskolininduced cyclic AMP formation (half-maximal effect observed at a concentration of 15 nM). The effect of compound 25 at the doses of 20 and 100 nM was blocked by a per se inactive dose (5 nM) of WIN55,212 and hence was mediated by CB₁ receptors (Figure 5, higher doses of 25 could not be tested as they would have required non inactive doses of WIN55,212-2 to be counteracted). Therefore, this compound behaved as a inverse agonist/ antagonist at CB₁ receptors. Also another compound that exhibited affinity for CB₁ receptors, albeit lower than compound 25, i.e., compound 40, also behaved as a inverse agonists/antagonist (half-maximal effect observed at a concentration of 150 nM, data not shown). We also attempted at performing functional assays for CB₂ receptors in cell lines constitutively expressing this receptor but could not get a response even with WIN55,-212-2 and, therefore, did not go further.

Conclusions

Our hypothesis that compounds active at cannabinoid receptors can be obtained by linking a stable and rigid structure typical of aromatic compounds, such THC, olivetol, and its analogues, to a flexible chain carrying an amidic "head", as in AEA, appeared to be correct and resulted in the synthesis of a new class of compounds. Some of the new compounds exhibited potent activity at CB₁ and CB₂ receptors and exhibited antagonist/ inverse agonist activity. As a result of these studies we are currently pursuing new experiments in order to further evaluate the functional activity of these molecules, to elucidate the requirements involved in the recognition and binding of their biological targets, and to evaluate their pharmacological activity in vivo, to direct our research toward the design and synthesis of ever more potent and selective compounds.

Experimental Section

Melting points were determined on a Kofler hot stage apparatus (K) or using a Mettler FPI apparatus (2 °C/min) (M) and are uncorrected. Elemental analyses of all synthesized compounds were performed by our analytical laboratory in a Perkin-Elmer elemental apparatus Mod. 240 for C, H, N and the data are within $\pm 0.4\%$ of the theoretical values. ¹H NMR spectra were recorded at 25 °C on a Brucker AC200F employing TMS as internal standard, while NOESY experiments have been performed with a Varian VRX 300 MHz and chemical shifts are expressed as δ (ppm). Mass spectral data were determined by direct insertion at 70 eV with a VG70 spectrometer. All compounds were checked for purity by TLC on Merck 60 F₂₅₄ silica plates. For column chromatography, Merck 60 silica gel, 230-400 mesh, was used. Final products were purified by a Biotage flash chromatography system with columns 12.25 mm, packed with KP-Sil, 60A, 32-63 μ M. Reagents were purchased from Sigma Aldrich Srl (Italy) and used as received unless otherwise stated.

General Method for Esters. A mixture of phenolic compound (5.0 mmol), anhydrous potassium carbonate (2.5 mmol), and potassium fluoride (5.0 mmol) in dry acetone was refluxed under nitrogen atmosphere and continuously stirred for half hour, and then a solution of the corresponding bromoalkylmethyl ester (5.0 mmol) in dry acetone was added, refluxing for another 48-72 h and checking the reaction by TLC. Afterward the reaction mixture was concentrated, diluted with water, and extracted with chloroform. The extracts were collected, dried, and evaporated under reduced pressure, and the residue was purified by column chromatography on silica gel. Yields, melting points, and chemical and spectroscopic data of all esters are available in the Supporting Information.

General Methods for the Final Products. Method A. Each ester (1.0 mmol) was refluxed in a methanolic/aqueous sodium hydroxide solution (0.1 M, 3 equiv) for 3 h. Then the reaction mixture was allowed to get room temperature, made acidic (pH 3-4) by adding diluted HCl, and finally extracted with ethyl acetate. The organic layer was dried and the solvent evaporated to yield the crude acid, which was thoroughly dried under vacuum before being subjected to the successive reaction without further manipulation. To a mixture of the crude acid (1.0 mmol), the appropriate amine (1.5 mmol), and 1-hydroxybenzotriazole (HOBt, 1.2 mmol) in dry dichloromethane or in dry acetonitrile, kept in an ice bath, was added dropwise a solution, in the same solvent, of 1-cyclohexyl-3-(2-morpholinoethyl)carbodiimide methyl p-toluensulfonate (CMC, 1.5 mmol) under nitrogen atmosphere and continuous stirring. The reaction mixture was left to room temperature, and stirring was continued 24 h. The solvent was removed under reduced pressure, and the residue was diluted with chloroform. The organic layer was washed with 5% aqueous NaHCO3 and then with 1 N HCl and dried. After drying and evaporation of the solvent, a crude residue was obtained and purified by column chromatography on silica gel.

Method B. Each ester (0.150 g) was dissolved, under nitrogen atmosphere and stirring, in ethanolamine (4 mL) and warmed at 120-130 °C for 4-6 h. The resulting mixture was diluted with water and extracted with chloroform, the extracts were washed with a solution of ammonium chloride, dried, evaporated, and the raw material was purified by column chromatography on silica gel.

6-(3-Pentadecylphenoxy)hexanoic Acid (2-Hydroxyethyl)amide (15).²¹ White solid (CHCl₃/MeOH = 47/3) (80% yield): mp 62.7 °C (M). 1H NMR (CDCl₃) δ (ppm): 7.14–7.10 (m, 1H), 6.75-6.66 (m, 3H), 5.95 (s br, 1H), 3.93 (t, 2H, J = 0.00)6.3 Hz), 3.70 (t, 2H, J = 4.9 Hz), 3.44-3.36 (m, 2H), 2.65 (s br, 1H), 2.54 (t, 2H, J = 7.7 Hz), 2.22 (t, 2H, J = 7.3 Hz), 1.81– 1.46 (mm, 8H), 1.44-1.24 (mm, 24H) 0.86 (t, 3H, J=6.2 Hz).Anal. (C₂₉H₅₁NO₃) C, H, N.

6-(3-Pentadecylphenoxy)hexanoic Acid Cyclopropylamide (16). White solid (CHCl₃, recrystallized from acetone/ ethyl ether) (50% yield): mp 67.8 °C (M). 1 H NMR (CDCl₃) δ (ppm): 6.65 (t, 1H, J = 7.7 Hz), 6.50 (s br, 1H), 6.26-6.17 (m, 3H), 3.46 (t, 2H, J = 6.3 Hz), 2.28-2.18 (m, 1H), 2.08 (t, 2H, J = 7.5 Hz), 1.69 (t, 2H, J = 7.3 Hz), 1.36–1.03 (mm, 8H), 1.0-0.79 (mm, 24H), 0.40 (t, 3H, J = 6.3 Hz), 0.24-0.15 (m, 2H), 0.11-0.02 (m, 2H). MS m/z: 458 [M + 1]⁺ (100). Anal. $(C_{30}H_{51}NO_2)$ C, H, N.

6-(3-Pentadecylphenoxy)hexanoic Acid (4-Hydroxy**phenyl)amide (17).** White solid (CHCl₃/MeOH = 47/3) (70% yield): mp 103.2 °C (M). 1 H NMR (CDCl₃) δ (ppm): 7.31 (half of AB_q, 2H, J = 8.7 Hz), 7.20-7.12 (m, 2H), 7.00 (s br, 1H), 6.79-6.68 (m, 4H), 3.95 (t, 2H, J = 7.1 Hz), 3.92 (s br, 1H), 2.55 (t, 2H, J = 7.6 Hz), 2.36 (t, 2H, J = 7.0 Hz), 1.80-1.69(m, 4H), 1.63-1.53 (m, 4H), 1.25-1.18 (mm, 24H), 0.87 (t, 3H, 24H),J = 6.6 Hz). MS m/z: 510 [M + 1]⁺, 1019 [2M + 1]⁺ (100). Anal. $(C_{33}H_{51}NO_3)$ C, H, N.

11-(3-Pentadecylphenoxy)undecanoic Acid (2-Hydroxyethyl)amide (18). White solid (CHCl₃/MeOH = 47/3) (75% yield): mp 83-85 °C (K). ¹H NMR (CDCl₃) δ (ppm): 7.18-7.11 (m, 1H), 6.74-6.67 (m, 3H), 5.85 (s br, 1H), 3.92 (t, 2H, J = 6.4 Hz), 3.71 (t, 2H, J = 4.7 Hz), 3.44–3.36 (m, 2H), 2.60 (s br, 1H), 2.55 (t, 2H, J = 7.5 Hz), 2.18 (t, 2H, J = 7.5 Hz), 1.78-1.51 (mm, 6H), 1.28-1.24 (mm, 36H), 0.86 (t, 3H, J =6.3 Hz). Anal. (C₃₄H₆₁NO₃) C, H, N.

11-(3-Pentadecylphenoxy)undecanoic Acid Cyclopropylamide (19). White needles (CHCl3, recrystallized from acetone) (91% yield): mp 82.0 °C (K). ¹H NMR (CDCl₃) δ (ppm): 7.15 (t, 1H, J = 8.1 Hz), 6.75-6.67 (m, 3H), 5.46 (s br, $1\overline{\text{H}}$), 3.92 (t, 2H, J = 6.5 Hz), 2.71–2.64 (m, 1H), 2.55 (t, 2H, J = 7.6 Hz), 2.09 (t, 2H, J = 7.5 Hz), 1.78–1.41 (mm, 10H), 1.27-1.24 (mm, 32H), 0.86 (t, 3H, J = 6.2 Hz), 0.80-0.70 (m, 2H), 0.55-0.46 (m, 2H). MS m/z: 1077 [2M + Na]⁺ (100). Anal. $(C_{35}H_{61}NO_2)$ C, H, N.

11-(3-Pentadecylphenoxy)undecanoic Acid (4-Hydroxy**phenyl)amide (20).** White solid (CHCl₃/MeOH = 47/3) (86% yield): mp 107.0 °C (M). 1 H NMR (CDCl₃) δ (ppm): 7.33 (half of AB_q, 2H, J = 7.5 Hz), 7.21-7.12 (m, 2H), 7.01 (s br, 1H), 6.97-6.54 (m, 4H), 3.93 (t, 2H, J = 6.5 Hz), 2.56 (t, 2H, J =7.7 Hz), 2.35 (t, 2H, J = 7.4 Hz), 1.79–1.60 (m, 8H), 1.30– $1.22 \text{ (mm, 34H)}, 0.87 \text{ (t, 3H, } J = 6.6 \text{ Hz)}. \text{ MS: } 579 \text{ [M]}^+, 1159$ $[2M + 1]^+$ (100). Anal. ($C_{38}H_{61}NO_3$) C, H, N.

6-(3-Hydroxy-5-pentylphenoxy)hexanoic Acid (2-Hy**droxyethyl)amide (21).** Pink oil (ethyl acetate/MeOH = 46/ 4) (65% yield). ¹H NMR (Acetone- d_6) δ (ppm): 8.43 (s, 1H), 7.33 (s br, 1H), 6.25-6.21 (m, 3H), 4.18 (s br, 1H), 3.86 (t, 2H, J = 6.3 Hz), 3.60-3.57 (m, 2H), 3.37-3.28 (m, 2H), 2.44 (t, 2H, J = 7.6 Hz), 2.22 (t, 2H, J = 7.3 Hz), 1.70 - 1.29 (mm, 12H), 0.85 (t, 3H, J = 6.5 Hz). Anal. ($C_{19}H_{31}NO_4$) C, H, N.

6-(3-Hydroxy-5-pentylphenoxy)hexanoic Acid Cyclo**propylamide (22).** Pale yellow oil (CHCl₃/MeOH = 48/2) (60%). ¹H NMR (CDCl₃) δ (ppm): 7.04 (s br, 1H), 6.29–6.24 (m, 3H), 5.87 (s br, 1H), 3.86 (t, 2H, J = 6.2 Hz), 2.74-2.65 (m, 1H), 2.46 (t, 2H, J = 7.6 Hz), 2.14 (t, 2H, J = 7.3 Hz), 1.75-1.24 (mm, 12H), 0.86 (t, 3H, J = 6.5 Hz), 0.78-0.69 (m, 2H), 0.56-0.47 (m, 2H). Anal. (C₂₀H₃₁NO₃) C, H, N.

6-(3-Hydroxy-5-pentylphenoxy)hexanoic Acid (4-Hydroxyphenyl)amide (23). White solid (CHCl₃/MeOH = 45/ 5) (75% yield): mp 69.2 °C (M). ¹H NMR (CDCl₃) δ (ppm): 7.33 (half of AB_q , 2H, J = 8.7 Hz), 7.02 (s br, 1H), 6.80 (half of AB_q , 2H, J = 8.7 Hz), 6.29-6.21 (m, 3H), 4.95 (s br, 1H), 3.92 (t,2H, J = 6.2 Hz), 2.49 (t, 2H, J = 7.6 Hz), 2.36 (t, 2H, J = 7.2 Hz)Hz), 1.83-1.76 (m, 4H), 1.61-1.53 (m, 4H), 1.33-1.25 (mm, 4H), 0.88 (t, 3H, J = 6.5 Hz). MS m/z: 386 [M + 1]⁺ (100), 771 $[2M + 1]^+$, 793 $[2M + Na]^+$. Anal. $(C_{23}H_{31}NO_4)$ C, H, N.

11-(3-Hydroxy-5-pentylphenoxy)undecanoic Acid (2-Hydroxyethyl)amide (24). Pasty pink solid (ethyl acetate/ MeOH = 44/6) (70% yield): mp 55–56 °C (K). ¹H NMR (DMSO) δ (ppm): 9.34 (s, 1H), 7.68 (s br, 1H), 6.13-6.09 (m, 3H), 4.57 (t, 1H, J = 5.3 Hz), 3.82 (t, 2H, J = 6.3 Hz), 3.39-3.30 (m, 2H), 3.11-3.03 (m, 2H), 2.39 (t, 2H, J=7.5 Hz), 2.02(t, 2H, J = 7.2 Hz), 1.63 - 1.45 (mm, 6H), 1.41 - 1.23 (mm, 16H),0.83 (t, 3H, J = 6.4 Hz). Anal. ($C_{24}H_{41}NO_4$) C, H, N.

11-(3-Hydroxy-5-pentylphenoxy)undecanoic Acid Cy**clopropylamide (25).** White solid (CHCl₃/MeOH = 47/3) (61% yield): mp 70.7 °C (M). ¹H NMR (CDCl₃) δ (ppm): 6.98 (s br, 1H), 6.28-6.26 (m, 3H), 5.79 (s br, 1H), 3.88 (t, 2H, J=6.3Hz), 2.72-2.67 (m, 1H), 2.47 (t, 2H, J = 7.6 Hz), 2.12 (t, 2H, $J=7.5~\rm{Hz}),\,1.75-1.49~\rm{(mm,\,8H)},\,1.28-1.26~\rm{(mm,\,14H)},\,0.87$ (t, 3H, $J=6.4~\rm{Hz}),\,0.79-0.70~\rm{(m,\,2H)},\,0.55-0.43~\rm{(m,\,2H)}.$ Anal. (C25H41NO3) C, H, N.

11-(3-Hydroxy-5-pentylphenoxy)undecanoic Acid (4-Hydroxyphenyl)amide (26). Pasty white solid (CHCl₃/MeOH = 47/3) (65% yield). ¹H NMR (CDCl₃) δ (ppm): 7.28 (half of AB_q, 2H, J = 9.0 Hz), 7.15 (s br, 1H), 6.74 (half of AB_q, 2H, J = 8.8 Hz), 6.29–6.25 (m, 3H), 3.90 (t, 2H, J = 6.4 Hz), 2.50 (t, 2H, J = 7.6 Hz), 2.32 (t, 2H, J = 7.4 Hz), 1.73–1.53 (m, 8H), 1.33–1.29 (m, 14H), 0.87 (t, 3H, J = 6.6 Hz). MS m/z: 456 [M + 1]⁺ (100), 911 [2M + 1]⁺, 933 [2M + Na]⁺. Anal. (C₂₈H₄₁NO₄) C, H, N.

16-(3-Hydroxy-5-pentylphenoxy)hexadecanoic Acid (2-Hydroxyethyl)amide (27). White solid (ethyl acetate/MeOH = 50/5; recrystallized from ethyl acetate) (57% yield): mp 84–85 °C (K). $^1\mathrm{H}$ NMR (DMSO) δ (ppm): 9.17 (s, 1H), 7.72 (s br, 1H), 6.17–6.13 (m, 3H), 4.61 (t, 1H, J=5.1 Hz), 3.86 (t, 2H, J=6.1 Hz), 3.40–3.33 (m, 2H), 3.15–3.09 (m, 2H), 2.42 (t, 2H, J=7.3 Hz), 2.06 (t, 2H, J=7.2 Hz), 1.66–1.52 (mm, 8H), 1.48–1.25 (mm, 24H), 0.87 (t, 3H, J=6.3 Hz). Anal. (C₂₉H_{51-NO₄}) C, H, N.

16-(3-Hydroxy-5-pentylphenoxy)hexadecanoic Acid Cyclopropylamide (28). White solid (CHCl₃/MeOH = 48/2) (88.5% yield): mp 91.0 °C (M). ¹H NMR (CDCl₃) δ (ppm): 6.29–6.24 (m, 3H), 5.52 (s br, 1H), 3.89 (t, 2H, J = 6.4 Hz), 2.72–2.67 (m, 1H), 2.48 (t, 2H, J = 7.6 Hz), 2.11 (t, 2H, J = 7.5 Hz), 1.76–1.66 (mm, 8H), 1.57–1.24 (mm, 24H), 0.87 (t, 3H, J = 6.5 Hz), 0.80–0.71 (m, 2H), 0.55–0.42 (m, 2H). Anal. (C₃₀H₅₁NO₃) C, H, N.

16-(3-Hydroxy-5-pentylphenoxy)hexadecanoic Acid (4-Hydroxyphenyl)amide (29). White solid (CHCl₃/MeOH = 46/4) (83.0% yield): mp 89.5 °C (M). ¹H NMR (DMSO) δ (ppm): 9.50 (s, 1H), 9.30 (s, 1H), 9.12 (s br, 1H), 7.29 (half of AB_q, 2H, J=8.4 Hz), 6.63 (half of AB_q, 2H, J=8.6 Hz), 6.12–6.07 (m, 3H), 3.80 (t, 2H, J=6.3 Hz), 2.37 (t, 2H, J=7.4 Hz), 2.14 (t, 2H, J=7.6 Hz), 1.70–1.15 (mm, 32H), 0.81 (t, 3H, J=6.4 Hz). Anal. (C₃₃H₅₁NO₄) C, H, N.

6-(3-Hydroxyphenoxy)hexanoic Acid (2-Hydroxyethyl)amide (30). Pale pink oil (ethyl acetate/MeOH = 50/2) (50.0% yield). 1 H NMR (CDCl₃) δ (ppm): 8.70 (s br, 1H), 7.41 (s br, 1H), 7.02 (t, 1H, J=8.3 Hz), 6.97–6.33 (m, 3H), 4.31 (s br, 1H), 3.87 (t, 2H, J=6.3 Hz), 3.62–3.57 (m, 2H), 3.35–3.27 (m, 2H), 2.23 (t, 2H, J=7.2 Hz), 1.74–1.58 (m, 4H), 1.50–1.43 (m, 2H). MS m/z: 268 [M + 1]⁺ (100), 290 [M + Na]⁺. Anal. (C₁₄H₂₁NO₄) C, H, N.

6-(3-Hydroxyphenoxy)hexanoic Acid Cyclopropylamide (31). White needles (CHCl₃/MeOH = 47/3) (86% yield): 102.5 °C (M). ¹H NMR (CDCl₃) δ (ppm): 7.12–7.04 (m, 1H), 6.43–6.40 (m, 3H), 6.11 (s, 1H), 5.63 (s br, 1H), 3.90 (t, 2H, J = 6.1 Hz), 2.74–2.67 (m, 1H), 2.14 (t, 2H, J = 7.3 Hz), 1.82–1.62 (mm, 4H), 1.55–1.37 (m, 2H), 0.80–0.70 (m, 2H), 0.50–0.47 (m, 2H). MS: 264 [M + 1]⁺, 527 [2M + 1]⁺, 549 [2M + Na]⁺ (100). Anal. (C₁₅H₂₁NO₃) C, H, N.

6-(3-Hydroxyphenoxy)hexanoic Acid (4-Hydroxyphenyl)amide (32). Pale pink solid (CHCl₃/MeOH = 45/5; recrystallized from ethyl acetate) (88.0% yield): 123.1 °C (M). ¹H NMR (DMSO) δ (ppm): 9.53 (s, 1H), 9.26 (s, 1H), 9.05 (s, 1H), 7.31 (half of AB_q, 2H, J=8.8 Hz), 6.95 (t, 1H, J=8.1 Hz), 6.64 (half of AB_q, 2H, J=8.7 Hz), 6.31–6.27 (m, 3H), 3.85 (t, 2H, J=6.3 Hz), 2.23 (t, 2H, J=7.2 Hz), 1.74–1.53 (m, 4H), 1.46–1.36 (m, 2H). Anal. (C₁₈H₂₁NO₄) C, H, N.

11-(3-Hydroxyphenoxy)undecanoic Acid (2-Hydroxyethyl)amide (33). Cream solid (Ethyl acetate/MeOH = 50/4) (34% yield): mp 72.1 °C (M). $^1\mathrm{H}$ NMR (CDCl_3) δ (ppm): 7.10 (t, 1H, J=8.0 Hz), 6.44–6.38 (m, 3H), 6.15 (s br, 1H), 5.96 (s br, 1H), 3.92 (t, 2H, J=6.2 Hz), 3.74–3.70 (m, 2H), 3.46–3.38 (m, 2H), 2.58 (s br, 1H), 2.19 (t, 2H, J=7.5 Hz), 2.02–1.56 (mm, 6H), 1.43–1.24 (mm, 10H). MS m/z: 338 [M + 1]+, 360 [M + Na]+, 697 [2M + Na]+ (100). Anal. (C $_{19}$ H $_{31}$ NO $_{4}$) C, H, N.

11-(3-Hydroxyphenoxy)undecanoic Acid Cyclopropilamide (34). White solid (Ethyl acetate) (90% yield): mp 92.8 °C (M). ¹H NMR (CDCl₃) δ (ppm): 7.08 (t, 1H, J=8.2 Hz), 6.50 (s br, 1H), 6.47–6.39 (m, 3H), 5.60 (s br, 1H), 3.91 (t, 2H,

J=6.3 Hz), 2.76-2.66 (m, 1H), 2.11 (t, 2H, J=7.2 Hz), 1.79-1.60 (mm, 4H), 1.46-1.27 (mm, 12H), 0.80-0.70 (m, 2H), 0.48-0.42 (m, 2H). MS $\emph{m/z}\colon 334$ [M + 1]+, 667 [2M + 1]+ (100), 689 [2M + Na]+. Anal. (C_{20}H_{31}NO_3) C, H, N.

11-(3-Hydroxyphenoxy)undecanoic Acid (4-Hydroxyphenyl)amide (35). White solid (CHCl₃/MeOH = 45/5) (79% yield): mp 121.2 °C (M). ¹H NMR (DMSO) δ (ppm): 9.50 (s, 1H), 9.26 (s, 1H), 9.05 (s, 1H), 7.30 (half of AB_q, 2H, J=8.7 Hz), 6.98 (t, 1H, J=8.0 Hz), 6.63 (half of AB_q, 2H, J=8.6 Hz), 6.31–6.26 (m, 3H), 3.83 (t, 2H, J=6.4 Hz), 2.19 (t, 2H, J=7.3 Hz), 1.62–1.56 (m, 4H), 1.53–1.24 (mm, 12H). Anal. ($C_{23}H_{31}NO_4$) C, H, N.

16-(3-Hydroxyphenoxy)hexadecanoic Acid (2-Hydroxyethyl)amide (36). White solid (Ethyl acetate/MeOH = 50/1) (35% yield): mp 94.5 °C (M). ¹H NMR (DMSO) δ (ppm): 9.51 (s, 1H), 7.91 (s br, 1H), 7.44 (t, 1H, J=8.2 Hz), 6.81–6.75 (m, 3H), 4.80 (t, 1H, J=5.5 Hz), 4.33 (t, 2H, J=6.4 Hz), 3.94–3.85 (m, 2H), 3.66–3.57 (m, 2H), 2.54 (t, 2H, J=7.0 Hz), 2.18–2.10 (m, 4H), 2.00–1.70 (mm, 22H). Anal. (C₂₄H₄₁NO₄) C, H, N

16-(3-Hydroxyphenoxy)hexadecanoic Acid Cyclopropylamide (37). White bright solid (Ethyl acetate/MeOH = 50/4; recrystallized Ethyl acetate) (78% yield): mp 105.6 °C (M). $^1\mathrm{H}$ NMR (DMSO) δ (ppm): 9.25 (s, 1H), 7.73 (s br, 1H), 6.98 (t, 1H, J=7.9 Hz), 6.31–6.26 (m, 3H), 3.83 (t, 2H, J=6.4 Hz), 2.58–2.51 (m, 1H), 1.94 (t, 2H, J=7.3 Hz), 1.64–1.59 (m, 2H), 1.41–1.20 (mm, 24H), 0.58–0.49 (m, 2H), 0.40–0.26 (m, 2H). Anal. (C25H41NO3) C, H, N.

16-(3-Hydroxyphenoxy)hexadecanoic Acid (4-Hydroxyphenyl)amide (38). White solid (CHCl₃/MeOH = 50/5) (71.5% yield): 123.6 °C (M). ¹H NMR (DMSO) δ (ppm): 9.57 (s, 1H), 9.32 (s,1H), 9.05 (s, 1H), 7.36 (half of AB_q, 2H, J = 8.6 Hz), 7.04 (t, 1H, J = 7.9 Hz), 6.69 (half of AB_q, 2H, J = 8.6 Hz), 6.37–6.32 (m, 3H), 3.89 (t, 2H, J = 6.3 Hz), 2.24 (t, 2H, J = 7.3 Hz), 1.69–1.57 (m, 4H), 1.50–1.27 (mm, 22H). Anal. ($C_{28}H_{41}NO_4$) C, H, N.

6-(3-Hydroxy-4-hexylphenoxy)hexanoic Acid (2-Hydroxyethyl)amide (39). Pale yellow solid (CHCl₃/MeOH = 46/4) (55% yield): mp 71.2 °C (M). ¹H NMR (Acetone- d_6) δ (ppm): 8.42 (s, 1H), 7.42 (s br, 1H), 6.93 (d, 1H, J=8.0 Hz), 6.45 (d, 1H, J=2.0 Hz), 6.32 (dd, 1H, J=2.0 Hz, J=8.0 Hz), 4.30 (t, 1H, J=5.2 Hz), 3.87 (t, 2H, J=6.3 Hz), 3.67-3.59 (m, 2H), 3.40-3.31 (m, 2H), 2.54 (t, 2H, J=7.5 Hz), 2.27 (t, 2H, J=7.3 Hz), 1.76-1.65 (m, 4H), 1.61-1.42 (mm, 4H), 1.32-1.30 (m, 6H), 0.88 (t, 3H, J=6.1 Hz). MS m/z: 352 [M + 1]+, 374 [M + Na]+ (100), 725 [2M + Na]+. Anal. (C₂₀H₃₃-NO₄) C, H, N.

6-(3-Hydroxy-4-hexylphenoxy)hexanoic Acid Cyclopropylamide (40). White bright solid (Ethyl acetate; recrystallized from n.hexane) (70% yield): mp 134.2 °C (M). ¹H NMR 300 MHz (DMSO) δ (ppm): 9.11 (s, 1H), 7.80 (d, 1H, J=3.6 Hz)), 6.85 (d, 1H, J=8.4 Hz), 6.31 (d, 1H, J=2.4 Hz), 6.24 (dd, 1H, J=2.4 Hz, J=8.4 Hz), 3.80 (t, 2H, J=6.6 Hz), 2.61–2.53 (m, 1H), 2.42 (t, 2H, J=7.3 Hz), 2.00 (t, 2H, J=7.1 Hz), 1.68–1.60 (m, 2H), 1.55–1.41 (m, 4H), 1.37–1.23 (mm, 8H), 0.83 (t, 3H, J=6.6 Hz), 0.60–0.53 (m, 2H), 0.36–0.31 (m, 2H). MS m/z: 348 [M + 1]+ (100), 695 [2M + 1]+. Anal. ($C_{21}H_{33}NO_3$) C, H, N.

6-(3-Hydroxy-4-hexylphenoxy)hexanoic Acid (4-Hydroxyphenyl)amide (41). White solid (CHCl₃/MeOH = 47/3) (79.5% yield): mp 128.4 °C. ¹H NMR (Acetone- d_6) δ (ppm): 8.82 (s, 1H), 8.00 (s br, 1H), 7.43 (half of ABq, 2H, J = 8.6 Hz), 6.92 (half of ABq, 2H, J = 8.6 Hz), 6.73 (d, 1H, J = 8.8 Hz), 6.40 (d, 1H, J = 2.7 Hz), 6.31 (dd, 1H, J = 2.5 Hz, J = 8.7 Hz), 3.87 (t, 2H, J = 6.4 Hz), 2.51 (t, 2H, J = 7.4 Hz), 2.32 (t, 2H, J = 7.2 Hz), 1.77-1.64 (m, 6H), 1.55-1.44 (m, 4H), 1.40-1.27 (m, 4H), 0.85 (t, 3H, J = 6.6 Hz). Anal. (C₂₄H₃₃NO₄) C, H, N.

11-(3-Hydroxy-4-hexylphenoxy)undecanoic Acid (2-Hydroxyethyl)amide (42). White solid (Ethyl acetate/MeOH = 50/4) (43% yield): mp 85.3 °C (M). ¹H NMR (CDCl₃) δ (ppm): 6.96 (d, 1H, J = 9.0 Hz), 6.40 – 6.35 (m, 2H), 5.96 (s br, 1H), 3.89 (t, 2H, J = 6.2 Hz), 3.73 (t, 2H, J = 5.0 Hz), 3.46 – 3.39 (m, 2H), 2.51 (t, 2H, J = 7.6 Hz), 2.21 (t, 2H, J = 7.5 Hz), 1.76 – 1.52 (mm, 8H), 1.45 – 1.22 (mm, 16H), 0.87 (t, 3H, J =

6.4 Hz). MS m/z: 422 [M + 1]⁺, 444 [M + Na]⁺ (100). Anal. (C₂₅H₄₃NO₄) C, H, N.

11-(3-Hydroxy-4-hexylphenoxy)undecanoic Acid Cyclopropylamide (43). White bright solid (Ethyl acetate; recrystallized from ethyl acetate/n.hexane) (81% yield): 112.3 °C (M). 1 H NMR 300 MHz (DMSO) δ (ppm): 9.10 (s, 1H), 7.77 (s br, 1H), 6.85 (d, 1H, J = 8.4 Hz), 6.30 (d, 1H, J = 2.4 Hz), $6.24 \, (dd, 1H, J = 2.4 \, Hz, J = 8.0 \, Hz), 3.81 \, (t, 2H, J = 6.4 \, Hz),$ 2.59-2.53 (m, 1H), 2.39 (t, 2H, J=7.4 Hz), 1.96 (t, 2H, J=7.4 Hz), J=7.7.4 Hz), 1.65–1.58 (m, 2H), 1.46–1.38 (m, 4H), 1.34–1.23 (mm, 18H), 0.83 (t, 3H, J = 6.4 Hz), 0.57-0.53 (m, 2H), 0.34-0.31(m, 2H). MS m/z: 418 [M + 1]⁺, 440 [M + Na]⁺ (100). Anal. (C₂₆H₄₃NO₃) C, H, N.

11-(3-Hydroxy-4-hexylphenoxy)undecanoic Acid (4-Hydroxyphenyl)amide (44). White solid (CHCl₃/MeOH = 48/2) (88.0% yield): mp 121.2 °C (M). ¹H NMR (DMSO) δ (ppm): 9.48 (s, 1H), 9.04 (s, 1H), 9.02 (s, 1H), 7.30 (half of AB_q, 2H, J = 8.8 Hz), 6.84 (d, 1H, J = 8.1 Hz), 6.63 (half of AB_q), 2H, J = 8.8 Hz), 6.29 (d, 1H, J = 2.1 Hz), 6.23 (dd, 1H, J = 2.1 Hz) 2.3 Hz, J = 8.2 Hz), 3.80 (t, 2H, J = 6.4 Hz), 2.38 (t, 2H, J = 6.4 Hz)7.4 Hz), 2.20 (t, 2H, J = 7.4 Hz), 1.65 - 1.40 (mm, 10H), 1.37 - 1.40 (mm, 10H)1.22 (mm, 14H), 0.82 (t, 3H, CH₃, J = 6.4 Hz). Anal. ($C_{29}H_{43}$ - NO_4) C, H, N.

16-(3-Hydroxy-4-hexylphenoxy)hexadecanoic Acid (2-Hydroxyethyl)amide (45). White solid (Ethyl acetate/MeOH = 50/1) (50% yield): mp 96.4 °C. ¹H NMR (DMSO) δ (ppm): 9.14 (s, 1H), 7.73 (s br, 1H), 6.88 (d, 1H, J = 8.3 Hz), 6.34 (d, 1.00 (d, 1.00 d)1H, J = 1.7 Hz, 6.27 (dd, 1H, J = 8.3 Hz, J = 1.7 Hz), <math>4.63 (t, J)1H, J = 5.2 Hz), 3.84 (t, 2H, J = 6.2 Hz), 3.57–3.40 (m, 2H), 3.15-3.06 (m, 2H), 2.43 (t, 2H, J = 7.2 Hz), 2.06 (t, 2H, J =7.3 Hz), 1.70-1.48 (mm, 8H), 1.44-1.25 (mm, 26H), 0.87 (t, 3H, J = 5.8 Hz). Anal. (C₃₀H₅₃NO₄) C, H, N.

16-(3-Hydroxy-4-hexylphenoxy)hexadecanoic Acid Cyclopropylamide (46). White bright solid (Ethyl acetate/ MeOH = 50/2; recrystallized from ethyl acetate) (94.5% yield): mp 114.3 °C (M). ¹H NMR (DMSO) δ (ppm): 9.15 (s, 1H), 7.80 (s br, 1H), 6.82 (d, 1H, J = 8.1 Hz), 6.34 (d, 1H, J =2.6 Hz), 6.20 (dd, 1H, J = 2.4 Hz, J = 8.3 Hz), 3.79 (t, 2H, J= 6.3 Hz), 2.55-2.48 (m, 1H), 2.37 (t, 2H, J = 7.4 Hz), 1.94 (t, 2H)2H, J = 7.3 Hz), 1.62-1.55 (m, 4H), 1.41-1.20 (mm, 30H), 0.82 (t, 3H, J = 6.0 Hz), 0.58 - 0.49 (m, 2H), 0.34 - 0.27 (m, 2H).Anal. (C₃₁H₅₃NO₃) C, H, N.

16-(3-Hydroxy-4-hexylphenoxy)hexadecanoic Acid (4-**Hydroxyphenyl)amide** (47). White solid (CHCl₃/MeOH = 45/5) (81% yield): mp 133.2 °C (M). $^1\mathrm{H}$ NMR (DMSO) δ (ppm): 9.50 (s, 1H), 9.06 (s, 1H), 9.04 (s, 1H), 7.30 (half of AB_q, 2H, J = 8.5 Hz), 6.83 (d, 1H, J = 8.0 Hz), 6.62 (half of AB_q, 2H, J = 8.5 Hz), 6.28 (s, 1H), 6.22 (d, 1H, J = 8.1 Hz), 3.80 (t, 2H, J = 6.2 Hz), 2.38 (t, 2H, J = 7.3 Hz), 2.18 (t, 2H, J = 7.1 Hz)Hz), 1.61–1.21 (mm, 34H), 0.81 (t, 3H, J=6.4 Hz). Anal. (C₃₄H₅₃NO₄) C, H, N.

6-(2-Hexyl-5-hydroxyphenoxy)hexanoic Acid (2-Hydroxyethyl)amide (48). Pale yellow thick oil (CHCl₃/MeOH = 46/4) (50% yield). ¹H NMR (Acetone- d_6) δ (ppm): 8.23 (s, 1H), 7.27 (s br, 1H), 6.88 (d, 1H, J = 8.0 Hz), 6.42 (d, 1H, J =2.0 Hz), 6.35-6.31 (m, 1H), 4.08 (s br, 1H), 3.91 (t, 2H, J =6.3 Hz), 3.61–3.58 (m, 2H), 3.36–3.28 (m, 2H), 2.50 (t, 2H, J = 7.4 Hz), 2.25 (t, 2H, J = 7.2 Hz), 1.85-1.67 (mm, 4H), 1.63-1.63 (mm, 4H)1.52 (m, 4H), 1.49 - 1.31 (mm, 6H), 0.88 (t, 3H, J = 6.4 Hz).MS m/z: 353 [M + D]⁺, 374 [M + Na]⁺, 725 [2M + Na]⁺ (100). Anal. (C₂₀H₃₃NO₄) C, H, N.

6-(2-Hexyl-5-hydroxyphenoxy)hexanoic Acid Cyclo**propylamide (49).** Pale yellow thick oil (Ethyl acetate) (80% yield). ¹H NMR 300 MHz (DMSO) δ (ppm): 9.06 (s, 1H), 7.80 (d, 1H, J = 3.9 Hz), 6.81 (d, 1H, J = 8.1 Hz), 6.30 (d, 1H, J = 8.1 Hz) $2.3~{\rm Hz}$), $6.21~{\rm (dd,~1H,}~J=2.2~{\rm Hz},~J=8.1~{\rm Hz}$), $3.82~{\rm (t,~2H,}~J=8.1~{\rm Hz}$ = 6.2 Hz), 2.61-2.52 (m, 1H), 2.38 (t, 2H, J = 7.1 Hz), 2.00 (t, 2H)2H, J = 7.1 Hz), 1.71 - 1.62 (m, 2H), 1.56 - 1.47 (m, 2H), 1.42 -1.34 (m, 4H), 1.27-1.23 (mm, 6H), 0.82 (t, 3H, J = 6.6 Hz),0.58-0.52 (m, 2H), 0.35-0.30 (m, 2H). MS m/z: 348 [M + 1]⁺ (100), 695 [2M + 1] $^+$. Anal. ($C_{21}H_{33}NO_3$) C, H, N.

6-(2-Hexyl-5-hydroxyphenoxy)hexanoic Acid (4-Hydroxyphenyl)amide (50). White solid ($CHCl_3/MeOH = 47/$ 3) (63% yield): mp 106.0 °C (M). 1 H NMR (Acetone- d_{6}) δ (ppm): 9.52 (s, 1H), 9.04 (s, 1H), 9.02 (s, 1H), 7.30 (half of AB_a, 2H, J = 8.8 Hz), 6.80 (d, 1H, J = 8.1 Hz), 6.62 (half of AB_q, 2H, J = 8.7 Hz), 6.28 (s, 1H), 6.20 (d, 1H, J = 8.0 Hz), 3.83 (t, 1H)2H, J = 6.0 Hz), 2.36 (t, 2H, J = 7.3 Hz), 2.23 (t, 2H, J = 7.1Hz), 1.70–1.57 (mm, 10H), 1.53–1.20 (m, 4H), 0.81 (t, 3H, J = 6.7 Hz). Anal. $(C_{24}H_{33}NO_4) \text{ C, H, N}$.

11-(2-Hexyl-5-hydroxyphenoxy)undecanoic Acid (2-Hydroxyethyl)amide (51). Pale yellow solid (Ethyl acetate/ MeOH = 50/4) (25% yield): mp 75.4 °C (M). ¹H NMR (CDCl₃) δ (ppm): 6.90 (d, 1H, J = 7.9 Hz), 6.38–6.36 (d, 1H, J = 2.0Hz), 6.31 (dd, 1H, J = 7.9 Hz, J = 2.0 Hz), 6.08 (s br, 1H), 3.87 (t, 2H, J = 6.3 Hz), 3.70 (t, 2H, J = 4.9 Hz), 3.43-3.36(m, 2H), 2.48 (t, 2H, J = 7.5 Hz), 2.17 (t, 2H, J = 7.6 Hz), 1.81-1.68 (m, 2H), 1.63-1.43 (mm, 4H), 1.40-1.27 (mm, 18H), 0.86 (t, 3H, J=6.3 Hz). MS m/z: 422 [M + 1]⁺ (100), 444 [M $+ Na]^+$, 843 [2M + 1]⁺. Anal. (C₂₅H₄₃NO₄) C, H, N.

11-(2-Hexyl-5-hydroxyphenoxy)undecanoic Acid Cy**clopropylamide (52).** White solid (CHCl₃/MeOH = 47/3) (90% yield): mp 57.3 °C (M). 1 H NMR 300 MHz (DMSO) δ (ppm): 9.05 (s, 1H), 7.76 (s br, 1H), 6.81 (d, 1H, J = 7.7 Hz), 6.29 (s, J = 7.7 Hz)1H), 6.20 (d, 1H, J = 8.1 Hz), 3.83 (t, 2H, J = 5.5 Hz), 2.58-2.54 (m, 1H), 2.38 (t, 2H, J = 7.3 Hz), 1.95 (t, 2H, J = 6.9 Hz),1.69-1.64 (m, 2H), 1.43-1.38 (m, 4H), 1.26-1.14 (mm, 18H), 0.82 (t, 3H, J = 6.1 Hz), 0.57 - 0.53 (m, 2H), 0.33 - 0.31 (m, 2H).MS $\emph{m/z}$: 418 [M + 1]⁺, 440 [M + Na]⁺ (100). Anal. (C₂₆H₄₃- NO_3) C, H, N.

11-(2-Hexyl-5-hydroxyphenoxy)undecanoic Acid (4-Hydroxyphenyl)amide (53). White solid (CHCl₃/MeOH = 47/3; recrystallized diethyl ether) (83% yield): mp 108.7 °C (M). ${}^{1}H$ NMR (Acetone- d_{6}) δ (ppm): 9.05 (s, 1H), 9.00 (s, 1H), 8.80 (s, 1H), 7.43 (half of AB_q , 2H, J = 8.8 Hz), 6.86 (d, 1H, J= 7.9 Hz), 6.73 (half of ABq, 2H, J = 8.8 Hz), 6.40 (d, 1H, J = 2.5 Hz), 6.30 (dd, 1H, J = 2.5 Hz, J = 7.9 Hz), 3.90 (t, 2H, J= 6.1 Hz), 2.48 (t, 2H, J = 7.4 Hz), 2.28 (t, 2H, J = 7.2 Hz), 1.80-1.60 (mm, 14H), 1.50-1.32 (mm, 10H), 0.86 (t, 3H, J =6.3 Hz). Anal. (C₂₉H₄₃NO₄) C, H, N.

16-(2-Hexyl-5-hydroxyphenoxy)hexadecanoic Acid (2-Hydroxyethyl)amide (54). White solid (Ethyl acetate/MeOH = 50/1) (60% yield): mp 77.2 °C (M). 1 H NMR (DMSO) δ (ppm): 9.00 (s, 1H), 7.64 (s br, 1H), 6.80 (d, 1H, J = 8.0 Hz), $6.28 \, (d, 1H, J = 1.7 \, Hz), 6.20 \, (dd, 1H, J = 8.0 \, Hz, J = 1.7 \, Hz),$ 4.53 (t, 1H, J = 5.2 Hz), 3.83 (t, 2H, J = 5.8 Hz), 3.40-3.33(m, 2H), 3.10-3.05 (m, 2H), 2.38 (t, 2H, J = 7.3 Hz), 2.00 (t, 2H, J = 7.4 Hz), 1.66-1.63 (m, 2H), 1.41-1.15 (mm, 32H), 0.82 (t, 3H, J = 5.9 Hz). Anal. ($C_{30}H_{53}NO_4$) C, H, N.

16-(2-Hexyl-5-hydroxyphenoxy)hexadecanoic Acid Cy**clopropylamide (55).** White solid (CHCl₃/MeOH = 47/3) (80% yield): mp 72.8 °C (M). 1 H NMR 300 MHz (DMSO) δ (ppm): 9.05 (s, 1H), 7.75 (s br, 1H), 6.81 (d, 1H, J = 8.1 Hz), 6.29 (d, J = 8.1 Hz)1H, J = 2.4 Hz), 6.20 (dd, 1H, J = 2.4 Hz, J = 8.1 Hz), 3.83 (t, 2H, J = 6.3 Hz), 2.56-2.51 (m, 1H), 2.38 (t, 2H, J = 7.2 Hz),1.95 (t, 2H, J = 7.2 Hz), 1.69 - 1.62 (m, 2H), 1.42 - 1.20 (mm, 2H)32H), 0.82 (t, 3H, J = 6.3 Hz), 0.56-0.52 (m, 2H), 0.34-0.31(m, 2H). Anal. (C₃₁H₅₃NO₃) C, H, N.

16-(2-Hexyl-5-hydroxyphenoxy)hexadecanoic Acid (4-**Hydroxyphenyl)amide** (56). White solid (CHCl₃/MeOH = 46/4) (70% yield): mp 82.4 °C (M). 1 H NMR (DMSO) δ (ppm): $9.45~(s, 1H), 9.05~(s, 1H), 9.02~(s, 1H), 7.30~(half of AB_q, 2H, J)$ = 8.7 Hz), 6.91 (d, 1H, J = 8.0 Hz), 6.75 (half of AB_q, 2H, J = 8.7 Hz), 6.35 (d, 1H, J=2.2 Hz), 6.29 (dd, 1H, J=2.2 Hz, J=2= 8.0 Hz), 3.88 (t, 2H, J = 6.3 Hz), 2.40 (t, 2H, J = 7.5 Hz), 2.33 (t, 2H, J = 7.5 Hz), 1.80 - 1.62 (m, 4H), 1.58 - 1.25 (mm, 4H)30H), 0.86 (t, 3H, J = 6.5 Hz). Anal. ($C_{34}H_{53}NO_4$) C, H, N.

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Supporting Information Available: Yields, melting points, and chemical and spectroscopic data of all synthesized esters; elemental analysis data of final products. This material is available free of charge via the Internet at http://pubs.acs.org.

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