



Nicotinamide-independent asymmetric bioreduction of C=C-bonds via disproportionation of enones catalyzed by enoate reductases

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ABSTRACT

The asymmetric bioreduction of activated C=C-bonds catalyzed by a single flavoprotein was achieved via direct hydrogen transfer from a sacrificial 2-enone or 1,4-dione as hydrogen donor without requirement of a nicotinamide cofactor. Due to its simplicity, this system has clear advantages over conventional FAD-recycling systems.

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1. Introduction

Disproportionation reactions generally furnish a 1:1 mixture of products and are often plagued by unfavorable equilibria, hence they are commonly considered as inefficient and are rarely used in organic synthesis.¹ In biology, the disproportionation of sulfur yielding hydrogen sulfide and sulfate and of hydrogen peroxide furnishing O₂ and H₂O represent the most dominant disproportionation processes in Nature.² The disproportionation of cyclohex-2-enone, forming equimolar amounts of cyclohexanone and phenol has been described for several flavoproteins from the old yellow enzyme (OYE) family, such as OYE isoenzymes 1–3 and estrogen-binding protein.³ In the context of these studies, this phenomenon has been considered either as a minor side reaction demonstrating the catalytic promiscuity⁴ of OYEs or as ‘aromatase’ activity of enoate reductases catalyzing the formation of the phenolic A-ring in steroids, such as 17β-estradiol, from the corresponding enone-precursor 19-nortestosterone.³ Overall, this reaction constitutes a flavin-dependent hydrogen transfer, during which an equivalent of [2H] is formally transferred from a cyclohex-2-enone (being oxidized) onto another one (being reduced). The oxidized product constitutes a conjugated dienone, which spontaneously tautomerises to form phenol, thereby providing a large driving force of ca.

–30 kcal/M for the disproportionation reaction. During this hydrogen-transfer reaction, the flavin-cofactor is recycled internally and no external nicotinamide cofactor is required for the reductive half-reaction.⁵ In nicotinamide-dependent systems, C=C-bonds are reduced at the expense of an external hydride donor,⁶ such as formate, glucose, glucose-6-phosphate or phosphite, which requires a second dehydrogenase enzyme, such as FDH, GDH, G6PDH⁷ or phosphite-DH,⁸ respectively. This technology is generally denoted as ‘coupled-enzyme-approach’, which depends on the concurrent operation of two independent redox enzymes for substrate reduction and co-substrate-oxidation, respectively.⁹ Aiming to reduce the complexity of these redox systems,¹⁰ considerable efforts have recently been devoted to the development of nicotinamide-independent electrochemical and light-driven recycling systems for reduced flavins, which take advantage of the direct transfer of a hydride (or electrons, respectively) from a donor onto the flavin.¹¹ In this context, the nicotinamide-independent disproportionation of enones is of appealing simplicity, since it requires only a single flavoprotein and represents a ‘coupled-substrate-approach’.¹²

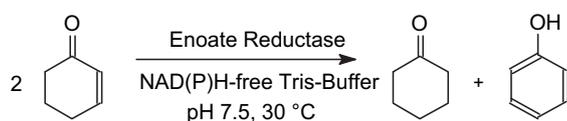
2. Results and discussion

In an initial screening, a set of cloned and overexpressed enoate reductases were tested for their catalytic activity in the disproportionation of cyclohex-2-enone (Scheme 1). To our delight, the desired disproportionation activity was observed in a variety of

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OYE-homologs. Although all of these enoate reductase-type proteins have been reported to reduce cyclohex-2-enone to cyclohexanone at the expense of NAD(P)H,¹³ the corresponding disproportionation activity was exceedingly low in 12-oxophytodienoic acid reductase (OPR) isoenzymes 1 & 3 and in *N*-ethylmaleimide-, morphinone-, and pentaerythritol tetranitrate reductase. Likewise, cyclohexenone reductase from *Zymomonas mobilis* showed only modest activity, whereas the OYE-homolog YqjM¹⁴ and OYE isoenzymes¹⁵ 1 and 2 from yeast were highly active. Surprisingly, OYE isoenzyme 3 was almost inactive. The divergent behavior of OYE isoenzymes is reflected by their structural relationship: Whereas both highly active isoenzymes 1 and 2 show an amino acid sequence identity of 92%, isoenzyme 3 is a more distant relative (sequence identity 80%). The disproportionation activity of estrogen-binding protein (EBP1)³ could be nicely reproduced using EBP1 cloned into *Escherichia coli*. By taking the strongest disproportionation activities as a lead, further experiments were performed using YqjM, OYE1, and OYE2.



Enzyme ^a	Conv. [%]
OPR1	<1
OPR3	<1
NEM-Reductase	<1
MOR-Reductase	<1
PETN-Reductase	0
<i>Zymomonas</i> NCR-Reductase	7
YqjM	85
OYE1	92
OYE2	75
OYE3	7
EBP1	45

Scheme 1. Disproportionation of cyclohex-2-ene catalyzed by enoate reductases.³OPR1, OPR3=oxophytodienoate reductase isoenzymes 1 and 3, respectively, from tomato;¹⁶ NEM-reductase=*N*-ethylmaleimide reductase;¹⁸ MOR-reductase=morphinone reductase;¹⁷ PETN-reductase=pentaerythritol tetranitrate reductase;¹⁸ *Z. mobilis* NCR reductase=nicotinamide-dependent cyclohexenone reductase;¹⁹ YqjM=OYE-homolog from *Bacillus subtilis*;¹⁴ OYE1–3=Old Yellow Enzyme isoenzymes from yeasts;¹⁵ EBP1=estrogen-binding protein,³ employed as cell-free extract of *E. coli* expressing EBP1.

In order to convert the scrambling-like hydrogen-transfer reaction between two identical cyclohexenone molecules into a useful *directed* redox process, a pair of suitable enone substrate/co-substrate—one *only* being reduced, the other *only* being oxidized—have to be coupled. During our previous studies on NAD(P)H-dependent enone reduction, we observed that α -substituted cyclic enones were quickly reduced, whereas an alkyl-substituent in the β -position

severely impeded the reaction rate.^{10,15} Hence, we envisaged that an α -substituted enone might act as H-acceptor, while a β -substituted analog would serve as H-donor.

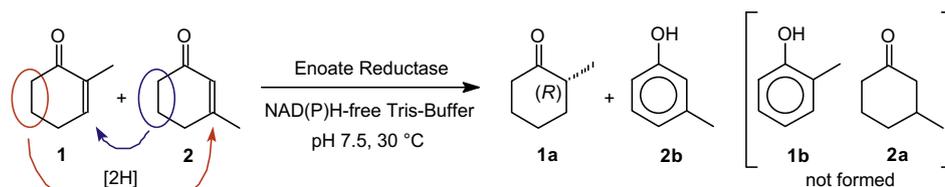
In order to test the viability of this concept, an equimolar amount of α - (1) and β -methylcyclohex-2-enone (2) were subjected to OYE1, OYE2, and YqjM in a nicotinamide-free buffer system (Scheme 2). The results of these experiments provided a clear proof-of-concept: Depending on the enzyme, the desired reduced α -methyl derivative 1a was formed in up to 48% conversion, while the oxidized β -methyl analog 2b was detected in approximately equimolar amounts.²⁰ In contrast, only trace amounts of the corresponding cross-hydrogen-transfer products 1b and 2a, which would arise from undesired oxidation of 1 and reduction of 2 were found, indicating that the directed hydrogen transfer indeed worked as envisaged.

Investigation of the optical purity and absolute configuration of 1a revealed that the product was formed in the same highly selective fashion as in the classic reduction-mode using NAD(P)H-recycling, ensuring that the chiral induction process of the enzymes remained unchanged.^{10b,15}

In order to test the applicability of this nicotinamide-free C=C-bond reduction system, we subjected two activated alkenes (3, 4), which are known to be readily reduced by enoate reductases in presence of NAD(P)H,^{10,15} to the hydrogen-transfer protocol in presence of equimolar amounts of β -methylcyclohex-2-enone (2) as hydrogen donor (Scheme 3). Whereas OYE1 and OYE2 showed only modest conversion, YqjM furnished the corresponding reduction products 3a and 4a in up to 22% conversion together with a stoichiometric amount of 3-methylphenol 2b. Within experimental errors, the enantiomeric excess of (*R*)-configured products 3a and 4a was identical to that of the nicotinamide-driven process, indicating that the enzymatic chiral induction remained intact.^{10,15}

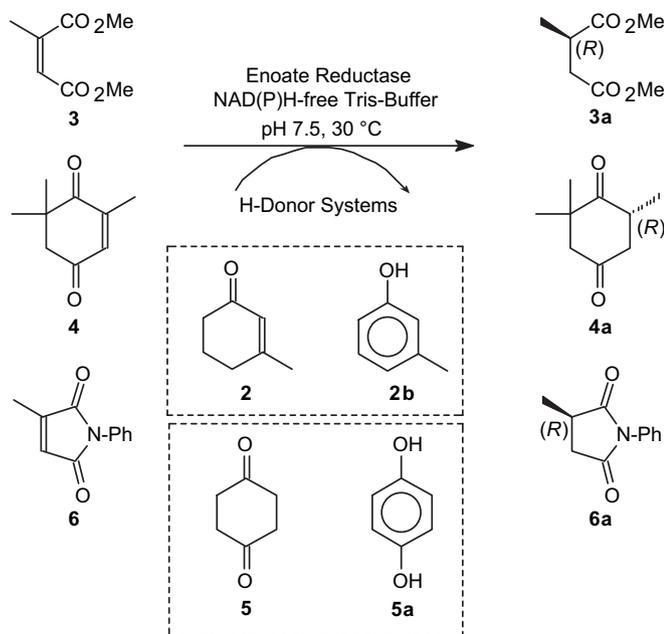
Since the use of equimolar amounts of 3-methylcyclohex-2-enone (2) as co-substrate would be economically unfavorable, a cheaper hydrogen donor was sought. After attempts using 1-indanone and hydroquinone failed, cyclohexane-1,4-dione (5)—yielding 1,4-dihydroxybenzene (hydroquinone, 5a) as oxidation product—was found to provide a suitable alternative. Using YqjM, all substrates showed enhanced conversion as compared to β -methylcyclohex-2-enone (2) as co-substrate.

At this early stage, this novel NAD(P)H-independent cofactor recycling system has not yet been fully optimized, particularly in view of the (co)-substrate concentrations²² and the overall conversions. Incomplete conversions might be attributed to a certain degree of enzyme inhibition, most presumably caused by the co-product phenol(s), which are known to form charge-transfer complexes with flavins.²³ Since the latter process is reversible, the removal of phenols by in-situ (co)-product removal (ISPR)²⁴ using



Enzyme	1a		2b	1b [%]		2a [%]	
	c [%]	[e.e. [%]]		c [%]	c [%]	c [%]	
OYE1	27	[85 (<i>R</i>)]	27%	<1	<1		
OYE2	16	[80 (<i>R</i>)]	19%	<1	<1		
YqjM	48	[91 (<i>R</i>)]	65%	<1	<1		

Scheme 2. Asymmetric hydrogen transfer between α - (1) and β -methylcyclohex-2-enone (2) catalyzed by enoate reductases.



Enzyme	Substrate	H-Donor	Product		Co-Product
			c [%]	e.e. [%]	
OYE1	3	2	3a 2	n.d.	2b 5
OYE2	3	2	3a 1	n.d.	2b 3
YqjM	3	2	3a 17	>99 (R)	2b 21
OYE1	4	2	4a 4	n.d.	2b 6
OYE2	4	2	4a 3	n.d.	2b 4
YqjM	4	2	4a 22	76 (R)	2b 25
OYE1	3	5	3a 3	n.d.	5b 5
OYE2	3	5	3a 4	n.d.	5b 6
YqjM	3	5	3a 20	>99 (R)	5b 19
YqjM	4	5	4a 35 ^a	66 (R)	5b 29
YqjM	6	5	6a 51 ^b	>99 (R)	5b 49

Scheme 3. Asymmetric bioreduction of activated alkenes via a coupled-substrate approach. n.d.=not determined. ^aA trace of 2,3-epoxy-2,6,6-trimethylcyclohexane-1,4-dione was formed ($\leq 3\%$), presumably due to spontaneous epoxidation of the C=C-bond by H₂O₂ derived via enzyme-catalyzed reduction of O₂.²¹ ^bA trace of aniline was detected as side product ($\leq 3\%$).

biphasic aqueous-organic solvent systems²⁵ is the first choice. This strategy has been successfully applied to industrial scale using (flavin-dependent) enzymatic Beyer–Villiger oxidation²⁶ and (cytochrome P₄₅₀-dependent) alkene epoxidation mediated by whole viable cells.²⁷

3. Conclusion

An novel substrate-coupled C=C-bond bioreduction system was developed, which depends only on a single flavoprotein and neither requires a second (dehydrogenase) recycling enzyme, nor a nicotinamide cofactor. Due to its simplicity, it has clear advantages over other nicotinamide-independent alternative systems, such as light-driven and electrochemical FAD-recycling systems.¹¹

4. Experimental

4.1. General

GC–MS analyses were performed with an Agilent 7890A GC system equipped with an Agilent 5975C mass-selective detector (electron impact, 70 eV) using a (5%-phenyl)-methylpolysiloxane phase column (Agilent HP-5 ms, 30 m, 250 μ m, 0.25 μ m). Helium was used as carrier gas (column flow: 2 mL/min). GC–FID analyses were carried out with a Varian 3800 by using H₂ as a carrier gas

(14.5 psi). HPLC analyses were performed by using a Shimadzu system equipped with a Chiralcel OD-H column (25 cm, 0.46 cm). NMR spectra were measured on a Bruker AMX spectrometer at 360 MHz.

2-Methyl-2-cyclohexen-1-one (**1**), 2-methylcyclohexan-1-one (**1a**), 3-methylcyclohexan-1-one (**2a**) and *rac*-levodione were provided by BASF (Ludwigshafen). Cyclohexanone, 2-cyclohexen-1-one, phenol, 3-methyl-2-cyclohexen-1-one (**2**), 1,4-cyclohexanedione (**5**), hydroquinone (**5a**), and *N*-phenyl-2-methylmaleimide (**6**) were purchased from Aldrich. Citraconic acid was purchased from Alfa Aesar and 4-ketoisophorone was from ABCR Co.

4.2. Synthesis of substrates and reference materials

4.2.1. *rac*-Methylsuccinic acid. Citraconic acid (105 mg, 0.81 mmol) was dissolved in THF/EtOH 50:50 (10 mL) and was hydrogenated at atmospheric pressure and room temperature in the presence of 10% Pd/C (5 mg) as catalyst. After 24 h, the mixture was filtered through Celite and evaporated yielding 99% of *rac*-**1b** (106 mg, 0.80 mmol). Mp=110–115 °C. ¹H NMR (D₂O): δ 3.62–3.64 (d, 3H, *J*=7.2 Hz), 4.96–5.15 (m, 2H), 5.29–5.34 (m, 1H). ¹³C NMR (D₂O): δ 18.7, 38.2, 39.6, 179.0, 182.8.¹⁵

4.2.2. *rac*-Dimethyl-2-methylsuccinate (3a**).** A solution of *rac*-methylsuccinic acid (32 mg, 0.24 mmol) in BF₃/MeOH (0.5 mL, 14%) was stirred at 100 °C for 1 h. H₂O (0.5 mL) was added and the reaction mixture was extracted with *n*-hexane (3 \times 1 mL). The combined organic layers were dried over Na₂SO₄, filtered and evaporated, yielding 46% of *rac*-(**3a**) (17 mg, 0.11 mmol). ¹H NMR (CDCl₃): δ 1.22–1.24 (d, 3H, *J*=7.2 Hz), 2.39–2.45 (dd, 1H, *J*=6.06 Hz, *J*=16.52 Hz), 2.72–2.79 (dd, 1H, *J*=8.15 Hz, *J*=16.51 Hz), 2.91–2.94 (m, 1H), 3.69 (s, 3H), 3.71 (s, 3H). ¹³C NMR (CDCl₃): δ 17.0, 35.7, 37.4, 51.7, 51.9, 172.3, 175.7.¹⁵

4.2.3. Citraconic acid dimethylester (3**).** Substrate **3** was synthesized according to the procedure described above starting from citraconic acid. ¹H NMR (CDCl₃): δ 2.06–2.07 (d, 3H, *J*=1.6 Hz), 3.73 (s, 3H), 3.83 (s, 3H), 5.86–5.87 (d, 1H, *J*=1.6 Hz). ¹³C NMR (CDCl₃): δ 20.5, 51.8, 52.4, 120.6, 145.7, 165.4, 169.4.¹⁵

4.2.4. *rac*-*N*-Phenyl-2-methylsuccinimide (6a**).** *N*-Phenyl-2-methylmaleimide (**6**, 50 mg, 0.27 mmol) was dissolved in EtOAc (5 mL) and was hydrogenated at atmospheric pressure at room temperature using 10% Pd/C (2.8 mg) as catalyst. After 24 h, the mixture was filtered through Celite and evaporated yielding 94% of *rac*-**6a** (48 mg, 0.25 mmol). ¹H NMR (CDCl₃): δ 1.47 (d, 3H, *J*=7 Hz), 2.52 (dd, 1H, *J*=17.4 Hz, *J*=4 Hz), 3.01–3.10 (m, 1H), 3.11 (dd, 1H, *J*=17.3 Hz, *J*=9.2 Hz), 7.29–7.51 (m, 5H).¹⁵

4.3. General procedure for the screening for enzymatic disproportionation of cyclohex-2-enone

An aliquot of the isolated enzyme OPR1, OPR3, YqjM, OYE1, OYE2, OYE3, *Z. mobilis* ER, NEM-Red, MOR-Red, and PETN-Red (protein purity >90%, protein content 90–110 μ g/mL) was added to a Tris–HCl buffer solution (0.8 mL, 50 mM, pH 7.5) containing cyclohex-2-enone (10 mM). The mixture was shaken at 30 °C and 120 rpm for 24 h and the products were extracted with EtOAc (2 \times 0.5 mL). The combined organic phases were dried (Na₂SO₄) and the resulting samples were analyzed on achiral GC. Products were identified by comparison with authentic reference materials via co-injection on GC–MS and achiral GC. Column: 6% Cyanopropyl-phenyl phase capillary column (Varian CP-1301, 30 m, 0.25 mm, 0.25 μ m), detector temperature 250 °C, split ratio 30:1; temperature program: 80 °C; hold 2 min.; rise to 120 °C with 5 °C/min. *T*_{Ret}: cyclohex-2-enone 2.97 min, cyclohexanone 2.43 min, phenol 4.98 min.

4.10. Determination of enantiomeric excess and absolute configuration

The enantiomeric excess of **1a** and **3a** was determined using a modified β -cyclodextrin capillary column (ChiralDEX B-TA, 40 m, 0.25 mm). Detector temperature 200 °C, injector temperature 180 °C, split ratio 25:1. Temperature program for **1a**: 80 °C hold 2 min, 5 °C/min to 105 °C, 10 °C/min, hold 4 min. Retention times: (R)-**1a** 6.34 and (S)-**1a** 6.47 min. Temperature program for **3a**: 90 °C hold 4 min, 3 °C/min to 115 °C, 30 °C/min to 180 °C. Retention times: (S)-**3b** 7.33 min; (R)-**3b** 7.45 min; The enantiomeric excess of **4a** was determined using a β -cyclodextrin capillary column (CP-Chirasil-DEX CB, 25 m, 0.32 mm, 0.25 μ m film). Temperature program for **4a**: 90 °C hold 2 min, 4 °C/min to 115 °C, 20 °C/min to 180 °C, hold 2 min. Retention times: (R)-**4a** 6.42; (S)-**4a** 6.74 min. The enantiomeric excess of **8b** was determined on HPLC using *n*-heptane/*i*-PrOH 95:5 (isocratic) at 18 °C and 1 mL/min. Retention times: (R)-**8b** 25.10 min; (S)-**8b** 29.15 min.¹⁵

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References and notes

- Notable exceptions are the Cannizzaro reaction ($2R-CHO \rightarrow R-CH_2OH + R-CO_2H$), Tishchenko reaction ($2R-CHO \rightarrow R-CO_2R$), Kornblum–DeLaMare rearrangement ($R^2CH-O-O-CR^3 \rightarrow R^1-CO-R^1 + R^3C-OH$), Meerwein–Ponndorf–Verley/Oppenauer reduction/oxidation ($R^1R^2CO + 2-PrOH \leftrightarrow R^1R^2CH-OH + acetone$), Boudouard reaction ($2CO \rightarrow CO_2 + C$) and the catalytic disproportionation of toluene ($2MePh \rightarrow benzene + xylene$), see: Banks, R. L. *J. Mol. Catal.* **1980**, *8*, 269–276; Abdal Kareem, M. A.; Chand, S.; Mishra, I. M. *J. Sci. Ind. Res.* **2001**, *60*, 319–327.
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