

# A Bioinspired Strategy for the Enantioselective Synthesis of Bicyclic Oxygen Heterocycles

Edith Rodriguez Venegas and Christine L. Willis\*



Cite This: <https://dx.doi.org/10.1021/acs.orglett.0c00425>



Read Online

ACCESS |



Metrics & More

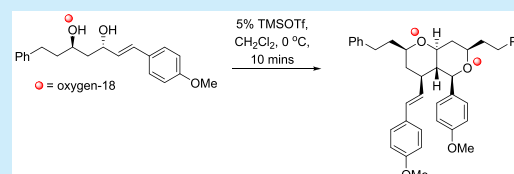


Article Recommendations



Supporting Information

**ABSTRACT:** A new strategy is described for the direct conversion of unsaturated 3,5-dihydroxy-diarylheptanoids to dimeric products assembled on *trans*-2,8-dioxabicyclo[4.4.0]decane frameworks. The key atom-economical acid-mediated coupling creates 2 rings and 4 new stereocenters in a single-pot process. Oxygen-18 labeling studies are in accord with reactions proceeding via a cascade mechanism involving carbocationic intermediates. This approach enabled the concise total syntheses of analogues of the natural product blepharocalyxin D in 4 steps from simple starting materials.



Many compounds assembled on fused heterocycles display potent bioactivities; hence, the development of efficient approaches for their synthesis is an important goal. Among them is a diverse family of diarylheptanoids which exhibit potent antiproliferative activities.<sup>1</sup> For example, Kadota and co-workers reported a series of related polyphenolic compounds from the seeds of *Alpinia blepharocalyx*,<sup>2</sup> a plant commonly used in Chinese traditional medicine. These natural products include linear diarylheptanoids (e.g., **1**, Figure 1), those assembled on a single tetrahydropyran/dihydropyran ring (e.g., **2** and calyxin L) as well as fused heterocyclic frameworks such as calyxin I.

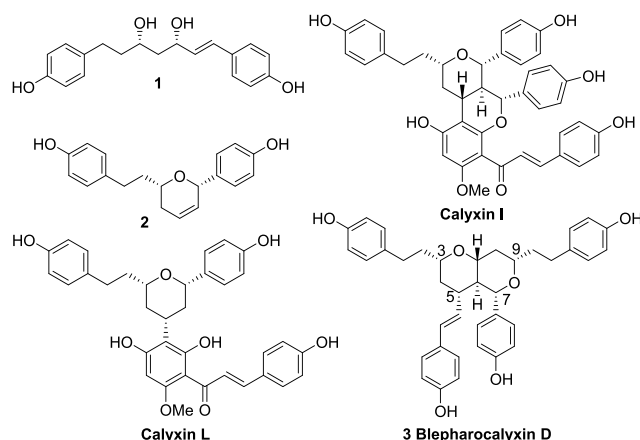
A diarylheptanoid of particular interest is blepharocalyxin D **3** isolated in small quantities (only 5 mg from 10 kg of seeds of *A. blepharocalyx*) which exhibits potent antiproliferative activity against murine colon 26-L5 carcinoma and human

HT 1080 fibrosarcoma cells.<sup>2e</sup> Structure–activity studies on such compounds have been hampered by the paucity of available natural products and analogues.

Blepharocalyxin D **3** is assembled on a *trans*-2,8-dioxabicyclo[4.4.4]decane with 4 equatorial side chains. Its structure was determined using spectroscopic methods<sup>2a</sup> and later confirmed by total synthesis. In the first reported synthesis, Lee and co-workers used two separate Prins cyclizations to construct each oxane ring giving the natural product in 17 steps and 0.9% overall yield.<sup>3</sup> Later we developed a new approach via reaction of methyl 3,3-dimethoxypropionate with  $\gamma,\delta$ -unsaturated alcohols to give the *trans*-fused bicyclic framework with creation of 2 rings and 4 stereocenters in a single pot. Further elaboration to introduce the C-9 side chain gave blepharocalyxin D in 15 steps and 8% overall yield.<sup>4</sup>

To date no biosynthetic studies have been reported for these dimeric diarylheptanoids from *A. blepharocalyx*. However, Kadota and co-workers speculated on their possible biogenesis based on cometabolites including diol **1** and dihydropyran **2** which may serve as biosynthetic building blocks to the more complex fused heterocyclic frameworks.<sup>2e</sup> This biosynthetic speculation gave inspiration for our development of a concise approach for the total synthesis of a series of related dimeric diarylheptanoids including (+)-blepharocalyxin D which are now reported alongside mechanistic studies.

To begin, racemic diol **4** was prepared using the approach of Cossy et al.<sup>5</sup> giving a mixture of diastereomers in favor of the

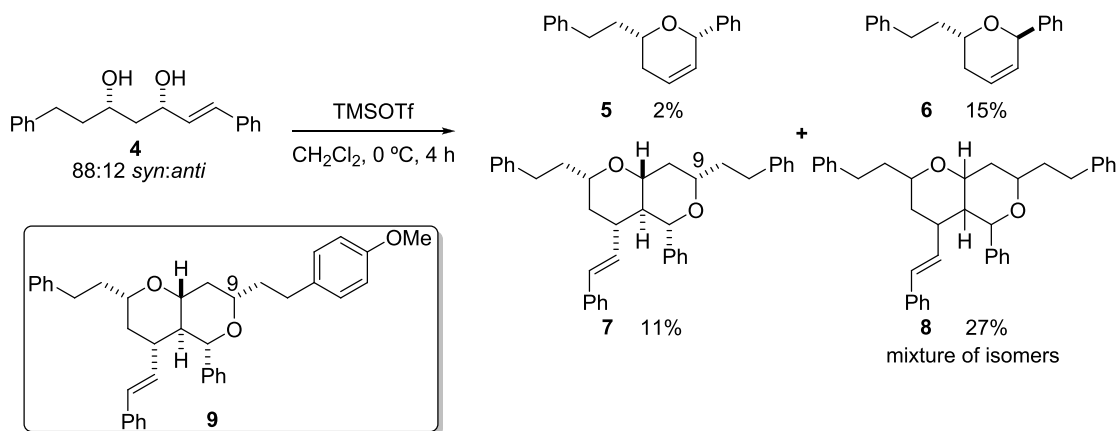


**Figure 1.** Examples of diarylheptanoids isolated from the seeds of *A. blepharocalyx*.

**Received:** February 2, 2020



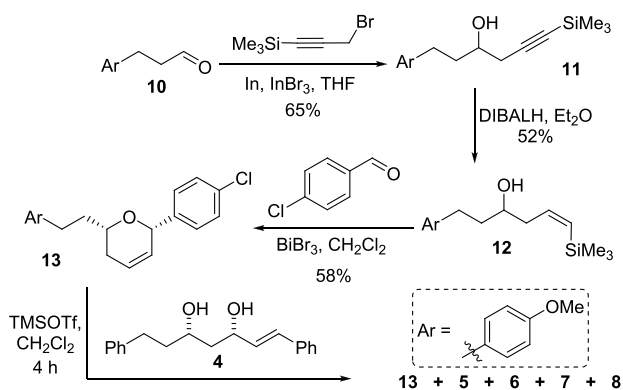
Scheme 1. Synthesis of Dimeric Diarylheptanoids from Racemic Diol 4



*syn*-isomer. Treatment of diol 4 with TMSOTf in dry dichloromethane at 0 °C gave a mixture of products which were separated by column chromatography giving dihydropyrans 5 and 6 in 2% and 15% yield, respectively (Scheme 1). In addition, several dimeric diarylheptanoids were formed including the desoxy-analogue 7 of blepharocalyxin D. The structure of 7 with the *trans* ring junction and 4 equatorial side chains was determined by extensive NMR studies and confirmed by comparison with data for a similar product 9 (with a 9-*p*-methoxyphenethyl side chain) for which the X-ray structure has been reported.<sup>4</sup>

To investigate if a dihydropyran is involved in the coupling process, dihydropyran 13 was synthesized with *p*-methoxyphenyl and *p*-chlorophenyl groups required to determine the origin of the aromatic rings in the dimeric products (Scheme 2). An indium-mediated propargylation of aldehyde 10<sup>6</sup> and

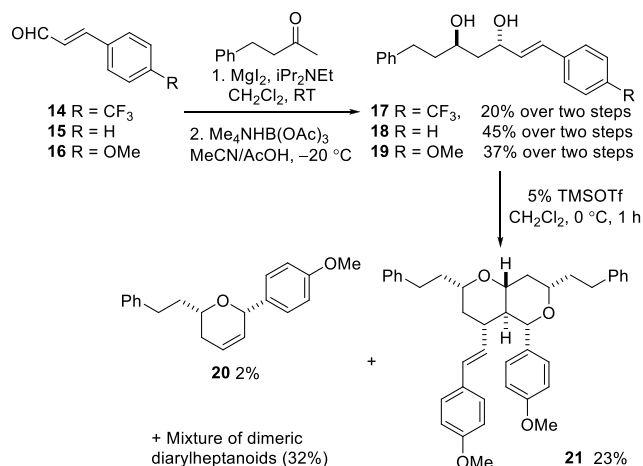
Scheme 2. Synthesis of Dihydropyran 13 and Reaction with Diol 4



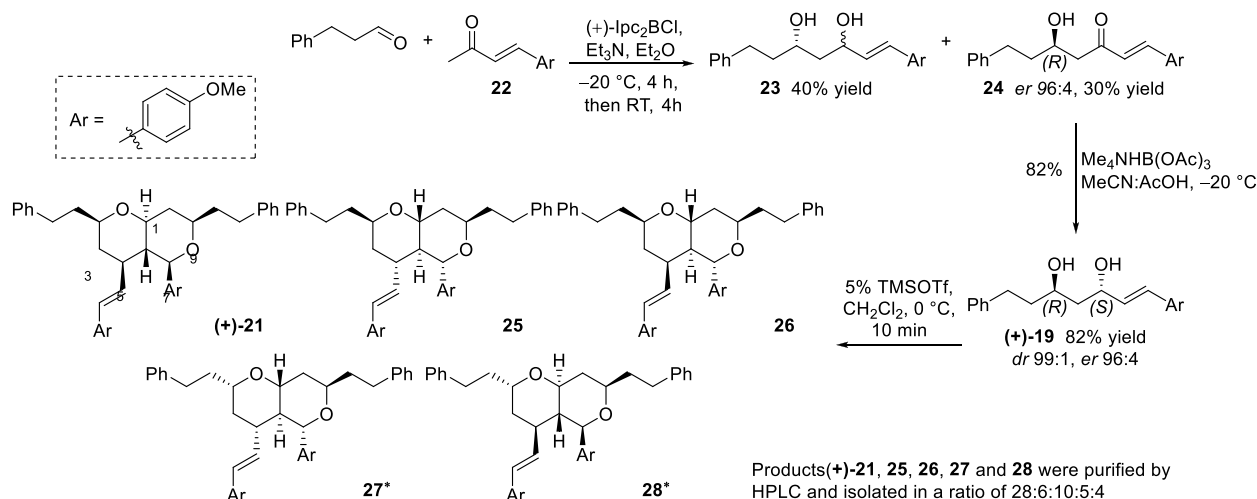
subsequent DIBALH reduction of alkyne 11 gave *Z*-vinylsilane 12. Prins cyclization of 12 with *p*-chlorobenzaldehyde in the presence of BiBr<sub>3</sub><sup>7</sup> gave dihydropyran 13 in 58% yield as a single diastereomer; the *syn* configuration was determined by <sup>1</sup>H NMR spectroscopy and comparison with reported analogues.<sup>8</sup> Treatment of a mixture of dihydropyran 13 and diol 4 with TMSOTf returned unchanged dihydropyran 13 and products (5–8) isolated previously from the starting diol 4. Hence, a dihydropyran is not involved in the synthesis of the dimeric diarylheptanoids. Optimization of the reaction conditions by varying the temperature and acid showed no

improvement on the original conditions of 5% TMSOTf in CH<sub>2</sub>Cl<sub>2</sub> at 0 °C. While both *syn* and *anti* diols gave dimeric products under the optimized conditions, a slightly higher yield of the blepharocalyxin analogue was generally observed in the case of the *anti*-diols (e.g., *syn*-diol 4 and *anti*-diol 18 gave 7 in 11% and 20% yield respectively).

Racemic diols 17–19 with variously *para*-substituted aromatic rings were prepared via an aldol reaction of 4-phenylbutan-2-one and aldehydes 14–16 with MgI<sub>2</sub> and DIPEA,<sup>9</sup> followed by reduction under Evans' conditions<sup>10</sup> (Scheme 3). This reaction sequence was simple to perform on a multigram scale.

Scheme 3. Two-Step Synthesis of Racemic *anti*-Diols and Reaction with TMSOTf

Diols 17, 18, and 19 were reacted separately under the same conditions (5% TMSOTf in CH<sub>2</sub>Cl<sub>2</sub>) at 0 °C. After 1 h, diol 17 with the electron-withdrawing *p*-trifluoromethylphenyl group simply returned starting material. In the case of diol 18, cyclic products were observed but ca. 50% of starting material was also recovered. In contrast, diol 19 with the *p*-methoxyphenyl group gave blepharocalyxin D analogue 21 in 23% yield along with a further 32% of mixed dimeric products. No starting material 19 was detected. These results are in accord with the proposal that the reaction proceeds via carbocationic intermediates stabilized by the presence of the electron-rich *p*-methoxyphenyl ring.

Scheme 4. Enantioselective Synthesis of *anti*-Diol (+)-19 and Reaction with TMSOTf (\*or the Enantiomers)

To prepare enantiomerically enriched diol (+)-19, dihydrocinnamaldehyde was coupled with ketone **22** in the presence of (+)-Ipc<sub>2</sub>BCl and triethylamine<sup>11</sup> at -20 °C for 4 h, giving hydroxyketone **24** with 75:25 *er* in 81% yield. To improve the *er*, the reaction mixture was then warmed to room temperature and monitored by chiral HPLC until one of the enantiomers had been almost fully consumed. Following column chromatography, (*R*)-hydroxyketone **24** was isolated with 96:4 *er* and diol **23** in 40% yield. Directed reduction of hydroxyketone **24** gave the required *anti*-diol (+)-19 in 82% yield with 99:1 *dr*.

Treatment of diol (+)-19 with 5% TMSOTf for just 10 min at 0 °C gave the novel analogue (+)-21 of the natural product blepharocalyxin D with the *trans*-2,8-dioxabicyclo[4.4.0]-decane framework and 4 equatorial side chains (Scheme 4). Further dimeric diarylheptanoids **25**–**28** were purified by HPLC, and their structures were determined by extensive NMR studies. It was evident that all four products were assembled on a *trans*-2,8-dioxabicyclodecane framework (coupling of 1-H to 6-H, *J* 10–11 Hz) with the 7-methoxyphenyl group equatorial in each case (coupling 6-H to 7-H, *J* 10–11 Hz). The structures of the isomers varied by having one or more axial side chains as confirmed from analysis of coupling constants combined with NOE studies (Supporting Information).

Furthermore, compounds **25**–**27**, each with a C-9 axial side chain, showed a characteristic downfield shift of the signal assigned to 9-H(eq) to  $\delta$  4.2 ppm from  $\delta$  3.5 (9-H<sub>ax</sub>) in blepharocalyxin analogue **21** and isomer **28** (Figure 2). The major products **21**, **25**, and **26** each arise from coupling two molecules of the major enantiomer (3*S*, 5*R*) of starting diol **19**. In contrast, the remaining two products **27** and **28** were isolated in very low yields, 3% and 1%, respectively, and originate from coupling the enantiomers of the starting material.

Results from the studies described herein using substrates with the electron-rich and electron-deficient aromatic rings (Scheme 3), combined with the array of products formed (Scheme 4) are in accord with the proposed mechanism for the formation of the dimeric diarylheptanoids illustrated in Scheme 5 for the major diastereomer **21**. Loss of the allylic hydroxyl group from diol **19** gives carbocation **I** stabilized by the aromatic ring (Ar). Coupling of diol **19** with **I** and loss of

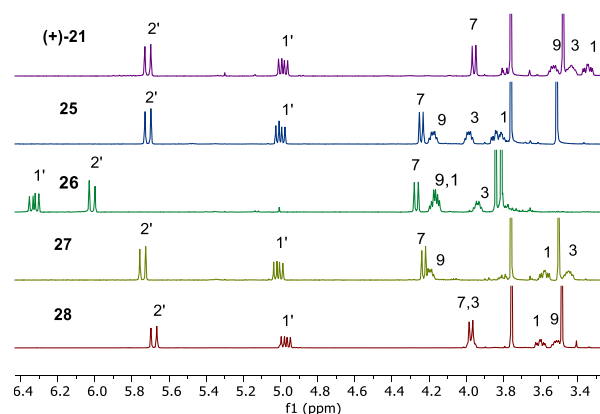
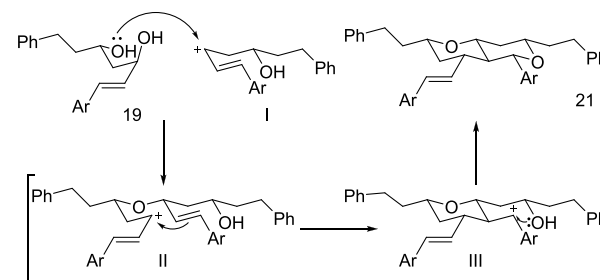


Figure 2. Comparison of the <sup>1</sup>H NMR spectra of compounds **21**, **25**–**28**.

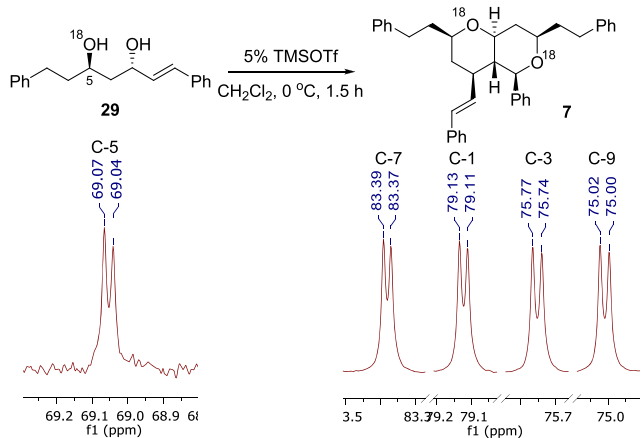
## Scheme 5. Proposed Mechanism for Formation of Dimeric Diarylheptanoids



the second allylic alcohol generates a further intermediate **II**. Intramolecular cyclization leads to formation of the new carbon–carbon bond and secondary carbocation **III** again stabilized by the aromatic ring. Finally, cyclization delivers the bicyclic product **21**. While the *trans*-fused rings with all equatorial side chains are preferred, other diastereomers may arise by non-stereoselective attack on the carbocationic intermediates.

An interesting feature of the proposed mechanism is that the allylic alcohols must be lost selectively from the diols in the coupling process; thus, both oxygens in the heterocycles will

originate from the 5-hydroxyl group of starting diol **19**. This proposal was verified using an oxygen-18 labeling experiment. [ $^{18}\text{O}$ ]-Diol **29** was prepared via the approach shown in Scheme 4 starting from [ $^{18}\text{O}$ ]-dihydrocinnamaldehyde (50% incorporation, prepared by exchange with  $\text{H}_2^{18}\text{O}$ ) and (*E*)-4-phenylbut-3-en-2-one. The incorporation of label was confirmed by the presence of two signals assigned to C-5 in the  $^{13}\text{C}$  NMR spectrum arising from the characteristic upfield shift (ca.  $\delta$  0.02 ppm) of  $^{13}\text{C}$ – $^{18}\text{O}$  compared with  $^{13}\text{C}$ – $^{16}\text{O}$  (Figure 3).



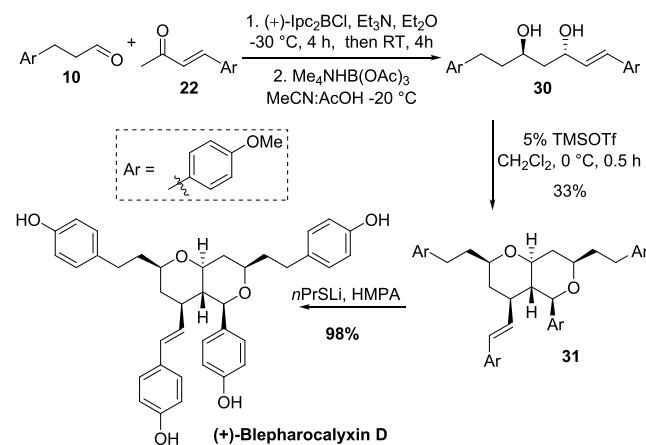
**Figure 3.** Diagnostic signals in the  $^{13}\text{C}$  NMR spectra of oxygen-18 labeled **29** and **7**.

[ $^{18}\text{O}$ ]-Diol **29** was treated under the standard TMSOTf-mediated reaction conditions and the major dimeric product **7** isolated by column chromatography. The  $^{13}\text{C}$  NMR spectrum clearly showed that the signals assigned to all four oxygenated carbons in the rings, C-1 ( $\delta$  79.13), C-3 ( $\delta$  75.77), C-7 ( $\delta$  83.39), and C-9 ( $\delta$  75.02), included an isotopically labeled upfield signal with retention of all the oxygen-18 label from the starting diol **29**, consistent with the proposed cascade mechanism (Scheme 5).

Finally, this new acid-mediated procedure was used in the total synthesis of the enantiomer of the natural product, blepharocalyxin D. The required substrate **30** for the key dimerization was prepared in 2 steps via an aldol reaction of aldehyde **10** and ketone **22** followed by directed reduction of the resultant  $\beta$ -hydroxyketone as shown in Scheme 6. Treatment of **30** with TMSOTf gave bicyclic product **31** which was deprotected using LiSPr/HMPA to give (+)-blepharocalyxin D in 4 steps from simple starting materials. The synthetic sample of (+)-blepharocalyxin D gave an optical rotation of  $[\alpha]_{\text{D}}^{25} +80.3$  (c. 0.7 MeOH), in accord for it being the enantiomer of the natural product.<sup>3a,4</sup>

In conclusion, we have developed a bioinspired approach for the efficient synthesis of bicyclic heterocycles assembled on a *trans*-2,8-dioxabicyclo[4.4.0]decane decorated with 4 side chains. A series of unsaturated 3,5-dihydroxydiarylheptanoids were prepared using aldol chemistry followed by a directed reduction of the resultant  $\beta$ -hydroxyketones. The key atom economic step involves an acid mediated (5% TMSOTf) coupling of these linear dihydroxy-diarylheptanoid to produce 2 oxane rings and 4 new stereocenters in one pot. Oxygen-18 labeling studies were in accord with the proposed cascade mechanism of dimerization. Several analogues of blepharocalyxin D have been prepared including the total synthesis of the

## Scheme 6. Total Synthesis of (+)-Blepharocalyxin D



enantiomer of the natural product which was achieved in 4 steps and 13% overall yield from simple starting materials, aldehyde **10** and enone **22**. In comparison with previous total syntheses of blepharocalyxin D,<sup>3a,4</sup> this concise approach significantly reduces the number of steps to such targets and will give access to a library of dimeric diarylheptanoids for further biological assessment.

## ■ ASSOCIATED CONTENT

### Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.0c00425>.

Experimental procedures, spectroscopic data and  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of all purified compounds (PDF)

## ■ AUTHOR INFORMATION

### Corresponding Author

Christine L. Willis – School of Chemistry, University of Bristol, Bristol BS8 1TS, United Kingdom; [orcid.org/0000-0002-3919-3642](https://orcid.org/0000-0002-3919-3642); Email: [Chris.Willis@bristol.ac.uk](mailto:Chris.Willis@bristol.ac.uk)

### Author

Edith Rodriguez Venegas – School of Chemistry, University of Bristol, Bristol BS8 1TS, United Kingdom

Complete contact information is available at: <https://pubs.acs.org/doi/10.1021/acs.orglett.0c00425>

### Notes

The authors declare no competing financial interest.

## ■ ACKNOWLEDGMENTS

We are grateful to CONACyT for a studentship to E.R.V. and the EPSRC for funding Equipment to Chemical Synthesis CDT (EP/K035746/1) for access to HPLC.

## ■ REFERENCES

- (1) (a) Zhang, W. J.; Luo, J. G.; Kong, L. Y. *World J. Tradit. Chinese Med.* **2016**, *2*, 26–41. (b) Ma, X. N.; Xie, C. L.; Miao, Z.; Yang, Q.; Yang, X. W. *RSC Adv.* **2017**, *7*, 14114–14144.
- (2) (a) Dong, H.; Chen, S. X.; Xu, H. X.; Kadota, S.; Namba, T. *J. Nat. Prod.* **1998**, *61*, 142–144. (b) Prasain, J. K.; Li, J. X.; Tezuka, Y.; Tanaka, K.; Basnet, P.; Dong, H.; Namba, T.; Kadota, S. *J. Nat. Prod.* **1998**, *61*, 212–216. (c) Gewali, M. B.; Tezuka, Y.; Banskota, A. H.; Ali, M. S.; Saiki, I.; Dong, H.; Kadota, S. *Org. Lett.* **1999**, *1*, 1733–

1736. (d) Tezuka, Y.; Gewali, M. B.; Ali, M. S.; Banskota, A. H.; Kadota, S. *J. Nat. Prod.* **2001**, *64*, 208–213. (e) Ali, M. S.; Tezuka, Y.; Banskota, A. H.; Kadota, S. *J. Nat. Prod.* **2001**, *64*, 491–496.
- (3) (a) Ko, H. M.; Lee, D. G.; Kim, M. A.; Kim, H. J.; Park, J.; Lah, M. S.; Lee, E. *Tetrahedron* **2007**, *63*, 5797–5805. (b) Ko, H. M.; Dong, G. L.; Kim, M. A.; Kim, H. J.; Park, J.; Lah, M. S.; Lee, E. *Org. Lett.* **2007**, *9*, 141–144.
- (4) Cons, B. D.; Bunt, A. J.; Bailey, C. D.; Willis, C. L. *Org. Lett.* **2013**, *15*, 2046–2049.
- (5) Cornil, J.; Gonnard, L.; Guérinot, A.; Reymond, S.; Cossy, J. *Eur. J. Org. Chem.* **2014**, *2014*, 4958–4962.
- (6) Dias, L. C.; De Lucca, E. C.; Ferreira, M. A. B.; Polo, E. C. *J. Braz. Chem. Soc.* **2012**, *23*, 2137–2158.
- (7) Lian, Y.; Hinkle, R. J. *J. Org. Chem.* **2006**, *71*, 7071–7074.
- (8) (a) Ali, M. S.; Tezuka, Y.; Banskota, A. H.; Kadota, S. *J. Nat. Prod.* **2001**, *64*, 491–496. (b) Kadota, S.; Tezuka, Y.; Prasain, J. K.; Ali, M. S.; Banskota, A. H. *Curr. Top. Med. Chem.* **2003**, *3*, 203–225.
- (9) Wei, H.-X.; Jasoni, R. L.; Shao, H.; Hu, J.; Paré, P. W. *Tetrahedron* **2004**, *60*, 11829–11835.
- (10) Evans, D. A.; Chapman, K. T.; Carreira, E. *J. Am. Chem. Soc.* **1988**, *110*, 3560–3578.
- (11) Paterson, I.; Goodman, J. M.; Anne Lister, M.; Schumann, R. C.; McClure, C. K.; Norcross, R. D. *Tetrahedron* **1990**, *46*, 4663–4684.