dimethylformamide (1 mL) was treated with 50% sodium hydride in oil (20 mg) for 20 min, and then with ethyl iodide (0.02 mL) for 20 min longer. The mixture was concentrated to dryness at 0.1 mm, taken up in ethyl acetate, washed with sodium bicarbonate solution, dried (Mg_2SO_4), concentrated to dryness, and triturated with ether to afford the ester as an off-white solid, mp 110–118 °C (30 mg, 31%). Similarly, 3s was prepared from 3a 2579

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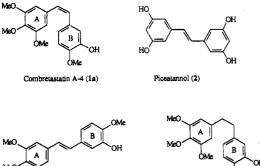
Synthesis and Evaluation of Stilbene and Dihydrostilbene Derivatives as Potential Anticancer Agents That Inhibit Tubulin Polymerization

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An array of cis-, trans-, and dihydrostilbenes and some N-arylbenzylamines were synthesized and evaluated for their cytotoxicity in the five cancer cell cultures A-549 lung carcinoma, MCF-7 breast carcinoma, HT-29 colon adenocarcinoma, SKMEL-5 melanoma, and MLM melanoma. Several cis-stilbenes, structurally similar to combretastatins, were highly cytotoxic in all five cell lines and these were also found to be active as inhibitors of tubulin polymerization. The most active compounds also inhibited the binding of colchicine to tubulin. The most potent of the new compounds, both as a tubulin polymerization inhibitor and as a cytotoxic agent, was (Z)-1-(4-methoxyphenyl)-2-(3,4,5-trimethoxyphenyl)ethene (5a). This substance was almost as potent as combretastatin A-4 (1a), the most active of the combretastatins, as a tubulin polymerization inhibitor. Compound 5a was found to be approximately 140 times more cytotoxic against HT-29 colon adenocarcinoma cells and about 10 times more cytotoxic against MCF-7 breast carcinoma cells than combretastatin A-4. However, 5a was found to be about 20 times less cytotoxic against A-549 lung carcinoma cells, 30 times less cytotoxic against SKMEL-5 melanoma cells, and 7 times less cytotoxic against MLM melanoma cells than combretastatin A-4. The relative potencies 5a > 8a > 6a for the cis, dihydro, and trans compounds, respectively, as inhibitors of tubulin polymerization are in agreement with the relative potencies previously observed for combretastatin A-4 (1a), dihydrocombretastatin A-4 (1c), and transcombretastatin A-4 (1b). The relative potencies 5a > 8a > 6a were also reflected in the results of the cytotoxicity assays. Structure-activity relationships of this group of compounds are also discussed.

Interest in the synthesis and evaluation of polymethoxylated stilbenes and dihydrostilbenes as potential anticancer agents stems from the discovery of many such natural products as antimitotic and antileukemic agents.¹⁻¹² This includes isolation of many stilbene derivatives, termed combretastatins, from the South African tree *Combretum caffrum*. Many of these combretastatins were found to be cytotoxic, with combretastatin A-4 (1a) the most potent.⁷



OMe Dihydrocombretastatin A-4 (1c)

This compound was found to cause mitotic arrest in L1210 murine leukemia cells, inhibit tubulin polymerization, and competitively inhibit the binding of radiolabeled colchicine to tubulin.⁷ It is presently being investigated under the sponsorship of the Cancer Research Campaign Clinical Trails scheme.¹³ A recent investigation of combretastatins revealed that combretastatin A-4 (1a) was active against

trans-Combretastatin A-4 (1b)

multidrug resistant (MDR) cancer cell lines.¹³ Combretastatin A-4 (1a) as well as its trans isomer 1b, its dihydro derivative 1c, and a number of related substances have been found to cause mitotic arrest in cells in culture at cytotoxic concentrations.^{1,2,4-9,13-17} Several other *trans*stilbenes and 1,4-diarylalkanes were also reported to be

- Pettit, G. R.; Singh, S. B.; Hamel, E.; Lin, C. M.; Alberts, D. S.; Garcia-Kendall, D. Experientia 1989, 45, 209.
- (2) Pettit, G. R.; Cragg, G. M.; Herald, D. L.; Schmidt, J. M.; Lohavanijaya, P. Can. J. Chem. 1982, 60, 1374.
- (3) Pettit, G. R.; Singh, S. B.; Niven, M. L.; Hamel, E.; Schmidt, J. M. J. Nat. Prod. 1987, 50, 119.
- (4) Pettit, G. R.; Singh, S. B.; Cragg, G. M. J. Org. Chem. 1985, 50, 3404.
- (5) Pettit, G. R.; Cragg, G. M.; Singh, S. B. J. Nat. Prod. 1987, 50, 386.
- (6) Pettit, G. R.; Singh, S. B. Can. J. Chem. 1987, 65, 2390.
- (7) Lin, C. M.; Singh, S. B.; Chu, P. S.; Dempcy, R. O.; Schmidt,
- J. M.; Pettit, G. R.; Hamel, E. Mol. Pharmacol. 1988, 34, 200. (8) Pettit, G. R.; Singh, S. B.; Schmidt, J. M.; Nivin, M. L.; Hamel,
- E.; Lin, C. M. J. Nat. Prod. 1988, 51, 517. (9) Itoh, Y.; Brossi, A.; Hamel, E.; Lin, C. M. Helv. Chem. Acta
- 1988, 71, 1199. (10) Bai, R.; Pettit, G. R.; Hamel, E. Biochem. Pharmacol. 1990, 39,
- 1941. (11) Ferrigni, N. R.; McLaughlin, J. L.; Powell, R. G.; Smith, C. R.,
- (11) Ferrigni, N. R.; McLaughin, J. L.; Fowen, R. G.; Shith, C. R., Jr. J. Nat. Prod. 1984, 47, 347.
- (12) Gill, M. T.; Bajaj, R.; Chang, C. J.; Nichols, D. E.; McLaughlin, J. L. J. Nat. Prod. 1987, 50, 36.
- (13) McGown, A. T.; Fox, B. W. Cancer Chemother. Pharmacol. 1990, 26, 79.
- (14) Lin, C. M.; Ho, H. H.; Pettit, G. R.; Hamel, E. Biochemistry 1989, 28, 6984.
- (15) Hamel, E.; Lin, C. M. Biochem. Pharmacol. 1983, 32, 3864.
 (16) McGown, A. T.; Fox, B. W. Anti-Cancer Drug Design 1989, 3,
- 249.
- (17) Jiang, J. B.; Hesson, D. P.; Dusak, B. A.; Dexter, D. L.; Kang, G. J.; Hamel, E. J. Med. Chem. 1990, 33, 1721.

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cytotoxic agents.^{7,18,19} Apart from these reports, many natural products possessing a trimethoxybenzene ring, e.g., colchicine, podophyllotoxin, and steganacin, were found to be potent cytotoxic agents and exert their antitumor property by their antitubulin character.^{1,2,4-9,13-17} Therefore, there has been considerable interest in the development of new cytotoxic and/or antitubulin agents based on these products as structural leads.^{7,9,20-22} Research in this direction has led to the discovery of several new anticancer agents and has provided insight into their structure-activity relationships.^{7,9,20-23}

Our interest in stilbenes began with piceatannol (2), a cytotoxic inhibitor of protein-tyrosine kinases,^{11,12,24} and the design and synthesis of superior inhibitors of such enzymes.^{25,26} The present paper describes the synthesis of an array of methoxylated stilbenes and related compounds as potential cytotoxic agents and their evaluation for the inhibition of tubulin polymerization and inhibition of binding of radiolabeled colchicine to tubulin.

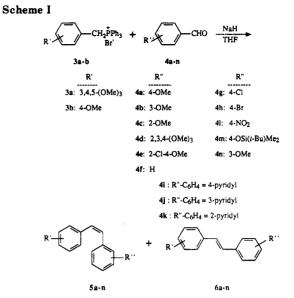
Our initial syntheses and evaluation of some cis- and trans-stilbenes as cytoxic agents showed that cis-stilbenes were much superior as cytotoxic agents when compared to the simultaneously obtained trans-stilbenes. Our new agents were thus clearly related to the potent antimitotic natural product derived from *Combretum caffrum*, combretastatin A-4.^{1,13,14} We therefore evaluated these initial compounds as inhibitors of tubulin-dependent reactions, and the results supported the conclusion that their mechanism of action involved inhibition of microtubule assembly.

These findings have led us to begin an extensive synthetic program and biological and biochemical evaluation of *cis*-stilbenes and dihydrostilbenes derived from them, since a number of the latter agents also show significant toxic effects on cells in culture. Our goals are the further definition of structure-activity relationships among this class of compounds with their target protein, tubulin, and the development of effective agents for the treatment of cancer. Our initial results with 16 *cis*-stilbenes, 17 dihydrostilbenes, and 36 related compounds are presented here.

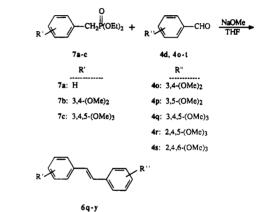
Chemistry

Wittig reaction of phosphonium bromides 3a,b with aryl aldehydes 4a-n in THF in the presence of sodium hydride followed by preparative thin layer chromatographic separation gave the corresponding *cis*-stilbenes 5a-n and *trans*-stilbenes 6a-n (Scheme I).²⁷ In general all these reactions gave the cis isomers as major components, and, except in a few cases, trans isomers were also isolated as

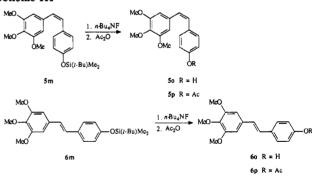
- (18) Klaus, M.; Mohr, P.; Weiss, E. Eur. Pat. Appl. EP 331,983, 1989.
- (19) Neiss, E. S.; Allen, L. M. Eur. Pat. Appl. EP 297,733, 1989.
 (20) Jardine, I. In Anticancer Agents Based on Natural Product Models; Cassady, J. M., Douros, J. J., Ed.; Academic Press:
- New York, 1980; p 319.
 (21) Batra, J. K.; Kang, G. J.; Jurd, L.; Hamel, E. Biochem. Pharmacol. 1988, 37, 2595.
- (22) Wang, Z.-Q.; Kuo, Y.-H.; Schnur, D.; Bowen, J. P.; Liu, S.-Y.; Han, F.-S.; Chang, J.-Y.; Cheng, Y.-C.; Lee, K.-H. J. Med. Chem. 1990, 33, 2660.
- (23) Denny, W. A. Drug Design Delivery 1988, 3, 99.
- (24) Geahlen, R. L.; McLaughlin, J. L. Biochem. Biophys. Res. Commun. 1989, 165, 241.
- (25) Cushman, M.; Chinnasamy, P.; Chakraborti, A. K.; Jurayj, J.; Geahlen, R. L.; Haugwitz, R. D. Int. J. Pept. Protein Res. 1990, 36, 538.
- (26) Cushman, M.; Nagarathnam, D.; Burg, D. L.; Geahlen, R. L. J. Med. Chem. 1991, 34, 798.
- (27) Asakawa, Y.; Tanikawa, K.; Aratani, T. Phytochemistry 1976, 15, 1057.



Scheme II



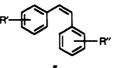
Scheme III



minor products in yields of over 10%. However, in the case of aryl aldehydes with a substituent at the 2-position (4c, 4d, and 4e) and pyridine-2-carboxaldehyde, cis isomers were obtained in very high yields, and the trans isomers were obtained in poor yields. With 2,3,4-trimethoxybenzaldehyde (4d), an isolable amount of the *trans*-stilbene was not obtained. *trans*-Stilbenes 6q-y were prepared by the Wittig-Horner reaction of the phosphonate esters 7a-c with the aryl aldehydes 4d and 4o-t in DMF using sodium methoxide as the base (Scheme II).²⁸ Under these reaction conditions, trans isomers were obtained exclusively. 4'-Hydroxystibenes 5o and 6o were prepared from the corresponding O-silyloxylated stilbenes 5m and 6m by the

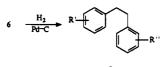
⁽²⁸⁾ Bachelor, F. W.; Loman, A. A.; Snowdon, L. R. Can. J. Chem. 1970, 48, 1554.

Table I. cis-Stilbenes



			cytotoxicity (ED ₅₀ in μ M)					
no.	R′	R″	A-549	MCF-7	HT-29	SKMEL-5	MLM	mp, °C
5a	3,4,5-(OMe) ₃	4-OMe	2.2×10^{-5}	1.2×10^{-6}	2.7 × 10 ⁻⁶	9.7×10^{-7}	9.3 × 10 ⁻⁶	oil
5b	3,4,5-(OMe) ₃	3-OMe	1.3×10^{-1}	1.4×10^{-1}	9.0×10^{-2}	6.0×10^{-2}	1.4	oil
5c	3,4,5-(OMe) ₃	2-OMe	1.1	1.3	8.7×10^{-1}	1.2	8.6	oil
5d	2,3,4-(OMe) ₃	4-OMe	9.7×10^{-1}	2.3×10^{-1}	1.0	1.1	10.9	55-7
5e	3,4,5-(OMe) ₃	2-Cl-4-OMe	5.1×10^{-2}	4.6×10^{-2}	6.6×10^{-2}	1.7×10^{-2}	1.4×10^{-1}	oil
5f	3,4,5-(OMe) ₃	н	1.7×10^{-1}	2.5×10^{-1}	8.4×10^{-2}	1.2×10^{-1}	>25	oil
5g	3,4,5-(OMe) ₃	4-Cl	8.0×10^{-2}	1.8×10^{-1}	5.0×10^{-2}	1.0×10^{-2}	1.7×10^{-1}	oil
5ĥ	3,4,5-(OMe) ₃	4-Br	1.1×10^{-2}	1.6×10^{-2}	8.2×10^{-3}	6.7 × 10 ⁻⁸	1.4×10^{-2}	oil
5i	1-(4-pyridyl)-2-(3,4,5-1	trimethoxyphenyl)ethene	14.2	17.0	12.9	14.4	>25	oil
5j	1-(3-pyridyl)-2-(3,4,5-trimethoxyphenyl)ethene		13.7	14.7	9.8	6.0	>25	oil
5k	1-(2-pyridyl)-2-(3,4,5-1	rimethoxyphenyl)ethene	>25	>25	>25	>25	>25	oil
51	3,4,5-(OMe) ₃	4-NO ₂	>25	>25	>25	>25	>25	140-2
5m	3,4,5-(OMe) ₃	$4-OSi(t-Bu)Me_2$	10.95	7.29	10.60	7.59	17.23	oil
5 n	3,4,5-(OMe) ₃	4-NMe ₂	4.1×10^{-3}	5.8×10^{-3}	8.1 × 10 ^{−3}	1.5×10^{-4}	9.4 × 10 ⁻³	oil
50	3,4,5-(OMe) ₃	4-0H	12.70	5.70	1.75	2.27	12.60	148-50
5p	3,4,5-(OMe) ₃	4-OAc	1.7	3.0×10^{-1}	6.0	6.0×10^{-1}	6.3	oil
1a		astatin A-4	1.2×10^{-6}	3.8×10^{-6}	1.2×10^{-6}	3.0×10^{-8}	1.4×10^{-5}	
	adriamycin		2.9×10^{-2}	3.1×10^{-2}	5.5 × 10 ⁻²	$3.2 imes 10^{-2}$	1.3×10^{-1}	

Scheme IV

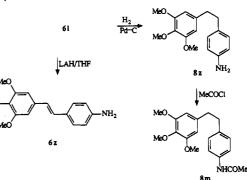


reaction of tetra-n-butylammonium fluoride in THF. In another set of experiments, 4'-acetoxystilbenes 5p and 6p were prepared by acetylation of 4'-hydroxystilbenes 50 and 60 (Scheme III). Cis or trans geometries of most of these compounds were confirmed by their characteristic coupling constants for the olefinic protons of about 12 Hz for cis and 16.0-16.5 Hz for trans isomers.²⁹ The two olefinic protons of compounds 5d, 5m, 6g, and 6p gave singlets and those of compounds 50 and 6b gave multiplets, and the geometries of these compounds were assigned relative to their isomers, which gave distinct doublets with characteristic coupling constants. Catalytic hydrogenation of (E)-stilbenes 6 at about 40 psi in the presence of 10%palladium on charcoal gave dihydrostilbenes 8 (Scheme IV). Lithium aluminum hydride reduction of (E)-4'nitro-3,4,5-trimethoxystilbene (61) provided (E)-4'amino-3,4,5-trimethoxystilbene (6z). Catalytic hydrogenation of compound 61 in EtOAc at 40 psi in the presence of 10% palladium on charcoal gave 4'-amino-3,4,5-trimethoxydihydrostilbene (8z), which on subsequent reaction with acetyl chloride gave the acetamido compound 8m (Scheme V). Scheme VI describes the general method adopted for the preparation of amides 11a-f and their subsequent reduction to substituted benzylamines 12a-f.

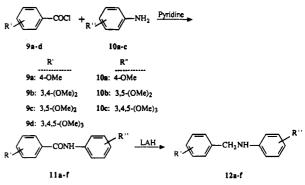
Results and Discussion

In the present study we synthesized 16 cis-stilbenes, 24 trans-stilbenes, 17 dihydrostilbenes, six N-phenylbenzamides, and six N-phenylbenzylamines and tested them against the five cancer cell cultures A-549 lung carcinoma, MCF-7 breast carcinoma, HT-29 colon adenocarcinoma, SKMEL-5 melanoma, and MLM melanoma. The cytotoxicity results are summarized in Tables I-IV. Among





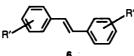




the 69 compounds tested in the present study, 11 of them, 5a, 5b, 5e-h, 5n, 6n, 8a, 8n, and 12a, were found to have significant cytotoxicity ($ED_{50} < 1 \mu M$ in at least three cell lines). In general, *cis*-stilbenes were more potent than the other groups of compounds, and (Z)-1-(4-methoxyphenyl)-2-(3,4,5-trimethoxyphenyl)ethene (5a) was the most potent of all. Taking compound 5a, a close analogue of combretastatin A-4 (1a), as the model compound for structure-activity relationship discussion, the presence of a 4-methoxy group in the B ring plays a very important role for this compound to be highly cytotoxic. Transfer of the 4-methoxy group in the B ring to the 3- or 2-position (compounds 5b and 5c) or substitution of it with H, NO₂, OSi(*t*-Bu)Me₂, OH, OAc (compounds 5f, 5l, 5m, 5o, and 5p) decreased the activity drastically. Similarly, intro-

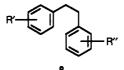
⁽²⁹⁾ Silverstein, E. M.; Bassler, G. C.; Morrill, T. C. Spectrometric Identification of Organic Compounds; John Wiley and Sons: New York, 1981; pp 264-265.

Table II. trans-Stilbenes



	R′	R″	cytotoxicity (ED ₅₀ in μ M)					
no.			A-549	MCF-7	HT-29	SKMEL-5	MLM	mp, °C
6a	3,4,5-(OMe) ₃	4-OMe	1.18	1.05	1.82	8.1×10^{-1}	2.07	152-538,39
6b	3,4,5-(OMe) ₃	3-OMe	9.8	12.2	7.3	10.5	>25	123-5
6c	3,4,5-(OMe) ₃	2-OMe	12.2	18.0	12.1	13.5	>25	oil
6e	3,4,5-(OMe) ₃	2-Cl-4-OMe	>25	>25	>25	>25	>25	oil
6f	3,4,5-(OMe) ₃	н	>25	>25	>25	>25	>25	105-6 ²⁸
6g	3,4,5-(OMe) ₃	4-Cl	>25	>25	>25	>25	>25	147 9
6 h	3,4,5-(OMe) ₃	4-Br	6.47	9.14	12.69	6.53	5.13	155-6
6i		rimethoxyphenyl)ethene	>25	>25	>25	>25	>25	247-8 ⁴⁷
6j	1-(3-pyridyl)-2-(3,4,5-t	rimethoxyphenyl)ethene	>25	>25	>25	>25	>25	103-548
61	3,4,5-(OMe) ₃	4-NO ₂	>25	>25	>25	>25	>25	192-4 ⁴⁰
6m	3,4,5-(OMe) ₃	4-OSi(t-Bu)Me ₂	>25	>25	>25	>25	>25	oil
6 n	3,4,5-(OMe) ₃	4-NMe ₂	6.1×10^{-3}	8.2×10^{-2}	6.9 × 10 ⁻³	4.6×10^{-3}	1.25×10^{-2}	114-5
60	3,4,5-(OMe) ₃	4-OH	>25	18.63	>25	11.55	24.15	188 -90⁸
6p	3,4,5-(OMe) ₃	4-OAc	9.7	9.6	5.4	4.6	13.0	oil
6q	3,4-(OMe) ₂	н	>25	>25	>25	>25	>25	106-8 ⁴¹
6r	2,3,4-(OMe) ₃	н	>25	>25	>25	>25	>25	7 9 -82
68	3,4,5-(OMe) ₃	3,5-(OMe) ₂	>25	>25	>25	>25	>25	132-4 ⁴⁹
6t	3,4,5-(OMe) ₃	2,3,4-(OMe) ₃	12.5	14.72	10.27	10.64	23.86	87-8
6u	3,4,5-(OMe) ₃	3,4,5-(OMe) ₃	>25	>25	>25	>25	>25	174-550
6v	3,4,5-(OMe) ₃	2,4,5-(OMe) ₃	>25	>25	>25	>25	>25	147-8
6w	2,4,5-(OMe) ₃	Н	>25	8.5	>25	>25	>25	81-2
6x	2,4,6-(OMe) ₃	н	>25	>25	>25	>25	>25	57 -9^{51,52}
6y	3,4-(OMe) ₂	3,5-(OMe) ₂	>25	>25	>25	>25	>25	65-7 ⁴⁹
6z	3,4,5-(OMe) ₃	NH ₂	>25	>25	>25	>25	>25	251-3
	adriamycin		2.9×10^{-2}	3.1×10^{-2}	5.5×10^{-2}	3.2×10^{-2}	1.3×10^{-1}	

Table III. Dihydrostilbenes

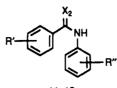


	R′	R″	cytotoxicity (ED ₅₀ in μ M)					
no.			A-549	MCF-7	HT-29	SKMEL-5	MLM	mp, °C
8a.	3,4,5-(OMe) ₃	4-0Me	1.8 × 10 ⁻⁴	1.6 × 10 ⁻⁴	2.5 × 10 ⁻⁴	1.4 × 10-4	1.8×10^{-4}	73-539
8b	3,4,5-(OMe) ₃	3-OMe	11.7	12.4	7.6	9.2	>25	oil
8c	3,4,5-(OMe) ₃	2-OMe	13.5	11.8	>25	20	>25	oil
8f	3,4,5-(OMe) ₃	н	>25	>25	>25	>25	>25	oil
8i	1-(4-pyridyl)-2-(3.4.5-tr	imethoxyphenyl)ethane	>25	>25	>25	>25	>25	oil
8j		imethoxyphenyl)ethane	12.2	>25	>25	>25	>25	oil ⁴⁸
8m	3,4,5-(OMe),	4-NĤCOČH ₃	>25	>25	>25	>25	>25	112-4
8n	3,4,5-(OMe) ₃	4-NMe ₂	8.3×10^{-2}	6.4×10^{-2}	7.7×10^{-2}	5.9×10^{-2}	1.2×10^{-1}	oil
80	3,4,5-(OMe) ₃	4-OH	>25	>25	>25	>25	>25	108-108
8p	3,4,5-(OMe)	4-OAc	19.0	>25	>25	>25	>25	oil
8q	3,4-(OMe) ₂	н	>25	>25	>25	>25	>25	oil
8 r	2,3,4-(OMe) ₃	н	>25	>25	>25	>25	>25	oil
88	3,4,5-(OMe) ₃	3,5-(OMe) ₂	>25	>25	>25	>25	>25	76-7 ⁴⁹
8t	3,4,5-(OMe) ₃	2,3,4-(OMe) ₃	>25	>25	>25	>25	>25	oil
8u	3,4,5-(OMe) ₃	3,4,5-(OMe) ₃	>25	>25	>25	>25	>25	137853
8v	3,4,5-(OMe) ₃	2,4,5-(OMe) ₃	>25	>25	>25	>25	>25	88-9
8z	3,4,5-(OMe) ₃	4-NH ₉	12.23	11.88	24.56	12.65	>25	84-5
le		dihydrocombretastatin A-4		3.3×10^{-1}	8.1×10^{-3}	2.1×10^{-3}	1.0×10^{-2}	
		mycin	1.0×10^{-2} 2.9×10^{-2}	3.1×10^{-2}	5.5×10^{-2}	3.2×10^{-2}	1.3×10^{-1}	

duction of a Cl group at the 2-position of 5a (compound 5e) decreased the cytotoxicity. However, when the methoxy group in the B ring was substituted with a Cl, Br, or NMe₂ group (compounds 5g, 5h, 5n), although the potency decreased they were still highly cytotoxic (ED_{50} < 10⁻¹ μ M). Rotating the three A-ring methoxy groups from the 3,4,5-positions to the 2,3,4-positions reduced the cytotoxicity by more than 5 orders of magnitude. In another modification we replaced the B ring of compound 5a with 4-, 3-, or 2-pyridyl rings (compounds 5i, 5j, 5k) and none of them were active ($ED_{50} > 10 \mu$ M). These results show that the exact locations of the four methoxy groups are very important features for the pronounced cytotoxicity of compound 5a and that changes in their locations result in decreased potency. In comparison with combretastatin A-4 (1a), compound 5a was found to be approximately 140 times more cytotoxic against HT-29 cells and about 10 times more cytotoxic against MCF-7 cells than combretastatin A-4 (1a). However, 5a was found to be about 20 times less cytotoxic against A-549 cells, 30 times cytotoxic against SKMEL-5, and 7 times less cytotoxic against MLM cells than combretastatin A-4 (1a).

Except for compound 6n, *trans*-stilbenes were essentially inactive. This includes tetramethylated piceatannol (6y) and its methoxylated derivatives 6s, 6t, 6u, and 6v. Only two dihydrostilbenes (compounds 8a and 8n) were found

Table IV. Benzamides and Benzylamines



11, 12

	R′	R″	cytotoxicity (ED ₅₀ in μ M)						
no.			$\overline{\mathbf{X}_2}$	A-549	MCF-7	HT-29	SKMEL-5	MLM	mp, °C
11a	3,4,5-(OMe) ₃	4-OMe	0	14.07	12.21	22.65	>25	>25	160-142
11b	3,4-(OMe) ₂	3,4,5-(OMe) ₃	0	>25	>25	>25	>25	>25	155-6
11c	4-OMe	$3,4,5-(OMe)_3$	0	>25	>25	>25	>25	>25	159-160
11 d	3,5-(OMe) ₂	4-OMe	0	9.02	5.47	14.52	11.88	>25	104-5
11e	4-OMe	3,5-(OMe) ₂	0	>25	>25	>25	>25	>25	105-6
11 f	3,4,5-(OMe) ₃	3,4,5-(OMe) ₃	0	>25	>25	>25	>25	>25	211-2
1 2a	$3,4,5-(OMe)_3$	4-OMe	H_2	1.9×10^{-3}	2.4×10^{-3}	1.0×10^{-3}	7.0×10^{-4}	1.6×10^{-3}	73-4
12b	3,4-(OMe) ₂	3,4,5-(OMe) ₃	H_2	>25	>25	>25	>25	>25	oil
12c	4-OMe	3,4,5-(OMe) ₃	H_2	8.3×10^{-1}	8.6×10^{-1}	2.24	1.13	>25	77-8
12d	3,5-(OMe) ₂	4-OMe	H_2	5.71	6.08	17.84	2.41	>25	oil
12e	4-OMe	3,5-(OMe) ₂	H_2	>25	>25	>25	>25	>25	oil
1 2f	3,4,5-(OMe) ₃	3,4,5-(OMe) ₃	H ₂	>25	>25	>25	>25	>25	127-8
		mycin	•	2.9×10^{-2}	3.1×10^{-2}	5.5×10^{-2}	3.2×10^{-2}	1.3×10^{-1}	

to be active, with 8a being the second most cytotoxic agent we prepared (ED₅₀ values about $2 \times 10^{-4} \mu$ M). Compound 8a was more cytotoxic than dihydrocombrestatin A-4 (1c) in all five cancer cell lines studied here. When the ethylene bridge in compound 8a was replaced with an amide or an aminomethylene linkage (compounds 11a, 11c, 12a, 12c, and other analogues), none of the amides had significant activity (ED₅₀ > 1 μ M), but the N-benzylamine derivative 12a possessing the closest structural analogy to 8a was active. 3,4,5-Trimethoxy-N-(4-methoxyphenyl)benzylamine (12a) was only an order of magnitude less cytotoxic than 8a (ED₅₀ in the $10^{-3} \mu$ M range).

The mechanism of action of the combretastatins has been shown to be at the microtubule level, since they cause cells to accumulate in apparent metaphase arrest and inhibit in vitro microtubule assembly.^{1,7} They bind specifically to tubulin, the major component of microtubules, at the colchicine binding site, since combretastatin A-4 (1a) has been shown to competitively inhibit the binding of radiolabeled colchicine to tubulin.¹⁴

Initial investigation of several of the synthetic compounds prepared here revealed that they do in fact cause mitotic arrest in cell culture. We therefore performed a detailed quantitative study of the effects of most of these substances on tubulin polymerization. With the exception of compounds 5p and 12d, noncytotoxic agents had minimal effects on polymerization (IC₅₀ values > 50 μ M), but significant inhibition of the reaction occurred with 10 of the 11 highly cytotoxic compounds and with compounds 5p and 12d. Tubulin polymerization and colchicine binding inhibition data of the compounds we prepared for this study are presented in comparison with simultaneously obtained inhibitory data for the effects of combretastatin A-4 (1a; cf. 5a), its trans isomer (1b, cf. 6a), and its dihydro derivative (1c; cf. 8a) (Table V). Data are presented as well for podophyllotoxin, a potent tubulin inhibitor that binds at the colchicine site,³⁰ and for thiocolchicine, a particularly potent colchicinoid that has reproducibly yielded the lowest IC_{50} value in the polymerization assay for agents binding to the colchicine binding site.^{31,32}

Table V. Effects of Compounds **5a**, **5b**, **5e**, **5f**, **5g**, **5h**, **5n**, **5p**, **6a**, **6n**, **8a**, **8n**, **12a**, **12c**, and **12d** on Tubulin Polymerization and on the Binding of Radiolabeled Colchicine to Tubulin^a

compd	tubulin polymerization IC ₅₀ (μ M) (± S. D.)	colchicine binding % inhibition
5a	2.2 (±0.07)	95
5b	8.8 (±1)	50
5e	3.5 (±0.3)	73
5f	36 (±1)	14
5 g	4.8 (±0.3)	55
5h	$3.1 (\pm 0.1)$	73
5n	3.4 (±0.1)	83
5p	29 (±5)	24
6a	>50	
6n	>50	
8a	7. 9 (±0.8)	65
8 n	29 (±1)	31
12a	23 (±0.5)	34
12c	>50	
12d	29 (±2)	39
combretastatin A-4 (1a)	1.9 (±0.2)	99
1b	>50	
1c	3.3 (±0.2)	79
podophyllotoxin	$2.1 (\pm 0.1)$	88
thiocolchicine	$1.4 (\pm 0.08)$	57

^a The IC₅₀ values for tubulin polymerization were determined as described in the text, with full details presented elsewhere.³¹ For the colchicine binding assay, reaction mixtures (in triplicate) contained 1 μ M tubulin, 5 μ M [³H]colchicine, and 5 μ M inhibitor and were incubated for 10 min at 37 °C prior to analysis. Further details have been described previously.³²

Compound 5a was the most potent of the new agents as an inhibitor of tubulin polymerization, with an IC_{50} value $(2.2 \ \mu M)$ essentially indistinguishable from those of combretastatin A-4 and podophyllotoxin and somewhat higher than that of thiocolchicine. This is in agreement both with 5a being the most cytotoxic of the new compounds and with its close similarity to combretastatin A-4 (1a) in its overall effects on the cell lines we evaluated. The difference in IC_{50} values between the two dihydrostilbenes 1c and 8a was more noticeable. The combretastatin A-4 analogue 1c had an IC₅₀ value of 3.3 μ M, only modestly lower than the IC_{50} value of combretastatin A-4, but the corresponding hydrogenation of 5a to yield 8a resulted in an almost 4-fold increase in the IC_{50} value, from 2.2 to 7.9 μ M. Similarly the modest reduction in activity in the cis-stilbene 5n as compared to combretastatin A-4 (1a) (3.5

⁽³⁰⁾ Wilson, L. Biochemistry 1970, 9, 4999.

⁽³¹⁾ Muzaffar, A.; Brossi, A.; Lin, C. M.; Hamel, E. J. Med. Chem. 1990, 33, 567.

⁽³²⁾ Kang, G.-J.; Getahun, Z.; Muzaffar, A.; Brossi, A.; Hamel, E. J. Biol. Chem. 1990, 265, 10255.

versus $1.9 \ \mu$ M) was not reflected in the dihydrostilbene analogue 8n, which had an IC₅₀ value of 29 μ M. *cis*-Stilbenes 5b, 5e, 5g, and 5h were also active as inhibitors of tubulin polymerization, while the other ten *cis*-stilbenes had little or no activity. It should be noted that, with the exception of the most potent agents (1a and 5a), there was only qualitative agreement between the tubulin polymerization and cell culture assays. For example, while dihydrocombretastatin A-4 (1c) was more effective than compound 8a as an inhibitor of tubulin polymerization, the latter agent was more cytotoxic with the cell lines studied here. Similarly, although the halogenated *cis*stilbenes 5e, 5g, and 5h were not much less active than 1a and 5a as inhibitors of tubulin polymerization, they were about 1000-fold less cytotoxic.

The cytotoxic compounds gave reproducible results in the tubulin polymerization assay with the exception of the trans-stilbenes 1b and 6n. Initial evaluation of these compounds in the tubulin polymerization assay yielded results concordant with the cytotoxicity data, although the apparent IC₅₀ value obtained in the polymerization assay for 6n was difficult to reproduce and that for 1b initially obtained in the current experiments was lower than that obtained previously.⁷ Compound 6a, a close analogue of 1b, did not inhibit tubulin polymerization at all, and this led us to reevaluate the inhibitory effects of 1b and 6n. We found that both 1b and 6n solutions increased in activity with storage, and that, when care was taken to evaluate the solutions immediately after their preparation, neither of the trans-stilbene was able to significantly inhibit tubulin polymerization. This suggested that both compounds were unstable in solution and that more active agents might be formed during their storage. The cytotoxic properties of these two agents may similarly result from chemical changes in solution. NMR analysis (500 MHz) of 6n in solution demonstrated significant formation of the cis isomer 5n. The ratio of 6n:5n was 1:1 after 24 h of the dissolution of pure 6n in DMSO at room temperature. In a separate analysis of the stability of compound 1b in DMSO at room temperature (well protected from light), ¹H NMR analysis over a period of 1 month at frequent intervals confirmed the formation of about 3% and 10% of the cis isomer (compound 1a) after 2 and 4 weeks, respectively.

Compounds 5a and 8a can be taken as standards for structure-activity comparisons of *cis*-stilbenes and dihydrostilbenes, respectively, in the tubulin polymerization assay. Without exception, when the same modified analogue was available in both series, a greater loss of activity occurred in the dihydrostilbene than in the analogous *cis*-stilbene (cf. 5b and 6b; 5f and 6f; 5n and 6n; 5p and 6p). No modification among the compounds we have prepared thus far improved on the antitubulin activity of either 5a or 8a.

In the cis-stilbene series, a shift of a single methoxy group in the A ring, from position 5 to position 2, yielded an inactive agent (5d). When the B-ring methoxy group was shifted from position 4' to position 3', there was a 4-fold drop in activity (compound 5b; IC₅₀, 8.8 μ M). When the B-ring methoxy group was eliminated, there was a much larger drop in activity (compound 5f; IC₅₀, 36 μ M), while its placement at 2'-position yielded the inactive compound 5c. Addition of a Cl at position 2' (compound 5e) or replacement of the methoxy group with a Cl (5g), Br (5h), or NMe₂ (5n) group resulted in small reductions in antitubulin activity. Demethylation of the 4'-methoxy group led to loss of activity (compound 5o), and its replacement with an acetoxy group yielded a weak inhibitor (compound **5p**; IC₅₀, 29 μ M).

Turning to the dihydrostilbene series, replacement of the B-ring methoxy group with an amino group (compound 8z) resulted in loss of activity, but some activity returned if the amino group was converted to a dimethylamino group (compound 8n; cf. 5n). Addition of one (compound 8s) or two (compounds 8t-v) additional methoxy groups to the B ring also resulted in loss of activity. The only enhancement of antitubulin activity in the 5a/8a structure yet obtained by modification of the substituents on either phenyl ring has been addition of a single hydroxy group at position 3' [as occurs in combretastatin A-4 (1a) and dihydrocombretastatin A-4 (1c)] or addition of two hydroxy groups in a vicinal diol arrangement at positions 2' and 3' (as occurs in combretastatin A-1 and B-1), as described previously.^{3,7}

Replacement of the ethylene bridge connecting the two aromatic rings in 8a with amide or aminomethylene units as represented by structures 11a, 11c, and 12c resulted in loss of inhibitory activity in the tubulin polymerization assay. On the other hand, replacement of the ethylene bridge of 8a with an aminomethylene unit with the alternative orientation shown in structure 12a resulted in a 3-fold loss of activity (increase in the IC₅₀ value from 7.9 μ M for compound 8a to 23 μ M for compound 12a). Comparing compound 12a to compound 12d indicates that only a small loss of activity occurs with elimination of the 4methoxy group of the A ring (IC₅₀ of 29 μ M without the methoxy group as opposed to 23 μ M).

Combretastatin A-4 (1a), compound 1c, and related compounds all inhibit the binding of radiolabeled colchicine to tubulin. We therefore evaluated the new active agents in this assay, too, and they were all found to inhibit the binding reaction. Relative activity as inhibitors of colchicine binding correlated well with their activity as inhibitors of tubulin polymerization. The mechanism of action of the new agents, like that of the combretastatins, thus appears to involve an interaction of the drug with the colchicine binding site of tubulin. Only compound 5a, however, approached the nearly total inhibition of colchicine binding observed with equimolar combretastatin A-4 (1a).

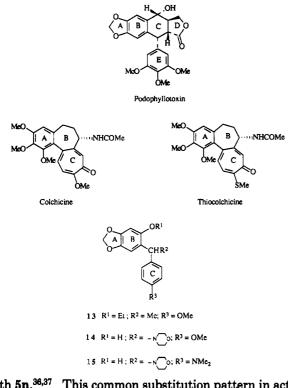
With the compounds described here, as with the combretastatins⁷ and other classes of antimitotic agents, 10,33there is only partial agreement between cytotoxicity and effects on tubulin, the presumptive target molecule. Seven of the most cytotoxic agents (**5a**, **5b**, **5e**, **5g**, **5h**, **5n**, and **8a**) were strong inhibitors of tubulin polymerization, and, except for the *trans*-stilbene **6n**, no compound inactive as inhibitor of tubulin polymerization had significant cytotoxic activity. Nevertheless, compounds **8n** and **12a** were strongly cytotoxic yet had only modest inhibitory effects on tubulin polymerization. Similarly, the structural differences between compounds **12a** and **12d** yielded only minor differences in antitubulin activity but resulted in major changes in their cytotoxic properties.

Besides the clear analogy of the compounds described here to the combretastatins, the activity observed in compound 5n, and to a lesser extent in compound 8n, suggests a relationship to the benzylbenzodioxole class of agents synthesized by Jurd.^{34,35} Among the active tubulin inhibitors were compounds 13, 14, and 15 (Chart I), with the latter having the dimethylamino substituent in common

- (34) Jurd, L.; Fye, R. L.; Morgan, J. J. J. Agric. Food Chem. 1979, 27, 1007.
- (35) Jurd, L. J. Heterocycl. Chem. 1985, 22, 993.

⁽³³⁾ Batra, J. K.; Powers, L. J.; Hess, F. D.; Hamel, E. Cancer Res. 1986, 46, 1889.

Chart I



with $5n.^{36,37}$ This common substitution pattern in active compounds makes it likely that the B ring of combretastatin derivatives is analogous to the C ring of the benzylbenzodioxole group of compounds.

The relative potencies 5a > 8a > 6a for these cis, dihydro, and trans compounds as inhibitors of tubulin polymerization are in agreement with the relative potencies previously observed⁷ for combretastatin A-4 (1a) and dihydrocombretastatin A-4 (1c) and our finding in the present study that freshly dissolved trans-combretastatin A-4 (1b) has little or no activity. We assume that the flexibility of the dihydro compound 8a allows it to adopt a conformation resembling the cis isomer 5a, which explains why 8a is more cytotoxic and potent as a tubulin polymerization inhibitor than the trans isomer 6a. The relative potencies 5a > 8a > 6a for these cis, dihydro, and trans compounds, respectively, as inhibitors of tubulin polymerization were also reflected in the results of the cytotoxicity assays. These relative potencies of 5a > 8a> 6a in the cytotoxicity assays are also in agreement with the relative cytotoxicities of 1a > 1c previously reported for L1210 murine leukemia cells in the combretastatin series, although in that study 1b was intermediate in cytotoxic activity between 1a and 1c.⁷

In summary, a series of stilbene derivatives has been prepared and tested for cytotoxicity in cancer cell cultures, for inhibition of tubulin polymerization, and for inhibition of colchicine binding to tubulin. The most potent of these substances was 5a, which was found to be almost as potent as combretastatin A-4 (1a), the most active of the combretastatins, as a tubulin polymerization inhibitor. Compound 5a was also determined to be slightly more cytotoxic than combretastatin A-4 (1a) in HT-29 colon adenocarcinoma and MCF-7 breast carcinoma cells and slightly less cytotoxic in A-549 lung carcinoma, SKMEL-5 melanoma, and MLM melanoma cells. It is clear from these results that the phenolic hydroxyl group of combretastatin A-4

(1a) is not a critical structural feature for either its anticancer activity or for its ability to inhibit tubulin polymerization. The relative potencies 5a > 8a > 6a for these cis, dihydro, and trans compounds, respectively, are in agreement with the previously established potencies for combretastatin A-4 (1a), dihydrocombretastatin A-4 (1a), and trans-combretastatin A-4 (1b) when tested for inhibition of tubulin polymerization. These studies contribute knowledge about the structure-activity relationships in the combretastatin series that may eventually allow the design and synthesis of a stilbene derivative having superior anticancer activity.

Experimental Section

Melting points were determined in capillary tubes on a Mel-Temp apparatus and are uncorrected. Spectra were obtained as follows: CI mass spectra on a Finnegan 4000 spectrometer; ¹H NMR spectra on a Chemagnetics A-200 or Nicolet QE-300 or Varian VXR-500S spectrometers with TMS as an internal standard in $CDCl_3$ or $DMSO-d_6$; IR spectra on a Beckman IR-33 spectrophotometer. Microanalyses were performed at the Purdue Microanalysis Laboratory, and all values were within $\pm 0.4\%$ of the calculated composition. All organic solvents were appropriately dried and/or purified prior to use. Diethyl benzylphosphonate (7a), aryl aldehydes 4a-t, and a 1 M solution of tetra-n-butylammonium fluoride in THF were obtained from commercial sources. Compounds 7b-c were prepared by the reaction of the corresponding benzyl bromides and triethyl phosphite.²⁸ Phosphonium bromides **3a**,**b** were prepared by stirring a mixture of triphenylphosphine and the corresponding benzyl bromides in toluene.²⁷ Combretastatin A-4 and its trans isomer were kindly provided by Prof. G. R. Pettit, Arizona State University. Compound 1c was prepared as described previously. Podophyllotoxin was obtained from Aldrich Chemical Co., and thiocolchicine was from Roussel-Uclaf. Preparative silica gel TLC plates (200 μ m) were purchased from Analtech.

General Procedure for the Preparation of (Z)-Stilbenes 6a-n and (E)-Stilbenes 5a-n. Sodium hydride (72 mg, 3 mmol) was added in portions to a well-stirred suspension of phosphonium bromide 3a,b (2.0 mmol) and aryl aldehyde (2.0 mmol) in benzene (20 mL) under an argon atmosphere at 0–5 °C, and the mixture was allowed to warm to room temperature. After an additional stirring for 16 h, excess sodium hydride was quenched by the addition of methanol (1 mL). Solvents from the reaction mixture were evaporated at reduced pressure, and the residue was purified by preparative thin layer chromatography using 5% EtOH in hexane as the eluent. Products 5d and 5l were obtained as solids, and all the other cis-stilbenes were obtained as viscous oils.

(Z)-1-(4-Methoxyphenyl)-2-(3,4,5-trimethoxyphenyl)ethene (5a): 400 mg; 66%; oil; ¹H NMR (CDCl₃, 500 MHz) δ 7.25 (d, J = 9 Hz, 2 H), 6.80 (d, J = 9 Hz, 2 H), 6.53 (d, J = 12Hz, 1 H), 6.51 (s, 2 H), 6.44 (d, J = 12 Hz, 1 H), 3.84 (s, 3 H), 3.79 (s, 3 H), 3.69 (s, 6 H); CIMS (isobutane) m/e 301 (MH⁺, 100). Anal. $(C_{18}H_{20}O_4)$ C, H.

(Z)-1-(3-Methoxyphenyl)-2-(3,4,5-trimethoxyphenyl)ethene (5b): 410 mg; 69%; oil; ¹H NMR (CDCl₃, 200 MHz) δ 7.18 (t, J = 7.9 Hz, 1 H), 6.91–6.83 (m, 2 H), 6.78–6.72 (m, 1 H), 6.58 (d, J = 12.2 Hz, 1 H), 6.50 (d, J = 12.2 Hz, 1 H), 6.49 (s, 2H), 3.83 (s, 3 H), 3.70 (s, 3 H), 3.67 (s, 6 H); ¹³C NMR (CDCl₃, 50 MHz) δ 159.94, 153.29, 139.63, 137.63, 132.84, 130.67, 130.22 129.63, 121.79, 114.24, 113.46, 106.42, 61.06, 56.01, 55.26; CIMS (isobutane) m/e 301 (MH⁺, 100). Anal. (C₁₈H₂₀O₄) C, H.

(Z)-1-(2-Methoxyphenyl)-2-(3,4,5-trimethoxyphenyl)ethene (5c): 440 mg; 73%; oil; ¹H NMR (CDCl₃, 500 MHz) δ 7.27-7.20 (m, 2 H), 6.90 (d, J = 8.4 Hz, 1 H), 6.82 (t, J = 8.4 Hz, 1 H), 6.65 (d, J = 12.2 Hz, 1 H), 6.54 (d, J = 12.2 Hz, 1 H), 6.47 (s, 2 H), 3.84 (s, 3 H), 3.82 (s, 3 H), 3.63 (s, 6 H); CIMS (isobutane) m/e 301 (MH⁺, 100). Anal. (C₁₈H₂₀O₄) C, H.

(Z)-1-(4-Methoxyphenyl)-2-(2,3,4-trimethoxyphenyl)ethene (5d): 460 mg; 77%; mp 55-7 °C; ¹H NMR (CDCl₃, 200 MHz) δ 7.19 (d, J = 8.9 Hz, 2 H), 6.94 (d, J = 8.7 Hz, 1 H), 6.75 (d, J = 8.9 Hz, 2 H), 6.52 (s, 2 H), 6.49 (d, J = 8.7 Hz, 1 H), 3.90(s, 3 H), 3.88 (s, 3 H), 3.84 (s, 3 H), 3.78 (s, 3 H); CIMS (isobutane) m/e 301 (MH⁺, 100). Anal. (C₁₈H₂₀O₄) C, H.

(Z)-1-(2-Chloro-4-methoxyphenyl)-2-(2,3,4-trimethoxy-

⁽³⁶⁾ Batra, J. K.; Jurd, L.; Hamel, E. Mol. Pharmacol. 1985, 27, 94.

⁽³⁷⁾ Batra, J. K.; Jurd, L.; Hamel, E. Biochem. Pharmacol. 1986, 35, 4013.

phenyl)ethene (5e): 420 mg; 63%; oil; ¹H NMR (CDCl₃, 500 MHz) δ 7.21 (d, J = 8.5 Hz, 1 H), 6.96 (d, J = 2.6 Hz, 1 H), 6.67 (dd, 1 H), 6.59 (d, J = 12.1 Hz, 1 H), 6.57 (d, J = 12.1 Hz, 1 H), 6.42 (s, 2 H), 3.82 (s, 3 H), 3.78 (s, 3 H), 3.66 (s, 6 H). Anal. (C₁₈H₁₉ClO₄) C, H.

(Z)-1-Phenyl-2-(3,4,5-trimethoxyphenyl)ethene (5f): 270 mg; 50%; oil; ¹H NMR (CDCl₃, 200 MHz) δ 7.35–7.25 (m, 5 H), 6.61 (d, J = 12.2 Hz, 1 H), 6.50 (d, J = 12.2 Hz, 1 H), 6.47 (s, 2 H), 3.83 (s, 3 H), 3.65 (s, 6 H); ¹³C NMR (CDCl₃, 50 MHz) δ 153.27, 137.86, 137.12, 132.84, 130.48, 130.36, 129.28, 128.62, 127.51, 106.35, 61.09, 55.96; CIMS (isobutane) m/e 271 (MH⁺, 100). Anal. (C₁₇H₁₈O₃) C, H.

(Z)-1-(4-Chlorophenyl)-2-(3,4,5-trimethoxyphenyl)ethene (5g): 307 mg; 50%; oil; ¹H NMR (CDCl₃, 200 MHz) δ 7.23 (s, 4 H), 6.55 (d, J = 12 Hz, H), 6.49 (d, J = 12 Hz, 1 H), 6.45 (s, 1 H), 3.84 (s, 3 H), 3.68 (s, 6 H); ¹³C NMR (CDCl₃, 50 MHz) δ 153.43, 137.77, 136.17, 133.19, 132.53, 131.20, 130.74, 128.96, 128.75, 106.26, 61.10, 56.05. Anal. (C₁₇H₁₇ClO₃) C, H.

(Z)-1-(4-Bromophenyl)-2-(3,4,5-trimethoxyphenyl)ethene (5h): 363 mg; 52%; oil; ¹H NMR (CDCl₃, 200 MHz) δ 7.38 (d, J = 8.6 Hz, 2 H), 7.16 (d, J = 8.6 Hz, 2 H), 6.56 (d, J = 12.1 Hz, 1 H), 6.47 (d, J = 12.1 Hz, 1 H), 6.44 (s, 2 H), 3.84 (s, 3 H), 3.68 (s, 6 H); ¹³C NMR (CDCl₃, 50 MHz) δ 153.42, 137.77, 136.63, 132.49, 131.71, 131.28, 131.02, 128.98, 121.29, 106.25, 61.09, 56.05; CIMS (isobutane) m/e 350 (93) 348 (MH⁺, 100). Anal. (C₁₇-H₁₇BrO₃) C, H.

(Z)-1-(4-Pyridyl)-2-(3,4,5-trimethoxyphenyl)ethene (5i): 277 mg; 51%; oil; ¹H NMR (CDCl₃, 500 MHz) δ 8.49 (d, J = 6.0 Hz, 2 H), 7.18 (d, J = 6.0 Hz, 2 H), 6.69 (d, J = 12.2 Hz, 1 H), 6.48 (d, J = 12.2 Hz, 1 H), 6.42 (s, 2 H), 3.84 (s, 3 H), 3.66 (s, 6 H); CIMS (isobutane) m/e 272 (MH⁺, 100). Anal. (C₁₆H₁₇NO₃) C, H.

(Z)-1-(3-Pyridyl)-2-(3,4,5-trimethoxyphenyl)ethene (5j): 292 mg; 54%; oil; ¹H NMR (CDCl₃, 500 MHz) δ 8.53 (s, 1 H), 8.43 (d, J = 4.8 Hz, 1 H), 7.60 (d, J = 7.9 Hz, 1 H), 7.18 (dd, J₁ = 4.8 Hz, J₂ = 7.9 Hz, 1 H), 6.67 (d, J = 12.2 Hz, 1 H), 6.53 (d, J = 12.2 Hz, 1 H), 6.41 (s, 2 H), 3.84 (s, 3 H), 3.67 (s, 6 H); CIMS (isobutane) m/e 272 (MH⁺, 100). Anal. (C₁₆H₁₇NO₃) C, H.

(Z)-1-(2-Pyridyl)-2-(3,4,5-trimethoxyphenyl)ethene (5k): 351 mg; 65%; oil; ¹H NMR (CDCl₃, 500 MHz) δ 8.64 (d, J = 4.7 Hz, 1 H), 7.58–7.54 (dt, $J_1 = 7.5$ Hz, $J_2 = 1.8$ Hz, 1 H), 7.32–7.30 (m, 1 H), 7.17–7.15 (m, 2 H), 6.79 (d, J = 12.4 Hz, 1 H), 6.58 (s, 2 H), 3.89 (s, 3 H), 3.74 (s, 6 H); CIMS (isobutane) m/e 272 (MH⁺, 100). Anal. (C₁₆H₁₇NO₃) C, H.

(Z)-1-(4-Nitrophenyl)-2-(3,4,5-trimethoxyphenyl)ethene (51): 170 mg; 27%; mp 140–142 °C; ¹H NMR (CDCl₃, 300 MHz) δ 8.15 (d, J = 8.6 Hz, 2 H), 7.56 (d, J = 8.6 Hz, 2 H), 7.05 (d, J = 12 Hz, 1 H), 6.95 (d, J = 12 Hz, 1 H), 6.71 (s, 2 H), 3.86 (s, 6 H), 3.82 (s, 3 H); CIMS (isobutane) m/e 316 (MH⁺, 100). Anal. (C₁₇H₁₇NO₅) C, H.

(Z)-1-[4-[(*tert*-Butyldimethylsilyl)oxy]phenyl]-2-(3,4,5trimethoxyphenyl)ethene (5m): 429 mg; 53%; oil; IR (Neat) 2980, 2960, 1610, 1580, 1520, 1470, 1420, 1360, 1330, 1270, 1140 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 7.17 (d, J = 8.0 Hz, 2 H), 6.73 (d, J = 8.0 Hz, 2 H), 6.51 (s, 2 H), 6.41 (s, 2 H), 3.84 (s, 3 H), 3.79 (s, 6 H), 0.99 (s, 9 H), 0.14 (s, 6 H). Anal. (C₂₃H₃₂O₄Si) C, H.

(Z)-1-[4-(Dimethylamino)phenyl]-2-(3,4,5-trimethoxyphenyl)ethenes (5n): 450 mg; 72%; oil; ¹H NMR (CDCl₃, 500 MHz) δ 7.22 (d, J = 8.9 Hz, 2 H), 6.61 (d, J = 8.9 Hz, 2 H), 6.58 (s, 2 H), 6.41 (d, J = 12.1 Hz, 1 H), 6.34 (d, J = 12.1 Hz, 1 H), 3.85 (s, 3 H), 3.72 (s, 6 H), 2.93 (s, 6 H); CIMS (isobutane) m/e314 (MH⁺, 100). Anal. (C₁₉H₂₃NO₃) C, H.

(E)-1-(4-Methoxyphenyl)-2-(3,4,5-trimethoxyphenyl)ethene (6a): 67 mg; 11%; mp 152-5 °C (lit.^{38,39} oil); ¹H NMR (CDCl₃, 500 MHz) δ 7.45 (d, J = 8.5 Hz, 2 H), 6.97 (d, J = 16.0 Hz, 1 H), 6.91 (d, J = 16.0 Hz, 1 H), 6.90 (d, J = 8.5 Hz, 2 H), 6.72 (s, 2 H), 3.87 (s, 6 H), 3.84 (s, 3 H), 3.82 (s, 3 H); CIMS (isobutane) m/e 301 (MH⁺, 100). Anal. (C₁₈H₂₀O₄) C, H.

(E)-1-(4-Nitrophenyl)-2-(3,4,5-trimethoxyphenyl)ethene (61): 280 mg; 44%; mp 192-4 °C (lit.⁴⁰ mp 186-187 °C).

(40) Schiemenz, G. P.; Thobe, J. Chem. Ber. 1966, 99, 2663.

(*E*)-1-[4-[(*tert*-Butyldimethylsilyl)oxy]phenyl]-2-(3,4,5trimethoxyphenyl)ethene (6m): 218 mg, 27%; oil; IR (neat) 2980, 2960, 1605, 1585, 1520, 1470, 1420, 1340, 1320, 1270, 1170, 1130, 1010 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 6.89 (d, *J* = 16.5 Hz, 1 H), 6.81 (d, *J* = 16.5 Hz, 1 H), 7.31 (d, *J* = 8.5 Hz, 2 H), 6.76 (d, *J* = 8.5 Hz, 2 H), 6.64 (s, 2 H), 3.84 (s, 3 H), 3.79 (s, 6 H), 0.92 (s, 9 H), 0.14 (s, 6 H). Anal. (C₂₃H₃₂O₄Si) C, H.

(E)-1-[4-(Dimethylamino)phenyl]-2-(3,4,5-trimethoxyphenyl)ethene (6n): 145 mg; 23%; mp 114-5 °C; IR (KBr) 3000, 2980, 2940, 2860, 1600, 1580, 1520, 1340, 1240, 1120, 960 cm⁻¹; ¹H NMR (DMSO- d_6 , 500 MHz) δ 7.42 (d, J = 8.85 Hz, 2 H), 7.10 (d, J = 16.3 Hz, 1 H), 6.92 (d, J = 16.3 Hz, 1 H), 6.86 (d, J = 8.8 Hz, 2 H), 6.61 (s, 2 H), 3.83 (s, 6 H), 3.67 (s, 3 H), 2.94 (s, 6 H); CIMS (isobutane) m/e 314 (MH⁺, 100), 313 (72). Anal. (C₁₉H₂₃NO₃) C, H.

(Z)-1-(4-Hydroxyphenyl)-2-(3,4,5-trimethoxyphenyl)ethene (50). A solution of *n*-Bu₄NF in THF (1 M, 2 mL, 2 mmol) was added to a stirred solution of silyl ether 5m (372 mg, 1 mmol) in THF (5 mL) at room temperature and the stirring was continued for 30 min. Solvent was removed at reduced pressure, the resulting residue was treated with 20 mL of water, and the product was extracted with EtOAc (2 × 20 mL). The EtOAc solution was dried (MgSO₄) and concentrated and the residue was crystallized from EtOAc/hexane to give 5n (217 mg, 76%): mp 148-150 °C; IR (KBr) 3440, 3020, 2940, 2840, 1610, 1580, 1510, 1420, 1330, 1230, 1160, 1120, 980, 790, 750 cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ 7.40 (bs, 1 H), 7.24 (d, J = 8.1 Hz, 2 H), 7.10 (d, J = 8.1 Hz, 2 H), 6.60–6.30 (m, 4 H), 3.80 (s, 6 H), 3.76 (s, 3 H); CIMS (isobutane) m/e 287 (MH⁺, 100). Anal. (C₁₇H₁₈O₄) C, H.

(E)-1-(4-Hydroxyphenyl)-2-(3,4,5-trimethoxyphenyl)ethene (60). With use of the same procedure as described for 50, compound 60 was prepared from 6m in 1-mmol scale (228 mg, 80%): mp 188-190 °C (lit.⁸ mp 188-90 °C).

General Procedure for the Preparation of Acetates 5p and 6p. A solution of n-Bu₄NF in THF (1 M, 2 mL, 2 mmol) was added to a solution of stilbenes 5m/6m (400 mg, 1 mmol) in THF (5 mL) and the mixture was stirred at 0 °C. After 30 min, acetic anhydride (0.5 mL) was added, and the stirring was continued at room temperature for 24 h. Solvents were evaporated at reduced pressure, and the residue was mixed with water (50 mL). The product was extracted with ether (2 × 25 mL), and the ether solution was washed with water (2 × 100 mL). Evaporation of the solvents and purification of the crude product by preparative TLC using 40% ethyl acetate in hexanes as the eluent afforded the desired products.

(Z)-1-(4-Åcetoxyphenyl)-2-(3,4,5-trimethoxyphenyl)ethene (5p): 93 mg; 28%; oil; ¹H NMR (CDCl₃, 500 MHz) δ 7.30 (d, J = 8.3 Hz, 2 H), 6.98 (d, J = 8.3 Hz, 2 H), 6.57 (d, J = 12.1 Hz, 1 H), 6.47 (d, J = 12.1 Hz, 1 H), 6.45 (s, 2 H), 3.83 (s, 3 H), 3.67 (s, 6 H), 2.29 (s, 3 H); CIMS (isobutane) m/e 329 (MH⁺, 100). Anal. (C₁₉H₂₀O₅) C, H.

(E)-1-(4-Acetoxyphenyl)-2-(3,4,5-trimethoxyphenyl)ethene (6p): 114 mg; 34%; oil; ¹H NMR (CDCl₃, 500 MHz) δ 7.51 (d, J = 8.7 Hz, 2 H), 7.09 (d, J = 8.7 Hz, 2 H), 6.99 (s, 2 H), 6.73 (s, 2 H), 3.92 (s, 6 H), 3.87 (s, 3 H), 2.31 (s, 3 H); CIMS (isobutane) m/e 329 (MH⁺, 100). Anal. (C₁₉H₂₀O₅) C, H.

General Procedure for the Preparation of 6q-y. A solution of phosphonate esters 7a-c (12 mmol) in dry DMF (10 mL) was added to a magnetically stirred solution of NaOMe (0.65 g, 12 mmol) in dry DMF (10 mL) at 0 °C, and the solution was stirred for 30 min. A solution of aldehyde 4d/4o-t (10 mmol) in dry DMF (10 mL) was added at 0 °C, and the reaction mixture was allowed to warm to room temperature over a period of 1.5 h. The mixture was heated at 95-100 °C for 1 h and left overnight at room temperature. The mixture was poured slowly onto crushed ice, and the precipitated solid was filtered, washed with water, dried, and crystallized from EtOAc-hexane.

(E)-1-(3,4-Dimethoxyphenyl)-2-phenylethene (6q): 1.64 g; 68%; mp 106-8 °C (lit.⁴¹ mp 111 °C).

(*E*)-1-Phenyl-2-(2,3,4-trimethoxyphenyl)ethene (6r): 2.34 g; 87%; mp 79–82 °C; IR (KBr) 3020, 3000, 2940, 2840, 1600, 1510, 1470, 1420, 1300, 1260, 1230, 1090, 1030, 1000, 980 cm⁻¹; ¹H NMR (CDCl₃, 500 MHz) δ 7.60–7.50 (m, 2 H), 7.40–7.20 (m, 5 H), 6.90

⁽³⁸⁾ Carroll, J. F.; Kulkowit, S.; McKervey, M. A. J. Chem. Soc., Chem. Commun. 1980, 507.

⁽³⁹⁾ Letcher, R. M.; Nhamo, L. R. M.; Gumiro, I. T. J. Chem. Soc., Perkin Trans. 1 1972, 206.

⁽⁴¹⁾ Dey, B. B.; Row, K. K. J. Indian Chem. Soc. 1925, 1, 277.

(d, J = 16.5 Hz, 1 H), 6.70 (d, J = 16.5 Hz, 1 H), 3.91 (s, 3 H), 3.90 (s, 3 H), 3.89 (s, 3 H); CIMS (isobutane) m/e 271 (MH⁺, 100). Anal. (C₁₇H₁₈O₃) C, H.

(E)-1-(4-Aminophenyl)-2-(3,4,5-trimethoxyphenyl)ethene (6z). Lithium aluminum hydride (76 mg, 2 mmol) was added to a solution of nitro stilbene 61 (270 mg, 0.87 mmol) in THF (25 mL), and the mixture was stirred at room temperature for 12 h. Solvent was evaporated at reduced pressure, and the residue was decomposed by careful addition of ice-water (20 mL) containing 2 mL of glacial acetic acid. The red solid formed was filtered and crystallized from CH₂Cl₂-ether to give 6z (200 mg, 82%): mp 251-3 °C; IR (KBr) 3440, 3400, 3000, 2920, 2820, 1600, 1580, 1340, 1240, 1130, 990, 950, 830 cm⁻¹; ¹H NMR (DMSO-d₆, 300 MHz) δ 7.87 (d, J = 8.3 Hz, 2 H), 7.59 (d, J = 8.3 Hz, 2 H), 7.08 (d, J= 16.0 Hz, 1 H), 7.01 (d, J = 16.0 Hz, 1 H), 6.71 (s, 2 H), 3.87 (s, 6 H), 3.82 (s, 3 H); CIMS (isobutane) m/e 286 (MH⁺, 100). Anal. (C₁₇H₁₉NO₃) C, H.

General Procedure for the Preparation of Dihydrostilbenes 8. A solution of stilbene 5 and 6 (1 mmol) in EtOAc (25 mL) was hydrogenated at 40 psi in the presence of 10% Pd-C (30 mg) until the uptake of hydrogen ceased (4 h). The solution was filtered and concentrated to obtain the dihydrostilbenes 8 almost as single components.

1-(4-Methoxyphenyl)-2-(3,4,5-trimethoxyphenyl)ethane (8a): 280 mg; 93%; mp 73–5 °C (lit.³⁸ oil); ¹H NMR (CDCl₃, 500 MHz) δ 7.10 (d, J = 8.5 Hz, 2 H), 6.83 (d, J = 8.5 Hz, 2 H), 6.36 (s, 2 H), 3.83 (s, 3 H), 3.82 (s, 6 H), 3.79 (s, 3 H), 2.80–2.90 (m, 4 H); ¹³C NMR CDCl₃, 50 MHz) δ 158.53, 153.62, 138.16, 136.68, 134.23, 129.96, 114.22, 105.88, 61.16, 56.31, 55.52, 38.85, 37.33; CIMS (isobutane) m/e 303 (MH⁺, 100). Anal. (C₁₈H₂₂O₄) C, H.

1-[4-(Dimethylamino)phenyl]-2-(3,4,5-trimethoxyphenyl)ethane (8n): 265 mg; 84%; oil; ¹H NMR (CDCl₃, 500 MHz) δ 7.08 (d, J = 8.7 Hz, 2 H), 6.72 (d, J = 8.7 Hz, 2 H), 6.38 (s, 2 H), 3.85 (s, 3 H), 3.83 (s, 6 H), 2.92 (s, 4 H), 2.82 (s, 6 H); CIMS (isobutane) m/e 316 (MH⁺, 100%). Anal. (C₁₉H₂₅NO₃) C, H.

1-(4-Aminophenyl)-2-(3,4,5-trimethoxyphenyl)ethane (8z). A solution of nitrostilbene 6l (250 mg, 0.8 mmol) in EtOAc (20 mL) was hydrogenated at 30 psi in the presence of 10% Pd-C (25 mg) at room temperature for 4 h, and the catalyst was filtered off. Evaporation of the solvent and crystallization of the residue from hexanes gave the amine 8z (180 mg, 80%): mp 84-5 °C; IR (KBr) 3450, 3400, 3020, 2920, 2840, 1600, 1580, 1520, 1330, 1240, 1120, 980 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 6.98 (d, J = 8.2 Hz, 2 H), 6.63 (d, J = 8.2 Hz, 2 H), 6.37 (s, 2 H), 3.83 (s, 9 H), 3.75 (bs, 2 H), 2.80 (s, 4 H); CIMS (isobutane) m/e 288 (MH⁺, 100). Anal. (C₁₇H₂₁NO₃) C, H.

1-(4-Acetamidophenyl)-2-(3,4,5-trimethoxyphenyl)ethane (8m). The amine 3z (0.574 g, 2 mmol) was dissolved in dry benzene (10 mL) containing triethylamine (0.5 mL), and the solution was cooled to 0 °C. Acetyl chloride (320 mg, 4 mmol) was added dropwise, and the solution was stirred for 30 min. The contents were poured into ice-cold water, and the mixture was extracted with ether (25 mL). The organic layer was washed with water and 5% sodium bicarbonate solution, and dried (MgSO₄), and the solvent was evaporated. The residue was crystallized from EtOAc-hexane (0.52 g, 79%): mp 112-4 °C; IR (KBr) 3450, 3000, 2930, 2840, 1670, 1600, 1580, 1510, 1340, 1120 cm⁻¹; ¹H NMR (CDCl₃, 200 MHz) δ 7.52 (bs, 1 H), 7.42 (d, J = 8.4 Hz, 2 H), 7.11 (d, J = 8.4 Hz, 2 H), 6.36 (s, 2 H), 3.82, (s, 3 H), 3.81 (s, 6 H), 2.85 (s, 4 H), 2.14 (s, 3 H); CIMS (isobutane) m/e 330 (MH⁺, 100). Anal. (C₁₉H₂₃NO₄) C, H.

General Procedure for Preparation of Benzamides 11a-f. Aroyl chloride 9a-d (20 mmol) was added to a stirred solution of substituted aniline 10a-c (20 mmol) in pyridine (50 mL) at room temperature, and the reaction mixture was stirred for 4 h and poured into a mixture of ice (400 g) and hydrochloric acid (100 mL). The precipitated product was filtered, washed with water, dried and recrystallized from CHCl₃-hexane.

3,4,5-Trimethoxy-N-(4-methoxyphenyl)benzamide (11a): 5.83 g; 92%; mp 160–161 °C (lit.⁴² mp 162 °C); ¹H NMR (CDCl₃, 500 MHz) δ 8.22 (bs, 1 H), 7.50 (d, J = 8.1 Hz, 2 H), 7.03 (s, 2 H), 6.83 (d, J = 8.1 Hz, 2 H), 3.85 (s, 3 H), 3.80 (s, 6 H), 3.77 (s, 3 H); CIMS (isobutane) m/e 318 (MH⁺, 100).

4. Methoxy-N-(3,4,5-trimethoxyphenyl)benzamide (11c): 5.60 g; 88%; mp 159–160 °C; IR (KBr) 3300, 2980, 2940, 1650, 1605, 1515, 1455, 1420, 1340, 1270, 1220, 1020, 1000 cm⁻¹; ¹H NMR (CDCl₃, 500 MHz) δ 8.18 (bs, 1 H), 7.60 (d, J = 8.0 Hz, 2 H), 6.90 (s, 2 H), 6.88 (d, J = 8.0 Hz, 2 H), 3.92 (s, 3 H), 3.80 (s, 6 H), 3.76 (s, 3 H); CIMS (isobutane) m/e 318 (MH⁺, 100). Anal. (C₁₇-H₁₉NO₅) C, H.

General Procedure for Preparation of N-Benzylanilines 12a-f. A solution of benzamide 11a-f (5 mmol) in THF (50 mL) was added to a well-stirred suspension of lithium aluminum hydride (0.285 g, 7.5 mmol) in dry THF (10 mL) at 0 °C under a nitrogen atmosphere, and the reaction mixture was allowed to warm to room temperature. After 4 h, the reaction mixture was poured onto ice (200 g), and the mixture was extracted with ether (3×20 mL). The combined extracts were washed with water and dried (K₂CO₃). Evaporation of ether from the solution afforded amines 12a-f almost as single products. Analytical samples of solid products were prepared by crystallization from ether-hexane, and liquids were purified by preparative thin-layer chromatography using 2% methanol in CHCl₃ as eluent.

3,4,5-Trimethoxy-N-(4-methoxyphenyl)benzylamine (12a): 1.42 g; 94%; mp 73-4 °C; IR (KBr) 3400, 2990, 2920, 2220, 1600, 1510, 1460, 1420, 1330, 1260, 1230, 1120, 1110, 1030, 1000 cm⁻¹; ¹H NMR (CDCl₃, 500 MHz) δ 6.78 (d, J = 8.6 Hz, 2 H), 6.62 (d, J = 8.6 Hz, 2 H), 6.61 (s, 2 H), 4.21 (s, 2 H), 3.86 (bs, 1 H), 3.84 (s, 9 H), 3.74 (s, 3 H); ¹³C NMR (CDCl₃, 50 MHz) δ 153.98, 152.88, 143.03, 137.50, 136.04, 115.37, 114.66, 104.77, 61.18, 56.39, 56.08, 50.00; CIMS (isobutane) m/e 304 (MH⁺, 100). Anal. (C₁₇H₂₁NO₄) C, H.

4-Methoxy-N-(3,4,5-trimethoxyphenyl)benzylamine (12c): 1.42 g; 94%; mp 77-8 °C; IR (KBr) 3380, 2980, 2960, 2940, 2820, 1605, 1580, 1520, 1460, 1440, 1255, 1225, 1130, 1110, 1010, 990 cm⁻¹; ¹H NMR (CDCl₃, 500 MHz) δ 7.29 (d, J = 8.6 Hz, 2 H), 6.88 (d, J = 8.6 Hz, 2 H), 5.87 (s, 2 H), 4.22 (s, 2 H), 3.82 (bs, 1 H), 3.80 (s, 3 H), 3.79 (s, 6 H), 3.76 (s, 3 H); CIMS (isobutane) m/e304 (MH⁺, 100). Anal. (C₁₇H₂₁NO₄) C, H.

Cytotoxicity Assays. An MTT colorimetric assay was employed according to the established procedure.^{43,44} After the addition of the samples to the cell cultures, the cells were incubated for 6 days before the MTT reagent was added. The assays were performed in the Purdue Cell Culture Laboratory. All of the compounds were initially tested once in each of the cell lines listed in Tables I–IV. The active compounds (ED₅₀ < 25 μ M) were tested again, and the values shown for these cytotoxic substances are the averages of two determinations.

Tubulin Polymerization and Colchicine Binding Assays. Electrophoretically homogeneous tubulin was purified from bovine brain as described previously.⁴⁵ Determination of IC₅₀ values for the polymerization of purified tubulin was performed as described in detail elsewhere.³¹ In brief, tubulin was preincubated at 37 °C with varying compound concentrations, reaction mixtures were chilled on ice, GTP (required for the polymerization reaction) was added, and polymerization was followed at 37 °C by turbidimetry at 350 nm in Gilford recording spectrophotometers equipped with electronic temperature controllers. Four instruments were used, and two control reaction mixtures were present in each experiment. The extent of polymerization after a 20-min incubation was determined (the values for the two controls were usually within 5% of each other). IC_{50} values were determined graphically. Active compounds were examined in at least three independent assays, while inactive compounds (defined as IC_{M}) value > 50 μ M) were examined in at least two independent experiments. The effect of agents on the binding of [³H]colchicine (obtained from Amersham) to tubulin was measured by the DEAE-filter technique as described previously.^{32,46}

- (43) Alley, M. C.; Scudiero, D. A.; Monks, A.; Hursey, M. L.; Czerwinski, M. J.; Fine, D. L.; Abbott, B. J.; Mayo, J. G.; Shoemaker, R. H.; Boyd, M. R. Cancer Res. 1988, 48, 589.
 (44) Margine T. J. Langued Meth. 1986, 65–55.
- 44) Mossman, T. J. Immunol. Meth. 1983, 65, 55.
- (45) Hamel, E.; Lin, C. M. Biochemistry 1984, 23, 4173.
- (46) Borisy, G. G. Anal. Biochem. 1972, 13, 373.
- (47) Baker, B. R.; Gibson, R. E. J. Med. Chem. 1971, 14, 1057.
- (48) Erdtman, H.; Rosengren, A. Acta. Chem. Scand. 1968, 22, 1475.

⁽⁴²⁾ Grammaticakis, P. Compt. Rend., Ser. C 1966, 262, 584.

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Registry No. 3a, 61240-20-8; **3b**, 1530-38-7; **4a**, 123-11-5; **4b**, 591-31-1; **4c**, 135-02-4; **4d**, 2103-57-3; **4e**, 54439-75-7; **4f**, 100-52-7; **4g**, 104-88-1; **4h**, 1122-91-4; **4i**, 872-85-5; **4j**, 500-22-1; **4k**, 1121-60-4; **4l**, 555-16-8; **4m**, 120743-99-9; **4n**, 120-21-8; **4o**, 120-14-9; **4p**, 7311-34-4; **4q**, 86-81-7; **4r**, 4460-86-0; **4s**, 830-79-5; **5a**, 134029-49-5;

- (49) Drews, S. E.; Fletcher, I. P. J. Chem. Soc., Perkin Trans. 1 1974, 961.
- (50) Sieber, R. H. Justus Liebigs Ann. Chem. 1969, 730, 31.
- (51) Loudon, G. M.; Noyce, D. S. J. Am. Chem. Soc. 1969, 91, 1433.
- (52) Molho, D.; Coillard, J. Bull. Soc. Chim. Fr. 1968, 78.
- (53) Iyoda, M.; Sakaitani, M.; Otsuka, H.; Oda, M. Chem. Lett. 1985, 127.

5b, 97399-87-6; 5c, 134029-50-8; 5d, 134029-51-9; 5e, 134029-52-0; 5f, 74809-44-2; 5g, 134029-53-1; 5h, 74809-51-1; 5i, 134029-54-2; 5j, 134029-55-3; 5k, 134029-56-4; 5l, 134029-57-5; 5m, 134029-58-6; 5n, 134029-59-7; 5o, 134029-60-0; 5p, 134029-61-1; 6a, 134029-62-2; 6b, 97399-88-7; 6c, 134029-63-3; 6e, 134054-80-1; 6f, 74809-43-1; 6g, 134029-64-4; 6h, 74809-50-0; 6i, 118160-75-1; 6j, 134029-65-5; 6l, 134029-66-6; 6m, 134029-67-7; 6n, 134029-68-8; 6o, 116519-00-7; 6p, 134029-69-9; 6q, 3892-92-0; 6r, 134029-70-2; 6s, 89946-15-6; 6t, 134029-71-3; 6u, 61240-22-0; 6v, 134029-72-4; 6w, 129047-45-6; 6x, 22887-34-9; 6y, 83088-26-0; 6z, 134029-73-5; 7a, 1080-32-6; 7b, 78055-65-9; 7c, 134029-74-6; 8a, 35450-81-8; 8b, 108853-15-2; 8c, 134029-75-7; 8f, 133462-62-1; 8i, 134029-76-8; 8j, 20934-93-4; 8m, 134029-77-9; 8n, 134029-78-0; 8o, 39499-95-1; 8p, 39500-11-3; 8g, 3892-93-1; 8r, 134029-79-1; 8s, 52961-91-8; 8t, 117048-69-8; 8u, 33284-74-1; 8v, 134029-80-4; 8z, 110997-90-5; 9a, 100-07-2; 9b, 3535-37-3; 9c, 17213-57-9; 9d, 4521-61-3; 10a, 104-94-9; 10b, 10272-07-8; 10c, 24313-88-0; 11a, 742-15-4; 11b, 134029-81-5; 11c, 134029-82-6; 11d, 134029-83-7; 11e, 134029-84-8; 11f, 748-31-2; 12a, 134029-85-9; 12b, 134029-86-0; 12c, 134029-87-1; 12d, 134029-88-2; 12e, 134054-81-2; 12f, 134029-89-3.

Supplementary Material Available: Physical and spectral data of compounds 6b,c,e,g-j,s-y, 8b,c,f,i,j,o-v, 11b,d-f and 12b,d-f (6 pages). Ordering information is given on any current masthead page.

Relaxant Activity of 6-Cyano-2,2-dimethyl-2*H*-1-benzopyran-4-carboxamides and -thiocarboxamides and Their Analogues in Guinea Pig Trachealis

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Structural modifications of the potassium channel activator cromakalim (1) are described in which the amide moiety at C-4 has been replaced by carboxamide and thiocarboxamide functions. Analogues in which the hydroxyl group at C-3 has been oxidized or removed are also disclosed. Such analogues display an interesting profile of smooth muscle relaxant activity in the guinea pig isolated trachea, not all of which appears to result from the opening of potassium channels, but few compounds retain useful in vivo activity. However, one compound in particular, 6-cyano-2,2-dimethyl-N-methyl-2H-1-benzopyran-4-thiocarboxamide (13) was shown to be a potent potassium channel activator in vitro and to provide prolonged protection to guinea pigs from the respiratory effects of inhaled histamine.

Introduction

Interest in the potassium channel activators has increased markedly over the past few years, having gained particular momentum following the classification of additional potassium channel subtypes and the identification of a number of blockers with subtype specificity.¹ Another major contributor to this growth was the discovery of cromakalim (1), a specific potassium channel activator,²



which is thought to exert its smooth muscle relaxant ac-

tivity through the opening of ATP-sensitive potassium channels and consequent hyperpolarization.^{3,4} While such activity was initially found in the smooth muscle of the vasculature, and thus demonstrated in models of hypertension, it is now recognized that potassium channel activators also have potential application for the treatment of other diseases involving smooth muscle, such as asthma and urinary incontinence.¹

Particular support for the utility of potassium channel activators in respiratory disorders results from the detailed study of cromakalim in animal models⁵ and preliminary

(5) Arch, J. R. S.; Buckle, D. R.; Bumstead, J.; Clarke, G. D.; Taylor, J. F.; Taylor, S. G. Evaluation of the Potassium Channel Activator Cromakalim (BRL 34915) as a Bronchodilator in the Guinea pig: Comparison with Nifedipine. Br. J. Pharmacol. 1988, 95, 763-770.

Robertson, D. W.; Steinberg, M. I. Potassium Channel Modulators: Scientific Applications and Therapeutic Promise. J. Med. Chem. 1990, 33, 1529-1541.

⁽²⁾ Hamilton, T. C.; Weir, S. W.; Weston, A. H. Comparison of the Effects of BRL 34915 and Verapamil on Electrical and Mechanical Activity in Rat Portal Vein. Br. J. Pharmacol. 1986, 88, 103-111.

⁽³⁾ Ashwood, V. A.; Buckingham, R. E.; Cassidy, F.; Evans, J. M.; Faruk, E. A.; Hamilton, T. C.; Nash, D. J.; Stemp, G.; Willcocks, K. Synthesis and Antihypertensive Activity of 4-(Cyclic amido)-2H-1-benzopyrans. J. Med. Chem. 1986, 29, 2194-2201.

⁽⁴⁾ Standen, N. B.; Quayle, J. M.; Davies, N. W.; Brayden, J. E.; Huang, Y.; Nelson, M. T. Hyperpolarising Vasodilators Activate ATP-sensitive K⁺ Channels in Arterial Smooth Muscle. *Science* 1989, 245, 177-180.