

Research Article Synthesis of Novel 1,2,4-Triazolyl Coumarin Derivatives as Potential Anticancer Agents

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Received 22 June 2018; Accepted 29 August 2018; Published 14 October 2018

Academic Editor: Gabriel Navarrete-Vazquez

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A series of novel coumarin derivatives carrying 1,2,4-triazole or 1,2,4-triazolo[3,4-*b*][1,3,4]thiadiazole moieties were prepared and evaluated *in vitro* as anticancer in the human colon cancer (HCT116) cell line. The derivatives **4c** and **8c** exhibited marked anticancer activity with IC₅₀ values 4.363 and 2.656 μ M, respectively. The molecular docking studies suggested possible interaction with tyrosine kinases (CDK2).

1. Introduction

Coumarin (2H-1-benzopyran-2-one; 2H-chromen-2-one) derivatives are a large class of important naturally occurring and synthetic oxygen-containing heterocycles. This type of benzopyrone structure enables such derivatives to interact with diversity of enzymes and receptors in organisms through weak bond interactions, thereby exhibiting wide potentiality as bioactive drugs [1]. More than 1300 coumarin derivatives were identified as secondary metabolites from plants, bacteria, and fungi [2]. Coumarin derivatives were early recognized as important agents for the prevention and treatment of diseases [3, 4]. Several coumarin-based derivatives are currently used as anticoagulant drugs and as rodenticides [5-7]. The naturally occurring coumarins such as novobiocin, chlorobiocin, and coumermycin have been found to be an unprecedented class of antibiotics, specifically against Gram-positive bacteria [8]. The natural coumarin derivatives possessing long-chain hydrocarbon substitutions such as ammoresinol and ostruthin [9], grandivittin, agasyllin, and osthole displayed significant antibacterial activity against clinically isolated Gram-positive and Gram-negative bacterial strains [10, 11]. In addition, the coumarin glycoside fraxin displayed free-radical scavenging effect and cell protective effect against hydrogen peroxide-mediated oxidative

stress [12, 13]. Meanwhile, the coumarin derivatives ensaculin [14] and AP2243 [15] are known acetylcholine esterase inhibitors which have been used clinically in treating Alzheimer's disease for a long time.

Coumarin compounds as potential anticancer agents have become a rapidly developing, extremely active, and attractive topic. The coumarin derivative STX 64, a potent estrogen antagonist, is currently under clinical trials as anticancer agents against breast carcinoma [16]. Moreover, auraptene has been reported as anticancer agent against the liver, skin, tongue, esophagus, and colon cancers [17] (Figure 1).

On the contrary, 1,2,4-triazole nucleus has been reported to constitute the essential pharmacophore of various therapeutically active agents possessing marked antiinflammatory [18, 19], antifungal [20, 21], and anticancer [22, 23] activities. Moreover, the condensed triazolo heterocycles such as 1,2,4-triazolo[3,4-*b*] [1,3,4]thiadiazoles were reported to exhibit significant antibacterial [24, 25] and anticancer [26] activities.

Based on the abovementioned biological activities of coumarin, 1,2,4-triazole, and 1,2,4-triazolo[3,4-*b*] [1,3,4] thiadiazole derivatives, we report herein the synthesis and evaluation of the anticancer activity of new series of coumarin derivatives carrying a 1,2,4-triazole or 1,2,4-triazolo [3,4-*b*] [1,3,4]thiadiazole moieties.



FIGURE 1: The structures of coumarin-based anticancer drugs.

2. Results and Discussion

2.1. Chemical Synthesis. The synthesis of the target triazolyl coumarin derivatives is outlined in Scheme 1. The starting material 2-(coumarin-4-yl) acetic acid 1 (2-(2-oxo-2Hchromen-4-yl) acetic acid was prepared via reaction of citric acid with phenol in sulfuric acid following the previously reported procedures [27-29]. 2-(Coumarin-4-yl)acetic acid 1 was reacted with thiocarbohydrazide 2 in refluxing phosphoryl chloride to yield the target 3-[(coumarin-4-yl) methyl]-4-amino-1H-1,2,4-triazole-5(4H)-thione 3 in 80% yield. The reaction of the aminotriazole derivative 3 with different aromatic aldehydes in propanol containing catalytic amount of acetic acid yielded the corresponding arylideneamino derivatives 4a-c. 4-Arylideneamino-1H-1,2,4triazole-5(4H)-thiones were reported to undergo oxidative cyclization to their 1,2,4-triazolo[3,4-b][1,3,4]thiadiazole analogues using different oxidizing agents such as iodine, nitrobenzene, and ammonium cerium(IV) nitrate [30]. Accordingly, the arylideneamino derivatives 4a-c were successfully cyclized to their 1,2,4-triazolo[3,4-b][1,3,4] thiadiazole analogues 5a-c by the action of iodine in acetonitrile. The 1,2,4-triazolo[3,4-b][1,3,4]thiadiazole analogues 5a-c were also independently synthesized in slightly lower yields by the reaction of the aminotriazole derivative 3 with the corresponding aromatic carboxylic acid via heating in phosphoryl chloride. The reaction of compound 3 with carbon disulfide in pyridine under reflux furnished 6mercapto-1,2,4-triazolo[3,4-*b*][1,3,4]thiadiazole analogue 6 in 78% yield. The mercapto derivative 6 was further reacted with iodomethane in dry N,N-dimethylformamide (DMF), in the presence of anhydrous potassium carbonate to yield the methylthio analogue 7. The reaction of the methylthio analogue 7 with different primary aromatic amines via prolonged heating in DMF resulted in replacement of the methylthio group with an arylamino substituent leading to the 6-arylamino derivatives 8a-c in reasonable yields (Scheme 1; Table 1). The structures of the newly synthesized compounds were established based on IR, ¹H NMR, ¹³C NMR, and mass spectral data.

2.2. In Vitro Anticancer Activity. The antitumor activity of compounds **4a-c**, **5a-c**, and **8a-c** was investigated against the human colorectal cancer cell line (HCT116) following the previously reported colorimetric cytotoxicity assay of Skehan et al. [31]. The results of the preliminary cytotoxic activity of compounds **4a-c**, **5a-c**, and **8a-c** and the anticancer drug doxorubicin [32] are shown in Table 2.

The analysis of the IC_{50} values on the HCT116 cell line (Table 2) revealed that the compounds **4c** and **8c** exhibited

a high anticancer activity (relative potency >50%) with IC₅₀ values 4.363 and 2.656 μ M, respectively. Meanwhile, compounds **4a**, **8a**, and **8b** displayed moderate anticancer activity (relative potency 10–20%) with IC₅₀ values 18.76, 25.630, and 15.296 μ M. Compounds **4b**, **5a**, **5b**, and **5c** were practically inactive (relative potency <10%).

2.3. Molecular Docking Studies. To understand the mechanism of action of the anticancer activities of the newly synthesized compounds, we carried out molecular docking, which is used to predict the binding mode of ligands within the binding site of target proteins [33]. Taking into consideration the previously reported tyrosine kinases (CDK2) inhibitory activity of the structurally related chromene anticancer agents genistein [34] and quercetin [35], we docked the synthesized compounds in CDK2 active site. To validate and specify the target protein for the antitumor activity of newly synthesized triazolyl coumarin derivatives, CDK2 protein was selected and downloaded from the protein data bank (PDB ID: 1KE8) [36]. Docking studies of compound 4c into the active site of CDK2 enzyme showed two hydrogen bonds with THR160 and GLU81 and a good electrostatic interaction with the active site (Figure 2).

Similarly, docking conformation of the active compound **8c** showed good interactions with the active site residues of this protein. Compound **8c** formed two hydrogen bond interactions between amino groups moiety, as it acts as a hydrogen bond donor with the side chain of HIE84 and ASP86 residues with strength of 45%. Furthermore, it showed Van der Waals interaction with PHE80 and PHE82 (Figure 3). Finally, there was a good correlation between the docking studies and the biological profiles.

3. Materials and Methods

Melting points (°C) were measured in open-glass capillaries using a Gallenkamp melting point apparatus and are uncorrected. IR spectra were recorded in potassium bromide discs on a Shimadzu FTIR 8101 PC infrared spectrophotometer. The NMR spectra were recorded on a BRUKER VX-500 NMR spectrometer. ¹H spectra were run at 500 MHz, and $^{13} {\rm \hat{C}}$ spectra were run at 125 MHz in deuterated dimethylsulphoxide (DMSO-d₆) using TMS as an internal standard. Electron impact mass spectra were recorded on a Shimadzu GCMS-QP 1000 EX mass spectrometer at 70 eV. Elemental analyses were performed using a Vario LIII CHNS analyzer, and the analytical data (C, H & N) were in agreement with the proposed structures within $\pm 0.4\%$ of the theoretical values. The biological evaluation of the products was carried out in the Medical Mycology Laboratory of the Regional Center for Mycology and Biotechnology of Al-Azhar University, Cairo, Egypt.

3.1. 3-[(Coumarin-4-yl)methyl]-4-amino-1H-1,2,4-triazole-5 (4H)-thione 3. Phosphoryl chloride (5 g) was added dropwise to an ice-cold mixture of 2-(coumarin-4-yl) acetic acid 1 (1.06 g, 0.01 mole) and thiocarbohydrazide 2 (2.04 g, 0.01 mole) with stirring, and the mixture was then heated under



SCHEME 1: Synthesis of the target triazolyl coumarin derivatives.

TABLE 1: Crystallization solvents, melting points, yield percentages, molecular formulae, and molecular weights of compounds **3**, **4a–c**, **5a–c**, **6**, and **7**.

3 — EtOH 226–229 80 C	$_{12}H_{10}N_4O_2S$ (274.30)
4a 4-OHC ₆ H ₄ DMF/EtOH 248-250 60 C	$_{19}H_{14}N_4O_3S$ (378.40)
4b 4-MeOC ₆ H ₄ DMF/EtOH 255–257 66 C	$_{20}H_{16}N_4O_3S$ (392.43)
4c 4-OH,3-MeOC ₆ H ₃ DMF/EtOH 240–243 54 C	$_{20}H_{16}N_4O_4S$ (408.43)
5a $4-OHC_6H_4$ DMF 271-273 74 C	$_{19}H_{12}N_4O_3S$ (376.39)
5b $4-\text{MeOC}_6\text{H}_4$ DMF 282–284 62 C	$_{20}H_{14}N_4O_3S$ (390.42)
5c 4-OH,3-MeOC ₆ H ₃ DMF 266–268 48 C	$_{20}H_{14}N_4O_4S$ (406.41)
6 – Dioxan 236–239 78 C	$H_{13}H_8N_4O_2S_2$ (316.36)
7 — EtOH 244–246 85 C ₁	$_{14}H_{10}N_4O_2S_2$ (330.38)
8a 4-OHC ₆ H ₄ DMF/EtOH 273–275 64 C	₁₉ H ₁₃ N ₅ O ₃ S (391.40)
8b 4-ClC ₆ H ₄ DMF/EtOH 264–267 75 C ₁₉	₉ H ₁₂ ClN ₅ O ₂ S (409.85)
8c $4-H_2NSO_2C_6H_4$ DMF/EtOH 283–285 55 C_1	$_{19}H_{14}N_6O_4S_2$ (454.48)

reflux for one hour. The excess phosphoryl chloride was distilled off under reduced pressure. The resulted residue was triturated with dry pyridine (2 mL), and ice-cold water (250 mL) was added and the mixture was kept for 10 minutes. The precipitated crude product was filtered, washed with water, and crystallized from ethanol to yield 2.2 g (80%) of **3** as light brown powder. IR (cm⁻¹): 3280 (NH), 3169, 3155 (NH₂), 2895 (Aliphatic CH), 1715 (C=O). ¹H NMR: δ 13.81 (s, 1H, NH), 7.64 (t, *J*=7.5 Hz, 1H, Coumarin-H), 7.63 (d, *J*=8.4 Hz, 1H, Coumarin-H), 7.47 (d, *J*=8.6 Hz, 1H), 7.34 (t, *J*=7.5 Hz, 1H, Coumarin-H), 5.20 (s, 2H, NH₂), 3.86 (s, 2H, CH₂). ¹³C NMR: δ 172.90

(C=S), 161.30 (C=O), 154.14, 150.14, 146.94, 132.55, 126.54, 124.62, 118.12, 117.10, 114.20 (Coumarin-C & Triazole C3), 29.67 (CH₂). EI-MS (m/z) = 274. Anal. calcd. for C₁₂H₁₀N₄O₂S, %: C, 52.54; H, 3.67; N, 20.43. Found, %: C, 52.36; H, 3.43; N, 20.28.

3.2. 3-[(Coumarin-4-yl)methyl]-4-arylideneamino-1H-1,2,4triazole-5(4H)-thione **4a-c**. A mixture of 3-[(coumarin-4-yl) methyl]-4-amino-1H-1,2,4-triazole-5(4H)-thione **3** (274 mg, 1 mmol) and the appropriate aromatic aldehyde (1 mmol), in isopropyl alcohol (25 mL) containing three drops of acetic

Relative potency* Comp. no. IC_{50} (μM) 4a 13.865 18.76% **4b** 44.704 5.82% 4c 4.363 59.62% 129.259 5a 2.01% 266.089 5b 0.98% 5c 39.133 6.65% **8**a 25.630 10.15% 8b 15.296 17.00% 97.93% 8c 2.656 Doxorubicin 2.601 100%

TABLE 2: Cytotoxic activity of compounds **4a–c**, **5a–c**, and **8a–c** against the human colorectal cancer cell line (HCT116).

*Compared to doxorubicin.

acid, was heated under reflux for 7 hours. On cooling, the precipitated crude products were filtered, dried, and recrystallized from DMF-EtOH to afford compounds **4a-c** as yellowish brown powders.

4a: IR (cm⁻¹): 3361 (OH), 3231 (NH), 3055 (Aromatic CH), 2860 (Aliphatic CH), 1720 (C=O), 1625 (N=CH). ¹H NMR: *δ* 13.68 (s, 1H, NH), 9.75 (s, 1H, OH), 8.31 (s, 1H, N=CH), 7.70 (t, J = 7.6 Hz, 1H, Coumarin-H), 7.44–7.50 (m, 4H, Coumarin-H & Ar–H), 7.23 (t, J = 7.6 Hz, 1H, Coumarin-H), 6.88 (d, J = 7.5 Hz, 2H, 2H, Ar–H), 6.24 (s, 1H, CH, Coumarin-H), 3.92 (s, 2H, CH₂). ¹³C NMR: *δ* 172.35 (C=S), 161.20 (C=O), 157.24, 154.27, 153.24, 146.07, 145.56, 132.42, 132.10, 127.09, 126.54, 124.21, 118.07, 117.08, 115.15, 114.44 (Coumarin-C, N=CH, Ar–C & Triazole C3), 30.15 (CH₂). EI-MS (m/z) = 378. Anal. calcd. for C₁₉H₁₄N₄O₃S, %: C, 60.31; H, 3.73; N, 14.81. Found, %: C, 60.26; H, 3.83; N, 14.38.

4b: IR (cm⁻¹): 3274 (NH), 3060 (Aromatic CH), 2880 (Aliphatic CH), 1710 (C=O), 1650 (C=N). ¹H NMR: δ 13.66 (s, 1H, NH), 8.32 (s, 1H, N=CH), 7.63 (t, J= 8.1 Hz, 1H, Coumarin-H), 7.54–7.45 (m, 4H, Coumarin-H), 7.07 (d, J= 8.4 Hz, 2H, Ar–H), 3.92 (s, 2H, CH₂), 3.82 (s, 3H, OCH₃). ¹³C NMR: δ 172.36 (C=S), 161.19 (C=O), 158.91, 154.06, 153.06, 146.08, 145.56, 132.40, 129.54, 128.11, 126.54, 124.26, 118.03, 117.08, 114.44, 113.69 (Coumarin-C, N=CH, Ar–C & Triazole C3), 55.33 (OCH₃), 30.17 (CH₂). EI-MS (m/z) = 392. Anal. calcd. for C₂₀H₁₆N₄O₃S, %: C, 61.21; H, 4.11; N, 14.28. Found, %: C, 60.86; H, 3.93; N, 14.16.

4c: IR (cm⁻¹): 3341 (OH), 3214 (NH), 3065 (Aromatic CH), 2886 (Aliphatic CH), 1715 (C=O), 1650 (C=N). ¹H NMR: δ 13.29 (s, 1H, NH), 8.96 (s, 1H, OH), 8.37 (s, 1H, N=CH), 7.63–7.46 (m, 4H, Coumarin-H & Ar–H), 7.33 (t, J=7.9 Hz, 1H, Coumarin-H), 7.13 (s, 1H, Coumarin-H), 7.10–6.87 (m, 1H, Ar–H), 6.32 (s, 1H, Coumarin-H), 3.93 (s, 2H, CH₂), 3.86 (s, 3H, OCH₃). ¹³C NMR: δ 172.33 (C=S), 161.20 (C=O), 154.06, 151.98, 147.71, 147.41, 146.08, 145.56, 132.29, 127.20,

126.57, 124.88, 124.44, 118.03, 117.09, 115.01, 114.43, 109.04 (Coumarin-C, N=CH, Ar–C & Triazole C3), 56.14 (OCH₃), 30.15 (CH₂). EI-MS (m/z) = 408. Anal. calcd. for C₂₀H₁₆N₄O₄S, %: C, 58.81; H, 3.95; N, 13.72. Found, %: C, 58.66; H, 3.73; N, 13.46.

3.3. 6-Aryl-3-[(coumarin-4-yl)methyl]-[1,2,4]triazolo[3,4-b] [1,3,4]thiadiazole **5a-c**

3.3.1. Method A. To a suspension of the arylideneamino analogues 4a-c (50 mmol), in dry acetonitrile (70 mL), iodine (1.27 g, 50 mmol) was added, the reaction mixture was heated under reflux, and the reaction progression was monitored by TLC using ethyl acetate:hexane (9:1) as mobile phase. Upon completion of the reaction, the reaction mixture was poured onto ice-cold water (200 mL), and a solution of sodium thiosulfate was added to destroy excess iodine. The greenish precipitate thus formed was filtered, washed with ethyl acetate, and recrystallized from DMF-EtOH.

3.3.2. Method B. A mixture of the appropriate carboxylic acid (50 mmol), 3-[(coumarin-4-yl)methyl]-4-amino-1*H*-1,2,4-triazole-5(4*H*)-thione **3** (1.37 g, 50 mmol), and phosphoryl chloride (5 mL) was heated under reflux for 1 hour. On cooling, crushed ice (100 g) was cautiously added, and the mixture was stirred for 30 minutes. The separated crude product was filtered, washed with water then with saturated sodium hydrogen carbonate solution and finally with water, dried, and crystallized from DMF-EtOH.

5a: IR (cm⁻¹): 3361 (OH), 3110 (Aromatic CH), 2885 (Aliphatic CH), 1720 (C=O), 1650 (C=N). ¹H NMR: δ 9.25 (s, 1H, OH), 7.77 (d, *J* = 8.4 Hz, 2H, Coumarin-H), 7.65–7.33 (m, 4H, Coumarin-H & Ar–H), 6.91 (d, *J* = 7.4 Hz, 2H, Ar–H), 6.29 (s, 1H, Coumarin-H), 4.14 (s, 2H, CH₂). ¹³C NMR: δ 165.54, 161.26, 159.55, 157.24, 154.41, 145.76, 142.95, 132.38, 132.18, 126.81, 124.21, 123.63, 117.90, 117.30, 117.04, 112.59 (Coumarin-C, Ar–C & Triazolo[3,4-*b*][1,3,4]thiadiazole-C), 30.02 (CH₂). EI-MS (*m*/*z*) = 376. Anal. calcd. for C₁₉H₁₂N₄O₃S, %: C, 60.63; H, 3.21; N, 14.89. Found, %: C, 60.36; H, 3.53; N, 15.16.

5b: IR (cm⁻¹): 3090 (Aromatic CH), 2980 (Aliphatic CH), 1710 (C=O), 1655 (C=N). ¹H NMR: *δ* 7.89 (d, J = 8.4 Hz, 2H, Ar–H), 7.65–7.64 (m, 2H, Coumarin-H), 7.49 (d, J = 8.3 Hz, 1H, Coumarin-H), 7.32 (t, J = 7.8 Hz, 1H, Coumarin-H), 7.03 (d, J = 8.4 Hz, 2H, Ar–H), 6.28 (s, 1H, Coumarin-H), 4.14 (s, 2H, CH₂), 3.88 (s, 3H, OCH₃). ¹³C NMR: *δ* 165.51, 161.33, 159.53, 157.24, 154.25, 145.79, 142.95, 132.37, 131.94, 126.84, 125.56, 124.02, 117.83, 117.02, 114.86, 112.62 (Coumarin-C, Ar–C & Triazolo[3,4-*b*][1,3,4]thiadiazole-C), 55.32 (OCH₃), 30.02 (CH₂). EI-MS (*m*/*z*) = 390. Anal. calcd. for C₂₀H₁₄N₄O₃S, %: C, 61.53; H, 3.61; N, 14.35. Found, %: C, 61.26; H, 3.33; N, 14.09.



FIGURE 2: (a) 2D and (b) 3D interactions of compound **4c** with ATP active site of CDK2, which showed hydrogen bond and electrostatic interactions.



FIGURE 3: (a) 2D and (b) 3D interactions of compound **8c** with ATP active site of CDK2, which showed hydrogen bond and electrostatic interactions.

5c: IR (cm⁻¹): 3380 (OH), 3098 (Aromatic CH), 2975 (Aliphatic CH), 1720 (C=O), 1655 (C=N). ¹H NMR: *δ* 8.81 (s, 1H, OH), 7.75 (d, *J* = 7.8 Hz, 1H, Coumarin-H), 7.62–7.38 (m, 3H, Coumarin-H), 7.33–6.97 (m, 2H, Ar–H), 6.31 (s, 1H, Coumarin-H), 4.13 (s, 2H, CH₂), 3.86 (s, 3H, OCH₃). ¹³C NMR: *δ* 166.96, 161.27, 157.28, 154.25, 149.20, 147.64, 145.80, 143.00, 132.30, 126.82, 125.94, 124.44, 124.17, 117.83, 117.02, 116.40, 115.14, 112.62 (Coumarin-C, Ar–C & Triazolo[3,4-*b*][1,3,4]thiadiazole-C), 55.96 (OCH₃), 30.04 (CH₂). EI-MS (*m*/*z*) = 406. Anal. calcd. for C₂₀H₁₄N₄O₄S, %: C, 59.11; H, 3.47; N, 13.79. Found, %: C, 59.26; H, 3.31; N, 13.49.

3.4. 3-[(Coumarin-4-yl)methyl]-6-mercapto[1,2,4]triazolo [3,4-b][1,3,4]thiadiazole 6. Carbon disulfide (2 mL) and few drops of triethylamine were added to a solution of 3-[(coumarin-4-yl)methyl]-4-amino-1H-1,2,4-triazole-5(4H)-thione 3 (2.74 g, 0.01 mol) in pyridine (30 mL), and the mixture was heated under reflux at 100°C with stirring for 6 hours. On cooling, the reaction mixture was poured onto

ice-cold water (100 mL), and the mixture was slightly acidified with hydrochloric acid. The precipitated crude product was filtered, washed with water, dried, and crystallized from dioxin to yield 2.47 g (78%) of **6** as orange crystals. IR (cm⁻¹): 2985 (Aliphatic CH), 1715 (C=O), 1660 (C=N). ¹H NMR: δ 7.69 (d, J = 7.9 Hz, 1H, Coumarin-H), 7.64 (t, J = 8.1 Hz, 1H, Coumarin-H), 7.50 (d, J = 8.1 Hz, 1H, Coumarin-H), 7.50 (d, J = 8.1 Hz, 1H, Coumarin-H), 7.50 (d, J = 8.1 Hz, 1H, Coumarin-H), 6.21 (s, 1H, Coumarin-H), 5.65 (s, 1H, SH), 4.12 (s, 2H, CH₂). ¹³C NMR: δ 161.74, 160.93, 156.21, 154.22, 145.79, 144.86, 132.18, 126.65, 124.07, 117.88, 117.01, 112.52 (Coumarin-C & Triazolo[3,4-*b*][1,3,4]thiadiazole-C), 29.71 (CH₂). EI-MS (*m*/*z*) = 316. Anal. calcd. for C₁₃H₈N₄O₂S₂, %: C, 49.36; H, 2.55; N, 17.71. Found, %: C, 49.26; H, 2.38; N, 17.49.

3.5. 3-[(Coumarin-4-yl)methyl]-6-methylthio[1,2,4]triazolo [3,4-b][1,3,4]thiadiazole 7. To a solution of 3-[(coumarin-4-yl)methyl]-6-mercapto[1,2,4]triazolo[3,4-b][1,3,4]thiadiazole **6** (1.58 g, 50 mmol) in DMF (10 mL), iodomethane (1.14 gm, 80 mmol) and anhydrous potassium carbonate

(0.69 gm, 50 mol) were added, and the mixture was stirred at room temperature for 4 hours. Water (15 mL) was added, and the mixture was stirred for further 30 minutes. The precipitated crude product was filtered, washed with water, dried, and crystallized from ethanol to yield 1.4 gm (85%) of 7 as yellow crystalline powder. IR (cm⁻¹): 2979 (Aliphatic CH), 1710 (C=O), 1640 (C=N). ¹H NMR: δ 7.70 (d, J = 7.8 Hz, 1H, Coumarin-H), 7.64 (t, J = 8.2 Hz, 1H, Coumarin-H), 7.50 (d, J = 8.2 Hz, 1H, Coumarin-H), 7.34 (t, J=7.9 Hz, 1H, Coumarin-H), 6.27 (s, 1H, Coumarin-H), 4.11 (s, 2H, CH₂), 2.67 (s, 3H, CH₃). ¹³C NMR: δ 166.94, 160.93, 154.24, 153.69, 145.79, 141.83, 131.92, 126.66, 124.39, 117.88, 117.01, 112.50 (Coumarin-C & Triazolo[3,4-b][1,3,4] thiadiazole-C), 30.30 (CH₂), 15.48 (CH₃). EI-MS (m/z) =330. Anal. calcd. for C₁₄H₁₀N₄O₂S₂, %: C, 50.90; H, 3.05; N, 16.96. Found, %: C, 50.66; H, 2.88; N, 17.19.

3.6. 3-[(Coumarin-4-yl)methyl]-6-arylamino[1,2,4]triazolo [3,4-b][1,3,4]thiadiazole **8a**-c. 4-Aminophenol, 4-chloroaniline, or 4-aminobenzenesulfonamide (50 mmol) was added to a solution of compound 7 (1.56 g, 50 mmol), in DMF (10 mL) containing few drops of 37% hydrochloric acid, and the mixture was heated under reflux for 10 hours. The excess DMF was then distilled off *in vacuo*, and the solid residue was triturated with water (20 mL). The separated solid precipitate was filtered, washed with water, and crystallized from DMF-Ethanol to yield compounds **8a**-c as reddish brown crystals.

8a: IR (cm⁻¹): 3351 (OH), 3244 (NH), 3065 (Aromatic CH), 2880 (Aliphatic CH), 1714 (C=O), 1650 (C=N). ¹H NMR: δ 9.95 (s, 1H, NH), 9.24 (s, 1H, OH), 7.70–7.40 (m, 3H, Coumarin-H), 7.33 (t, *J* = 8.1 Hz, 1H, Coumarin-H), 7.01 (d, *J* = 8.4 Hz, 2H, Ar–H), 6.81 (d, *J* = 8.4 Hz, 2H, Ar–H), 6.20 (s, 1H, Coumarin-H), 4.12 (s, 2H, CH₂). ¹³C NMR: δ 161.25, 159.00, 156.71, 154.25, 153.86, 145.79, 142.43, 133.77, 132.37, 126.84, 124.89, 123.98, 117.83, 117.02, 115.27, 112.62 (Coumarin-C, Ar–C & Triazolo[3,4-*b*][1,3,4]thiadiazole-C), 29.84 (CH₂). EI-MS (*m*/*z*) = 391. Anal. calcd. for C₁₉H₁₃N₅O₃S, %: C, 58.30; H, 3.35; N, 17.89. Found, %: C, 58.61; H, 3.18; N, 17.59.

8b: IR (cm⁻¹): 3242 (NH), 3066 (Aromatic CH), 2890 (Aliphatic CH), 1718 (C=O), 1655 (C=N) ¹H NMR: δ 9.94 (s, 1H, NH), 7.70–7.44 (m, 5H, Coumarin-H & Ar–H), 7.33 (t, *J* = 7.9 Hz, 1H, Coumarin-H), 6.82 (d, *J* = 8.4 Hz, 2H, Ar–H), 6.20 (s, 1H, Coumarin-H), 4.13 (s, 2H, CH₂). ¹³C NMR: δ 161.26, 159.00, 156.71, 154.25, 145.80, 142.45, 138.97, 132.25, 128.94, 127.98, 126.85, 124.44, 123.01, 117.82, 117.04, 112.62 (Coumarin-C, Ar–C & Triazolo[3,4-*b*][1,3,4]thiadiazole-C), 29.84 (CH₂). EI-MS (*m*/*z*) = 411 (39%), 409 (100%). Anal. calcd. for C₁₉H₁₂ClN₅O₂S, %: C, 55.68; H, 2.95; N, 17.09. Found, %: C, 55.41; H, 3.15; N, 17.29.

8c: IR (cm⁻¹): 3256 (NH), 3179, 3165 (NH₂), 3085 (Aromatic CH), 2885 (Aliphatic CH), 1716 (C=O), 1659 (C=N), 1333 (SO₂). ¹H NMR: δ 10.06 (s, 1H, NH), 7.80 (d, J = 8.5 Hz, 2H, Ar–H), 7.71–7.48 (m, 3H,

Coumarin-H), 7.33 (t, J = 7.9 Hz, 1H, Coumarin-H), 7.06 (d, 2H, J = 8.5 Hz, 2H, Ar–H), 6.40 (s, 2H, NH₂), 6.18 (s, 1H, Coumarin-H), 4.13 (s, 2H, CH₂). ¹³C NMR: δ 161.30, 159.00, 157.16, 154.25, 145.77, 142.93, 140.36, 136.91, 132.21, 127.81, 126.84, 124.11, 120.93, 117.76, 117.07, 112.60 (Coumarin-C, Ar–C & Triazolo[3,4-*b*] [1,3,4]thiadiazole-C), 29.83 (CH₂). EI-MS (*m*/*z*) = 454. Anal. calcd. for C₁₉H₁₄N₆O₄S₂, %: C, 50.21; H, 3.10; N, 18.49. Found, %: C, 50.51; H, 3.25; N, 18.29.

4. Conclusions

In the current study, a novel series of triazolyl coumarin derivatives were synthesized and evaluated as anticancer agents against human colorectal cancer cell line (HCT116). The compounds **4c** and **8b** exhibited marked cytotoxic activity with 59.62 and 97.93% relative potency, respectively, compared to the potent anticancer drug doxorubicin. The molecular docking studies of the active compounds revealed that these compounds might act via inhibition of tyrosine kinases (CDK2). The active compounds are considered to be good candidates as newer anticancer agents, and further studies including preparation of newer analogues and toxicity testing are required for optimization of the activity which are being undertaken.

Data Availability

The data used to support the findings of this study are available from the corresponding author upon request.

Conflicts of Interest

The authors declare that they have no conflicts of interest.

Acknowledgments

The authors would like to thank the Deanship of Scientific Research, Princess Nourah Bint Abdulrahman University, for funding this study (Research Project No. 37-S-178).

References

- X.-M. Peng, G. L. V. Damu, and C.-H. Zhou, "Current developments of coumarin compounds in medicinal chemistry," *Current Pharmaceutical Design*, vol. 19, no. 21, pp. 3884–3930, 2013.
- [2] K. N. Venugopala, V. Rashmi, and B. Odhav, "Review on natural coumarin lead compounds for their pharmacological activity," *BioMed Research International*, vol. 2013, Article ID 963248, 14 pages, 2013.
- [3] C. Kontogiorgis, A. Detsi, and D. Hadjipavlou-Litina, "Coumarin-based drugs: a patent review (2008-present)," *Expert Opinion on Therapeutic Patents*, vol. 22, no. 4, pp. 437-454, 2012.
- [4] S. H. Bairagi, P. P. Salaskar, S. D. Loke, N. N. Surve, D. V. Tandel, and M. D. Dusara, "Medicinal significance of coumarins: a review," *International Journal of Pharmaceutical Research*, vol. 4, pp. 16–19, 2012.
- [5] K. P. Link, "The discovery of dicumarol and its sequels," *Circulation*, vol. 19, no. 1, pp. 97–107, 1959.

- [6] M. S. Bhatia, K. B. Ingale, P. B. Choudhari, N. M. Bhatia, and R. L. Sawant, "Application quantum and physicochemical molecular descriptors utilizing principal components to study mode of anticoagulant activity of pyridyl chromen-2-one derivatives," *Bioorganic & Medicinal Chemistry*, vol. 17, no. 4, pp. 1654–1662, 2009.
- [7] N. H. Holford, "Clinical pharmacokinetics and pharmacodynamics of warfarin: understanding the dose-effect relationship," *Clinical Pharmacokinetics*, vol. 11, no. 6, pp. 483–504, 1986.
- [8] M. E. Riveiro, N. D. Kimpe, A. Moglioni et al., "Coumarins: old compounds with novel promising therapeutic perspectives," *Current Medicinal Chemistry*, vol. 17, no. 13, pp. 1325–1338, 2010.
- [9] K. Hodák, V. Jakesová, and V. Dadák, "On the antibiotic effects of natural coumarins. VI. The relation of structure to the antibacterial effects of some natural coumarins and the neutralization of such effects," *Cesko-Slovenska Farmacie*, vol. 16, pp. 86–91, 1967.
- [10] C. A. J. Erdelmeier and O. Sticher, "Coumarin derivatives from *Eryngium campestre*," *Planta Medica*, vol. 51, no. 5, pp. 407–409, 1985.
- [11] A. Basile, S. Sorbo, V. Spadaro et al., "Antimicrobial and antioxidant activities of coumarins from the Roots of *Ferulago campestris* (Apiaceae)," *Molecules*, vol. 14, no. 3, pp. 939–952, 2009.
- [12] I. Kostova, S. Bhatia, P. Grigorov et al., "Coumarins as antioxidants," *Current Medicinal Chemistry*, vol. 18, no. 25, pp. 3929–3951, 2011.
- [13] W. K. Whang, H. S. Park, and I. H. Ham, "Natural compounds, fraxin and chemicals structurally related to fraxin protect cells from oxidative stress," *Experimental & Molecular Medicine*, vol. 37, no. 5, pp. 436–446, 2005.
- [14] X. Zhou, X. B. Wang, T. Wang, and L. Y. Kong, "Design, synthesis, and acetylcholinesterase inhibitory activity of novel coumarin analogues," *Bioorganic & Medicinal Chemistry*, vol. 16, no. 17, pp. 8011–8021, 2008.
- [15] L. Piazzi, A. Cavalli, F. Colizzi et al., "Multi-target-directed coumarin derivatives: hAChE and BACE1 inhibitors as potential anti-Alzheimer compounds," *Bioorganic & Medicinal Chemistry Letters*, vol. 18, no. 1, pp. 423–426, 2008.
- [16] S. J. Stanway, A. Purohit, L. W. L. Woo et al., "Phase I study of STX 64 (667 Coumate) in breast cancer patients: the first study of a steroid sulfatase inhibitor," *Clinical Cancer Research*, vol. 12, no. 5, pp. 1585–1592, 2006.
- [17] M. Curini, G. Carvotto, F. Epifano, and G. Giannone, "Chemistry and biological activity of natural and synthetic prenyloxycoumarins," *Current Medicinal Chemistry*, vol. 13, no. 2, pp. 199–222, 2006.
- [18] L. Navidpour, H. Shafaroodi, K. Abdi et al., "Design, synthesis, and biological evaluation of substituted 3-alkylthio-4,5-diaryl-4H-1,2,4-triazoles as selective COX-2 inhibitors," *Bioorganic & Medicinal Chemistry*, vol. 14, no. 8, pp. 2507– 2517, 2006.
- [19] A. A. El-Emam and T. M. Ibrahim, "Synthesis and anti-inflammatory and analgesic activity of some 3-(1-adamantyl)-4-substituted-5-mercapto-1,2,4-triazoles," *Arzneimittel-Forschung/Drug Research*, vol. 41, pp. 1260– 1264, 1991.
- [20] A. C. Pasqualotto, K. O. Thiele, and L. Z. Goldani, "Novel triazole antifungal drugs: focus on isavuconazole, ravuconazole and albaconazole," *Current Opinion in Investigational Drugs*, vol. 11, pp. 165–174, 2010.

- [21] X. Chai, J. Zhang, S. Yu et al., "Design, synthesis, and biological evaluation of novel 1-(1H-1,2,4-triazole-1-yl)-)-2-(2,4-difluorophenyl)-3-substituted benzylamino-2-propanols," *Bioorganic & Medicinal Chemistry Letters*, vol. 19, no. 6, pp. 1811–1814, 2009.
- [22] Y. Hou, J. Sun, Z. Pang et al., "Synthesis and antitumor activity of 1,2,4-triazoles having 1,4-benzodioxan fragment as a novel class of potent methionine aminopeptidase type II inhibitors," *Bioorganic & Medicinal Chemistry*, vol. 19, no. 20, pp. 5948–5954, 2011.
- [23] A. T. Mavrova, D. Wesselinova, Y. A. Tsenov, and P. Denkova, "Synthesis, cytotoxicity and effects of some 1,2,4-triazole and 1,3,4-thiadiazole derivatives on immunocompetent cells," *European Journal of Medicinal Chemistry*, vol. 44, no. 1, pp. 63–69, 2009.
- [24] Y. Kotaiah, K. Nagaraju, N. Harikrishna, C. V. Rao, L. Yamini, and M. Vijjulatha, "Synthesis, docking and evaluation of antioxidant and antimicrobial activities of novel 1,2,4-triazolo[3,4-b][1,3,4]thiadiazol-6-yl)selenopheno[2,3-d]pyrimidines," European Journal of Medicinal Chemistry, vol. 75, pp. 195–202, 2014.
- [25] S. N. Swamy, Basappa, B. S. Priya et al., "Synthesis of pharmaceutically important condensed heterocyclic 4,6disubstituted-1,2,4-triazolo-1,3,4-thiadiazole derivatives as antimicrobials," *European Journal of Medicinal Chemistry*, vol. 41, no. 4, pp. 531–538, 2006.
- [26] I. Khan, S. Zaib, A. Ibrar, N. S. Rama, J. Simpson, and J. Iqbal, "Synthesis, crystal structure and biological evaluation of some novel 1,2,4-triazolo[3,4-b]-1,3,4-thiadiazoles and 1,2,4-triazolo[3,4-b]-1,3,4-thiadiazines," *European Journal of Medicinal Chemistry*, vol. 78, pp. 167–177, 2014.
- [27] T. Besson, B. Joseph, P. Moreau, M. C. Viaeud, G. Coudert, and G. Guillaumet, "Synthesis and fluorescent properties of new heterobifunctional fluorescent probes," *Heterocycles*, vol. 34, pp. 273–291, 1992.
- [28] M. Cacic, M. Trkovnik, F. Cacic, and E. Has-Schon, "Synthesis and antimicrobial activity of some derivatives on the basis (7hydroxy-2-oxo-2*H*-chromen-4-yl)-acetic acid hydrazide," *Molecules*, vol. 11, no. 2, pp. 134–147, 2006.
- [29] C. Charitos, G. Kokotos, and C. Tzougraki, "Bifunctional coumarin derivatives in solution and solid phase synthesis of fluorogenic enzyme substrates," *Journal of Heterocyclic Chemistry*, vol. 38, no. 1, pp. 153–158, 2001.
- [30] A. A. EI-Emam, M. A. Moustafa, H. I. El-Subbagh, and M. B. EI-Ashmawy, "Triazoles and fused triazoles, III: facile and efficient synthesis of 2,5-disubstituted-s-triazolo[3,4-b]-1,3,4-thiadiazoles," *Monatshefte für Chemie*, vol. 121, pp. 221–225, 1990.
- [31] P. Skehan, R. Storeng, D. Scudiero et al., "New colorimetric cytotoxicity assay for anticancer-drug screening," *Journal of the National Cancer Institute*, vol. 82, no. 13, pp. 1107–1112, 1990.
- [32] O. Tacar, P. Sriamornsak, and C. R. Dass, "Doxorubicin: an update on anticancer molecular action, toxicity and novel drug delivery systems," *Journal of Pharmacy and Pharmacology*, vol. 65, no. 2, pp. 157–170, 2013.
- [33] S. A. K. Tanoli, N. U. Tanoli, S. Usmani, and A. G. Ferreira, "The exploration of interaction studies of smaller size, mostly ignored yet intrinsically inestimable molecules towards BSA: an example of STD and DOSY NMR," *Central European Journal of Chemistry*, vol. 12, no. 3, pp. 332–340, 2014.
- [34] T. Akiyama, J. Ishida, S. Nakagawa et al., "Genistein, a specific inhibitor of tyrosine-specific protein kinases,"

Journal of Biological Chemistry, vol. 262, pp. 5592-5595, 1987.

- [35] Y. Graziani, E. Erikson, and R. L. Erikson, "Differential inhibitory effects of various flavonoids on the activities of reverse transcriptase and cellular DNA and RNA polymerases," *European Journal of Biochemistry*, vol. 135, no. 3, pp. 583–589, 1983.
- [36] Protein Data Bank, http://www.rcsb.org/pdb.





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