Journal Pre-Proof

Bis-coumarins; Non-Cytotoxic Selective Urease Inhibitors and Antiglycation Agents

Uzma Salar, Arsalan Nizamani, Fizza Arshad, Khalid Mohammed Khan, Muhammed Imran Fakhri, Shahnaz Perveen, Nessar Ahmed, M. Iqbal Choudhary

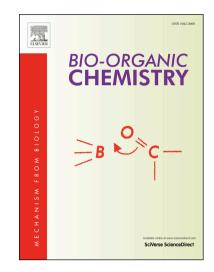
PII: S0045-2068(19)30717-5

DOI: https://doi.org/10.1016/j.bioorg.2019.103170

Reference: YBIOO 103170

To appear in: Bioorganic Chemistry

Received Date: 5 May 2019 Revised Date: 24 July 2019 Accepted Date: 30 July 2019



Please cite this article as: U. Salar, A. Nizamani, F. Arshad, K. Mohammed Khan, M. Imran Fakhri, S. Perveen, N. Ahmed, M. Iqbal Choudhary, *Bis*-coumarins; Non-Cytotoxic Selective Urease Inhibitors and Antiglycation Agents, *Bioorganic Chemistry* (2019), doi: https://doi.org/10.1016/j.bioorg.2019.103170

This is a PDF file of an article that has undergone enhancements after acceptance, such as the addition of a cover page and metadata, and formatting for readability, but it is not yet the definitive version of record. This version will undergo additional copyediting, typesetting and review before it is published in its final form, but we are providing this version to give early visibility of the article. Please note that, during the production process, errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

© 2019 Published by Elsevier Inc.

Bis-coumarins; Non-Cytotoxic Selective Urease Inhibitors and Antiglycation Agents

Uzma Salar,^{a,b} Arsalan Nizamani,^a Fizza Arshad,^a Khalid Mohammed Khan,*a,^e Muhammed Imran Fakhri,^a Shahnaz Perveen,^c Nessar Ahmed,^d M. Iqbal Choudhary^{a,b,f}

^aH. E. J. Research Institute of Chemistry, International Center for Chemical and Biological Sciences, University of Karachi, Karachi-75270, Pakistan

^bDr. Panjwani Center for Molecular Medicine and Drug Research, International Center for Chemical and Biological Sciences, University of Karachi, Karachi-75270, Pakistan ^cPCSIR Laboratories Complex, Karachi, Shahrah-e-Dr. Salimuzzaman Siddiqui, Karachi-75280, Pakistan

^dCentre for Biomedicine, School of Healthcare Science, Manchester Metropolitan University, Manchester M1 5GD, United Kingdom

^eDepartment of Clinical Pharmacy, Institute for Research and Medical Consultations (IRMC), Imam Abdulrahman Bin Faisal University, P.O. Box 31441, Dammam, Saudi Arabia ^fDepartment of Biochemistry, King Abdulaziz University, Jeddah-214412, Saudi Arabia

Abstract: The current study is concerned with the identification of lead molecules based on the bis-coumarin scaffold having selective urease inhibitory and antiglycation activities. For that purpose, bis-coumarins (1-44) were synthesized and structurally characterized by different spectroscopic techniques. Eight derivatives 4, 8-10, 14, 17, 34, and 40 demonstrated urease inhibition in the range of IC₅₀ = $4.4 \pm 0.21 - 115.6 \pm 2.13 \mu M$, as compared to standard thiourea $(IC_{50} = 21.3 \pm 1.3 \mu M)$. Especially, compound 17 $(IC_{50} = 4.4 \pm 0.21 \mu M)$ was found to be fivefold more potent than the standard. Kinetic studies were also performed on compound 17 in order to identify the mechanism of inhibition. Kinetic studies revealed that compound 17 is a competitive inhibitor. Antiglycation activity was evaluated using glycation of bovine serum albumin by methylglyoxal in vitro. Compounds 2, 11-13, 16, 17, 19-22, 35, 37, and 42 showed good to moderate antiglycation activities with IC₅₀ values of 333.63 - 919.72 μ M, as compared to the standard rutin (IC₅₀ = 294.46 \pm 1.5 μ M). Results of both assays showed that the compounds with urease inhibitory activity did not show any antiglycation potential, and vice versa. Only compound 17 showed dual inhibition potential. All compounds were also evaluated for cytotoxicity. Compounds 17, 19, and 37 showed a weak toxicity towards 3T3 mouse fibroblast cell line. All other compounds were found to be non-cytotoxic. Urease inhibition is an approach to treat infections caused by ureolytic bacteria whereas inhibition of glycation of proteins is a strategy to avoid late diabetic complications. Therefore, these compounds may serve as leads for further research.

1

^{*}Corresponding Author: khalid.khan@iccs.edu; drkhalidhej@gmail.com; Tel. 00922134824910; Fax. 00922134819018

Keywords: *Bis*-coumarins; urease; glycation; advanced glycation end products; peptic ulcer; diabetes; non-cytotoxic.

Introduction

Bis-coumarins are biologically active pharmacophores, initially isolated from natural sources [1,2]. Several biological activities are associated with *bis*-coumarins, such as α -glucosidase [3], urease [4], nucleotide pyrophosphatases-1 [5], and DNA polymerase β -lyase inhibitory activities [6]. *Bis*-coumarins are also reported to possess anticoagulant, and hemorrhagic properties [7]. However, there is still a need to explore this class for a wide spectrum of pharmacological activities.

Urease (amidohydrolase EC 3.5.1.5) is a metalloenzyme, contains nickel in its active site. It catalyzes the hydrolysis of urea into carbon dioxide and ammonia [8,9]. This enzyme synthesizes by numerous plants, animals, bacteria, and other organisms [10]. Hyperactivity of urease is harmful to human and animal health, as well as for the agricultural sector. Urease is a key virulence in the pathogenesis of urolithiasis, urinary catheter encrustation, pyelonephritis, hepatic coma, and hepatic encephalopathy [11,12]. It also participates in the pathologies caused by ureolytic bacteria *Helicobacter pylori* (HP). It facilitates bacteria to survive in stomach at acidic pH during initial colonization. Therefore, it plays an important role in the pathologies of ulcers (gastric and peptic), and cancer [13-16]. In the agriculture sector, hyperactivity of urease leads to considerable economic and environmental damage by liberating aberrantly large quantities of ammonia into the atmosphere, during the process of urea fertilization [17]. Therefore, it is important to develop strategies based on urease inhibition to solve the problems caused by urease producing bacteria.

Glycation is a non-enzymatic reaction in which reducing sugars non-enzymatically bind with the amino terminal of proteins *via* a nucleophilic addition reaction, ultimately giving rise to advanced glycation end products (AGEs). Haemoglobin, serum albumin, collagen, elastin, and crystalline are common proteins that undergo glycation. Changes in their structures and functions lead to different abnormalities such as atherosclerosis, neuropathy, diabetic retinopathy, diabetic nephropathy, *etc*. This process is cumulatively called glycation stress [18,19]. In this process, reactive intermediates such as methylglyoxal (MG) are more prone to bind with amino groups as compared to their carbohydrate precursors. Eighty percent (80%) of blood proteins is serum albumin, which is more likely to be glycated [20]. As a result of complex rearrangements,

substitution, and addition reactions of glycated proteins, AGEs are produced in the body which change the functions of proteins, and accumulate with time in different tissues [21-23]. Many late diabetic complications, such as retinopathy, nephropathy, cataracts, atherosclerosis, and osteoporosis are due to the glycation of vital proteins, and accumulation of AGEs [24]. The inhibition of glycation process plays a pivotal role in the prevention of many late diabetic complications. Therefore, it is important to find inhibitors for glycation.

Bis-coumarins have not yet been reported for their antiglycation activity. Figure-1 showed that chromone ring, a positional isomer of coumarin, is the main scaffold of rutin which encouraged us to evaluate the compounds **1-44** for their antiglycation activity. Furthermore, we have previously reported *bis*-coumarins for urease inhibitory activity [4]. New members of this series were thus evaluated to identify more potent urease inhibitors [Figure-1]. In brief, forty-four derivatives were synthesized and evaluated for their urease inhibitory, and antiglycation activities. After knowing the selective potential of compounds, cytotoxicity was also checked. To the best of our knowledge, except compounds **6**, **14**, **16**, **17**, **19**, **20**, **23-26**, **28**, and **30-32** [25-29], the rest of the compounds were identified as new.

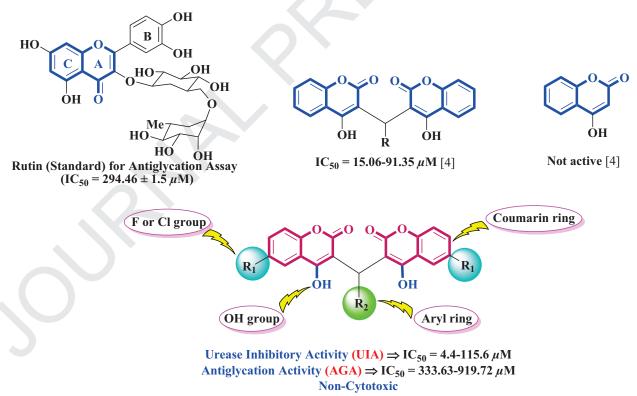


Figure-1: Structural similarity between rutin and *bis*-coumarins 1-44 as a rationale of the current study.

Results and Discussion

Chemistry

Bis-coumarin derivatives **1-44** were synthesized by reacting 6-fluoro-4-hydroxy, 4-hydroxy, and 6-chloro-4-hydroxy coumarins with a variety of benzaldehydes in the presence of tetraethylammonium bromide (TEAB) as a catalyst. Reactions were performed in distilled water (Scheme-1) and checked periodically by TLC analysis. Precipitates of products were obtained in good yields (Table-1). Compounds were structurally identified by various spectroscopic analyses such as ¹H- and ¹³C-NMR as well as FAB-, ESI-, and HRESI-MS.

Scheme-1: Synthesis of bis-coumarin derivatives 1-44.

Characteristic spectral features of representative compound 21

Structure elucidation of a new compound **21** is presented here as an example. 1 H- and 13 C-NMR spectra of compound **21** were recorded in DMSO- d_6 . Characteristic signal of methine proton resonated as a singlet at $\delta_{\rm H}$ 5.60, also confirmed the formation of the *bis*-coumarin scaffold. Amongst the protons of coumarin nucleus, H-7 and H-7' resonated at $\delta_{\rm H}$ 7.53 as triplet, and showed *ortho* coupling (t, $J_{7,6/7',6'} = J_{7,8/7',8'} = 6.9$ Hz) with adjacent protons, H-6/H-6' and H-8/H-8'. Similarly, H-6 and H-6' appeared as a triplet at $\delta_{\rm H}$ 7.26, and also showed *ortho* coupling (t, $J_{6,5/6',5'} = J_{6,7/6',7'} = 7.5$ Hz) with neighbouring protons. H-5 and H-5' appeared as the most downfield signal at $\delta_{\rm H}$ 7.82 with *ortho* coupling (d, $J_{5,6/5',6'} = 6.3$ Hz) with the adjacent H-6 and H-6'. H-8 and H-8' appeared at $\delta_{\rm H}$ 7.27, *ortho* coupled (d, $J_{8,7/8',7'} = 8.1$ Hz) with H-7 and H-7'. H-2'' and H-6'' of ring R₂ resonated as a broad singlet at $\delta_{\rm H}$ 7.12 (Figure-2).

7.27 (d,
$$J_{8,7} = 8.1 \text{ Hz}$$
)

7.27 (d, $J_{8,7} = 8.1 \text{ Hz}$)

7.27 (d, $J_{8,7} = 8.1 \text{ Hz}$)

H

7.53 (t, $J_{7,6} = J_{7,8} = 6.9 \text{ Hz}$)

H

7.53 (t, $J_{7,6} = J_{7,8} = 6.9 \text{ Hz}$)

7.53 (t, $J_{7,6} = J_{7,8} = 6.9 \text{ Hz}$)

7.54 (t, $J_{6,5} = J_{6,7} = 7.5 \text{ Hz}$)

7.55 (t, $J_{7,6} = J_{7,8} = 6.9 \text{ Hz}$)

7.56 (t, $J_{6,5} = J_{6,7} = 7.5 \text{ Hz}$)

7.26 (t, $J_{6,5} = J_{6,7} = 7.5 \text{ Hz}$)

7.26 (t, $J_{6,5} = J_{6,7} = 7.5 \text{ Hz}$)

7.26 (t, $J_{6,5} = J_{6,7} = 7.5 \text{ Hz}$)

7.26 (t, $J_{6,5} = J_{6,7} = 7.5 \text{ Hz}$)

7.27 (d, $J_{8,7} = 8.1 \text{ Hz}$)

7.53 (t, $J_{7,6} = J_{7,8} = 6.9 \text{ Hz}$)

7.26 (t, $J_{6,5} = J_{6,7} = 7.5 \text{ Hz}$)

130.2

124.0

164.2 O

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

167.2

1

Figure-2: ¹H- and ¹³C-NMR chemical shifts of compound 21.

Total 16 signals of carbon (eleven methine and fourteen quarternary carbons) appeared in 13 C-NMR broad-band decoupled spectrum (DMSO- d_6). Most downfield signal of C-2 and C-2', which are the quaternary carbons of lactone moiety, appeared at δ_C 167.2. Similarly, other quaternary C-10/C-10', and C-4/C-4' resonated at δ_C 164.2 and 152.4 as the downfield signals due to the adjacent electronegative oxygen atom. An upfield characteristic signal resonated at δ_C 35.0, corresponding to the methine CH carbon.

The structure of compound **21** was further confirmed with the help of FAB (Neg.)-MS, ESI-MS, and HRESI-MS. FAB (Neg.)-MS of compound **21** displayed the peaks at m/z = 584 [M-H]⁻¹, 586 [M+2-H]⁻¹ and 588 [M+4-H]⁻¹ which indicated the presence of two bromine atoms. Similarly, ESI-MS of compound **21** displayed the [M+H]⁺, [M+2+H]⁺, and [M+4+H]⁺ at m/z 585, 587, and 589, respectively. HR-(ESI)MS showed the [M+H]⁺ at m/z 584.9208, corresponding to the formula $C_{25}H_{15}Br_2O_7$ Calcd (584.9184). The UV spectrum of compound **21** showed absorption at 298 nm, characteristic of coumarin moiety. The structure of new compound **21** was thus deduced unambiguously.

Bioactivities in vitro

Forty-four *bis*-coumarin derivatives were evaluated for urease inhibitory, and antiglycation activities (Table-1). Eight compounds **4**, **8-10**, **14**, **17**, **34**, and **40**, showed a good to moderate urease inhibitory activity in the range of $IC_{50} = 4.4 \pm 0.21$ -115.6 $\pm 2.13 \mu M$, as compared to standard thiourea ($IC_{50} = 21.3 \pm 1.3 \mu M$). Whereas, compounds **2**, **11-13**, **16**, **17**, **19**, **20-22**, **35**, **37**, and **42** demonstrated good to moderate antiglycation activities in the range of $IC_{50} = 333.63 - 919.72 \mu M$, as compared to standard rutin ($IC_{50} = 294.46 \pm 1.5 \mu M$). *Bis*-coumarins have never been reported before for its antiglycation activity. It is worth mentioning that the compounds showed selective activity in both assays. *Bis*-coumarins which showed urease inhibitory activity

did not give the antiglycation potential, and *vice versa*. All compounds were also checked for their cytotoxicity and largely found to be non-cytotoxic.

Table-1: *In vitro* urease inhibitory activity, antiglycation activity, and cytotoxicity of *bis*-coumarins (1-44).

Compounds	R ₂	Urease Inhibitory Activity IC ₅₀ ± SEM ^a	Antiglycation Activity IC ₅₀ ± SEM ^a	Cytotoxicity $IC_{50} \pm SEM^{a}$
	Cate	egory A $(R_1 = F)$	1030 - 22111	1030 = 22111
1	OMe 5" OMe	NA ^b	NA ^b	NT°
2	6" F 5" 3" OMe	NA ^b	378.03 ± 0.75	>30
3	Br 5" 3" OMe	NAb	NA ^b	NT°
4	6" C1 5" 3"	87.0 ± 0.88	NA ^b	NT°
5	OMe Br 4"	NA ^b	NA ^b	NT°
6	6" 2" 3" Me	NA ^b	NA ^b	NT°
7	OMe MeO 4"	NAb	NA ^b	NT°
8	6" 2" 2" 3"" OMe 6" 5"'	30.3 ± 1.01	NAb	NT°

9	6" 2" 3" CF ₃	79.8 ± 1.01	NA ^b	NT°
10	6" 2" CF ₃	193.7 ± 0.66	NAb	NT°
11	6" OH F 4"	NA ^b	399.47 ± 1.88	>30
12	OH Cl 4"	NA ^b	465.33 ± 1.99	>30
13	OH 5" OH	NAb	443.64 ± 0.57	>30
14	6" 2" Br OMe	187.3 ± 1.75	NA ^b	NT°
15	OMe MeO 3"	NAb	NAb	$ m NT^c$
Category B $(R_1 = H)$				
16	OH 5" OH	NAb	351.85 ± 3.56	>30
17	OH OH	4.4 ± 0.21	374.45 ± 1.21	18.70 ± 0.57
18	Me S" Me	NA ^b	NAb	NT°

19	OH Cl 4"	NA ^b	397.07 ± 2.21	>30
20	OH 5" OH	NAb	434.71 ± 2.73	19.42 ± 0.79
21	Br OH	NAb	432.58 ± 0.76	>30
22	Cl 2" Cl	NA ^b	428.84 ± 2.44	>30
23	OMe MeO 4"	NA ^b	NA ^b	NT°
24	OMe 5"OMe	NAb	NAb	NT°
25	6" 2" OMe OEt	NA ^b	NA ^b	NT°
26	OMe 5" OMe	NA ^b	NA ^b	NT°
27	6" F 5" 3" Br	NA ^b	NA ^b	NT°
28	Br MeO 3" OMe	NA ^b	NA ^b	NT°
22 23 24 25 26	Br OH OH OH OH OH OMe OMe OMe OMe	NA ^b NA ^b NA ^b NA ^b	1428.84 ± 2.44 100	>30 NTc NTc NTc

29	6" 2" 2" 3"" 4""	NAb	NAb	NT°
30	6" 2" OMe OH	NAb	NAb	NT°
31	6" 2" 3" SMe	NA ^b	NA ^b	NT°
	Cate	gory $C(R_1 = Cl)$		
32	6" 2" Br OMe	NA ^b	NAb	NT°
33	6" 2" 2" 3"" 4"" 5""	NAb	NA ^b	NT°
34	OH OH	115.6 ± 2.13	NAb	NT°
35	Br OH	NA ^b	744.82 ± 5.64	>30
36	7" OH 6" 3"	NAb	NA ^b	NT°
37	HO OMe	NAb	919.72 ± 1.98	21.11 ± 1.27
38	OMe Br 4"	NA ^b	NAb	NT°

39	OH HO 4"	NAb	NA ^b	NT°
40	6" 2" Br OH	36.5 ± 1.37	NAb	NT°
41	6" Br F 4"	NA ^b	NA ^b	NT°
42	MeO OMe	NAb	333.63 ± 1.98	>30
43	6" F 5" 3" OMe	NAb	NAb	NT°
44	Br OMe OH	NAb	NA ^b	NT°
	Thiourea ^d	21.3 ± 1.3		
Standards	Rutine		294 ± 1.5	-
CEM (Ct. 1 1	Cycloheximide ^f		- d (G) 1 1 :	0.26 ± 0.1

SEM^a (Standard error of mean); NA^b (Not Active); NT^c (Non Toxic); Thiourea^d (Standard inhibitor of urease inhibitory activity); Rutin^e (Standard inhibitor of antiglycation activity); Cycloheximide^f (Standard inhibitor for cytotoxicity)

Structure-activity relationship (SAR)

All structural features of *bis*-coumarin derivatives such as coumarin ring, substitutions on coumarin rings (R_1) , and aryl ring (R_2) apparently playing their role in the inhibitory activity, and variation in the activity can be attributed to the different substitution pattern on coumarin, and aryl rings (Figure-3).

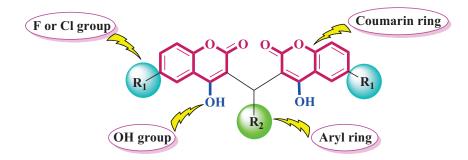


Figure-3: General structural features of bis-coumarin.

Results presented in Table-1 indicate that almost all compounds selectively exhibited their potential in both assays, and found to be non-toxic as well.

Bis-coumarins **1-44** were divided into three categories in order to understand the structure-activity pattern. In category "A", fluoro groups are present at C-6 and C-6' of *bis*-coumarin i.e. **1-15**. Category "B" compounds **16-31** has no substitution on both coumarin rings, and category "C" has chloro groups at C-6 and C-6' *i.e.* compounds **32-44**.

In category "A", non-cytotoxic compound 8 (IC₅₀ = $30.3 \pm 1.01 \,\mu\text{M}$) with 3"-benzyloxy and 4"methoxy substitutions showed comparable urease inhibitory activity with the standard thiourea $(IC_{50} = 21.3 \pm 1.3 \,\mu\text{M})$, and did not show any antiglycation potential. The good urease inhibitory activity of the compound 8 could be due to π - π stacking of the benzyloxy group with the active site of urease enzyme. Activity of compound 8 was compared with the activity of non-cytotoxic compound 14 (IC₅₀ = 187.3 \pm 1.75 μ M) which has a bromo group instead of a benzyloxy at C-3", showed six times less urease inhibitory activity. This indicated the involvement of benzyloxy in the inhibitory activity. Similarly, compound 14 did not show any antiglycation potential. It is worth mentioning that both compounds 8 and 14 showed selective activity towards the urease, as well as being non-cytotoxic. Other compounds with different positional combination of methoxy and bromo groups such as compounds 3, 5, and 15, did not show any urease inhibitory and antiglycation activities. However, all three were found to be non-cytotoxic. The activity of compound 3 can be compared with compound 2 (IC₅₀ = 378.03 \pm 0.75 μ M) which has a fluoro group, instead of a bromo at C-2", showed an antiglycation potential comparable to the standard rutin (IC₅₀ = 294.46 \pm 1.5 μ M). It showed that fluoro group on aryl ring plays an important role in the antiglycation activity (Figure-4).

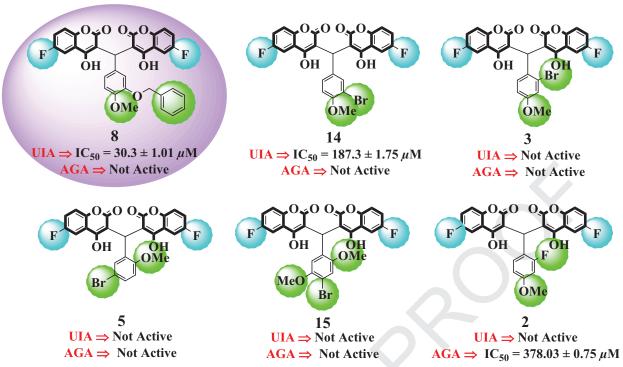


Figure-4: Structure-activity relationship of compounds 2, 3, 5, 8, 14, and 15.

In the same category "A", compounds with the halogen substitutions (**4**, **9**, and **10**), and the compounds with combinations of halogen and hydroxy groups (**11** and **12**), showed a good selectivity and non-cytotoxicity. For example, compound **4** ($IC_{50} = 87.0 \pm 0.88 \,\mu\text{M}$) with 2",4"-dichloro substitutions showed a selective urease inhibitory activity, and no antiglycation activity. The two chloro groups on aryl ring might form some polar interactions with the active site of the urease enzyme. Similarly, another compound **9** ($IC_{50} = 79.8 \pm 1.01 \,\mu\text{M}$) with a 4"-trifluoromethyl group, also showed urease inhibitory activity but no antiglycation activity. This might be that trifluoromethyl group establish polar interaction with the active site of urease enzyme. However, its positional isomer **10** ($IC_{50} = 193.7 \pm 0.66 \,\mu\text{M}$) with a 4"-trifluoromethyl group, showed two times lower activity. It showed that C-3" substitution is not favourable for urease inhibition (Figure-5).



Figure-5: Structure-activity relationship of compounds 4, 9, and 10.

Compounds with the combination of halogen and hydroxy, such as **11** (OH and F *para* to each other) and **12** (OH and Cl *para* to each other) showed no urease inhibition, however, they did show antiglycation activity. Similarly, compound **13** (IC₅₀ = 443.64 \pm 0.57 μ M) with 2",4"-dihydroxy substitutions also showed antiglycation potential (Figure-6). Thus hydroxy group is apparently playing an important role in the antiglycation activity. Phenolic OH known to have radical scavenging activity through phenolic OH groups, along with other structural features which may contribute to the inhibition of protein glycation.



Figure-6: Structure-activity relationship of compounds 11-13.

Involvement of hydroxy groups in the antiglycation activity was further deduced by examining antiglycation activity pattern of compounds **16** (IC₅₀ = 351.85 ± 3.56 μ M), **17** (IC₅₀ = 374.45 ± 1.21 μ M), and **20** (IC₅₀ = 434.71 ± 2.73 μ M) belong to category "B". Compound **16** (IC₅₀ = 351.85 ± 3.56 μ M), having a 2,3-dihydroxy substitution, was selectively active against glycation but no urease inhibitory activity. The antiglycation activity of compound **16** can be compared with compound **20** (IC₅₀ = 434.71 ± 2.73 μ M) which has a 2,4-dihydroxy group, instead of 2,3-dihydroxy, and showed lower antiglycation activity. This shows that the positions of OH substitution also plays a role in the activity. However, incorporation of one or more hydroxy group in compound **17** (IC₅₀ = 374.45 ± 1.21 μ M) enhanced the antiglycation potential to a considerable

level. In addition, compound 17 was found to be a potent urease inhibitor, with five times more activity than the standard thiourea (IC₅₀ = $21.3 \pm 1.3 \mu$ M). Interestingly, this is the only compound which showed dual inhibitory activity against urease and glycation (Figure-7).

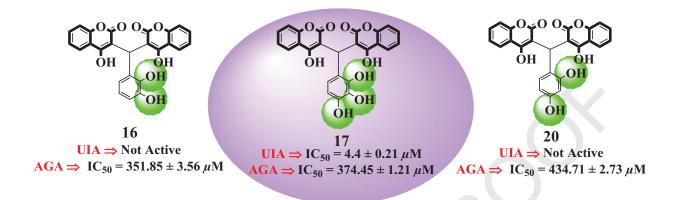


Figure-7: Structure-activity relationship of compounds 16, 17, and 20.

Other compounds with the halogen substitutions such as compound 22 with 3",5"-dichloro substitutions, showed an antiglycation activity (IC₅₀ = 428.84 \pm 2.44 μ M) but no urease inhibitory activity. Compounds having the combination of halogen (Cl and Br) with the hydroxy group such as 19 and 21, did not display any urease inhibition, however, both compounds 19 (IC₅₀ = 432.58 \pm 0.76 μ M) and 21 (IC₅₀ = 397.07 \pm 2.21 μ M) showed antiglycation activity (Figure-8). All three compounds were found to be non-cytotoxic.



In category "C", compound **40** (IC₅₀ = 36.5 \pm 1.37 μ M) having 3″-bromo-4″-hydroxy substitutions, showed a urease inhibitory activity comparable to standard thiourea (IC₅₀ = 21.3 \pm 1.3 μ M) and found non-cytotoxic. The hydroxy group is apparently making the hydrogen bonding interaction as well as the bromo group involved in some polar interaction with the active site of the enzyme. Activity of compound **40** can be compared with analogs **44** and **35** which have additional methoxy and bromo group, respectively, at 5″-position of R₂, a complete loss of activity

was observed. The inactivity of the compounds might be due to the additional groups which create the steric hindrance for the hydroxy group to bind with the enzyme active site. However, compound **35** (IC₅₀ = 744.82 \pm 5.64 μ M) showed selective antiglycation activity and also found to be non-cytotoxic. Furthermore, non-cytotoxic compounds **42** (IC₅₀ = 333.63 \pm 1.98 μ M) and **37** (IC₅₀ = 919.72 \pm 1.98 μ M) showed a selective good to moderate antiglycation activity. Compound **42** has a 3",5"-dibromo-4"-hydroxy substitution while compound **37** is with 4"-bromo-3",5"-dimethoxy substitutions at phenyl ring R₂. It is worthy to note that compounds **35**, **37**, and **42**, are 3",4",5"-trisubstituted analogs which demonstrated the selective antiglycation potential. Another compound **34** (IC₅₀ = 115.6 \pm 2.13 μ M) having a 3",4",5"-trihydroxy group, showed selective but weak urease inhibitory activity. Its activity compared with the structurally similar and most potent compound **17** (IC₅₀ = 4.4 \pm 0.21 μ M). Compound **34** might have a conformation which does not fit well into the active site of enzyme. Similarly, additional chloro groups at C-6 and C-6' positions are not playing their role in the activity (Figure-9).

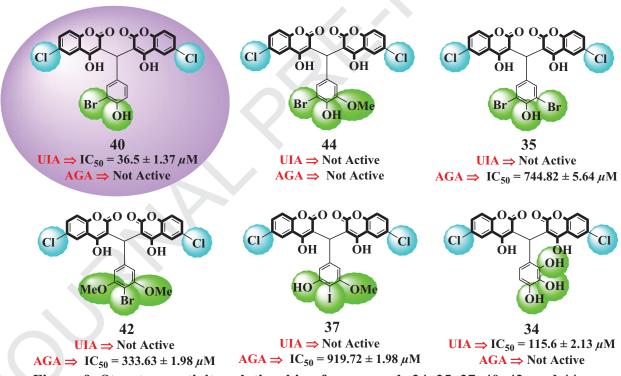


Figure-9: Structure-activity relationship of compounds 34, 35, 37, 40, 42, and 44.

In brief, it was observed that all derivatives, except 17, were found to be selective and non-cytotoxic. Our next goal is to place heterocycles on *bis*-coumarin skeleton in order to see their effects on selectivity and cytotoxicity.

Conclusion

Synthetic *bis*-coumarins **1-44** were evaluated for their urease inhibitory and antiglycation activities. Seven derivatives **4**, **8-10**, **14**, **34**, and **40** showed selective urease inhibition, whereas, twelve analogs **2**, **11-13**, **16**, **17**, **19-22**, **35**, **37**, and **42** demonstrated antiglycation potential. Only compound **17** showed dual inhibition. All compounds were largely found to be non-cytotoxic. Newly identified compounds, based on *bis*-coumarin scaffold, may serve as leads for future research for more powerful, non-cytotoxic, and selective agents against urease and protein glycation.

Experimental

Materials and methods

Thin layer chromatography (TLC) was performed on pre-coated silica gel aluminum plates (Kieselgel 60 F-254, 0.20 mm, Merck, Darmstadt, Germany). Chromatograms were visualized by using a handhold UV lamp at 254, and 365 nm or placing in iodine vapors. Fast atom bombardement mass spectra (FAB MS) were recorded on a Finnigan MAT-311A (Germany) (70 eV) spectrometers, electrospray ionization mass spectra (ESI-MS, HRESI-MS) were recorded on a QSTAR XL LCMS-MS, and ABSciex (Germany) (50 kV) mass spectrometers, and the data are tabulated as m/z. 1 H- and 13 C-NMR spectroscopic analysis was performed on an Avance Bruker (Germany) AM spectrometer 300 MHz machine. Splitting patterns for 1 H-NMR spectra were as follows, s (singlet); d (doublet); t (triplet); m (multiplet). Chemical shifts are reported in δ (ppm) and coupling constants are given in Hz. All solvents and reagents were of reagent grade, and used directly without purification. Melting points of the compounds were determined on Büchi-M560 melting point apparatus.

General procedure for the syntheses of bis-coumarin derivatives 1-44

6-Fluoro-4-hydroxy /6-chloro-4-hydroxy /4-hydroxy coumarin (1 mmol), and a variety of aromatic aldehydes (0.5 mmol), as well as 10 mol% of tetraethylammonium bromide (TEAB) were dissolved in distilled water (15 mL) in a 100 mL round-bottommed flask. The reaction mixture was refluxed for 2 h. Periodic TLC was taken to check the progress of reaction. Resulting precipitates were filtered, and washed with distilled water. This afforded products **1-44** in high yields.

3,3'-((2",3"-Dimethoxyphenyl)methylene)bis(6-fluoro-4-hydroxy-2H-chromen-2-one) (1)

White Solid; Yield: 82%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, Acetone- d_6): δ 11.25 (s, 2H, 2OH), 7.67 (d, $J_{5,6/5',6'}$ = 8.7 Hz, 2H, H-5, H-5'), 7.50 (d, $J_{7,8/7',8'/8,7/8',7'}$ = 5.4 Hz, 4H, H-7, H-8, H-7', H-8'), 7.01 (m, 3H, H-4", H-5", H-6"), 6.22 (s, 1H, -CH-), 3.81 (s, 3H, OCH_{3a}), 3.49 (s, 3H, OCH_{3b}); ¹³C-NMR (75 MHz, DMSO- d_6): δ 165.8, 165.8, 163.8, 168.3, 159.3, 159.3, 156.1, 152.3, 148.5, 146.4, 136.0, 124.4, 122.3, 121.1, 121.0, 120.9, 118.1, 117.8, 117.5, 117.3, 110.5, 109.3, 108.9, 104.5, 59.3, 55.4, 32.7; FAB (Pos.)-MS m/z = 509 [M+H]⁺; ESI-MS m/z = 509 [M+H]⁺; HRESI-MS Calcd for $C_{27}H_{19}F_{2}O_{8}$ [M+H]⁺: m/z = 509.1047, found 509.1090.

3,3'-((2"-Fluoro-4"-methoxyphenyl)methylene)*bis*(6-fluoro-4-hydroxy-2*H*-chromen-2-one) (2)

White Solid; Yield: 87%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, DMSO- d_6): δ 11.31 (bd.s, 2H, 2OH), 7.68 (d, $J_{5,6F/5',6F}$ = 8.7 Hz, 2H, H-5, H-5'), 7.52 (d, $J_{8,7/8',7'/6'',5''}$ = 5.4 Hz, 3H, H-8, H-8', H-6''), 7.32 (m, 1H, H-3"), 6.75 (m, 3H, H-7, H-7', H-5"), 6.19 (bd.s, 1H, -CH-), 3.80 (s, 3H, OCH₃); ¹³C-NMR (75 MHz, DMSO- d_6): δ 160.2, 159.9, 159.9, 159.2, 159.2, 156.7, 149.7, 148.5, 129.1, 120.0, 119.7, 119.3, 118.7, 118.4, 117.4, 117.3, 114.9, 114.8, 113.6, 112.1, 109.5, 109.2, 108.5, 101.5, 55.5, 28.3; FAB (Neg.)-MS m/z = 495 [M-H]-1; ESI-MS m/z = 477 [M+H-HF]+; HRESI-MS Calcd for $C_{26}H_{15}F_2O_7$ [M+H-HF]+: m/z = 477.0785, Found 477.0787.

$3,3'-((2''-Bromo-4''-methoxyphenyl)methylene) bis (6-fluoro-4-hydroxy-2H-chromen-2-one) \\ (3)$

White Solid; Yield: 79%; M.p.: 230-232 °C; ¹H-NMR (300 MHz, Acetone- d_6): δ 11.47 (bd.s, 2H, 2OH), 7.68 (d, $J_{5,6F/5',6F}$ = 8.1 Hz, 2H, H-5, H-5'), 7.52 (bd.s, 5H, H-7, H-8, H-7', H-8', H-3"), 7.32 (d, $J_{6'',5''}$ = 8.1 Hz, 1H, H-6"), 7.02 (d, $J_{5'',6''}$ = 8.1 Hz, 1H, H-5"), 6.15 (bd.s, 1H, -CH-), 3.86 (bd.s, 3H, OCH₃); ¹³C-NMR (75 MHz, DMSO- d_6): δ 160.1, 160.1, 159.5, 159.5, 156.9, 149.9, 148.8, 129.2, 125.2, 120.0, 119.8, 119.4, 118.9, 118.6, 117.6, 117.4, 114.8, 114.6, 113.7, 112.3, 109.4, 109.2, 108.4, 101.6, 55.7, 28.5; FAB (Neg.)-MS m/z = 556 [M-H]-¹, 558 [M+2-H]-¹; ESI-MS m/z = 557 [M+H]+, 559 [M+2+H]+; HRESI-MS Calcd for $C_{26}H_{15}BrF_{2}O_{7}$ [M+H]+: m/z = 557.0047, Found 557.0010.

3,3'-((2",4"-Dichlorophenyl)methylene)bis(6-fluoro-4-hydroxy-2H-chromen-2-one) (4)

White Solid; Yield: 81%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, Acetone- d_6): δ 7.68 (dd, $J_{5,7//5',7'}$ = 2.1 Hz, $J_{5,6F/5',6F'}$ = 7.8 Hz, 2H, H-5, H-5'), 7.57 (d, $J_{6'',5''}$ = 8.4 Hz, 1H, H-6"), 7.49 (m, 4H, H-8, H-8', H-3", H-5"), 7.39 (dd, $J_{7,5/7',5'}$ = 2.1 Hz, $J_{7,8/7',8'}$ = 6.6 Hz, 2H, H-7, H-7'), 6.24 (s, 1H, -CH-); ¹³C-NMR (75 MHz, DMSO- d_6): δ 160.4, 160.1, 159.3, 157.1, 156.5, 156.1, 148.2, 148.0, 141.5, 135.4, 132.5, 131.7, 130.0, 126.6, 125.0, 124.9, 119.1, 119.0, 115.2, 115.0, 111.4, 111.3, 100.2, 100.0, 28.2; FAB (Neg.)-MS m/z = 516 [M-H]-¹, 518 [M+2-H]-¹; ESI-MS m/z = 517 [M+H]+; HRESI-MS Calcd for $C_{25}H_{13}Cl_2F_2O_6$ [M+H]+: m/z = 517.0057, Found 517.0071.

3,3'-((5"-Bromo-2"-methoxyphenyl)methylene)*bis*(6-fluoro-4-hydroxy-2*H*-chromen-2-one) (5)

White Solid; Yield: 77%; M.p.: 245-247 °C; ¹H-NMR (300 MHz, Acetone- d_6): δ 11.21 (bd.s, 2H, 2OH), 7.67 (d, $J_{5,6F/5',6F'}$ = 8.7 Hz, 2H, H-5, H-5'), 7.51 (d, $J_{8,7/8',7'/6'',4''}$ = 5.7 Hz, 3H, H-8, H-8', H-6"), 7.43 (bd.s, 2H, H-7, H-7'), 6.96 (d, $J_{3'',4''/4'',3''}$ = 8.7 Hz, 2H, H-3", H-4"), 6.14 (s, 1H, -CH-), 3.61 (s, 3H, OCH_{3a}); ¹³C-NMR (75 MHz, DMSO- d_6): δ 165.8, 165.8, 163.5, 159.3, 157.5, 156.5, 156.2, 148.5, 148.5, 133.1, 131.1, 129.2, 124.2, 120.8, 120.7, 118.3, 117.9, 117.6, 117.4, 113.2, 111.4, 109.3, 109.0, 103.7, 55.8, 32.9; FAB (Neg.)-MS m/z = 556 [M-H]-1, 558 [M+2-H]-1; ESI-MS m/z = 557 [M+H]+, 559 [M+2+H]+; HRESI-MS Calcd for $C_{26}H_{16}BrF_{2}O_{7}$ [M+H]+: m/z = 557.0047, Found 557.0010.

3,3'-(p-Tolylmethylene)bis(6-fluoro-4-hydroxy-2H-chromen-2-one) (6)

White Solid; Yield: 76%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, Acetone- d_6): δ 11.43 (s, 2H, 2OH), 7.68 (d, $J_{5,6F/5',6F}$ = 8.7 Hz, 2H, H-5, H-5'), 7.54 (d, $J_{8,7/8',7'/7,8}$ = 5.7 Hz, 3H, H-7, H-8, H-8'), 7.21 (d, $J_{2'',3''/6'',5''/7',8'}$ = 8.1 Hz, 3H, H-7', H-2", H-6"), 7.14 (d, $J_{3'',2''/5'',6''}$ = 8.1 Hz, 2H, H-3", H-5"), 6.10 (bd.s, 1H, -CH-), 2.29 (s, 3H, CH₃); FAB (Neg.)-MS m/z = 461 [M-H]⁻¹; ESI-MS m/z = 463 [M+H]⁺; HRESI-MS Calcd for $C_{26}H_{17}F_{2}O_{6}$ [M+H]⁺: m/z = 463.0993, Found 463.1022.

3,3'-((2",5"-Dimethoxyphenyl)methylene)bis(6-fluoro-4-hydroxy-2H-chromen-2-one) (7)

White Solid; Yield: 72%; M.p.: 240-242 °C; ¹H-NMR (300 MHz, Acetone- d_6): δ 11.20 (s, 2H, 2OH), 7.67 (d, $J_{5,6F/5',6F}$ = 8.7 Hz, 2H, H-5, H-5'), 7.50 (d, $J_{7,8/7',8'/8,7/8',7'}$ = 5.7 Hz, 4H, H-7, H-8, H-7', H-8'), 6.91 (m, 3H, H-3", H-4", H-6"), 6.10 (s, 1H, -CH-), 3.68 (s, 3H, OCH_{3a}), 3.52 (s, 3H, CH_{3b}); ¹³C-NMR (75 MHz, DMSO- d_6): δ 165.1, 165.1, 163.7, 163.7, 159.3, 159.3, 156.2, 152.6,

148.5, 146.3, 131.4, 125.3, 120.5, 120.4, 118.3, 118.0, 117.6, 117.5, 116.4, 111.9, 109.6, 109.3, 108.9, 104.4, 56.2, 55.0, 33.0; FAB (Neg.)-MS m/z = 507 [M-H]⁻¹; ESI-MS m/z = 509 [M+H]⁺; HRESI-MS Calcd for $C_{27}H_{19}F_2O_8$ [M+H]⁺: m/z = 509.1047, Found 509.1064.

3,3'-((3"-(Benzyloxy)-4"-methoxyphenyl)methylene)bis(6-fluoro-4-hydroxy-2H-chromen-2-one) (8)

White Solid; Yield: 80%; M.p.: 200-202 °C; ¹H-NMR (300 MHz, Acetone- d_6): δ 11.41 (s, 2H, 2OH), 7.66 (d, $J_{5,6F/5',6F} = 8.4$ Hz, 2H, H-5, H-5'), 7.54 (bd.d, $J_{8,7/8',7'/7,8/7',8'} = 5.4$ Hz, 4H, H-7, H-8, H-7', H-8'), 7.27 (m, 5H, Ar-H), 6.96 (bd.s, 1H, H-2"), 6.93 (d, $J_{5'',6''} = 7.5$ Hz, 1H, H-5''), 6.87 (d, $J_{6'',5''} = 8.0$ Hz, 1H, H-6''), 6.05 (s, 1H, -CH-), 4.97 (s, 2H, -CH₂-), 3.80 (s, 3H, OCH₃); ¹³C-NMR (100 MHz, DMSO- d_6): δ 162.7, 162.7, 161.6, 161.6, 159.4, 159.4, 149.5, 148.2, 148.2, 146.9, 136.6, 135.4, 128.7, 128.7, 127.5, 127.2, 127.2, 125.3, 125.3, 122.0, 119.2, 119.2, 115.0, 115.0, 114.2, 112.4, 111.5, 111.5, 100.3, 100.3, 71.4, 56.2, 28.4; FAB (Neg.)-MS m/z = 582 [M-H]-¹; ESI-MS m/z = 585 [M+H]+; HRESI-MS Calcd for $C_{33}H_{23}F_2O_8$ [M+H]+: m/z = 585.1360, Found 585.1332.

3,3'-((4"-(Trifluoromethyl)phenyl)methylene)bis(6-fluoro-4-hydroxy-2H-chromen-2-one) (9)

White Solid; Yield: 84%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.53 (d, $J_{5,6F/5',6F}$ = 8.4 Hz, 2H, H-5, H-5'), 7.49 (dd, $J_{3'',5''/5'',3''}$ = 3.0 Hz, $J_{3'',2''/5'',6''}$ = 6.0 Hz, 2H, H-3", H-5"), 7.37 (m, 4H, H-8, H-8', H-2", H-6"), 7.29 (d, $J_{7,8/7',8'}$ = 8.4 Hz, 2H, H-7, H-7'), 6.29 (bd.s, 1H, -CH-); ¹³C-NMR (100 MHz, DMSO- d_6): δ 162.8, 162.8, 161.4, 161.4, 159.6, 159.6, 148.0, 148.0, 147.6, 128.9, 128.9, 128.4, 125.3, 125.3, 125.1, 125.1, 124.3, 119.4, 119.4, 115.5, 115.5, 111.6, 111.6, 100.9, 100.9, 28.4; FAB (Neg.)-MS m/z = 515 [M-H]-¹; ESI-MS m/z = 517 [M+H]+; HRESI-MS Calcd for $C_{26}H_{14}F_5O_6$ [M+H]+: m/z = 517.0710, Found 517.0755.

3,3'-((3"-(Trifluoromethyl)phenyl)methylene)*bis*(6-fluoro-4-hydroxy-2*H*-chromen-2-one) (10)

White Solid; Yield: 77%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, Acetone- d_6): δ 11.52 (bd.s, 2H, 2OH), 7.69 (m, 4H, H-5, H-8, H-5', H-8'), 7.55 (m, 6H, H-7, H-7', H-2", H-4", H-5", H-6"), 6.30 (s, 1H, -CH-); ¹³C-NMR (100 MHz, DMSO- d_6): δ 162.8, 162.8, 161.4, 161.4, 159.6, 159.6, 148.0, 148.0, 142.7, 131.2, 130.7, 128.7, 127.3, 125.3, 125.3, 124.5, 122.2, 119.4, 119.4, 115.5,

115.5, 111.6, 111.6, 100.9, 100.9, 28.5; FAB (Neg.)-MS m/z = 515 [M-H]⁻¹; ESI-MS m/z = 517 [M+H]⁺; HRESI-MS Calcd for $C_{26}H_{14}F_5O_6$ [M+H]⁺: m/z = 517.0710, Found 517.0755.

3,3'-((5''-Fluoro-2''-hydroxyphenyl)methylene) bis(6-fluoro-4-hydroxy-2H-chromen-2-one) (11)

White Solid; Yield: 73%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.88 (m, 1H, H-5), 7.75 (m, 1H, H-5'), 7.61 (m, 2H, H-8, H-8'), 7.46 (m, 3H, H-7, H-7', H-6"), 7.19 (m, 1H, H-4"), 6.98 (m, 1H, H-3"), 5.69 (s, 1H, -CH-); ¹³C-NMR (100 MHz, DMSO- d_6): δ 162.2, 161.9, 160.5, 160.1, 159.5, 159.0, 155.2, 151.6, 149.5, 148.8, 125.6, 125.1, 122.5, 119.9, 119.5, 117.6, 117.2, 117.0, 114.8, 114.7, 109.7, 109.4, 101.6, 101.4, 28.5; FAB (Neg.)-MS m/z = 462 [M-H-HF]-1; ESI-MS m/z = 465 [M+H-H₂O]+; HRESI-MS Calcd for $C_{25}H_{12}F_3O_6$ [M+H-H₂O]+: m/z = 465.0585, Found 465.0549.

3,3'-((5"-Chloro-2"-hydroxyphenyl)methylene)*bis*(6-fluoro-4-hydroxy-2*H*-chromen-2-one) (12)

White Solid; Yield: 79%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, Acetone- d_6): δ 7.84 (m, 2H, H-5, H-5'), 7.50 (m, 7H, H-7, H-8, H-7', H-8', H-3", H-4", H-6"), 5.79 (s, 1H, -CH-); ¹³C-NMR (100 MHz, DMSO- d_6): δ 162.5, 162.2, 160.4, 160.0, 159.6, 159.3, 151.4, 149.5, 148.9, 126.7, 125.7, 125.2, 122.6, 119.8, 119.4, 117.7, 117.4, 117.1, 114.7, 114.3, 109.8, 109.3, 101.7, 101.5, 28.6; FAB (Neg.)-MS m/z = 497 [M-H]⁻¹; 499 [M+2-H]⁻¹; ESI-MS m/z = 499 [M+H]⁺; HRESI-MS Calcd for $C_{25}H_{13}ClF_2O_7$ [M+H]⁺: m/z = 499.0318, Found 499.0336.

3,3'-((2",4"-Dihydroxyphenyl)methylene)bis(6-fluoro-4-hydroxy-2H-chromen-2-one) (13)

White Solid; Yield: 81%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, Acetone- d_6): δ 7.84 (dd, $J_{5,7}$ = 2.1 Hz, $J_{5,6F}$ = 8.7 Hz, 1H, H-5), 7.76 (d, $J_{5',7'}$ = 2.4 Hz, $J_{5',6F}$ = 8.7 Hz, 1H, H-5'), 7.52 (m, 4H, H-7, H-8, H-7', H-8'), 7.10 (d, $J_{6'',5''}$ = 8.4 Hz, 1H, H-6"), 6.82 (bd.s, 1H, H-3"), 6.65 (m, 1H, H-5"), 5.62 (bd.s, 1H, -CH-); ¹³C-NMR (75 MHz, DMSO- d_6): δ 160.2, 159.9, 159.5, 157.3, 156.7, 156.3, 148.5, 148.3, 129.1, 119.9, 119.6, 119.5, 119.2, 118.7, 118.6, 118.4, 118.3, 115.0, 114.9, 113.1, 111.9, 108.6, 108.3, 102.9, 28.2; FAB (Neg.)-MS m/z = 461 [M-H-H₂O]-¹; ESI-MS m/z = 463 [M+H-H₂O]+; HRESI-MS Calcd for $C_{25}H_{13}F_{2}O_{7}$ [M+H-H₂O]+: m/z = 463.0629, Found 463.0616.

3,3'-((3"-Bromo-4"-methoxyphenyl)methylene)*bis*(6-fluoro-4-hydroxy-2*H*-chromen-2-one) (14)

White Solid; Yield: 85%; M.p.: 210-212 °C; ¹H-NMR (300 MHz, Acetone- d_6): δ 11.47 (s, 2H, 2OH), 7.69 (d, $J_{5,6F/5',6F} = 8.7$ Hz, 2H, H-5, H-5'), 7.54 (d, $J_{7,8/8,7/7',8'/8',7'} = 6.6$ Hz, 4H, H-7, H-8, H-7', H-8'), 7.52 (bd.s, 1H, H-2"), 7.32 (bd.d, $J_{6",5"} = 8.1$ Hz, 1H, H-6"), 7.02 (d, $J_{5",6"} = 8.4$ Hz, 1H, H-5"), 6.14 (s, 1H, -CH-), 3.87 (s, 3H, OCH₃); ¹³C-NMR (75 MHz, DMSO- d_6): δ 166.5, 166.4, 164.1, 159.4, 156.2, 153.1, 148.7, 135.4, 130.8, 127.2, 120.8, 120.7, 118.6, 118.2, 117.7, 117.6, 112.3, 110.0, 109.4, 109.1, 103.7, 56.1, 35.2; FAB (Neg.)-MS m/z = 555 [M-H]-¹, 557 [M+2-H]-¹; ESI-MS m/z = 557 [M+H]+; HRESI-MS Calcd for $C_{26}H_{16}BrF_{2}O_{7}$ [M+H]+: m/z = 557.0047, Found 557.0046.

3,3'-((4"-Bromo-2",5"-dimethoxyphenyl)methylene)*bis*(6-fluoro-4-hydroxy-2*H*-chromen-2-one) (15)

White Solid; Yield: 83%; M.p.: 205-207 °C; ¹H-NMR (300 MHz, Acetone- d_6): δ 11.30 (s, 2H, 2OH), 7.67 (d, $J_{8,7/8',7'}$ = 8.4 Hz, 2H, H-8, H-8'), 7.49 (d, $J_{7,8/8,7/7',8'/8',7'}$ = 5.4 Hz, 4H, H-7, H-8, H-7', H-8'), 7.14 (s, 1H, H-3"), 7.10 (s, 1H, H-6"), 6.13 (s, 1H, -CH-), 3.72 (s, 3H, OCH_{3b}), 3.57 (s, 3H, OCH_{3a}); ¹³C-NMR (75 MHz, DMSO- d_6): δ 165.7, 165.6, 163.6, 163.5, 159.3, 159.2, 151.8, 149.6, 148.8, 148.5, 131.3, 125.3, 125.2, 120.8, 120.7, 118.3, 117.9, 117.6, 117.4, 116.0, 114.4, 109.3, 109.0, 107.5, 104.0, 56.5, 56.5, 33.0; FAB (Neg.)-MS m/z = 586 [M-H]-¹, 588 [M+2-H]-¹; ESI-MS m/z = 587 [M+H]+, 589 [M+2+H]+; HRESI-MS Calcd for $C_{27}H_{18}BrF_{2}O_{8}$ [M+H]+: m/z = 587.0153, Found 587.0111.

3,3'-((2",3"-Dihydroxyphenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (16)

White Solid; Yield: 82%; M.p.: 110-112 °C; ¹H-NMR (300 MHz, Acetone- d_6): δ 8.64 (bd.s, 1H, OH), 8.36 (d, $J_{5,6}$ = 8.1 Hz, 1H, H-5), 8.06 (d, $J_{5',6'}$ = 8.1 Hz, 1H, H-5'), 7.72 (t, $J_{7,6}$ = $J_{7,8}$ = 8.4 Hz, 1H, H-7), 7.61 (t, $J_{7',6'}$ = $J_{7',8'}$ = 7.2 Hz, 1H, H-7'), 7.49 (t, $J_{6,5}$ = $J_{6,7}$ = 7.2 Hz, 1H, H-6), 7.39 (d, $J_{8,7}$ = 8.7 Hz, 1H, H-8), 7.39 (t, $J_{6',5'}$ = $J_{6',7'}$ = 8.7 Hz, 1H, H-6'), 7.24 (d, $J_{8',7'}$ = 8.1 Hz, 1H, H-8'), 6.98 (t, $J_{5'',4''}$ = $J_{5'',6''}$ = 7.8 Hz, 1H, H-5''), 6.88 (d, $J_{6'',5''}$ = 6.6 Hz, 1H, H-6''), 6.76 (d, $J_{4'',5''}$ = 7.5 Hz, 1H, H-4''), 5.67 (s, 1H, -CH-); 13 C-NMR (75 MHz, DMSO- d_6): δ 160.5, 152.1, 151.9, 151.9, 145.0, 145.0, 132.3, 132.3, 132.0, 132.0, 125.0, 124.3, 124.3, 123.9, 123.9, 123.2, 123.2, 118.2, 116.3, 116.1, 116.1, 115.4, 114.0, 28.7; FAB (Neg.)-MS m/z = 425 [M-H-H₂O]⁻¹; ESI-MS

 $m/z = 427 \text{ [M+H-H₂O]}^+$; HRESI-MS Calcd for C₂₅H₁₆O₈ [M+H-H₂O]⁺: m/z = 427.0845, Found 427.0868.

3,3'-((2",3",4"-Trihydroxyphenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (17)

White Solid; Yield: 77%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, Acetone- d_6): δ 8.36 (d, $J_{5,6}$ = 8.1 Hz, 1H, H-5), 8.36 (bd.s, 1H, OH), 8.08 (bd.s, 1H, OH), 8.05 (d, $J_{5',6'}$ = 6.6 Hz, 1H, H-5'), 7.72 (t, $J_{7,6} = J_{7,8} = 8.7$ Hz, 1H, H-7), 7.60 (t, $J_{7',6'} = J_{7',8'} = 7.2$ Hz, 1H, H-7'), 7.49 (t, $J_{6,5} = J_{6,7} = 7.2$ Hz, 1H, H-6), 7.40 (d, $J_{6'',5''} = 8.7$ Hz, 1H, H-6"), 7.38 (t, $J_{6',5'} = J_{6',7'} = 8.1$ Hz, 1H, H-6'), 7.24 (d, $J_{5'',6''} = 8.1$ Hz, 1H, H-5"), 6.66 (m, 2H, H-8, H-8'), 5.60 (s, 1H, -CH-); ¹³C-NMR (100 MHz, DMSO- d_6): δ 160.5, 160.5, 156.1, 152.0, 152.0, 151.8, 151.8, 145.5, 145.5, 138.7, 132.9, 132.9, 132.2, 131.9, 124.3, 124.3, 123.8, 123.4, 117.3, 116.2, 116.1, 114.1, 112.2, 28.3; FAB (Neg.)-MS m/z = 441 [M-H-H₂O]⁻¹; ESI-MS m/z = 443 [M+H-H₂O]⁺; HRESI-MS Calcd for $C_{25}H_{15}O_8$ [M+H-H₂O]⁺: m/z = 443.0766, Found 443.0744.

3,3'-((2",3"-Dimethylphenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (18)

White Solid; Yield: 73%; M.p.: 203-205 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.81 (d, $J_{5,6/5',6'}$ = 7.5 Hz, 2H, H-5, H-5'), 7.49 (t, $J_{7,6} = J_{7,8} = J_{7',6'} = J_{7',8'} = 6.9$ Hz, 2H, H-7, H-7'), 7.24 (d, $J_{8,7/8',7'}$ = 8.1 Hz, 2H, H-8, H-8'), 7.24 (t, $J_{6,5} = J_{6,7} = J_{6',5'} = J_{6',7'} = 8.1$ Hz, 2H, H-6, H-6'), 7.14 (d, $J_{4'',5''} = 8.4$ Hz, 1H, H-4"), 6.80 (m, 2H, H-5", H-6"), 6.02 (s, 1H, -CH-), 2.16 (s, 3H, CH₃), 2.01 (s, 1H, CH₃); ¹³C-NMR (100 MHz, DMSO- d_6): δ 162.2, 162.2, 161.4, 161.4, 152.6, 152.6, 136.8, 135.7, 135.4, 128.2, 128.2, 127.5, 125.4, 125.3, 125.3, 123.4, 123.4, 121.4, 116.6, 116.6, 112.2, 112.2, 100.0, 100.0, 28.5, 19.3, 17.2; FAB (Neg.)-MS m/z = 439 [M-H]⁻¹; ESI-MS m/z = 441 [M+H]⁺; HRESI-MS Calcd for $C_{27}H_{21}O_6$ [M+H]⁺: m/z = 441.1338, Found 441.1328.

3,3'-((5"-Chloro-2"-hydroxyphenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (19)

White Solid; Yield: 78%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, DMSO- d_6): δ 8.10 (d, $J_{5,6}$ = 7.5 Hz, 1H, H-5), 8.04 (bd.d, 1H, H-5'), 7.73 (t, $J_{7,6} = J_{7,8} = 8.7$ Hz, 1H, H-7), 7.62 (t, $J_{7',6'} = J_{7',8'} = 8.4$ Hz, 1H, H-7'), 7.51 (m, 6H, H-6, H-8, H-6', H-8', H-3", H-4"), 7.18 (s, 1H, H-6"), 5.70 (bd.s, 1H, -CH-); FAB (Neg.)-MS m/z = 461 [M-H]-1, 463 [M+2-H]-1; ESI-MS m/z = 445 [M+H-H₂O]+; HRESI-MS Calcd for $C_{25}H_{14}ClO_6$ [M+H-H₂O]+: m/z = 445.0478, Found 445.0453.

3,3'-((2",4"-Dihydroxyphenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (20)

White Solid; Yield: 82%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, DMSO- d_6): δ 9.68 (s, 1H, OH), 8.07 (d, $J_{5,6} = 6.9$ Hz, 1H, H-5), 7.95 (bd.s, 1H, H-5'), 7.69 (t, $J_{7,6} = J_{7,8} = 7.2$ Hz, 1H, H-7), 7.52 (t, $J_{7',6'} = J_{7',8'} = 7.2$ Hz, 1H, H-7'), 7.48 (t, $J_{6,5} = J_{6,7} = 8.1$ Hz, 1H, H-6), 7.43 (d, $J_{8,7} = 7.5$ Hz, 1H, H-8), 7.25 (m, 2H, H-6', H-8'), 7.00 (d, $J_{6",5"} = 8.4$ Hz, 1H, H-6"), 6.67 (d, $J_{3",5"} = 2.1$ Hz, 1H, H-3"), 6.55 (dd, $J_{5",3"} = 2.1$ Hz, $J_{5",6"} = 6.0$ Hz, 1H, H-5"), 5.57 (s, 1H, -CH-); FAB (Neg.)-MS m/z = 425 [M-H-H₂O]⁻¹; ESI-MS m/z = 427 [M+H-H₂O]⁺; HRESI-MS Calcd for C₂₅H₁₅O₇ [M+H-H₂O]⁺: m/z = 427.0817, Found 427.0816.

3,3'-((3",5"-Dibromo-4-hydroxyphenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (21)

White Solid; Yield: 87%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.82 (d, $J_{5,6/5',6'}$ = 6.3 Hz, 2H, H-5, H-5'), 7.53 (t, $J_{7,6} = J_{7,8} = J_{7',6'} = J_{7',8'} = 6.9$ Hz, 2H, H-7, H-7'), 7.27 (d, $J_{8,7/8',7'}$ = 8.1 Hz, 2H, H-8, H-8'), 7.26 (t, $J_{6,5} = J_{6,7} = J_{6',5'} = J_{6',7'} = 7.5$ Hz, 2H, H-6, H-6'), 7.12 (bd.s, 2H, H-2", H-6"), 6.16 (s, 1H, -CH-); ¹³C-NMR (75 MHz, DMSO- d_6): δ 167.2, 167.2, 164.2, 164.2, 152.4, 152.4, 148.1, 148.1, 136.6, 131.3, 131.3, 130.2, 130.2, 124.0, 124.0, 123.1, 123.1, 119.3, 119.3, 115.6, 115.6, 111.7, 102.9, 102.9, 35.0; FAB (Neg.)-MS m/z = 584 [M-H]⁻¹, 586 [M+2-H]⁻¹; ESI-MS m/z = 585 [M+H]⁺, 587 [M+2+H]⁺, 589 [M+4+H]⁺; HRESI-MS Calcd for C₂₅H₁₅Br₂O₇ [M+H]⁺: m/z = 584.9184, Found 584.9208.

3,3'-((3",5"-Dichlorophenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (22)

White Solid; Yield: 82%; M.p.: 232-234 °C; ¹H-NMR (300MHz, Acetone- d_6): δ 8.14 (d, $J_{5,6}$ = 6.3 Hz, 1H, H-5), 8.08 (d, $J_{5',6'}$ = 6.6 Hz, 1H, H-5'), 7.76 (t, $J_{7,6}$ = $J_{7,8}$ = 8.4 Hz, 1H, H-7), 7.62 (t, $J_{7',6'}$ = $J_{7',8'}$ = 7.5 Hz, 1H, H-7'), 7.55 (t, $J_{6,5}$ = $J_{6,7}$ = 7.5 Hz, 1H, H-6), 7.52 (d, $J_{2'',6''/6'',2''}$ = 2.4 Hz, 1H, H-2", H-6''), 7.43 (t, $J_{6',5'}$ = $J_{6',7'}$ = 8.4 Hz, 1H, H-6'), 7.38 (d, $J_{8,7}$ = 7.2 Hz, 1H, H-8), 7.32 (d, $J_{4'',2''/4'',6''}$ = 2.4 Hz, 1H, H-4"), 7.27 (d, $J_{8',7'}$ = 8.1 Hz, 1H, H-8'), 5.74 (s, 1H, -CH-); ¹³C-NMR (100 MHz, DMSO- J_{6}): δ 162.7, 162.7, 161.6, 161.6, 152.4, 152.4, 145.2, 135.7, 135.7, 128.4, 128.4, 126.8, 126.8, 126.3, 125.3, 125.3, 123.4, 123.4, 116.5, 116.5, 116.3, 116.3, 100.1, 100.1, 28.4; FAB (Neg.)-MS J_{6} = 476 [M-H]-¹, 478 [M+2-H]-¹; ESI-MS J_{6} = 479 [M+H]+, 481 [M+2+H]+; HRESI-MS Calcd for J_{6} C₂₅H₁₅Cl₂O₆ [M+H]+: J_{6} = 479.0089, Found 479.0105.

3,3'-((2",5"-Dimethoxyphenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (23)

White Solid; Yield: 74%; M.p.: 181-183 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.87 (dd, $J_{5,7/5',7'}$ = 1.2 Hz, $J_{5,6/5',6'}$ = 6.6 Hz, 2H, H-5, H-5'), 7.56 (t, $J_{7,6} = J_{7,8} = J_{7',6'} = J_{7',8'} = 8.4$ Hz, 2H, H-7, H-7'), 7.31 (d, $J_{8,7/8',7'}$ = 7.8 Hz, 2H, H-8, H-8'), 7.30 (t, $J_{6,7} = J_{6,8} = J_{6',7'} = J_{6',8'} = 7.5$ Hz, 2H, H-6, H-6'), 6.79 (d, $J_{4'',3''}$ = 8.4 Hz, 1H, H-4''), 6.71 (s, 1H, H-6''), 6.71 (d, $J_{3'',4''}$ = 8.7 Hz, 1H, H-3''), 6.19 (s, 1H, -CH-), 3.60 (s, 3H, OCH_{3a}), 3.48 (s, 3H, OCH_{3b}); ¹³C-NMR (75 MHz, DMSO- d_6): δ 165.1, 163.9, 152.7, 152.1, 151.6, 131.1, 131.1, 123.7, 123.7, 123.2, 123.2, 118.5, 116.2, 115.6, 115.6, 111.8, 109.6, 104.2, 56.1, 55.0, 32.9; FAB (Neg.)-MS m/z = 470 [M-H]-¹; ESI-MS m/z = 473 [M+H]+; HRESI-MS Calcd for $C_{27}H_{20}O_8$ [M+H]+: m/z = 473.1236, Found 473.1198.

3,3'-((2",3"-Dimethoxyphenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (24)

White Solid; Yield: 76%; M.p.: 190-192 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.85 (dd, $J_{5,7/5',7'}$ = 1.2 Hz, $J_{5,6/5',6'}$ = 6.9 Hz, 2H, H-5, H-5'),7.53 (t, $J_{7,6} = J_{7,8} = J_{7',6'} = J_{7',8'} = 8.4$ Hz, 2H, H-7, H-7'), 7.28 (d, $J_{8,7/8',7'}$ = 8.4 Hz, 2H, H-8, H-8'), 7.27 (t, $J_{6,7} = J_{6,8} = J_{6',7'} = J_{6',8'} = 7.5$ Hz, 2H, H-6, H-6'), 6.89 (m, 3H, H-4", H-5", H-6"), 6.29 (s, 1H, -CH-), 3.71 (s, 3H, OCH_{3b}), 3.44 (s, 3H, OCH_{3a}); 13 C-NMR (75 MHz, DMSO- d_6): δ 165.0, 164.0, 152.2, 152.0, 146.5, 135.1, 131.2, 131.2, 123.7, 123.7, 123.3, 123.3, 122.5, 120.8, 118.5, 115.7, 115.7, 110.7, 104.6, 59.2, 55.4, 32.7; FAB (Neg.)-MS m/z = 471 [M-H⁺]-¹. FAB (Neg.)-MS m/z = 471 [M-H]-¹; ESI-MS m/z = 473 [M+H]⁺; HRESI-MS Calcd for C_{27} H₂₁O₈ [M+H]⁺: m/z = 473.1236, Found 473.1207.

3,3'-((4"-Ethoxy-3"-methoxyphenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (25)

White Solid; Yield: 78%; M.p.: 195-197 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.81 (dd, $J_{5,7/5',7'}$ = 1.5 Hz, $J_{5,6/5',6'}$ = 6.3 Hz, 2H, H-5, H-5'), 7.50 (t, $J_{7,6} = J_{7,8} = J_{7',6'} = J_{7',8'}$ = 6.9 Hz, 2H, H-7, H-7'), 7.25 (d, $J_{8,7/8',7'}$ = 8.1 Hz, 2H, H-8, H-8'), 7.23 (t, $J_{6,7} = J_{6,8} = J_{6',7'} = J_{6',8'} = 7.2$ Hz, 2H, H-6, H-6'), 6.73 (d, $J_{5'',6''}$ = 8.7 Hz, 1H, H-5''), 6.64 (s, 1H, H-2''), 6.60 (d, $J_{6'',5''}$ = 8.7 Hz, 1H, H-6''), 6.18 (s, 1H, -CH-), 3.92 (q, 2H, OCH₂-), 3.50 (s, 3H, OCH₃), 1.29 (t, 3H, CH₃); FAB (Neg.)-MS m/z = 485 [M-H]-1; ESI-MS m/z = 487 [M+H]+; HRESI-MS Calcd for $C_{28}H_{23}O_8$ [M+H]+: m/z = 487.1392, Found 487.1385.

3,3'-((2",4"-Dimethoxyphenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (26)

White Solid; Yield: 72%; M.p.: 197-199 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.82 (d, $J_{5,6/5',6'}$ = 6.9 Hz, 2H, H-5, H-5'), 7.50 (t, $J_{7,6} = J_{7,8} = J_{7',6'} = J_{7',8'} = 8.4$ Hz, 2H, H-7, H-7'), 7.25 (d, $J_{8,7/8',7'} = 8.1$ Hz, 2H, H-8, H-8'),7.22 (t, $J_{6,7} = J_{6,8} = J_{6',7'} = J_{6',8'} = 7.8$ Hz, 2H, H-6, H-6'), 7.06 (d, $J_{6'',5''} = 8.1$ Hz, 1H, H-6"), 6.39 (m, 2H, H-3", H-5"), 6.12 (s, 1H, -CH-), 3.68 (s, 3H, OCH_{3b}), 1.29 (t, 3H, OCH_{3a}); ¹³C-NMR (100 MHz, DMSO- d_6): δ 164.4, 163.9, 158.8, 158.0, 152.0, 131.2, 131.2, 128.7, 123.6, 123.3, 123.3, 120.9, 118.2, 115.7, 115.7, 104.6, 103.8, 98.5, 55.5, 55.0, 32.2; FAB (Neg.)-MS m/z = 470 [M-H]-¹; ESI-MS m/z = 473 [M+H]+; HRESI-MS Calcd for $C_{27}H_{21}O_8$ [M+H]+: m/z = 473.1236, Found 473.1230.

3,3'-((4"-Bromo-2"-fluorophenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (27)

White Solid; Yield: 83%; M.p.: 245-247 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.81 (d, $J_{5,6/5',6'}$ = 6.6 Hz, 2H, H-5, H-5'), 7.52 (t, $J_{7,6} = J_{7,8} = J_{7',6'} = J_{7',8'} = 8.4$ Hz, 2H, H-7, H-7'), 7.27 (m, 7H, H-6, H-8, H-6', H-8', H-5", H-6"), 6.24 (s, 1H, -CH-); ¹³C-NMR (100 MHz, DMSO- d_6): δ 162.9, 162.5, 162.2, 160.4, 160.0, 151.4, 150.9, 132.6, 128.4, 127.2, 125.7, 125.2, 121.5, 120.5, 119.8, 119.4, 117.7, 117.4, 114.7, 114.3, 109.8, 109.3, 101.7, 101.5, 28.6; FAB (Neg.)-MS m/z = 507 [M-H]⁻¹, 509 [M+2-H]⁻¹; ESI-MS m/z = 509 [M+H]⁺, 511 [M+2+H]⁺; HRESI-MS Calcd for $C_{25}H_{15}BrFO_6$ [M+H]⁺: m/z = 509.0036, Found 509.0032.

3,3'-((2"-Bromo-4",5"-dimethoxyphenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (28)

White Solid; Yield: 85%; M.p.: 200-202 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.83 (d, $J_{5,6/5',6'}$ = 6.6 Hz, 2H, H-5, H-5'), 7.26 (t, $J_{7,6} = J_{7,8} = J_{7',6'} = J_{7',8'} = 7.2$ Hz, 2H, H-7, H-7'), 7.26 (d, $J_{8,7/8',7'} = 8.7$ Hz, 2H, H-8, H-8'), 7.25 (t, $J_{6,7} = J_{6,8} = J_{6',7'} = J_{6',8'} = 8.1$ Hz, 2H, H-6, H-6'), 7.01 (s, 1H, H-3"), 6.97 (s, 1H, H-6"), 5.97 (s, 1H, -CH-), 3.70 (s, 3H, OCH_{3a}), 1.29 (t, 3H, OCH_{3b}); ¹³C-NMR (100 MHz, DMSO- d_6): δ 190.1, 165.3, 163.5, 152.2, 147.6, 147.1, 133.0, 131.2, 131.2, 123.7, 123.7, 123.3, 123.3, 118.4, 116.0, 115.7, 115.7, 114.4, 113.4, 110.5, 104.1, 55.7, 55.6, 37.9; FAB (Neg.)-MS m/z = 549 [M-H]-¹, 551 [M+2-H]-¹; ESI-MS m/z = 551 [M+H]+, 553 [M+2+H]+; HRESI-MS Calcd for $C_{27}H_{20}BrO_8$ [M+H]+: m/z = 551.0341, Found 551.0327.

3,3'-((3"-(Benzyloxy)-4"-methoxyphenyl)methylene)*bis*(4-hydroxy-2*H*-chromen-2-one) (29)

White Solid; Yield: 81%; M.p.: 210-212 °C; ¹H-NMR (300 MHz, Acetone- d_6): δ 11.44 (s, 1H, OH), 8.01 (d, $J_{5,6/5',6'} = 6.9$ Hz, 2H, H-5, H-5'), 7.74 (t, $J_{7,6} = J_{7,8} = J_{7',6'} = J_{7',8'} = 7.2$ Hz, 2H, H-7, H-7'), 7.48 (t, $J_{6,7} = J_{6,8} = J_{6',7'} = J_{6',8'} = 8.1$ Hz, 2H, H-6, H-6'), 7.48 (d, $J_{8,7/8',7'} = 7.8$ Hz, 2H, H-8, H-8'), 7.26 (m, 2H, H-2"', H-6"'), 7.20 (m, 3H, H-3"', H-4"', H-6"'), 6.97 (bd.s, 1H, H-2"'), 6.93 (d, $J_{6'',5''} = 7.2$ Hz, 1H, H-6"), 6.87 (d, $J_{5'',6''} = 7.2$ Hz, 1H, H-5"), 6.04 (s, 1H, -CH-), 4.96 (s, 2H, -CH₂-), 3.80 (s, 3H, OCH₃); ¹³C-NMR (100 MHz, DMSO- d_6): δ 162.9, 162.9, 161.7, 161.7, 152.6, 152.6, 149.8, 146.6, 136.5, 135.7, 128.7, 128.7, 128.4, 128.4, 127.4, 127.2, 127.2, 125.5, 125.5, 123.2, 123.2, 122.4, 116.7, 116.7, 116.2, 116.2, 114.2, 112.4, 100.4, 100.4, 71.2, 28.7, 56.2; FAB (Neg.)-MS m/z = 546 [M-H]-¹; ESI-MS m/z = 549 [M+H]+; HRESI-MS Calcd for $C_{33}H_{25}O_8$ [M+H]+: m/z = 549.1549, Found 549.1550.

3,3'-((4"-Hydroxy-3"-methoxyphenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (30)

White Solid; Yield: 84%; M.p.: 255-257 °C; ¹H-NMR (300 MHz, Acetone- d_6): δ 11.45 (s, 2H, 2OH), 8.02 (d, $J_{5,6/5',6'}$ = 7.2 Hz, 2H, H-5, H-5'), 7.75 (t, $J_{7,6} = J_{7,8} = J_{7',6'} = J_{7',8'}$ = 7.5 Hz, 2H, H-7, H-7'), 7.51 (m, 4H, H-6, H-8, H-6', H-8'), 6.95 (s, 1H, H-2"), 6.77 (s, 2H, H-5", H-6"), 6.08 (s, 1H, -CH-), 3.67 (s, 3H, OCH₃); FAB (Neg.)-MS m/z = 457 [M-H]⁻¹; ESI-MS m/z = 459 [M+H]⁺; HRESI-MS Calcd for $C_{26}H_{19}O_{8}$ [M+H]⁺: m/z = 459.1079, Found 459.1053.

3,3'-((4"-(Methylthio)phenyl)methylene)bis(4-hydroxy-2H-chromen-2-one) (31)

White Solid; Yield: 77%; M.p.: 260-262 °C; ¹H-NMR (300 MHz, Acetone- d_6): δ 11.50 (s, 2H, 2OH), 8.03 (d, $J_{5,6/5',6'}$ = 8.4 Hz, 2H, H-5, H-5'), 7.76 (t, $J_{7,6} = J_{7,8} = J_{7',6'} = J_{7',8'} = 6.9$ Hz, 2H, H-7, H-7'), 7.49 (m, 4H, H-6, H-8, H-6', H-8'), 7.29 (d, $J_{2",3"/6",5"} = 8.1$ Hz, 2H, H-2", H-6"), 7.23 (d, $J_{3",2"/5",6"} = 8.7$ Hz, 2H, H-3", H-5"), 6.10 (s, 1H, -CH-), 2.46 (s, 3H, SCH₃); FAB (Neg.)-MS m/z = 457 [M-H]⁻¹; ESI-MS m/z = 459 [M+H]⁺; HRESI-MS Calcd for $C_{26}H_{19}O_6S$ [M+H]⁺: m/z = 459.0902, Found 459.0922.

3,3'-((3"-Bromo-4"-methoxyphenyl)methylene)*bis*(6-chloro-4-hydroxy-2*H*-chromen-2-one) (32)

White Solid; Yield: 85%; M.p.: 204-206 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.73 (d, $J_{5,7/5',7'}$ = 2.4 Hz, 2H, H-5, H-5'), 7.55 (dd, $J_{7,5/7',5'}$ = 2.4 Hz, $J_{7,8/7',8'}$ = 9.0 Hz, 2H, H-7, H-7'), 7.32 (d, $J_{8,7/8',7'}$ = 8.7 Hz, 2H, H-8, H-8'), 7.15 (bd.s, 1H, H-2"), 7.04 (bd.d, $J_{6",5"}$ = 8.4 Hz, 1H, H-6"), 6.92 (d, $J_{5",6"}$ = 8.7 Hz, 1H, H-5"), 6.15 (s, 1H, -CH-), 3.76 (s, 3H, OCH₃); ¹³C-NMR (100 MHz, DMSO- d_6): δ 166.3, 163.9, 153.1, 151.0, 135.3, 130.8, 130.8, 130.7, 127.2, 127.2, 123.2, 123.2, 121.1, 117.7, 117.7, 112.3, 110.0, 103.8, 56.0, 32.2; FAB (Neg.)-MS m/z = 587 [M-H]-¹; 589 [M+2-H]-¹; 591 [M+4-H]-¹; ESI-MS m/z = 599 [M+H]+; HRESI-MS Calcd for $C_{26}H_{15}BrCl_2O_7$ [M+H]+: m/z = 588.9456, Found 588.9479.

3,3'-((3"-(Benzyloxy)phenyl)methylene)bis(6-chloro-4-hydroxy-2H-chromen-2-one) (33)

White Solid; Yield: 80%; M.p.: 192-194 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.72 (d, $J_{5,7/5',7'}$ = 2.4 Hz, 2H, H-5, H-5'), 7.54 (dd, $J_{7,5/7',5'}$ = 2.4 Hz, $J_{7,8/7',8'}$ = 6.3 Hz, 2H, H-7, H-7'), 7.31 (m, 4H, H-8, H-8', H-3''', H-5'''), 7.24 (m, 3H, H-2''', H-4''', H-6'''), 7.09 (t, $J_{5'',4''}$ = $J_{5'',6''}$ = 8.1 Hz, 1H, H-5''), 6.74 (d, $J_{6'',5''}$ = 8.1 Hz, 1H, H-6''), 6.64 (m, 2H, H-2'', H-4''), 6.18 (s, 1H, -CH-), 4.94 (s, 3H, OCH₂); 13 C-NMR (75 MHz, DMSO- d_6): δ 166.1, 166.1, 164.0, 164.0, 158.3, 151.0, 151.0, 143.2, 137.2, 130.8, 130.8, 128.8, 128.2, 128.2, 127.7, 127.6, 127.6, 127.2, 127.2, 123.2, 123.2, 121.1, 119.3, 119.3, 117.7, 117.7, 113.7, 110.9, 104.0, 104.0, 69.1, 36.2; FAB (Neg.)-MS m/z = 585 [M-H]-¹, 587 [M+2-H]-¹; ESI-MS m/z = 587 [M+H]+; HRESI-MS Calcd for C_{32} H₂₁Cl₂O₇ [M+H]+: m/z = 587.0664, Found 587.0666.

3,3'-((2",3",4"-Trihydroxyphenyl)methylene)*bis*(6-chloro-4-hydroxy-2*H*-chromen-2-one) (34)

White Solid; Yield: 86%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, DMSO- d_6): δ 9.26 (bd.s, 1H, OH), 9.06 (bd.s, 1H, OH), 8.56 (d, $J_{5,7}$ = 2.7 Hz, 2H, H-5), 7.95 (bd.s, 1H, H-5'), 7.72 (d, $J_{7,8}$ = 8.7 Hz, 1H, H-7), 7.58 (d, $J_{7',8'}$ = 9.0 Hz, 1H, H-7'), 7.47 (d, $J_{8,7}$ = 9.0 Hz, 1H, H-8), 7.31 (d, $J_{8',7'}$ = 8.4 Hz, 1H, H-8'), 6.55 (d, $J_{6'',5''}$ = 8.4 Hz, 1H, H-6"), 6.46 (d, $J_{5'',6''}$ = 8.4 Hz, 1H, H-5"), 5.56 (s, 1H, -CH-); ¹³C-NMR (100 MHz, DMSO- d_6): δ 166.0, 165.7, 164.1, 163.6, 151.4, 151.0, 148.6, 147.4, 133.9, 131.3, 130.9, 129.5, 129.2, 126.7, 126.4, 124.5, 122.9, 122.5, 118.9, 118.6, 116.7, 100.5, 100.2, 108.6, 30.6; FAB (Neg.)-MS m/z = 509 [M-H-H₂O]⁻¹; ESI-MS m/z = 511 [M+H-H₂O]⁺; HRESI-MS Calcd for $C_{25}H_{14}Cl_{2}O_{9}$ [M+H-H₂O]⁺: m/z = 511.0093, Found 511.0062.

3,3'-((3",5"-Dibromo-4"-hydroxyphenyl)methylene)*bis*(6-chloro-4-hydroxy-2*H*-chromen-2-one) (35)

White Solid; Yield: 82%; M.p.: > 300 °C dec.; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.74 (d, $J_{5,7/5',7'}$ = 2.4 Hz, 2H, H-5, H-5'), 7.55 (dd, $J_{7,5/7',5'}$ = 2.4 Hz, $J_{7,8/7',8'}$ = 6.3 Hz, 2H, H-7, H-7'), 7.32 (d, $J_{8,7/8',7'}$ = 8.7 Hz, 2H, H-8, H-8'), 7.13 (s, 2H, H-2", H-6"), 6.13 (s, 1H, -CH-); ¹³C-NMR (100 MHz, DMSO- d_6): δ 166.4, 166.4, 164.3, 164.3, 151.3,150.8, 150.8, 139.0, 134.5, 134.5, 131.0, 131.0, 129.5, 129.5, 126.8, 126.8, 122.9, 122.9, 118.8, 118.8, 110.3, 110.3, 100.2, 100.2, 35.2; FAB (Neg.)-MS m/z = 651 [M-H]-¹; 653 [M+2-H]-¹; ESI-MS m/z = 653 [M+H]+; HRESI-MS Calcd for $C_{25}H_{12}Br_2Cl_2O_7$ [M+H]+: m/z = 652.8405, Found 652.8436.

3,3'-((2"-Hydroxynaphthalen-1"-yl)methylene)*bis*(6-chloro-4-hydroxy-2*H*-chromen-2-one) (36)

White Solid; Yield: 86%; M.p.: 252-254 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 10.83 (s, 2H, 2OH), 9.13 (s, 1H, OH), 8.66 (d, $J_{5'',6''/8'',7''}$ = 8.4 Hz, 2H, H-5", H-8"), 8.32 (d, $J_{4'',3''}$ = 9.3 Hz, 1H, H-4"), 8.11 (d, $J_{3'',4''}$ = 7.8 Hz, 1H, H-3"), 7.79 (t, $J_{6'',5''}$ = $J_{6'',7''}$ = 7.5 Hz, 1H, H-6"), 7.67 (m, 4H, -CH-, H-5, H-5', H-7"), 7.51 (dd, $J_{7,5/7',5'}$ = 2.4 Hz, $J_{7,8/7',8'}$ = 8.7 Hz, 2H, H-7, H-7'), 6.94 (d, $J_{8,7/8',7'}$ = 8.7 Hz, 2H, H-8, H-8'); ¹³C-NMR (75 MHz, DMSO- d_6): δ 190.4, 158.1, 156.9, 154.3, 139.6, 139.6, 135.1, 135.1, 134.2, 134.2, 129.9, 129.4, 129.4, 129.2, 129.0, 129.0, 128.8, 128.8, 127.5, 126.4, 125.7, 122.9, 122.5, 122.5, 118.9, 118.9, 116.5, 112.7, 28.6; FAB (Neg.)-MS m/z = 440 [M-H-2H₂O-2Cl]-¹; ESI-MS m/z = 529 [M+H-H₂O]+; HRESI-MS Calcd for C₂₉H₁₆Cl₂O₇ [M+H-H₂O]+: m/z = 529.0245, Found 529.0268.

3,3'-((4"-Hydroxy-3"-iodo-5"-methoxyphenyl)methylene)bis(6-chloro-4-hydroxy-2H-chromen-2-one) (37)

White Solid; Yield: 80%; M.p.: 234-236 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.74 (d, $J_{5,7/5',7'}$ = 2.7 Hz, 2H, H-5, H-5'), 7.54 (dd, $J_{7,5/7',5'}$ = 2.7 Hz, $J_{7,8/7',8'}$ = 6.3 Hz, 2H, H-7, H-7'), 7.31 (d, $J_{8,7/8',7'}$ = 9.0 Hz, 2H, H-8, H-8'), 6.90 (bd.s, 1H, H-2"), 6.64 (bd.s, 1H, H-6"), 6.12 (s, 1H, -CH-), 3.57 (s, 3H, OCH₃); ¹³C-NMR (75 MHz, DMSO- d_6): δ 166.1, 166.1, 163.8, 163.8, 150.9, 150.9, 146.7, 143.9, 134.4, 130.7, 130.7, 127.8, 127.8, 127.2, 127.2, 123.1, 123.1, 121.0, 117.7, 117.7, 111.2, 104.0, 104.0, 84.1, 55.9, 35.4; FAB (Neg.)-MS m/z = 651 [M-H]-1, 653 [M+2-H]-1; ESI-MS m/z = 653 [M+H]+; HRESI-MS Calcd for $C_{26}H_{16}Cl_2IO_8$ [M+H]+: m/z = 652.9267, Found 652.9280.

3,3'-((5"-Bromo-2"-methoxyphenyl)methylene)*bis*(6-chloro-4-hydroxy-2*H*-chromen-2-one) (38)

White Solid; Yield: 81%; M.p.: 251-253 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.73 (d, $J_{5,7/5',7'}$ = 2.4 Hz, 2H, H-5, H-5'), 7.52 (dd, $J_{7,5/7',5'}$ = 2.7 Hz, $J_{7,8/7',8'}$ = 6.0 Hz, 2H, H-7, H-7'), 7.29 (d, $J_{8,7/8',7'}$ = 8.7 Hz, 2H, H-8, H-8'), 7.23 (d, $J_{3'',4''}$ = 6.9 Hz, 1H, H-3"), 7.21 (s, 1H, H-6"), 6.80 (d, $J_{4'',3''}$ = 8.4 Hz, 1H, H-4"), 6.15 (s, 1H, -CH-), 3.53 (s, 1H, OCH₃); ¹³C-NMR (100 MHz, DMSO- d_6): δ 166.2, 166.0, 163.7, 163.5, 157.7, 150.7, 150.5, 134.6, 131.2, 131.0, 129.8, 129.6, 126.7, 126.5, 129.4, 118.8, 118.6, 115.5, 113.6, 100.3, 100.1, 123.4, 122.8, 122.6, 56.2, 29.7; FAB (Neg.)-MS m/z = 587 [M-H]-1, 589 [M+2-H]-1, 591 [M+4-H]-1; ESI-MS m/z = 599 [M+H]+; HRESI-MS Calcd for $C_{26}H_{15}BrCl_2O_7$ [M+H]+: m/z = 588.9456, Found 588.9478.

3,3'-((2",5"-Dihydroxyphenyl)methylene)bis(6-chloro-4-hydroxy-2H-chromen-2-one) (39)

White Solid; Yield: 76%; M.p.: 274-276 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 9.37 (s, 1H, OH), 8.07 (d, $J_{5,7} = 2.1$ Hz, 1H, H-5), 8.00 (bd.s, 1H, H-5'), 7.74 (dd, $J_{7,5} = 2.4$ Hz, $J_{7,8} = 6.3$ Hz, 1H, H-7), 7.62 (d, $J_{7',8'} = 7.5$ Hz, 1H, H-7'), 7.49 (d, $J_{8,7} = 8.7$ Hz, 1H, H-8), 7.35 (d, $J_{8',7'} = 8.7$ Hz, 1H, H-8'), 7.21 (d, $J_{3'',4''} = 8.7$ Hz, 1H, H-3"), 6.67 (d, $J_{4'',3''} = 6.0$ Hz, 1H, H-4"), 6.58 (bd.s, 1H, H-6"), 5.62 (bd.s, 1H, -CH-); ¹³C-NMR (100 MHz, DMSO- d_6): δ 166.2, 165.9, 164.3, 164.0, 151.3, 151.0, 149.2, 148.7, 131.4, 131.1, 129.3, 129.0, 126.5, 126.2, 124.3, 122.7, 122.5, 118.7, 118.4, 117.2, 116.7, 114.6, 104.6, 104.3, 30.1; FAB (Neg.)-MS m/z = 493 [M-H-H₂O]-¹; 495 [M+2-H]-¹; ESI-MS m/z = 495 [M+H]+; HRESI-MS Calcd for $C_{25}H_{14}Cl_2O_8$ [M+H]+: m/z = 495.0038, Found 495.0065.

3,3'-((3"-Bromo-4"-hydroxyphenyl)methylene)*bis*(6-chloro-4-hydroxy-2*H*-chromen-2-one) (40)

White Solid; Yield: 75%; M.p.: 248-250 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 9.84 (bd.s, 2H, 2OH),7.73 (d, $J_{5,7/5',7'}$ = 2.4 Hz, 2H, H-5, H-5'), 7.55 (dd, $J_{7,5/7',5'}$ = 2.7 Hz, $J_{7,8/7',8'}$ = 6.0 Hz, 2H, H-7, H-7'), 7.31 (d, $J_{8,7/8',7'}$ = 8.7 Hz, 2H, H-8, H-8'), 7.06 (bd.s, 1H, H-2"), 6.88 (d, $J_{6",5"}$ = 8.7 Hz, 1H, H-6"), 6.76 (d, $J_{5",6"}$ = 8.4 Hz, 1H, H-5"), 6.11 (s, 1H, -CH-); ¹³C-NMR (100 MHz, DMSO- d_6): δ 166.1, 163.9, 163.9, 151.5, 151.5, 151.0, 133.5, 133.5, 130.8, 130.5, 127.2, 127.2, 127.1, 123.2, 123.2, 121.0, 121.0, 117.7, 117.7, 116.0, 108.7, 103.9, 103.9, 35.2; FAB (Neg.)-MS m/z = 573 [M-H]-¹, 575 [M+2-H]-¹; ESI-MS m/z = 575 [M+H]+; HRESI-MS Calcd for $C_{25}H_{13}BrCl_2O_7$ [M+H]+: m/z = 574.9300, Found 574.9326.

$3,3'-((2''-Bromo-5''-fluorophenyl)methylene) \emph{bis} (6-chloro-4-hydroxy-2\emph{H}-chromen-2-one) \end{41}$

White Solid; Yield: 79%; M.p.: 273-275 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.74 (d, $J_{5,7/5',7'}$ = 2.4 Hz, 2H, H-5, H-5'), 7.55 (dd, $J_{7,5/7',5'}$ = 2.7 Hz, $J_{7,8/7',8'}$ = 6.0 Hz, 2H, H-7, H-7'), 7.49 (m, 1H, H-3"), 7.31 (d, $J_{8,7/8',7'}$ = 8.7 Hz, 2H, H-8, H-8'), 7.11 (dd, $J_{6'',4'''}$ = 3.0 Hz, $J_{6'',F}$ = 7.8 Hz, 1H, H-6"), 6.99 (m, 1H, H-4"), 5.97 (s, 1H, -CH-); ¹³C-NMR (100 MHz, DMSO- d_6): δ 166.0, 166.0, 163.4, 163.4, 161.6, 150.9, 150.9, 142.4, 133.2, 131.4, 131.4, 129.7, 129.7, 126.8, 126.8, 122.7, 122.7, 119.3, 118.7, 118.6, 114.5, 104.2, 104.2, 31.8; FAB (Neg.)-MS m/z = 574 [M-H]-¹, 576 [M+2-H]-¹, 578 [M+4-H]-¹; ESI-MS m/z = 579 [M+H]+; HRESI-MS Calcd for $C_{25}H_{13}BrCl_2FO_6$ [M+H]+: m/z = 576.9256, Found 576.9296.

3,3'-((4"-Bromo-3",5"-dimethoxyphenyl)methylene)*bis*(6-chloro-4-hydroxy-2*H*-chromen-2-one) (42)

White Solid; Yield: 82%; M.p.: 276-278 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.74 (d, $J_{5,7/5',7'}$ = 2.4 Hz, 2H, H-5, H-5'), 7.54 (dd, $J_{7,5/7',5'}$ = 2.7 Hz, $J_{7,8/7',8'}$ = 6.3 Hz, 2H, H-7, H-7'), 7.31 (d, $J_{8,7/8',7'}$ = 8.7 Hz, 2H, H-8, H-8'), 6.45 (s, 2H, H-2", H-6"), 6.19 (s, 1H, -CH-), 3.58 (s, 6H, 2OCH₃); ¹³C-NMR (100 MHz, DMSO- d_6): δ 160.0, 160.0, 159.2, 159.2, 155.2, 155.2, 150.6, 150.6, 149.7, 132.2, 132.2, 131.8, 131.8, 129.1, 129.1, 128.7, 128.0, 118.6, 118.6, 117.7, 117.7, 112.2, 112.2, 101.5, 55.5, 55.5, 28.2; FAB (Neg.)-MS m/z = 617 [M-H]-¹, 619 [M+2-H]-¹, 621 [M+4-H]-¹; ESI-MS m/z = 619 [M+H]+; HRESI-MS Calcd for $C_{27}H_{17}BrCl_2O_8$ [M+H]+: m/z = 618.9562, Found 618.9536.

3,3'-((2"-Fluoro-4"-methoxyphenyl)methylene)bis(6-chloro-4-hydroxy-2H-chromen-2-one) (43)

White Solid; Yield: 75%; M.p.: 243-245 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 8.13 (d, $J_{5,7}$ = 2.4 Hz, 1H, H-5), 7.99 (bd.s, 1H, H-5'), 7.74 (dd, $J_{7,5}$ = 2.8 Hz, $J_{7,8}$ = 6.0 Hz, 1H, H-7), 7.60 (bd.d, $J_{7',8'}$ = 7.2 Hz, 1H, H-7'), 7.50 (d, $J_{8,7}$ = 8.8 Hz, 1H, H-8), 7.33 (d, $J_{8',7'}$ = 9.2 Hz, 1H, H-8'), 7.08 (d, $J_{6'',5''}$ = 8.4 Hz, 1H, H-6"), 7.03 (d, $J_{3'',5''}$ = 2.4 Hz, 1H, H-3"), 6.73 (dd, $J_{5'',3''}$ = 2.4 Hz, $J_{5'',6''}$ = 6.3 Hz, 1H, H-5"), 5.61 (bd.s, 1H, -CH-), 3.77 (s, 1H, OCH₃); ¹³C-NMR (75 MHz, DMSO- d_6): δ 166.0, 163.8, 163.8, 155.9, 151.0, 151.0, 143.0, 131.2, 131.2, 130.8, 130.8, 127.2, 127.2, 123.1, 123.1, 120.8, 118.3, 117.8, 117.8, 104.0, 104.0, 103.9, 96.7, 56.0, 36.7; FAB (Neg.)-MS

m/z = 527 [M-H]⁻¹, 529 [M+2-H]⁻¹; ESI-MS m/z = 529 [M+H]⁺; HRESI-MS Calcd for $C_{26}H_{15}Cl_2FO_7$ [M+H]⁺: m/z = 529.0257, Found 529.0257.

3,3'-((3"-Bromo-4"-hydroxy-5"-methoxyphenyl)methylene)bis(6-chloro-4-hydroxy-2H-chromen-2-one) (44)

White Solid; Yield: 87%; M.p.: 282-284 °C; ¹H-NMR (300 MHz, DMSO- d_6): δ 7.74 (d, $J_{5,7/5',7'}$ = 2.4 Hz, 2H, H-5, H-5'), 7.54 (dd, $J_{7,5/7',5'}$ = 2.4 Hz, $J_{7,8/7',8'}$ = 6.3 Hz, 2H, H-7, H-7'), 7.31 (d, $J_{8,7/8',7'}$ = 8.7 Hz, 2H, H-8, H-8'), 6.70 (bd.s, 1H, H-6"), 6.62 (bd.s, 1H, H-2"), 6.13 (s, 1H, -CH-), 3.58 (s, 3H, OCH₃); ¹³C-NMR (100 MHz, DMSO- d_6): δ 166.1, 166.1, 163.8, 163.8, 150.9, 150.9, 148.1, 141.4, 133.4, 130.7, 130.7, 127.2, 127.2, 123.1, 123.1, 122.1, 121.0, 121.0, 117.7, 117.7, 110.2, 108.8, 103.9, 103.9, 56.0, 35.6; FAB (Neg.)-MS m/z = 604 [M+H]⁺, 606 [M+2+H]⁺ 608 [M+4+H]⁺; HRESI-MS Calcd for C₂₆H₁₆BrCl₂O₈ [M+H]⁺: m/z = 604.9405, Found 604.9448.

Urease inhibitory assay

Urease (EC 3.5.1.5) enzyme solution (25 μ L) and 55 μ L buffers, containing 100 mM urea, were incubated at 30 °C for 15 min along with 5 μ L (1 mM concentration) test compounds in 96-well plates. Ammonia production was measured in order to determine urease activity by utilizing the indophenols method by Weatherburn [30]. Briefly, 45 μ L phenol reagent (1% w/v phenol and 0.005% w/v sodium nitroprusside) and 70 μ L alkali reagent (0.5% w/v NaOH and 0.1% active chloride NaOCl) were added to each well. By using a microplate reader (Spectra Max, Molecular Devices, USA), increase in absorbance was recorded at 630 nm after 50 min. All reactions were carried out thrice in a final volume of 200 μ L. The change in absorbance per min was processed by serving SoftMax Pro software (Molecular Devices, USA). All the assays were carried out at pH 8.2 (0.01 M K₂HPO₄.3H₂O, 1 mM EDTA and 0.01 M LiCl). Thiourea was used as the standard inhibitor of urease. Percentage inhibitions were calculated from the following formula:

In vitro BSA-MG Anti-glycation Assay

In this assay, BSA (Bovine serum albumin) of concentration (10 mg/mL) and (14 mM) methylglyoxal were prepared in phosphate buffer 0.1 M of pH 7.4, containing 3 mM sodium azide (NaN₃) which was used as antimicrobial agent. 1 mM test compounds were dissolved in the

DMSO. Assay was performed in triplicate. Reaction mixtures contained 50 μ L BSA, 50 μ L methylglyoxal, 20 μ L test compound, and 80 μ L phosphate buffer of pH 7.4. The reaction mixture was incubated for 9 days at 37 °C under aseptic conditions. Fluorescence was recorded for excitation and emission at 330, and 420 nm on a microtitre plate reader (Spectra Max M5, Molecular Devices, CA, USA) [31,32]. The percent inhibition of AGE formation was calculated by using the following formula:

% Inhibition= (1-Fluorescence of test sample/Fluorescence of the control) x 100

The IC₅₀ values were determined at different concentrations (1000-50 μ M) of test compounds with the help of EZ-FIT Enzyme Kinetics Program (Perrella Scientific Inc., Amherst, USA). Rutin was used as a standard. In protein model system (BSA-MG glycation model), it showed an IC₅₀ of 294 \pm 1.5 μ M.

3T3 Cell based (mouse Fibroblast) cytotoxicity assay

Standard MTT (3-[4,5-dimethylthiazole-2-yl]-2,5-dimethyltetrazolium bromide) colorimetric assay was used to evaluate the cytotoxicity of thirteen *bis*-coumarin derivatives with antiglycation properties in 96-well flat-bottomed microplates. 3T3 Cells (mouse fibroblasts) in Dulbecco's modified eagle's medium was cultured for this purpose, supplemented with streptomycin, penicillin, and 5% fetal bovine serum (FBS) by using a flask placed in a 5% $\rm CO_2$ incubator at 37 $\rm ^{\circ}C$. The growth of cells was harvested exponentially by using hemocytometer, the harvested cells was counted and in a particular medium was diluted. Cell cultures was prepared with a required concentration and plated onto 96-well plates. After overnight incubation, medium was removed and fresh medium was added with different concentrations of the compound. After 72 h, MTT was added to each well, and incubation was continued for 4 h. Subsequently, 100 μ L of DMSO was added to each well to solubilize formazan-MTT adduct, formed by the action of enzyme mitochondrial dehydrogenase. The extent of MTT reduction to formazan within cells was calculated by measuring the absorbance at 540 nm by using a microplate ELISA reader [33].

Statistical analysis

All experiments were conducted using (SpectraMax M5, molecular Devices, CA. USA). Different softweres were used to analyze the results i.e (MS-Excel, SoftMaxPro 4.8, and GraphPad Prism-6.0). In the end IC₅₀ values of active compounds were examined by using EZ-FIT, Enzyme kinetics software (Perrella Scientific, Inc., USA)

Acknowledgment: The authors are thankful to the Higher Education Commission (HEC), Pakistan, for providing financial support under "National Research Program for Universities", for Project No. 20-1910. One of us (Arsalan Nizamani) also acknowledges Higher Education Commission, Pakistan, for financial support through "Indigenous 5000 Scholarship Programme Batch-VII".

References

- [1] Link, K. P. The anticoagulant 3, 3-methyl-bis-4-hydroxycoumarin. Federation *Proceedings*, **1945**, *4*, 176-182.
- [2] Sengupta, P.; Sen, M.; Karuri, P. Structure of gerberinol novel dimethyldicoumarol from *Gerbera lanuginosa* benth. *Journal of Indian Chemical Society*, **1985**, *62*, 916-919.
- [3] Khan, K. M.; Rahim, F.; Wadood, A.; Kosar, N.; Taha, M.; Lalani, S.; Khan, A.; Fakhri, M. I.; Junaid, M.; Rehman, W.; Khan, M.; Perveen, S.; Sajid, M.; Choudhary, M. I. Synthesis and molecular docking studies of potent α-glucosidase inhibitors based on *bis*-coumarin skeleton. *European Journal of Medicinal Chemistry*, **2014**, *81*, 245-252.
- [4] Khan, K. M.; Iqbal, S.; Lodhi, M. A.; Maharvi, G. M.; Ullah, Z.; Choudhary, M. I.; Attaur-Rahman; Perveen, S. *Bis*-coumarin: new class of urease inhibitors; economical synthesis and activity. *Bioorganic & Medicinal Chemistry*, **2004**, *12*, 1963-1968.
- [5] Choudhary, M. I.; Fatima, N.; Khan, K. M.; Jalil, S.; Iqbal, S.; Atta-ur-Rahman, New *bis*-coumarin derivatives-cytotoxicity and enzyme inhibitory activities. *Bioorganic & Medicinal Chemistry*, **2006**, *14*, 8066-8072.
- [6] Li, S. S.; Gao, Z.; Feng, X.; Hecht, S. M. *Bis*-coumarin derivatives from *Edgeworthia* gardneri that inhibit the lyase activity of DNA polymerase-β. *Journal of Natural Products*, **2004**, *67*, 1608-1610.
- [7] Campbell, H. A.; Link, K. P. The isolation and crystallization of the hemorrhagic agent. *Journal of Biological Chemistry*, **1941**, *138*, 21-33.
- [8] Witkowska, D.; Rowinska-Zyrek, M.; Valensin, G.; Kozlovski, H. Specific poly-histidyl and poly-cysteil protein sites involved in Ni²⁺ homeostasis in *Helicobacter pylori*. Impact

- of Bi³⁺ ions on Ni²⁺ binding to proteins. Structural and thermodynamic aspects. *Coordination Chemistry Reviews*, **2012**, *256*, 133-148.
- [9] Habala, L.; Roller, A.; Matuška, M.; Valentová, J.; Rompel, A.; Devínsky, F. Complexes of *N*-hydroxyethyl-*N*-benzimidazolylmethylethylenediaminediacetic acid with copper(II) and cobalt(II): Preparation, crystal structure and urease inhibitory activity, *Inorganica Chimica Acta*, **2014**, *421*, 423–426.
- [10] Krajewska, B. Ureases I. Functional, catalytic and kinetic properties: A review. *Journal of Molecular Catalysis B: Enzymatic*, **2009**, *59*, 9-21.
- [11] Mobley, H. L. T.; Island, M. D.; Hausinger, R. P. Molecular biology of microbial ureases. *Microbiological Reviews*, **1995**, *59*, 451-480.
- [12] Mobley, H. L. T.; Hausinger, R. P. Microbial ureases: significance, regulation, and molecular characterization. *Microbiological Reviews*, **1989**, *53*, 85-108.
- [13] Stingl, K.; Altendorf, K.; Bakker, E.P. Acid survival of *Helicobacter pylori*: how does urease activity trigger cytoplasmic pH homeostasis. *Trends in Microbiology*, **2002**, *10*, 70-74.
- [14] Futagami, S.; Takahashi, H.; Norose, Y.; Nagata, K.; Kobayashi, M.; Nomura, T. Analysis of immune response to *Helicobacter pylori*; identification of the protein recognized by anti-*Helicobacter pylori* antibodies from sera of patients with gastroduodenal diseases. *The Japanese Society of Gastroenterology*, **1994**, *91*, 2202-2213.
- [15] Tanaka, T.; Kawase, M.; Tani, S. Urease inhibitory activity of simple *α,β*-unsaturated ketones. *Life Sciences*, **2003**, *73*, 2985-2990.
- [16] Xu, Y. P.; Qin, J.; Sun, S. M.; Liu, T. T.; Zhang, X. L.; Qian, S.S.; Zhu, H. L. Synthesis, crystal structures, molecular docking and urease inhibitory. *Inorganica Chimica Acta*, **2014**, *423*, 469-476.
- [17] Bremenr, J. M. Recent research on problems in the use of urea as a nitrogen fertilizer. *Fertilizer Research*, **1995**, *42*, 321-329.
- [18] Ansari, N.; Rasheed, Z. Non-enzymatic glycation of proteins: from diabetes to cancer. Biochemistry (Moscow) Supplement Series B: Biomedical Chemistry, 2009, 3, 335-342.

- [19] Vlassara, H.; Palace, M. Diabetes and advanced glycation endproducts. *Journal of Internal Medicine*, **2002**, *251*, 87-101.
- [20] Seo, S.; Karboune, S.; L' Hocine, L.; Yaylayan, V. Characterization of glycated lysozyme with galactose, galactooligosaccharides and galactan: Effect of glycation on structural and functional properties of conjugates. *LWT-Food Science and Technology*, **2013**, *53*, 44-53.
- [21] Li, W.; Zheng, H.; Bukuru, J.; De Kimpe, N. Natural medicines used in the traditional Chinese medical system for therapy of diabetes mellitus. *Journal of Ethnopharmacology*, **2004**, *92*, 1-21.
- [22] Ulrich, P.; Cerami, A. Protein glycation, diabetes, and aging. *Recent Progress in Hormone, Research*, **2001**, *56*, 1-51.
- [23] Cho, S. J.; Roman, G.; Yeboah, F.; Konishi, Y. The road to advanced glycation end products: a mechanistic perspective. *Current Medicinal Chemistry*, **2007**, *14*, 1653-1671.
- [24] Ramkissoon, J.S.; Mahomoodally, F.M.; Nessar. A.; Subratty, H.A. Natural inhibitors of advanced glycation end-products. *Nutrition and Food Science*, **2012**, 6, 397-404.
- [25] Weatherburn, M. Phenol-hypochlorite reaction for determination of ammonia. *Analytical Chemistry*, **1967**, *39*, 971-974.
- [26] Mercep, M.; Malnar, I.; Hrvacic, B.; Markovic, S.; Sucic, A. F.; Bosnjak, B.; Klonkay, A. C.; Rupcic, R.; Hutinec, A.; Elenkov, I. J. *Bis*-(coumarin) compounds with antiinflammatory activity and their preparation, pharmaceutical compositions and use in the treatment of asthma and inflammatory diseases. PCT Int. Appl. 2006, WO 2006111858 A2 20061026.
- [27] Li, F.; Hu, J.; Wang, J. Coumarin derivative as antioxidant and its preparation. *Faming Zhuanli Shenqing*, **2014**, CN 103601710 A 20140226.
- [28] Hamdi, N.; Puerta, M. C.; Valerga, P. Synthesis, structure, antimicrobial and antioxidant investigations of dicoumarol and related compounds. *European Journal of Medicinal Chemistry*, **2008**, *43*, 2541-2548.

- [29] Muratovic, S.; Duric, K.; Veljovic, E.; Osmanovic, A.; Softic, D.; Zavrsnik, D. Synthesis of *bis*-coumarin derivatives as antimicrobial agents. *Asian Journal of Pharmaceutical and Clinical Research*, **2013**, *6*, 131-134.
- [30] Weatherburn, M. Phenol-hypochlorite reaction for determination of ammonia. *Analytical Chemistry*, **1967**, *39*, 971-974.
- [31] Choudhary, M.I.; Adhikari, A.; Rasheed, S.; Marasini; B.P.; Hussain, N.; Kaleem, W.A.; Atta-ur-Rahman, Cyclopeptide alkaloids of *Ziziphus oxyphylla* Edgw as novel inhibitors of α-glucosidase enzyme and protein glycation. *Phytochemistry Letters*, **2011**, *4*, 404-406.
- [32] Wu, C.H.; Yen, G.C. Inhibitory effect of naturally occurring flavonoids on the formation of advanced glycation endproducts. *Journal of Agricultural and Food Chemistry*, **2005**, 53, 3167-3173.
- [33] Dimas, K.; Demetzos, C.; Marsellos, M.; Sotiriadou, R.; Malamas, M.; Kokkinopolos, D. Cytotoxic activity of labdane type diterpenes against human leukemic cell lines *in vitro*. *Planta Medica*, **1998**, *64*, 208-211.

Graphical Abstract

Bis-coumarins; Non-Cytotoxic Selective Urease Inhibitors and Antiglycation Agents

Uzma Salar, ^{a,b} Arsalan Nizamani, ^a Fizza Arshad, ^a Khalid Mohammed Khan, ^{a,e} Muhammed Imran Fakhri, ^a Shahnaz Perveen, ^c Nessar Ahmed, ^d M. Iqbal Choudhary ^{a,b,f}

^aH. E. J. Research Institute of Chemistry, International Center for Chemical and Biological Sciences, University of Karachi, Karachi-75270, Pakistan

^bDr. Panjwani Center for Molecular Medicine and Drug Research, International Center for Chemical and Biological Sciences, University of Karachi, Karachi-75270, Pakistan

^cPCSIR Laboratories Complex, Karachi, Shahrah-e-Dr. Salimuzzaman Siddiqui, Karachi-75280, Pakistan

^dCentre for Biomedicine, School of Healthcare Science, Manchester Metropolitan University, Manchester M1 5GD, United Kingdom

^eDepartment of Clinical Pharmacy, Institute for Research and Medical Consultations (IRMC), Imam Abdulrahman Bin Faisal University, P.O. Box 31441, Dammam, Saudi Arabia ^fDepartment of Biochemistry, King Abdulaziz University, Jeddah-214412, Saudi Arabia

$$R_1$$
 OH R_2 H R_2 Reflux R_1 OH R_2 OH R_2 OH R_2 OH R_3 OH R_4 OH R_5 OH R_5 OH R_6 OH R_7 OH R_8 OH

Scheme-1: Synthesis of bis-coumarin derivatives 1-44.

Rutin (Standard) for Antiglycation Assay (IC₅₀ = 294.46 \pm 1.5 μ M)



Urease Inhibitory Activity (UIA) \Rightarrow IC₅₀ = 4.4-115.6 μ M Antiglycation Activity (AGA) \Rightarrow IC₅₀ = 333.63-919.72 μ M Non-Cytotoxic

Research Highlights

- Synthesis of forty four *bis*-coumarin derivatives.
- Evaluation of synthetic compounds **1-44** for urease inhibitory and antiglycation activities.
- Compound 17 was found to possess potent urease inhibitory activity.
- ➤ Kinetic studies revealed the compound 17 as competitive inhibitor.
- > Several compounds showed good to moderate antiglycation potential.