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# Theranostic Pt(IV) Conjugate with Target Selectivity for Androgen Receptor

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Supporting Information

ABSTRACT: It is difficult to diagnose and treat castrationresistant prostate cancer (CRPC) which occurs due to the overexpression of androgen receptor (AR). Because there is a high level of AR in CRPC, we designed and prepared three Pt(IV)based prodrugs targeting AR. Among them, compound 3, a three-in-one hybrid (an AR binding ligand, a cisplatin unit, and a coumarin moiety), was found to display satisfactory AR binding affinity and antagonist activity against androgen receptor, which could also be effectively internalized and visualized in LNCaP (AR+) cells. Due to its AR affinity, 3 selectively accumulated in greater quantities in LNCaP (AR+) cells than in PC-3 (AR-) cells. Moreover, compound 3 exhibited excellent anticancer activ-



ity superior to cisplatin. These results highlight the targeting theranostic application of Pt(IV) prodrugs.

# INTRODUCTION

Prostate cancer (PC) is now one of the leading causes of cancer deaths in men worldwide.<sup>1,2</sup> Depending on the diagnosis, treatment may involve surgical removal of the prostate gland or androgen depletion therapy (ADT). At present, nonsteroidal antiandrogens such as flutamide, bicalutamide, and nilutamide are the most commonly prescribed ADT drugs whose mechanism of action consists of competitively inhibiting androgen receptor (AR) activity associated with PC growth, division, and survival (Figure 1).<sup>3</sup>



Figure 1. Examples of nonsteroidal antiandrogens reported in the literature.

Although these therapies are initially effective at suppressing tumor growth and improving survival, the tumor will eventually progress and develop into castration-resistant prostate cancer (CRPC), which is almost always incurable and often accompanied by poor prognosis and metastatic spreading, with a median survival range from 18 to 24 months.7 Therefore, effective diagnosis and therapeutic treatments are needed to treat CRPC.

AR is a ligand-dependent transcription factor that is vital for the normal development and maintenance of the prostate. However, AR-mediated gene expression is also believed to act as an important driver throughout prostate cancer progression.<sup>8-10</sup> Although the mechanisms responsible for the progression of PC to CRPC are not well-known, it has become clear that AR protein is essential for CRPC to adapt to the low levels of androgens.<sup>11</sup> It has been reported that the expression level of AR is approximately 6-fold higher in castration-resistant than hormone-sensitive prostate cancer.<sup>11,12</sup> Hence, AR could be recognized as an attractive target for the treatment of CRPC.<sup>13</sup> For example, Koch and co-workers have reported an androgen receptor-targeted nonsteroidal ligand for the tumor-specific delivery of a doxorubicinformaldehyde conjugate.<sup>14</sup> Oyelere et al. studied a series of heterobifunctional conjugates by conjugating nonsteroidal antiandrogen ligands with histone deacetylase inhibitors.<sup>15</sup> Several promising agents including enzalutamide have been applied to improve CRPC patient survival;<sup>16</sup> however, CRPC has evolved to be able to reactivate AR despite continued androgen depletion therapy.<sup>1</sup> Therefore, discovery of more potent drugs beyond only acting on AR is needed.

Cisplatin (CDDP) is one of the most efficient drugs used for cancer chemotherapy. It binds to nuclear DNA, and then causes the blockage of replication and transcription, which ultimately leads to cellular apoptosis.<sup>18,19</sup> However, the clinical efficiency

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Figure 2. Chemical structures of cisplatin, cis-steroidal Pt(II), and several known Pt(IV) complexes.

of cisplatin is limited by several side effects such as nephrotoxicity, nerotoxicity, and ototoxicity mainly due to its nonspecificity.<sup>20,21</sup> To overcome the nonspecificity of cisplatin, Rodger et al. developed a nonconventional Pt(II) complex that conjugates a steroidal androgen to target tumor cells (Figure 2).<sup>22</sup> Over the past decade, the development of inert Pt(IV) prodrugs has been a promising strategy because the axial position of Pt(IV) complexes could be modified by various functional groups.<sup>23–25</sup> On this basis, Lippard et al. reported a method to deliver cisplatin to prostate cancer cells by constructing Pt(IV)-encapsulated prostate-specific membrane antigen (PSMA), providing an avenue for systemic targeted therapy against this cancer.<sup>26</sup> Even though several Pt-based conjugates intended for prostate cancer treatment have been reported thus far,<sup>22,27</sup> there are few reports to our knowledge describing the use of Pt-based complexes to conjugate a nonsteroidal unit targeting AR.

In order to introduce an AR-binding ligand to the axial position of Pt(IV) complexes and to develop selective and potent drugs to treat CRPC, here we report a few scaffolds composed of Pt(IV) complexes and nonsteroidal cyanonilutamide,<sup>14,15,28</sup> which are expected to target PC via AR to endow the Pt(IV) prodrugs with favorable accumulation in cancer cells. In addition to assessing the cellular uptake and accumulation of Pt, a fluorescent agent (coumarin) was also introduced in a Pt(IV) prodrug to implement real-time imaging so that the cell uptake and targeting efficiency could be simultaneously visualized.

# RESULTS AND DISCUSSION

**Design of an AR Ligand.** To endow the Pt(IV) complex with AR-targeting ability, we designed a cyanonilutamide derivative (L-1) as the ligand (Scheme 1). Initial docking studies of L-1, cyanonilutamide and testosterone (an endogenous steroid that binds AR) were carried out to explore the binding capability in the cavity of helix-12 (H12, at the C-terminus of the

AR LBD).<sup>29</sup> The binding modes of these compounds were depicted in Figure 3 and the Surflex docking scores obtained were summarized in Table S1. The docking scores were 5.51 for L-1, 3.64 for cyanonilutamide, and 8.15 for testosterone, where higher scores indicated greater binding affinity. Although weaker than testosterone, the binding affinity of L-1 was greater than its parent compound cyanonilutamide. In Figure 3C, testosterone in complex with AR LBD showed that the carbonyl group had a hydrogen bond interaction with ARG752 and the hydroxyl group had a hydrogen bond with ASN705, contributing to the closed conformation of AR.<sup>30</sup> L-1, possessing a similar binding mode to cyanonilutamide, could also fit into the binding site by the formation of the key hydrogen bond between the cyano group and ARG752 (Figure 3A,B). The docking results suggested that L-1 had a competitive mechanism of action at the same site as testosterone, which might disturb the proper positioning of H12 and repress the activation of AR.

**Synthesis.** Three novel Pt(IV) prodrugs conjugated with L-1 were designed and synthesized (Scheme 1). Cyanonilutamide (6) was prepared according to the former literature report.<sup>28</sup> Compound 7 was synthesized via an alkylation of 6 with ethyl bromoacetate in the presence of sodium hydride. Then L-1 was obtained by the hydrolysis of 7. To explore the effect of ligands on the activity, complexes 1 and 2 were axially coordinated with chloridion and hydroxyl, respectively. Besides, a coumarin derivative (11) was introduced to the axial position of 2 to obtain a theranostic prodrug 3. All new compounds were unambiguously characterized by NMR spectroscopy and ESI-MS. The purity of Pt(IV) complexes 1–3 were confirmed to be more than 95% by HPLC.

**Cyclic Voltammetry.** To show how easily these compounds can be reduced to its Pt(II) equivalents, we measured the reduction potentials of complexes 1-3. In cyclic voltammograms (Figure S1, see Supporting Information), all complexes exhibited

Scheme 1. Synthesis of Pt(IV) Conjugates



irreversible reduction processes with the reduction potentials of 1-3 being -0.435, -0.455, and -0.467 V, respectively. The result indicated that the complex with an axial chloridion ligand (1) possessed the lowest electrochemical reducibility and the complex with an axial carboxyl ligand (3) possessed the highest electrochemical reducibility.

HPLC Analyses on the Stability and Reduction of **Complexes 1–3.** The solution stability and the reduction of complexes were examined by HPLC at different times. As shown in Figures 4A and S18-S20, all Pt(IV) complexes were stable in medium RPMI-1640 (10% fetal bovine serum) and PBS after 24 h. To further confirm whether the complex could be reduced to release its Pt(II) equivalent with the AR-targeting ligand and compare the reducing rate in the presence of ascorbic acid, each compound was investigated by HPLC in a mixture solvent of acetonitrile/water containing ascorbic acid. As seen in Figure S19, the absorption peak of 1 decreased upon reacting with VC and the peak of the AR-targeting ligand increased simultaneously as the time passed ( $T_{1/2}$ : 2.3 h, Figure 4). The similar phenomenon was observed when 2 was incubated with VC apart from the slower reducing rate  $(T_{1/2}: 8.8 \text{ h})$ . As for 3, the reduction could also be observed with the longest time ( $T_{1/2}$ : 13.9 h), as shown by the presence of the AR-targeting ligand and coumarin motiety. Under the same condition, the experimental data of complexes 1-3 were all fitted well to a monoexponential decay function as described by previous reports.<sup>31–33</sup> The rate constants  $(k_{obs})$  of 1-3 were 0.3010 h<sup>-1</sup>, 0.0792 h<sup>-1</sup>, and 0.0501  $h^{-1}$ , respectively. The chemical reduction rate of 2 was

slower than 1, indicating that an axial chlorido ligand in the coordination sphere would facilitate the electron transfer better than an axial hydroxyl ligand. In contrast, 3 displayed the slowest chemical reduction rate, demonstrating that carboxylates do not facilitate the electron transfer from the reducing agent to the Pt(IV) center.<sup>34</sup> However, no peak of the supposed cisplatin species was observed in HPLC chromatograms under the test condition, which was due to its weak chromophore in the UV detecting condition. It was noted that the reducing rate trend of complexes monitored by HPLC was in accordance with that measured by cyclic voltammetry. These observations revealed that the Pt(IV) complexes were stable in PBS and could be reduced to release the axial ligand(s) in the presence of ascorbic acid.

**AR Binding Affinity.** In order to explore the AR binding affinity of the new cyanonilutamide derivatives, the fluorescence polarization (FP) based binding assay by competition with the fluorescent tracer was carried out. As shown in Table 1, such derivatives retained the ability to bind AR directly, while cisplatin had no apparent binding affinity. The values of AR affinity for cyanonilutamide and bicalutamide were 4.42 and 2.51  $\mu$ M, respectively, a little different from those in the previous report.<sup>15</sup> Considering that the values were in the same order, our results were convincing as well. L-1 showed the highest binding affinity with an IC<sub>50</sub> value of 3.75  $\mu$ M among the new cyanonilutamide derivatives. Meanwhile, 1–3 displayed a little weaker binding affinities with IC<sub>50</sub> values in the range from 5.49 to 7.58  $\mu$ M. The tiny decline might be owing to the bigger size of the Pt(IV)



Figure 3. Molecular docking of androgen receptor (AR) in complex with L-1 (A), cyanonilutamide (B), and testosterone (C).

complexes, which made them harder to insert the ligand-binding pocket.

Binding of molecules to AR may lead to antagonist activity.<sup>15</sup> To assess the antiandrogenic activity of the compounds, we evaluated their pharmacological properties in androgen responsive element (ARE)-driven luciferase reporter assay in 293T cells. The AR luciferase assays were examined in the treatment of 1 nM AR agonist R1881 along with  $10 \,\mu$ M tested compounds, respectively, and the antagonist activity was measured as inhibition of R1881 induced luciferase expression. All tested compounds showed excellent AR antagonist activity. However, there was no apparent correlation between the effect on antagonistic activity and binding affinity of these compounds. Overall, it was determined that L-1 and 1–3 all possessed potent binding affinity and displayed antagonist activity against AR.

UV Absorption and Fluorescence Spectroscopy. UV absorption and fluorescence spectra of all Pt(IV) compounds were measured. In Figure 5, complexes 1-3 exhibited an absorption band at 224 nm. As for 3, an extra absorption band at 320 nm was observed, which is typical for the coumarin group.

When excited at 320 nm, 3 ( $\Phi = 0.072$ ) displayed an emission band at 397 nm. However, its fluorescence quantum yield was weaker than the coumarin ligand 11 ( $\Phi = 0.32$ ). It's wellknown that Pt(IV) complexes could usually release the axial ligand after reduced by ascorbic acid or GSH, which were overexpressed in cancer cells.<sup>24</sup> Thus, we assumed that prodrug 3 could exhibit strong fluorescence after uptaken into cancer cells.

**Confocal Microscopy.** With an aim to explore the potential of prodrug 3 as targeted theranostic in cells, LNCaP (AR+) and PC-3 (AR-) were employed for cell imaging to demonstrate the fluorescence response by laser scanning confocal microscopy. As shown in Figure S21, PC-3 (AR-) cells treated with 3 or 11 both showed weak fluorescence. Significantly, LNCaP cells exhibited strong fluorescence intensity upon incubation with 3 for 4 h, confirming the remarkable cellular uptake of 3 by LNCaP cells (Figure 6). Even though the fluorescence in PC-3 cells was weaker than that in LNCaP cells, we could observe the internalization of compound 3, responsible for the death of cells, in both the cells. On the other hand, the cells showed faint



**Figure 4.** (A) Stability of 1–3 in medium RPMI-1640 (10% fetal bovine serum). Time-dependent reductions of 1 (B), 2 (C), and 3 (D) upon VC on HPLC (1–3 were 0.2 mM and VC was 1 mM). The experimental data of complexes 1–3 were fitted well to a monoexponential decay function (equation:  $y = y_0 + A_1 \exp(-x/t_1)$ ,  $k_{obs} = \ln(2)/t_1$ ).

Table 1. AR Binding Affinity and AR Antagonist Activity of Bicalutamide, Cyanonilutamide, Testosterone, Cisplatin, L-1, and complexes 1–3

compound	$IC_{50}$ ( $\mu M$ )	inhibition % <sup><i>a</i></sup> (10 $\mu$ M)
bicalutamide	$2.51 \pm 0.06$	85.4
cyanonilutamide	$4.42 \pm 0.12$	77.8
L-1	$3.75 \pm 0.10$	51.2
1	$5.49 \pm 0.14$	73.5
2	$6.13 \pm 0.10$	66.3
3	$7.58 \pm 0.19$	70.1
testosterone	$0.0095 \pm 0.0006$	$\mathrm{NT}^{b}$
cisplatin	ND <sup>c</sup>	ND

<sup>*a*</sup>Inhibition rate was shown as a ratio to the R1881 control and the data represent the mean of at least three independent experiments. <sup>*b*</sup>NT, not tested. <sup>*c*</sup>ND, not determined.

fluorescence under the same treatment with 11 (the precursor of 3). As the stronger fluorescence often means higher cellular uptake, we could deem that the uptake of 3 was more than that of 11 in LNCaP cells. The results proved that the AR targeted ligand could significantly increase the uptake of 3 in AR overexpressed LNCaP cells and the conjugate could meet the requirement for cell imaging.

**Cellular Uptake.** Since the cytotoxic effect of Pt-based anticancer drugs is highly dependent on cellular uptake and accumulation, we explored the effect of the AR-ligand conjugation on the accumulation of the Pt(IV) complexes. For this purpose, the cellular uptake of the Pt(IV) complexes was determined by inductively coupled plasma mass spectrometry (ICP-MS) in both ARpositive LNCaP and AR-negative PC-3 cells. As shown in Figure 7, the uptake of three Pt(IV) complexes did not exhibit apparent improvement compared with that of cisplatin in PC-3 (AR–) cells. However, in LNCaP (AR+) cells, complexes 1–3 accumulated in a higher extent than cisplatin. This suggested that the conjugation of the AR ligand with Pt(IV) units resulted in the increased cellular uptake, which might account for their higher cytotoxicity in AR-positive LNCaP cells. Moreover, complexes 1–3 preferentially accumulated in LNCaP (AR+) cells compared with PC-3 (AR–) cells, but in contrast, there was almost no difference in the control group of cisplatin. The amount of the platinum uptake of complex 3 was in accordance with the result in cell imaging experiments. These data clearly demonstrated that AR-targeted conjugates displayed the greater abilities of accumulation and selectivity owing to the acquired AR binding affinity.

In Vitro Cytotoxicity. To investigate cell-type selectivity and potency, we estimated the antiproliferative activity of these compounds in LNCaP (AR+) and PC-3 (AR-) cells. The in vitro cytotoxicity of the novel compounds against the two cell lines was evaluated by the MTT assay with cisplatin, bicalutamide and cyanonilutamide as reference controls. The corresponding IC<sub>50</sub> values (concentration required to reduce viability to 50%) obtained after 72 h exposure were summarized in Table 2. Bicalutamide showed moderate antiproliferative activity in LNCaP cells and no apparent activity in PC-3 cells, whereas cyanonilutamide had no activity in both cells. Not surprisingly, the IC<sub>50</sub> value of cisplatin in LNCaP cells was higher than that in PC-3 cells (SI = 0.75), which was consistent with the previous report.<sup>26</sup> This result indicated that cisplatin did not show any pronounced AR dependency in its cytotoxicity to either cell lines. However, it was gratifying to observe that all three complexes are



Figure 5. (A) Absorption of complexes 1–3 and 11. (B) Fluorescence spectra of 3 and 11. Compounds were dissolved in PBS containing 5% DMF at the concentration of 0.1 mM.



**Figure 6.** Confocal microscopy images of LNCaP cells treated with 20  $\mu$ M compound **11** (d–f), 20  $\mu$ M prodrug **3** (g–i), or untreated control cells (a–c). Fluorescence from DAPI staining nucleus appears as blue signals (Ex: 340 nm; Em: 450–470 nm) in the left column while fluorescence from **11** or **3** appears as green signals (Ex: 355 nm; Em: 490–520 nm) in the middle column. The right column was the merge of both above. Scale bar: 30  $\mu$ m.



Figure 7. Cellular uptake of cisplatin and 1-3 in PC-3 and LNCaP cells after 12 h of incubation.

generally more cytotoxic against AR-dependent LNCaP cells than them against AR-independent PC-3 cells. More significantly,

Table 2. In Vitro Cytotoxicity of Cisplatin, Bicalutamide, Cyanonilutamide, L-1, 11, and Complexes 1–3

	$IC_{50}$ ( $\mu$ M)		
compound	PC-3 (-)	LNCaP (+)	SI <sup>a</sup>
cisplatin	$7.31 \pm 0.29$	$9.78 \pm 0.48$	0.75
bicalutamide	>50	$47.39 \pm 1.85$	
cyanonilutamide	>50	>50	
L-1	>50	>50	
11	>50	>50	
1	$6.88 \pm 0.51$	$3.15 \pm 0.27$	2.18
2	$5.79 \pm 0.39$	$2.87 \pm 0.15$	2.01
3	$2.84 \pm 0.25$	$1.02 \pm 0.08$	2.78

 $^a{\rm SI}$  (Selectivity Index) is defined as  ${\rm IC}_{\rm 50}$  value in PC-3 cells/IC\_{\rm 50} value in LNCaP cells.

all Pt(IV) complexes displayed enhanced activity superior to cisplatin. The results demonstrated that the ability of these



Figure 8. Flow cytometry analysis for apoptosis (A) and cell cycle distribution (B) of LNCaP cells induced by cisplatin or 3 at 20 µM for 24 h.

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AR-Pt(IV) conjugates to bind to AR has indeed conferred celltype selectivity and improved the antitumor effect remarkably. Notably, the theranostic prodrug 3, conjugated with a AR-targeting ligand and the coumarin motiety simultaneously, was  $9.6 \times$  more active than cisplatin and exhibited the best selectivity (SI = 2.78).

Cell Cycle and Apoptosis Analysis. Since prodrug 3 displayed the best antitumor activity in vitro, we chose 3 as the optimal candidate for further therapeutic mechanism investigation. To evaluate whether the inhibition of the cancer cell proliferation was associated with apoptosis, an Annexin V-FITC/PI staining assay was carried out in LNCaP cells. The flow cytometry analysis showed that cisplatin achieved an apoptosis rate of 71.6% (including the early and late apoptosis). In contrast, 3 could achieve an apoptosis rate of 83.0%, much superior to that of cisplatin apparently (Figure 8A). Additionally, we also assessed the effect of 3 on cell cycle by measuring DNA content. LNCaP cells were still treated with 20  $\mu$ M cisplatin and 3 for 24 h, and then cell cycle distribution was tested. As shown in Figure 8B, the percentage of sub-G1 phase, which is considered as a biomarker for DNA damage and related to the presence of apoptosis, increased apparently when treated with cisplatin or 3. Moreover, the cell cycle slightly changed after incubation with cisplatin: the percentage of G0/G1 phase decreased from 51.27% to 47.34%, the percentage of S phase increased from 30.49% to 33.71%, while the percentage of G2/M phase showed no obvious variations, respectively. However, the LNCaP cells were significantly arrested at S phase (40.37%) after incubation with 3, revealing that 3 possessed the stronger cell cycle arrest than cisplatin.

Western Blot Analysis. To gain more insights into the molecular mechanisms the activity of complex 3, we measured

the level of proteins related to the regulation of apoptosis by Western blotting. In Figure 9, the combined treatment with



Figure 9. Western blotting analysis of the expression of related proteins.

cisplatin increased the expression of pro-apoptotic gene Bax and suppressed the expression of antiapoptotic gene Bcl-2. Excitingly, **3** showed a stronger effect on Bax and Bcl-2 genes than cisplatin, confirming that apoptosis was conspicuously promoted by **3**. In addition, **3** also induced the proteolytic cleavage of PARP and obviously decreased expression level of caspase-3, while cisplatin displayed little influence on it. These results proved that the Pt(IV) conjugate **3** might possess a unique mode of action to kill cancer cells different from cisplatin owing to the conjugation with both the AR-targeting ligand and coumarin moiety.

# CONCLUSIONS

Herein, we have described a type of novel AR-targeted anticancer agents produced by the conjugation of a cyanonilutamide ligand to Pt(IV) complexes derived from cisplatin and evaluated their biological activity. Electrochemical characteristics and HPLC studies revealed that the order of the chemical reduction rates of these complexes was 1 > 2 > 3. Competitive experiments showed that all AR-Pt(IV) conjugates displayed satisfactory AR binding affinity and antagonist activity against androgen receptor. Additionally, there was greater accumulation of our complexes, especially 3, in LNCaP (AR+) cells than in PC-3 (AR-) cells. Such differences in accumulation contributed to the selective antitumor activity in vitro. In addition, the internalization of prodrug 3 could be easily visualized by confocal microscopy due to its fluorescence property, hinting at its potential as a targeting theranostic agent. Further mechanistic research revealed that 3 arrested the cell cycle at S phase and dramaticlly increased the apoptosis.

Our three-in-one hybid 3 with its innovative design is a Pt(IV) prodrug that can simultaneously bind to AR and emit fluorescence. The AR-targeted theranostic agent is able to add the AR overpressed tumor selectivity of cisplatin and offer a promising strategy to treat CRPC.

## EXPERIMENTAL SECTION

Materials and Instrument. All chemicals and solvents were of analytical reagent grade and used without further purification, unless noted specifically. c,c,t-[Pt(NH<sub>3</sub>)<sub>2</sub>Cl<sub>3</sub>(OH)] and c,c,t-[Pt(NH<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub>(OH)<sub>2</sub>] were prepared according to literature reports.<sup>22</sup> The purity of all compounds used in the biological studies was ≥95%. All antibodies were purchased from Santa Cruz Biotechnology. All cancer cell lines were obtained from Jiangsu KeyGEN BioTECH company (China). Cell cycle and apoptosis experiments were measured by flow cytometry (FAC Scan, Becton Dickenson) and analyzed by Cell Quest software. <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were recorded in DMSO- $d_6$  on a Bruker 300 MHz spectrometer. <sup>195</sup>Pt NMR spectra were measured in DMSO-d<sub>6</sub> with a Bruker 600 MHz spectrometer. Platinum contents were determined by Inductively Coupled Plasma-Mass Spectrometer (ICP-MS, Optima 5300DV, PerkinElmer, USA). Mass spectra were measured by an Agilent 6224 ESI/TOF MS instrument. Elemental analyses of C, H, and N used a Vario MICRO CHNOS elemental analyzer (Elementar).

**Chemistry. Synthesis, and Characterization of 4.** 4-Fluoro-2-(trifluoromethyl)benzonitrile (4.02 g, 21.3 mmol) was added to a mixture of hydantoin (13.6 g, 106.3 mmol) and potassium carbonate (4.40 g, 31.9 mmol) in DMF (60 mL) and stirred at 45 °C under argon for 48 h. Reaction mixture was diluted in ethyl acetate and washed three times with water. The organic layer was dried over sodium sulfate, filtered and concentrated in vacuo. Column chromatography (eluent 30:1 DCM/Methanol) gave 6 as a white solid (4.62 g, 74%). <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  1.42 (s, 6H), 8.01–8.04 (d, 1H), 8.18 (s, 1H), 8.29–8.32 (d, 1H), 8.84 (s, 1H) ppm.

**Synthesis of 8 (L-1).** Compound 6 (5.00 g, 16.8 mmol) was dissolved in DMF (40 mL) under argon, followed by addition of NaH (60% in mineral oil, 1.00 g, 25.2 mmol) and stirring for 2 h at room temperature. Then ethyl bromoacetate (4.22 g, 25.2 mmol) was added and the reaction was stirred for 5 h at 55 °C. To the reaction was added EtOAc (200 mL) and the mixture was successively washed with brine (5 × 150 mL) and H<sub>2</sub>O (3 × 125 mL). The organic layer was dried over sodium sulfate, filtered and concentrated in vacuo. Column chromatography (eluent 5:1 PE/EtOAc) gave 7 as a white solid (5.21 g, 81%). Compound 7 (5.01 g, 13.1 mmol) was dissolved in MeOH (40 mL), followed by addition of NaOH (2.0 g, 50.4 mmol) and stirring for 2 h at 50 °C. The reaction mixture was concentrated, dissolved in 150 mL of H<sub>2</sub>O and adjusted the pH value to acidity. The precipitate was filtered and dried in vacuo to get 8 as a white solid (4.55 g, 97%). ESI-HRMS:  $m/z [M - H]^- = 354.0699$ . <sup>1</sup>H NMR (300 MHz,

DMSO- $d_6$ )  $\delta$  1.45 (s, 6H), 4.14 (s, 2H), 8.04–8.07 (d, 1H, J = 8.5 Hz), 8.20 (s, 1H), 8.30–8.33 (d, 1H, J = 8.3 Hz), 12.93 (s, 1H) ppm.

Synthesis of 11. 7-Hydroxycoumarin (1.62 g, 10 mmol), potassium carbonate (6.91 g, 50 mmol), bromoacetic acid (6.95 g, 50 mmol), and acetone (250 mL) were added into a 500 mL round-bottom flask. Then catalytic amount of potassium iodide was added. The mixture was refluxed for 20 h. The reaction was monitored by TLC. After reaction finished, water (200 mL) was added and the pH was adjusted to ~3 by 5% HCl aqueous. Then acetone was removed under reduced pressure at room temperature. The mixture was filtered and the solid was washed by water (30 mL × 3) and Et<sub>2</sub>O (30 mL × 3). The white solid was dried, yield 89%. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  4.82 (s, 2H), 6.82–6.32 (d, 1H, J = 9.5 Hz), 6.94–6.97 (m, 2H), 7.62–7.65 (d, 1H, J = 9.2 Hz), 7.97–8.00 (d, 1H, J = 9.5 Hz), 13.09 (s, 1H) ppm.

Synthesis of 1. A solution of TBTU (105.6 mg, 0.33 mmol) and 8 (117.2 mg, 0.33 mmol) in 10 mL of anhydrous DMF was stirred at room temperature under N2 atmosphere. After 10 min, TEA (33.3 mg, 0.33 mmol) was added and the reaction was stirred for 15 min, c,c, t-[Pt(NH<sub>3</sub>)<sub>2</sub>Cl<sub>3</sub>(OH)] (115.8 mg, 0.33 mmol) was then added and the reaction mixture was stirred at room temperature for 12 h. The solvent was then removed by evaporation under reduced pressure. Column chromatography (eluent 15:1 DCM/methanol) gave 1 as a pale yellow solid (111.6 mg, 49%). ESI-HRMS: calcd for m/z [M - H]<sup>-</sup>, 687.9946; found, 687.9851. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ )  $\delta$  1.45 (s, 6H), 4.10 (s, 2H), 6.20 (m, 6H), 8.04-8.07 (d, 1H), 8.19 (s, 1H), 8.30-8.33 (d, 1H) ppm. <sup>13</sup>C NMR (75 MHz, DMSO-d<sub>6</sub>) δ 22.87, 42.72, 62.27, 107.38, 115.70, 121.37 (J = 274.6 Hz), 124.46 (J = 274.6 Hz) 4.8 Hz), 130.44, 131.52 (J = 33.6 Hz), 136.70, 137.74, 153.23, 175.26, 175.95 ppm. <sup>195</sup>Pt NMR (129 MHz, DMSO-d<sub>6</sub>) δ 552 ppm. Anal. Calcd for C<sub>15</sub>H<sub>17</sub>Cl<sub>3</sub>F<sub>3</sub>N<sub>5</sub>O<sub>4</sub>Pt: C, 26.12; H, 2.48; N, 10.15%. Found: C, 26.35; H, 2.59; N, 10.03%.

Synthesis of 2. A solution of TBTU (105.6 mg, 0.33 mmol) and 8 (117.2 mg, 0.33 mmol) in 40 mL of anhydrous DMF was stirred at room temperature under a N<sub>2</sub> atmosphere. After 10 min, TEA (33.3 mg, 0.33 mmol) was added and the reaction was stirred for 15 min c,c, t-[Pt(NH<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub>(OH)<sub>2</sub>] (110.3 mg, 0.33 mmol) was then added and the reaction mixture was stirred at 55 °C for 12 h. The solvent was then removed by evaporation under reduced pressure. Column chromatography (eluent 8:1 DCM/Methanol) gave 2 as a yellow solid (50.3 mg, 22.7%). ESI-HRMS: calcd for  $m/z [M - H]^{-}$ , 670.0285; found, 670.0168. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>) δ 1.44 (s, 6H), 4.00 (s, 2H), 5.83-6.08 (m, 6H), 8.04-8.06 (d, 1H, J = 8.1 Hz), 8.19 (s, 1H), 8.31–8.33 (d, 1H, J = 8.1 Hz) ppm. <sup>13</sup>C NMR (75 MHz, DMSO- $d_6$ )  $\delta$  22.87, 42.78, 62.19, 107.29, 115.68, 121.35 (J = 273.3 Hz), 124.38 (J = 4.9 Hz), 130.35, 131.50 (J = 32.5 Hz), 136.66, 137.20, 153.16, 175.35, 175.95 ppm. <sup>195</sup>Pt NMR(129 MHz, DMSO- $d_6$ )  $\delta$  1065 ppm. Anal. Calcd for  $C_{15}H_{18}Cl_2F_3N_5O_5Pt:$  C, 26.84; H, 2.70; N, 10.43%. Found: C, 26.65; H, 2.79; N, 10.19%.

Synthesis of 3. A solution of TBTU (35.3 mg, 0.11 mmol) and 11 (24.2 mg, 0.11 mmol) in 10 mL of anhydrous DMF was stirred at room temperature under a N<sub>2</sub> atmosphere. After 10 min, TEA (11.2 mg, 0.11 mmol) was added and the reaction was stirred for 15 min. Complex 2 (73.7 mg, 0.11 mmol) was then added and the reaction mixture was stirred at 25 °C for 12 h. The solvent was then removed by evaporation under reduced pressure. Column chromatography (eluent 15:1 DCM/ methanol) gave 3 as a yellow solid (52.3 mg, 54.5%). ESI-HRMS: calcd for  $m/z [M - H]^-$ , 872.0551; found, 872.0421. <sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>)  $\delta$  1.45 (s, 6H), 4.12 (s, 2H), 4.78 (s, 2H), 6.27-6.30 (s, 1H, J = 10.0 Hz), 6.50 (m, 6H), 6.93–6.95 (d, 1H, J = 8.7 Hz), 7.01 (s, 1H), 7.59–7.61 (d, 1H, J = 8.4 Hz), 7.97–8.00 (d, 1H, J = 9.6 Hz), 8.04-8.07 (d, 1H, J = 8.5 Hz), 8.19 (s,1H), 8.31-8.33 (d, 1H, J = 8.4 Hz) ppm.  $^{13}\mathrm{C}$  NMR (75 MHz, DMSO- $d_6)$   $\delta$  22.81, 41.77, 62.24, 65.01, 102.17, 107.37, 113.02, 113.05, 113.47, 115.67, 121.34 (J =272.6 Hz), 124.42 (J = 4.6 Hz), 129.75, 130.38, 131.51 (J = 31.5 Hz), 136.69, 137.12, 144.81, 153.22, 155.74, 160.83, 161.75, 175.20, 175.27, 176.10 ppm. <sup>195</sup>Pt NMR(129 MHz, DMSO-d<sub>6</sub>) δ 1231 ppm. Anal. Calcd for  $C_{26}H_{24}Cl_2F_3N_5O_9Pt:$  C, 35.75; H, 2.77; N, 8.02%. Found: C, 35.92; H, 2.90; N, 7.86%.

**Cyclic Voltammetry.** The cyclic voltammetry (CV) was measured on PGSTAT101 (Autolab, Metrohm) in a range of potentials from

+1.0 V to -1.0 V, with a three electrode setup comprising a glassy carbon working electrode, platinum wire auxiliary electrode and a calomel reference electrode. Complexes 1–3 were dissolved in PBS containing 5% DMF at the concentration of 20 mM. The scan rate was 100 mV·s<sup>-1</sup> at 25 °C, KCl (0.1 M, pH 7.0) was used as a background electrolyte.

Stability of Complexes 1–3 in PBS Buffer and Medium RPMI-1640. The stability of Pt(IV) complexes in a phosphate buffer saline (PBS) and RPMI-1640 supplemented with 10% fetal bovine serum was investigated by HPLC. The incubation was generated by adding different complexes into PBS buffer or RPMI-1640, which was performed at 25 °C at different times. Reversed phase HPLC was implemented on a 250 × 4.5 mm ODS column and the HPLC profiles were recorded on UV detection at 210 nm. Mobile phase consisted of acetonitrile/water, and flow rate was 1.0 mL/min. The samples were taken for HPLC analysis after filtration by 0.45  $\mu$ m filter.

Reduction Ability of Pt(IV) Complexes 1–3 Treated with Ascorbic Acid. The Reduction of Pt(IV) complexes was carried out using ascorbic acid in acetonitrile/water and reduction products were examined at 25 °C by HPLC. Reversed phase HPLC was implemented on a 250 × 4.5 mm ODS column and the HPLC profiles were recorded on UV detection at 210 nm. Mobile phase consisted of acetonitrile/ water (80:20, v/v), and flow rate was 1.0 mL/min. The samples were taken for HPLC analysis after filtration by 0.45  $\mu$ m filter.

**UV–Vis Spectral test.** The UV absorption of complexes 1-3 in PBS containing 5% DMF at the concentration of 0.1 mM was recorded on a Shimadzu UV2600 instrument equipped with a thermostatically controlled cell holder. The wavelength range is 210-400 nm. All the experiments were studied at  $37 \pm 0.1$  °C, the solvent absorption was deducted as the background.

**Determination of Quantum Yields.** Quantum yields were determined at 25 °C, Rhodamine B ( $\Phi = 0.59$ ) in ethanol was used as a standard. The absorption of Rhodamine B was adjusted to the same value (abs < 0.1) as that of fluorescent molecules. Excitation was chosen at 320 nm; the emission spectra were corrected and integrated from 340 to 520 nm. The quantum yields were calculated with the following equation:  $\Phi_{\text{sample}} = \Phi_{\text{standard}} (F_{\text{sample}}/F_{\text{standard}}) (A_{\text{sample}}/A_{\text{standard}})$ .  $\Phi$  is the quantum yield, *F* is the integration of emission intensity, and *A* is the absorbance value at excitation wavelength.

Preparation of Stock Solutions for Cellular Studies. Organic compounds and Pt(IV) complexes were dissolved in DMF to a final concentration of 20 mM and serially diluted prior to testing. The final DMF concentration in culture medium did not exceed 0.4%. Stock solution of cisplatin was prepared in water and diluted directly into culture medium.

**Cell Culture.** PC-3 (Human prostatic cancer cell line, AR–) and LNCaP (Human prostatic cancer cell line, AR+) were maintained in the logarithmic phase at 37 °C in a 5% carbon dioxide atmosphere using the following monolayer culture media containing 10% fetal bovine serum (FBS), 100  $\mu$ g/mL of penicillin, and 100  $\mu$ g/mL of streptomycin.

Molecular Docking Analysis. All the docking studies were carried out using Sybyl-X 2.0 on a Windows workstation. The initial coordinates for AR were taken from the crystal structure of AR in complex with testosterone (PDB: 2AM9).<sup>24,25</sup> The AR-targeting ligand L-1, cyanonilutamide and the endogenous steroid testosterone were selected for the docking studies. The 3D structures of these selected compounds were first built using Sybyl-X 2.0 sketch followed by energy minimization using the MMFF94 force field and Gasteiger-Marsili charges. We employed Powell's method for optimizing the geometry with a distance dependent dielectric constant and a termination energy gradient of 0.005 kcal/mol. All the selected compounds were automatically docked into the binding pocket of AR by an empirical scoring function and a patented search engine in the Surflex docking program. Before the docking process, the natural ligand was extracted; the water molecules were removed from the crystal structure. Subsequently, the protein was prepared by using the Biopolymer module implemented in Sybyl. The polar hydrogen atoms were added. The automated docking manner was applied in the present work.

**Confocal Microscopy.** Cellular localization of tested compounds in LNCaP and PC-3 cells was determined by using confocal microscopy.  $0.5 \times 10^6$  cells were seeded on the 35 mm glass bottom dishes (MatTek). The cells were treated and incubated with tested compounds at 37 °C under 5% CO<sub>2</sub> for 4 h. Then DAPI was added during the final 15 min of the incubation. The cells were washed by phosphate buffered saline (PBS) and then imaged after further incubation in colorless serum-free media for 15 min. Fluorescence images were taken using a confocal laser scanning microscope (Zeiss LSM 700, Zeiss, Germany). Fluorescence from **11** or **3** appears as green signals (Ex: 355 nm; Em: 490–520 nm), while that from DAPI staining nucleus appears as blue signals (Ex: 340 nm; Em: 450–470 nm). Scale bar, 30  $\mu$ m.

**AR Ligand Binding Affinity.** The fluorescence polarization technique was used to analyze the binding of L-1, 1–3, cisplatin, cyanonilutamide, and bicalutamide to the androgen receptor using the PolarScreen AR Competitor Assay, Green (lifetechnologies, A15880) according to the manufacturer's instructions. Briefly, the assay entails titration of the test compound against a preformed complex of Fluormone AL Green and the AR-LBD (GST). The assay mixture was allowed to equilibrate at room temperature in 384-well black plates for 4 h, after which the fluorescence polarization values were measured in a SpectraMax Paradigm Multi-Mode Detection Platform (Molecular Devices) using an excitation wavelength of 485 nm and an emission wavelength of 535 nm. Data analysis for the ligand binding assays was performed using Prism software (GraphPad Software, Inc.).

Luciferase Assay. Assay of androgenic activity was performed by means of ARE luciferase reporter assay using HEK 293T cells. Cells were sown in a 150 cm<sup>2</sup> flask (Corning), and cultured in culture medium (DMEM medium containing 10% Dextran Charcoal (DCC)-Fetal Bovine Serum (FBS), 2 mM glutamine) for 24 h pcDNA3.1-AR, pRLSV40, and pMMTV-Luc vector containing luciferase gene bound at the downstream of an AR promoter derived from Mouse Mammary Tumor Virus (MMTV) were cotransfected by using Lipofectamine 2000. After culturing at 37 °C in a 5% CO<sub>2</sub> atmosphere for 4 h, these cells were harvested and plated in a 96 well plate (10000 cells/well) and cultured for 2 h. A total of 24 h after addition of the sample (final concentration, 10 µM) and 1 nM R1881, cells were harvested with 20  $\mu$ L of cell passive lysis buffer (Promega), and the firefly and Renilla luciferase activities were determined with a Dual Luciferase Asaay Kit (Promega) by measuring luminescence with a Wallac Micro-Beta scintillation counter (PerkinElmer Life Sciences). The data were obtained in triplicate and expressed as inhibition rate over the R1881 control. Inhibition% = 1 -  $(RLU_{test} - RLU_{blank})/(RLU_{R1881} - RLU_{blank}) \times$ 100%. RLU = relative light unit.

**Cytotoxicity Measurement.** The in vitro activity of the synthesized compounds was determined by the MTT method. The cultured cells with better vitality were transferred to a 96-well plate so that the density of the cells was 5000 per well, and they were incubated overnight. Then, the compounds were dissolved in DMF and diluted with medium to various concentrations (the final concentration of DMF was less than 0.4%). After being incubated at 37 °C for 72 h, cells were stained with 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2*H*-tetrazolium bromide (MTT; 5 mg/mL) for another 4 h, and then dissolved with 150  $\mu$ L of DMSO. The UV absorption intensity was detected with an ELISA reader at 490 nm. The IC<sub>50</sub> values were calculated by SPSS software after three parallel experiments.

**Cellular Uptake Test.** LNCaP and PC-3 cells with good activity were seeded in 6-well plates at 37 °C in 5% CO<sub>2</sub>. After the cell density reached 80%, cisplain or complexes **1–3** was added to each well at a concentration of 20  $\mu$ M, and the plates were incubated for 12 h. Then, cells were collected and washed three times with ice-cold PBS, followed by centrifugation for 10 min and resuspension in PBS (1 mL). Then, 100  $\mu$ L of suspension was taken out for measuring the cell density. The remaining cells were digested by HNO<sub>3</sub> (200  $\mu$ L, 65%) at 65 °C for 10 min. The results were measured by ICP-MS after three parallel experiments.

**Apoptosis Analysis.** LNCaP cells were used in the apoptosis experiment. Cisplatin was the positive control at a concentration of 20  $\mu$ M. Specific operations were as follows: LNCaP cells with good activity (5 × 10<sup>5</sup> cells per plate) were transferred to six-well plates and cultured overnight in 5% CO<sub>2</sub> at 37 °C. Drugs were added, which were diluted to a concentration of 20  $\mu$ M. After 24 h, the cells were digested

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with trypsin and washed twice with cold PBS. Then, cells were collected by centrifugation (2000 rpm, 5 min). After that, cells were resuspended in binding buffer (10 mM HEPES, 140 mM NaCl, 2.5 mM CaCl<sub>2</sub>, pH 7.4) and incubated with annexin V-VAPC (100 ng/mL) and then with propidium iodide (2  $\mu$ g/mL) for 15 min in the dark at room temperature. At last, the fluorescence of cells was detected by an annexin V-APC/7-AAD apoptosis detection kit (Roche) according to the manufacturer's protocol, and cells were analyzed by a computer station running Cell Quest software.

**Cell Cycle Measurement.** LNCaP cells with good vitality were transferred into six-well plates, with a density of 10000 per well, and cultured overnight at 37 °C. Then, 20  $\mu$ M of the tested compounds were incubated with cells for 24 h. All adherent and floating cells were collected and washed twice with PBS. Then, the cells were fixed with 70% EtOH at 4 °C for 24 h. After that, fixed cells were washed with PBS. After being centrifuged, cells were stained with 50  $\mu$ g/mL propidium iodide solution containing 100  $\mu$ g/mL RNase at 37 °C for 0.5 h. The sample (at least 1 × 10<sup>4</sup> cells) was measured by flow cytometry (FAC Scan, Becton Dickenson) using Cell Quest software and recording propidium iodide (PI) in the FL2 channel.

Western Blot. The active preferred LNCaP cells were seeded until the cell density reached 80%. Then 20  $\mu$ M of the compounds were added, and the cells were cultured for 12 h at 37 °C. Proteins were extracted by lysis buffer. The concentration of protein was measured by the BCA (bicinchoninic acid) assay with a Varioskan multimode microplate spectrophotometer (Thermo, Waltham, MA). Then equal amounts of protein (20 mg/Lane) were separated by 10% sodium dodecyl sulfate-polyacrylamide gel electrophoresis (SDS-PAGE). After that, protein samples were transferred onto polyvinylidene difluoride (PVDF) Immobilon-P membrane (Bio-Rad) with a transblot apparatus (Bio-Rad). The blots, blocked with 5% nonfat milk in TBST (Trisbuffered saline plus 0.1% Tween 20) for 1 h, were incubated with primary antibodies diluted in PBST overnight at 4 °C. After that, the membrane was washed with PBST three times and incubated with IRDye 800 conjugated secondary antibody for 1 h at 37 °C. Detection was performed by an Odyssey scanning system (Li-COR, Lincoln, Nebraska).

# ASSOCIATED CONTENT

#### **S** Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.inorgchem.8b00083.

Diagrams of ESI-MS, <sup>1</sup>H, <sup>13</sup>C and <sup>195</sup>Pt NMR spectra. Cyclic voltammograms of complexes 1-3. HPLC analyses on the stability and the released ability of Pt(IV) complexes under reduction with ascorbic acid (PDF).

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#### Notes

The authors declare no competing financial interest.

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# ABBREVIATIONS

DMSO dimethyl sulfoxide; DCM dichloromethane; TEA triethylamine; DMF *N*,*N*'-dimethylformamide; PBS phosphate buffer solution; TBTU *O*-(benzotriazol-1-yl)-*N*,*N*,*N*',*N*'-tetra-methyluronium tetrafluoroborate; ICP-MS inductively coupled plasma mass spectrometry; VC ascorbic acid; CV cyclic voltammetry; AR androgen receptor

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