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Original article

Design and synthesis of novel coumarin analogs and their nematicidal activity against five phytonematodes

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ABSTRACT

The presence of hydroxyl groups at the C4 and C7 positions in coumarin backbone has been proposed as a potential modification site for providing excellent bioactivity according to previous studies. A series of novel coumarin derivatives were rationally designed and synthesized by use of a complex catalytic system for a targeted modification at the above sites. These derivatives were assayed for nematicidal activity. As predicted, the derivatization enhanced the activity of the coumarins against five nematodes. Compounds **7b**, **9a**, **10c** and **11c** showed significant strong nematicidal broad spectrum activity against all tested nematodes. Compound **10c** was the most effective with the lowest LC_{50} values against *Meloidogyne incognita* (5.1 μ mol/L), *Ditylenchus destructor* (3.7 μ mol/L), *Bursaphelenchus mucronatus* (6.4 μ mol/L), *Bursaphelenchus B. xylophilus* (2.5 μ mol/L) and *Aphelenchoides besseyi* (3.1 μ mol/L), respectively. A brief investigation on the structure–activity relationships (SAR) revealed that the targeted modification by a C7 hydroxyl was optimum compared with that of a C4 hydroxyl and that the coupling chain length was crucial for the nematicidal activity.

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1. Introduction

Plant-parasitic nematodes have been one of the most notorious plant pathogens worldwide [1,2]. Thousands of crops and trees are susceptible and the disease caused by the phytonematodes results in billions of agricultural losses annually [3]. Chemical methods, combined with agriculture practice, have been the primary way for the nematode control [4–7]. To lessen environmental toxicity, pesticide residues and nematode resistance, the development of new control substitutes has become an urgent and challenging task [8]. Natural products and their derivatives provide a promising treasury for the identification of modern pesticides [9]. As an alternative to a large screening program for the identification of new active materials, a rational program of structural modification of known active compounds can be more efficient and equally useful.

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Coumarins are widely available promising natural compounds for modification due to their broad bioactivities [10–12]. Some simple coumarins, furocoumarines and dicoumarolums, display excellent nematicidal activity and their skeletons have drawn interest for the development of efficient nematicides [13,14]. In our previous study, 7-hydroxycoumarin was isolated from Stellera chamaejasme and had been discovered to show nematicidal activity against Bursaphelenchus xylophilus and Bursaphelenchus mucronatus. In recent years, the structure-activity relationships (SAR) of hydroxycoumarin derivatives have been intensively conducted and indicated that C4 and C7 position modifications in the backbone could enhance their tumor cytotoxicity, termiticidal and bactericidal activities [15-18]. Therefore, we were prompted to design and develop coumarinbased nematicides by modification at these positions (Fig. 1). In a pilot study, we coupled 4-hydroxycoumarin (1) and 7-hydroxy-4-methylcoumarin (2) with alkyl bromides, and a significant difference in activity was observed. Based on this, we proposed to further develop modifications of 4-hydroxycoumarin (1) and 7-hydroxy-4-methylcoumarin (2) by coupling coumarin with functional moieties at the C4 and C7 hydroxyls, and five

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Fig. 1. Structures of coumarins and their modification sites.

phytonematodes were applied to evaluate their nematicidal activities.

2. Experimental

All starting chemicals were of analytical reagent and used without purification. Reactions were monitored by precoated TLC plates (Silica Gel 60 F_{254} . Qingdao Haiyang Chemical Co., Ltd., Qingdao, China), and the spots were visualized by ultraviolet (UV) illumination. Silica gel (200–300 mesh) (Qingdao Haiyang Chemical Co., Ltd.) was used for column chromatography. Melting points were tested with a X-4 melting point apparatus (Beijing Tech Instrument Co., Ltd., China). The structural ¹H NMR and ¹³C NMR spectra were performed on a Bruker AM-400BB instrument (Bruker, Karlsruhe, Germany) with TMS as internal standard, operating at 400 MHz. The chemical shift values are on a δ scale and the coupling constant values (*J*) are in Hertz. ESI-HRMS was recorded using a Bruker micrOTOF-Q II.

2.1. Synthesis of target compounds (3a-12b)

The general synthesis is illustrated in Scheme 1. Compounds **3a–c** were prepared by the reaction of 4-hydroxycoumarin (1) with 1, 2-dibromoethane, 1, 3-dibromopropane and 1, 4-dibromobutane, respectively, in the presence of an alkaline catalyst [19]. Compounds **9a-c** were similarly prepared with the dibromides. 6-Methoxy-4-methylquinolone (5) and 6-hydroxy-4-methylquinolone (6) were synthesized via a Knorr reaction of ethyl acetoacetate with anisidine in the presence of H_2SO_4 [20]. The bromoalkoxy derivatives of 4-hydroxycoumarins (4a-c) and 7-hydroxy-4-methylcoumarins (10a-c) were prepared through the bromoalkylation of 4-hydroxycoumarin (1) and 7-hydroxy-4methylcoumarin (2) with 1, 2-dibromoethane, 1,3-dibromopropane and 1,4-dibromobutane, respectively. N-Alkylation is classically realized with halogen derivatives under alkaline condition, but with simply potassium hydroxide as a catalyst, the coupling reaction between alkyl bromides, coumarins (4a-c) and 6methoxy-4-methylquinolone (5) was unsuccessful. Finally, a complex catalyst system of KOH, KI and tetrabutyl ammonium bromide (TBAB) was developed to prepare compounds 7a-c in high yield. Compounds **11a–c** were then prepared by the reaction between 6-methoxy-4-methyl-quinolone (5) and 7-bromoalkoxy-4-methylcoumarins (**10a-c**). 6-Hydroxy-4-methylquinolone (**6**) coupled with appropriate 4-bromoalkoxycoumarins (4a and b) in the presence of K₂CO₃, KI and TBAB to yield compounds **8a** and **8b**, and the reaction with 7-bromoalkoxy-4-methylcoumarins (10b and \mathbf{c}) gave **12a** and \mathbf{b} . The coumarin analogs were analyzed by ¹H NMR, ¹³C NMR and HR-ESI-MS. The purity of all test compounds was above 95% (determined by HPLC). All the compounds were characterized and the data was listed in Supporting information. Here, compound **11b** was taken as an example to show the typical structure of coumarin analogs.

4-Methyl-7-[3-(6-methoxy-4-methylquinolin-2-on-1-yl)propyloxy]benzopyran-2-one **(11b):** White solid, yield 63%. mp 184-185 °C. Its molecular formula was determined to be $C_{24}H_{23}NO_5$ from the HRESIMS data at *m/z* 406.2 [M + H]⁺. ¹H NMR (400 MHz, CDCl₃): δ 2.32-2.35 (m, 2H, H-12), 2.38 (s, 3H, CH₃-4'), 2.58 (s, 3H,

CH₃-4), 3.92 (s, 3H, OCH₃-6), 4.23 (t, 2H, J = 6.0 Hz, H-13), 4.63 (t, 2H, J=6.0 Hz, H-11), 6.12 (s, 1H, H-3'), 6.76 (s, 1H, H-3), 6.83 (d, 1H, / =2.4 Hz, H-8'), 6.87 (dd, 1H, / =8.8 Hz, 2.4 Hz, H-6), 7.14 (d, 1H, *J* =2.4 Hz, H-8), 7.25-7.28 (m, 2H, H-7', H-5), 7.46 (d, 1H, *J* = 8.8 Hz, H-5'). ¹³C NMR (100 MHz, CDCl₃): δ 18.68, 18.89, 28.95, 30.95, 55.97, 65.46, 101.43, 103.39, 111.91, 112.07, 112.73, 113.12, 113.53, 120.26, 125.47, 125.57, 125.93, 128.95, 152.54, 155.29, 156.04, 160.44, 161.34, 162.07. The ¹H NMR spectrum showed the two methyl protons and one methoxyl protons. The protons at δ 6.12-7.46 indicated the moieties of coumarin and quinolone. The presence of the protons at δ 2.32-2.35 (H-12), 4.23 (H-13) and 4.63 (H-11) revealed the existence of linker between the two moieties. The results of ¹³C NMR were in accordance with that of ¹H NMR spectrum. Therefore, the compound **11b** was elucidated as 4-Methyl-7-[3-(6-methoxy-4-methylquinolin-2-on-1-yl)propyloxy]benzopyran-2-one.

2.2. Nematicidal assay

Five prevalent nematodes, Meloidogyne incognita, Ditylenchus destructor, Bursaphelenchus xylophilus, Bursaphelenchus mucronatus and Aphelenchoides besseyi, were used to evaluate the nematicidal spectrum and potential of the synthesized compounds. B. xylophilus and B. mucronatus, isolated from Pinus massoniana, were supplied by Dr. Han Zhengmin (College of Forest Resources and Environment, Nanjing Forestry University). M. incognita, D. destructor and A. besseyi were provided by Dr. Lin Maosong (College of Plant Protection, Nanjing Agricultural University). D. destructor was grown on potato dextrose Agar (PDA) media containing a strain of Fusarium solani in 9-cm Petri dishes. M. incognita, B. xylophilus, B. mucronatus and A. besseyi were cultured on PDA media with Botrytis cinerea Pers. All nematodes were stored at 28 °C and subcultured before bioassay. With successive observing under a microscope, [2s were collected and suspended in distilled water for the experiments. The tested compounds were dissolved in dimethylsulfoxide (DMSO) to obtain stock solutions, which were diluted with distilled water containing Tween-20 to prepare working solutions. 300 µL of dilutions and J2s suspension (containing about 60-80 J2s) were added into 24-well plates within 24 h. The final concentrations of DMSO and Tween-20 were kept under 0.5% and 0.05% of volume in each well, at which concentration levels, the motility of nematodes exposed was similar to those of a blank control. Distilled water and a solution of DMSO and Tween-20, at concentrations equivalent to those in the treatment wells, was used as a control. Abamectin was used as a positive control [21]. The plates were covered and parafilmed to prevent evaporation, and then incubated in the dark at 28 °C for 72 h. Juveniles were delivered to clean water and observed with a microscope at $40 \times$ magnification (Shanghai Optical Instrument Factory, Shanghai, China). Those immotile in a straight or "L" shape and not recovering in clean water were defined as dead. (Fig. 2) The experiment was conducted twice with three replicates. The values were determined as percentage corrected mortality (±standard deviation) according to the Schneider-Orelli formula:

Corrected mortality% = [(mortality% in treatment – mortality% in control)/(100 – mortality% in control)] \times 100.

The compounds with strong activity were tested further under a series of concentrations to calculate their LC_{50} values. Probit analysis was used to estimate LC_{50} [22,23].

3. Results and discussion

The target coumarin analogs were efficiently prepared in the presence of the complex catalytic system. Their nematicidal activity was evaluated under a series of concentration and the LC_{50}

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Scheme 1. Synthesis of coumarin derivatives. Reagents and conditions: (a) $K_2CO_3/KOH/acetone$, reflux; (b) ethyl acetoacetate, reflux; (c) 80% H₂SO₄, 95 °C; (d) K₂CO₃/acetone, reflux; (e) KOH/KI/TBAB/methylbenzene, 90 °C; (f) K₂CO₃/KI/TBAB/methylbenzene, 90 °C.

Meloidogyne incognita	Ditylenchus destructor	Bursaphelenchus mucronatus	Bursaphelenchus xylophilus	Aphelenchoides besseyi
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Fig. 2. Representative pictures of five plant-parasitic nematodes immersed in the solution of compound 9a at the concentration of 100 μ mol/L for 72 h (observed with microscope at $100 \times$ magnification).

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 Table 1

 Nematicidal activity of coumarin analogs against five plant-parasitic nematodes (calculated at 72 h).

Comp.	LC ₅₀ values (95% CI)/µmol/L						
	M. incognita	D. destructor	B. mucronatus	B. xylophilus	A. besseyi		
1	261.1 (150.0-318.7)	333.5 (220.2-412.1)	329.7 (262.3-483.0)	168.9 (92.1-251.9)	406.7 (307.0-539.8)		
2	>1000	>1000	>1000	>1000	>1000		
3a	>1000	>1000	>1000	>1000	>1000		
3b	>1000	>1000	>1000	>1000	>1000		
3c	940.3 (737.0-1043.5)	286.2 (140.8-351.1)	>1000	744.9 (612.1-955.4)	>1000		
4a	>1000	>1000	>1000	>1000	>1000		
4b	>1000	>1000	>1000	>1000	>1000		
4c	>1000	>1000	>1000	>1000	>1000		
7a	544.4 (195.7-893.2)	436.2 (284.5-663.9)	329.4 (152.6-556.3)	812.9 (464.3-1241.1)	>1000		
7b	64.0 (20.5-127.0)	52.9 (38.8-84.7)	97.9 (72.7-123.2)	103.2 (112.9-427.3)	95.2 (29.4-276.9)		
7c	>1000	>1000	>1000	>1000	>1000		
8a	883.5 (786.8-980.3)	761.8 (510.5-913.8)	>1000	913.1 (565.5-1260.8)	>1000		
8b	>1000	>1000	>1000	>1000	>1000		
9a	37.4 (10.9-67.2)	35.8 (15.7–75.3)	60.9 (31.3-86.7)	190.9 (75.9–260.2)	121.2 (63.9-184.6)		
9b	>1000	>1000	>1000	>1000	>1000		
9c	>1000	>1000	>1000	>1000	>1000		
10a	491.1 (233.1-785.3)	907 (555.0-1255.2)	652.8 (385.8-819.6)	910.5 (486.6-1334.5)	893.7 (618.8-1168.5)		
10b	762.6 (432.2-955.1)	593.7 (247.6-912.1)	256.6 (221.2-535.9)	731.6 (386.7-944.0)	907.9 (555.0-1293.7)		
10c	5.1 (2.0-9.3)	3.7 (1.2-8.5)	6.4 (2.4–9.2)	2.5 (0.3-6.2)	3.1 (1.0-7.3)		
11a	>1000	>1000	>1000	>1000	>1000		
11b	114.5 (98.8–196.3)	214.2 (126.8-301.5)	367.6 (302.6-432.5)	410.2 (333.6-566.4)	384.5 (337.8-453.9)		
11c	42.4 (19.8-70.2)	68.0 (41.2-86.9)	77.8 (42.1–101.4)	145.5 (72.6-232.1)	120.7 (65.4-201.2)		
12a	385.0 (343.3-447.8)	380.6 (343.2-433.9)	430.0 (385.9-497.6)	381.8 (322.2-495.1)	441.6 (400.3-502.5)		
12b	335.9 (279.5-430.4)	275.3 (233.1-335.3)	330.6 (288.8-391.5)	393.3 (315.2-561.3)	>1000		
Abamectin ^a	0.50 (0.15-1.08)	0.62 (0.12-1.36)	0.38 (0.12-0.94)	0.47 (0.11-0.94)	0.26 (0.14-1.22)		

^a Abamectin was used as a positive control.

values were calculated (Table 1). As predicted, the modification on the hydroxyl at C4 and C7 positions led to the identification of promising lead compounds. Among the derivatives, compounds 7b, 9a, 10c and 11c showed significant strong nematicidal activity, with a broad spectrum against all tested nematodes. Compound 10c was the most effective with lowest LC₅₀ values against *M.* incognita (5.1 μmol/L), *D.* destructor (3.7 μmol/L) and B. mucronatus (6.4 µmol/L), B. xylophilus (2.5 µmol/L) and A. besseyi (3.1 µmol/L) respectively. The rest of the derivatives exhibited moderate or weak nematicidal activity. Clearly, the type and position of the substituents were intensively responsible for the activity expression. Different substituents resulted in varied activity. When the coumarin moiety was replaced with that of a quinolone, a change of activity occurred. For the C4 position, substitution of the coumarin moieties with quinolone groups significantly increased (7b) or decreased (7a and c) the nematicidal activity. For the C7 position, compounds **9a-c** showed significantly different activity compared with their quinolone derivatives (11ac). Beside the effect of a particular functional group, the position effect was also significant. Most 7-hydroxy-4-methylcoumarin derivatives showed improved nematicidal activity, whereas only one among all the 4-hydroxycoumarin derivatives (7b) exhibited better activity. Thus, the C7 hydroxyl is more likely to be a better modification site for the preparation of promising compounds.

It is notable that the length of linkers between two terminal moieties was intensively responsible for the activity. For the linkers between two 4-hydroxycoumarins (1), the change of chain length had little impact on the activity of compounds (**3a–c**), with **3a** and **3b** exhibiting none activity and **3c** just showing weak nematicidal activity to specific species. Three-carbon chain lengths (**7b**) were most active among the linkers between 4-hydroxycoumarin (1) and quinolone, and four-carbon atom lengths (**7c**) induced the loss of activity. For the coupling of two 7-hydroxy-4-methylcoumarins (2), the chain length of two carbon atoms (**9a**) was optimum, and chain lengths of three carbon atoms (**9b**) or four (**9c**) resulted in the great loss of activity. However, concerning to compounds **10–12**, it was found a positive correlation between linker lengths and activities

among quinolone coupled 7-hydroxy-4-methylcoumarins (**11a-c**). This was not observed among compounds 10a-c, but they all showed strong activities with four carbon chains. The chain length had little influence on the activity expression of compounds 12a and b. It was worth noting that brombutyl substituted 7-hydroxy-4-methylcoumarin (10c) exerted outstanding nematicidal activity, whereas bromopropyl (10b) and bromoethyl (10a) derivatives showed weak activity, which suggested that the four-carbon chain length was the most active. About mono-coumarin analogs and bi-coumarin ones, it was found that, as a mono-coumarin analog, brombutyl coumarin (10c) exhibited better activities than its bi-coumarin analog (9c), but the different results were observed between bromoethyl analog (10a) and bi-coumarin one (9a). This indicated that the better activity of compound 10c was determined by the combination of coumarin moiety, bromo atoms and butyl chain. Taken together, the length of the alkyl chain was a key factor for the activity, but the optimum length was not fixed and was depended upon the moiety to which it was linked.

4. Conclusion

In summary, a series of coumarin-based compounds were designed and synthesized with a targeted derivatization of the C4 and C7 hydroxyl groups. Their activity against five prevalent nematodes was systematically investigated to reveal structure–activity relationship (SAR). The results showed that the expression of activity was determined by a combination of carbon chain link lengths and functional moieties, and that an optimum link chain length has been established. Compounds **7b**, **9a**, **10c** and **11c** exhibited the most efficient broad spectrum activity and could be promising for development as novel nematicides.

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Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.cclet.2016.01.029.

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