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# A simple procedure for the synthesis of novel 3-(benzofur-2-yl)pyrazole-based heterocycles

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**Abstract** 3-(Benzofuran-2-yl)-1-phenyl-1*H*-pyrazole-4carbaldehyde was used, as a precursor for the synthesis of a series of novel heterocycles by facile reactions with 3-oxo-3-phenylpropanenitrile, 2-cyanoethanethioamide, and various hydrazides such as cyanoacetohydrazide, acetohydrazide, and carbohydrazide derivatives. The structures of the products were confirmed by various spectroscopic methods along with the X-ray crystal structures.

#### **Graphical Abstract**



**Keywords** 2-Substituted benzofuran · Pyrazole-4carbaldehyde · Carbohydrazide · Heterocycles · Synthesis

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#### Introduction

Compounds containing benzofuran moieties are important constituent of many fused heterocycles in which many derivatives have been isolated from natural sources and show various biological activities (Abdel-Wahab et al. 2009; Asif 2016; Dawood et al. 2006; Hiranita et al. 2010; Hiremathad et al. 2015; Khanam 2015; Naik et al. 2015; Rickli et al. 2015; Saha et al. 2011). Various synthetic approaches are known to afford substituted benzofurans (Abdel-Aziem 2015; Abd El-Wahab et al. 2011; Abu-Hashem et al. 2014; Kadieva and Oganesyan 1997; Kumar and Sharma 2016; Liu et al. 2016; Weissberger and Taylor 1974; Wu et al. 2016; Yeung 2012; Zhou et al. 2010). The most common traditional methods involve the reaction of salicylaldehyde and chloroacetic acid (Burgstahler and Worden 1966), Perkin rearrangement of coumarin (Perkin 1871), and cyclization of 2-alkyne phenols (Fürstner and Davies 2005). The most recent examples of high yielding synthesis of 2-alkyl/benzyl benzofurans involve the cyclization of 2-styryl phenols at room temperature in acetonitrile for 1-3 h in the presence of stoichiometric amounts of (diacetoxyiodo) benzene (Singh and Wirth 2012), cyclization of 1-allyl-2-allyloxybenzenes at 90 °C in toluene for 3 h in the presence of a ruthenium catalyst (van Otterlo et al. 2005), cross-coupling of 2-iodophenols and substituted acetylene in the presence of palladium nanoparticles in a mixture of methanol and acetonitrile at 60 °C (Mandali and Chand 2015), and cyclization of 2-(1hydroxyprop-2-ynyl)phenols in the presence of palladium acetate and cesium carbonate as catalysts in acetonitrile (Rajesh et al. 2015). In addition, 2-arylbenzofurans can be obtained in high yields from the reaction of arylboronic acids or potassium aryl trifluoroborates (two mole equivalents) and aliphatic nitriles in aqueous tetrahydrofuran in

the presence of palladium acetate as a catalyst and excess trifluoroacetic acid (Wang et al. 2013a, b). Moreover, the reaction of phenols with diols or bromoalkynes at high temperatures in the presence of a catalyst gave high yields of 2-substituted benzo[b]furans (Lee et al. 2012; Wang et al. 2011).

Recently, we have reported the synthesis of a variety of heterocycles (Abdel-Wahab et al. 2016; Abdel-Megeed et al. 2012a, b; Baashen et al. 2016; Bekheit et al. 2016) as a part of our program in preparation of bioactive heterocycles. The current work deals with the synthesis of novel benzofur-2-ylpyrazoles in which 3-(benzofuran-2-yl)-1-phenyl-1*H*-pyrazole-4-carbaldehyde was used as the starting material.

#### **Results and discussion**

The reaction of equimolar quantities of 1 (Kumar et al. 2007; El-Zahar et al. 2009) with a molar equivalent of 2 in dry ethanol under reflux condition for 3 h, in the presence of piperidine as a base, gave the corresponding benzofuran 3 in 76% yield (Scheme 1). In an attempt to synthesize pyridine-3,5-dicarbonitrile derivative 5 through Michael addition of 2-cyanoethanethioamide (4) to 3, the unexpected product 6 was obtained instead in 72% yield (Scheme 1). In addition, product 6 was obtained in 84% yield from direct reaction between 1 and 4.

IR spectrum of **6** shows a characteristic absorption band  $(2213 \text{ cm}^{-1})$  due to the nitrile group. In addition, its <sup>1</sup>H NMR spectrum shows an exchangeable singlet corresponding to the NH<sub>2</sub> protons (10.22 ppm). The structure of **6** was established further by the X-ray crystallography (Fig. 1).

Formation of 6 could involve a retro-aldol reaction that takes place first for 3 to produce intermediates 7 and 8. Intermediate 8 yields the anion of 2 as a leaving group along with intermediate 9. Reaction of 9 with the thioamide anion, in the sense of an aldol reaction, affords 6 via intermediate 10 (Scheme 2).

Reaction of 1 with 2-cyanoacetohydrazide (11; two mole equivalents) in dry, boiling methanol containing acetic acid gave benzofuran 12 in 83% yield (Scheme 3). Compound 12 was previously synthesized under similar reaction condition but in acetic acid (Abdel-Aziem 2015). However, the physical and spectroscopic data for compound 12 are not exactly the same as the reported ones. For example, the <sup>1</sup>H NMR spectrum of 12 shows a characteristic singlet (4.21 ppm) due to the methylene protons compared to 3.98 ppm for the reported one. In addition, the electron impact mass spectrum of 12 shows an intense molecular ion peak at m/z 369 compared to m/z 368 (M – 1) for the reported one Condensation of 12 with 1 in boiling methanol in the presence of piperidine gave substituted benzofuran 13 in 80% yield (Scheme 3). The



**Scheme 1** Synthetic routes to new thioamide derivative 6



Fig. 1 X-ray structure of compound 6

Scheme 2 Proposed

compound 6

region. The <sup>1</sup>H NMR spectra of **15** show singlet signals (2.10-2.36 ppm) due to the methyl protons .

Reaction of 1 with acetohydrazide 16 gave the corresponding hydrazone 17 in 86% yield (Scheme 5). The  $^{1}$ H NMR spectrum of 17 shows a characteristic singlet (8.74 ppm) and an exchangeable singlet (11.43 ppm) due to the CH=N and NH protons, respectively.

Finally, the reaction of 1 with carbohydrazide 18 gave the corresponding hydrazone 19 in 85% yield (Scheme 6). The <sup>1</sup>H NMR spectrum of **19** shows a singlet (8.87 ppm) and an exchangeable singlet (10.22 ppm) due to the CH=N proton and NH proton, respectively. In addition, the X-ray crystallography (Fig. 2) confirmed the structure of 19.

#### **Experimental**

### General

electron impact mass spectrum of 13 shows an intense molecular ion peak at m/z 639.

In a similar manner, reaction of 1 with cyanoacetohydrazides 14a-c in boiling methanol containing piperidine afforded acrylohydrazides 15a-c in 84-87% yields (Scheme 4). The IR spectra of 15 show the presence of the carbonitrile groups that appear within the  $2217-2220 \text{ cm}^{-1}$ 

The melting points were taken on Electrothermal IA 9000 series digital melting point apparatus. The IR spectra (KBr) were recorded on a Shimadzu CVT-04 spectrophotometer. The <sup>1</sup>H NMR spectra were measured in DMSO- $d_6$  on a JEOL E.C.A-500 MHz spectrometer in which tetramethylsilane was used as internal standard. The mass spectra were determined using a Varian MAT CH-5 spectrometer (70 eV). Compounds 1 (Kumar et al. 2007), 14a (Mohareb



BH = piperidine



Scheme 3 Synthetic access to new 2-cyanoacrylohydrazide 10



Scheme 5 Synthetic route to novel acetohydrazide 15

et al. 2009), **14b** (Rida et al. 2005), **14c** (Zhang et al. 2012), **16** (Mullican et al. 1993), **18** (Abdel-Wahab et al. 2008) were synthesized based on the reported procedures. The

Crystallographic data for compounds **6** and **19** were deposited at the Cambridge Crystallographic Data Center as CCDC1531680 and CCDC1531681, respectively.

compound 16



Scheme 6 Synthetic access to novel carbohydrazide 16



Synthesis of 3-(benzofuran-2-yl)-1-phenyl-1H-pyrazole-4carbaldehyde (1)

Phosphorus oxychloride (20 mL) was added dropwise to a stirred dimethyl formamide (150 mL) at 0-5 °C. (1-Benzofuran-2-yl)ethylidene-2-phenyl hydrazine (23.53 g, 0.0940 mol) was added to the mixture in a portion-wise manner. The mixture was stirred overnight at room temperature, poured onto ice-cold water and neutralized with ammonium hydroxide solution (5%). The solid produced was filtered, dried, and crystallized from acetic acid to give 1 (24.39 g, 0.0846 mmol; 90%), Mp 160-162 °C (lit. 116-118 °C; Kumar et al. 2007; lit. 160-162 °C; El-Zahar et al. 2009).

Synthesis of 3-(3-(benzofuran-2-yl)-1-phenyl-1H-pyrazol-4-yl)-2-benzoylacrylonitrile (3)

A mixture of 1 (1.44 g, 5.00 mmol) and 2 (0.72 g, 5.00 mmol), piperidine (0.3 mL, 3.04 mmol), and dry EtOH (30 mL) was refluxed for 3 h. The solid obtained was filtrated, dried, and crystallized from EtOH to give pure 3 (1.58 g, 3.80 mmol; 76%). Mp 255-256 °C. IR (KBr)  $v_{\text{max}}$ /cm<sup>-1</sup> 1650 (C=O), 2227 (CN). <sup>1</sup>H NMR:  $\delta$ 7.41-7.98 (m, 15H, Ar-H), 8.60 (s, 1H, CH), 9.31 (s, 1H, CH-pyrazole). <sup>13</sup>C NMR:  $\delta$  106.6, 109.3, 111.2, 114.9, 116.8, 119.9, 121.9, 123.8, 125.8, 127.6, 128.6, 129.1, 129.9, 130.0, 132.9, 134.6, 136.0, 138.3, 145.8, 146.6, 148.0, 154.4, 190.2. MS (EI) m/z (%): 415 (M<sup>+</sup>, 32), 57

(100). Anal. Calcd. for  $C_{27}H_{17}N_3O_2$  (415.44): C, 78.06; H, 4.12; N, 10.11%. Found: C, 78.23; H, 4.40; N, 10.35%.

# Synthesis of 3-(3-(benzofuran-2-yl)-1-phenyl-1H-pyrazol-4-yl)-2-cyanoprop-2-enethioamide (**6**)

Method A A mixture of 3 (0.83 g, 2.00 mmol), 4 (0.2 g, 2.00 mmol), piperidine (0.3 mL, 3.04 mmol), and dry EtOH (30 mL) was refluxed for 5 h. The solid obtained was filtrated, dried, and crystallized from DMF to give pure 6 (0.53 g, 1.44 mmol; 72%). Mp 205-206 °C. IR (KBr) v<sub>max</sub>/cm<sup>-1</sup> 1342 (C=S), 2213 (CN), 3286–3343 (NH<sub>2</sub>). <sup>1</sup>H NMR:  $\delta$  7.34–8.00 (m, 10H, Ar–H), 8.48 (s, 1H, CH), 9.43 (s, 1H, CH-pyrazole), 10.22 (s, exch., 2H, NH<sub>2</sub>). <sup>13</sup>C NMR:  $\delta$  106.9, 111.6, 114.8, 116.8, 119.4, 119.7, 121.9, 123.7, 125.7, 127.9, 127.9, 128.3, 129.5, 129.8, 129.9, 137.8, 138.5, 144.9, 147.7, 154.5, 191.4. MS (EI) m/z (%): 371  $(M^+ + 1, 43), 370 (M^+, 56), 57 (100)$ . Anal. Calcd. for C<sub>21</sub>H<sub>14</sub>N<sub>4</sub>OS (370.43): C, 68.09; H, 3.81; N, 15.12%. Found: C, 68.32; H, 3.98; N, 15.53%. Selected crystallographic data: orange plates,  $C_{21}H_{14}N_4OS$ , FW = 370.42, T = 296(2) k,  $\lambda = 0.71073$  Å, triclinic, space group P1, a = 5.8807(8) Å, b = 9.2390(11) Å, c = 16.4270(2) Å,  $\alpha = 89.882(10)^{\circ}$ ,  $\beta = 82.806(10)^{\circ}, \quad \gamma = 77.795(11)^{\circ},$  $V = 865.2(2) \text{ Å}^3, Z = 2, \rho_{\text{calc.}} = 1.421 \text{ Mg/m}^{-3}, \text{ crystal}$ size =  $0.45 \times 0.29 \times 0.03$  mm,  $m = 0.021 \text{ mm}^{-1}$ , collected = 3550, reflections independent reflections = 5319, R = 0.075, wR = 0.196, R(int) = 0.033. CDC1531680.

Method B A mixture of 1 (0.57 g, 2.00 mmol), 4 (0.2 g, 2.00 mmol), piperidine (0.3 mL, 3.04 mmol), and dry EtOH (30 mL) was refluxed for 3 h. The solid obtained was filtrated, dried, and crystallized from EtOH to give pure **6** (0.61 g, 1.64 mmol; 84%). The spectroscopic and analytical data were similar to those obtained for **6** using method A.

# N'-((3-(Benzofuran-2-yl)-1-phenyl-1H-pyrazol-4yl)methylene)-2-cyanoacetohydrazide (**12**)

A mixture of **1** (1.44 g, 5.00 mmol), **11** (0.50 g, 5.00 mmol), glacial AcOH (0.5 mL, 8.74 mmol), and dry MeOH (20 mL) was refluxed for 1 h. The formed solid was collected by filtration and dried to give pure **12** (1.53 g, 4.14 mmol; 83%). Mp 215–217 °C (lit 228–230 °C; Abdel-Aziem 2015). IR (KBr)  $v_{max}/cm^{-1}$  1686 (C=O), 1612 (C=N), 2215 (CN), 3282 (NH). <sup>1</sup>H NMR:  $\delta$  4.21 (s, 2H, CH<sub>2</sub>), 7.39–8.10 (m, 10H, Ar–H), 8.61 (s, 1H, CH), 9.33 (s, 1H, CH-pyrazole), 11.78 (s, exch., 1H, NH). MS (EI) m/z (%): 370 (M<sup>+</sup> + 1, 100), 369 (M<sup>+</sup>, 85). Anal. Calcd. for C<sub>21</sub>H<sub>15</sub>N<sub>5</sub>O<sub>2</sub> (369.38): C, 68.28; H, 4.09; N, 18.96%. Found: C, 68.39; H, 4.18; N, 19.21%.

3-(3-(Benzofuran-2-yl)-1-phenyl-1H-pyrazol-4-yl)-N'-((3-(benzofuran-2-yl)-1-phenyl-1H-pyrazol-4-yl)methylene)-2cyanoacrylohydrazide (13)

A mixture of **12** (0.37 g, 1.00 mmol), **1** (0.29 g, 1.00 mmol), piperidine (0.3 mL, 3.04 mmol), and dry EtOH (30 mL) was refluxed for 1 h. The solid product formed on cooling was collected by filtration and dried to give pure 13 (0.51 g, 0.80 mmol; 80%). Mp 237-238 °C. IR (KBr) v<sub>max</sub>/cm<sup>-1</sup> 1598 (C=O), 2223 (CN), 3280 (NH). <sup>1</sup>H NMR:  $\delta$  7.41–8.10 (m, 20H, Ar–H), 8.45 (s, 1H, CH), 8.56 (s, 1H, CH), 8.95 (s, 1H, CH-pyrazole), 9.34 (s, 1H, CH-pyrazole), 11.13 (s, exch., 1H, NH). <sup>13</sup>C NMR: 158.0, 155.2, 154.9, 150.2, 147.4, 145.7, 143.0, 141.4, 140.4, 139.9, 136.3, 130.3, 130.23, 130.19, 129.0, 128.8, 128.0, 126.9, 126.2, 125.7, 125.5, 123.9, 123.3, 119.5, 118.8, 111.7, 108.3, 106.7, 106.6. MS (EI) m/z (%): 640 (M<sup>+</sup> + 1, 42), 639 (M<sup>+</sup>, 100). Anal. Calcd. for C<sub>39</sub>H<sub>25</sub>N<sub>7</sub>O<sub>3</sub> (639.66): C, 73.23; H, 3.94; N, 15.33%. Found: C, 73.47; H, 4.09; N, 15.52%.

*Synthesis of* **15** A mixture of **1** (0.29 g, 1.00 mmol), **14** (1.00 mmol), piperidine (0.3 mL, 3.04 mmol), and dry EtOH (30 mL) was refluxed for 2 h. The solid formed on cooling was filtrated and dried to give pure **15**.

# 3-(3-(Benzofuran-2-yl)-1-phenyl-1H-pyrazol-4-yl)-2cyano-N'-(1-(furan-2-yl)ethylidene)acrylohydrazide (**15a**)

Yield 84% (0.39 g, 0.84 mmol); Mp 235–237 °C. IR (KBr)  $v_{max}$ /cm<sup>-1</sup> 1608 (C=O), 2220 (CN), 3240 (NH). <sup>1</sup>H NMR:  $\delta$  2.36 (s, 3H, CH<sub>3</sub>), 6.89–7.97 (m, 13H, Ar–H), 8.53 (s, 1H, CH), 9.35 (s, 1H, CH-pyrazole), 10.55 (s, exch., 1H, NH).<sup>13</sup>C NMR:  $\delta$  11.7, 103.9, 106.2, 109.1, 110.8, 115.8, 111.6, 112.9, 119.3, 121.3, 123.3, 124.6, 125.1, 126.2, 127.2, 127.8, 129.4, 137.9, 139.7, 141.6, 143.2, 144.9, 147.4 153.9, 190.2. MS (EI) m/z (%): 462 (M<sup>+</sup> + 1, 100), 461 (M<sup>+</sup>, 35). Anal. Calcd. for C<sub>27</sub>H<sub>19</sub>N<sub>5</sub>O<sub>3</sub> (461.47): C, 70.27; H, 4.15; N, 15.18%. Found: C, 70.38; H, 4.31; N, 15.37%.

# 3-(3-(Benzofuran-2-yl)-1-phenyl-1H-pyrazol-4-yl)-N'-(1-(benzofuran-2-yl)ethylidene)-2-cyanoacrylohydrazide (15b)

Yield 86% (0.41 g, 0.86 mmol); Mp 240-242 °C. IR (KBr)  $v_{max}/cm^{-1}$  1605 (Cld=O), 2218 (CN), 3228 (NH). <sup>1</sup>H NMR:  $\delta$  2.35 (s, 3H, CH<sub>3</sub>), 7.28–8.00 (m, 15H, Ar–H), 8.67 (s, 1H, CH), 9.30 (s, 1H, CH-pyrazole), 11.05 (s, exch., 1H, NH). <sup>13</sup>C NMR:  $\delta$  11.9, 103.8, 106.3, 110.8, 114.3, 119.3, 121.3, 122.9, 123.1, 123.1, 125.1, 127.2, 127.3, 127.8, 129.0, 129.4, 137.9, 144.2, 147.2, 147.4, 152.8, 159.9, 190.4. MS (EI) m/z (%): 512 (M<sup>+</sup> + 1, 100). Anal. Calcd. for C<sub>31</sub>H<sub>21</sub>N<sub>5</sub>O<sub>3</sub> (511.53): C, 72.79; H, 4.14; N, 13.69%. Found: C, 72.99; H, 4.37; N, 13.96%.

3-(3-(Benzofuran-2-yl)-1-phenyl-1H-pyrazol-4-yl)-2cyano-N'-(1-(pyridin-2-yl)ethylidene)acrylohydrazide (15c)

Yield 87% (0.39 g, 0.87 mmol); Mp 262–263 °C. IR (KBr)  $\nu_{max}/cm^{-1}$  1608 (C=O), 2227 (CN), 3228 (NH). <sup>1</sup>H NMR:  $\delta$  2.10 (s, 3H, CH<sub>3</sub>), 7.31-8.35 (m, 14H, Ar–H), 8.73 (s, 1H, CH), 9.29 (s, 1H, CH-pyrazole), 11.13 (s, exch., 1H, NH). MS (EI) *m*/*z* (%): 472 (M<sup>+</sup>, 45), 191 (100). Anal. Calcd. for C<sub>28</sub>H<sub>20</sub>N<sub>6</sub>O<sub>2</sub> (472.50): C, 71.17; H, 4.27; N, 17.79%. Found: C, 71.29; H, 4.42; N, 17.96%.

Synthesis of **17** and **19** A mixture of **1** (0.29 g, 1.00 mmol), appropriate carbohydrazide **16** or **18** (1.00 mmol), glacial AcOH (0.5 mL, 8.74 mmol), and dry MeOH (20 mL) was refluxed for 1 h. The formed solid was filtrated, dried, and crystallized from DMF.

### N'-((3-(Benzofuran-2-yl)-1-phenyl-1H-pyrazol-4yl)methylene)-2-(naphthalen-2-yloxy)acetohydrazide (17)

Yield 86% (0.42 g, 0.86 mmol); Mp 208–210 °C. <sup>1</sup>H NMR:  $\delta$  4.62 (s, 2H, CH<sub>2</sub>), 7.31–7.98 (m, 17H, Ar–H), 8.74 (s, 1H, CH), 9.13 (s, 1H, CH-pyrazole), 11.43 (s, exch., 1H, NH).MS (EI) *m*/*z* (%): 486 (M<sup>+</sup>, 100). Anal. Calcd. for C<sub>30</sub>H<sub>22</sub>N<sub>4</sub>O<sub>3</sub> (486.52): C, 74.06; H, 4.56; N, 11.52%. Found: C, 74.31; H, 4.67; N, 11.75%.

5-(Benzofuran-2-yl)-N'-((3-(benzofuran-2-yl)-1-phenyl-1H-pyrazol-4-yl)methylene)-1-phenyl-1H-pyrazole-3carbohydrazide (**19**)

Yield 85% (0.50 g, 0.85 mmol); Mp 270–271 °C. <sup>1</sup>H NMR: 7.12–7.91 (m, 20H, Ar–H), 8.87 (s, 1H, CH), 9.13 (s, 1H, CHpyrazole), 9.35(s, 1H, CH-pyrazole), 10.22 (s, exch., 1H, NH). MS (EI) *m/z* (%): 588 (M<sup>+</sup>, 100). Anal. Calcd. for C<sub>36</sub>H<sub>24</sub>N<sub>6</sub>O<sub>3</sub> (588.61): C, 73.46; H, 4.11; N, 14.28%. Found: C, 73.62; H, 4.37; N, 14.64%. Selected crystallographic data: orange plates, C<sub>39</sub>H<sub>31</sub>N<sub>7</sub>O<sub>4</sub>, FW = 661.71, *T* = 293(2) K,  $\lambda = 0.71073$  Å, triclinic, space group P1, *a* = 9.8227(5) Å, *b* = 11.4055(6) Å, *c* = 29.4496(14) Å, α = 91.753(4)°,  $\beta = 97.043(4)°$ ,  $\gamma = 93.894(4)°$ , *V* = 3264.3(3) Å<sup>3</sup>, *Z* = 4,  $\rho_{calc.} = 1.346$  Mg/m<sup>-3</sup>, crystal size = 0.38 × 0.24 × 0.12 mm, *m* = 0.090 mm<sup>-1</sup>, reflections collected = 28397, independent reflections = 15581, *R* = 0.058, wR = 0.149, *R*(int) = 0.030. CCDC1531681.

### Conclusion

Several novel heterocycles having a 3-(benzofur-2yl)pyrazole moiety were synthesized in high yields using simple and convenient procedures. Moreover, the synthetic processes involved required only commercially available starting materials.

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