

Design, synthesis and anti-acetylcholinesterase evaluation of some new pyrazolo[4,3-*e*]-1,2,4-triazolo[1,5-*c*]pyrimidine derivatives

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Abstract The target new hybrid molecule types pyrazolo[4,3-*e*]-1,2,4-triazolo[1,5-*c*]pyrimidines phosphonates **4** and 2-(coumarin-3''-yl)-7-phenylpyrazolo[4,3-*e*]-1,2,4-triazolo[1,5-*c*]pyrimidines **5** were prepared via Michaelis–Arbuzov rearrangement (Arbuzov reaction) of pyrazolo-triazolopyrimidines chloride **3a–c**, with trialkyl phosphate and Knoevenagel reaction of 2-cyanomethyl derivatives **3d–f** with salicylic aldehyde, respectively. The precursors **3** were obtained in two steps starting from aminopyrazole **1**. Target compounds **4** and **5** were completely characterized by ¹H NMR, ¹³C NMR, ³¹P NMR, IR and HRMS. The anti-acetylcholinesterase activity of compounds **4** and **5** was evaluated, and results found indicated that they have possessed significant activities (IC₅₀ = 1.73–39.86 μM), and the preliminary SAR of these compounds was investigated.

Keywords Aminopyrazole · Phosphoric compounds · Coumarinic compounds · Anti-acetylcholinesterase activity

Introduction

The study of heterocyclic compounds such as five- and six-membered rings containing nitrogen has a known considerable development due to the revealing of their varied effects in diverse domains. In this framework, pyrazoles,

triazoles and pyrimidines, as well as their condensed derivatives, are very attractive targets from both a theoretical and a synthetic point of view. Moreover, they have been the subject of many chemical and biological studies on account of their pharmacological activity, such as anti-inflammatory, analgesic (Amin *et al.*, 2009), cytotoxicity, antitumor (Rashad *et al.*, 2010) and adenosine receptor antagonists (Baraldi *et al.*, 2012). In addition, some of them have shown anti-acetylcholinesterase activity (Fig. 1a–c) (Zhi *et al.*, 2008).

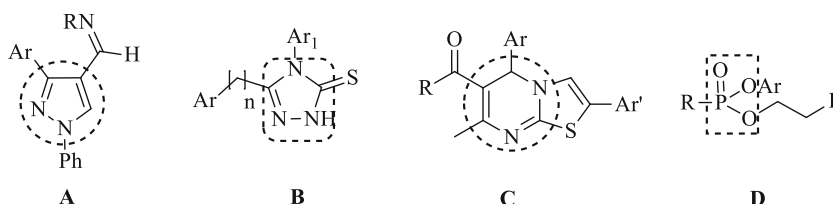
On the other hand, diverse biological activity (Weiqin *et al.*, 2006; Mado *et al.*, 2014) of heterocyclic phosphonates containing the P–C bond has for a long time attracted considerable synthetic and pharmacological interest (Ali *et al.*, 2001). Some of them exert their biological action on arthropods by attacking the system of neural transmission and inhibiting the function of acetylcholinesterase (Fig. 1d) (Fest and Schmidt 1973; Troev 2006). Thus, a large number of new phosphonate derivatives have been prepared hitherto with special attention to nitrogen heterocyclic compounds (Francesca *et al.*, 2001; Yanchang *et al.*, 2002; Francisco *et al.*, 2003; Lise *et al.*, 2007).

Furthermore, coumarins are a class of compounds, which occupies a special role in nature. They have attracted intense interest in recent years because of their diverse pharmacological properties like anti-HIV (Ma *et al.*, 2008), anticoagulant (Kidane *et al.*, 2004), antibacterial (Appendino *et al.*, 2004), antioxidant (Kontogiorgis and Hadjipavlou-Litina 2004), cytotoxic (Musa *et al.*, 2008) and particularly acetylcholinesterase inhibitors (Fig. 1e) (Razavi *et al.*, 2013; Nam *et al.*, 2014).

In addition, the synthesis of new hybrid molecules is a new concept in the field of drug development based on the association between pharmacophoric fragments of different bioactive substances to produce new hybrid compounds

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Fig. 1 Previously reported AChEI compounds

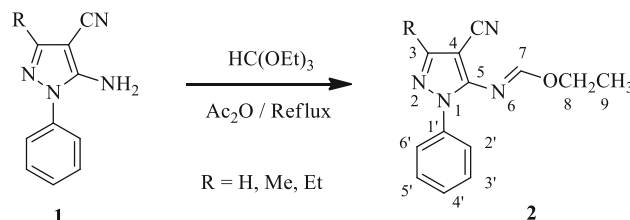
with improved efficiency and affinity. In this context and as a continuation of our previous work on the synthesis of new fused pyrimidine scaffolds (Rahmouni *et al.*, 2014a, b), we report here the synthesis of some new hybrid molecules **4** and **5** bearing in their structures fragments described as anti-acetylcholinesterase agents, as indicated above, such as pyrazole, pyrimidine, triazole and phosphoric or coumarinic systems. Compounds **4** and **5** were evaluated for their anti-acetylcholinesterase activity, and preliminary SAR of these compounds was investigated (Table 1).

Results and discussion

Chemistry

Our key intermediate was the α -functionalized iminoethers type **2**, which were synthesized from aminopyrazole **1** (Scheme 1), since such system constitutes a commonly used building block for the construction of a variety of polyheterocycles (Zaki 1998; El-Agordy *et al.*, 2001).

For this purpose and as described in schemes 2, 4 and 5, our approach to the target systems **4** and **5** was firstly started by the construction of the pyrazolotriazolopyrimidine skeleton type **3** via the intramolecular cyclocondensation reaction of **2**. In fact, these intermediates, which possess two reactive sites: a cyano group and an imidic carbon, were made to react with appropriate acid hydrazide under ethanol reflux to give desired fused tricyclic triazolopyrimidines. Plausible pathway involves two successive nucleophilic additions of $-\text{NH}_2$ group on the imidic carbon and on the cyano function followed by

**Scheme 1** Synthetic route of compounds **2**

dehydrocyclization to give triazolopyrazolopyrimidines **3a–f** (Scheme 2).

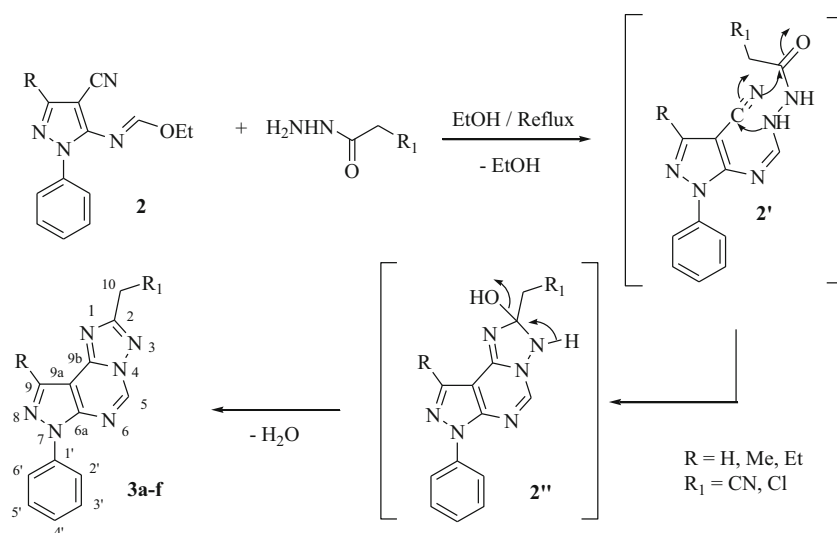
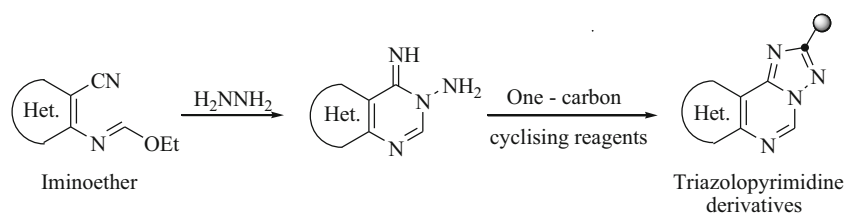
We note that triazolopyrimidine derivatives, such compounds **3**, have generally been prepared, in two steps, via reaction of α -functionalized iminoethers with hydrazine followed by cyclocondensation reaction with one-carbon cyclizing reagents, namely acids, acid chlorides, orthoesters, CSI and DMFDMA (Romdhane *et al.*, 2003, 2008; Scheme 3).

To access new pyrazolotriazolopyrimidine phosphonates containing the P–C bond **4a–f**, we thought about Michaelis–Arbuzov rearrangement (Arbuzov reaction) which is known as one of the most extensively investigated methods and is widely used to prepare phosphonates, by formation of P–C bond through reaction of an aryl/alkyl halide and trialkyl phosphite (Kosolapov *et al.*, 1950). In fact, formation of new phosphonated compounds **4a–f** in good yields (Table 2) was carried out via the reaction of pyrazolotriazolopyrimidines chloride **3a–c**, with trialkyl phosphite.

In order to determine the best conditions for the preparation of pyrazolotriazolopyrimidines phosphonates **4**, we examined the reaction under various conditions by changing the solvent (toluene, xylene), and we found that

Table 1 Yields of pyrazolotriazolopyrimidines **3a–f**

Compounds	R	R ₁	Yields (%)
3a	H	Cl	76
3b	Me	Cl	82
3c	Et	Cl	74
3d	H	CN	68
3e	Me	CN	80
3f	Et	CN	82

Scheme 2 Synthetic route of compounds **3a–f****Scheme 3** Synthesis of triazolopyrimidine derivatives, in two steps**Table 2** Yields of pyrazolotriazolopyrimidines dialkylphosphonates **4a–f**

	R	R ₂	Toluene/reflux Time (h)/yield (%)	Xylene/reflux Time (h)/yield (%)	P(OR ₂) ₃ /reflux Time (h)/yield (%)
4a	H	Me	48/35	24/42	6/88
4b	H	Et	48/30	24/47	7/90
4c	Me	Me	48/38	24/50	6/92
4d	Me	Et	48/36	24/46	7/98
4e	Et	Me	48/40	24/52	6/96
4f	Et	Et	48/31	24/45	7/94

the best yields (Table 2) were obtained when compounds **3a–c** were reacted under reflux with an excess of trialkyl phosphite used at the same time as a solvent. The reaction was conducted until TLC indicated that the starting materials have been completely converted to products **4**.

The structures of compounds **4** have been assigned from their analytical data, IR, ¹H NMR, ¹³C NMR, ³¹P NMR and mass spectrometry (ES-HRMS). In fact, the ¹H NMR spectra of compounds **4** showed, in addition to the signals corresponding to the protons introduced by compounds **3**, a new doublet ($J = 21.3\text{--}21.6$ Hz) at 3.58–3.66 ppm attributable to the methylene group (H-10) bearing the phosphoryl moiety. The large splitting was a result of a ² J coupling of being directly bonded to the phosphorus. We also detected the presence of a doublet at 3.82–3.86 ppm

($J = 12.3\text{--}12.4$ Hz), when R₂ = Me, as a result of protons of the methyl group coupled to the phosphorus atom and a multiplet at 4.07–4.11 ppm, when R₂ = Et, due to the coupling of protons of the ethyl group to the phosphorus atom.

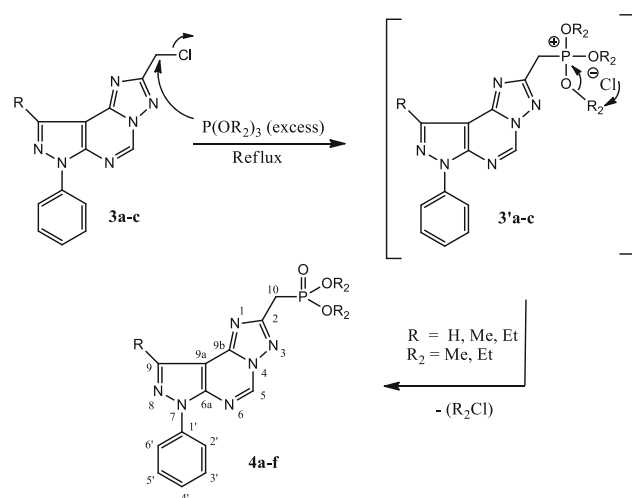
The ¹³C NMR spectra of these compounds were also in agreement with the proposed structures. In fact, in addition to the signals corresponding to the carbons introduced by the intermediate **3**, we observed a doublet at 27.5–28.2 ppm with a very large coupling constant (161.3–161.5 Hz) as a result of being directly bonded of C-10 to the phosphorus atom and two doublets at 62.1–62.7 ppm ($J = 24.3\text{--}24.4$ Hz) and 161.1–162.2 ppm ($J = 33.3\text{--}33.5$ Hz) due to the coupling of the phosphorus atom with the methoxy carbon and C-2, respectively. For compounds **4b,d,e** a new doublet at 16.5–16.8 ppm ($J = 23.4\text{--}23.6$ Hz) was observed

resulting of the coupling between the methyl carbon of the ethoxy group and the phosphorus atom. The ^{31}P NMR spectra of compounds **4a–f** showed a signal corresponding to the phosphonyl group at 21.6–24.6 ppm. The infrared spectra also revealed the presence of an absorbance due to the P = O band at 1230–1238 cm^{-1} (Scheme 4).

In the second part of this study, we have converted the intermediates **2** into pyrazolo[4,3-*e*]-1,2,4-triazolo[1,5-*c*]pyrimidine-2-acetonitriles **3d–f** by reaction with cyanoacetic acid hydrazide, which was obtained by addition of hydrazine hydrate to ethyl cyanoacetate in ethanol with stirring at 0 °C, under reflux of toluene (Gorobets *et al.*, 2004) (Scheme 2), in order to subsequently treat compounds **3d–f** with salicylaldehyde via Knoevenagel condensation reaction, to access to the coumarin derivatives **5** (Scheme 5).

The formed triazolopyrimidines **3d–f** were characterized by their IR, ^1H and ^{13}C spectra. In fact, the IR spectra of compounds **3d–f** showed an absorbance band at 2225–2230 cm^{-1} due to the cyano group. The ^1H NMR spectra of compounds **3d–f** showed, in addition to the signals of protons introduced by the intermediate **2**, the appearance of a new signal due to the methylene group ($-\text{CH}_2-\text{CN}$) at 4.08–4.10 ppm. The ^{13}C NMR of **3d–f** showed the appearance of new signals at 18.8–19.1, 116.8–117.5 and 164.5–165.1 ppm relative to $-\text{CH}_2-\text{CN}$, $-\text{CN}$ and C_2 , respectively.

The 2-cyanomethyl derivatives **3d–f** were treated with salicylic aldehyde, via Knoevenagel reaction (Bylov *et al.*, 1999; Kovalenko *et al.*, 2000); under reflux of ethanol, an addition product formed **3'**, from which reaction of an aqueous solution of hydrochloric acid gave the 2-(coumarin-3''-yl)-14(Aryl)-14*H*-naphto[2,1-*b*]pyrano[3,2-*e*][1,2,4]triazolo[1,5-*c*] pyrimidines **5a–c** (scheme 5).



Scheme 4 Synthetic route of pyrazolotriazolopyrimidine phosphonates **4a–f**

The reaction was conducted until TLC indicated that the starting materials have been completely converted to products **5**. The IR spectra of these compounds revealed the presence of an absorbance band due to C = O at 1740–1747 cm^{-1} and the disappearance of any absorption frequency in the CN region. Further, the ^1H NMR spectra of compounds **5** showed the disappearance of the singlet relative to the methylene protons (H-10) of compounds **3d–f** (4.08–4.10 ppm) and the appearance of new signals, attributable to protons of the coumarin moiety, of which chemical shifts and multiplicities are in agreement with the proposed structure. Analysis of ^{13}C NMR spectra of these compounds showed, in addition to the signals relative to the coumarin moiety carbons, the disappearance of two signals in the region of 18.8–19.1 ppm and 116.8–117.5 ppm attributable to $-\text{CH}_2-\text{CN}$ and $-\text{CN}$ in compounds **3d–f**, respectively.

The ES-HRMS showed essentially the correct protonated molecular peak $[\text{M} + \text{H}]^+$ for all examined compounds **3–5**.

Biological activity

Anti-acetylcholinesterase activity

Inhibition of acetylcholinesterase (AChE), the key enzyme in the breakdown of acetylcholine, is considered one of the treatment strategies against several neurological disorders such as Alzheimer's disease, senile dementia, ataxia and myasthenia gravis (Orhan *et al.*, 2006; Howes *et al.*, 2003). Only compounds **4a–f** and **5a–c** were analysed on what concerns their acetylcholinesterase inhibition activity (Table 3) using an adaptation of the method described in the literature (Ferreira *et al.*, 2006).

The results indicated in Table 3 showed that all the tested heterocycles gave significant activity. Compared to those given in the literature for crude pure products (Mata *et al.*, 2007), we can say that the prepared derivatives **4** and **5** are considered good acetylcholinesterase inhibitors. Moreover, it has been found that the methylated phosphonate derivatives **4a,c,e** were more active than their ethylated analogous **4b,d,f**.

The greatest inhibitory activity was exhibited by the phosphonated compound **4a** ($\text{R} = \text{H}$, $\text{R}_2 = \text{Me}$, $\text{IC}_{50} = 1.73 \pm 0.05 \mu\text{M}$), which was found almost twenty-three times more active than (*E*)-anethole ($\text{IC}_{50} = 39.86 \pm 0.1 \mu\text{M}$) used as a standard compound, but it is about two times less active than tacrine ($\text{IC}_{50} = 0.86 \pm 0.02 \mu\text{M}$), a drug used to treat Alzheimer.

On the other hand, it has been shown that the activity of these derivatives depends on the nature of R . In **4a,c,e** ($\text{R}_2 = \text{Me}$), the acetylcholinesterase inhibition decreases from $\text{R} = \text{H}$ (**4a**) to $\text{R} = \text{Et}$ (**4e**). The same phenomena has been observed with **4b,d,f** ($\text{R}_2 = \text{Et}$).

Scheme 5 Synthetic route of coumarinic pyrazolotriazolopyrimidine derivatives **5a–c**

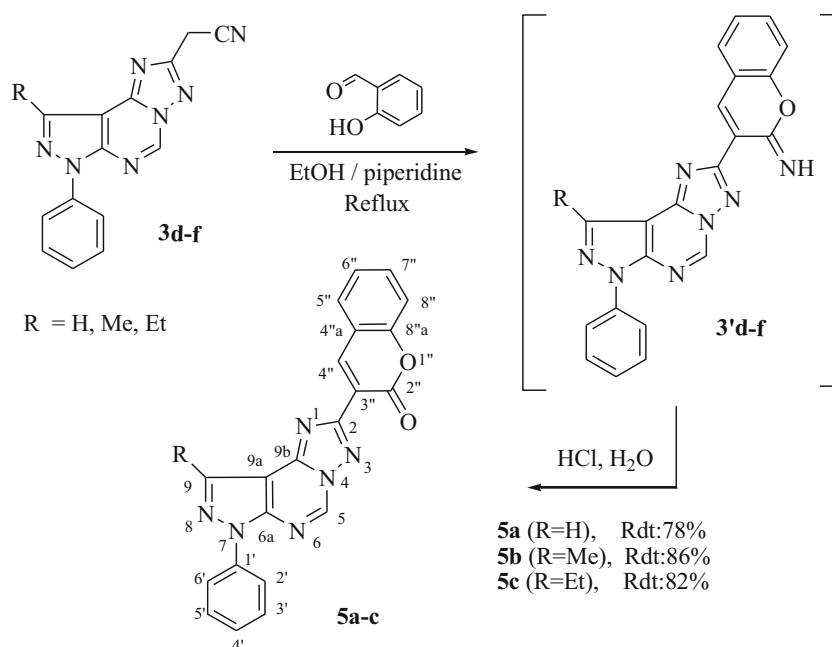


Table 3 Acetylcholinesterase inhibition capacity represented by IC₅₀ (μM)^a of compounds **4**, **5** and standard compounds ((*E*)-anethole and tacrine)

Compound	Acetylcholinesterase inhibition capacity represented by IC ₅₀ (μM)
4a	1.73 ± 0.05
4b	11.71 ± 0.18
4c	4.65 ± 0.20
4d	15.32 ± 0.22
4e	6.50 ± 0.20
4f	19.03 ± 0.13
5a	10.31 ± 0.17
5b	26.54 ± 0.21
5c	36.54 ± 0.20
(<i>E</i>)-Anethole	39.86 ± 0.10
Tacrine	0.86 ± 0.02

^a Averages ± SD were obtained from three different experiments

IC₅₀ values represent the concentration of inhibitor required to decrease enzyme activity by 50 % and are the mean of two independent measurements, each performed in triplicate

For coumarinic compounds, it has been shown also that the activity of these derivatives depends on the nature of R; in fact, the acetylcholinesterase inhibition decreases from R = H (**5a**) to R = Et (**5c**). Compound **5c** (R = Et) was found to be the less inhibitor (IC₅₀ = 36.54 ± 0.2 μM).

Inhibition of AChE by compounds **4a–f** can be explained by forming covalent adducts with the reactive Ser200 hydroxyl group, which prevents ACh hydrolysis. In fact, the serine hydroxyl group, deactivated by the phosphorylated moiety of varying degrees that depends on the groups

attached to the phosphorus atom (Ahmed *et al.*, 2013; Gonçalves *et al.*, 2011a; Bajqar 2004;), is no longer able to participate in the hydrolysis of ACh. The anti-AChE potential of phosphonate compounds **4a–f** can be explained by the possible intervention of the serine from the AChE via its hydroxy group as indicated in Scheme 6.

This proposed mechanism in addition to the probable steric hindrance due to the ethyl group (R₂ = Et) in the phosphonate moiety, compared to that induced by the methyl group, may explain the relatively high activity of the compounds **4a,c,e** (R₂ = Me).

On the other hand, and according to some previous studies (Alipour *et al.*, 2012; Razavi *et al.*, 2013), we can explain the inhibition activity of coumarin derivatives **5a–c**, by formation of an additional π–π interaction between the coumarin moiety and some active sites of the AChE resulting more stability of the formed ligand which prevents more hydrolysis of ACh.

All these findings allowed to note that the covalent bond formation between the Ser200 hydroxyl group of AChE and the phosphorus atom in compounds **4a–f** is more efficient than the π–π interaction between AChE and the coumarin system in derivatives **5a–c**. However, the nature of coumarin attached to the triazole may also explain this difference in activity.

Conclusion

In conclusion, this work reports the synthesis of new hybrid compounds types pyrazolo[4,3-*e*]-1,2,4-triazolo[1,5-*c*]pyrimidines phosphonate **4** and coumarinic pyrazolo[4,3-

e]-1,2,4-triazolo[1,5-*c*]pyrimidines **5** via Michaelis–Arbuzov rearrangement (Arbuzov reaction) of pyrazolotriazolopyrimidines chloride **3a–c**, with trialkyl phosphate and Knoevenagel reaction of 2-cyanomethyl derivatives **3d–f** with salicylic aldehyde, respectively. Precursors **3** were obtained in two steps starting from aminopyrazole **1**. The anti-acetylcholinesterase activity of the synthesized hybrid compounds **4** and **5** was evaluated. It has been found that the methylated phosphonates **4a,c,e** were the most active ones ($IC_{50} = 1.73–6.50 \mu M$).

Finally, we can say that the prepared derivatives **4** and **5** are considered good acetylcholinesterase inhibitors; this encouraged us to pursue this study in order to more explain the preliminary structure–activity relationship (SAR) of these compounds.

Experimental section

Chemistry

All reactions were monitored by TLC using aluminium sheets of SDS silica gel 60 F₂₅₄, 0.2 mm. Melting temperatures were determined on an electrothermal 9002 apparatus and were reported uncorrected. NMR spectra were recorded on a Bruker AC-300 spectrometer at 300 MHz (¹H), 75 MHz (¹³C) and 120 MHz (³¹P). All chemical shifts were reported as δ values (ppm) relative to residual non-deuterated solvent. IR spectra were recorded on FTS-6000 BIO-RAD apparatus. Mass spectra were obtained with ESI-TOF (LCT, Waters) using the reflectron mode in the positive ion mode. The starting materials **1** and **2** were prepared according to the literature (Rashad *et al.*, 2005; Al-Afaleq *et al.*, 2001).

General procedure for the synthesis of ethyl N-(4-cyano-1-phenyl-1H-pyrazol-5-yl) formimidate **2a–c**

A mixture of compound **1** (10 mmol) and triethylorthoformate (3 mL) in acetic anhydride (25 mL) was refluxed for 1 h. After cooling, the precipitated product was filtered off and washed thoroughly with ethanol and recrystallized from ethanol to yield **2** as white solid.

Ethyl N-(4-cyano-1-phenyl-1H-pyrazol-5-yl)formimidate (**2a**)

White solid; Yield: 65 %; m.p.: 120–122 °C (ethanol); IR (KBr, cm^{-1}) ν : 1620 ($-C=N$), 2228 ($-CN$); ¹H NMR (CDCl₃, 300 MHz): δ 1.34 (t, 3H, $J = 7.6$ Hz), 4.31 (q, 2H, $J = 7.6$ Hz), 7.33–7.60 (m, 5H, H_{arom}), 8.23 (s, 1H, H_4), 8.72 (s, 1H, H_7); ¹³C NMR (DMSO-*d*₆, 75 MHz): δ 13.5 (C_8), 64.3 (C_9), 82.4 (C_4), 115.3 (CN), 124.6 ($C_{2',6'}$), 127.3 ($C_{4'}$), 129.7 ($C_{3',5'}$), 138.3 ($C_{1'}$), 150.1 (C_3), 151.4 (C_5), 160.6 (C_7).

Ethyl N-(4-cyano-3-methyl-1-phenyl-1H-pyrazol-5-yl)formimidate (**2b**)

White solid; Yield: 65 %; m.p.: 120–122 °C (ethanol); IR (KBr, cm^{-1}) ν : 1625 ($-C=N$), 2225 ($-CN$); ¹H NMR (CDCl₃, 300 MHz): δ 1.35 (t, 3H, $J = 7.5$ Hz), 2.39 (s, 3H), 4.31 (q, 2H, $J = 7.5$ Hz), 7.31–7.61 (m, 5H, H_{arom}), 8.36 (s, 1H, H_7); ¹³C NMR (DMSO-*d*₆, 75 MHz): δ 13.1 (C_8), 13.9, 64.2 (C_9), 81.7 (C_4), 114.7 (CN), 123.9 ($C_{2',6'}$), 127.6 ($C_{4'}$), 128.8 ($C_{3',5'}$), 138.1 ($C_{1'}$), 150.0 (C_3), 151.7 (C_5), 160.3 (C_7).

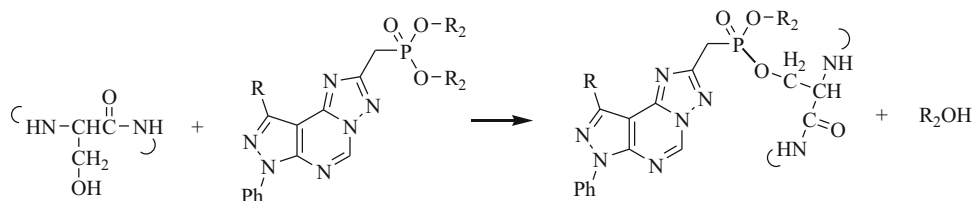
Ethyl N-(4-cyano-3-ethyl-1-phenyl-1H-pyrazol-5-yl)formimidate (**2c**)

White solid; Yield: 65 %; m.p.: 120–122 °C (ethanol); IR (KBr, cm^{-1}) ν : 1625 ($-C=N$), 2228 ($-CN$); ¹H NMR (CDCl₃, 300 MHz): δ 1.32 (t, 3H, $J = 7.3$ Hz), 1.34 (t, 3H, $J = 7.6$ Hz), 2.70 (q, 2H, $J = 7.3$ Hz), 4.45 (q, 2H, $J = 7.6$ Hz), 7.35–7.63 (m, 5H, H_{arom}), 8.46 (s, 1H, H_7); ¹³C NMR (DMSO-*d*₆, 75 MHz): δ 14.2 (C_9), 14.8, 64.5, 65.1 (C_8), 82.4 (C_4), 116.3 (CN), 124.7 ($C_{2',6'}$), 127.6 ($C_{4'}$), 129.5 ($C_{3',5'}$), 138.5 ($C_{1'}$), 150.3 (C_3), 151.1 (C_5), 160.2 (C_7).

General procedure for the synthesis of 2-chloromethyl-7-phenylpyrazolo[4,3-*e*]-1,2,4-triazolo[1,5-*c*]pyrimidines **3a–c**

To a solution of iminoethers **2** (1 mmol) in ethanol (30 mL), 2-chloroaceto-hydrazide hydrochloride (1.1 mmol) and

Scheme 6 Blocking-up reaction of serine hydroxyl by the phosphorylated moiety



triethylamine (1.1 mmol) were added. The reaction mixture was boiled at reflux for 4 h, and after cooling, it is poured into ice water (50 mL) to give a white solid which was crystallized from ethanol.

2-chloromethyl-7-phenylpyrazolo[4,3-e]-1,2,4-triazolo[1,5-c]pyrimidine (3a)

White solid; Yield: 76 %; m.p.: 184–186 °C (ethanol); IR (KBr, cm^{-1}) ν : 1620 ($-\text{C}=\text{N}$); ^1H NMR (CDCl_3 , 300 MHz): δ 4.85 (s, 2H, $-\text{CH}_2\text{Cl}$), 7.45–8.15 (m, 5H, H_{arom}), 8.58 (s, 1H, H_9), 9.15 (s, 1H, H_5); ^{13}C NMR ($\text{DMSO}-d_6$, 75 MHz): δ 37.8 ($-\text{CH}_2\text{Cl}$), 104.4 (C_{9a}), 122.4 ($\text{C}_{2',6'}$), 127.4 ($\text{C}_{4'}$), 129.5 ($\text{C}_{3',5'}$), 138.1 ($\text{C}_{1'}$), 138.5 (C_5), 146.5 (C_{9b}), 149.0 (C_{6a}), 149.3 (C_9), 164.6 (C_2), ES-HRMS $[\text{M} + \text{H}]^+$ calcd for $(\text{C}_{13}\text{H}_{10}\text{ClN}_6)^+$: 285.0577, found: 285.0569.

2-chloromethyl-9-methyl-7-phenylpyrazolo[4,3-e]-1,2,4-triazolo[1,5-c]pyrimidine (3b)

White solid; Yield: 82 %; m.p.: 192–194 °C (ethanol); IR (KBr, cm^{-1}) ν : 1615 ($-\text{C}=\text{N}$); ^1H NMR (CDCl_3 , 300 MHz): δ 2.91 (s, 3H, $-\text{CH}_3$), 4.83 (s, 2H, $-\text{CH}_2\text{Cl}$), 7.47–8.21 (m, 5H, H_{arom}), 9.20 (s, 1H, H_5); ^{13}C NMR ($\text{DMSO}-d_6$, 75 MHz): δ 14.1 ($-\text{CH}_3$), 37.6 ($-\text{CH}_2\text{Cl}$), 103.9 (C_{9a}), 121.9 ($\text{C}_{2',6'}$), 126.8 ($\text{C}_{3'}$), 128.9 ($\text{C}_{3',5'}$), 138.2 ($\text{C}_{1'}$), 138.8 (C_5), 145.9 (C_{9b}), 148.9 (C_{6a}), 149.1 (C_9), 165.1 (C_2), ES-HRMS $[\text{M} + \text{H}]^+$ calcd for $(\text{C}_{14}\text{H}_{12}\text{ClN}_6)^+$: 299.0734, found: 299.0740.

2-chloromethyl-9-ethyl-7-phenylpyrazolo[4,3-e]-1,2,4-triazolo[1,5-c]pyrimidine (3c)

White solid; Yield: 74 %; m.p.: 196–198 °C (ethanol); IR (KBr, cm^{-1}) ν : 1617 ($-\text{C}=\text{N}$); ^1H NMR (CDCl_3 , 300 MHz): δ 1.54 (t, 3H, $J = 7.5$ Hz), 3.24 (q, 2H, $J = 7.5$ Hz), 4.87 (s, 2H, $-\text{CH}_2\text{Cl}$), 7.36–8.10 (m, 5H, H_{arom}), 9.14 (s, 1H, H_5); ^{13}C NMR ($\text{DMSO}-d_6$, 75 MHz): δ 12.9, 22.0, 37.4 ($-\text{CH}_2\text{Cl}$), 102.5 (C_{9a}), 122.3 ($\text{C}_{2',6'}$), 127.5 ($\text{C}_{4'}$), 129.3 ($\text{C}_{3',5'}$), 138.2 ($\text{C}_{1'}$), 138.4 (C_5), 146.4 (C_{9b}), 149.1 (C_{6a}), 149.2 (C_9), 164.5 (C_2), ES-HRMS $[\text{M} + \text{H}]^+$ calcd for $(\text{C}_{15}\text{H}_{14}\text{ClN}_6)^+$: 313.0890, found: 313.0899.

General procedure for the synthesis of 7-phenylpyrazolo[3,2-e][1,2,4]triazolo[1,5-c]pyrimidine-2-acetonitriles 3d–f

A mixture of **2** (1 mmol) and cyanoacetic hydrazide (1.2 mmol) in absolute ethanol (20 mL) was refluxed for 10 h; after cooling, the precipitated product was filtered off and crystallized from ethanol to give **3d–f**.

7-Phenylpyrazolo[4,3-e]-1,2,4-triazolo[1,5-c]pyrimidine-2-acetonitrile (3d)

White solid; Yield: 68 %; m.p.: 172–174 °C (ethanol); IR (KBr, cm^{-1}) ν : 2225 ($-\text{CN}$); ^1H NMR (CDCl_3 , 300 MHz): δ 4.08 (s, 2H, $-\text{CH}_2-\text{CN}$), 7.42–8.13 (m, 5H, H_{arom}), 8.60 (s, 1H, H_9), 9.13 (s, 1H, H_5); ^{13}C NMR ($\text{DMSO}-d_6$, 75 MHz): δ 18.8 ($-\text{CH}_2-\text{CN}$), 104.4 (C_{9a}), 117.3 ($-\text{CN}$), 122.2 ($\text{C}_{2',6'}$), 127.4 ($\text{C}_{4'}$), 129.5 ($\text{C}_{3',5'}$), 138.1 ($\text{C}_{1'}$), 138.5 (C_5), 146.5 (C_{9b}), 149.0 (C_{6a}), 149.3 (C_9), 164.6 (C_2), ES-HRMS $[\text{M} + \text{H}]^+$ calcd for $(\text{C}_{14}\text{H}_{10}\text{N}_7)^+$: 276.0916, found: 276.0908.

9-Methyl-7-phenylpyrazolo[4,3-e]-1,2,4-triazolo[1,5-c]pyrimidine-2-acetonitrile (3e)

White solid; Yield: 80 %; m.p.: 176–178 °C (ethanol); IR (KBr, cm^{-1}) ν : 2228 ($-\text{CN}$); ^1H NMR (CDCl_3 , 300 MHz): δ 2.91 (s, 3H, $-\text{CH}_3$), 4.10 (s, 2H, $-\text{CH}_2-\text{CN}$), 7.47–8.21 (m, 5H, H_{arom}), 9.20 (s, 1H, H_5); ^{13}C NMR ($\text{DMSO}-d_6$, 75 MHz): δ 14.3 ($-\text{CH}_3$), 18.8 ($-\text{CH}_2-\text{CN}$), 104.2 (C_{9a}), 117.5 ($-\text{CN}$), 122.3 ($\text{C}_{2',6'}$), 126.5 ($\text{C}_{4'}$), 129.2 ($\text{C}_{3',5'}$), 138.4 ($\text{C}_{1'}$), 139.3 (C_5), 145.3 (C_{9b}), 148.3 (C_{6a}), 149.7 (C_9), 164.4 (C_2), ES-HRMS $[\text{M} + \text{H}]^+$ calcd for $(\text{C}_{15}\text{H}_{12}\text{N}_7)^+$: 290.1074, found: 290.1066.

9-Ethyl-7-phenylpyrazolo[4,3-e]-1,2,4-triazolo[1,5-c]pyrimidine-2-acetonitrile (3f)

White solid; Yield: 82 %; m.p.: 180–182 °C (ethanol); IR (KBr, cm^{-1}) ν : 2230 ($-\text{CN}$); ^1H NMR (CDCl_3 , 300 MHz): δ 1.54 (t, 3H, $J = 7.5$ Hz), 3.24 (q, 2H, $J = 7.5$ Hz), 4.10 (s, 2H, $-\text{CH}_2-\text{CN}$), 7.17–8.21 (m, 5H, H_{arom}), 9.17 (s, 1H, H_5); ^{13}C NMR ($\text{DMSO}-d_6$, 75 MHz): δ 12.9 ($-\text{CH}_2\text{CH}_3$), 18.8 ($-\text{CH}_2-\text{CN}$), 22.3 ($-\text{CH}_2\text{CH}_3$), 102.3 (C_{9a}), 116.8 ($-\text{CN}$), 122.6 ($\text{C}_{2',6'}$), 128.1 ($\text{C}_{4'}$), 129.8 ($\text{C}_{3',5'}$), 136.5 ($\text{C}_{1'}$), 137.3 (C_5), 146.7 (C_{9b}), 148.7 (C_{6a}), 150.7 (C_9), 165.7 (C_2), ES-HRMS $[\text{M} + \text{H}]^+$ calcd for $(\text{C}_{16}\text{H}_{14}\text{N}_7)^+$: 304.1563, found: 304.1557.

General procedure for the synthesis of pyrazolotriazolopyrimidine dialkyl phosphonates 4a–f

In a typical procedure, the solution of pyrazolotriazolopyrimidine **3a–c** (1 mmol) and an excess of trialkylphosphite (10 mL) was refluxed for 6 h. The reaction evolution was checked by TLC. When all the starting material was consumed, the mixture was cooled to room temperature, and then, the precipitate formed was filtered, dried and purified by silica gel column chromatography eluted with petroleum ether–ethyl acetate (7:3).

Dimethyl ((7-phenyl-7H-pyrazolo[4,3-e][1,2,4]triazolo[1,5-c]pyrimidin-2-yl)methyl)phosphonate (4a)

White solid; Yield: 88 %; mp 248–250 °C; IR (KBr, cm^{-1}) ν : 1625 (C = N), 1235 (P = O), 975 (P–O–C); ^1H NMR (300 MHz, $\text{DMSO-}d_6$): δ 3.62 (d, $^2J_{\text{P-H}} = 21.3$ Hz, 2H, H_{10}), 3.86 (d, $J_{\text{P-H}} = 12.3$ Hz, 6H, 2 P–O– CH_3), 7.48–8.09 (m, 5 $\text{H}_{\text{arom.}}$), 8.72 (s, 1H, H_9), 9.65 (s, 1H, H_5); ^{13}C NMR (75 MHz, $\text{DMSO-}d_6$): δ 27.6 (d, $^1J_{\text{P-C}} = 161.3$ Hz, C_{10}), 62.4 (d, $^2J_{\text{P-C}} = 24.3$ Hz, P–O– CH_3), 103.5 (C_{9a}), 122.6 ($\text{C}_{2',6'}$), 128.1 ($\text{C}_{4'}$), 129.7 ($\text{C}_{3',5'}$), 133.7 ($\text{C}_{1'}$), 138.4 (C_5), 140.6 (C_{9b}), 146.2 (C_{6a}), 148.1 (C_9), 161.2 (d, $^2J_{\text{P-C}} = 33.3$ Hz, C_2); ^{31}P NMR (121 MHz, CDCl_3): δ 24.6; ES-HRMS $[\text{M} + \text{H}]^+$ calcd. For ($\text{C}_{15}\text{H}_{16}\text{N}_6\text{O}_3\text{P}$) $^+$: 359.0943; found: 359.0937.

Diethyl ((7-phenyl-7H-pyrazolo[4,3-e][1,2,4]triazolo[1,5-c]pyrimidin-2-yl)methyl)phosphonate (4b)

White solid; Yield: 90 %; mp 253–255 °C; IR (KBr, cm^{-1}) ν : 1620 (C = N), 1238 (P = O), 978 (P–O–C); ^1H NMR (300 MHz, $\text{DMSO-}d_6$): δ 1.23 (t, $J_{\text{H-H}} = 7.2$ Hz, 6H, 2 P–O– $\text{CH}_2\text{--CH}_3$), 3.64 (d, $^2J_{\text{P-H}} = 21.3$ Hz, 2H, H_{10}), 4.07 (m, 4H, 2 P–O– $\text{CH}_2\text{--CH}_3$), 7.41–8.11 (m, 5 $\text{H}_{\text{arom.}}$), 8.73 (s, 1H, H_9), 9.70 (s, 1H, H_5); ^{13}C NMR (75 MHz, $\text{DMSO-}d_6$): δ 16.7 (d, $^3J_{\text{P-C}} = 23.4$ Hz, P–O– $\text{CH}_2\text{--CH}_3$), 28.1 (d, $^1J_{\text{P-C}} = 161.5$ Hz, C_{10}), 62.8 (d, $^2J_{\text{P-C}} = 24.4$ Hz, P–O– $\text{CH}_2\text{--CH}_3$), 103.9 (C_{9a}), 123.5 ($\text{C}_{2',6'}$), 128.0 ($\text{C}_{4'}$), 130.1 ($\text{C}_{3',5'}$), 134.7 ($\text{C}_{1'}$), 138.8 (C_5), 141.3 (C_{9b}), 146.4 (C_{6a}), 149.3 (C_9), 161.4 (d, $^2J_{\text{P-C}} = 33.3$ Hz, C_2); ^{31}P NMR (121 MHz, CDCl_3): δ 21.8; ES-HRMS $[\text{M} + \text{H}]^+$ calcd. For ($\text{C}_{17}\text{H}_{20}\text{N}_6\text{O}_3\text{P}$) $^+$: 387.1256; found: 387.1248.

Dimethyl ((9-methyl-7-phenyl-7H-pyrazolo[4,3-e][1,2,4]triazolo[1,5-c]pyrimidin-2-yl)methyl)phosphonate (4c)

White solid; Yield: 92 %; mp 260–262 °C; IR (KBr, cm^{-1}) ν : 1618 (C = N), 1234 (P = O), 973 (P–O–C); ^1H NMR (300 MHz, $\text{DMSO-}d_6$): δ 2.87 (s, 3H, $-\text{CH}_3$), 3.61 (d, $^2J_{\text{P-H}} = 21.4$ Hz, 2H, H_{10}), 3.84 (d, $J_{\text{P-H}} = 12.4$ Hz, 6H, 2 P–O– CH_3), 7.43–8.12 (m, 5 $\text{H}_{\text{arom.}}$), 9.63 (s, 1H, H_5); ^{13}C NMR (75 MHz, $\text{DMSO-}d_6$): δ 14.3 ($-\text{CH}_3$), 28.1 (d, $^1J_{\text{P-C}} = 161.4$ Hz, C_{10}), 62.2 (d, $^2J_{\text{P-C}} = 24.5$ Hz, P–O– CH_3), 103.8 (C_{9a}), 122.8 ($\text{C}_{2',6'}$), 128.3 ($\text{C}_{4'}$), 129.5 ($\text{C}_{3',5'}$), 133.4 ($\text{C}_{1'}$), 138.6 (C_5), 141.1 (C_{9b}), 146.3 (C_{6a}), 149.2 (C_9), 162.1 (d, $^2J_{\text{P-C}} = 33.4$ Hz, C_2); ^{31}P NMR (121 MHz, CDCl_3): δ 24.5; ES-HRMS $[\text{M} + \text{H}]^+$ calcd. For ($\text{C}_{16}\text{H}_{18}\text{N}_6\text{O}_3\text{P}$) $^+$: 373.1100; found: 373.1108.

Diethyl ((9-methyl-7-phenyl-7H-pyrazolo[4,3-e][1,2,4]triazolo[1,5-c]pyrimidin-2-yl)methyl)phosphonate (4d)

White solid; Yield: 98 %; mp 258–260 °C; IR (KBr, cm^{-1}) ν : 1622 (C = N), 1235 (P = O), 975 (P–O–C); ^1H NMR (300 MHz, $\text{DMSO-}d_6$): δ 1.24 (t, $J_{\text{H-H}} = 7.3$ Hz, 6H, 2 P–O– $\text{CH}_2\text{--CH}_3$), 2.91 (s, 3H, $-\text{CH}_3$), 3.66 (d, $^2J_{\text{P-H}} = 21.6$ Hz, 2H, H_{10}), 4.11 (m, 4H, 2 P–O– $\text{CH}_2\text{--CH}_3$), 7.44–8.09 (m, 5 $\text{H}_{\text{arom.}}$), 9.68 (s, 1H, H_5); ^{13}C NMR (75 MHz, $\text{DMSO-}d_6$): δ 14.7 ($-\text{CH}_3$), 16.5 (d, $^3J_{\text{P-C}} = 23.6$ Hz, P–O– $\text{CH}_2\text{--CH}_3$), 27.5 (d, $^1J_{\text{P-C}} = 161.5$ Hz, C_{10}), 62.7 (d, $^2J_{\text{P-C}} = 24.4$ Hz, P–O– $\text{CH}_2\text{--CH}_3$), 104.1 (C_{9a}), 122.3 ($\text{C}_{2',6'}$), 128.1 ($\text{C}_{4'}$), 129.7 ($\text{C}_{3',5'}$), 133.6 ($\text{C}_{1'}$), 138.7 (C_5), 141.0 (C_{9b}), 146.2 (C_{6a}), 148.3 (C_9), 161.6 (d, $^2J_{\text{P-C}} = 33.5$ Hz, C_2); ^{31}P NMR (121 MHz, CDCl_3): δ 22.0; ES-HRMS $[\text{M} + \text{H}]^+$ calcd. For ($\text{C}_{18}\text{H}_{22}\text{N}_6\text{O}_3\text{P}$) $^+$: 401.1413; found: 401.1406.

Diethyl ((9-methyl-7-phenyl-7H-pyrazolo[4,3-e][1,2,4]triazolo[1,5-c]pyrimidin-2-yl)methyl)phosphonate (4e)

White solid; Yield: 96 %; mp 268–270 °C; IR (KBr, cm^{-1}) ν : 1620 (C = N), 1232 (P = O), 977 (P–O–C); ^1H NMR (300 MHz, $\text{DMSO-}d_6$): δ 1.51 (t, 3H, $J = 7.5$ Hz), 3.18 (q, 2H, $J = 7.5$ Hz), 3.64 (d, $^2J_{\text{P-H}} = 21.6$ Hz, 2H, H_{10}), 3.82 (d, $J_{\text{P-H}} = 12.3$ Hz, 6H, 2 P–O– CH_3), 7.52–8.11 (m, 5 $\text{H}_{\text{arom.}}$), 9.62 (s, 1H, H_5); ^{13}C NMR (75 MHz, $\text{DMSO-}d_6$): δ 11.7, 21.7, 28.2 (d, $^1J_{\text{P-C}} = 161.3$ Hz, C_{10}), 62.2 (d, $^2J_{\text{P-C}} = 24.3$ Hz, P–O– CH_3), 103.4 (C_{9a}), 122.8 ($\text{C}_{2',6'}$), 127.9 ($\text{C}_{4'}$), 130.1 ($\text{C}_{3',5'}$), 133.2 ($\text{C}_{1'}$), 137.9 (C_5), 141.1 (C_{9b}), 146.4 (C_{6a}), 147.9 (C_9), 161.4 (d, $^2J_{\text{P-C}} = 33.3$ Hz, C_2); ^{31}P NMR (121 MHz, CDCl_3): δ 24.3; ES-HRMS $[\text{M} + \text{H}]^+$ calcd. For ($\text{C}_{17}\text{H}_{20}\text{N}_6\text{O}_3\text{P}$) $^+$: 387.1256; found: 387.1248.

Diethyl ((9-ethyl-7-phenyl-7H-pyrazolo[4,3-e][1,2,4]triazolo[1,5-c]pyrimidin-2-yl)methyl)phosphonate (4f)

White solid; Yield: 94 %; mp 266–268 °C; IR (KBr, cm^{-1}) ν : 1625 (C = N), 1230 (P = O), 975 (P–O–C); ^1H NMR (300 MHz, $\text{DMSO-}d_6$): δ 1.25 (t, $J_{\text{H-H}} = 7.3$ Hz, 6H, 2 P–O– $\text{CH}_2\text{--CH}_3$), 1.46 (t, 3H, $J = 7.5$ Hz), 3.10 (q, 2H, $J = 7.5$ Hz), 3.63 (d, $^2J_{\text{P-H}} = 21.6$ Hz, 2H, H_{10}), 4.10 (m, 4H, 2 P–O– $\text{CH}_2\text{--CH}_3$), 7.30–8.10 (m, 5 $\text{H}_{\text{arom.}}$), 9.61 (s, 1H, H_5); ^{13}C NMR (75 MHz, $\text{DMSO-}d_6$): δ 13.8, 16.6 (d, $^3J_{\text{P-C}} = 23.6$ Hz, P–O– $\text{CH}_2\text{--CH}_3$), 26.3, 28.1 (d, $^1J_{\text{P-C}} = 161.3$ Hz, C_{10}), 62.5 (d, $^2J_{\text{P-C}} = 24.3$ Hz, P–O– $\text{CH}_2\text{--CH}_3$), 103.1 (C_{9a}), 122.0 ($\text{C}_{2',6'}$), 127.5 ($\text{C}_{4'}$), 129.7 ($\text{C}_{3',5'}$),

138.5 (C_{1'}), 140.4 (C₅), 142.9 (C_{9b}), 146.5 (C_{6a}), 148.5 (C₉), 161.4 (d, $^2J_{P-C}$ = 33.3 Hz, C₂); ^{31}P NMR (121 MHz, CDCl₃): δ 23.3; ES-HRMS [M + H]⁺ calcd. For (C₁₉H₂₄N₆O₃P)⁺: 415.1569; found: 415.1563.

General procedure for the synthesis of 2-(coumarin-3''-yl)pyrazolo[4,3-e]-1,2,4-triazolo[1,5-c]pyrimidine 5a-c

A mixture of 2-cyanomethyl derivatives **3d–f** (1 mmol) and salicylic aldehyde (1 mmol) in ethanol (20 mL) was refluxed for 1 h in the presence of few drops of piperidine. The intermediate **3'd–f** obtained after filtration was reacted with a mixture of ethanol/water/concentrated hydrochloric acid (20:2:2, v:v:v) at reflux for 3 h, and after cooling to room temperature, the precipitated product was filtered off and recrystallized from ethanol to give **5a–c**.

2-(Coumarin-3''-yl)-7-phenylpyrazolo[4,3-e]-1,2,4-triazolo[1,5-c]pyrimidine (5a)

White solid; Yield: 78 %; m.p.: 272–274 °C (ethanol); IR (KBr, cm⁻¹) ν : 1740 (CO); ^1H NMR (CDCl₃, 300 MHz): δ 7.40–8.10 (m, 8H, H_{arom}), 8.21 (d, 1H, H_{arom}, J = 8.1 Hz), 8.62 (s, 1H, H₉), 9.11 (s, 1H, H₅), 9.21 (s, 1H, H_{4'}); ^{13}C NMR (DMSO-*d*₆, 75 MHz): δ 103.7 (C_{9a}), 121.7 (C_{8''}), 121.8 (C_{2',6'}), 122.7 (C_{4a''}), 125.7 (C_{6''}), 126.8 (C_{5''}), 127.1 (C_{4'}), 128.4 (C_{7''}), 129.3 (C_{3''}), 130.3 (C_{3',5'}), 137.8 (C_{1'}), 139.2 (C₅), 146.1 (C_{4''}), 147.2 (C_{9b}), 149.2 (C_{6a}), 149.5 (C₉), 150.1 (C_{8a'}), 164.2 (C₂), 165.3 (CO). ES-HRMS [M + H]⁺ calcd for (C₂₁H₁₃N₆O₂)⁺: 381.1022, found: 381.1015.

2-(Coumarin-3''-yl)-9-methyl-7-phenylpyrazolo[4,3-e]-1,2,4-triazolo[1,5-c] pyrimidine (5b)

White solid; Yield: 86 %; m.p.: 280–282 °C (ethanol); IR (KBr, cm⁻¹) ν : 1745 (CO); ^1H NMR (CDCl₃, 300 MHz): δ 2.90 (s, 3H, –CH₃), 7.51–8.12 (m, 8H, H_{arom}), 8.22 (d, 1H, H_{arom}, J = 8.1 Hz), 8.63 (s, 1H, H₉), 9.15 (s, 1H, H₅), 9.24 (s, 1H, H_{4'}); ^{13}C NMR (DMSO-*d*₆, 75 MHz): δ 14.2 (–CH₃), 104.5 (C_{9a}), 121.8 (C_{8''}), 122.8 (C_{2',6'}), 123.3 (C_{4a''}), 124.8 (C_{6''}), 126.1 (C_{5''}), 127.6 (C_{4'}), 128.7 (C_{7''}), 129.1 (C_{3''}), 129.7 (C_{3',5'}), 138.2 (C_{1'}), 139.1 (C₅), 146.7 (C_{4''}), 147.2 (C_{9b}), 149.3 (C_{6a}), 150.1 (C₉), 150.7 (C_{8a'}), 163.2 (C₂), 164.2 (CO). ES-HRMS [M + H]⁺ calcd for (C₂₂H₁₅N₆O₂)⁺: 395.1172, found: 395.1178.

2-(Coumarin-3''-yl)-9-ethyl-7-phenylpyrazolo[4,3-e]-1,2,4-triazolo[1,5-c] pyrimidine (5c)

White solid; Yield: 82 %; m.p.: 284–286 °C (ethanol); IR (KBr, cm⁻¹) ν : 1747 (CO); ^1H NMR (CDCl₃, 300 MHz): δ

1.52 (t, 3H, J = 7.5 Hz), 3.22 (q, 2H, J = 7.5 Hz), 7.43–8.13 (m, 8H, H_{arom}), 8.41 (d, 1H, H_{arom}, J = 8.1 Hz), 8.71 (s, 1H, H₉), 9.09 (s, 1H, H₅), 9.31 (s, 1H, H_{4'}); ^{13}C NMR (DMSO-*d*₆, 75 MHz): δ 12.4, 22.3, 104.8 (C_{9a}), 121.1 (C_{8''}), 122.5 (C_{2',6'}), 123.1 (C_{4a''}), 124.7 (C_{6''}), 127.1 (C_{5''}), 127.8 (C_{4'}), 128.5 (C_{7''}), 129.8 (C_{3''}), 130.3 (C_{3',5'}), 138.6 (C_{1'}), 139.3 (C₅), 146.3 (C_{4''}), 146.6 (C_{9b}), 150.2 (C_{6a}), 151.2 (C₉), 152.8 (C_{8a'}), 165.3 (C₂), 166.8 (CO). ES-HRMS [M + H]⁺ calcd for (C₂₃H₁₇N₆O₂)⁺: 409.1327, found: 409.1321.

Biological activity

Determination of AChE inhibitory activity

Cholinesterase (ChE) inhibitory activity was measured using Ellman's method (Ellman *et al.*, 1961), with modifications (Moyo *et al.*, 2010). In this study, 50 μL of 0.1 M sodium phosphate buffer (pH 8.0), 25 μL of AChE solution, 25 μL of each compound, and 125 μL of DTNB [5,50–dithiobis (2-nitrobenzoic acid)] were added in a 96-well microplate and incubated for 15 min at 25 °C. All compounds were re-suspended in the DMSO followed by dilution in the buffer so that the DMSO does not exceed 1 %. 25 μL of a solution of acetylthiocholine iodide was added, and the final mixture incubated, for 15 min, at 25 °C. The hydrolysis of acetylthiocholine iodide was monitored by the formation of the yellow 5-thio-2-nitrobenzoate anion as a result of the reaction of DTNB with thiocholines, catalysed by enzymes at a wavelength of 412 nm. The concentration of the compounds which caused 50 % inhibition of the AChE activity (IC₅₀) was calculated by nonlinear regression analysis. The percentage of inhibition was calculated from $(1 - S/E) \times 100$, where E and S were the respective enzyme activity without and with the test samples, respectively.

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