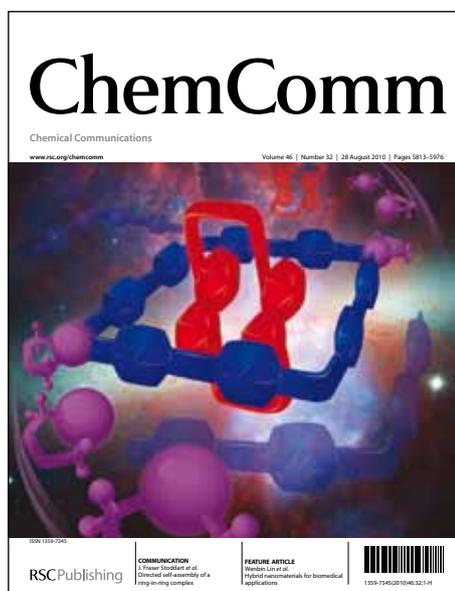


ChemComm

Accepted Manuscript

This article can be cited before page numbers have been issued, to do this please use: W. Y. Yu, W. Chan and Z. Zhou, *Chem. Commun.*, 2013, DOI: 10.1039/C3CC44769C.



This is an *Accepted Manuscript*, which has been through the RSC Publishing peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, which is prior to technical editing, formatting and proof reading. This free service from RSC Publishing allows authors to make their results available to the community, in citable form, before publication of the edited article. This *Accepted Manuscript* will be replaced by the edited and formatted *Advance Article* as soon as this is available.

To cite this manuscript please use its permanent Digital Object Identifier (DOI®), which is identical for all formats of publication.

More information about *Accepted Manuscripts* can be found in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics contained in the manuscript submitted by the author(s) which may alter content, and that the standard [Terms & Conditions](#) and the [ethical guidelines](#) that apply to the journal are still applicable. In no event shall the RSC be held responsible for any errors or omissions in these *Accepted Manuscript* manuscripts or any consequences arising from the use of any information contained in them.

Cite this: DOI: 10.1039/c0xx00000x

View Article Online

DOI: 10.1039/C3CC44769C

www.rsc.org/xxxxxx

ARTICLE TYPE

Palladium-Catalyzed Oxidative Direct C-H/C-H Cross Coupling of Anilides with β -Keto Esters

Wai-Wing Chan, Zhongyuan Zhou and Wing-Yiu Yu*

Received (in XXX, XXX) Xth XXXXXXXXXX 20XX, Accepted Xth XXXXXXXXXX 20XX

DOI: 10.1039/b000000x

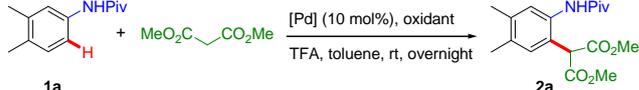
A Pd(II)-catalyzed oxidative direct C(sp²)-H/C(sp³)-H cross coupling of anilides with α -dicarbonyl compounds with Mn(OAc)₃·2H₂O as oxidant is reported and this protocol provides a facile access to α -aryl malonates and β -keto esters in good yields and regioselectivity.

Transition metal-catalyzed oxidative cross coupling reactions for C-C bond formation via twofold direct C-H activation (termed cross dehydrogenative coupling, CDC) are attracting extensive investigation.¹ With the coupling of unfunctionalized hydrocarbons, the CDC reaction is a promising approach for the design of highly atom-economical transformations. In this regard, palladium has been the catalyst of choice for many investigations. Since the pioneering works by Fujiwara and co-workers on the Pd-catalyzed oxidative aromatic C-H alkenylation,² remarkable advances have been made by many research groups on dehydrogenative C(sp²)-H/C(sp²)-H cross-coupling of (hetero)arenes with alkenes³ or arenes⁴ via Pd(II)/Pd(0) catalysis. However, the analogous examples of structurally different hydrocarbons involving C(sp²)-H/C(sp³)-H bond cross coupling are sparse in the literature.⁵

Recently, we⁶ and others⁷ reported the Pd-catalyzed oxidative arene acylations based on cross coupling of arenes with aldehydes with *tert*-butyl hydroperoxide (TBHP) as oxidant. The acylation reactions should involve two distinct C-H activation processes: (1) arene C-H palladation to form arylpalladium(II) complexes, and (2) oxyradical-mediated hydrogen atom abstraction of the aldehyde C-H bond to form acyl radicals. The subsequent coupling of the acyl radicals with the arylpalladium complexes should bring about the C-C bond formation. Inspired by these results, we envisioned that the analogous cross coupling reaction of arylpalladium(II) complex with a α -dicarbonylalkyl radical [\bullet CH(COR)₂] would lead to the formation of α -aryl dicarbonyl compounds.⁸ Herein we report a mild Pd(II)-catalyzed direct α -arylation of 1,3-dicarbonyls with anilides by cross dehydrogenative coupling reaction using Mn(OAc)₃·2H₂O as stoichiometric oxidant.

At the outset, we examined the reaction of 3,4-dimethyl-*N*-pivalanilide (**1a**, 0.2 mmol) and dimethyl malonate (6 equiv.) in the presence of Pd(OAc)₂ (10 mol%), trifluoroacetic acid (TFA) (0.5 equiv.) and TBHP (2 equiv.) as oxidant with dioxane or toluene as solvent. No significant product

formation was observed despite standing the mixture for

Table 1 Reaction Optimizations^a


Entry	Pd catalyst	Malonate (equiv.)	Oxidant (equiv.)	TFA (equiv.)	Yield ^b (%)
1 ^c	Pd(OAc) ₂	6	TBHP (2)	0.5	0
2 ^d	Pd(OAc) ₂	6	TBHP (2)	0.5	0
3	Pd(OAc) ₂	6	Mn(OAc) ₃ (1)	0.5	17
4	Pd(TFA) ₂	6	Mn(OAc) ₃ (1)	0.5	18
5	Pd(OTs) ₂ (MeCN) ₂	6	Mn(OAc) ₃ (1)	0.5	16
6	Pd(OAc) ₂	3	Mn(OAc) ₃ (1)	2	41
7 ^e	Pd(OAc) ₂	2 × 3 equiv.	[Mn] (2 × 50 mol%)	2 × 3 equiv.	72
8 ^e	Pd(OAc) ₂	2 × 3 equiv.	[Mn] (2 × 50 mol%)	2 × 3 equiv.	80
9 ^{e,f}	Pd(OAc) ₂	3 × 3 equiv.	[Mn] (3 × 50 mol%)	2 × 3 equiv.	85
10 ^e	none	3 × 3 equiv.	[Mn] (3 × 50 mol%)	2 × 3 equiv.	0

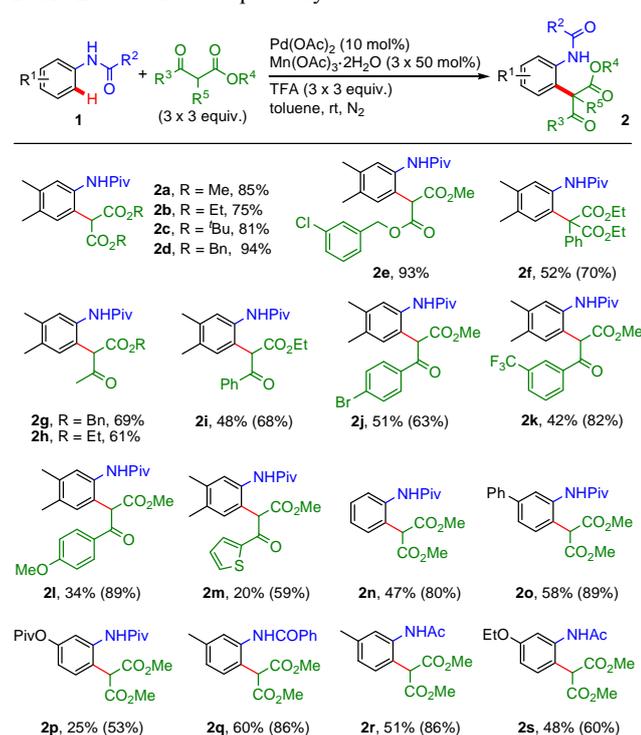
^a Reaction conditions: **1a** (0.2 mmol), dimethyl malonate, Pd catalyst (10 mol%), TBHP or Mn(OAc)₃·2H₂O as oxidant, TFA, toluene (1.5 mL) at room temperature under N₂ for overnight. ^b Yields are determined by ¹H NMR. ^c Dioxane was used as solvent. ^d 80 °C instead of rt. ^e Batchwise addition interval 4 h. ^f Isolated yield.

overnight at room temperature or at 80 °C (Table 1, entries 1 – 2). Gratifyingly, when Mn(OAc)₃·2H₂O (1 equiv.) was employed as oxidant, the “**1a** + dimethyl malonate” reaction at room temperature furnished the arylmalonate **2a** in 17% yield (entry 3). The molecular structure of a di-*tert*-butyl arylmalonate derivative **2c** has been established by X-ray crystallography (see Figure S44 in ESI). Other palladium catalysts such as Pd(TFA)₂ and Pd(OTs)₂(MeCN)₂ did not give significant improvement (entries 4 – 5). After several trials, we found that employing 2 equiv. of TFA would improve the product yield to 41% (entry 6). In order to achieve a sustained supply of the malonate radicals over the reaction time, reagents were added to the reaction mixture in a batchwise fashion. Thus, when **1a** was treated with dimethyl malonate (3 × 3 equiv./4 h), Pd(OAc)₂ (10 mol%), Mn(OAc)₃·2H₂O (3 × 50 mol%/4 h) and TFA (3 × 3 equiv./4 h) in toluene (1.5 mL) at room temperature, **2a** was obtained in 85% yield (entry 9). Notably, no **2a** formation was observed in the absence of the Pd catalyst (entry 10).

With the optimized conditions in hand, various malonates

and keto esters were examined for the Pd-catalyzed direct C-H

Table 2 Substrate scope study^{a,b}



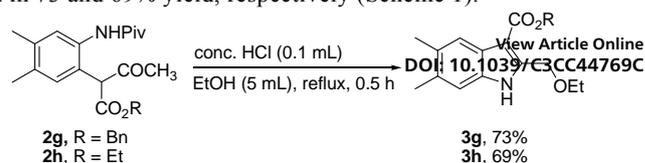
^a Reaction conditions: **1** (0.2 mmol), 1,3-dicarbonyls (3 × 3 equiv. / 4 h), Pd(OAc)₂ (10 mol%), Mn(OAc)₃·2H₂O (3 × 50 mol% / 4 h), TFA (3 × 3 equiv. / 4 h), toluene (1.5 mL) at room temperature under N₂ for overnight. ^b Isolated yields; percentage yield based on anilide conversion is given in parentheses.

arylation (Table 2). A series of dialkyl malonates such as ethyl, *tert*-butyl malonates would successfully transform **1a** to the corresponding arylmalonates **2b** and **2c** in good yields. When malonates bearing more sterically hindered benzyl group were used as coupling partner, **2d** and **2e** were obtained in 93–94% yields. Notably, diethyl phenylmalonate was also effective for the coupling reaction, and α,α -biarylmalonate **2f** bearing a quaternary carbon was obtained in 52% yield. Based on their smaller *pK*_a's comparing to the malonates,⁹ β -keto esters should be effective coupling partners. Notwithstanding, the corresponding coupling with **1a** produced **2g–2m** in 20–69% yields. The lower anilide conversion was attributed to the incompatibility of the rate of arene C-H palladation and the Mn-mediated radical generation.

The scope of the anilides was also investigated, and the Pd-catalyzed direct C-H activation was found to be sensitive to the substituent effects. For example, pivalanilides bearing other *meta*-substituents (H, Ph and OPiv) reacted with dimethyl malonate to give **2n–2p** in 25–58% yields. As anticipated, benzamido and acetamido are capable directing groups for the *ortho* C-H coupling reactions to give **2q–2s** in 48–60% yields.

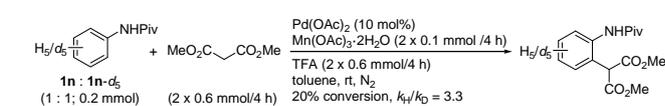
Next, we sought to apply this coupling reaction for the synthesis of indoles. For instance, treating **2g** and **2h** (0.2 mmol) with concentrated HCl (0.1 mL) and EtOH (5 mL) under reflux for 0.5 h afforded the unprotected indole **3g** and

3h in 73 and 69% yield, respectively (Scheme 1).

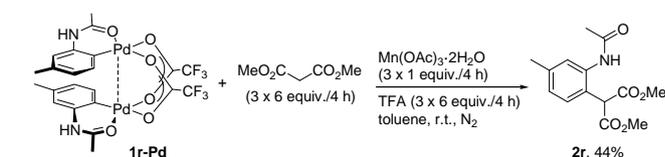


Scheme 1 Synthesis of unprotected indole **3g** and **3h**.

The Pd-catalyzed oxidative coupling reaction exhibits a primary KIE (k_H/k_D) of 3.3 based on the competitive experiments using an equimolar amount of **1n** and **1n-d₅** (Scheme 2).¹⁰ In addition, a cyclopalladated complex **Pd-1r** was prepared according to the literature.^{4b} In this work, the stoichiometric reaction of **Pd-1r** and dimethyl malonate in presence of Mn(OAc)₃·2H₂O and TFA in toluene at room temperature produced the corresponding product **2r** in 44% yield (Scheme 3). This result suggested that the Pd-catalyzed oxidative coupling reaction is likely to proceed via the cyclopalladated complex, which is formed by the rate-limiting arene C-H activation.



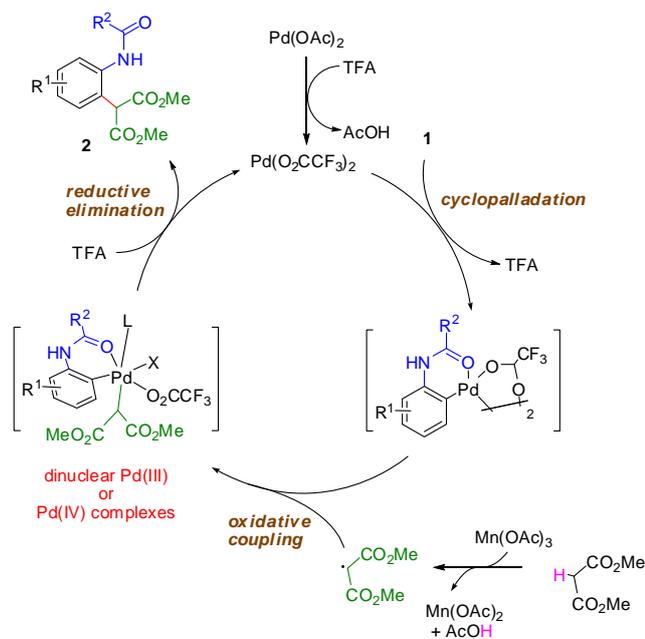
Scheme 2 Kinetic isotope effect study.



Scheme 3 Stoichiometric reaction of **Pd-1r** with dimethyl malonate.

Scheme 4 depicts a plausible mechanism for the Pd-catalyzed direct C-H α -arylation of the 1,3-dicarbonyl compounds. In the presence of an excess amount of TFA, Pd(OAc)₂ should undergo ligand exchange to form Pd(TFA)₂. Coordination of **1** with Pd(TFA)₂, followed by the rate-limiting C-H activation, afforded the six-membered palladacycle with the elimination of TFA. Since oxidation of dicarbonyl compounds by Mn(OAc)₃·2H₂O is known to generate enolate radicals,¹¹ we hypothesized that the enolate radicals would couple with the palladacycle to give some either the dinuclear Pd(III)¹² or Pd(IV)¹³ complexes. Reductive elimination of the high valent Pd species would give the desired arylmalonate.

In conclusion, we have described a Pd-catalyzed direct C-H/C-H cross coupling of 1,3-dicarbonyl compounds and anilides. With Mn(OAc)₃·2H₂O as oxidant, arylmalonates were obtained in good yields and regioselectivity. A mechanism involving the coupling of cyclopalladated complex with carboradical is proposed. The reaction can be performed under mild conditions without addition of base and provides a easy route to α -aryl carbonyl compounds. Comprehensive mechanistic studies are underway.



Notes and references

State Key Laboratory for Chirosciences and Department of Applied Biology and Chemical Technology, The Hong Kong Polytechnic University, Hung Hom, Kowloon, Hong Kong. E-mail: wing-yiu.yu@polyu.edu.hk

† Electronic Supplementary Information (ESI) available: Detailed experimental procedure, ^1H and ^{13}C NMR spectra and analytical data for all the compounds; crystallographic data for **2c**. CCDC 946973. For ESI and crystallographic in CIF and other electronic format see DOI: 10.1039/b000000x/

We acknowledge the financial support of the Hong Kong Research Grants Council (PolyU 5031/09P).

- 1 For selected reviews, see: (a) C. S. Yeung and V. M. Dong, *Chem. Rev.*, 2011, **111**, 1215; (b) C. Liu, H. Zhang, W. Shi and A. Lei, *Chem. Rev.*, 2011, **111**, 1780; (c) W. J. Yoo and C. J. Li, in *C-H activation, Topics in Current Chemistry*, ed. J. Q. Yu and Z. Shi, 2010, vol. 292, pp. 281; (d) C. J. Scheuermann, *Chem. Asian J.*, 2010, **5**, 436; (e) C. J. Li, *Acc. Chem. Res.*, 2009, **42**, 335. For reviews of Pd-catalyzed C-H/C-H coupling, see: (f) B.-J. Li and Z.-J. Shi, *Chem. Soc. Rev.*, 2012, **41**, 5588; (g) T. W. Lyons and M. S. Sanford, *Chem. Rev.*, 2010, **110**, 1147; (h) J. Le Bras and J. Muzart, *Chem. Rev.*, 2011, **111**, 1170; (i) S. H. Cho, J. Y. Kim, J. Kwak and S. Chang, *Chem. Soc. Rev.*, 2011, **40**, 5068.
- 2 I. Moritani and Y. Fujiwara, *Tetrahedron Lett.*, 1967, **8**, 1119.
- 3 For recent examples of Pd-catalyzed cross-coupling of (hetero)arenes with alkenes, see: (a) X. Cong, J. You, G. Gao and J. Lan, *Chem. Commun.*, 2013, **49**, 662; (b) P. Gandeepan and C.-H. Cheng, *J. Am. Chem. Soc.*, 2012, **134**, 5738; (c) S. R. Kandukuri and M. Oestreich, *J. Org. Chem.*, 2012, **77**, 8750; (d) M. Yu, Y. Xie, C. Xie and Y. Zhang, *Org. Lett.*, 2012, **14**, 2164; (e) Y.-Y. Liu, R.-J. Song, C.-Y. Wu, L.-B. Gong, M. Hu, Z.-Q. Wang, Y.-X. Xie and J.-H. Li, *Adv. Synth. Catal.*, 2012, **354**, 347; (f) C. Huang, B. Chattopadhyay and V. Gevorgyan, *J. Am. Chem. Soc.*, 2011, **133**, 12406; (g) A. Garcia-Rubia, B. Urones, R. G. Arrayás and J. C. Carretero, *Angew. Chem. Int. Ed.*, 2011, **50**, 10927; (h) K. M. Engle, D. H. Wang and J. Q. Yu, *Angew. Chem. Int. Ed.*, 2010, **49**, 6169.
- 4 For recent examples of Pd-catalyzed cross-coupling of (hetero)arenes with arenes, see: (a) F. Yang, F. Song, W. Li, J. Lan and J. You, *RSC Advances*, 2013, **3**, 9649; (b) M. Min and S. Hong, *Chem. Commun.*, 2012, **48**, 9613; (c) G. Wu, J. Zhou, M. Zhang, P. Hu and W. Su, *Chem. Commun.*, 2012, **48**, 8964; (d) H. Li, R.-Y. Zhu, W.-J. Shi, K.-H. He and Z.-J. Shi, *Org. Lett.*, 2012, **14**, 4850; (e) X. Wang,

- 5 D. Leow and J.-Q. Yu, *J. Am. Chem. Soc.*, 2011, **133**, 13864; (f) J. Karthikeyan and C.-H. Cheng, *Angew. Chem. Int. Ed.*, 2011, **50**, 9880; (g) C. S. Yeung, X. Zhao, N. Borduas and V. M. Dong, *Chem. Sci.*, 2010, **1**, 331; (h) X. Zhao, C. S. Yeung and V. M. Dong, *Chem. Sci.*, 2010, **1**, 331; (i) L. Ackermann, R. Vicente and A. R. Kapai, *Angew. Chem. Int. Ed.*, 2009, **48**, 9792; (j) G. Brasche, J. García-Fortanet and S. L. Buchwald, *Org. Lett.*, 2008, **10**, 2207; (k) D. R. Stuart and K. Fagnou, *Science*, 2007, **316**, 1172; (l) K. L. Hull and M. S. Sanford, *J. Am. Chem. Soc.*, 2007, **129**, 11904.
- 6 (a) L. Meng, K. Wu, C. Liu and A. Lei, *Chem. Commun.*, 2013, **49**, 5853; (b) Y. Shang, X. Jie, J. Zhou, P. Hu, S. Huang and W. Su, *Angew. Chem. Int. Ed.*, 2013, **52**, 1299; (c) C. Pierre and O. Baudoin, *Tetrahedron*, 2013, **69**, 4473; (d) S. Ghosh, S. De, B. N. Kakde, S. Bhunia, A. Adhikary and A. Bisai, *Org. Lett.*, 2012, **14**, 5864; (e) M. Leskinen, K.-T. Yip, A. Valkonen and P. M. Pihko, *J. Am. Chem. Soc.*, 2012, **134**, 5750; (f) K.-T. Yip, R. Y. Nimje, M. V. Leskinen and P. M. Pihko, *Chem. Eur. J.*, 2012, **18**, 12590; (g) M. Wasa, K. M. Engle and J.-Q. Yu, *J. Am. Chem. Soc.*, 2010, **132**, 3680; (h) B. Liégault and K. Fagnou, *Organometallics*, 2008, **27**, 4841.
- 7 (a) C.-W. Chan, Z. Zhou and W.-Y. Yu, *Adv. Synth. Catal.*, 2011, **353**, 2999; (b) C.-W. Chan, Z. Zhou, A. S. C. Chan and W.-Y. Yu, *Org. Lett.*, 2010, **12**, 3926.
- 8 (a) S. Sharma, J. Park, E. Park, A. Kim, M. Kim, J. H. Kwak, Y. H. Jung and I. S. Kim, *Adv. Synth. Catal.*, 2013, **355**, 332; (b) F. Szabó, J. Daru, D. Simkó, T. Z. Nagy, A. Stirling and Z. Novák, *Adv. Synth. Catal.*, 2012, **355**, 685; (c) C. Pan, X. Jia and J. Cheng, *Synthesis*, 2012, **44**, 677; (d) C. Li, L. Wang, P. Li and W. Zhou, *Chem. Eur. J.*, 2011, **17**, 10208; (e) Y. Wu, B. Li, F. Mao, X. Li and F. Y. Kwong, *Org. Lett.*, 2011, **13**, 3258.
- 9 For the biological importance of α -aryl carbonyl compounds, see: (a) J. Magano and J. R. Dunetz, *Chem. Rev.*, 2011, **111**, 2177; (b) F. Bellina and R. Rossi, *Chem. Rev.*, 2010, **110**, 1082; (c) C. C. C. Johansson and T. J. Colacot, *Angew. Chem. Int. Ed.*, 2010, **49**, 676; (d) G. C. Lloyd-Jones, *Angew. Chem. Int. Ed.*, 2002, **41**, 953.
- 10 The pK_a of $\text{CH}_2(\text{CO}_2\text{Et})_2$ and $\text{MeCOCH}_2\text{CO}_2\text{Et}$ in DMSO at 25 °C is 16.4 and 14.2 respectively. See: F. G. Bordwell, *Acc. Chem. Res.*, 1988, **21**, 4563.
- 11 Other Pd-catalyzed chelation-assisted C-H functionalization reactions typically exhibit primary KIE values ranging from 2.2 to 6.7, for examples, see: (a) P. Gandeepan and C.-H. Cheng, *Org. Lett.*, 2013, **15**, 2084; (b) L.-S. Zheng, K. Chen, G. Chen, B.-J. Li, S. Luo, Q.-Y. Guo, J.-B. Wei and Z.-J. Shi, *Org. Lett.*, 2013, **15**, 10; (c) K. J. Stowers and M. S. Sanford, *Org. Lett.*, 2009, **11**, 4584; (d) L. V. Desai, K. J. Stowers, M. S. Sanford, *J. Am. Chem. Soc.*, 2008, **130**, 13285; (e) X. Chen, C. E. Goodhue, J.-Q. Yu, *J. Am. Chem. Soc.*, 2006, **128**, 12634.
- 12 For reviews, see: (a) A. S. Demir and M. Emrullhuglu, *Curr. Org. Syn.*, 2007, **4**, 321; (b) B. B. Snider, *Chem. Rev.* 1996, **96**, 339; (c) B. B. Snider, *Tetrahedron*, 2009, **65**, 10738; For selected examples of Mn-mediated radical reactions, see: (d) V. Bhat, K. A. Mackey and V. H. Rawal, *Org. Lett.*, 2011, **13**, 3214; (e) T. Tsubusaki, H. Nishino, *Tetrahedron*, 2009, **65**, 9448; (f) K.-P. Chen, H.-Q. Lee, Y.-C. Cheng and C.-P. Chuang, *Org. Biomol. Chem.*, 2009, **7**, 4074; (g) J. Magolen and M. A. Kerr, *Org. Lett.*, 2006, **8**, 4561; (h) A. Citterio, D. Fancelli, C. Finzi, L. Pesce, *J. Org. Chem.*, 1989, **54**, 2713.
- 13 (a) D. C. Powers, M. A. L. Geibel, J. E. M. N. Klein and T. Ritter, *J. Am. Chem. Soc.*, 2009, **131**, 17050; (b) D. C. Powers and T. Ritter, *Nat. Chem.*, 2009, **1**, 302.
- 14 (a) N. R. Deprez and M. S. Sanford, *J. Am. Chem. Soc.* 2009, **131**, 11234; (b) J. M. Racowski, A. R. Dick and M. S. Sanford, *J. Am. Chem. Soc.* 2009, **131**, 10974; (c) S. R. Whitfield and M. S. Sanford, *J. Am. Chem. Soc.* 2007, **129**, 15142.