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The di-*t*-butylsilylene protecting group as a bridging unit in linear and macrocyclic bis-malonates for the regioselective multifunctionalization of C_{60}



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ABSTRACT

Fullerene *equatorial* bis-adducts have been prepared with high regioselectivity by a double Bingel reaction between [60]fullerene and di-*t*-butylsilylene-tethered bis-malonates. Macrocyclic bis-malonates incorporating di-*t*-butylsilylene moieties have also been prepared and used to functionalize C_{60} in multiple Bingel cyclopropanations. Fullerene bis-adducts with a *cis*-2 addition pattern and tris-adducts with an *e,e,e* addition pattern have been thus obtained. Finally, the bridging di-*t*-butylsilylene is not only a directing group for the cyclization step, it is also a protecting group that can be readily cleaved to afford the corresponding acyclic fullerene polyols.

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Fullerene chemistry has generated unprecedented stereochemical problems¹ and the regioselective polyfunctionalization of C₆₀ through multiple addition reactions remains a major challenge. Actually, mono-functionalized C₆₀ derivatives possess nine different 6-6 bonds (bonds at the junction between two six-membered rings, see Fig. 1) and mixtures of regioisomers are obtained by successive reactions at the C_{60} core.^{2,3} The synthesis of fullerene tris-adducts is even more challenging as 46 isomers differing by the relative position of the three addends on the carbon sphere are theoretically possible. The first efficient methodologies allowing the regioselective preparation of C₆₀ multi-adducts have been reported by Diederich and co-workers and are based on cyclization reactions on the fullerene scaffold.⁴ These seminal contributions have inspired most of the research efforts devoted toward the preparation of fullerene multi-adducts with a well-defined stereochemistry.^{5–8} As part of this research, we have recently developed an expeditious synthesis of fullerene tris-adducts based on a threefold Bingel reaction between C₆₀ and t-butyl(trialkoxy)silane derivatives bearing three malonate substituents.⁸ The silane unit is at the same time a directing group allowing the control of the stereochemistry of the tris-addition on the C₆₀ sphere during the cyclization step and a protecting group that can be readily cleaved to generate the corresponding acyclic fullerene derivatives. Based on this finding, we became interested in using the di-*t*-butylsilylene protecting group as a bridging unit in linear and macrocyclic bis-malonates for the regioselective bis-functionalization of C_{60} . Upon cyclization, a desilylation reaction provides the corresponding acyclic polyol derivatives. The possibility to easily open the bridging unit upon cyclization is a clear advantage to generate new fullerene building blocks with a controlled distribution of functional groups on their surfaces.

As shown in Scheme 1, the preparation of bis-malonates 2a-b was first achieved from $1a-b^{8b}$ to validate the possibility of using di-*t*-butylsilylene bridged bis-malonates for the regioselective bis-functionalization of C₆₀. Treatment of compounds 1a-b (2 equiv) with di-*t*-butylsilyl bis(trifluoromethanesulfonate) ($tBu_2Si(OTf)_2$, 1 equiv) in DMF in the presence of pyridine (pyr) gave the corresponding bis-malonates 2a-b. Fullerene derivatives 3a-b were then prepared by taking advantage of the versatile regioselective reaction developed in the group of Diederich,^{5a} which led to macrocyclic bis-adducts of C₆₀ by a cyclization reaction at the C sphere with bis-malonate derivatives in a double Bingel⁹ cyclopropanation. Reaction of 2a with C₆₀, I₂, and 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) in toluene afforded the corresponding cyclization product 3a in 23% yield. Similarly, compound 3b was obtained in 48% yield from bis-malonate 2b and C₆₀ under the same conditions.



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Figure 1. Addition of a second addend to a C_{2v} -symmetrical C_{60} mono-adduct can in principle lead to nine different regioisomeric bis-adducts. Relative to the first addend, the second one can be located either in the same hemisphere (*cis*), in the opposite hemisphere (*trans*) or on the equatorial belt (*e*). For identical addends, a second attack onto the *e*-edge or *e*-face positions leads to identical products.

Fullerene bis-adducts **3a-b** were characterized by ¹H and ¹³C NMR, UV-vis, and IR spectroscopies.¹⁰ In addition, their structure was confirmed by MALDI-TOF mass spectrometry showing the expected pseudo-molecular ion peaks at m/z 1210.2 for **3a** ([M+H]⁺, calcd for C₈₂H₃₇O₁₀Si: 1210.3) and 1237.3 for **3b** ([M+H]⁺, calcd for $C_{84}H_{41}O_{10}Si$: 1237.3). For both **3a** and **3b**, the molecular symmetry (C_1) deduced from the ¹H and ¹³C NMR spectra suggests an equatorial addition pattern. This is also consistent with molecular modeling studies (see ESI). For fullerene derivatives bearing two malonate addends, the relative position of the two cyclopropane rings on the C₆₀ core suggested by the molecular symmetry is conveniently confirmed by UV/vis spectroscopy. Effectively, the absorption spectra of C₆₀ bis-adducts are highly dependent on the addition pattern and characteristic for each of the regioisomers.⁵ The absorption spectra of bis-adducts **3a** and **3b** are similar thus showing that the relative position of the two cyclopropane rings on the fullerene core is identical for both compounds. As shown in Figure 2, the absorption spectrum recorded in CH₂Cl₂ for compound **3b** is broad in the visible region with two shoulders (398 and 410 nm) and a quite sharp band at ca. 420 nm. Indeed, the UV/vis spectrum of 3b clearly reveals the diagnostic features



Scheme 1. Reagents and conditions: (i) tBu₂Si(OTf)₂, DMF, pyr, rt, 12 h (**2a**: 43%; **2b**: 52%); (ii) C₆₀, DBU, I₂, PhMe, -15 °C, 1 h (**3a**: 23%, **3b**: 48%); (iii) BF₃·Et₂O, CH₂Cl₂, CH₃CN, rt 12 h (**4a**: 91%; **4b**: 74%).



Figure 2. Absorption spectra (CH_2Cl_2) of fullerene derivatives **3b**, **20**, **21**, and **22** showing significant differences as a function of the number of addends. In the case of bis-adducts **3b** and **21**, characteristic features of e (**3b**) and *cis-2* (**21**) bis-addition patterns are clearly observed.

previously reported for analogous *equatorial* C_{60} bis-adducts.⁵ Therefore, in addition to the C_1 symmetry deduced from their ¹H and ¹³C NMR spectra, the absorption spectra of **3a** and **3b** allowed us to definitively elucidate their stereochemistry. It is worth noting that the *equatorial* addition pattern is C_s symmetric but owing to the macrocyclic structure of compounds **3a**–**b**, the symmetry is reduced.

Finally, different reaction conditions for the cleavage of the connecting di-*t*-butylsilylene protecting group in **3a–b** were tested (Scheme 1). Reaction of **3a–b** with HF-pyridine in THF gave diols **4a–b** in good yields (80–90%). Desilylation conditions using an excess of BF₃·Et₂O (20 equiv) in CH₂Cl₂/CH₃CN were also found to be effective for the preparation of **4a–b** from **3a–b**.¹¹ From a practical point of view, the use of BF₃·Et₂O is by far more convenient as these conditions do not require HF-resistant equipment. Furthermore, traces of pyridine were particularly difficult to remove in samples of **4a–b** prepared by treatment of **3a–b** with HF.pyridine.

Following the successful preparation of *equatorial* C_{60} bis-adducts from silylene-tethered bis-malonates, a similar synthetic strategy was used for the preparation of an *equatorial* fullerene bis-adducts substituted with two symmetrically substituted malonate addends (Scheme 2). Compound **6** was obtained in an almost quantitative yield by heating 3-chloro-1-propanol (**5**) at 120 °C in the presence of 2,2-dimethyl-1,3-dioxane-4,6-dione (Meldrum's acid). Treatment of **6** (1 equiv) with an excess of 1,3-propanediol (4 equiv) under esterification conditions using *N*-(3-dimethylaminopropyl)-*N*'-ethylcarbodiimide hydrochloride (EDC) and 4-dimethylaminopyridinium tosylate (DPTS) afforded **7** in 50% yield. Subsequent treatment with $tBu_2Si(OTf)_2/pyr$ in DMF provided bis-malonate **8** in 51% yield. The C_1 -symmetrical macrocyclic



Scheme 2. Reagents and conditions: (i) Meldrum's acid, 120 °C, 3 h (99%); (ii) 1,3-propanediol, EDC, DPTS, CH_2Cl_2 , 0 °C to rt, 12 h (50%); (iii) $tBu_2Si(OTf)_2$, DMF, pyr, rt, 4 h (51%); (iv) C₆₀, DBU, I₂, PhMe, -15 °C, 1 h (40%); (v) BF₃.Et₂O, CH₂Cl₂, CH₃CN, rt 12 h (95%); (vi) TsCl, pyr, CH₂Cl₂, 0 °C to rt, 48 h (77%); (vii) *n*Bu₄NCl, THF, rt, 4 h (95%).

fullerene *e*-bis-adduct **9** was then obtained in 40% yield by reaction of **8** with C_{60} in the presence of DBU and I_2 . Desilylation with BF_{3_-} . Et₂O followed by reaction of the resulting **10** with *p*-toluenesulfonyl chloride (TsCl) in the presence of pyridine gave **11**. Finally, treatment of bis-tosylate **11** with an excess of tetra-*n*-butylammonium chloride (*n*Bu₄NCl) in THF yielded **12** in 95% yield.

The equatorial addition pattern of fullerene bis-adducts 9-12 was confirmed by both their absorption spectra and their symmetry deduced from their ¹H and ¹³C NMR spectra. Whereas derivatives 9, 10, and 11 bearing unsymmetrically substituted malonate addends are all C₁-symmetrical compounds, bis-adduct 12 with its two $C(CO_2(CH_2)_3Cl)_2$ subunits is C_s symmetric as deduced from a careful analysis of its ¹³C NMR spectrum (Fig. 3). Out of the 32 expected fullerene resonances, three are observed at δ = 70.1, 71.3, and 71.4 ppm (sp³ C atoms) and 28 between δ = 138.7 and 147.4 ppm (sp² C atoms). It is also worth noting that two of the resonances seen in the sp² region show half intensity as well as two of the three resonances observed for the fullerene sp^3 C atoms. Actually, the latter observations are unambiguous proofs for the C_s symmetry of compound **12**. Effectively, for such a fullerene derivative, there are 28 pairs of equivalent C atoms and four unique ones. Furthermore, out of the three possible C_{s} symmetrical addition patterns (*cis-2*, *e* and *trans-4*), the only one for which 3 resonances are expected for the sp³ fullerene C atoms is the equatorial bis-adduct. This is further confirmed by the observation of three resonances for the carbonyl C atoms (δ = 163.4, 163.35 and 163.25 ppm). Therefore, the preparation of compound 12 provided definitive and non-ambiguous conclusions about the stereochemistry of fullerene derivatives 9, 10, and 11. With their e fullerene addition pattern, these compounds are ideally suited for the preparation of fullerene hexa-adducts with an octahedral addition pattern.¹² This new methodology will be therefore very useful to produce new fullerene scaffolds bearing different malonate addends in a 4:2 ratio.

The concept of using cleavable di-*t*-butylsilylene-tethered bismalonates for the regioselective functionalization of C_{60} was then extended to cyclo-oligomalonates (Scheme 3). The selective mono-protection of 1,3-propanediol (**13**) was carried out by treatment with Ag₂O and *p*-methoxybenzyl chloride (PMBCl) according to the conditions reported by Bouzide and Sauvé.¹³ Treatment of the resulting mono-protected derivative **14** with $tBu_2Si(OTf)_2$ and imidazole in DMF provided **15** in 68% yield. The PMB protecting groups in **15** were conveniently removed by treatment with



Figure 3. ¹³C NMR spectrum (CDCl₃, 100 MHz) of compound **12**. Inset: detailed views showing the resonances of the carbonyl groups (left), the fullerene sp² C atoms (center) and the fullerene sp³ C atoms (right); * indicates the two sp² fullerene C atoms showing half intensity signals.

2,3-dichloro-5,6-dicyanobenzoquinone (DDQ) in CH₂Cl₂ containing a small amount of water. The di-t-butylsilylene remained unchanged and key building block 16 was thus obtained in good yields (86%). Reaction of diol 16 with malonyl chloride in the presence of 4-dimethylaminopyridine (DMAP) under pseudo high dilution conditions afforded a mixture of cyclooligomers and polymers. Macrocycles 17, 18, and 19 were isolated in a pure form in 24%, 15%, and 3% yield, respectively, by repeated chromatographic separations.¹⁴ While the ¹H and ¹³C NMR spectra of **17**, **18**, and **19** were in full agreement with their cyclooligomeric structures, it was not possible to conclude about their ring-size based on the NMR data. In the case of **18** and **19**, the proposed structures were confirmed by MALDI-TOF mass spectrometry. For the smallest member, high level of fragmentation prevented the detection of the molecular ion peak. However, based on the elution order of the compounds and on their relative yields, compound **17** is likely the cyclomonomeric derivative. To gain further evidence about the proposed structure for 17, a Bingel reaction was performed to generate the C_{2v} symmetrical fullerene mono-adduct **20**. The structure of **20** was confirmed by MALDI-TOF mass spectrometry as well as by ¹H and ¹³C NMR, IR, and UV/vis (Fig. 2) spectroscopies thus showing that compound **17** was effectively the cyclomonomalonate.

Reaction of the 28-membered ring macrocyclic bis-malonate 18 with C_{60} in the presence of iodine and DBU gave the *cis*-2 C_{60}

bis-adduct **21** with an excellent regioselectivity.¹⁵ Indeed, the only isolable by-product was a bis-fullerene derivative resulting from the reaction of **18** with two molecules of C_{60} but no regioisomeric bis-adducts of **21** could be isolated from the reaction mixture. The ¹³C NMR spectrum of compound **21** is shown in Figure 4. As discussed in the case of 12, the presence of half intensity signals for some of the resonances of the fullerene C atoms seen in the sp² region is an unambiguous signature for a C_s symmetrical structure. In addition, the presence of only two resonances for the fullerene sp³ C atoms suggests that this bis-adduct 21 could be either a cis-2 or a trans-4 regioisomer. The UV-vis spectrum of 21 is indeed fully consistent with a cis-2 C₆₀ bis-adduct (Fig. 2). The stereochemistry of compound 21 was thus elucidated. To further support this structural assignment, computational studies were also performed. The molecular geometry of the various regioisomers (*cis-2*, *cis-3*, e. trans-3 and trans-4) was optimized with Spartan'10 Macintosh Parallel Edition (Wavefunction Inc., USA) at the AM1 semi-empirical level. The cis-2 isomer was effectively found to be noticeably lower in energy when compared to all the other regioisomers $(\Delta E > 75 \text{ kJ/mol}).$

The reaction of the cyclo-tris-malonate **19** with C_{60} was also carried out. The reaction was highly regioselective and only the fullerene tris-adduct with an *e,e,e*-addition pattern was thus obtained. Compound **22** was isolated with a remarkable 61% yield. The relative



Scheme 3. Reagents and conditions: (i) PMBCl, Ag₂O, CH₂Cl₂, 48 h (80%); (ii) tBu₂Si(OTf)₂, DMF, imidazole, rt, 12 h (68%); (iii) DDQ, CH₂Cl₂, H₂O, rt, 6 h (86%); (iv) malonyl chloride, DMAP, CH₂Cl₂, rt, 48 h (17: 24%; 18: 15%; 19: 3%); (v) C₆₀, DBU, I₂, PhMe, rt, 1 h (20: 46%; 21: 27%; 22: 61%).



Figure 4. (A) ¹³C NMR spectrum (CDCl₃, 100 MHz) of compound **21** (\blacklozenge : cyclohexane); inset: detailed view showing the resonances of the fullerene sp² C atoms; * indicates the sp² fullerene C atoms showing half intensity signals. (B) ¹³C NMR spectrum (CDCl₃, 100 MHz) of compound **22** showing its threefold symmetry (\blacklozenge : cyclohexane); inset: detailed view showing the resonances of the fullerene sp² C atoms.

position of the three cyclopropane rings on the C_{60} core were determined based on the symmetry (C_3) deduced from the ¹H and ¹³C NMR spectra. ¹⁶ As shown in Figure 4, the number of resonances observed in the ¹³C NMR spectrum of **22** is fully consistent with a three-fold symmetrical compound. The cherry-red color and the UV/vis spectrum of **22** (Fig. 2) are also in complete agreement with the proposed *e,e,e*-addition pattern for compound **22**.⁸

As shown in Scheme 4, the preparation of macrocylic oligo-malonates was also carried out from 1,4-butanediol (**23**) by following the synthetic route developed for the preparation of compounds **17–19**. Mono-protection of **23** by treatment with Ag₂O/PMBCI followed by silylation ($tBu_2Si(OTf)_2$ /imidazole) and deprotection (DDQ) afforded diol **26**. Reaction of **26** with malonyl chloride in the presence of DMAP gave the cyclomonomalonate **27** as the main product. Its structure was confirmed by MALDI-TOF mass spectrometry. Higher cyclooligomers were also obtained but in very low yields. This result prompted us to develop a stepwise synthetic route that will prevent the formation of cyclomonomalonates (Scheme 4). Treatment of **25** with 1 equiv of DDQ provided the mono-protected derivative **28** in 47% yield. Subsequent reaction with malonyl chloride in the presence of DMAP gave malonate **29**. The choice of the appropriate protecting groups for the two terminal alcohol functions of **29** was the key to this synthesis. The deprotection conditions must not be acidic to preserve the bridging di-*t*-butylsilylene groups and may not be basic to preserve the ester functions.¹⁷ The PMB protecting groups being removed under



Scheme 4. Reagents and conditions: (i) PMBCl, Ag₂O, CH₂Cl₂, 48 h (95%); (ii) tBu₂Si(OTf)₂, DMF, imidazole, rt, 12 h (60%); (iii) DDQ, CH₂Cl₂, H₂O, rt, 6 h (91%); (iv) malonyl chloride, DMAP, CH₂Cl₂, rt, 48 h (51%); (v) DDQ (1 equiv), CH₂Cl₂, H₂O, rt, 6 h (from **25**: 47%); (vi) malonyl chloride, DMAP, CH₂Cl₂, rt, 12 h (67%); (vii) DDQ, CH₂Cl₂, rt, 2 h (68%); (viii) malonyl chloride, DMAP, CH₂Cl₂, rt, 48 h (51%); (v) C₆₀, DBU, I₂, PhMe, rt, 1 h (39%); (x) BF₃·Et₂O, CH₂Cl₂, CH₃CN, rt, 12 h (88%).

neutral conditions by treatment with DDQ, both the ester functions and the di-*t*-butylsilylene groups remained effectively intact and diol **30** was obtained in a good yield (68%). Reaction of **30** with malonyl chloride under pseudo high dilution conditions gave cyclobismalonate **31** in 51% yield. Macrocycle **31**, which has a ring size of 32 atoms, was characterized by NMR spectroscopy and mass spectrometry.

Reaction of C_{60} with macrocyclic bis-malonate **31**, I₂, and DBU afforded **32** in 39% yield. As discussed for **21**, the relative position of the two cyclopropane rings on the fullerene core was determined based on the C_s molecular symmetry deduced from the NMR spectra¹⁵ and on the diagnostic features of a *cis*-2 bis-addition pattern seen in the absorption spectrum of **32**. Finally, treatment of compound **32** with an excess of BF₃.Et₂O gave the C_s -symmetrical fullerene *cis*-2 bis-adduct **33** in 88% yield. With its four alcohol functions, compound **33** is a valuable building block for further chemical modifications.

In conclusion, the reaction of linear and macrocyclic di-*t*-butylsilylene-tethered bis-malonate with C_{60} gave access to fullerene bis-adducts with an excellent regioselectivity. The bridging di-*t*butylsilylene is not only a directing group during the cyclization step, it is also a protecting group that can be readily cleaved to afford the corresponding acyclic fullerene polyols. By systematically changing the length and the rigidity of the spacer units linking the malonate subunits to the di-*t*-butylsilylene moieties, fullerene bisadducts with different addition patterns should become easily accessible thus opening new perspectives in the chemistry of multifunctionalized fullerene derivatives.

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Supplementary data

Supplementary data associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.tetlet.2013.09. 024.

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- DBU (5 equiv) was added to a solution of C_{60} (1 equiv), appropriate bis-10 malonate (**2a** or **2b**; 1 equiv) and I_2 (2.5 equiv) in toluene (2 mL/mg of C₆₀) at -15 °C. After 1 h at -15 °C, the resulting mixture was directly filtered on a plug of SiO_2 (first eluted with toluene to eliminate the unreacted C_{60} then with CH₂Cl₂) and concentrated. Column chromatography (SiO₂, CH₂Cl₂/cyclohexane 3:2) gave the corresponding fullerene bis-adduct (3a or 3b). Compound 3a (59 mg, 23%): dark brown solid: IR (neat): 1744 (C=O): UV/vis (CH₂Cl₂): 310 (5h, 34800), 368 (sh, 9900), 398 (sh, 3900), 410 (sh, 2500), 421 (2400), 479 (2700); ¹H NMR (300 MHz, CDCl₃): 4.77 (m, 1H), 4.58–3.99 (m, 10H), 3.89 (m, 1H), 1.44 (t, *J* = 7 Hz, 6H), 1.09 (s, 9H) 0.94 (s, 9H). ¹³C NMR (75 MHz, CDCl₃): 164.0, 163.6, 163.5, 163.1, 148.7, 147.4, 146.6, 146.5, 146.4, 146.3, 146.2, 145.8, 145.6, 145.5, 145.4, 145.3, 145.1, 145.0, 144.9, 144.8, 144.7, 144.6, 144.4, 144.2, 144.1, 143.9, 143.8, 143.7, 143.6, 143.5, 143.2, 143.1, 142.2, 142.1, 142.0, 141.8, 141.6, 141.5, 140.6, 140.4, 140.0, 139.8, 139.0, 138.6, 71.1, 70.4, 68.1, 66.7, 63.6, 63.5, 62.4, 60.8, 53.8, 51.2, 28.1, 27.7, 21.8, 21.4, 14.3; MALDI-TOF-MS: 1210.2 $(100\%, [M+H]^{+}, calcd for C_{82}H_{37}O_{10}Si: 1210.3), 1164.7 (21\%, [M-OEt]^{+}, calcd for C_{80}H_{32}O_{9}Si: 1165.2). Compound$ **3b**(332 mg, 48%): dark brown solid; IR (neat):1751 (C=O); UV/vis (CH₂Cl₂): 310 (sh, 38670), 361 (sh, 13450), 398 (sh, 3850), 409 (sh, 2410), 421 (2260), 479 (sh, 2740); ¹H NMR (300 MHz, CD₂Cl₂): 4.81-4.71 (m, 2H), 4.57–4.36 (m, 6H), 3.98–3.84 (m, 4H), 2.06–1.96 (m, 4H), 1.44 (t, J = 7 Hz, 6H), 1.06 (s, 9H), 0.94 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): 164.1, 163.6, 163.3 (2C), 148.8, 147.3, 147.0, 146.5, 146.4, 146.2, 146.1, 145.8, 145.5, 145.4, 145.3, 145.2, 145.1, 145.0, 144.9, 144.8, 144.7, 144.6, 144.5, 144.3, 144.1, 143.9, 142.7, 144.6, 144.5, 144.3, 144.1, 143.9, 144.7, 144.6, 144.5, 144.3, 144.1, 143.9, 144.7, 144.6, 144.5, 144.3, 144.1, 143.9, 144.7, 144.6, 144.5, 144.3, 144.1, 143.9, 144.7, 144.6, 144.5, 144.3, 144.1, 143.9, 144.7, 144.6, 144.5, 144.3, 144.1, 143.9, 144.7, 144.6, 144.5, 144.3, 144.1, 143.9, 144.7, 144.6, 144.5, 144.3, 144.1, 143.9, 144.7, 144.6, 144.5, 144.3, 144.1, 143.9, 144.7, 144.6, 144.5, 144.5, 144.7, 144.6, 144.5, 144.7, 144.6, 144.5, 144.7, 144.6, 144.5, 144.7, 144.6, 144.5, 144.7, 144.6, 144.5, 144.7, 144.6, 144.5, 144.7, 144.6, 144.5, 144.7, 144.6, 144.5, 144.7, 144.6, 144.5, 144.7, 144.6, 144.5, 144.7, 144.6, 144.5, 144.7, 144.6, 144.5, 144.5, 144.7, 144.6, 144.5, 144.5, 144.7, 144.6, 144.5, 144.5, 144.7, 144.6, 144.5, 144. 143.7, 143.6, 143.5, 143.2, 143.1, 142.5, 142.0, 141.9, 141.8, 141.7, 141.5, 141.2, 140.5, 139.7, 138.8, 138.6, 71.7, 70.5, 64.2, 63.5, 63.4, 63.3, 60.4, 59.8, 54.0, 51.6, 32.2, 31.7, 27.8, 27.0, 21.3, 14.3; MALDI-TOF-MS: 1237.3 ([M+H]⁺, calcd for C₈₄H₄₁O₁₀Si: 1237.3).
- 11. BF₃·Et₂O (20 equiv) was added to a solution of the appropriate fullerene derivative (3a-b; 1 equiv) in CH₂Cl₂/CH₃CN (2:1) at room temperature. The

resulting mixture was stirred overnight at room temperature. A saturated NaHCO₃ aqueous solution was added and the aqueous layer extracted with CH₂Cl₂. The organic layer was washed with water, dried (MgSO₄) and concentrated. Column chromatography (SiO₂, CH₂Cl₂/MeOH 96:4) gave **4a-b**. Compound **4a** (36 mg, 91%): brown solid; IR (neat): 3443 (br, OH), 1737 (C=O); UV/vis (CH₂Cl₂): 251 (20040), 309 (sh, 7600), 399 (1300), 410 (sh, 830), 422 (790), 479 (980); ¹H NMR (400 MHz, CDCl₃): 4.60–4.49 (m, 8H), 3.97 (m, 4H), 2.06 (br s, 1H), 2.09 (br s, 1H), 1.47 (t, *J* = 7 Hz, 6H). MALDI-TOF-MS: 1091.5 (3450 (br, OH), 1730 (C=O); UV/vis (CH₂O₁₀: 1068.1). Compound **4b** (61 mg, 74%): brown solid; IR (neat): 3450 (br, OH), 1740 (C=O). UV/vis (CH₂Cl₂): 306 (sh, 21900), 331 (sh, 7400), 397 (sh, 2350), 409 (sh, 1550), 421 (1450), 478 (sh, 1600). ¹H NMR (400 MHz, CDCl₃): 4.61–4.48 (m, 8H), 3.80 (m, 4H), 2.04 (m, 4H), 1.94 (br s, 2H), 1.46 (t, *J* = 7 Hz, 3H); MALDI-TOF-MS: 1096.1 ([M]⁺, calcd for C₇₆H₂₄O₁₀: 1096.1).

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- 14. A solution of DMAP (1.42 g, 11.60 mmol) in anhydrous CH₂Cl₂ (100 mL) was added dropwise within 3 h to a stirred solution of **16** (1.56 g, 5.30 mmol) and malonyl chloride (0.57 mL, 5.83 mmol) in anhydrous CH₂Cl₂ (480 mL). After 18 h, the mixture was filtered on a plug of SiO₂ (CH₂Cl₂) and evaporated. Column chromatography (SiO₂, CH₂Cl₂/cyclohexane 80:20 to CH₂Cl₂) gave **17** (450 mg, 24%), **18** (290 mg, 15%) and **19** (60 mg, 3%). Compound **17**: colorless oil; IR (neat): 1735 (C=O); ¹H NMR (400 MHz, CDCl₃): 4.29 (t, *J* = 6 Hz, 4H), 3.86 (t, *J* = 6 Hz, 4H), 3.37 (s, 2H), 1.91 (m, 4H), 1.01 (s, 18H); ¹³C NMR (100 MHz, CDCl₃): 166.6, 62.4, 60.0, 41.6, 31.7, 27.8, 20.8. Compound **18**: colorless oil; IR (neat): 1736 (C=O); ¹H NMR (400 MHz, CDCl₃): 4.29 (t, *J* = 6 Hz, 8H), 3.37 (s, 4H), 1.89 (m, 8H), 1.00 (s, 36H); ¹³C NMR (100 MHz, CDCl₃): 166.6, 62.4, 60.0, 41.6, 31.7, 27.8, 21.1; MALDI-TOF-MS: 720.9 ([M]^{*}, calcd for C₃₄He₄O₁₂Si₂: 720.4). Compound **19**: colorless oil. ¹H NMR (400 MHz, CDCl₃): 4.22 (m, 12H), 3.84 (t, *J* = 6 Hz, 12H), 3.30 (s, 6H), 1.83 (m, 12H), 0.93 (s, 54H); ¹³C NMR (100 MHz, CDCl₃): 1.22 (m, 12H), 3.84 (t, *J* = 6 Hz, 12H), 3.30 (s, 6H), 1.83 (m, 12H), 0.93 (s, 54H); ¹³C NMR (100 MHz, CDCl₃): 4.22 (m, 12H), 3.84 (t, *J* = 6 Hz, 12H), 3.30 (s, 6H), 1.83 (m, 12H), 0.93 (s, 54H); ¹³C NMR (100 MHz, CDCl₃): 166.6, 61.5, 60.1, 41.5, 31.7, 27.8, 21.1; MALDI-TOF-MS: 1103.5 (100%, [M+Na]^{*}, calcd for C₅₁H₉₆O₁₈Si₃Na: 1103.6), 1081.5 (47%, [M+H]^{*}, calcd for C₅₁H₉₇O₁₈Si₃: 1081.6).
- 1. Solution of C₆₀ (1 equiv), and 1, equival, 1, equiv
- 16. DBU (0.15 mL, 1.00 mmol) was added to a solution of C_{60} (108 mg, 0.15 mmol), **19** (58 mg, 0.15 mmol), and I_2 (131 mg, 0.52 mmol) in toluene (400 mL). After 1 h at room temperature, the resulting mixture was filtered on a plug of SiO₂ (toluene then CH₂Cl₂) and concentrated. Column chromatography (SiO₂, CH₂Cl₂/cyclohexane 50:50 to CH₂Cl₂) gave **22** (55 mg, 61%) as a cherry-red glassy solid; IR (neat): 1751 (C=0); UV/vis (CH₂Cl₂): 252 (169115), 281 (107581), 304 (sh, 86077), 482 (6208), 564 (sh, 1819); ¹H NMR (400 MHz, CDCl₃): 4.70 (m, 3H), 4.60 (m, 3H), 4.36 (m, 6H), 3.95–3.77 (m, 12H), 1.94 (m, 12H), 0.98 (s, 36H), 0.90 (s, 36H); ¹³C NMR (100 MHz, CDCl₃): see Figure 4; MALDI-TOF-MS: 1797.0 ([M+H]⁺, calcd for C₁₁₁H₉₁O₁₈Si₃: 1796.5).
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