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Selective Vinylogous Reactivity of Carbene Intermediate in Gold-Catalyzed Alkyne Carbocyclization: Synthesis of Indenols

Cheng Zhang,^{†, ‡,§} Hongli Li,^{‡,§} Chao Pei,[‡] Lihua Qiu,[‡] Wenhao Hu,^{*,†} Xiaoguang Bao,^{*,‡} and Xinfang $Xu^{*,†,\ddagger}$

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ABSTRACT: A gold-catalyzed carbocyclization of alkynes with a pendant diazo group that is completed by reaction with a protic nucleophile for the synthesis of indenol derivatives with a tertiary center is described. Mechanistic studies and DFT calculations indicate that the transformation is initiated by a gold-promoted *5-endo-dig* carbocyclization to form the key intermediate vinyl gold-carbene, which is intercepted by an unprecedented vinylogous addition and followed by external protic nucleophile assisted protodeauration. Furthermore, in this catalytic alkyne transformation, various nucleophiles, including water, commercially available 1°, 2° and 3° alcohols, menthol, D-galactose, cholesterol, steroid, *etc.*, all perform well under these mild conditions to produce the corresponding indenol derivatives in high yields with structural diversity.

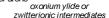
KEYWORDS: gold-catalysis, 5-endo-dig carbocyclization, diazo compound, gold carbene, indenol

INTRODUCTION

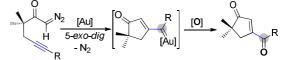
Indenols are important structural motifs found in a number of natural compounds¹ and biologically active molecules.²⁻⁴ For example, these compounds show antiproliferative activity²⁴ and antimycobacterial activity;^{2b} some are used as antitumor antibiotics,^{2c} molecular probes,³ and for the treatment of chronic obstructive pulmonary disease (COPD)^{4a,4b} and CNS disorders.^{4c} Consequently, the development of effective methods for the construction of indenol skeletons with structural diversity is an attractive subject in organic synthesis.⁵⁻⁸ Of these approaches, transition-metal-catalyzed carbocyclization of alkynes with carbonyl group has been established as a powerful synthetic method for the direct construction of indenols with a tertiary carbon center.⁶ For example, advances in this direction via C-H bond activation are reported by Cheng, Glorius,^{7b} and Zhao,^{7c} independently. However, the methods for the straightforward synthesis of the ether and ester variants, which are difficult to prepare from corresponding tertiary indenols, are quite limited.⁸ Thus, the development of efficient, diversity-oriented approaches is a useful endeavor and remains a challenge.

Recently, gold-catalyzed alkyne transformations have experienced explosive development for the effective construction of C-C and C-X bonds. Intermediates possessing carbene/carbenoid character have been postulated and verified experimentally,^{9,10} including both the α -carbonyl¹¹ and α imino¹² gold carbene species (Scheme 1a). In comparison with Rh¹³ and Cu¹⁴ catalyzed metal carbene transformations, most of the gold-catalyzed reactions of diazo compounds proceed through carbene reactivity,¹⁵ including cyclopropanation and cyclopropenation,¹⁶ X-H insertion,¹⁷ and others.¹⁸ However, (a) Gold-catalyzed alkyne transformation via carbenoid species:





(c) Gold-catalyzed carbenoid oxidation:



(d) This work: viny logous position addition



Scheme 1. Carbenoid vs Vinylogous Reactivity in Gold-Catalysis

the formation of a distinct Au=C bond has been demonstrated to have dramatic differences in both reactivity and selectivity.¹⁹ For example, X. Shi and J. Zhang have observed an exclusive electrophilic aromatic substitution of arenes with α -

diazoesters, independently.²⁰ Moreover, the ylide or zwitterionic intermediates, which were generated from gold carbenoid species with corresponding nucleophiles, have been successfully intercepted in various tandem processes (Scheme 1b), 21 such as dihydrofuran 21a and inden 21b synthesis. On the other hand, Toste has disclosed a gold-catalyzed oxidative rearrangement of alkyne-tethered diazoketone, and a vinyl gold-carbenoid intermediate was proposed in this report which was formed via a gold-catalyzed 5-exo-dig carbocyclization (Scheme 1c).²² Inspired by these milestone works on the goldcatalyzed carbenoid reactions and as the continuation of our interest in the carbene/alkyne metathesis (CAM) transformations,²³ we are intrigued by the possibility that using the nucleophilic carbon of a diazo compound addition with a goldactivated alkyne to form the vinyl gold carbene species III (Scheme 1d), instead of an N-oxide or azide to generate the α carbonyl and a-imino carbenoid intermediates, respectively (Scheme 1a). New transformations via interception of this intermediate III, especially the vinylogous reactivity that hasn't been reported in gold-carbenoid transformations,²⁴ could be envisioned.²⁵ However, preferential activation of the carboncarbon triple bond with the gold catalyst could be highly challenging because the competitive direct carbenoid formation with the coexisting diazo group shows priority according to previous reports (Scheme 1a, and 1b).¹⁵⁻²¹ Herein, we present our recent results in this direction, the gold-catalyzed 5-endodig carbocyclization of o-alkynylphenyl diazoacetates 1 occurs in the presence of a protic nucleophile, which is found to be the key factor to selectively facilitate the catalytic transformation. This unprecedented reaction not only provides a direct and efficient access for the synthesis of indenol derivatives in high to excellent yields with structural diversity, but also presents the first example of selective vinylogous reactivity in comparison with the disclosed gold carbenoid reactions. Functional groups, especially alkenyl and azide, which are described as reactive toward the metal carbenoid or in goldpromoted alkyne transformations,12 remain untouched in this work, and the synthetic utility of these products with a remaining azide group for the construction of tetracyclic frameworks is also demonstrated.

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RESULTS AND DISCUSSION

Following our initial hypothesis (Scheme 1d), a variety of gold complexes were initially evaluated for the carbocyclization reaction of 1a, which could be easily prepared from obromophenyl acetates through Sonogashira coupling and diazo transfer reactions in high yields.²⁶ It is surprising that, in andichloroethane (DCE) hydrous at 80 °C JohnPhosAu(CH₃CN)SbF₆ (3) did not catalyze any transformation of 1a, and a majority of the material was recovered after 3 hours under these conditions (Table 1, entry 1). However, when the commercially available DCE was directly used as the solvent, the indenol product 2a was isolated in 45% yield (entry 2). Water was found to have a considerable impact on the catalytic activity in this gold-catalyzed reaction, which not only acted as a reagent but was also essential to promoting the catalytic transformation. The reaction gave comparable results in the presence of 5-10 equivalents of water (entries 3 and 4) at 60-80 °C (entries 3 and 5), and the best results were obtained in the presence of 5 equivalents of water at 80 °C in DCE (entry 3, 92% isolated yield). Decreasing the catalyst

Table 1. Condition Optimization^a

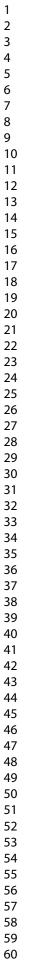
	N ₂ L COOMe cat (x mol %) H ₂ O (n equiv.) DCE, 80 °C	HO Za	,CO₂Me ≻—Ph
	OH O CO ₂ Me CO ₂ Me A Ph B Ph	CO ₂ Me	CO ₂ Me
entry	cat (x mol %)	H ₂ O (n equiv.)	yields (%) ^b 2a/A/B/C
1^d	JohnPhosAu(CH ₃ CN)SbF ₆ 3 (5.0)	-	-/-/-
2^{e}	3 (5.0)	-	45 ^{<i>f</i>} /-/-/-
3	3 (5.0)	5	92 ^{<i>f</i>} /-/-/-
4	3 (5.0)	10	85 ^{<i>f</i>} /< 5/-/-
5 ^g	3 (5.0)	5	87 ^{<i>f</i>} /-/-/-
6	3 (2.0)	5	60 ^{<i>f</i>} /-/-/-
7	JohnPhosAuCl (5.0) + AgNTf ₂ (5.0)	5	89 ^{<i>f</i>} /-/-/-
8	JohnPhosAuCl (5.0)	5	16/< 10/30 ^{<i>f</i>} /18 ^{<i>c</i>}
9	AuCl ₃ (5.0)	5	< 10/75 ^{<i>f</i>} /11/-
10	AgSbF ₆ (5.0)	5	-/32 ^f /35 ^f /-
11	Rh ₂ (OAc) ₄ (2.0)	5	-/75 ^{<i>f</i>} /11/0
12	Cu(hfacac)2:H2O (5.0)	5	-/-/30 ^f /60 ^c

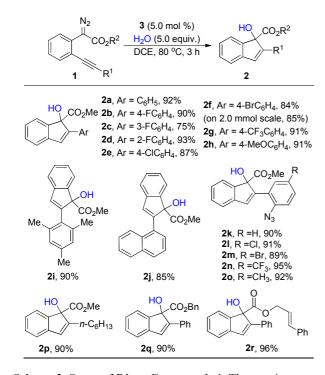
^aReaction conditions: to a solution of the catalyst in dry DCE (2.0 mL), was added the solution of **1a** (0.2 mmol) in DCE (2.0 mL) at the indicated temperature under argon atmosphere and the reaction was stirred for 3 h under these conditions. ^bYields were determined by proton NMR of the crude reaction mixture based on the yields of isolated products. ^{cl}Solated as a mixture of **C** and **C**². ^dMost of **1a** was recovered. ^cCommercially available DCE was used as solvent and partial of **1a** was reserved. ^fIsolated yields. ^gThe reaction was carried at 60 °C. DCE = dichloroethane.

loading resulted in moderate yields (entry 6). Varying the counterion of gold catalyst from NTf₂⁻ to SbF₆⁻ shown little effect (entry 7). Notably, this reaction was performed without slow addition of the diazo compound *via* syringe pump, and the observed chemoselectivity is unique for gold catalysis because no indenol product **2a** was generated in the presence of silver, dirhodium, or copper catalysts (entries 10-12). Byproducts **A**, **B**, **C**, and **C'** were generated in these cases *via* the corresponding metal carbene reaction pathway.²⁷

SCOPE

Under the optimal reaction conditions, a wide variety of 2substituted indenols have been generated with this method (Scheme 2). The impact of substituents on the alkyne motifs of **1a** was initially investigated (\mathbb{R}^1), Electron-neutral, electronwithdrawing, electron-donating, and bulky groups were all tolerated in this transformation, and provided the corresponding products in high to excellent yields (**2a-2i**, 75-93%). Moreover, comparable results were obtained when the reaction was carried out on a 2.0 mmol scale (**2f**, results in parenthesis). The naphthyl- and alkyl-substituted substrates also underwent

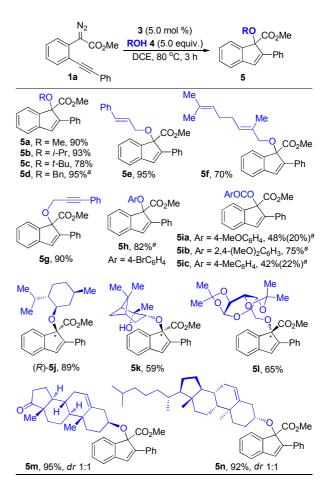




Scheme 2. Scope of Diazo Compounds 1. The reactions were carried out on a 0.2 mmol scale in DCE at 80 °C for 3 h.

the reaction smoothly, leading to indenois 2j and 2p in 85% and 90% yields, respectively. Notably, the azide group, which was reported to form the α -imino gold carbene with alkyne in the presence of gold catalyst,¹² remained untouched under these conditions, and corresponding products were isolated in > 89% yields (2k-2o). These results not only showed the preferential reactivity of diazo group over the azide, but also offered the potential synthetic utility of these products with a remaining azide group. Subsequently, the ester part of the diazoacetates was investigated. It is worth mentioning that diazoacetates bearing the groups, including benzyl and alkenyl groups which were described as reactive toward the metal carbene for C-H insertion and cyclopropanation reactions, respectively, were also well-tolerated in this reaction and afforded indenol products in > 90% yields (2q and 2r). The structures of products 2f and 2n were unambiguously confirmed by signal-crystal X-ray analysis.

Encouraged by the above promising results of this goldcatalyzed reaction in the presence of water, we envisioned that the catalytic system may also facilitate the carbocyclization of alkyne with other challenging substrates (Scheme 3). To our delight, the reaction showed broad substrate generality, including alkyl alcohol, cinnamyl alcohol, geraniol, and phenylpropargyl alcohol all promoted the reaction smoothly and provided the corresponding ether products 5a-5g with a tertiary carbon center in > 70% yields. It is worth mentioning that with the tert-butyl alcohol, the reaction provided the corresponding product 5c with an ethereal bond flanked by two quaternary centers in 78% yields. Phenol was well-tolerated under these reaction conditions, giving the cyclization product 5h in 82% yield at room temperature. Notably, benzoic acid derivatives, which are more acidic than water or alcohol, were also compatible, giving the corresponding cyclization products



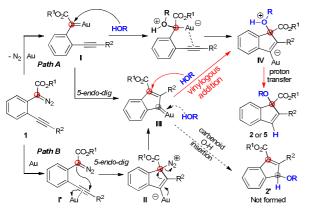
Scheme 3. Scope of Protic Nucleophiles 4. The reactions were carried out on a 0.2 mmol scale in DCE at 80 °C for 3 h. ^{*a*} The reactions were carried out at room temperature for 5 h, and the data in parentheses are yields of corresponding carbene O-H insertion products, see SI for details.

5ia-5ic in fairly good yields contaminated with the direct O-H insertion product in the cases with 5ia and 5ic. Lower reaction temperature was necessary to ensure the high yield in the cases with phenol and benzoic acids to avoid the direct O-H insertion. Furthermore, chiral protic nucleophiles, including Lmenthol, (+)-2,3-pinanediol, and even sugar derivative diacetone-D-galactose, all proceeded smoothly and selectively to afford the ether products 5j, 5k, and 5l, each as a signal enantiomer in 89%, 59%, and 65% yields, respectively. The competition experiment has been carried out in the presence of equal amount of isopropanol and tert-butanol, and only selective addition product with secondary alochol was formed (see Figure S4 in the SI for details). And the chiral alcohol itself contributed to the high stereoselectivity in this transformation. The stereochemistry of the newly formed chiral center in 5j was unambiguously determined to be R by single-crystal Xray analysis,²⁸ and the structure **51** of **51** and was tentatively assigned by the comparison of optical rotations.²⁶ It is worth mentioning that this is the first example that provided the direct access for the synthesis of optical pure indenol ethers with tertiary center.⁸ In addition, two biomolecules, а dehydroepiandrosterone and cholesterol, were also tested in

this reaction, and the indene moiety was successfully linked to these compounds and the products were obtained in > 92% yields with 1:1 dr (**5m** and **5n**). These results showed significant potential of this underexplored pattern of reactivity for the selective modification of bioactive molecules which possess a hydroxyl group, especially for the *late-stage modification* of natural products or pharmaceuticals.

MECHANISTIC DISCUSSION

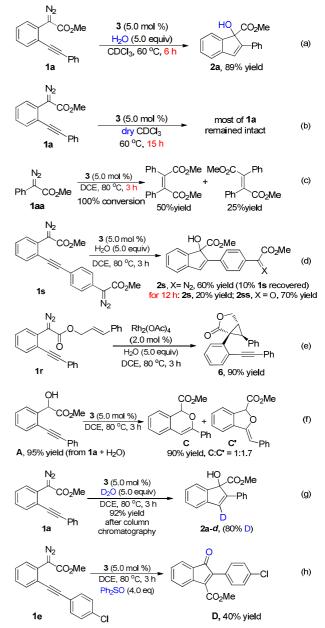
According to the previously reported gold-catalyzed transformations, two potential catalytic pathways may be involved in this reaction, including direct gold-carbene formation of the diazo group,¹⁵⁻²¹ and classic alkyne activation with the π -acidic carbophilic gold catalyst (Scheme 4, Path A and B).⁹⁻¹² To



Scheme 4. Proposed Mechanism

gain insight into the initial step, control reactions of 1a and 1aa were conducted individually in the presence of gold catalyst 3 Although 2a was isolated in 89% yield when the reaction was conducted at 60 °C in the presence of water (Scheme 5a), the majority of 1a was recovered after 15 hours at 60 °C in dry CDCl₃ (Scheme 5b and Figure S1). On the other hand, complete decomposition of 1aa occurred within 3 hours in DCE, leading to the formation of dimerized products as a Z/Emixture (Scheme 5c and Figure S2). Interestingly, the additional diazo group on the substrate 1s survives under these reaction conditions, and affords the cyclized product 2s in 60% yields with 10% of material recovered (Scheme 5d, decomposition of 2s occurs slowly under these reaction conditions to give the corresponding carbonyl product 2ss, see SI for details).26 These results suggest that the activation of the alkyne has priority in this case compared with the goldcatalyzed diazo group decomposition, and similar preferential reactivity was observed by Doyle and co-authors in goldcatalysis.²⁹ On the other hand, in the presence of Rh₂(OAc)₄ instead of gold catalyst 3, the cyclopropanation product 6 is formed from 1r in 90% yield via a corresponding metal carbene intermediate, which is distinctly different from the product formed in the gold catalysis (Scheme 5e vs 2r in Scheme 2). Furthermore, the O-H insertion product A is produced as the only product from 1a in the presence of degassed water catalyzed by $Rh_2(OAc)_4$, and the cyclization products C and C' could be obtained from A when catalyzed by 3 (Scheme 5f).²⁷ All these results lead to the conclusion that direct formation of the gold carbenoid intermediate I (Scheme 4, Path A) was not the initial step in this reaction. The deuterated product **2a**-*d* was isolated in 92% yields in the presence of D_2O after column chromatography (Scheme 5g). In addition, the carbenoid oxidation product **D** was isolated in 40% yield in the presence of diphenylsulfoxide instead of water (Scheme 5h),²² which indicated the possibility of the formation of corresponding vinyl gold carbenoid species **III** in this transformation (Scheme 4, Path B). And the remaining question is the preferential addition between the vinylogous position (Scheme 4, red arrows) to give the product **2** and the carbenoid O-H insertion to produce the regioisomer **2'** which is not detected in this transformation (dashed arrows).

To further understand the reaction mechanism, comparison experiments of **1a** with phenylpropargyl alcohol **4g** catalyzed by gold and silver separately or in combination are carried out (Scheme 6). Under the standard conditions, which means with



Scheme 5. Control Experiments

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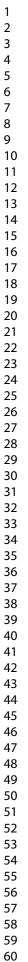
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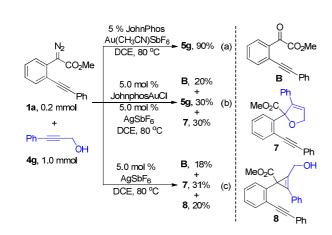
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Scheme 6. Comparison Experiments of Au- and/or Ag-Catalysis

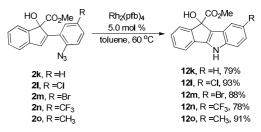
the gold catalyst **3** alone, the reaction forms the indenol derivate **5g** as the only product in 90% yield (Scheme 6a). However, by combining of gold with silver, which is supposed to form **3** *in situ*, only resulted in the formation of product **5g** in 30% yield, and contaminated with compounds **B** and **7** in 20% and 30% yields, respectively (Scheme 6b). And these two by-products are generated through the metal carbene intermediate.²¹ Additionally, when the reaction is catalyzed by AgSbF₆ alone, three identifiable products **B**, **7**, and **8** are isolated in combined high yields, instead of **5g** (Scheme 6c). These results clearly illustrate that the current reaction goes through a reaction pathway that is different from the reported transformations, which does not involve the direct formation of metal carbene intermediate at the diazo group (Scheme 4, Path A).

Density functional theory (DFT) calculations were carried out to gain a mechanistic insight on the formation of indenol derivatives.²⁶ The formation of Au(I)-carbenoid intermediate (I) could be achieved by the gold catalyzed activation of the diazo moiety of 1a and synchronously release N₂ (Figure S5, Path A).^{19,30} The predicted energy barrier for Path A is 29.6 kcal/mol relative to separated 1a and the catalyst. Alternatively, Path B leading to the cyclized Au(I)-carbenoid intermediate (III) via the alkyne activation with the π -acidic carbophilic gold catalyst was explored. The calculated ΔG^{\neq} for the cyclization step is 27.1 kcal/mol, which is 2.5 kcal/mol lower in energy than that of the direct activation of diazo moiety to form the gold carbenoid I (Figure S5 Path B). Therefore, the computational results suggest that it is more feasible for 1a to undergo the Au(I) promoted 5-endo-dig carbocyclization followed by N₂ dissociation to afford the gold carbenoid intermediate III. After the generation of III, the predicted energy barrier for the vinylogous position addition step is 19.5 kcal/mol relative to separated III and methanol (Figure S6). The proton transfer step could be facilitated by another molecule of methanol to complete the proton migration with a lower energy barrier (see Figure S6 vs S7).²

Based on the above experimental studies and DFT calculations, a possible reaction mechanism was proposed (Scheme 4, Path B). Initially, the gold-promoted 5-endo-dig carbocyclization of 1 leads to the key intermediate vinyl gold carbenoid III via II, followed by preferential vinylogous position addition and external protic nucleophile assisted protodeauration through IV to deliver the indenol derivatives 2 or 5. Moreover, the ³¹P NMR analysis results by combination of gold catalyst 3 (0.02 mmol) and 1a (0.02 mmol) in the absence of protic nucleophile suggested the formation of a relative stable complex(s) under these conditions (see Figure S3 in SI), and the reaction could not complete a catalytic cycle in the absence of the protic nucleophile under current conditions. It should be noted that other possible reaction pathway(s), such as combination of path A and B, couldn't be totally ruled out so far.

UTILITY

To demonstrate the synthetic utility of the current method, further transformations of carbocyclization products **2** were conducted. For example, **2a** was readily converted into the indanone product under basic aqueous conditions followed by treatment with HCl.²⁶ Notably, with the azide product **2k-2o**, the C-N bond formation occurred in the presence of a rhodium(II) carboxylates catalyst, and the tetracyclic fused indole products **12** precipitated out as solid and obtained after filtration in 78-93% isolated yields without further purification (Scheme 7).³¹ Other derivations, including reduction and



Scheme 7. Derivatization of Indenol Products 2

Sonogashira coupling reaction with bromo-substituted product **2f** were also conducted, and the corresponding products were obtained in 69% and 89% yields, respectively.²⁶

CONCLUSIONS

In summary, we have disclosed an unprecedented goldcatalyzed carbocyclization reaction of alkyne-tethered diazo compounds with protic nucleophiles, which provides a direct access for the synthesis of indenol derivatives under mild reaction conditions with broad substrate scope in high to excellent yields. Various protic nucleophiles, including water, commercially available alcohols, menthol, steroid, etc., are all well tolerated under these conditions to produce the corresponding indenol derivatives with structural diversity. Mechanistic studies and DFT calculations suggest that the vinyl gold carbenoid is the key intermediate in this transformation, and the following selective interception of this intermediate with protic nucleophile presents the first example of vinylogous reactivity in comparison with the disclosed gold carbenoid reactions. This underexplored pattern of reactivity shows significant untapped potential in selective modification of privileged bioactive molecules with a hydroxyl group, and its innovative applications beyond organic synthesis could be expected.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge via the Internet at http://pubs.acs.org.

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Experimental procedure, the ¹ H and ¹³C NMR spectra of all the products (PDF), crystallographic data for **2f**, **2n** and **5j** (CIF), and computational details.

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Author Contributions

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Notes

The authors declare no competing financial interest.

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