# REGIOSELECTIVE ALKYLATION *via* TRIALKYLSTANNYLATION: ME-THYL α-D-GLUCOPYRANOSIDE\*

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### ABSTRACT

Partial stannylation of methyl  $\alpha$ -D-glucopyranoside with (Bu<sub>3</sub>Sn)<sub>2</sub>O, and subsequent alkylation with benzyl bromide, allyl bromide, and trityl chloride, afforded the 2,6-disubstituted derivative as one of the major products in each case.

# INTRODUCTION

Partial stannylation of hydroxyl groups with  $(Bu_3Sn)_2O$  has been found to provide an efficient approach to the regioselective acylation of methyl hexosides<sup>2</sup>. The stannyl method was also employed for the regioselective benzylation of mannopyranosides, to afford intermediates crucial for the synthesis of branched mannooligosaccharides of biological importance<sup>3</sup>.

Partially protected glucopyranosides bearing benzyl, allyl<sup>4</sup>, or trityl groups may be regarded as versatile intermediates for the synthesis of complex, glycan chains involving glucopyranosyl residues. As part of a project on the regioselective transformation of carbohydrates via stannylation, we now report the regioselective benzylation, allylation, and tritylation of methyl  $\alpha$ -D-glucopyranoside.

# **RESULTS AND DISCUSSION**

It has been reported<sup>2</sup> that stannylation of methyl  $\alpha$ -D-glucopyranoside 1 with 1.5 molar equivalents of  $(Bu_3Sn)_2O$ , and subsequent treatment with benzoyl chloride at room temperature affords methyl 2,3,6-tri-O-benzoyl- $\alpha$ -D-glucopyranoside and methyl 2,6-di-O-benzoyl- $\alpha$ -D-glucopyranoside (19) in 18.4 and 81.4% yield, respectively. Formation of the major product (19) and the minor tribenzoate was explained as occurring through the intermediacy of the regioselectively stannylated methyl  $\alpha$ -D-glucopyranosides 2 and 6, respectively.

With the expectation of obtaining partially alkylated D-glucopyranosides, the reaction of partially stannylated methyl  $\alpha$ -D-glucopyranosides with benzyl bromide

<sup>\*</sup>Glucan Synthesis, Part I. For a preliminary communication, see ref. 1.

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Structure	Carbon atom							
	I ( <sup>1</sup> J <sub>CH</sub> )	2	3	4	5	6	Me	OCH <sub>2</sub> R(at O-n)
1 (D <sub>2</sub> O)	99.6	71.9	73.6	70.1	71.6	61.2	55.3	
8 (3,6-Bn <sub>2</sub> )	99.2 (168.0)	72.4	82.5	69.9	70.7	69.6	55.2	73.5 O-6
···-/	•		(+8.9)ª	:		(+8.4)		74.8 O-3
9 (2,6-Bn <sub>2</sub> )	97.8 (167.9)	79.2	73.1	70.0	70.7	69.4	55.1	73.6 O-6
		(+7.3)				(+8.2)		73.1 O-2
10 (4,6-Bn <sub>2</sub> )	99.2 (170.9)	72.5	75.0	77.4	70.1	68.5	55.1	73.5 O-6
				(7.3)		(7.3)		74.6 O-4
16 (3,4-Bn <sub>2</sub> )	99.1 (168.5)	72.8	82.8	77.0	70.9	61.6	55.1	75.2 O-3
			(+9.2	(+6.9)				74.7 0-4
11 (6-Bn)	100.6 (167.8)	73.2	75.1	72.1	71.5	70.7	55.2	73.6 O-6
(CD <sub>3</sub> COCD <sub>3</sub> )						(+9.5)		
26 2-Bn	97.8 (167.2)	79.3	73.0	70.5	70.6	62.2	55.3	73.0 O-2
	•	(+7.4)						
12 (3,6-All <sub>2</sub> )	99.3 (168.5)	72.1	82.0	69.9	70.5	69.4	55.1	72.4 O-6
	-		(+8.4)			(÷8.2)		73.6 O-3
13 (2,6-All <sub>2</sub> )	97.5 (167.2)	78.7	72.8	69.8	70.6	69.2	55.1	72.4 O-6
		(+6.8)				(+8.0)		71.8 O-2
14 (6-All)	99.3 (168.9)	71.7	74.2	70.3	70.3	69.2	55.1	72.4 O-6
	•					(+8.0)		
21 (3,4,6-Tr <sub>3</sub> )	95.6 (157.5)	70.1	79.1 <sup>6</sup>	69.80	71.4 <sup>b</sup>	61.5	56.9	88.9
								88.2
								85.5
24 (2,6-Tr <sub>2</sub> )	98.4 (169.7)	74.1	72.8 <sup>b</sup>	69.3 <i>°</i>	72.20	64.1	54.5	87.7
								86.9

13C CHEMICAL-SHIFTS OF PARTIALLY ALKYLATED METHYL 2-D-GLUCOPYRANOSIDES IN CDCI2

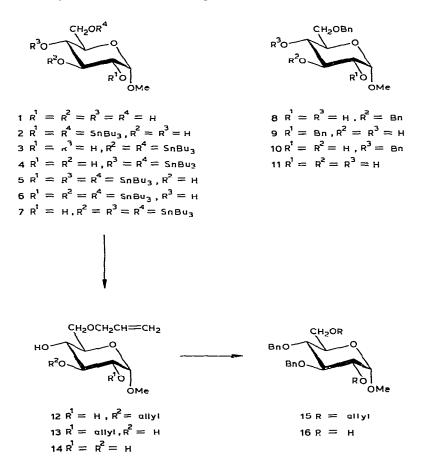
"Values in parentheses correspond to the  $\alpha$ -effect of alkylation. "The assignments are tentative.

was first examined. Stannylation of 1 with 1.5 molar equivalents of  $(Bu_3Sn)_2O$ , and subsequent reaction of the partially stannylated intermediates with benzyl bromide at 80–90°, gave the dibenzyl ethers 8, 9, and 10, and the monobenzyl ether 11, in 4.5, 30.5, 6.0, and 48.6% yield, respectively. The structures of 8, 9, 10, and 11 were assigned by <sup>13</sup>C-n.m.r. data (see Table I). Signal assignments for each carbon atom of the substituted methyl  $\alpha$ -D-glucopyranosides were based on the chemical shifts reported<sup>5</sup> for the corresponding carbon atoms of 1. The signals for methylene carbon atoms of the benzyl groups were distinguished by proton, off-resonance decoupling. Carbon atoms bearing a benzyloxy group appeared at lower field due to the  $\alpha$ -effect of alkylation<sup>5</sup>; signals for C-6 and C-3 of 8 appeared at  $\delta$  69.6 and 82.5, for C-6 and C-2 of 9 at  $\delta$  69.4 and 79.2, for C-6 and C-4 of 10 at  $\delta$  68.5 and 77.4, and for C-6 of 11 at  $\delta$  70.7. These di-O-benzyl and mono-O-benzyl derivatives had been prepared, in the conventional way, by Küster and Dyong<sup>6</sup> in 1975.

The isolation of the three dibenzyl ethers 8, 9, and 10 in the ratios of 3:20:4, together with the 6-O-monobenzyl derivative 11, showed the occurrence of only

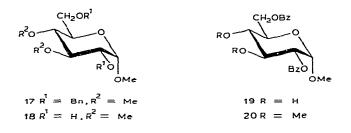
TABLE I

moderate regioselectivity for benzylation, as compared with benzoylation, of partially stannylated intermediates, and indicated the presence of an equilibrium among such partially stannylated intermediates as 2, 3, 4, 5, 6, and 7 under the reaction conditions employed. The intermediates 2, 5, and 6 may lead to formation of the major product 9, whereas 3, 6, and 7, and 4, 5, and 7, may lead to production of the two minor products, 8 and 10, respectively.

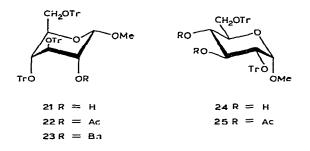


The reaction of partially stannylated intermediates with allyl bromide led to isolation of the diallyl ethers 12 and 13 in 8.0 and 41.8% yield, respectively, along with the monoallyl ether 14 in 23% yield. The structures of 12 and 13 were also assigned from <sup>13</sup>C-n.m.r. data: signals for C-6 and C-3 of 12 appeared at  $\delta$  69.4 and 82.0, whereas those for C-6 and C-2 of 13 appeared at  $\delta$  69.2 and 78.7. The  $\beta$ -effect of alkylation was observed for the signal of C-1 of 13, which was shielded by 2.1 p.p.m. compared with that of 1, confirming the introduction of an allyl group on O-2 of 13. The structure of 14 was assigned by the presence of a deshielded signal for C-6 at  $\delta$  69.2 in the <sup>13</sup>C-n.m.r. spectrum.

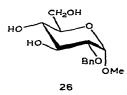
The major product, 13, was transformed into the dibenzyl ether 16 in two steps, namely, benzylation of 13 with NaH-benzyl bromide, and deallylation<sup>7</sup> with 10% Pd-C in EtOH-AcOH-H<sub>2</sub>O, in 62% overall yield. The <sup>13</sup>C-n.m.r. data for 16 confirmed the structure assigned, by revealing two deshielded signals, at  $\delta$  82.8 and 77.0, for C-3 and C-4, respectively.



The structure of the 2,6-dibenzyl ether 9 was confirmed by chemical correlation with authentic methyl 2,6-dibenzoate 19. Thus, 9 was treated with MeI–NaH to give 17 in 91.4% yield, and 19 was treated with  $CH_2N_2$  in  $CH_2Cl_2$  in the presence of  $BF_3$ -ether<sup>8</sup>, to give dimethyl ether 20 in 64.3% yield. Hydrogenolysis of 17, and *O*-debenzoylation of 20, afforded the same methyl 3,4-di-*O*-methyl- $\alpha$ -D-glucopyranoside (18).



Partial tritylation of 1 via stannylated intermediates was also examined. On treatment of partially stannylated 1 with trityl chloride at 65°, the 3,4,6-tritrityl (21) and the 2,6-ditrityl ether (24) were isolated in 37.3 and 53.6% yield, respectively. The structure of 21 was assigned by chemical transformation into the monobenzyl ether 26 via benzylation and hydrolysis. The structure of 26 was determined from the <sup>13</sup>C-n.m.r. spectrum, which contained a deshielded signal at  $\delta$  79.3 for C-2, and a shielded signal at  $\delta$  97.8 for C-1 due to the  $\beta$ -effect of alkylation at O-2. The <sup>13</sup>C-n.m.r.



spectrum of the 3,4,6-tritrityl ether 21 contained three signals for methine carbon atoms of triphenylmethyl groups, at  $\delta$  88.9, 88.2, and 85.5, which confirmed the introduction of three trityl groups into 1. The signal for the anomeric carbon atom appeared at  $\delta$  95.6 with  ${}^{1}J_{CH}$  157.5 Hz, indicating that the anomeric hydrogen atom was axially oriented. The <sup>1</sup>H-n.m.r. spectrum of 21 also revealed a deshielded signal for the axially oriented H-1, at  $\delta$  5.02, with w<sub>hh</sub> 3.5 Hz, compared with the one for the equatorially oriented H-1 of 24, which appeared at  $\delta$  3.88. These chemical and spectral data proved that 21 was the 3,4,6-tri-O-trityl derivative of 1, and that the favored conformation in CDCl<sub>3</sub> is  ${}^{1}C_{4}$ . The  ${}^{1}J_{CH}$  values of the signals for the anomeric carbon atoms of the acetate 22 ( $\delta$  93.8) and the benzyl ether 23 ( $\delta$  95.7) were 159.9 and 156.3 Hz, respectively, in agreement with their  ${}^{1}C_{4}$  conformation.

The <sup>13</sup>C-n.m.r. spectrum of the major product **24** contained two methine signals, at  $\delta$  87.7 and 86.9, and the <sup>1</sup>H-n.m.r. spectrum of the diacetate **25** showed two deshielded signals, at  $\delta$  5.58 and 4.61, both as triplets, with J 9.5 Hz for H-3 and H-4, respectively, thus showing the 2,6-substitution pattern in **24**.

In conclusion, by employing the stannylation–alkylation method, benzyl, allyl, and trityl groups have been selectively introduced at O-2 and O-6 of methyl  $\alpha$ -D-glucopyranoside.

#### EXPERIMENTAL

General. — Melting points were determined with a Yanagimoto micro meltingpoint apparatus, and are uncorrected. Optical rotations were determined with a Perkin-Elmer Model 141 polarimeter for solutions in CHCl<sub>3</sub> at 25°, unless otherwise noted. Column chromatography was performed in columns of Silica Gel Merck (70-230 mesh; E. Merck, Darmstadt, Germany). Thin-layer chromatography was conducted on precoated plates (layer thickness, 0.25 mm; E. Merck, Darmstadt, Germany) of Silica Gel 60  $F_{254}$ . I.r. spectra were recorded with an EPI-G2 Hitachi spectrophotometer, as KBr discs for the crystalline samples, and as neat films for the liquid samples. <sup>1</sup>H-N.m.r. spectra were recorded with a Varian HA-100 n.m.r. spectrometer, using tetramethylsilane as the internal standard. <sup>13</sup>C-N.m.r. spectra were recorded with a JNM-FX 100FT n.m.r. spectrometer operated at 25.05 MHz. The values of  $\delta_{\rm C}$  and  $\delta_{\rm H}$  are expressed in p.p.m. downwards from the internal standard for solutions in CDCl<sub>3</sub>, unless otherwise noted.

Methyl 3,6-di-O-benzyl- $\alpha$ -D-glucopyranoside (8), methyl 2,6-di-O-benzyl- $\alpha$ -Dglucopyranoside (9), methyl 4,6-di-O-benzyl- $\alpha$ -D-glucopyranoside (10), and methyl 6-O-benzyl- $\alpha$ -D-glucopyranoside (11). — A mixture of finely powdered 1 (970 mg, 5 mmol) and (Bu<sub>3</sub>Sn)<sub>2</sub>O (4.5 g, 7.5 mmol) in toluene (50 mL) was stirred under reflux for 14 h with continuous azeotropic removal of water. The clear solution was evaporated *in vacuo*, and a mixture of the residue with benzyl bromide (10 mL) was stirred for 2 days at 80–90° under argon, and evaporated; the residue was chromatographed on SiO<sub>2</sub> (100 g) with 1:1 toluene-EtOAc, to give crude 8 (120 mg), which was rechromatographed on SiO<sub>2</sub> (30 g) with 20:1 CH<sub>2</sub>Cl<sub>2</sub>-acetone, to give pure<sup>6</sup> 8 (85 mg, 4.5%),  $[\alpha]_D$  +79.2° (c 3.50);  $R_F$  0.55 in 1:3 toluene-EtOAc and 0.25 in 20:1 CH<sub>2</sub>Cl<sub>2</sub>-acetone;  $\delta_H$ : 7.45-7.30 (s, 10 H, 2 benzyl), 4.96 and 4.76 (AB, J 12 Hz, CH<sub>2</sub>Ph), 4.75 (d, 1 H, J 3 Hz, H-1), 4.58 (s, 2 H, CH<sub>2</sub>Ph), and 3.42 (s, 3 H, OMe).

Further elution with 1:3 toluene–EtOAc afforded<sup>6</sup> 9 (571 mg, 30.5%), m.p. 80–82° (iPr<sub>2</sub>O),  $[\alpha]_{\rm D}$  +58.7° (*c* 0.73);  $R_{\rm F}$  0.45 in 1:3 toluene–EtOAc;  $\delta_{\rm H}$ : 7.32 (s, 5 H, benzyl), 7.29 (s, 5 H, benzyl), 4.67 (s, 2 H, CH<sub>2</sub>Ph), 4.63 (d, 1 H, J 4 Hz, H-1), 4.57 (s, 2 H, CH<sub>2</sub>Ph), and 3.32 (s, 3 H, OMe).

Anal. Calc. for C<sub>21</sub>H<sub>26</sub>O<sub>6</sub>: C, 67.36; H. 7.00. Found: C, 67.21; H, 7.07.

Further elution with the same solvent afforded<sup>6</sup> 10 (113 mg, 6.0%), m.p. 70–73° (iPr<sub>2</sub>O),  $[\alpha]_D$  +109.0° (c 0.465);  $R_F$  0.30 in 1:3 toluene–EtOAc;  $\delta_H$ : 7.30 (s, 5 H, benzyl), 7.24 (s, 5 H, benzyl), and 3.38 (s, 3 H, OMe).

Anal. Calc. for C<sub>21</sub>H<sub>26</sub>O<sub>6</sub>: C, 67.36; H, 7.00. Found: C, 66.90; H, 6.95.

Further elution, with 10:1 CH<sub>2</sub>Cl<sub>2</sub>-MeOH, gave **11** (690 mg, 48.6%), m.p. 58–61° (Me<sub>2</sub>CO–iPr<sub>2</sub>O),  $[\alpha]_{D}$  + 104.7° (*c* 0.425);  $R_{F}$  0.25 in 10:1 CH<sub>2</sub>Cl<sub>2</sub>-MeOH;  $\delta_{H}$ : 7.29 (s, 5 H, benzyi), 4.71 (d, 1 H, *J* 4 Hz, H-1), 4.57 (s, 2 H, CH<sub>2</sub>Ph), and 3.37 (s, 3 H, OMe).

Anal. Calc. for C14H20O6: C, 59.14; H, 7.09. Found: C, 58.88; H, 7.11.

Methyl 3,6-di-O-allyl- $\alpha$ -D-glucopyranoside (12), methyl 2,6-di-O-allyl- $\alpha$ -D-glucopyranoside (13), and methyl 6-O-allyl- $\alpha$ -D-glucopyranoside (14). — Compound 1 (3.84 g, 20 mmol) was stannylated with (Bu<sub>3</sub>Sn)<sub>2</sub>O (18 g, 30 mmol) in toluene (100 mL) for 2 h at 140°. The clear solution was evaporated *in vacuo*, and a mixture of the residue with allyl bromide (50 mL) was stirred for 10 days at 80° under argon, cooled, and evaporated *in vacuo*; the residue was chromatographed on SiO<sub>2</sub> (200 g) with 1:3 toluene–EtOAc, to give 12 (425 mg, 8.0%),  $[\alpha]_D + 103.1°$  (c 0.390);  $R_F 0.47$  in 1:3 toluene–EtOAc;  $\delta_H$ : 6.2–5.7 (m, 2 H, 2 CH<sub>2</sub>-CH=CH<sub>2</sub>), 5.4–5.1 (m, 4 H, 2 -CH=CH<sub>2</sub>), 4.75 (d, 1 H, J 3 Hz, H-1), 4.4–4.3 (m, 2 H, -CH<sub>2</sub>-CH=CH<sub>2</sub>), 4.1–4.0 (m, 2 H<sub>2</sub> -CH=CH<sub>2</sub>), and 3.44 (s, 3 H, OMe).

Further elution with same solvent afforded 13 (2.265 g, 41.8%),  $[\alpha]_D + 106.5^{\circ}$  (c 0.755);  $R_F$  0.28 in 1:3 toluene–EtOAc;  $\delta_H$ : 6.2–5.7 (m, 2 H, 2 -CH<sub>2</sub>-CH=CH<sub>2</sub>), 5.4–5.1 (m, 4 H, 2 -CH<sub>2</sub>-CH=CH<sub>2</sub>), 4.82 (d, 1 H, J 4 Hz, H-1), 4.2–4.0 (m, 4 H, -CH<sub>2</sub>-CH=CH<sub>2</sub>), and 3.41 (s, 3 H, OMe).

Anal. Calc. for C<sub>13</sub>H<sub>22</sub>O<sub>6</sub>: C, 56.92; H, 8.08. Found: C, 56.79; H, 8.00.

Further elution, with 20:1 CH<sub>2</sub>Cl<sub>2</sub>-MeOH, gave **14** (1.09 g, 23%),  $[\alpha]_D$ +133.0° (c 0.87);  $R_F$  0.13 in 20:1 CH<sub>2</sub>Cl<sub>2</sub>-MeOH;  $\delta_H$ : 6.1-5.7 (m, 1 H, CH<sub>2</sub>-CH= CH<sub>2</sub>), 5.4-5.1 (m, 2 H, CH<sub>2</sub>-CH=CH<sub>2</sub>), 4.74 (d, 1 H, J 4 Hz, H-1), 4.1-4.0 (m, 2 H, -CH<sub>2</sub>-CH=CH<sub>2</sub>), and 3.40 (s, 3 H, OMe).

Anal. Calc. for C<sub>10</sub>H<sub>18</sub>O<sub>6</sub>: C, 51.27; H, 7.75. Found: C, 50.90; H, 7.77.

Methyl 2,6-di-O-allyl-3,4-di-O-benzyl- $\alpha$ -D-glucopyranoside (15). — To a solution of 13 (1.577 g, 5.8 mmol) in HCONMe<sub>2</sub> (20 mL) was added, portionwise, NaH (50%; 863 mg, 17.4 mmol), and the mixture was stirred for 30 min at 20°. To the cooled mixture was added, dropwise, benzyl bromide (2.1 mL, 17.4 mmol) at  $-5^{\circ}$ . The mixture was stirred for 2 h at  $-5^{\circ}$ , and for 1 h at 20°, and then, to this mixture

was added, dropwise, methanol (2 mL) at 0° (to consume the excess of NaH). Processing, and chromatography on SiO<sub>2</sub> (100 g) with 5:1 toluene–EtOAc, gave 15 (2.6 g, 98%),  $[\alpha]_D$  +40.3° (c 0.34);  $R_F$  0.70 in 3:1 toluene–EtOAc;  $\delta_H$ : 7.4–7.2 (m, 10 H, 2 benzyl), 6.2–5.6 (m, 2 H, 2 CH<sub>2</sub>-CH=CH<sub>2</sub>), 5.4–5.1 (m, 4 H, 2 CH<sub>2</sub>-CH=CH<sub>2</sub>), 5.0–4.5 (2 AB q, 4 H, 2 CH<sub>2</sub>Ph), 4.29 (d, 1 H, J 4 Hz, H-1), 4.2–3.9 (m, 4 H, 2 CH<sub>2</sub>-CH=CH<sub>2</sub>), and 3.40 (s, 3 H, OMe).

Anal. Calc. for C<sub>27</sub>H<sub>34</sub>O<sub>6</sub>: C, 71.34; H, 7.54. Found: C, 71.32; H, 7.53.

Methyl 3,4-di-O-benzyl- $\alpha$ -D-glucopyranoside (16). — A mixture of 15 (2.7 g, 5.94 mmol) and 10% Pd-C (0.3 g) in EtOH (20 mL)–AcOH (10 mL)–H<sub>2</sub>O (10 mL) was stirred for 68 h at 75°, cooled, filtered (Celite), and the filtrate evaporated *in vacuo*. The residue was chromatographed on SiO<sub>2</sub> (100 g) with 1:1 toluene–EtOAc, to give crystalline 16 (1.40 g, 63.1%)<sup>6.9</sup>, m.p. 105–106° (CH<sub>2</sub>Cl<sub>2</sub>–iPr<sub>2</sub>O),  $[\alpha]_D$  +101.3° (c 0.545);  $R_F$  0.32 in 1:3 toluene–EtOAc;  $\delta_H$ : 7.28 (bs, 10 H, 2 benzyl), 4.87 and 4.63 (AB q, 2 H, J 10 Hz, CH<sub>2</sub>Ph), 4.88 (s, 2 H, CH<sub>2</sub>Ph), 4.73 (d, 1 H, J 3 Hz, H-1), and 3.40 (s, 3 H, OMe).

Anal. Calc. for C<sub>21</sub>H<sub>26</sub>O<sub>6</sub>: C, 67.37; H, 7.00. Found: C, 67.19; H, 6.98.

Methyl 2,6-di-O-benzyl-3,4-di-O-methyl- $\alpha$ -D-glucopyranoside (17). — To a solution of 9 (230 mg, 0.5 mmol) in HCONMe<sub>2</sub> (2 mL) was added NaH (50%; 100 mg, 2 mmol, washed with hexane), and the mixture was stirred for 30 min at 15–20°. To this mixture was added MeI (0.5 mL) at  $-5^{\circ}$ , and the mixture was stirred for 2 h at -5 to 0°. Processing, and chromatography on SiO<sub>2</sub> (25 g) with 10:1 toluene–EtOAc, afforded 17 (226 mg, 91.4%),  $[\alpha]_D$  +52.0° (c 0.535);  $R_F$  0.62 in 3:1 toluene–EtOAc;  $\delta_H$ : 3.63 (s, 3 H, OMe), 3.44 (s, 3 H. OMe), and 3.32 (s, 3 H, OMe).

Anal. Calc. for C<sub>23</sub>H<sub>30</sub>O<sub>6</sub>: C, 68.63; H, 7.51. Found: C, 68.23; H, 7.35.

Methyl 2,6-di-O-benzoyl-3,4-di-O-methyl- $\alpha$ -D-glucopyranoside (20). — A solution of 19 (160 mg, 0.4 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (5 mL) containing 1 drop of BF<sub>3</sub>-ether was treated with a solution of CH<sub>2</sub>N<sub>2</sub> [prepared from 2 g of MeN(NO)CONH<sub>2</sub>] in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) according to a reported method<sup>8</sup>. Processing, and chromatography on SiO<sub>2</sub> (20 g) with 4:1 toluene–EtOAc, afforded 20 (110 mg, 64.3%), [ $\alpha$ ]<sub>D</sub> +117.5° (c 0.435);  $R_F$  0.66 in 3:1 toluene–EtOAc;  $\delta_H$ : 3.59 (s, 3 H, OMe), 3.58 (s, 3 H, OMe), and 3.37 (s, 3 H, OMe).

Anal. Calc. for C<sub>23</sub>H<sub>26</sub>O<sub>8</sub>: C, 64.17; H, 6.09. Found: C, 64.14; H, 6.14.

*Methyl* 3,4-di-O-methyl- $\alpha$ -D-glucopyranoside (18). — (A). A solution of 17 (176 mg) in EtOH (10 mL) was hydrogenolyzed in the presence of 10% Pd-C (100 mg), to give 18,  $[\alpha]_D$  +176.6° (c 0.475);  $R_F$  0.12 in 1:3 toluene–EtOAc;  $\delta_H$ : (3:1 CDCl<sub>3</sub>–Me<sub>2</sub>SO-d<sub>6</sub>): 4.69 (d, 1 H, J 3 Hz, H-1), 3.64 (s, 3 H, OMe), 3.55 (s, 3 H, OMe), and 3.40 (s, 3 H, OMe).

(B). A solution of 20 (60 mg) in MeOH (3 mL)-0.1M NaOMe (0.2 mL) was stirred for 14 h at 55°. Processing, and chromatography on SiO<sub>2</sub> (10 g) with 1:3 toluene-EtOAc, gave 18 (24 mg), which was identified by <sup>1</sup>H-n.m.r. spectroscopy with the sample prepared by method (A).

Methyl 3,4,6-tri-O-trityl-a-D-glucopyranoside (21) and methyl 2,6-di-O-trityl-a-

D-glucopyranoside (24). — Compound 1 (4.8 g, 25 mmol) was stannylated with  $(Bu_3Sn)_2O$  (23 g, 38 mmol) in toluene (100 mL). The clear solution resulting was stirred in the presence of added TrCl (21 g, 76 mmol) for 36 h at 65°. The mixture was evaporated *in vacuo*, and the residue, in CHCl<sub>3</sub> (200 mL), was stirred with<sup>10</sup> aq. KF (20 g in 100 mL) for 5 h at room temperature. Precipitated Bu<sub>3</sub>SnF was filtered off, and the filtrate was dried (MgSO<sub>4</sub>), and evaporated. The residual oil was chromatographed on SiO<sub>2</sub> (550 g) with 20:1:0.2 toluene–EtOAc–Et<sub>3</sub>N, to give 21 (8.5 g, 37.3%),  $[\alpha]_D + 1.07^\circ$  (c 0.75);  $R_F$  0.6 in 5:1 toluene–EtOAc;  $\delta_H$ : 7.3–6.8 (m, 45 H, trityl), 5.02 (bs, 1 H, w<sub>L3</sub> 3.5 Hz, H-1), and 3.83 (s, 3 H, OMe).

Anal. Calc. for C<sub>6.1</sub>H<sub>56</sub>O<sub>6</sub>: C, 83.45; H, 6.13. Found: C, 83.37; H, 6.16.

Further elution, with 10:1:0.1 toluene–EtOAc–Et<sub>3</sub>N, afforded **24** (9.0 g, 53.6%),  $[\alpha]_{D}$  + 32.8° (c 0.90);  $R_{F}$  0.3 in 5:1 toluene–EtOAc;  $\delta_{H}$ : 7.60–7.0 (m, 30 H, trityl), 4.00 (t, 1 H, J 9 Hz, H-3), 3.88 (d, 1 H, J 3 Hz, H-1), 3.64 (m, 1 H, H-5), 3.36 (q, 1 H, J 3,10 Hz, H-2), and 3.28 (s, 3 H, OMe).

Anal. Calc. for C45H42O6: C, 79.62; H, 6.24. Found: C, 79.51; H, 6.25.

*Methyl 2-O-acetyl-3,4,6-tri-O-trityl-* $\alpha$ -D-glucopyranoside (22). — Compound 21 (0.5 g) in pyridine (4 mL)–Ac<sub>2</sub>O (2 mL)–4-dimethylaminopyridine (0.1 g) was stirred for 1 h at 50°. Processing gave monoacetate 22,  $[\alpha]_D$  –23.6° (c 0.50);  $R_F$  0.50 in 10:1 toluene–EtOAc;  $\delta_{H}$ : 5.02 (d, 1 H, J 3 Hz, H-1), 4.84 (bs, 1 H, w<sub>hh</sub> 8 Hz, H-2), 3.66 (s, 3 H, OMe), and 2.04 (s, 3 H, OAc);  $\delta_C$  93.8 (<sup>1</sup>J<sub>CH</sub> 159.9 Hz, C-1), 88.4, 87.2 and 85.6 (3 methines of 3 trityl), 78.3, 70.7, 68.7, 68.3, 61.9 (C-6), 56.8 (OMe), and 21.1 (Ac).

*Methyl 3,4-di*-O-*acetyl-2,6-di*-O-*trityl-α*-D-*glucopyranoside* (25). — Compound **24** (0.7 g) in pyridine (6 mL)–Ac<sub>2</sub>O (3 mL) was stirred for 40 h at 15–20°; processing gave diacetate **25**;  $\delta_{\rm H}$ : 5.58 (t, 1 H, J 9.5 Hz, H-3), 4.61 (t, 1 H, J 9.5 Hz, H-4), 4.00 (d, 1 H, J 2.4 Hz, H-1), 3.85 (m, 1 H, H-5), 3.52 (q, 1 H, J 2.4, 9.5 Hz, H-2), 3.39 (s, 3 H, OMe), 1.85 (s, 3 H, Ac), and 1.76 (s, 3 H, Ac).

*Methyl* 2-O-*benzyl-3,4,6-tri*-O-*trityl-α*-D-*glucopyranoside* (23). — Compound 21 (0.30 g) in HCONMe<sub>2</sub> (10 mL) was benzylated with benzyl bromide and NaH, to afford 23 (447 mg, 81.4%),  $[\alpha]_{\rm D}$  +20.0° (c 0.75);  $R_{\rm F}$  0.49 in 10:1 toluene–EtOAc;  $\delta_{\rm H}$ : 4.92 (d, 1 H, J 3.5 Hz, H-1), 3.68 (s, 3 H, CMe);  $\delta_{\rm C}$ : 95.7 (<sup>1</sup>J<sub>CH</sub> 156.3 Hz, C-1), 78.3, 75.7, 73.3 (OCH<sub>2</sub>Ph), 7'.5, 69.5, 61.9 (C-6), and 56.7 (OMe).

Anal. Calc. for C<sub>71</sub>H<sub>2</sub>,O<sub>6</sub>: C, 84.36; H, 6.14. Found: C, 84.72; H, 6.18.

*Methyl* 2-O-*benzyl*- $\alpha$ -D-glucopyranoside (26). — A solution of 23 (182 mg, 0.2 mmol) in 80% aq. AcOH (20 mL) was stirred for 1 h at 80°. Evaporation in vacuo, and chromatography of the residue on SiO<sub>2</sub> (25 g) with EtOAc, afforded crystalline 26 (62 mg, 81.3%); m.p. 118.5–120°,  $[\alpha]_{\rm D}$  +85.3° (c 0.375);  $R_{\rm F}$  0.2 in EtOAc;  $\delta_{\rm H}$ : 7.36 (s, 5 H, benzyl), 4.67 (d, 1 H, J 3 Hz, H-1), 4.68, 4.66, 4.60, and 4.56 (2 H, CH<sub>2</sub>Ph), 3.36 (s, 3 H, OMe), 2.70 (bs, 1 H, OH), and 2.12 (bs, 1 H, OH). Anal. Calc. for C<sub>14</sub>H<sub>20</sub>O<sub>6</sub>: C, 59.16; H, 7.04. Found: C, 58.93; H, 7.06.

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#### REFERENCES

- 1 T. OGAWA AND M. MATSUI, Carbohydr. Res., 62 (1978) c1-c4.
- 2 T. OGAWA AND M. MATSUI, Carbohydr. Res., 56 (1977) c1-c6; Tetrahedron, 37 (1981) 2363-2369.
- 3 T. OGAWA, K. KATANO, AND M. MATSUI, *Carbohydr. Res.*, 64 (1978) C3-C9; T. OGAWA, K. KATANO, K. SASAJIMA, AND M. MATSUI, *Tetrahedron*, 37 (1981) 2779-2786.
- 4 P. A. GENT AND R. GIGG, J. Chem. Soc. Perkin Trans. 1, (1974) 1446-1455, 1835-1839; (1975) 361-363.
- 5 H. J. KOCH AND A. S. PERLIN, Carbohydr. Res., 15 (1970) 403-410; D. E. DORMAN AND J. D. ROBERTS, J. Am. Chem. Soc., 92 (1970) 1355-1361; P. A. J. GORIN AND M. MAZUREK, Can. J. Chem., 53 (1975) 1212-1223; T. E. WALKER, R. E. LONDON, T. W. WHALEY, R. BARKER, AND N. A. MATWIYOFF, J. Am. Chem. Soc., 98 (1976) 5807-5813.
- 6 J. M. KÜSTER AND I. DYONG, Justus Liebigs Ann. Chem., (1975) 2179-2189.
- 7 R. Boss AND R. SCHEFFOLD, Angew. Chem., 88 (1976) 578-579.
- 8 I. O. MASTRONARDI, S. M. FLEMATTI, J. O. DEFERRARI, AND E. G. GROS, *Carbohydr. Res.*, 3 (1966) 177–183; J. O. DEFERRARI, E. G. GROS, AND I. O. MASTRONARDI, *ibid.*, 4 (1967) 432–434.
  9 A. LIPTÁK, I. JODÁL, AND P. NÁNÁSI, *Carbohydr. Res.*, 44 (1975) 1–11.
- 10 J. E. LEIBNER AND J. JACOBUS, J. Org. Chem., 44 (1979) 449–450; D. MILSTEIN AND J. K. STILLE, J. Am. Chem. Soc., 100 (1978) 3636–3638.