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Antagonists of *myo*-Inositol 3,4,5,6-Tetrakisphosphate Allow Repeated Epithelial Chloride Secretion

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Abstract—Cystic fibrosis (CF) patients suffer from a defect in hydration of mucosal membranes due to mutations in the cystic fibrosis transmembrane regulator (CFTR), an apical chloride channel in mucosal epithelia. Disease expression in CF knockout mice is organ specific, varying with the level of expression of calcium activated Cl- channels (CLCA). Therefore, restoring transepithelial Cl⁻ secretion by augmenting alternate Cl⁻ channels, such as CLCA, could be beneficial. However, CLCA-mediated Cl⁻ secretion is transient, due in part to the inhibitory effects of *myo*-inositol 3,4,5,6-tetrakisphosphate [Ins $(3,4,5,6)P_4$]. This suggests that antagonists of $Ins(3,4,5,6)P_4$ could be useful in treatment of CF. We have, therefore, synthesized a series of membrane-permeant Ins(3,4,5,6)P₄ derivatives, carrying alkyl substituents on the hydroxyl groups and screened them for effects on Cl⁻ secretion in a human colonic epithelial cell line, T₈₄. While membrane-permeant Ins(3,4,5,6)P₄ derivatives had no direct effects on carbacholstimulated Cl⁻ secretion, Ins(3,4,5,6)P₄ derivatives, but not enantiomeric Ins(1,4,5,6)P₄ derivatives, reversed the inhibitory effect of Ins $(3,4,5,6)P_4$ on subsequent thapsigargin activation of Cl⁻ secretion. The extent of the antagonistic effect of the Ins $(3,4,5,6)P_4$ derivatives varied with the position of the alkyl substituents. Derivatives with a cyclohexylidene ketal or a butyl-chain at the 1position reversed the Ins(3,4,5,6)P₄-mediated inhibition of Cl⁻ secretion by up to 96 and 85%, respectively, whereas butylation of the 1- and 2-position generated a reversal effect of only 65%. Derivatives carrying the butyl chain only at the 2-position showed no antagonistic effect. These data: (1) Support the hypothesis that $Ins(3,4,5,6)P_4$ stereospecifically inhibits Ca^{2+} activated Cl^- secretion and that $Ins(3,4,5,6)P_4$ mediates most, if not all of the cholinergic-mediated inhibition of chloride secretion in T_{84} cells; (2) Demonstrate Ins(3,4,5,6)P₄-mediated inhibition can be completely reversed with rationally designed membrane-permeant Ins(3,4,5,6)P4 antagonists; (3) Demonstrate that a SAR for membrane-permeant Ins(3,4,5,6) P4 antagonists can be generated and screened in a physiologically relevant cell-based assay; (4) Indicate that $Ins(3,4,5,6)P_4$ derivatives could serve as a starting point for the development of therapeutics to treat cystic fibrosis.

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Introduction

Transepithelial chloride flux by polarized epithelia controls a variety of physiological processes, such as intestinal and pancreatic secretion and renal functions.¹ For example, Cl⁻ secretion regulates the fluid balance of the intestinal lumen,² and is a driving force for the hydration of airway epithelia.³ Given this key role in physiological functions, defects in Cl⁻ transport can result in life-threatening diseases, such as diarrhea and cystic fibrosis (CF).⁴

Transepithelial Cl⁻ secretion is gated through multiple apical Cl⁻ channels. One of these channels, the cystic

fibrosis transmembrane regulator (CFTR) can be triggered by elevating cAMP levels, and consequently PKA-dependent phosphorylation.⁵ The vast majority of CF patients suffer from a defect in hydration of mucosal membranes due to mutations in the gene coding for the CFTR. Mucosal epithelia also express Cl⁻ channels other than the CFTR such as the outwardly rectifying Cl⁻ channel (ORCC), calcium-activated Cl⁻ channels (CLCA) and volume-regulated Cl⁻ channels (ClC). While the ORCC appears to be controlled by the CFTR and therefore is dysfunctional in CF,^{6–9} active CLCA are reportedly more abundant in CF tissue.^{10,11} A number of studies indicate that phenotypes with increased activity of alternate Cl⁻ channels such as the CLCA correlate to milder clinical manifestations.^{9,12–15} Therefore, in CF, cAMP-mediated Cl- transport is defective,^{16,17} leaving Cl⁻ secretion via the Ca²⁺

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dependent pathway intact and even enhanced. Thus, it has been hypothesized that alternate ion channels may compensate for defects in CFTR function and could be therapeutically useful. There are compelling arguments for pursuing artificial activation of alternate Cl⁻ channels to counteract CF pathophysiology. This has lead to tests with Ca²⁺-elevating agents such as purinergic agonists in the treatment of CF.18 Currently, two compounds are in clinical development that elevate intracellular calcium [(Ca²⁺)_i] and thereby modulate Cl⁻ secretion; INS365, a PY2Y receptor agonist and duramycin, an antibiotic that triggers an increase in intracellular Ca²⁺ levels. However, an increase in $[(Ca^{2+})_i]$ does not always lead to Cl- secretion. Activation of phospholipase C (PLC) can lead to a transient activation of Cl⁻ secretion through the *myo*-inositol 1,4,5-trisphosphate $[Ins(1,4,5)P_3]/[Ca^{2+}]_i$ pathway, but also promotes a long-term inhibitory feedback, preventing Cl⁻ secretion from being sustained.¹⁹ We have demonstrated that the intracellular signaling molecule, $Ins(3,4,5,6)P_4$, becomes elevated after prolonged PLC activation and 'uncouples' Cl- secretion from the rise in intracellular Ca²⁺ in mucosal epithelia.²⁰ This regulatory role for $Ins(3,4,5,6)P_4$ has been confirmed by several investigators.^{21,22}

The observations that cholinergic but not histaminergic stimulation uncoupled the Ca²⁺-mediated Cl⁻ secretion (CaMCS), and the inhibitory action of membrane-permeant derivative of Ins(3,4,5,6)P₄, have led to the conclusion that $Ins(3,4,5,6)P_4$ is the endogenous negative regulator of this conductance.^{20,23} Studies by Xie et al. employing whole cell patch clamp technique with intracellular perfusion of inositol tetrakisphosphates indicated that $Ins(3,4,5,6)P_4$ modulated an apically located $Ca^{2+}/calmodulin$ kinase regulated Cl^{-} channel in T_{84} cells.^{22,24} Furthermore, Ca^{2+} -dependent Cl^- channels reconstituted in planar lipid bilayers could also be regulated by low levels of $Ins(3,4,5,6)P_4$ in the absence of phosphatase activities.²⁵ Electrophysiological analysis of the effects of $Ins(3,4,5,6)P_4$ in the CFPAC-1 cell line, originally derived from a CF patient suffering from the Δ F508 defect,²⁶ provided further evidence for an important and widespread intracellular function of Ins(3,4,5,6)P₄ as a negative regulator of Cl⁻ secretion.21,27,28

The pharmaceutical development of activators of Ca²⁺mediated Cl⁻ secretion in CF epithelia could be limited by the inhibitory action of Ins(3,4,5,6)P₄ on Cl⁻ transport. Therefore, there is an urgent need to understand the mechanism by which Ins(3,4,5,6)P₄ inhibits CaMCS, specifically identifying the downstream targets of Ins(3,4,5,6)P₄. Accordingly, we have synthesized a set of membrane-permeant Ins(3,4,5,6)P₄ derivatives to study the interaction of Ins(3,4,5,6)P₄ with its targets, and more importantly—to find antagonists of the inhibitory action of Ins(3,4,5,6)P₄ (Fig. 1). Prodrug approaches, in which the highly polar phosphate groups are masked, are commonly used to help deliver biologically active molecules to the cytosol.²⁹ In fact, membrane-permeant derivatives of inositol polyphosphates have been successfully employed in the past,^{20,30–36} and have been demonstrated to elevate, for instance, intracellular $Ins(3,4,5,6)P_4$ levels.²⁰ Preferred masking groups were acyloxymethyl groups. Here we demonstrate that some of these $Ins(3,4,5,6)P_4$ derivatives are potent antagonists of this inhibition following carbachol stimulation. Furthermore, the selection of modified $Ins(3,4,5,6)P_4$ derivatives enables us to elucidate some of the structural requirements for the putative $Ins(3,4,5,6)P_4$ interactions to its targets.

Results

Design and synthesis

We have previously shown that modifications of the hydroxyl groups of Ins(3,4,5,6)P₄ selectively influence the ability of membrane-permeant $Ins(3,4,5,6)P_4$ derivatives to inhibit carbachol-induced Cl⁻-secretion. In that study we demonstrated that a 2-deoxy derivative was a partial agonist for Ins(3,4,5,6)P₄, inhibiting Clsecretion, while the corresponding 1-deoxy derivative showed no effect on Cl--secretion.³⁶ Therefore, the complete deletion of a hydroxyl group eliminated both the hydrogen bonding donor and acceptor potential for interaction with targets. In the present study, we sought derivatives that would provide the acceptor potential but are void of the hydrogen donor properties. Apart from the compounds mentioned above, some deoxyfluoro-Ins(3,4,5,6)P₄ derivatives were previously prepared.³⁷ Some of these compounds are suspected to have effects on protein phosphatases.²²

We prepared a series of membrane-permeant myo-inositol 3,4,5,6-tetrakisphosphate derivatives alkylated at the hydroxyl groups, namely 1-O-butyl-2-O-butyrylmyo-inositol 3,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester $[1-O-Bu-2-O-Bt-Ins(3,4,5,6)P_4/AM]$ (1), 2-O-butyl-1-O-butyryl-myo-inositol 3,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester [2-O-Bu-1-O-Bt-Ins $(3,4,5,6)P_4/AM$ (2), 1,2-di-O-butyl-myo-inositol 3,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester $[1,2-di-O-Bu-Ins(3,4,5,6)P_4/AM]$ (3), and the corresponding enantiomeric compounds 3-O-Bu-2-O-Bt- $Ins(1,4,5,6)P_4/AM$ (*ent*-1), 2-O-Bu-3-O-Bt- $Ins(1,4,5,6)P_4/AM$ (ent-2) and 2,3-O-Bu-Ins(1,4,5,6)P_4/ AM (ent-3), respectively³⁸ (Fig. 1).

Synthesis of the Ins(3,4,5,6)P₄ derivative *rac*-1,2-*O*-cyclohexylidene-*myo*-inositol 3,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester (*rac*-4) started from the common precursor *rac*-1,2-*O*-cyclohexylidene-*myo*-inositol (*rac*-5) (Scheme 1). The phosphorylation of *rac*-5 proceeded smoothly using the method of Yu and Frazer-Reid³⁹ to give the fully protected tetrakisphosphate *rac*-6 after purification by preparative HPLC. Deprotection of the phosphoric acid triester was achieved by catalytic hydrogenolysis with Pd/C in ethanol. Equal amounts of ethyl-*N*,*N*-diisopropylamine (DIEA) with respect to the number of deprotected phosphates produced during the reaction prevented the otherwise common hydrolysis of the acid labile cyclohexylidene group. After filtration and drying *rac*-7 was obtained as its DIEA salt. *rac*-7 was



Figure 1. Structures of *myo*-inositol 3,4,5,6-tetrakisphosphate derivatives. Bioactivatable protecting groups are shown in red: 1-*O*-butyl-2-*O*-butyryl-*myo*-inositol 3,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester (1) and the enantiomer 3-*O*-butyl-2-*O*-butyryl-*myo*-inositol 1,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester (2), 1,2-di-*O*-butyl-*myo*-inositol 3,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester (2), 1,2-di-*O*-butyl-*myo*-inositol 3,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester (2), 3,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester (*ac*-4).



Scheme 1. (i) (a) (BnO)₂PN*i*Pr₂, 1*H*-tetrazole, MeCN; (b) -40 °C, AcOOH; (ii) H₂, Pd/C (10%), ethanol, DIEA, 6d; (iii) AMBr, DIEA, MeCN.

alkylated with acetoxymethyl bromide (AMBr) in the presence of DIEA to give the uncharged, bioactivatable octakis(acetoxymethyl) ester *rac*-4.

The synthesis strategy (Scheme 2) for mono- and dibutyl ether derivatives followed those previously reported for similar compounds.⁴⁰ In brief, enantiomerically pure 3,4,5,6-tetra-O-benzyl-myo-inositol (8) and 1,4,5,6-tetra-O-benzyl-myo-inositol (ent-8), respectively, were regioselectively alkylated with 1-butyl iodide at the 1/3-OH group via a cyclic dibutylstannyl intermediate. The remaining hydroxyl group was esterified with butyric anhydride. After removal of the benzyl groups, standard phosphorylation/deprotection procedures gave the tetrakisphosphates, which were alkylated with acetoxymethyl bromide, to yield the uncharged octakis(acetoxymethyl) esters 1 and ent-1, respectively. In case of 2-OH alkylated compounds, the initial alkylation of the 1/ 3-OH group was performed with allyl iodide, followed by the alkylation of the 2-OH group with 1-butyl iodide in the presence of sodium hydride. Selective removal of the allyl ether with tris(triphenyl)phosphin rhodium(I) chloride, followed by treatment with trifluoroacetic acid, gave alcohols 9 and (ent-9), respectively. Butyrylation and subsequent hydrogenolysis resulted in tetrols 11 and (ent-11), respectively. Phosphorylation and deprotection steps were performed as described

above to finally give the octakis(acetoxymethyl) esters 2 and *ent-*2, respectively. The 1,2-di-*O*-butyl-*myo*-inositol (24) and 2,3-di-*O*-butyl-*myo*-inositol (*ent-*24) precursors were synthesized by alkylation of 8 and *ent-*8, respectively, with butyl iodide in the presence of NaH in DMF, followed by hydrogenolysis. The reaction sequence to the octakis(acetoxymethyl) esters 3 and *ent-*3, respectively, was performed as described above for the other tetrakisphosphates.

Finally, the monobutyrates 13, ent-13, 19, and ent-19 were hydrolyzed with KOH to give the potassium salts of 1-O-butyl-myo-inositol 3,4,5,6-tetrakisphosphate (14) and 3-O-butyl-myo-inositol 1,4,5,6-tetrakisphosphate (ent-14) as well as 2-O-butyl-myo-inositol 3,4,5,6-tetrakisphosphate (22) and 2-O-butyl-myo-inositol 1,4,5,6-tetrakisphosphate (ent-22), respectively. Combined with the 1,2-di-O-butyl-myo-inositol 3,4,5,6-tetrakisphosphate (26) and its enantiomer ent-26 a set of six orthogonally alkylated tetrakisphosphates is now available for in vitro studies.

Effect of inositol 3,4,5,6-tetrakisphosphate derivatives on calcium-mediated chloride secretion

The colonic epthelia cell line T_{84} , an established model for the study of Cl⁻ secretion, was grown to confluence



Scheme 2. (i) (a) Bu_2SnO , toluene, reflux, soxhlet with molecular sieves, 3Å; (b) butyl iodide, CsF, DMF; (ii) Bt_2O , pyridine, DMAP; (iii) H_2 , Pd/C (10%), AcOOH; (iv) (a) (BnO)_2PNiPr_2, tetrazole, MeCN; (b) -40 °C, AcOOH; (v) AMBr, DIEA, MeCN; (vi) (a) Bu_2SnO , toluene, reflux, soxhlet with molecular sieves, 3Å; (b) allyl iodide, CsF, DMF; (vii) butyl iodide, NaH, DMF; (viii) (a) tris(triphenylphosphine)rhodium(I) chloride, EtOH/ H_2O ; (b) TFA. For details on intermediates, see text and Experimental. Only the enantiomers of the Ins(3,4,5,6)P4 series are shown.

on polycarbonate membranes in Snapwells (Costar). The monolayers were mounted into modified Ussing chambers and Cl- secretion was measured as short circuit current (I_{sc}) as described before.³⁶ We have previously reported that enantiomerically pure 1,2di-O-butyryl-Ins(3,4,5,6)P₄/AM was capable of inhibiting charbachol-stimulated CaMCS by about 50% and that of thapsigargin by about 60%.30 To compare the inhibitory effects of membrane-permeant derivatives of $Ins(3,4,5,6)P_4$ and $Ins(1,4,5,6)P_4$ on carbachol-stimulated CaMCS, cells were pre-incubated with $600 \,\mu\text{M}$ of the membrane-permeant derivatives 1– 3, respectively, for 30 min prior to mounting, or with $800\,\mu\text{M}$ of the racemic compound *rac*-4. After an additional 10 min, 100 µM carbachol was added to the solution bathing the basolateral surface and the Isc was monitored. We choose to stimulate CaMCS with carbachol and not with histamine due to the higher magnitude of the effect.²³ However, no significant change in Isc was observed although some of the derivatives led to a slighty (10–15%) higher I_{sc} compared to control. These results indicate that the $Ins(3,4,5,6)P_4$ -derivatives 1–3 and derivative rac-4 did not inhibit CaMCS in T_{84} cells.

Effect of membrane-permeant Ins(3,4,5,6)P₄-derivatives on carbachol-induced inhibition of chloride secretion

The response of T_{84} cells to carbachol is followed by a prolonged period in which subsequent CaMCS is suppressed.²³ This inhibition of the Cl⁻ secretory response is due, at least in part, to carbachol-induced elevation in intracellular Ins(3,4,5,6)P₄ levels, uncoupling Cl⁻ secretion from $[Ca^{2+}]_i$. This effect could be simulated by

using the membrane-permeant $Ins(3,4,5,6)P_4$ derivative, $Bt_2Ins(3,4,5,6)P_4/AM$, which generated intracellular $Ins(3,4,5,6)P_4$ levels of about $4\,\mu M.^{20}$ In this way, we tested whether the membrane-permeant derivatives 1–4 could reverse carbachol-induced inhibition of Cl⁻ secretion.

This protocol could potentially verify whether carbachol-mediated inhibition of Cl- secretion is due in its entirety to the elevation of Ins(3,4,5,6)P4 or whether other processes initiated through carbachol also contribute to the inhibition. To test this possibility, T_{84} monolayers were preincubated with 600 µM extracellular 1-O-Bu-2-O-Bt-Ins $(3,4,5,6)P_4/AM$ (1) for 30 min prior to mounting. 25 min after stimulation with carbachol, 1 µM thapsigargin was added to both halves of the Ussing chamber in order to elevate $[Ca^{2+}]_i$ and subsequently stimulate Cl⁻ secretion (Fig. 2). As is shown in Figure 3, 1-O-Bu-2-O-Bt-Ins(3,4,5,6)P₄/AM (1) abolished the inhibitory effect of carbachol on thapsigargin-stimulated Cl⁻ secretion by about 88%, while the enantiomeric $Ins(1,4,5,6)P_4/AM$ derivative (ent-1) was ineffective (Fig. 3). Lower concentrations $(200 \,\mu\text{M})$ showed no significant inhibition (n = 4, data not shown). To investigate whether the other butylated compounds were equally effective, cells were preincubated with similar amounts of 2-O-Bu-1-O-Bt-Ins(3,4,5,6)P₄/AM (2) (Fig. 2) or 1,2-di-O-Bu-Ins $(3,4,5,6)P_4/AM$ (3) and I_{sc} was monitored under the same conditions. As shown in Figure 3, 1,2-di-O-Bu-Ins $(3,4,5,6)P_4/AM$ (3) reversed the carbachol-mediated inhibition of CaMCS by about 65% compared to control, whereas 2-O-Bu-1-O-Bt- $Ins(3,4,5,6)P_4/AM$ (2) was ineffective (Figs 2 and 3). The enantiomeric Ins(1,4,5,6)P₄ derivatives ent-2 and ent-3



Figure 2. Time course of the reversed $Ins(3,4,5,6)P_4$ inhibition of thapsigargin-stimulated Cl⁻-secretion by membrane-permeant $Ins(3,4,5,6)P_4$ derivatives. T_{84} monolayers were incubated for 30 min with (A) **1**, (B) **2**, (C) **3** where indicated, or vehicle (DMSO/5% Pluronic) prior to mounting into Ussing chambers. Cl⁻ secretion was measured as short circuit current (ΔI_{sc}). Carbachol (10⁻⁴ M) was added to the basolateral side to induce elevated Ins(3,4,5,6)P_4 levels and a transient onset of Cl⁻-secretion.²⁰ After 25 min, thapsigargin (10⁻⁶ M) was added to the basolateral and the apical side to stimulate Cl⁻-secretion a second time. Data reflect monitoring every 4s and are the average of six experiments.

were also unable to reverse the inhibition of Cl⁻ secretion (Fig. 3). The racemic compound 1,2-O-cyclohexylidene-Ins(3,4,5,6)P₄/AM (*rac*-4) reversed the carbachol-induced inhibition of CaMCS by about 95% compared to control, at an extracellular concentration of 800 μ M (Figs 2 and 3), while lower doses (400 μ M) were not effective (n=4, data not shown). These data suggest that the Ins(3,4,5,6)P₄ derivatives 1, 3, and *rac*-4, but not 2, are able to bind to the putative Ins(3,4,5,6)P₄-binding site, preventing Ins(3,4,5,6)P₄ from inhibiting CaMCS.

Effect of membrane-permeant Ins(3,4,5,6)P₄ derivatives on intracellular calcium levels

We have previously demonstrated that $Ins(3,4,5,6)P_4$ inhibits Ca^{2+} -dependent Cl^- secretion without altering intracellular Ca^{2+} mobilization.²⁰ Therefore, it was unlikely that the reversal of the inhibition could be due to an effect on intracellular calcium levels ($[Ca^{2+}]_i$). As expected, ratio imaging experiments with Fura-2-loaded T_{84} cells showed unchanged $[Ca^{2+}]_i$ levels after treatment with 600 μ M extracellular for the compounds 1–3 or 800 μ M for the derivative *rac*-4, respectively (*n*=2, data not shown). Furthermore, the rise in $[Ca^{2+}]_i$ in response to thapsigargin was not altered by any of the derivatives (*n*=4, data not shown).

Discussion

The recent discovery of $Ins(3,4,5,6)P_4$ as an endogenous negative regulator of CaMCS in T_{84} cells may assist in the design of therapeutic agents which modulate fluid secretion in diseased epithelia of patients suffering from hypersecretory or hyposecretory disorders such as secretory diarrhea or CF.

However, the downstream effector in the $Ins(3,4,5,6)P_4$ signaling pathway, the $Ins(3,4,5,6)P_4$ target(s), has not yet been identified. Photoaffinity labeled derivatives or affinity columns may help to characterize these proteins in the future. Knowledge of the canonical ligand/protein interaction properties of $Ins(3,4,5,6)P_4$ with its putative



Figure 3. Ins(3,4,5,6)P₄/AM derivatives, but not Ins(1,4,5,6)P₄/AM derivatives reversed Ins(3,4,5,6)P₄ inhibition of thapsigargin-stimulated Cl--secretion in T₈₄ cells. Monolayers were preincubated with the indicated membrane-permeant tetrakisphosphate acetoxymethyl ester for 30 min prior to mounting into the Ussing chamber. CaMCS was first stimulated with carbachol (10^{-4} M) and after 25 min for a second time with thapsigargin (10⁻⁶M). Data are mean peak $\Delta I_{sc} \pm SEM$ expressed as% control for six experiments. Control values represent the response to thapsigargin in co-incubated monolayers. Significant differences are denoted by probability values derived by Students' two-tailed *t*-test (* p < 0.03; *** p < 0.04; *** p < 0.009). Concentration of the InsP₄ derivatives were 1-*O*-Bu-2-*O*-Bt-Ins(3,4,5,6)P₄/AM (1, 600 µM); 2-O-Bu-1-O-Bt-Ins(3,4,5,6)P₄/AM (2, 600 µM); 1,2-di-O-Bu-Ins(3,4,5,6)P₄/AM (3, 600 µM); rac-1,2-cyclo-Ins(3,4,5,6)P₄/AM (rac-4, 800 µM), 3-O-Bu-2-O-Bt-Ins(1,4,5,6)P₄/AM (ent-1, 600 µM); 2-O-Bu-3-O-Bt-Ins(1,4,5,6)P₄/AM (ent-2, 600 µM); 2,3-di-O-Bu-Ins(1,4,5,6)P₄/AM (ent-3, 600 µM).

binding sites will enable us to identify more potent ligands. The modified membrane-permeant $Ins(3,4,5,6)P_4$ -derivatives presented here now enable us to map the unidentified binding-sites in the natural environment of the living cell with transepithelial Cl⁻secretion as a read-out.

Derivatives for mapping the $Ins(3,4,5,6)P_4$ -binding sites require the systematic modification of each of the functional groups, namely the hydroxyl and phosphate moieties to alter particular sites of interaction with the target. In this study, we modified the interaction potential of the hydroxyl groups by simple alkylation since previous studies have shown that the proper orientation of the phosphate moieties is essential to exhibit physiological function.^{20,24} Alkylation of the respective hydroxyl group yielded $Ins(3,4,5,6)P_4$ derivatives, which exhibit the same stereochemistry, and an unchanged acceptor potential for hydrogen bonding. All alkylated $Ins(3,4,5,6)P_4$ -derivatives (1–4) were unable to inhibit CaMCS. Therefore, it appears that the hydrogen bonding donor potential of the hydroxyl groups is essential for the ability to inhibit CaMCS. Furthermore, we previously reported that a 1-deoxy-derivative, where the donor and acceptor potential is missing, was incapable of inhibiting Cl⁻ secretion, indicating that the hydrogen bonding acceptor potential of the 1-hydoxyl group is critical. In contrast, the 2-hydroxyl group was shown to play a relatively minor role (Fig. 4), because the 2-deoxy $Ins(3,4,5,6)P_4$ -derivative was partially active. We can therefore conclude that the lack of an effect of the 1-O-

Bt-2-O-Bu-Ins(3,4,5,6)P₄-derivative (2) on Cl⁻ secretion resulted from steric hindrance of the butyl group at the 2-position rather than from loss of hydrogen bonding donor potential.

Of course, the lack of effect on Cl⁻ secretion does not necessarily reflect the binding properties to protein(s). We investigated the competition of our derivatives with naturally produced Ins(3,4,5,6)P₄ after carbachol stimulation. Since carbachol-mediated inhibition of thapsigargin-induced Cl⁻ secretion was essentially abolished by pretreatment with 1-O-Bu-2-O-Bt-Ins(3,4,5,6)P₄/AM (1) as well as rac-1,2-O-cyclohexylidene-Ins $(3,4,5,6)P_4$ AM (rac-4), compounds modified at the 1-hydroxyl position seemed to retain the ability to bind to the $Ins(3,4,5,6)P_4$ target protein thereby antagonizing the inhibitory effect of $Ins(3,4,5,6)P_4$. 1,2-Di-O-Bu- $Ins(3,4,5,6)P_4/AM$ (3) had a slightly weaker effect and 2-O-Bu-1-O-Bt-Ins $(3,4,5,6)P_4$ /AM (2) was inactive. Thus, a flexible bulky substituent at the 2-position appeared to abolish the binding properties of the $Ins(3,4,5,6)P_4$ derivative probably due to steric hindrance (Fig. 4). On the other hand, the more rigid positioning of the cyclohexylidene ketal in rac-4 did not restrict binding, sugdiscrimination gesting steric factors of the Ins(3,4,5,6)P₄-binding site. The binding ability of the 1-O-butyl compound (1) to the $Ins(3,4,5,6)P_4$ -binding site points to the role of the 1-hydroxyl group as a hydrogen bonding acceptor rather than a donor group (Fig. 4). In addition, the lipophilicity of the butyl group (or the cyclohexylidene ketal) seems to promote binding as the sterically hindered 1,2-di-O-Bu-Ins(3,4,5,6)P₄ derivative (3) is also an antagonist, albeit with lower potency (Fig. 3). All enantiomeric $Ins(1,4,5,6)P_4/AM$ -derivatives (ent-1 through ent-3) did not prevent $Ins(3,4,5,6)P_4$ from inhibiting Cl⁻ secretion (Fig. 3), again indicating the importance of the proper orientation of the phosphates. In addition to the above findings, the complete reversal of the carbachol/Ins(3,4,5,6)P₄-mediated inhibition of the thapsigargin-induced Cl⁻ secretion suggests that carbachol inhibits subsequent CaMCS exclusively via $Ins(3,4,5,6)P_4$, with negligible contribution from other negative regulators.

For the development of future tools such as tethered compounds, the 1-hydroxyl group appears to be the most obvious choice to link a (hydrophobic) spacer to $Ins(3,4,5,6)P_4$. Furthermore, membrane-permeant deri-



Figure 4. Relative importance of the hydroxyl groups of $Ins(3,4,5,6)P_4$ with respect to binding and inhibitory properties.

vatives of fluorescently labeled $Ins(3,4,5,6)P_4$ -derivatives may help to locate the $Ins(3,4,5,6)P_4$ targets in living cells.

The antagonists identified in this study can serve as lead compounds for the development of pharmaceuticals, which can augment CaMCS by inactivating a feedback inhibitory mechanism. Such agents may overcome the chronically reduced electrolyte and water flux of epithelia from CF-patients. An important step for future optimization and drug development would be the identification of the putative $Ins(3,4,5,6)P_4$ -binding proteins.

In summary, we have demonstrated that modified membrane-permeant derivatives are useful tools for probing inositol phosphate/protein interaction in intact living cells. This methodology enabled us to identify the critical nature of the hydrogen bonding acceptor potential of the 1-hydroxyl group for binding to the $Ins(3,4,5,6)P_4$ binding site. Furthermore, we were able to identify specific antagonists of $Ins(3,4,5,6)P_4$ prior to the identification of the $Ins(3,4,5,6)P_4$ targets. The preparation of additional derivatives in their membranepermeant form might both increase our knowledge of the important messenger $Ins(3,4,5,6)P_4$ and open new possibilities for developing treatments for CF.

Experimental

All chemical reagents were obtained in the highest purity available. Where necessary, solvents were dried and/ or distilled before use. Acetonitrile was distilled from phosphorus(V) oxide and stored over molecular sieves (3Å). Ethyl-*N*,*N*-di*iso*propylamine (DIEA, from Merck) was dried over sodium wire. Palladium on charcoal (10%) was from Acros Organics. Dibenzyl N,N-diisopropylphosphoramidite, tetrazole, peracetic acid (32%), dibutyltin oxide, and acetoxymethyl bromide were from Aldrich. 4-(Dimethylamino)pyridine (DMAP) and cesium fluoride (CsF) were from Fluka. Thapsigargin was purchased from Alexis Biochemicals, La Jolla, CA, USA. Carbachol was obtained from ICN, Irvine, CA, USA. Fura-2 acetoxymethyl ester (fura-2/ AM) was purchased from Calbiochem, La Jolla, CA, USA. Cell culture membrane inserts (Snapwell, 0.4 µm pore size polycarbonate) were obtained from Corning Costar Corporation (Cambridge, MA, USA). All other reagents were at least reagent grade and were obtained commercially.

HPLC was performed on a LDC/Milton Roy Constametric III pump with a LDC/Milton Roy UV Monitor D (254 nm) or a Knauer refractive index detector. The analytical column was a Merck Hibar steel tube (250 × 4 mm) filled with RP 18 material (Merck, LiChrosorb; 10 μ m). Preparative HPLC was performed using a Shimadzu LC 8A pump with a preparative LDC UV III Monitor (254 nm) or a Knauer refractive index detector and a Merck Prepbar steel column (250 × 50 mm) filled with RP 18 material (Merck, LiChrospher 100, 10 μ m). The eluents were methanol–water mixtures; compositions are given in % methanol (MeOH).

¹H NMR and ³¹P NMR spectra were recorded on a Bruker DPX 200 or a Bruker AM 360 spectrometer. Chemical shifts were measured in ppm relative to tetramethylsilane for ¹H NMR spectra and external 85% H_3PO_4 for ³¹P NMR spectra. Coupling constants (J) are given in Hz. Mass spectra were recorded using a Finnigan Mat 8222 mass spectrometer with fast atom bombardment (FAB) ionization. Peak intensities are given on a scale from 0 to 100. High resolution masses (HRMS) were determined relative to known compounds with a mass not differing more than 10%. Where indicated, high resolution masses were determined with direct chemical ionisation (DCI) and matched with peaks from perfluorinated kerosins (PFK) giving a maximum difference of 5 ppm. Meltings points (uncorrected) were determined using a Büchi B-540 apparatus. Optical rotations were measured at the sodium D-line in a 10-cm cell with a Perkin-Elmer 1231 polarimeter. Ultrafiltration of the palladium/carbon catalyst was performed with a Sartorius filtration apparatus SM 162 01 using filters from regenerated cellulose (Sartorius, SM 116 04). Element analyses were performed by Mikroanalytisches Labor Beller, Göttingen, Germany.

Cell culture

All studies were performed using monolayers of the T_{84} cell line, and cells from passage numbers 20–29 only. Methods for the maintenance of T_{84} cells for use in transepthelial electrolyte transports studies have been described previously.⁴¹ In brief, T_{84} cells were grown in Dulbecco's modified Eagle's/F12 media (JRH, Lenexa, KS, USA) supplemented with 5% newborn calf serum (Hyclone, Logan, UT, USA) and 50 U/mL each of penicillin/streptomycin (Core Cell Culture Facility, UCSD). Medium was replaced every third day. For the measurement of chloride secretion, 10⁶ cells were seeded onto microporous filter inserts (see above) and maintained for 8–12 days prior to experiments in order to develop confluent monolayers with stable transepithelial resistance.

Chloride secretion

Chloride secretion was measured as short circuit current (I_{sc}) across monolayers of T_{84} cells, mounted in modified Ussing chambers (Physiologic Instruments, San Diego, CA, USA), bathed with Ringer solution warmed to 37 °C and gassed continuously with 95% O₂, 5% CO₂ at a rate of 30-35 mL/min, as described before.36 The spontaneous potential difference across the monolayer was short-circuited with a voltage clamp (Model VCC MC6, Physiologic Instruments, San Diego, CA, USA). Short circuit current (I_{sc}) and conductance were recorded at 4-s intervals using Acquire and Analyze Software 1.2 (Physiologic Instruments, San Diego, CA, USA). Increased I_{sc} stimulated through cholinergic pathways or through calcium-mobilizing agonists in T_{84} cells are fully reflective of Cl⁻ secretion.⁴² The Ringer's solution contained (in mM) 140 Na⁺, 5.2 K⁺, 1.2 Ca²⁺, 0.8 Mg^{2+} , 119.8 Cl⁻, 25 HCO₃⁻, 2.4 H₂PO₄⁻, 0.4 HPO₄⁻, and 10 glucose and was adjusted to pH 7.4.

Preincubation of T_{84} cells with membrane-permeant InsP₄-derivatives

Confluent T_{84} cells on Snapwell inserts were rinsed with 1 mL of Ringer's solution on each side of the Snapwell. 100 µL of Ringer's solution containing inositol tetrakisphosphate octakis(acetoxymethyl) ester and 2 µL of DMSO/5% Pluronic 127 was added to the basolateral reservoir. The cells were incubated at 37 °C for 30 min. The incubation mixture was discarded and the cells were washed with Ringer's solution (1 mL) and were ready for mounting. Cells for control experiments were incubated only with 100 µL of Ringer's solution with 2 µL of DMSO/5% Pluronic 127.

Synthesis

Enantiomerically pure 3,4,5,6-tetra-*O*-benzyl-*myo*-inositol (8) and 1,4,5,6-tetra-*O*-benzyl-*myo*-inositol (*ent*-8) were synthesized as described before.⁴⁰ *rac*-1,2-*O*-Cyclohexylidene-*myo*-inositol (*rac*-5) was synthesized by the method of Angyal and Tate.⁴³

General procedure for phosphorylation

The selectively protected *myo*-inositol derivative and tetrazole were dissolved under argon in dry acetonitrile before dibenzyl *N*,*N*-di*iso*propylphosphoramidite was added. After stirring at room temperature for the indicated time, the reaction mixture was cooled to $-40 \,^{\circ}\text{C}$ and peracetic acid (32% v/w; 1 mol equiv for each mol equiv of phosphoramidite) was added. After the mixture reached room temperature the solvent was removed under reduced pressure and the residual oil was purified by preparative HPLC to give the desired inositol tetra-kisphosphate derivative.

General procedure for removing the benzyl groups by hydrogenolysis

Either the fully protected *myo*-inositol tetrakisphosphate or the tetra-O-benzyl-*myo*-inositol derivative, respectively, was vigorously stirred with palladium on charcoal (10%; at least 0.1 mol palladium for each mol of benzyl groups) suspended in acetic acid under a hydrogen atmosphere in a self-built hydrogenation apparatus for the indicated time. The catalyst was removed by ultrafiltration and the filtrate was freezedried to give the respective product.

General procedure for the introduction of acetoxymethyl esters

The thoroughly dried inositol tetrakisphosphate derivative was suspended in dry acetonitrile under argon before dry DIEA (2.25 mol DIEA for each mol of hydroxy groups) and acetoxymethyl bromide (1 mol equiv for each mol equiv of DIEA) were added. After stirring the mixture at room temperature in the dark for 4 days, all volatile components were evaporated under reduced pressure and the crude residue was purified by preparative HPLC, with the solvent specified to give the inositol tetrakisphosphate octakis(acetoxymethyl) ester as a clear gum. rac-1,2-O-Cyclohexylidene-myo-inositol 3,4,5,6-tetrakis (dibenzyl)phosphate (rac-6). A solution of rac-1,2-Ocyclohexylidene-mvo-inositol (rac-5) (130 mg, 500 µmol) and tetrazole (350 mg, 5.00 mmol) in acetonitrile (6 mL) was treated with dibenzyl N,N-diisopropylphosphoramidite³⁹ (1.68 mL, 5.00 mmol) for 26 h and subsequently oxidized with peracetic acid at -40 °C. After the mixture warmed to room temperature the solvent was removed under reduced pressure and the residual oil was purified by preparative HPLC (93% MeOH; 40 mL/min; $t_{\text{R}} = 22.35 \text{ min}$) to give the fully protected compound rac-6 (306 mg, 47%) as a colorless gum. ¹H NMR (CDCl₃, 360 MHz): δ 1.20–1.80 (10H, m, CH₂ $[C_6H_{10}]$, 4.26 (1H, dd, J = 6.01, 6.01 Hz, H-1), 4.66 (1H, dd, J=6.01, 3.43 Hz, H-2), 4.76–4.81 (2H, m, H-3, H-5), 4.92–5.17 (18H, m, CH₂Ph, H-4, H-6), 7.20–7.40 (40H, m, CH₂Ph). ³¹P NMR (CDCl₃, ¹H decoupled, 145.8 MHz): δ -1.69 (1 P, s), -1.59 (1 P, s), -1.52 (1 P, s), -1.00 (1 P, s). MS: m/z (-ve ion FAB) 1209 $[(M-C_7H_7^+)^-, 1], 277 [OPO(OBn)_2^-, 100].$ HRMS (DCI): m/z 1209.312±10 ppm (M-C₇H₇⁺)⁻ (calcd for C₆₁H₆₅O₁₈P₄, 1209.312).

rac-1,2-O-Cyclohexylidene-myo-inositol 3,4,5,6-tetrakisphosphate (rac-7). rac-6 (91 mg, 70 µmol) was dissolved in dry ethanol (4 mL) before dry ethyl-N,N-diisopropylamine (95 µL, 560 µmol) was added, followed by palladium (10%) on charcoal (84 mg, 800 µmol). After stirring the solution at room temperature for 6 days under a hydrogen atmosphere the catalyst was removed by filtration and the filtrate was dried to give the ethyl-N,N-diisopropylammonium salt of tetrakisphosphate rac-7 (108 mg, 96%). ¹H NMR (D₂O, 360 MHz): δ 1.30–1.90 $(10H, m, CH_2 [C_6H_{10}]), 4.01(1H, ddd, J=9.08, 9.08)$ 8.23 Hz, H-5), 4.18-4.26 (2H, m, H-1, H-3), 4.32-4.40 (2H, m, H-4, H-6), 4.55 (1H, dd, J=4.26, 3.69 Hz, H-2). ³¹P NMR (D₂O, ¹H decoupled, 145.8 MHz): $\delta - 0.24$ (1 P, s), 0.10 (1 P, s), 0.85 (1 P, s), 0.90 (1 P, s). MS: m/z (+ve ion FAB) 581 [(M+H)⁺, 80], 81 [PO(OH)₂⁺, 100]. MS: m/z (-ve ion FAB) 579 [(M-H⁺)⁻, 100].

rac-1,2-O-Cyclohexylidene-myo-inositol 3,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester (rac-4). DIEA $(205\,\mu\text{L}, 1.20\,\text{mmol})$ and acetoxymethyl bromide (120 µL, 1.20 mmol) were added under argon to a suspension of compound rac-7 (108 mg, 66 µmol) in acetonitrile (2 mL). After stirring the reaction mixture in the dark for 4 days all volatile compounds were evaporated off under reduced pressure and the crude residue was purified by preparative HPLC (68% MeOH, 40 mL/ min, $t_{\rm R} = 19.35$) to give compound rac-4 (50 mg, 65%) as a syrup. ¹H NMR (toluene- d_8 , 360 MHz): δ 1.20–1.75 (10H, m, CH₂ [C₆H₁₀]), 1.81–1.92 (24H, 8 s, $8 \times OAc$), 4.28 (1H, dd, J = 5.51, 5.51 Hz, H-1). 4.77 (1H, dd, J=5.51, 3.54 Hz, H-2), 4.91-4.99 (2H, m, H-5, H-6), 5.07 (1H, ddd, J=8.66, 8.27, 3.54 Hz, H-3), 5.17 (1H, ddd, J=9.05, 8.66, 7.08 Hz, H-4), 5.65–5.94 (16H, m, CH₂OAc). ³¹P NMR (toluene- d_8 , ¹H decoupled, 145.8 MHz): δ -4.56 (1 P, s), -4.07 (1 P, s), -3.67 (1 P, s), -3.55 (1 P, s). MS: m/z (-ve ion FAB) 1083 $[(M-CH_2OAc^+)^-, 35], 241 [OPO(OCH_2OAc)_2^-, 100].$ HRMS: m/z 1157.168 [(M+H)⁺] (calcd for C₃₆H₅₇O₃₄P₄, 1157.167).

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D-3,4,5,6-Tetra-O-benzyl-1-O-butyl-myo-inositol (9). Dry 8 (250 mg, 463 µmol) and dry dibutyltin oxide (116 mg, 467 µmol) were heated under reflux in dry toluene (100 mL) in a Soxhlet apparatus with activated molecular sieves (3A) for 18h. The reaction mixture was cooled to room temperature and evaporated to dryness under diminished pressure. CsF (140 mg, 926 µmol) was added to the residual oil, and the mixture was kept under high vacuum for 2 h. The residual syrup was dissolved in dry DMF (10 mL) under argon and 1butyl iodide (300 µL, 2.62 mmol) was added. After stirring the solution for 48 h, HPLC analysis (90% MeOH; 1.5 mL/min; $t_{\rm R} = 7.43$ min) showed no further reaction. Excess of 1-butyl iodide and DMF were removed in high vacuum. The crude product was chromatographed by preparative HPLC (93% MeOH; 40 mL/min; $t_{\rm R} = 22.30$ min) to give compound 9 (175 mg, 74%) as a colorless solid. Mp: 75.4–75.9 °C (from methanol). $[\alpha]_D^{2+}$ $+8.4^{\circ}$ (c 0.98 in CHCl₃). ¹H NMR (CDCl₃, 360 MHz): δ 0.91 (3H, t, J=7.25 Hz, CH₃), 1.34–1.44 (2H, m, CH₂), 1.57–1.65 (2H, m, CH₂), 2.46 (1H, s (br), OH), 3.23 (1H, dd, J=9.31, 2.33 Hz, H-1), 3.43 (1H, dd, J=9.78, 2.80 Hz, H-3), 3.45 (1H, dd, J=9.78, 9.31 Hz, J = 6.99, H-4), 3.57 (1H, 6.52 Hz, dt, $OCH_2CH_2CH_2CH_3),$ 3.57 (1H, dt, J = 6.99 Hz, OCH₂CH₂CH₂CH₃), 3.91 (1H, dd, J=9.31, 9.31 Hz, H-5), 3.99 (1H, dd, J=9.78, 9.31 Hz, H-6), 4.27 (1H, dd, J = 2.80, 2.33 Hz, H-2), 4.71–4.94 (8H, m, CH_2Ph), 7.25–7.39 (20H, m, CH₂Ph). MS: m/z (+ve ion FAB) 597 [(M+H)⁺, 1], 91 [C₇H₇⁺, 100]. Anal. (C₃₈H₄₄O₆) C: calcd 76.48; found 76.87; H: calcd 7.43, found 7.50.

D-1,4,5,6-Tetra-O-benzyl-3-O-butyl-myo-inositol (ent-9). A similar reaction with the diol ent-8 gave compound ent-9. $[\alpha]_D^{24} - 8.7^\circ$ (c 1.01 in CHCl₃). Spectral data were in accordance with those obtained for enantiomer 9.

D-3,4,5,6-Tetra-O-benzyl-1-O-butyl-2-O-butyryl-myo-inositol (10). A solution of 9 (178 mg, 298 µmol), butyric anhydride (210 µL, 596 µmol), and DMAP (38 mg, 30 µmol) in dry pyridine (3 mL) was stirred at room temperature for 18 h. The solvents were evaporated under high vacuum to give an oil. Residual pyridine was removed by evaporating three times with octane. The residue was dissolved in *tert*-butyl methyl ether (10 mL) and was washed once with phosphate buffer (10 mL), once with sodium hydrogen carbonate (10 mL), once with sodium hydrogen sulfate (10 mL), once again with phosphate buffer (10 mL) and then with brine (10 mL). The organic layer was dried over Na₂SO₄ and filtered. Evaporation of the solvent gave pure **10** (194 mg, 98%) as a colorless oil. $[\alpha]_D^{24} - 10.4^{\circ}$ (c 1.97 in CHCl₃). ¹H NMR (CDCl₃, 360 MHz): δ 0.91 (3H, t, J = 7.25 Hz, CH₃), 0.98 (3H, t, *J* = 7.46 Hz, CH₃), 1.32–1.42 (2H, m, CH_2), 1.50–1.60 (2H, m, CH_2), 1.69 (2H, tq, J=7.46, 7.46 Hz, $O(O)CCH_2CH_2CH_3$), 2.39 (2H, t, J = 7.46 Hz, $O(O)CCH_2CH_2CH_3)$, 3.31 (1H, dd, J=9.66, 2.63 Hz, H-1). 3.44 J = 9.07, 6.81 Hz. (1H, dt. OCH₂CH₂CH₂CH₃), 3.48 (1H, dd, J=9.66, 3.07 Hz, H-3), 3.49 (1H, dd, J = 9.66, 9.22 Hz, H-5), 3.67 (1H, dt, J = 8.78, 6.81 Hz, OCH₂CH₂CH₂CH₃), 3.81 (1H, dd, J = 9.66, 9.66 Hz, H-4, 3.86 (1H, dd, J = 9.66, 9.22 Hz, H-6), 4.51–4.93 (8H, m, CH_2Ph), 5.83 (1H, dd, J=3.07, 2.63 Hz, H-2), 7.26–7.36 (20H, m, CH₂*Ph*). MS: m/z (+ve ion FAB) 667 [(M+H)⁺, 21], 91 [C₇H₇⁺, 100]. HRMS: m/z 667.364 [(M+H)⁺] (calcd for C₄₂H₅₁O₇, 667.363).

D-1,4,5,6-Tetra-*O*-benzyl-3-*O*-butyl-2-*O*-butyryl-myo-inositol (*ent*-10). Compound *ent*-9 was butyrylated as described above for the other enantiomer to give compound *ent*-10. $[\alpha]_D^{24}$ + 10.7° (*c* 2.05 in CHCl₃). HRMS: m/z 667.367 [(M + H)⁺] (calcd for C₄₂H₅₁O₇, 667.363). ¹H NMR and MS data were in accordance with those of enantiomer 10.

D-1-O-Butyl-2-O-butyryl-myo-inositol (11). Compound 10 (178 mg, 267 µmol) was hydrogenated with palladium (10%) on charcoal under hydrogen as described in the general procedure to give tetrol 11 (81 mg, 99%) as a solid after freeze-drying. Mp 125-125.9 °C (from MeOH). $[\alpha]_D^{24} + 26.5$ ° (c 2.0 in MeOH). ¹H NMR $(DMSO-d_6, \bar{3}60 \text{ MHz}): \delta 0.84 (3H, t, J = 7.38 \text{ Hz}, CH_3),$ $0.90 (3H, t, J = 7.38 \text{ Hz}, \text{CH}_3), 1.21 - 1.31 (2H, m, \text{CH}_2),$ 1.36-1.44 (2H, m, CH₂), 1.54 (2H, tq, J=7.38, 7.00 Hz, 2.24 (2H, t, J = 7.00 Hz, $O(O)CCH_2CH_2CH_3),$ $O(O)CCH_2CH_2CH_3)$, 2.97 (1H, dd, J=8.94, 8.55 Hz, H-5), 3.10 (1H, dd, J=9.72, 2.72 Hz, H-1), 3.25–3.35 (4H, m, H-3, H-4, H-6, OCH₂CH₂CH₂CH₃), 3.50 (1H, dt, J=8.94, 6.60 Hz, OCH₂CH₂CH₂CH₃), 4.85 (4H, s (br), OH), 5.36 (1H, dd, J=2.72, 2.33 Hz, H-2). MS: m/ z (+ve ion FAB) 307 [(M+H)⁺, 21], 71 [Bt⁺, 100]; m/z(-ve ion FAB) 305 [(M-H⁺)⁻, 27], 87 [BtO⁻, 100]. Anal. (C₁₄H₂₆O₇) C: calcd 54.89; found 54.45; H: calcd 8.55, found 8.56.

D-3-O-Butyl-2-O-butyryl-*myo***-inositol** (*ent***-11**). A similar reaction of the fully protected compound *ent***-10** afforded tetrol *ent***-11**. $[\alpha]_D^{24} - 26.6^\circ$ (*c* 0.76 in MeOH). ¹H NMR and MS data were in accordance with those obtained for enantiomer **11**.

D-1-O-Butyl-2-O-butyryl-myo-inositol 3,4,5,6-tetrakis(dibenzyl)phosphate (12). A solution of tetrol 11 (63 mg, 206 µmol) and tetrazole (174 mg, 2.47 mmol) in acetonitrile (2 mL) was treated with dibenzyl N,N-diisopropylphosphoramidite (834 µL, 2.47 mmol) for 18 h, oxidized with peracetic acid, and worked up as described above. Purification by preparative HPLC (93% MeOH; 40 mL/ min; $t_{\rm R} = 26.45$ min) gave compound **12** (165 mg, 60%) as a colorless gum. $[\alpha]_{\rm D}^{24} - 4.9^{\circ}$ (c 1.08 in CHCl₃). ¹H NMR (CDCl₃, 360 MHz): δ 0.77 (3H, t, J = 7.27 Hz, CH₃), 0.93 (3H, t, J=7.27 Hz, CH₃), 1.12–1.21 (2H, m, CH₂), 1.31–1.46 (2H, m, CH₂), 1.62 (2H, tq, J=7.26, 7.26 Hz, O(O)CCH₂CH₂CH₃), 2.24 (2H, t, J = 7.26 Hz, $O(O)CCH_2CH_2CH_3)$, 3.26 (1H, dt, J=8.23, 5.81 Hz, $OCH_2CH_2CH_2CH_3$), 3.37 (1H, dd, J=9.20, 2.90 Hz, H-1), 3.44 (1H, dt, J = 8.23, 7.26 Hz, OCH₂CH₂CH₂CH₂CH₃), 4.35 (1H, ddd, J = 9.69, 9.69, 2.42 Hz, H-3), 4.53 (1H, ddd, J = 9.69, 9.69, 9.69 Hz, H-5), 4.68 (1H, ddd, $J = 9.69, 9.69, 9.20 \,\mathrm{Hz}, \mathrm{H-6}, 4.91 \,(\mathrm{1H}, \mathrm{ddd}, \mathrm{H})$ J = 9.84, 9.69, 9.69 Hz, H-4, 4.92-5.02 (16H, m, CH_2Ph), 5.91 (1H, dd, J = 2.90, 2.42 Hz, H-2), 7.11–7.30 (40H, m, CH₂Ph). ³¹P NMR (CDCl₃, ¹H decoupled, 145.8 MHz): δ -1.81 (1 P, s), -1.18 (1 P, s), -0.72 (1 P, s), -0.66 (1 P, s). MS: m/z (+ve ion FAB) 1347 $[(M+H)^+, 8]$, 91 $[C_7H_7^+, 100]$. HRMS: m/z 1347.416 $[(M+H)^+]$ (calcd for $C_{70}H_{79}O_{19}P_4$, 1347.417).

D-3-O-Butyl-2-O-butyryl-myo-inositol 1,4,5,6-tetrakis (dibenzyl)phosphate (*ent*-12). Tetrol *ent*-11 was phosphitylated and oxidized as described above for compound 12 to give the fully protected phosphate *ent*-12. $[\alpha]_D^{24}$ +4.8° (*c* 2.09 in CHCl₃). HRMS (DCI): *m/z* 1255.345±10 ppm $[(M-C_7H_7^+)^-]$ (calcd for $C_{63}H_{79}O_{19}P_4$, 1255.345). ¹H NMR, ³¹P NMR, and MS data were in accordance with those obtained for enantiomer 12.

D-1-O-Butyl-2-O-butyryl-*myo*-inositol 3,4,5,6-tetrakisphosphate (13). Compound 12 (160 mg, 118 µmol) was hydrogenated with palladium (10%) on carbon as described in the general procedure to give compound 13 (73 mg, 99%) as a solid after freeze-drying. $[\alpha]_D^{24} + 4.1^\circ$ (c 1.04 in H₂O, pH 1.6). ¹H NMR (D₂O, 360 MHz): δ $0.81 (3H, t, J = 7.30 \text{ Hz}, \text{CH}_3), 0.91 (3H, t, J = 7.30 \text{ Hz},$ CH₃), 1.22–1.30 (2H, m, CH₂), 1.42–1.49 (2H, m, CH₂), 1.62 (2H, tq, J = 7.30, 7.30 Hz, O(O)CCH₂CH₂CH₃), 2.36-2.54 (2H, m, O(O)CCH₂CH₂CH₃), 3.56 (1H, dt, $J = 9.49, 6.46 \text{ Hz}, \text{ OC}H_2\text{C}H_2\text{C}H_2\text{C}H_3), 3.63 (1\text{H}, \text{dt}, \text{dt})$ $J = 9.36, 6.74 \text{ Hz}, \text{ OC}H_2\text{C}H_2\text{C}H_2\text{C}H_3), 3.70 (1\text{H}, \text{ dd}, \text{C}H_2\text{C}H_2\text{C}H_3)$ J=9.74, 2.62 Hz, H-1), 4.27 (1H, ddd, J=9.36, 9.36, 9.36 Hz, H-5), 4.31 (1H, ddd, J=9.74, 9.74, 2.25 Hz, H-3), 4.39 (1H, ddd, J=9.74, 9.36, 9.36 Hz, H-6), 4.49 (1H, ddd, J=9.74, 9.36, 9.00 Hz, H-4), 5.75 (1H, dd, J = 2.62, 2.25 Hz, H-2), ³¹P NMR (D₂O, ¹H decoupled, 145.8 MHz): δ -0.10 (2 P, s), 0.50 (1 P, s), 0.80 (1 P, s). MS: m/z (+ve ion FAB) 627 [(M+H)⁺, 4], 71 [Bt⁺, 100]. MS: m/z (-ve ion FAB) 625 [(M-H⁺)⁻, 100]. HRMS: m/z625.025 $[(M-H^+)^-]$ (calcd for C₁₄H₂₉O₁₉P₄, 625.025).

D-3-O-Butyl-2-O-butyryl-*myo***-inositol** 1,4,5,6-tetrakis**phosphate** (*ent*-13). A similar reaction with the fully protected compound *ent*-12 afforded the free acid *ent*-13 after freeze-drying. $[\alpha]_D^{24} - 4.1^\circ$ (*c* 0.78 in H₂O, pH 1.6). HRMS: *m/z* 625.023 [(M-H⁺)⁻] (calcd for C₁₄H₂₉O₁₉P₄, 625.025). ¹H NMR and MS data were in accordance with those obtained for enantiomer 13.

D-1-O-Butyl-myo-inositol 3,4,5,6-tetrakisphosphate (14). An aqueous solution of compound **13** (17 mg, 27 µmol) was treated with 1 M KOH ($260 \,\mu$ L) to adjust the pH to 12.8. The solution was stirred at room temperature for 2 days. The reaction mixture was directly poured onto an ion-exchange column (Dowex 50 WX 8, H⁺) for purification. Lyophilization gave compound 14 (14 mg, 94%). $[\alpha]_D^{24}$ + 4.9° (c 0.53 in H₂O, pH 1.6). ¹H NMR $(D_2O, 360 \text{ MHz})$: $\delta 0.78 (3H, t, J = 7.44 \text{ Hz}, CH_3), 1.26$ $(2H, tq, J=7.44, 7.44 Hz, CH_2), 1.43-1.52$ (2H, m, m) CH_2), 3.42 (1H, dd, J=9.60, 2.78 Hz, H-1), 3.52 (1H, dt, J = 9.71, 6.74 Hz, OCH₂CH₂CH₂CH₃), 3.59 (1H, dt, $J = 9.81, 6.74 \text{ Hz}, \text{ OCH}_2\text{CH}_2\text{CH}_2\text{CH}_3), 4.16 (1\text{H}, \text{ ddd}, \text{H})$ J = 9.60, 9.60, 2.78 Hz, H-3), 4.21 (1H, ddd, J = 9.37, 9.37, 9.37 Hz, H-5), 4.35 (1H, dd, J=2.78, 2.78 Hz, H-2), 4.39 (1H, ddd, J=9.60, 9.60, 9.37 Hz, H-6), 4.46 (1H, ddd, J=9.60, 9.60, 9.37 Hz, H-4). ³¹P NMR (D₂O, ¹H decoupled, 145.8 MHz): δ 0.10 (2 P, s), 0.70 (1 P, s), 1.05 (1 P, s); m/z (-ve ion FAB) 555 [(M-H⁺)⁻, 100].

HRMS: m/z 554.979 [(M-H⁺)⁻] (calcd for $C_{10}H_{23}O_{18}P_4$, 554.984).

D-3-*O***-Butyl-***myo***-inositol 1,4,5,6-tetrakisphosphate** (*ent***-14**). The butyryl group of *ent***-13** was hydrolyzed by the same method described above to give the tetrakisphosphate *ent***-14**. $[\alpha]_D^{24} - 4.6^\circ$ (*c* 0.35 in H₂O, pH 1.6). HRMS: m/z 554.982 [(M–H⁺)⁻] (calcd for C₁₀H₂₃O₁₈P₄, 554.984). ¹H NMR, ³¹P NMR, and MS data were in accordance with those obtained for enantiomer 14.

D-1-O-Butyl-2-O-butyryl-*myo*-inositol 3,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester (1). DIEA (131 μ L, 768 μ mol) and acetoxymethyl bromide (77 μ L, 768 µmol) were added to a suspension of compound 13 (30 mg, 47 µmol) in acetonitrile (2 mL) as described in the general procedure. Purification by preparative HPLC (73% MeOH, 37.5 mL/min, $t_R = 19.30$) gave compound 1 (31 mg, 55%) as a colorless syrup. $[\alpha]_D^{24}$ -2.0° (c 1.00 in toluene). ¹H NMR (toluene- d_8 , 360 MHz): δ 0.85 (3H, t, J=7.48 Hz, CH₃), 0.97 (3H, t, $J = 7.48 \text{ Hz}, \text{ CH}_3$, 1.39 (2H, tq, $J = 7.48, 7.28 \text{ Hz}, \text{ CH}_2$), 1.51–1.67 (4H, m, 2 \times CH₂), 1.75–1.75 (24H, 8 s, 8 \times OAc), 2.08–2.13 (2H, m, CH₂), 3.07 (1H, dd, J=9.45, 2.76 Hz, H-1), 3.46 (1H, dt, J = 7.68, 5.91 Hz, $OCH_2CH_2CH_2CH_3$), 3.62 (1H, dt, J = 7.74, 7.68 Hz, OCH₂CH₂CH₂CH₃), 4.62 (1H, ddd, J=9.84, 9.84, 2.76 Hz, H-3), 4.67 (1H, ddd, J=9.84, 9.84, 9.45 Hz, H-5), 4.80 (1H, ddd, J=9.45, 9.45, 9.16 Hz, H-6), 5.04 (1H, ddd, J=9.84, 9.84, 9.45 Hz, H-4), 5.63-5.96 (16H, m, CH₂OAc), 6.00 (1H, dd, J = 2.76, 2.76 Hz, H-2). ³¹P NMR (toluene- d_8 , ¹H decoupled, 145.8 MHz): δ -5.14 (1 P, s), -4.49 (1 P, s), -4.06 (1 P, s), -3.99 (1 P, s).MS: m/z (+ve ion FAB) 1131 [(M-CH₂OAc⁺+2H)⁺, 58], 987 [(M-3 CH₂OAc⁺ + 4H)⁺, 100], m/z (-ve ion $[(M-CH_2OAc^+)^-,$ 1129 FAB) 38], 241 $[OPO(OCH_2OAc)_2^-, 100]$. HRMS: m/z 1131.189 $[(M-CH_2OAc^++2H)^+]$ (calcd for $C_{35}H_{59}O_{33}P_4$, 1131.189).

D-3-O-Butyl-2-O-butyryl-*myo***-inositol** 1,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester (*ent*-1). Alkylation of the phosphate *ent*-21 as described above afforded the octakis(acetoxymethyl) ester *ent*-1. $[\alpha]_D^{24}$ +1.9° (*c* 1.46 in toluene). HRMS: *m/z* 1129.171 [(M-CH₂OAc⁺)⁻] (calcd for C₃₅H₅₇O₃₃P₄, 1129.173). ¹H NMR, ³¹P NMR, and MS data were in accordance with those obtained for enantiomer 1.

D-1-O-AllyI-3,4,5,6-tetra-O-benzyI-*myo***-inositol** (15). Dry **8** (690 mg, 1.28 mmol) and dry dibutyltin oxide (324 mg, 1.3 mmol) were heated to reflux in dry toluene (150 mL) in a Soxhlet apparatus with activated molecular sieves (3 Å) for 20 h. The reaction mixture was cooled to room temperature and evaporated to dryness under diminished pressure. CsF (388 mg, 2.56 mol) was added to the residual oil, and the mixture was kept under high vacuum for 2 h. The residual syrup was dissolved in dry DMF (10 mL) under argon and 1-allyl iodide (329 μ L, 3.58 mmol) was added. After stirring the solution for 20 h, HPLC analysis (95% MeOH; 1.5 mL/min; $t_{\rm R}$ = 3.20 min) showed no further reaction. Excess

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of 1-allyl iodide and DMF were removed in high vacuum. The crude product was chromatographed by (90% preparative HPLC MeOH; $40 \,\mathrm{mL/min};$ $t_{\rm R} = 26.15 \,{\rm min}$) to give compound 15 (448 mg, 60%) as a solid. Mp 71–72 °C (from methanol). $[\alpha]_D^{24}$ +4.6° (*c* 0.98 in CHCl₃). ¹H NMR (CDCl₃, 360 MHz): δ 3.30 (1H, dd, J=9.73, 3.10 Hz, H-1), 3.41 (1H, dd, J=9.73, 2.65, H-3), 3.44 (1H, dd, J=9.51, 9.51, H-5), 3.95 (1H, dd, J=9.73, 9.51 Hz, H-6), 3.99 (1H, dd, J=9.73, 2.51 Hz, H-4), 4.18 (1H, dd, J=1.65, 1.33 Hz, OCH₂CHCH₂), 4.19 (1H, dd, J=2.65, 1.33 Hz, OCH₂CHCH₂), 4.20 (1H, dd, J=3.10, 2.65 Hz, H-2), 4.71-4.92 (8H, m, CH_2Ph), 5.19 (1H, ddt, J=11.50, 1.33, 1.33 Hz, OCH₂CHCH₂), 5.29 (1H, ddt, J=16.91, 2.65, 1.33 Hz, OCH₂CHCH₂), 5.94 (1H, dddd, J=16.91, 11.50, 1.33, 1.33 Hz, OCH₂CHCH₂), 7.26–7.37 (20H, m, CH₂Ph). MS: m/z (+ve ion FAB) 581 [(M+H)⁺, 1], 91 [C₇H₇⁺, 100], m/z (-ve ion FAB) 580 [(M-H⁺)⁻, 6], 489 $[(M-C_7H_7^+)^-, 100].$

D-3-O-Allyl-1,4,5,6-tetra-O-benzyl-*myo***-inositol** (*ent***-15**). A similar reaction and workup of the diol *ent***-8** gave compound *ent***-15**. $[\alpha]_D^{24}$ -3.9° (*c* 0.82 in CHCl₃). ¹H NMR and MS data were in accordance with those obtained for enantiomer **15**.

D-1-O-Allyl-3,4,5,6-tetra-O-benzyl-2-O-butyl-myo-inosi-

tol (16). Sodium hydride (46 mg, 1.92 mmol) was added to a stirred solution of 15 (445 mg, 767 µmol) in dry DMF (5mL) at room temperature in the dark. The mixture was stirred for 5h, after which 1-butyl iodide (306 µL, 2.68 mmol) was added. The suspension was stirred for 18 h at 80 °C when HPLC (95% MeOH; 1.5 mL/min; $t_{\rm R}$ = 7.35) showed a complete reaction. Excess of 1-butyl iodide and DMF were evaporated off under reduced pressure. The mixture was then dissolved in tert-butyl methyl ether (40 mL) and washed once with phosphate buffer (20 mL), aq sodium dithionate (20 mL) and brine (20 mL), successively. The organic layer was dried over Na₂SO₄, filtered, and the ether was evaporated off to give an oil. The crude oil was purified by HPLC (93% preparative MeOH; $40 \,\mathrm{mL/min};$ $t_{\rm R} = 37.10$ min) to give the title compound 16 (458 mg, 94%) as an oil. $[\alpha]_{\rm D}^{24} + 1.5^{\circ}$ (c 2.29 in CHCl₃). ¹H NMR (CDCl₃, 360 MHz): δ 0.95 (3H, t, J = 7.27 Hz, CH₃), 1.39–1.49 (2H, m, CH₂), 1.58–1.66 (2H, m, CH₂), 3.24 (1H, dd, J=9.69 2.42 Hz, H-1), 3.36 (1H, dd, J=9.93)2.42, H-3), 3.46 (1H, dd, J = 9.45, 9.45, H-5), 3.78 (2H, t, J = 6.54 Hz, $OCH_2CH_2CH_2CH_3$), 3.89 (1H, dd, J=2.42, 2.42 Hz, H-2), 3.98 (1H, dd, J=9.45, 9.45 Hz, H-6), 4.03 (1H, dd, J=9.45, 9.45 Hz, H-4), 4.15 (1H, dd, $J = 5.81, 1.00 \text{ Hz}, \text{ OCH}_2\text{CHCH}_2), 4.16 (1\text{H}, \text{dd}, J = 6.06),$ 1.00 Hz, OCH₂CHCH₂), 4.72–4.95 (8H, m, CH₂Ph), 5.19 (1H, ddt, J = 10.66, 1.45, 1.45 Hz, OCH₂CHCH₂), 5.32 (1H, ddt, J = 16.95, 1.45, 1.45 Hz, OCH₂CHCH₂), 5.95 (1H, dddd, J = 16.95, 10.66, 1.45, 1.45 Hz, OCH₂CHCH₂), 7.26–7.40 (20H, m, CH₂Ph). MS: m/z (+ ve ion FAB) 637 $[(M+H)^+, 1]$, 91 $[C_7H_7^+, 100]$. HRMS: m/z 637.373 [(M+H)⁺] (calcd C₄₁H₄₉O₆, 637.353).

D-3-O-Allyl-1,4,5,6-tetra-O-benzyl-2-O-butyl-myo-inositol (ent-16). A similar reaction and workup of the com-

pound *ent*-**15** gave compound *ent*-**16**. $[\alpha]_D^{24} - 1.6^\circ$ (*c* 1.87 in CHCl₃). HRMS: m/z 637.357 $[(M + H)^+]$ (calcd C₄₁H₄₉O₆, 637.353). ¹H NMR and MS data were in accordance with those obtained for enantiomer **16**.

D-3,4,5,6-Tetra-O-benzyl-2-O-butyl-myo-inositol (17). Tris(triphenylphosphine)-rhodium(I) chloride (140 mg, 150 µmol) and DIEA (25 µL, 140 µmol) were added to a suspension of 16 (458 mg, 720 µmol) in 50% ethanol (90 mL). The suspension was heated to reflux for 7 h. The mixture was cooled to room temperature before trifluoroacetic acid (7 mL) was added and the solution was stirred for additional 24 h, when HPLC analysis (90% MeOH; 1.5 mL/min; $t_R = 4.03 \text{ min}$) showed no more starting material. After neutralization with aq NH₃ (2 N) the solvents were evaporated under reduced pressure to give a syrup. The syrup was dissolved in tertbutyl methyl ether (40 mL) and washed once with phosphate buffer (20 mL) and brine (20 mL), successively. The organic layer was dried over Na₂SO₄, filtered and the ether was evaporated off to give an oil. The crude oil was purified by preparative HPLC (92% MeOH; 40 mL/min; $t_{\rm R} = 27.40$ min) to give 17 (275 mg, 74%) as a solid. Mp 112.3–113.1 °C (from MeOH). $[\alpha]_{\rm P}^{24} + 24.6^{\circ}$ (c 0.97 in CHCl₃). ¹H NMR (CDCl₃, 360 MHz): $\delta 0.92 (3H, t, J = 7.36 \text{ Hz}, \text{ CH}_3), 1.32-1.43$ (2H, m, CH₂), 1.53–1.61 (2H, m, CH₂), 3.41 (1H, dd, J=9.96, 2.60 Hz, H-1), 3.45 (1H, dd, J=9.52, 2.60 Hz, H-3), 3.46 (1H, dd, J=9.31, 9.31 Hz, H-5), 3.61 (1H, dt, J = 9.09, 6.49 Hz, OC H_2 CH $_2$ CH $_2$), 3.74 (1H, dd, J=9.52, 9.31 Hz, H-4), 3.86 (1H, dd, J=2.60, 2.60 Hz, H-2), 3.95 (1H, dt, J = 8.95, 6.49 Hz, $OCH_2CH_2CH_2$), 3.98 (1H, dd, J=9.96, 9.31 Hz, H-6), 4.79–4.94 (8H, m, CH_2Ph), 7.26–7.36 (20H, m, CH_2Ph). MS: m/z (+ve ion FAB) 597 [(M+H)⁺, 1], 91 [C₇H₇⁺, 100]. MS: m/z (-ve ion FAB) 595 $[(M-H^+)^-, 100]$, $[(M-C_7H_7^+)^-, 20]$. Anal. (C₃₈H₄₄O₆) C: calcd 76.48; found 76.52; H: calcd 7.43, found 7.40.

D-1,4,5,6-Tetra-*O*-benzyl-2-*O*-butyl-myo-inositol (*ent*-17). A similar reaction of compound *ent*-16 gave *ent*-17. $[\alpha]_D^{24} - 24.9^\circ$ (*c* 1.00 in CHCl₃). ¹H NMR and MS data were in accordance with those obtained for enantiomer 17.

D-3,4,5,6-Tetra-O-benzyl-2-O-butyl-1-O-butyryl-myo-inositol (18). A solution of alcohol 17 (178 mg, 298 µmol) in dry pyridine (4 mL) was treated with butyric anhydride (158 µL, 447 µmol) and DMAP (38 mg, 29 µmol) and stirred at room temperature. After 18h, HPLC analysis (90% MeOH; 1.5 mL/min, $t_{\text{R}} = 6.40 \text{ min}$) showed no more starting material. The reaction mixture was evaporated under reduced pressure to give a crude oil. To remove residual pyridine the oil was dissolved in octane and evaporated three times. The residue was dissolved in tert-butyl methyl ether (20 mL) and was washed once with phosphate buffer (10 mL), once with sodium hydrogen carbonate (10 mL), once with sodium hydrogen sulfate (10 mL), once again with phosphate buffer (10 mL), and then with brine (10 mL). The organic layer was dried over Na₂SO₄ and filtered. Evaporation of the solvent gave pure **18** (176 mg, 89%) as a colorless oil. $[\alpha]_D^{24}$ -15.4° (c 1.00 in CHCl₃). ¹H NMR (CHCl₃, 360 MHz): δ 0.91 (3H, t, J=7.21 Hz, CH₃), 0.93 (3H, t, J=7.21 Hz, CH₃), 1.35–1.56 (4H, m, 2 × CH₂), 1.58–1.72 (2H, m, CH₂), 2.21–2.26 (2H, m, CH₂), 3.48 (1H, dd, J=9.61, 2.40 Hz, H-3), 3.51 (1H, dd, J=9.66, 9.37 Hz, H-5), 3.52 (2H, tq, J=8.97, 6.25 Hz, O(O)CCH₂CH₂CH₃), 3.76 (2H, t, J=9.13, 6.25 Hz, O(O)CCH₂CH2CH₃), 3.94 (1H, dd, J=2.40, 2.40 Hz, H-2), 4.00 (1H, dd, J=9.61, 9.61 Hz, H-4), 4.02 (1H, dd, J=9.85, 9.31 Hz, H-6), 4.73 (1H, dd, J=9.85, 2.40 Hz, H-1), 4.65–4.93 (8H, m, CH₂Ph), 7.25–7.34 (20H, m, CH₂Ph). MS: m/z (+ve ion FAB) 667 [(M + H)⁺, 1], 91 [C₇H₇⁺, 100]. HRMS: m/z 667.368 [(M + H)⁺] (calcd for C₄₂H₅₁O₇, 667.363).

D-1,4,5,6-Tetra-*O***-benzyl-***2***-***O***-butyrl-***3***-***O***-butyryl-***myo***-inositol** (*ent***-18**). Compound *ent***-17** was butyrylated as described above for the other enantiomer to give compound *ent***-18**. $[\alpha]_D^{24} + 15.9^\circ$ (*c* 1.10 in CHCl₃). HRMS: m/z 667.362 [(M + H)⁺] (calcd for C₄₂H₅₁O₇, 667.363). ¹H NMR and MS data were in accordance with those of enantiomer **18**.

D-2-O-Butyl-1-O-butyryl-myo-inositol (19). Compound 18 (170 mg, 255 µmol) was hydrogenated with palladium (10%) on carbon under hydrogen as described in the general procedure to give tetrol 19 (75 mg, 97%) as a solid after freeze-drying. Mp 131.2–131.8 °C (from ethanol). $[\alpha]_D^{24}$ +41.9° (c 1.10 in MeOH). ¹H NMR (DMSO-*d*₆, 360 MHz): δ 0.87 (3H, t, *J* = 7.16 Hz, CH₃), 0.89 (3H, t, J = 7.33 Hz, CH₃), 1.28–1.48 (4H, m, 2 × CH₂), 1.52–1.62 (2H, m, CH₂), 2.24–2.36 (2H, m, CH₂), 2.96 (1H, dd, J=9.21, 9.21 Hz, H-5), 3.26 (1H, dd, J=9.55, 2.39 Hz, H-3), 3.35 (1H, dd, J=9.55, 9.21 Hz, H-4), 3.44 (1H, dt, J=9.43, 6.39 Hz, $OCH_2CH_2CH_2CH_3$), 3.52 (1H, dd, J = 10.23, 9.21 Hz, H-6), 3.57 (1H, dd, J = 2.39, 2.39 Hz, H-2), 4.47 (1H, dt, $J = 9.21, 6.39 \text{ Hz}, \text{ OC}H_2\text{C}H_2\text{C}H_2\text{C}H_3), 4.47 (1\text{H}, \text{dd}, \text{C}H_2\text{C}H_2\text{C}H_3)$ J=10.23, 2.39 Hz, H-1), 4.71 (2H, s (br), OH), 4.82 (1H, s (br), OH), 4.86 (1H, s (br), OH), MS: m/z (+ve ion FAB) 307 [(M + H)⁺, 100]. MS: m/z (-ve ion FAB) 305 [(M-H⁺)⁻, 34], 87 [BtO⁻, 100]. Anal. (C₁₄H₄₄O₆) C: calcd 54.89; found 54.90; H: calcd 8.55, found 8.51.

D-2-O-Butyl-3-O-butyryl-*myo***-inositol** (*ent***-19**). A similar reaction of the fully protected compound *ent***-18** afforded tetrol *ent***-19**. $[\alpha]_D^{24} - 40.5^\circ$ (*c* 1.00 in MeOH). ¹H NMR and MS data were in accordance with those obtained for enantiomer **19**.

D-2-*O*-Butyl-1-*O*-butyryl-*myo*-inositol 3,4,5,6-tetrakis(dibenzyl)phosphate (20). A solution of compound 19 (55 mg, 178 μmol) and tetrazole (152 mg, 2.15 mmol) in acetonitrile (2 mL) was treated with dibenzyl *N*,*N*-di*iso*-propylphosphoramidite (726 μL, 2.15 mmol) for 22 h, then oxidized with peracetic acid, and worked up as described. Purification by preparative HPLC (92% MeOH; 40 mL/min; $t_{\rm R}$ = 29.00 min) gave compound 20 (192 mg, 80%) as a colorless oil. [α]_D²⁴ - 3.4° (*c* 1.02 in CHCl₃). ¹H NMR (CDCl₃, 360 MHz): δ 0.79 (3H, t, *J* = 7.30 Hz, CH₃), 0.86 (3H, t, *J* = 7.30 Hz, CH₃), 1.23-1.38 (2H, m, CH₂), 1.41–1.57 (4H, m, 2 × CH₂), 2.01–2.17(2H, m, CH₂), 3.54 (1H, dt, *J* = 8.52, 7.74 Hz, OCH₂CH₂CH₂CH₃), 3.60 (1H, dt, *J* = 8.52, 7.74 Hz,

OC H_2 CH₂CH₂CH₃), 4.13 (1H, dd, J = 2.52, 2.52 Hz, H-2), 4.22 (1H, ddd, J = 9.73, 9.73, 2.52 Hz, H-3), 4.43 (1H, ddd, J = 9.51, 9.51, 9.51 Hz, H-5), 4.44 (1H, ddd, J = 9.51, 9.51, 9.28 Hz, H-6), 4.89 (1H, ddd, J = 9.73, 9.73, 9.51 Hz, H-4), 4.91–5.08 (17H, m, CH₂Ph, H-1), 7.13–7.27 (40H, m, CH₂Ph). ³¹P NMR (CDCl₃, ¹H decoupled, 145.8 MHz): δ – 1.64 (1 P, s), -1.42 (1 P, s), -0.76 (1 P, s), -0.53 (1 P, s). MS: m/z (+ve ion FAB) 1347 [(M+H)⁺, 33], 71 [Bt⁺, 100], m/z (–ve ion FAB) 1255 [(M–C₇H₇⁺)⁻, 16], 277 [OPO(OBn)₂⁻, 100]. HRMS: m/z 1347.418 [(M+H)⁺] (calcd for C₇₀H₇₉O₁₉P₄, 1347.417).

D-2-O-Butyl-3-O-butyryl-*myo***-inositol 1,4,5,6-tetrakis(dibenzyl)phosphate (***ent***-20)**. Compound *ent***-19** was phosphitylated and oxidized as described above for compound **19** to give the fully protected phosphate *ent*-**20**. $[\alpha]_D^{24} + 3.1^\circ$ (*c* 1.10 in CHCl₃). HRMS (DCI): *m/z* 1255.354±10 ppm $[(M-C_7H_7^+)^-]$ (calcd for $C_{63}H_{71}O_{19}P_4$, 1255.354). ¹H NMR and MS data were in accordance with those obtained for enantiomer **20**.

D-2-O-Butyl-1-O-butyryl-*myo*-inositol 3,4,5,6-tetrakisphosphate (21). Compound 20 (190 mg, 141 µmol) was hydrogenated with palladium (10%) on carbon as described in the general procedure to give title compound 21 (87 mg, 99%) as a solid after freeze-drying. $[\alpha]_{D}^{24}$ +9.9° (c 1.10 in H₂O, pH 1.6). ¹H NMR (D₂O, 360 MHz): δ 0.75 (3H, t, *J* = 7.15 Hz, CH₃), 0.76 (3H, t, $J = 7.31 \text{ Hz}, \text{ CH}_3$, 1.19–1.29 (2H, m, CH₂), 1.38– 1.53 (4H, m, 2 \times CH₂), 2.30 (2H, t, J=7.47, $O(O)CCH_2CH_2CH_3)$, 3.58 (1H, dt, J=9.64, 6.44 Hz, $OCH_2CH_2CH_2CH_3$), 3.67 (1H, dt, J=9.64, 6.36 Hz, $OCH_2CH_2CH_2CH_3$, 4.01 (1H, dd, J = 2.54, 2.23 Hz, H-2), 4.22 (1H, ddd, J=9.22, 9.22, 8.90 Hz, H-5), 4.23 (1H, ddd, J=9.54, 9.54, 2.54 Hz, H-3), 4.43 (1H, ddd, J=9.90, 9.22, 9.22 Hz, H-6), 4.48 (1H, ddd, J=9.54, 9.54, 9.54 Hz, H-4), 4.94 (1H, dd, J=9.90, 2.23 Hz, H-1). ³¹P NMR (D₂O, ¹H decoupled, 145.8 MHz): δ -0.20 (2 P, s), 0.40 (1 P, s), 0.50 (1 P, s). MS: m/z (+ve ion FAB) 627 $[(M+H)^+, 8]$, 71 $[Bt^+, 100]$, m/z (-ve ion FAB) 625 $[(M-H^+)^-, 35]$, 79 $[OP(O)_2^-, 100]$. HRMS: m/z 625.030 [(M-H⁺)⁻] (calcd for C₁₄H₂₉O₁₉P₄, 625.025).

D-2-O-Butyl-3-O-butyryl-*myo***-inositol 1,4,5,6-tetrakisphosphate** (*ent-21*). A similar reaction with the fully protected substrate *ent-20* afforded the free acid *ent-21* after freeze-drying. $[\alpha]_D^{24} - 9.7^\circ$ (*c* 1.00 in H₂O, pH 1.6). HRMS: m/z 625.027 [(M-H⁺)⁻] (calcd for C₁₄H₂₉O₁₉P₄, 625.025). ¹H NMR and MS data were in accordance with those obtained for enantiomer **21**.

D-2-*O***-Butyl-***myo***-inositol 3,4,5,6-tetrakisphosphate (22).** Compound **21** (33 mg, 52 μmol) was treated with 1 M KOH (453 μL) to adjust the pH-value to 12.8. The solution was stirred at room temperature for 2 days. The reaction mixture was directly poured onto an ion-exchange column (Dowex 50 WX 8, H⁺) for purification. Lyophilization gave compound **22** (27 mg, 93%). $[\alpha]_D^{24} - 1.1^\circ$ (*c* 0.89 in H₂O, pH 1.6). ¹H NMR (D₂O, 360 MHz): δ 0.77 (3H, t, *J*=7.37 Hz, CH₃), 1.21–1.31 (2H, m, CH₂), 1.39–1.52 (2H, m, CH₂), 3.63 (1H, dt, J=9.30, 6.18 Hz, OCH₂CH₂CH₂CH₃), 3.67 (1H, dd, J=10.00, 2.63 Hz, H-1), 3.75 (1H, dt, J=9.30, 6.56 Hz, OCH₂CH₂CH₂CH₃), 3.94 (1H, dd, J=2.63, 2.37 Hz, H-2), 4.13 (1H, ddd, J=9.21, 9.21, 9.21 Hz, H-5), 4.17 (1H, ddd, J=9.47, 9.47, 2.37 Hz, H-3), 4.30 (1H, ddd, J=9.47, 9.47, 9.21 Hz, H-4), 4.43 (1H, ddd, J=10.00, 9.73, 9.21 Hz, H-6). ³¹P NMR (D₂O, ¹H decoupled, 145.8 MHz): δ -0.18 (1 P, s), 0.55 (2 P, s), 0.95 (1 P, s). MS: m/z (+ve ion FAB) 557 [(M+H⁺)⁺, 100], m/z(-ve ion FAB) 555 [(M-H⁺)⁻, 100]. HRMS: m/z554.983 [(M-H⁺)⁻] (calcd for C₁₀H₂₃O₁₈P₄, 554.984).

D-2-O-Butyl-*myo***-inositol 1,4,5,6-tetrakisphosphate** (*ent*-**22**). The butyryl group of substrate *ent***-21** was hydrolyzed by the method described above to give the tetrakisphosphate *ent*-**22**. $[\alpha]_D^{24} + 1.6^\circ$ (*c* 0.60 in H₂O, pH 1.6). HRMS: *m*/*z* 554.985 $[(M-H^+)^-]$ (calcd for $C_{10}H_{23}O_{18}P_4$, 554.984). ¹H NMR and MS data were in accordance with those obtained for enantiomer **22**.

D-2-O-Butyl-1-O-butyryl-*myo*-inositol 3,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester (2). DIEA $(182\,\mu\text{L}, 1.06\,\text{mmol})$ and acetoxymethyl bromide (107 μ L, 1.06 mmol) were added to a suspension of compound **21** (37 mg, 59 µmol) in acetonitrile (2 mL) as described in the general procedure. Purification by preparative HPLC (72%) MeOH, $40 \,\mathrm{mL/min}$, $t_{\rm R} = 21.12 \text{ min}$) gave compound 2 (33 mg, 46%) as a syrup. $\left[\alpha\right]_{D}^{24}$ +1.1° (c 1.07 in toluene). ¹H NMR (toluene- d_8 , 360 MHz): δ 0.93 (3H, t, J = 7.28 Hz, CH₃), 0.97 (3H, t, J=7.48 Hz, CH₃), 1.39–1.49 (2H, m, CH₂), 1.51– 1.60 (2H, m, CH₂), 1.77–1.86 (24H, 8 s, $8 \times OAc$), 2.39 $(1H, dt, J=16.67, 7.38 Hz, CH_2), 2.63 (1H, dt,$ $J = 16.67, 7.68, CH_2$, 3.78 (1H, dt, J = 9.45, 6.30 Hz, $OCH_2CH_2CH_2CH_3$), 3.87 (1H, dt, J=9.06, 6.30 Hz, OCH₂CH₂CH₂CH₃), 4.40 (1H, dd, J=2.36, 2.36 Hz, H-2), 4.68 (1H, ddd, J = 9.55, 9.55, 2.36 Hz, H-3), 4.79 (1H, ddd, J=9.55, 9.45, 9.45 Hz, H-5), 5.03 (1H, ddd, J = 9.55, 9.55, 9.55 Hz, H-4, 5.09 (1H, ddd, J = 10.04, 9.55, 9.55 Hz, H-6), 5.21 (1H, dd, J=10.04, 2.36 Hz, H-1), 5.58–5.93 (16H, m, CH₂OAc). ³¹P NMR (toluene-d₈, ¹H decoupled, 145.8 MHz): δ -4.42 (1 P, s), -3.98 (1 P, s), -3.48 (2 P, s). MS: m/z (+ve ion FAB) $[(M-CH_2OAc^++2H)^+, 44], 987$ 1131 [(M-3)] $CH_2OAc^+ + 4H)^+$, 100], m/z (-ve ion FAB) 1129 $[(M-CH_2OAc^+)^-, 18], 241 [OPO(OCH_2OAc)_2^-, 100].$ HRMS: m/z 1131.187 [(M-CH₂OAc⁺+2H)⁺] (calcd for C₃₅H₅₉O₃₃P₄, 1131.189).

D-2-*O*-Butyl-3-*O*-butyryl-*myo*-inositol 1,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester (*ent*-2). Alkylation of the phosphate *ent*-23 as described above afforded the octakis(acetoxymethyl) ester *ent*-2. $[\alpha]_D^{24}$ -1.3° (*c* 0.60 in toluene). HRMS: *m/z* 1129.177 [(M-CH₂OAc⁺)⁻] (calcd for C₃₅H₅₇O₃₃P₄, 1129.173). ¹H NMR and MS data were in accordance with those obtained for enantiomer 2.

D-3,4,5,6-Tetra-O-benzyl-1,2-di-O-butyl-myo-inositol (23). Sodium hydride (13 mg, 522 μ mol) was added to a stirred solution of 8 (94 mg, 174 μ mol) in dry DMF (3 mL) at room temperature in the dark. The mixture was stirred for 5 h, after which 1-butyl iodide (120 μ L,

1.04 mmol) was added. The suspension was stirred for 36 h at 80 °C, until HPLC (95% MeOH; 1.5 mL/min; $t_{\rm R} = 7.26$) showed predominantly one product. Excess of 1-butyl iodide and DMF were evaporated under reduced pressure. The mixture was then dissolved in *tert*-butyl methyl ether (30 mL) and washed once with phosphate buffer (10 mL), aq sodium dithionate (10 mL) and brine (10 mL), successively. The organic layer was dried over Na₂SO₄, filtered, and the ether was evaporated off to give an oil. The crude oil was purified by HPLC (95% preparative MeOH, $40 \,\mathrm{mL/min}$, $t_{\rm R} = 27.24$ min) to give the title compound **23** (100 mg, 88%) as an oil. $[\alpha]_{\rm D}^{24} + 1.3^{\circ}$ (*c* 1.00 in CHCl₃). ¹H NMR (CDCl₃, 360 MHz): δ 0.92 (6H, t, *J*=7.52 Hz, CH₃), 1.36–1.47 (4H, m, 2 \times CH₂), 1.55–1.63 (4H, m, 2 \times CH₂), 3.14 (1H, dd, J=9.57, 2.28 Hz, H-3), 3.34 (1H, dd, J=10.02, 2.28, H-1), 3.43 (1H, dd, J=9.34, 9.34, H-5), 3.52 (1H, dt, J=9.11, 6.60 Hz, OCCH₂CH₂CH₃), 3.60 (1H, dt, J=9.11, 6.60 Hz, OCCH₂CH₂CH₃), 3.75 $(2H, t, J=6.60 \text{ Hz}, \text{ OC}H_2\text{C}H_2\text{C}H_2\text{C}H_3), 3.89 (1H, dd,$ J = 2.28, 2.28 Hz, H-2), 3.92 (1H, dd, J = 9.57, 9.34 Hz, H-4), 4.00 (1H, dd, J = 10.02, 9.34 Hz, H-6), 4.71–4.93 (8H, m, CH₂Ph), 7.24–7.39 (20H, m, CH₂Ph). MS: m/z (+ve ion FAB) 653 [(M + H)⁺, 1], 91 [C₇H₇⁺, 100]. HRMS: m/z $653.383 (M + H)^+$ (calcd for C₄₂H₅₃O₆ 653.384).

D-1,4,5,6-Tetra-O-benzyl-2,3-di-O-butyl-myo-inositol (ent-23). A similar reaction of the compound ent-8 gave compound ent-23. $[\alpha]_D^{24} - 1.2^\circ$ (c 2.20 in CHCl₃). Spectral data were in accordance with those obtained for enantiomer 23.

D-1,2-Di-*O*-butyl-*myo*-inositol (24). Compound 23 (100 mg, 153 µmol) was hydrogenated with palladium (10%) on carbon under hydrogen as described in the general procedure to give tetrol 19 (40 mg, 92%) as a solid after freeze-drying. Mp 143.7-144.5 °C (from ethanol). $[\alpha]_D^{24}$ + 18.7° (*c* 0.80 in MeOH). ¹H NMR (DMSO-*d*₆, 360 MHz): δ 0.89 (3H, t, *J* = 7.52 Hz, CH₃), $0.91 (3H, t, J = 7.38 Hz, CH_3), 1.24-1.39 (4H, m, CH_2),$ 1.41-1.53 (4H, m, CH₂), 2.88 (1H, dd, J=9.55, 2.63 Hz, H-3), 2.93 (1H, dd, J=9.93, 2.63 Hz, H-1), 3.14 (1H, dt, $J = 9.51, 6.42 \text{ Hz}, \text{ OCH}_2\text{CH}_2\text{CH}_2\text{CH}_3), 3.31 (1\text{H}, \text{ dd},$ J = 9.52, 9.52 Hz, H-5), 3.39–3.67 (6H, m, 3 × OCH₂CH₂CH₂CH₃, H-2, H-4, H-6), 4.43 (1H, s, OH), 4.52 (1H, s, OH), 4.59 (2H, s (br), OH), MS: m/z (+ve ion FAB) 293 [(M+H)⁺, 100]. MS: m/z (-ve ion FAB) 291 [$(M-H^+)^-$, 100], Anal. ($C_{14}H_{28}O_6$) C: calcd 57.51; found 56.32; H: calcd 9.65, found 9.91.

D-2,3-Di-O-butyl-*myo***-inositol** (*ent***-24**). A similar reaction and workup of the fully protected compound *ent***-23** afforded tetrol *ent***-24**. $[\alpha]_D^{24} - 18.9^\circ$ (*c* 1.36 in MeOH). ¹H NMR and MS data were in accordance with those obtained for enantiomer **24**.

D-1,2-Di-*O***-butyl***-myo***-inositol 3,4,5,6-Tetrakis(dibenzyl)-phosphate (25).** A solution of compound **24** (40 mg, 137 μ mol) and tetrazole (115 mg, 1.64 mmol) in acetonitrile (2 mL) was treated with dibenzyl *N*,*N*-di*iso*propylphosphoramidite (552 μ L, 1.64 mmol) for 18 h, oxidized with peracetic acid, and worked up as described in the general procedures. Purification by preparative HPLC (93% MeOH; 40 mL/min; $t_{\rm R} = 26.05$ min) gave compound **25** (133 mg, 73%) as an oil. $[\alpha]_D^{24} - 2.3^\circ$ (c 1.00 in CHCl₃). ¹H NMR (CDCl₃, 360 MHz): δ 0.79 (3H, t, $J = 7.31 \text{ Hz}, \text{ CH}_3$, 0.86 (3H, t, $J = 7.31 \text{ Hz}, \text{ CH}_3$), 1.13– 1.38 (4H, m, $2 \times CH_2$), 1.41–1.56 (4H, m, $2 \times CH_2$), 3.24 (1H, dd, J=9.89, 2.15 Hz, H-1), 3.36 (1H, dt, J = 8.31, 7.41 Hz, OCH₂CH₂CH₂CH₃), 3.41 (1H, dt, J = 8.31, 5.87 Hz, OCH₂CH₂CH₂CH₃), 3.60 (1H, dt, $J = 8.88, 6.55 \text{ Hz}, \text{ OC}H_2\text{C}H_2\text{C}H_2\text{C}H_3), 3.71 (1\text{H}, \text{dt}, \text{dt})$ J=8.88, 6.55 Hz, OCH₂CH₂CH₂CH₃), 4.16 (1H, ddd, J=9.89, 9.89, 2.47 Hz, H-3), 4.21 (1H, dd, J=2.47, 2.15 Hz, H-2), 4.48 (1H, ddd, J = 9.67, 9.67, 9.67 Hz, H-5), 4.79 (1H, ddd, J = 9.89, 9.67, 9.67 Hz, H-6), 4.95 (1H, ddd, J=9.89, 9.67, 9.67 Hz, H-4), 4.96–5.09 (16H, m, CH₂Ph), 7.12–7.32 (40H, m, CH₂Ph). ³¹P NMR (CDCl₃, ¹H decoupled, 145.8 MHz): δ -1.82 (1 P, s), -1.59 (1 P, s), -0.81 (1 P, s), -0.68 (1 P, s). MS: m/z(+ve ion FAB) 1333 $[(M+H)^+, 2], 91 [C_7H_7^+, 100], m/z$ (-ve ion FAB) 1241 $[(M-C_7H_7^+)^-, 8]$, 277 $[OPO(OBn)_2^-, 100]$. HRMS: m/z 1333.416 $[(M + H^+)^+]$ (calcd for $C_{70}H_{81}O_{19}P_4$, 1333.417).

D-2,3-Di-O-butyl-*myo***-inositol 1,4,5,6-tetrakis(dibenzyl)-phosphate** (*ent-25*). Compound *ent-24* was phosphitylated and oxidized as described above for compound **25** to give the fully protected phosphate *ent-25*. $[\alpha]_D^{24}$ +2.5° (*c* 1.15 in CHCl₃). HRMS (DCI): *m/z* 1241.3748±10 ppm [(M-C₇H₇⁺)⁻] (calcd for C₆₃H₇₃O₁₈P₄, 1241.3748). ¹H NMR and MS data were in accordance with those obtained for enantiomer **25**.

D-1,2-Di-O-butyl-myo-inositol 3,4,5,6-tetrakisphosphate (26). Compound 25 $(134 \text{ mg}, 100 \mu \text{mol})$ was hydrogenated with palladium (10%) on carbon as described in the general procedure to give title compound 26 (52 mg, 87%) as a solid after freeze-drying. $[\alpha]_D^{24} + 4.1^\circ$ (c 1.10 in H₂O, pH 1.6). ¹H NMR (D₂O, 360 MHz): δ 0.70 (3H, t, $J = 7.37 \text{ Hz}, \text{ CH}_3$, 0.71 (3H, t, $J = 7.37 \text{ Hz}, \text{ CH}_3$), 1.13– 1.24 (4H, m, $2 \times CH_2$), 1.32–1.46 (4H, m, $2 \times CH_2$), 3.39 (1H, dd, J=9.47, 2.33 Hz, H-1), 3.45 (1H, dt, J = 8.59, 8.16 Hz, $OCH_2CH_2CH_2CH_3$), 3.51 (1H, dt, J = 8.59, 5.79 Hz, OCH₂CH₂CH₂CH₃), 3.56 (1H, dt, J = 9.47, 6.31 Hz, OC H_2 CH $_2$ CH $_2$ CH $_3$), 3.65 (1H, dt, $J = 9.47, 6.84 \text{ Hz}, \text{ OC}H_2\text{C}H_2\text{C}H_2\text{C}H_3), 4.02-4.08 (2H,$ m, H-2, H-3), 4.11 (1H, ddd, J=9.47, 9.47, 9.47 Hz, H-5), 4.27 (1H, ddd, J=9.47, 9.47, 9.47 Hz, H-4), 4.39 (1H, ddd, J = 9.47, 9.21, 9.21 Hz, H-6). ³¹P NMR (D₂O, ¹H decoupled, 145.8 MHz): $\delta - 0.69$ (1 P, s), -0.41 (2 P, s), -0.12 (1 P, s). MS: m/z (+ve ion FAB) 613 $[(M + H)^+, 15], 81 [PO(OH)_2^+, 100], m/z (-ve ion FAB)$ 611 [(M–H⁺)⁻, 90]. 79 [OP(O)₂⁻, 100]. HRMS: m/z $611.043 [(M-H^+)^-]$ (calcd for C₁₄H₃₁O₁₈P₄, 611.046).

D-2,3-Di-O-butyl-*myo***-inositol 1,4,5,6-tetrakisphosphate** (*ent-***26**). A similar reaction with the fully protected substrate *ent-***25** afforded the free acid *ent-***26** after freeze-drying. $[\alpha]_D^{24} - 4.3^\circ$ (*c* 1.00 in H₂O, pH 1.6). HRMS: m/z 611.047 [(M-H⁺)⁻] (calcd for C₁₄H₃₁O₁₈P₄, 611.046). ¹H NMR and MS data were in accordance with those obtained for enantiomer **26**.

D-1,2-Di-O-butyl-myo-inositol 3,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester (3). DIEA ($187 \mu L$,

1.00 mmol) and acetoxymethyl bromide $(111 \,\mu L,$ 1.00 mmol) were added to a suspension of compound 26 $(34 \text{ mg}, 55 \mu \text{mol})$ in acetonitrile (2 mL) as described in the general procedure. Purification by preparative HPLC $(73\% \text{ MeOH}, 40 \text{ mL/min}, t_{\text{R}} = 20.40 \text{ min})$ gave compound 3 (42 mg, 65%) as a syrup. $[\alpha]_{D}^{24} - 2.3^{\circ}$ (c 1.03 in toluene). ¹H NMR (toluene- d_8 , 360 MHz): δ 0.92 (3H, t, $J = 7.28 \text{ Hz}, \text{ CH}_3$, 0.99 (3H, t, $J = 7.48 \text{ Hz}, \text{ CH}_3$), 1.33– 1.47 (4H, m, 2 × CH₂), 1.48–1.54 (2H, m, CH₂), 1.61–1.69 (2H, m, CH₂), 1.77–1.87 (24H, 8 s, 8 × OAc), 3.11 (1H, dd, J=9.84, 2.36 Hz, H-1). 3.46 (1H, dt, J=8.79, 7.28 Hz, $OCH_2CH_2CH_2CH_3$), 3.56 (1H, dt, J=8.79, 7.28 Hz, $OCH_2CH_2CH_2CH_3$), 3.84 (1H, t, J=6.50 Hz, 2 × OCH₂CH₂CH₂CH₃), 4.34 (1H, dd, J=2.36, 2.36 Hz, H-2), 4.49 (1H, ddd, J=9.45, 9.45, 2.36 Hz, H-3), 4.68 (1H, ddd, J=9.65, 9.65, 9.65 Hz, H-5), 4.90 (1H, ddd, J=9.65, 9.45, 9.45 Hz, H-4), 5.06 (1H, ddd, J=9.84, 9.65, 9.65 Hz, H-6), 5.68–5.96 (16H, m, CH₂OAc). ³¹P NMR (toluene- d_8 , ¹H decoupled, 145.8 MHz): δ -5.12 (1 P, s), -3.98 (1 P, s), -3.69 (1 P, s), -3.42 (1 P, s).MS: m/z (+ve ion FAB) 1189 [(M+H)⁺, 3], 1045 $[(M-2 \times CH_2OAc^+ + 3H)^+, 100], m/z$ (-ve ion FAB) 1115 $[(M-CH_2OAc^+)^-, 30], 241 [OPO(OCH_2OAc)_2^-,$ 100]. HRMS: m/z 1117.210 [(M-CH₂OAc⁺+2H)⁺] (calcd for $C_{35}H_{61}O_{32}P_4$, 1117.210).

D-2,3-Di-*O*-butyl-*myo*-inositol 1,4,5,6-tetrakisphosphate octakis(acetoxymethyl) ester (*ent-3*). Alkylation of the phosphate *ent-26* as described above afforded the octakis(acetoxymethyl) ester *ent-3*. $[\alpha]_D^{24} + 2.5^{\circ}$ (*c* 1.10 in toluene). HRMS: *m/z* 1115.192 [(M-CH₂OAc⁺)⁻] (calcd for C₃₅H₅₉O₃₂P₄, 1115.194). ¹H NMR and MS data were in accordance with those obtained for enantiomer **3**.

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