



# A red-emitting fluorescent probe imaging release of calcium ions from lysosome induced by chloroquine based on a photochromic crowned spiropyrans

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## ABSTRACT

Calcium ions play an important role in intracellular signaling, metabolism and a wide range of cellular processes. An abnormal calcium concentration may cause many adverse symptoms, such as physiological responses in obesity, immune responses and pathological responses in Alzheimer's disease. Therefore, the development of specific fluorescent probes for monitor  $\text{Ca}^{2+}$  in living organism is of great significance. Herein, a red-emitting probe **SP-CE** was successfully designed and synthesized based on spiropyran derivatives as the fluorophore and aza-15-crown-5 as the sensing moiety for the recognition of calcium ions. Under the ultraviolet light irradiation, **SP-CE** was transformed to merocyanine (**MC**) form, which releasing negative oxygen ions, thereby achieving the recognition of calcium ions with the collaboration of crown ethers. According to the literature, chloroquine (**CQ**) could stimulate the up-regulation of pH in the sub-organelle lysosome in cells, resulting in changes of the calcium ions abundance in the cytoplasm. Interestingly, **SP-CE** could be used to visualize the occurrence of the above mentioned physiological processes in living cells. In addition, the probe **SP-CE** has been successfully employed for labeling calcium ions in living zebrafish model.

## 1. Introduction

$\text{Ca}^{2+}$  plays a major role that transmits in intracellular signaling and a variety of intracellular processes [1–4]. However, dysregulation of intracellular  $\text{Ca}^{2+}$  signaling is related to physiological responses in obesity, immune responses and pathological responses in Alzheimer's disease [5–11]. Therefore, it is important to track the dynamic changes of calcium concentration in cytosolic and understand the  $\text{Ca}^{2+}$  signal pathways in cells [12]. In addition, according to the literature reported that chloroquine (**CQ**) could cause calcium ion release by increasing lysosomal pH. Importantly, blockading lysosomal calcium ions release disrupts **CQ**-induced M1 macrophages polarisation, suggesting that lysosomal calcium release is necessary [13]. Therefore, a specific and ultra-fast detection tool is urgently needed for visualizing the fluctuation of calcium ion abundance in living cells and *in vivo*, which is essential for preventing related diseases by formulating corresponding intervention strategies.

Fluorescent probes based on small organic molecules combined with fluorescence imaging technology, using their unique advantages of good specificity, high sensitivity, and in-situ monitoring, have become important research methods in modern life sciences and disease diagnosis [14–17]. To date, great efforts have been made to monitor metal ions based on crown ether recognition sites [18,19]. For example, Heng's research group designed a light-controllable spiropyran reversible sensor optical probe for calcium ion detection [20]. Subsequently, Yin's research group designed a fluorescent probe based on photochromic crown ether spiropyran for lithium ion detection [21]. However, these fluorescent probes for monitoring metal ions were rarely exploited at the level of cellular imaging.

Spiropyran (**SP**) is one of the most common photochromic compounds, under UV/visible light irradiation, which could be reversibly converted from **SP** form to **MC** form [22–24]. "Hard" metal ions could be chelated with crown ethers by using lone pairs from oxygen effectively [25]. Due to these favourable characteristics, we designed and

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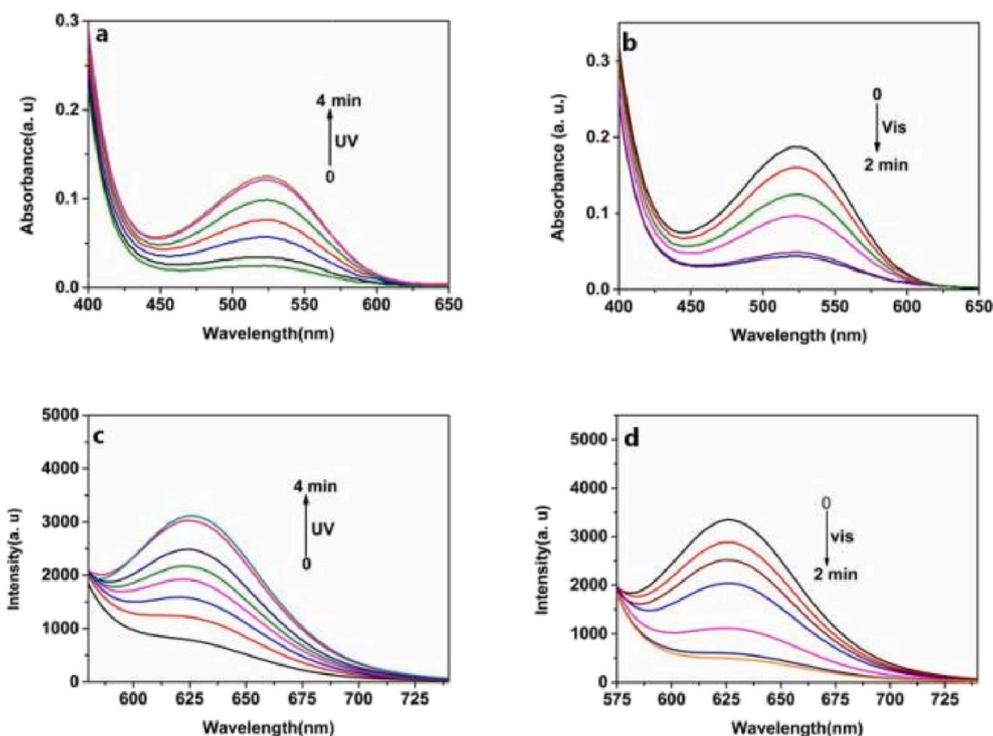
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**Fig. 1.** Spectroscopic analysis of photochromic SP-CE. Fig. 1(a) and (b) show that the UV-Vis absorption spectrum change of SP-CE (0.1 mM) under the ultraviolet light irradiation and visible light irradiation in  $\text{H}_2\text{O}:\text{CH}_3\text{CN} = 4:1$  (v/v). Fig. 1(c) and (d) show the fluorescence spectral change of SP-CE (0.1 mM) under ultraviolet light irradiation and visible light irradiation.

synthesized a red-emitting fluorescent probe (SP-CE), using spiropyran derivatives as the fluorophore and aza-15-crown-5 as the recognition site for calcium ions. Upon adding calcium ions into  $\text{H}_2\text{O}:\text{CH}_3\text{CN} = 4:1$  (v/v), the crown ether partly chelated with calcium ions to form corresponding complex, and induced the hydroxyl in the MC to negative oxygen ions, which enhanced the coordination ability with calcium ions. Further cells and zebrafish experiments clearly revealed that SP-CE could be used to monitor calcium ions in biological systems. Additionally, the probe could detect the calcium ions released from lysosome to cytoplasm due to chloroquine-induced pH changes in the lysosome.

## 2. Experiments

### 2.1. Synthesis of SP-CE

#### 2.1.1. Compound 1

Compound 1 was prepared according to literature [22]. mp,  $176\text{ }^\circ\text{C}$ – $178\text{ }^\circ\text{C}$ . IR ( $\text{cm}^{-1}$ ): 3309, 2973, 2529, 1720, 1594, 1469, 1400, 1360, 1349, 1270, 1207, 1030, 939, 757. ESI-MS  $m/z$ :  $[\text{C} + \text{H}]^+$  Calcd for  $\text{C}_{14}\text{H}_{18}\text{NO}_2^+$  232.13321, Found (Fig. S1).

#### 2.1.2. Compound 2

Related synthetic steps for compound 2 were described in the support information.

#### 2.1.3. Compound 3

Compound 2 (0.581 g, 1 mmol), EDCl (0.767 g, 4 mmol) and 1-hydroxybenzotriazole (0.54 g, 4 mmol) were dissolved in 15 mL of anhydrous  $N,N$ -dimethylformamide solution, the mixture was reacted for 30 min under the protection of argon at  $0\text{ }^\circ\text{C}$ . Then 1-aza-15-crown-5-ether (0.58 g, 1 mmol) and triethylamine (0.3 mL) were added to the reaction solution, and the mixture continued to react at room temperature for 24 h. After the reaction was completed, the reactants were poured into distilled water (100 mL) to continue stirring, and the resulting precipitate was filtered, washed with distilled water and dried

in vacuum. Finally, a dark gray solid was obtained (0.30 g, 52.3%). A similar substance was reported in this patent. We referred to the synthesis method of SP-CE in this patent [26]. mp,  $103\text{ }^\circ\text{C}$ – $106\text{ }^\circ\text{C}$ . IR ( $\text{cm}^{-1}$ ): 2970, 2865, 2362, 1706, 1641, 1609, 1508, 1484, 1460, 1331, 1270, 1126, 1081, 1029, 940, 912, 804, 747, 679.  $^1\text{H}$  NMR (600 MHz,  $\text{CD}_3\text{CN}$ )  $\delta$  8.10 (s, 1H), 8.03 (d,  $J = 8.9$  Hz, 1H), 7.18 (t,  $J = 7.6$  Hz, 1H), 7.14 (d,  $J = 7.2$  Hz, 1H), 7.07 (d,  $J = 10.4$  Hz, 1H), 6.86 (t,  $J = 7.4$  Hz, 1H), 6.76 (d,  $J = 9.0$  Hz, 1H), 6.68 (d,  $J = 7.7$  Hz, 1H), 5.97 (d,  $J = 10.4$  Hz, 1H), 3.66–3.34 (m, 22H), 2.72 (dt,  $J = 15.0, 7.4$  Hz, 1H), 2.58–2.52 (m, 1H), 1.27 (s, 3H), 1.15 (s, 3H) (Fig. S3).  $^{13}\text{C}$  NMR (151 MHz,  $\text{CD}_3\text{CN}$ )  $\delta$  171.24, 159.32, 146.59, 141.14, 136.18, 128.10, 127.77, 125.64, 122.76, 122.01, 121.81, 119.41, 119.06, 115.37, 107.01, 106.83, 70.54, 70.44, 69.93, 69.75, 69.57, 69.49, 68.83, 52.56, 50.07, 48.67, 39.73, 31.65, 25.26, 18.95. ESI-MS of probe SP-CE  $m/z$ :  $[\text{SP-CE}]^+$  calcd for 582.28129;  $[\text{SP-CE} + \text{Na}]^+$  calcd for 604.26308 (Fig. S3).

### 2.2. Materials

Related solvents and apparatus are described in the support information.

### 2.3. Fluorescence imaging

#### 2.3.1. Cell imaging

HeLa cells and HepG2 cells were cultured in DMEM medium containing 12% fetal bovine serum and 1% double antibody at  $37\text{ }^\circ\text{C}$  and 5% carbon dioxide incubator. When the cells were in the logarithmic growth phase, HeLa cells and HepG2 cells were inoculated in a special laser confocal culture dish, and continue culturing for another 24 h until the cells adhered to the wall.

The probe SP-CE ( $10\text{ }\mu\text{M}$ ) was dissolved in phosphate buffered saline (PBS) (2 mL), and then added to a confocal culture dish inoculated with HeLa cells and HepG2 cells and incubated at  $37\text{ }^\circ\text{C}$  for 20 min. In the experiment for detecting exogenous calcium ions, the HeLa cells and HepG2 cells loaded with SP-CE were incubated in PBS at  $37\text{ }^\circ\text{C}$  for 20

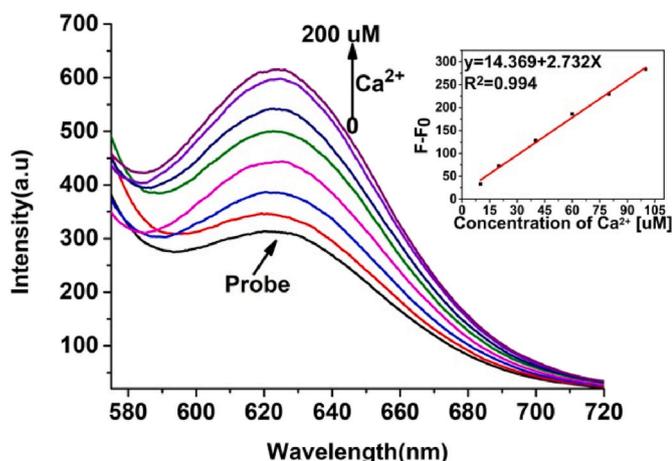


Fig. 2. Fluorescence spectrum responses of SP-CE (10  $\mu\text{M}$ ) to calcium ions (0–200  $\mu\text{M}$ ), in  $\text{H}_2\text{O}:\text{CH}_3\text{CN} = 4:1$  (v/v)  $\lambda_{\text{ex}} = 535$  nm,  $\lambda_{\text{em}} = 622$  nm, slit: 10 nm/10 nm; Inserted: The working curve of SP-CE (10  $\mu\text{M}$ ) in the presence of various concentrations of  $\text{Ca}^{2+}$  (0–100  $\mu\text{M}$ ),  $\lambda_{\text{ex}} = 535$  nm, slit: 10 nm/10 nm.

min, then incubated with calcium ions (100  $\mu\text{M}$ ) for 15 min, washed twice with PBS (2 mL), and then imaged. In the experiment for detecting endogenous calcium ions, HeLa cells and HepG2 cells loaded with SP-CE and CQ (100  $\mu\text{M}$ ) were incubated in PBS at 37  $^\circ\text{C}$  for 40 min, washed with PBS (2 mL) twice, and then imaged. The incubated cells were imaged under a UV lamp irradiation for 4 min.

### 2.3.2. Zebrafish culture and imaging

The probe SP-CE (20  $\mu\text{M}$ ) was added to E3 embryo medium containing zebrafish (5 days old) and incubated at 29  $^\circ\text{C}$  for 20 min and then imaged. In the experiment to detect exogenous calcium ions, the zebrafish was incubated with SP-CE (20  $\mu\text{M}$ ) for 20 min, then incubated with calcium ions (100  $\mu\text{M}$ ) for 20 min, then imaged. After incubation, the zebrafish was irradiated under UV light (365 nm) for 4 min.

## 3. Results and discussion

### 3.1. Study on the photochromic properties of probe SP-CE

We studied the photochromic properties of the SP-CE. The probe SP-CE (0.1 mM) showed a strong UV absorption band at 530 nm under the UV lamp 365 nm irradiation (0–4 min). However, when the SP-CE was irradiated with visible light (0–2 min), the UV absorption spectrum signal returned to the initial state, indicating that the MC form was converted into SP form. At the same time, when irradiated with 365 nm ultraviolet lamp, the fluorescence intensity of the SP-CE (0.1 mM) at 622 nm also increased with time (0–4 min). Nevertheless, it dropped to its original state under visible light irradiation, and the progress supported the UV–Vis spectrum analysis results.

### 3.2. SP-CE fluorescence spectrometric titration responds to calcium ions

The spectral properties of SP-CE and its response to calcium ions were obtained in  $\text{H}_2\text{O}:\text{CH}_3\text{CN} = 4:1$  (v/v) (see Fig. 1). Upon adding (0–200  $\mu\text{M}$ ) calcium ions to the detection solution containing probe SP-CE (10  $\mu\text{M}$ ), the solution changed from light pink to dark pink and the corresponding spectrum mainly showed an upward trend at near 620 nm, as depicted in Fig. 2. We plotted the maximum emission intensity of SP-CE at  $\lambda_{\text{ex}} = 535$  as a function of calcium ions concentration. The probe exhibited a good linear range and low detection limit (0.570  $\mu\text{M}$ ) for calcium ions (0–100  $\mu\text{M}$ ) (Fig. 2 Inserted). The detection limit of the probe was calculated according to the definition of IUPAC ( $\text{CDL} = 3\sigma/k$ ) [27,28].

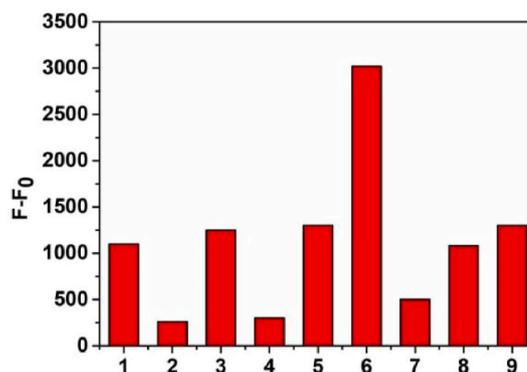
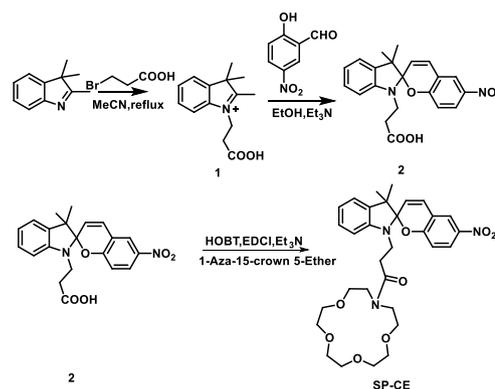
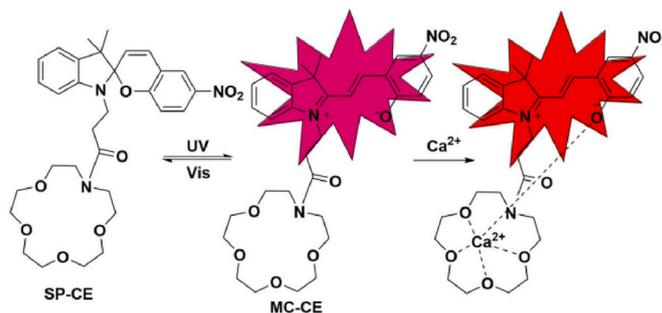


Fig. 3. Fluorescence response of probe SP-CE (100  $\mu\text{M}$ ) towards various biologically relevant species (500  $\mu\text{M}$ ) at 622 nm, (1)  $\text{Zn}^{2+}$ , (2)  $\text{Cu}^{2+}$ , (3)  $\text{Na}^+$ , (4)  $\text{Ag}^+$ , (5)  $\text{Mg}^{2+}$ , (6)  $\text{Ca}^{2+}$ , (7)  $\text{Fe}^{2+}$ , (8)  $\text{Li}^+$ , (9)  $\text{K}^+$ .



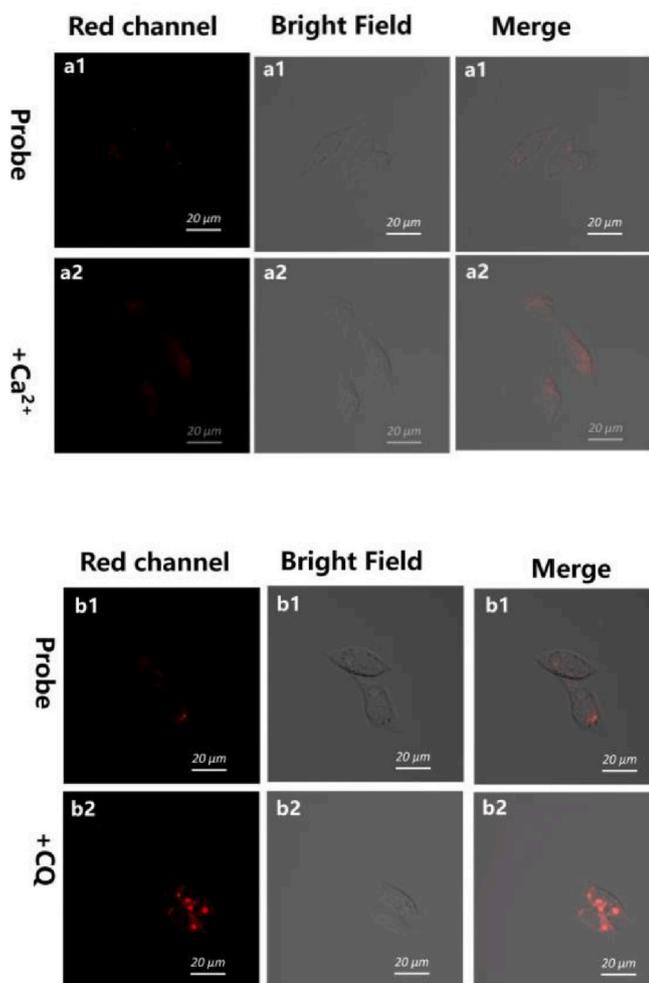
Scheme 1. Synthesis route for SP-CE.



Scheme 2. Mechanism diagram of the SP-CE response  $\text{Ca}^{2+}$ .

### 3.3. Selectivity of SP-CE

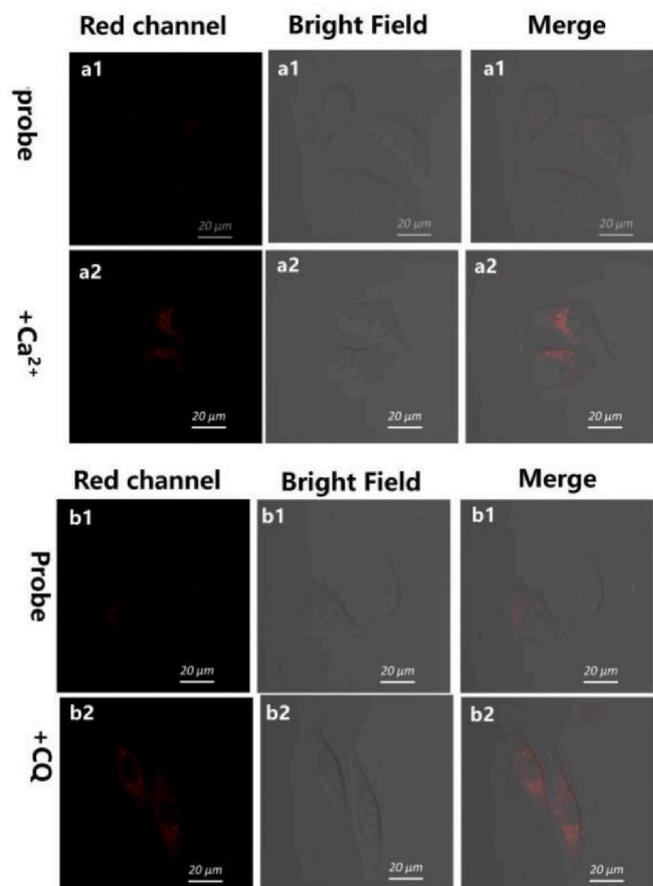
We then tested the selection recognition of the SP-CE to calcium ions under  $\text{H}_2\text{O}:\text{CH}_3\text{CN} = 4:1$  (v/v). As shown in Fig. 3, after adding calcium ions, the emission intensity increased significantly at 622 nm. Whereas upon the addition of other ions (500  $\mu\text{M}$ ) ( $\text{Zn}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{Na}^+$ ,  $\text{Ag}^+$ ,  $\text{Mg}^{2+}$ ,  $\text{Fe}^{2+}$ ,  $\text{Li}^+$ ,  $\text{K}^+$ ), the emission intensity did not increase remarkably. Therefore, the experiment proved that SP-CE has good selectivity to calcium ions. In addition, We tested the effect of different pH on the fluorescence responses of SP-CE to calcium ions. Notably, the fluorescence changes of SP-CE to  $\text{Ca}^{2+}$  reached a maximum at around pH = 7. It was further confirmed that SP-CE could recognize  $\text{Ca}^{2+}$  in physiological environment (Fig. S5).



**Fig. 4.** Confocal images of the SP-CE responds to exogenous  $\text{Ca}^{2+}$  and endogenous  $\text{Ca}^{2+}$  in HepG2 cells. (a1) HepG2 cells were incubated with SP-CE ( $10 \mu\text{M}$ ) for 20 min (a2) HepG2 cells were incubated with SP-CE ( $10 \mu\text{M}$ ) for 20 min, and then incubated with exogenous calcium ions ( $0.1 \text{ mM}$ ) for 15 min (b1) HepG2 cells were incubated with SP-CE ( $10 \mu\text{M}$ ) for 20 min (b2) HepG2 cells were incubated with SP-CE ( $10 \mu\text{M}$ ) and CQ ( $100 \mu\text{M}$ ) for 40 min. Red channel: emission wavelength  $622 \pm 20 \text{ nm}$  (excitation wavelength  $535 \text{ nm}$ ) Scale bar:  $20 \mu\text{M}$ .

### 3.4. Proposed mechanism

The response mechanism of SP-CE for detecting calcium ions was described in Scheme 2. Besides, as shown in Scheme 2, the SP-CE structure was easily transformed into MC-CE structure under UV irradiation, and the presence of calcium ions would further promoted the transformation of SP-CE structure into MC-CE through the combined action of negative oxygen anions and crown ether. Therefore, the fluorescence intensity was significantly enhanced after adding calcium ions. The corresponding reaction product was characterized by ESI-MS. The SP-CE acetonitrile solution ( $1 \text{ mL}$ ,  $1 \text{ mM}$ ) was mixed with  $\text{CaCl}_2$  aqueous solution under the ultraviolet light irradiation for 4 min, and then sample were measured by a mass spectrometer. As shown in Fig. S4, an obvious strong peak indicates that MC-CE is stably complexed with calcium ions. Due to the influence of the negative oxygen ion of spiro pyran on the binding efficiency of calcium ions, we used Prism-8 software to perform a non-linear regression fitting for calcium ions, as shown in Fig. S6, the stability constant of the SP-CE combining with calcium ions was  $1.12 \times 10^4 \text{ M}^{-1}$  in  $\text{H}_2\text{O}:\text{CH}_3\text{CN} = 4:1$ . Therefore, the selective binding of MC-CE isomer for calcium ions was attributed to the electrostatic interaction with *p*-nitrophenolate and the suitability of the

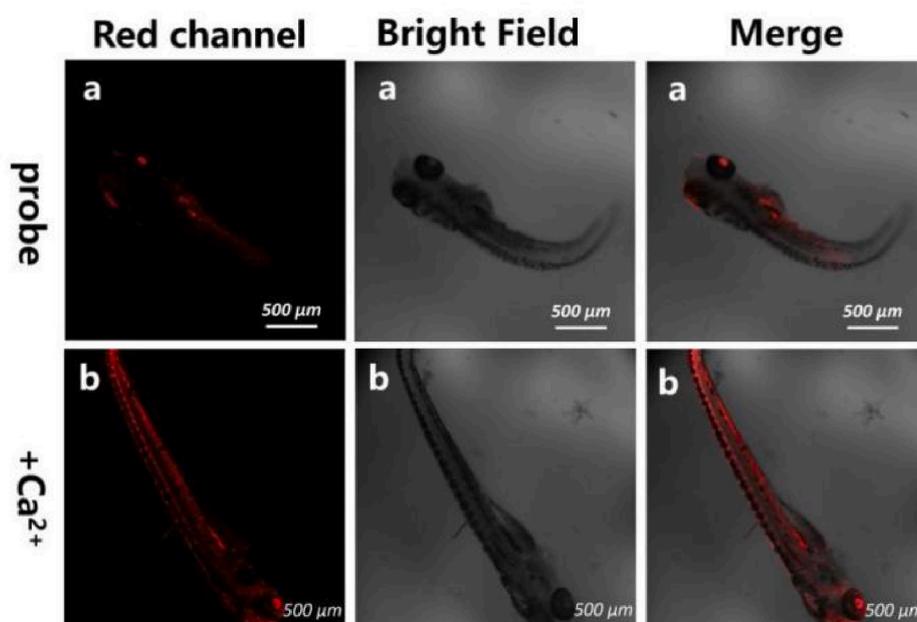


**Fig. 5.** Confocal images of the SP-CE responds to exogenous  $\text{Ca}^{2+}$  and endogenous  $\text{Ca}^{2+}$  in HeLa cells. (a1) HeLa cells were incubated with probe SP-CE ( $10 \mu\text{M}$ ) for 20 min (a2) HeLa cells were incubated with SP-CE ( $10 \mu\text{M}$ ) for 20 min, then treated with exogenous calcium ions ( $0.1 \text{ mM}$ ) for 15 min (b1) HeLa cells were incubated with SP-CE ( $10 \mu\text{M}$ ) for 20 min (b2) HeLa cells were incubated with SP-CE ( $10 \mu\text{M}$ ) and CQ ( $100 \mu\text{M}$ ) for 40 min. Red channel: emission wavelength  $622 \pm 20 \text{ nm}$  (excitation wavelength  $535 \text{ nm}$ ) Scale bar:  $20 \mu\text{M}$ .

size of alkali metal ions and crown ethers. (see Scheme 1)

### 3.5. Cellular imaging

We evaluated the SP-CE response to calcium ions in cell imaging experiments. After HepG 2 cells with SP-CE ( $10 \mu\text{M}$ ) were incubated for 20 min and then irradiated with UV light (4 min), a weak red fluorescence signal appeared in the red channel (Fig. 4a1). Furthermore, we also tracked to the exogenous calcium ions by fluorescence imaging. HepG2 cells were incubated with SP-CE ( $10 \mu\text{M}$ ) for 20 min and then incubated with calcium ions ( $0.1 \text{ mM}$ ) for 15 min and irradiated with UV light (4 min). A similar fluorescence phenomenon appeared in the red fluorescence signal, and it was greatly enhanced. (Fig. 4a2). Meanwhile, we also investigated the sensing performance of probe SP-CE for endogenous calcium ions in living cells. We used CQ drugs that change the lysosomal pH value, causing calcium ions to be released into the cytoplasm. As shown in Fig. 4b2, HepG2 cells were cultured with the SP-CE ( $10 \mu\text{M}$ ) and CQ ( $100 \mu\text{M}$ ) for 40 min, a significant red fluorescence signal was enhanced in the red fluorescence channel. At the same time, SP-CE was studied in HeLa cells (Fig. 5). These results implied that SP-CE could be considered as a cell-permeable molecular tool to examine and visualize dynamic equilibrium of calcium ions in cells.



**Fig. 6.** Confocal images of probe SP-CE response to exogenous  $\text{Ca}^{2+}$  in zebrafish. (a) The zebrafish was incubated with the SP-CE ( $20 \mu\text{M}$ ) for 20 min, and then irradiated with UV (365 nm) light for 4 min. (b) The SP-CE-loaded ( $20 \mu\text{M}$ ) zebrafish incubated for 20 min then incubated with  $\text{Ca}^{2+}$  ( $100 \mu\text{M}$ ) for 20 min, and then irradiated with UV (365 nm) light for 4 min. Red channel: emission wavelength  $622 \pm 20 \text{ nm}$  (excitation wavelength 535 nm) Scale bar:  $500 \mu\text{m}$ .

### 3.6. Zebrafish imaging

Next, we explored the feasibility of using the probe SP-CE ( $20 \mu\text{M}$ ) for imaging low concentration of calcium in zebrafish. As shown in Fig. 6a, the zebrafish was incubated with the probe SP-CE ( $20 \mu\text{M}$ ) for 20 min and then irradiated with UV lamp at 365 nm for 4 min, a weak red fluorescence signal appeared in the red channel. Simultaneously, when the zebrafish was incubated with SP-CE ( $20 \mu\text{M}$ ) for 20 min, then incubated with calcium ions ( $100 \mu\text{M}$ ) for 20 min, under the ultraviolet light irradiation for 4 min, the red fluorescence signal was enhanced in the red channel (Fig. 6b). Therefore, these imaging findings demonstrate the probe can effectively image calcium ions in vivo.

## 4. Conclusions

In summary, a red-emitting fluorescent probe SP-CE was explored to detect the endogenous and exogenous calcium ions in cells. SP-CE response mechanism was based on calcium ion induce the isomerized spiropyran MC to release negative oxygen ions, which could be specifically combined with crown ethers to achieve specific detection for calcium ions. Besides, further cells and zebrafish experiments clearly revealed that SP-CE could be used for monitoring calcium ions in biological system. Moreover, we applied CQ drugs to promote the release of calcium ions into the cytoplasm and to detect the level of calcium ion in the cytoplasm.

### Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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### Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.dyepig.2021.109467>.

### Author contributions

**Yingchun Wu:** Investigation.  
**Fangjun Huo:** Conceptualization, Writing-original draft.  
**Weijie Zhang:** Validation.  
**Jianbin Chao:** Project administration.  
**Caixia Yin:** Funding acquisition, Writing-review & editing.

### Declaration of interest statement

We have no any interest conflict.

### References

- [1] Fedrizzi L, Lim D, Carafoli E. Calcium and signal transduction. *Biochem Mol Biol Educ* 2008;36:175–80.
- [2] Scharff O, Foder B. Regulation of cytosolic calcium in blood cells. *Physiol Rev* 1993;73:547–82.
- [3] Komatsu H, Miki T, Citterio D. Single molecular multianalyte ( $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ ) fluorescent probe and applications to bioimaging. *J Am Chem Soc* 2005;127:10798–9.
- [4] Berridge MJ, Bootman MD, Roderick HL. Calcium signalling: dynamics, homeostasis and remodeling. *Nat Rev Mol Cell Biol* 2003;4:517–29.

- [5] LaFerla FM. Calcium dyshomeostasis and intracellular signalling in alzheimer's disease. *Nat Rev Neurosci* 2002;3:862–72.
- [6] Grynkiewicz G, Poenie M, Tsien RY. A new generation of  $\text{Ca}^{2+}$  indicators with greatly improved fluorescence properties. *Biol Chem* 1985;260:3440–50.
- [7] Harootunian AT, Kao JPY, Paranjpe S, Tsien RY. Generation of calcium oscillations in fibroblasts by positive feedback between calcium and IP3. *Science* 1991;251:75–8.
- [8] Chambers J, Ames RS, Bergsma D. Melanin-concentrating hormone is the cognate ligand for the orphan G-protein-coupled receptor SLC-1. *Nature* 1999;400:261–5.
- [9] Micu I, Ridsdale A, Zhang LQ. Real-time measurement of free  $\text{Ca}^{2+}$  changes in CNS myelin by two-photon microscopy. *Nat Med* 2007;13:874–9.
- [10] Skokos D, Shakhar G, Varma R, Waite JC. Peptide-MHC potency governs dynamic interactions between T cells and dendritic cells in lymph nodes. *Nat Immunol* 2007;8:835–44.
- [11] Kuchibhotla KV, Lattarulo CR, Hyman BT. Astrocytes in alzheimer's. *Sci Signal* 2009;323:1211–5.
- [12] Minta A, Kao J, Tsien RY. Fluorescent indicators for cytosolic calcium based on rhodamine and fluorescein chromophores. *J Biol Chem* 1989;264:8171–8.
- [13] Chen DG, Xie J, Fiskesund R. Chloroquine modulates antitumor immune response by resetting tumor-associated macrophages toward M1 phenotype. *Nat Commun* 2018;9:873.
- [14] Xing M, Wang K, Wu X, Ma S, Cao D, Guan R, Liu Z. A coumarin chalcone ratiometric fluorescent probe for hydrazine based on deprotection, addition and subsequent cyclization mechanism. *Chem Commun* 2019;55:14980.
- [15] Cao D, Liu Z, Verwilt P, Koo S, Jangjili P, Kim J, Lin W. Coumarin-Based small-molecule fluorescent chemosensors. *Chem Rev* 2019;119:10403.
- [16] Niu L, Guan Y, Chen Y, Wu L, Tung C, Yang Q. BODIPY-Based ratiometric fluorescent sensor for highly selective detection of glutathione over cysteine and homocysteine. *J Am Chem Soc* 2012;134:18928.
- [17] Liu Y, Ren DD, Zhang JJ, Li H, Yang XF. A fluorescent probe for hydrazine based on a newly developed 1-indanone-fused coumarin scaffold. *Dyes Pigments* 2019;162:112–9.
- [18] Rurack K, Kollmannsberger M, Daub J. A selective and sensitive fluoroionophore for Hg (II), Ag (I), and Cu (II) with virtually decoupled fluorophore and receptor units. *J Am Chem Soc* 2000;122:968–9.
- [19] Kim HM, Jeong MY, Ahn HC. Two-photon sensor for metal ions derived from azacrown ether. *J Org Chem* 2004;69:5749–51.
- [20] Heng S, Mak AM, Kostecki R. Photoswitchable calcium sensor: 'On'-'Off' sensing in cells or with microstructured optical fibers. *Sensor Actuator B Chem* 2017;252:965–72.
- [21] Yin CX, Li EZ, Kang J. A prospective material for the highly selective extraction of lithium ions based on a photochromic crowned spirobenzopyran. *J Mater Chem B* 2019;7:903–7.
- [22] Zhang WJ, Huo FJ, Yin CX. Photocontrolled single-/dual-site alternative fluorescence probes distinguishing detection of  $\text{H}_2\text{S}/\text{SO}_2$  in vivo. *Org Lett* 2019;21:5277–80.
- [23] Zhu JJ, Gao Q, Tong QX, Wu GF. Fluorescent probes based on benzothiazole-spiropyran derivatives for pH monitoring in vitro and in vivo. *Spectrochim Acta A* 2020;225:117506.
- [24] Zhang WJ, Huo FJ, Yue YK. Heat stroke in cell tissues related to sulfur dioxide level is precisely monitored by light-controlled fluorescent probes. *J Am Chem Soc* 2020;142:3262–8.
- [25] Katakya R, Nicholson PE, Parker D, Covington AK. Comparative performance of 14-crown-4 derivatives as lithium-selective electrodes. *Analyst* 1991;116:135–40.
- [26] Sasaki H, Ito Y, Kawashima K, Mitsuya T and Hoshi N, Reusable printing plate containing photochromic latent image-forming substance, manufacture of printing plate, and printing apparatus using the same, Japanese Patent 2002274075.
- [27] Zhang WJ, Liu T, Huo FJ, Ning P. Reversible ratiometric fluorescent probe for sensing bisulfate/ $\text{H}_2\text{O}_2$  and its application in zebrafish. *Anal Chem* 2017;89:8079–83.
- [28] Zhang WJ, Huo FJ, Liu T. Ratiometric fluorescence probe for hydrazine vapor detection and biological imaging. *J Mater Chem B* 2018;6:8085–9.