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Discovery and characterization of AZD6738, a potent inhibitor of ataxia telangiectasia mutated and rad3 related (ATR) kinase with application as an anti-cancer agent

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ABSTRACT

The kinase ataxia telangiectasia mutated and rad3 related (ATR) is a key regulator of the DNA-damage response and the apical kinase which orchestrates the cellular processes that repair stalled replication forks (replication stress) and associated DNA double-strand breaks. Inhibition of repair pathways mediated by ATR in a context where alternative pathways are less active is expected to aid clinical response by increasing replication stress. Here we describe the development of the clinical candidate **2** (AZD6738), a potent and selective sulfoximine morpholino-pyrimidine ATR inhibitor with excellent preclinical physicochemical and pharmacokinetic (PK) characteristics. Compound **2** was developed improving aqueous

solubility and eliminating CYP3A4 time-dependent inhibition starting from the earlier described inhibitor **1** (AZ20). The clinical candidate **2** has favorable human PK suitable for once or twice daily dosing and achieves biologically effective exposure at moderate doses. Compound **2** is currently being tested in multiple Phase I/II trials as an anti-cancer agent.

INTRODUCTION

Human cells are constantly exposed to DNA-damage events as a result of environmental and endogenous factors. In order to suppress genomic instability, an integrated group of biological pathways collectively called the DNA-damage response (DDR), has evolved to recognize, signal, and promote the repair of damaged DNA.^{1, 2} DNA-damage leads to cell death if sufficiently high and left unrepaired and is the concept behind DDR inhibition for cancer therapy. Tumor cells are sensitized to DDR based therapies through a combination of relatively rapid proliferation and DDR pathways that may already be functionally compromised. Ataxia telangiectasia and Rad3-related (ATR) is a serine/threonine-protein kinase belonging to the phosphatidylinositol 3-kinase-related kinase (PIKK) family of proteins and is a key regulator of DNA replication stress response (RSR) and DNA-damage activated checkpoints.^{3, 4} Replication stress, a hallmark of cancer,⁵ may occur in tumors through oncogene drivers or induced exogenously through treatment with DNA-damaging drugs or ionizing radiation (IR). Persistent replication stress leads to DNA breaks which if left unresolved are highly toxic to cells. In recent years potent and selective inhibitors of ATR (Scheme 1) have been developed

from orthogonal chemical series demonstrating preclinical *in vivo* proof of concept. These pivotal compounds and studies have been extensively reviewed,⁶⁻⁹ and reveal synthetic lethality of ATR inhibitors on tumors with p53-mutations or Ataxia telangiectasia mutated (ATM) loss-of-function,^{10, 11} as well as synergy in combination with a broad range of replication stress inducing chemotherapy agents such as platinums,¹² ionizing radiation,^{13, 14} and with novel agents such as the PARP inhibitor olaparib.¹⁵





We have previously described a series of potent and selective ATR inhibitors, exemplified by **1** (AZ20), from the sulfonylmethyl morpholino-pyrimidine series.¹⁶ Compound **1**, and close analogues,¹⁷ were shown to inhibit the growth of ATM-deficient xenograft models at well tolerated doses. However, we did not consider compound **1** of sufficient quality for further development due to low aqueous solubility and high-risk for drug-drug interactions (DDI) resulting from Cytochrome P450 3A4 (CYP3A4) time-dependent inhibition (TDI). In this report we describe our further studies to identify ATR inhibitors with the requisite properties suitable for clinical development that led to the discovery of **2** (AZD6738). The sulfoximine morpholino-pyrimidine **2**, along with the aminopyrazine **3** (berzosertib, M-6620 / VX-970), originating from Vertex and licensed to Merck KGaA, and most recently the naphthyridine **4** (BAY 1895344) from Bayer,¹⁸ have entered human studies. These compounds are being explored in early-phase clinical trials as single-agents and in combination with standard of care (SOC) and novel agents.^{8, 9}

RESULTS

Compounds 1, 5, 6, 8, 9 and 10 (Table 1) and intermediates 39, 46 – 49, 70 were prepared as described previously.¹⁶ Compounds 7, 17 and 18 were prepared as shown in Scheme 2 and compounds 11 - 16 were prepared as shown in Scheme 3. Intermediates 43 - 45 were prepared starting from the dichloropyrimidine 39. Suzuki coupling with 2-cyclohexenyl-4,4,5,5-tetramethyl-1,3,2-dioxaborolane led to 40 while S_NAr reaction with 8-oxa-3-azabicyclo[3.2.1]octane and 3-oxa-8-azabicyclo[3.2.1]octane afforded 41 and 42 respectively with no evidence of substitution at the 2-position. Cyclopropanation with 1,2-dibromoethane and strong base afforded 43 - 45. The 2-arylpyrimidine test compounds were synthesized by Suzuki coupling between the 2-chloropyrimidine substrates 43 - 48 and the corresponding

boronic acid or esters which were either purchased or prepared from the corresponding aryl bromides using literature methods.¹⁹

1-*H*-benzimidazole test compounds 19 - 30 (Table 4) were made as shown in Scheme 4 and 5. Cyclopropanation of led to 50 where reaction with 3(R)-methyl morpholine followed by sodium tungstate catalysed oxidation of the sulfide proceeded well to afford without overoxidation of the pyrimidine ring. Reaction with 1*H*-benzo[*d*]imidazol-2-amine or *N*-methyl-1*H*-benzo[*d*]imidazol-2-amine led directly to compounds and 21 respectively; compound 19 was derivatized to the *N*-acetyl 20. Benzimidazoles 22 - 30 were made starting from the 2chloropyrimidine intermediate (Scheme 5). Buchwald-Hartwig coupling with the appropriate substituted 2-nitroaniline utilising the xantphos ligand system afforded intermediates 52 - 60. Reduction of nitro to amino was achieved under indium catalyzed transfer hydrogenation conditions or with zinc in acetic acid to give the corresponding anilines 61 - 69 which were then cyclized using cyanogen bromide to afford compounds 22 - 30.

The iodobenzyl compound **70** was used as the starting point for the synthesis of sulfoxides **31**, **32** and sulfoximines **2**, **33** – **36** (Scheme 6). Displacement of the iodide in **70** with sodiumthiomethoxide gave sulfide **71** in high yield which was then oxidized to the corresponding sulfoxide **R/S-72** with sodium metaperiodate. Compounds were either made as a mixture of diastereoisomers and separated by chiral chromatography or prepared starting with the appropriate chirally pure sulfoxide. The mixture of sulfoxide diastereoisomers **R/S-72** were readily separable by chiral chromatography or vapor diffusion crystallization to afford **R-72** as a white crystalline solid and **S-72** as an oil. The stereochemistry of **R-72** was confirmed

by X-ray structure (Supplementary Figure S2). Later, a biocatalytic process to the chiral sulfoxide **R-72** was developed from **71** on large scale.²⁰ Cyclopropanation led to **R/S-73** followed by Suzuki coupling to afford the test compounds **31** and **32**. The sulfoximine moiety was introduced starting from the sulfoxides via rhodium catalyzed nitrene insertion.²¹ This proceeded with complete retention of stereochemistry and worked equally well either starting from the mixture of sulfoxide diastereoisomers to afford **R/S-74** or from single diastereoisomer **R-72** to afford **R-74**. Cyclopropanation resulted in rapid removal of the trifluoroamide group, to give **R/S-75** followed by Suzuki reaction led to test compounds **2**, **33** and **34** or reaction with *N*-methyl-benzo[*d*]imidazole-2-amine led to **35** and **36**.

Scheme 2^a



^{*a*} Reagents: (a) compound **40**: (Ph₃P)₄Pd, 2-(3,6-dihydro-2*H*-pyran-4-yl)-4,4,5,5tetramethyl-1,3,2-dioxaborolane, Cs₂CO₃, 1,4-dioxane, water, rt; compounds **41** and **42**: 8-oxa-3-azabicyclo[3.2.1]octane or 3-oxa-8-azabicyclo[3.2.1]octane respectively, Et₃N, DCM, rt; (b) compound **43**: 1,2-dibromoethane, NaH, DMF, 0 °C \rightarrow rt; compounds **44** and **45**: 1,2dibromoethane, 50% NaOH (aq.), tetraoctylammonium bromide, DCM, rt; (c) compound **7**: (Ph₃P)₂PdCl₂, 1*H*-indol-4-yl boronic acid, Na₂CO₃ (aq.), 4:1 DME:water, microwave, 110 °C; compounds **17** and **18**: (Ph₃P)₄Pd, 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1*H*pyrrolo[2,3-*c*]pyridine, Na₂CO₃ (aq.), 1,4-dioxane, 95 °C.

Scheme 3^a

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$$\begin{array}{c} 46 \text{ R1} = 100 \text{ R1} \\ 6 \text{ R1} = 100 \text{ R1} \\ 8 \text{ R1} = 100 \text{ R1}$$

^{*a*} Reagents: (a) compound 11: $(Ph_3P)_4Pd$, 1*H*-pyrrolo[2,3-*b*]pyridine-4-ylboronic acid, Na₂CO₃ (aq.), 1,4-dioxane, 90 °C; compound 12: 1*H*-benzo[*d*]imidazol-2-amine, Na₂CO₃, DMA, 160 °C, microwave; compound 13: (Ph₃P)₂PdCl₂, 1*H*-pyrrolo[2,3-*b*]pyridine-4vlboronic acid, Na₂CO₃ (aq.), 4:1 DME:water, 110 °C, microwave; compound 14: 2dicyclohexylphosphino-2',4',6'-triisopropylbiphenyl, bis(dibenzylideneacetone)palladium(0), 4-bromo-1*H*-pyrrolo[2,3-*c*]pyridine, KOAc, bis(pinacolato)diboron, dioxane, 100°C followed 100 °C; compound **46**, $(Ph_3P)_4Pd$, Na_2CO_3 (aq.), compound 15: 1,1'by bis(diphenylphosphino)ferrocenedichloropalladium, 4-bromo-1*H*-pyrrolo[2,3-*c*]pyridine, KOAc, bis(pinacolato)diboron, dioxane, 95 °C followed by compound 47, (Ph₃P)₄Pd, Na₂CO₃ (aq.), 95 °C; compound 16: (Ph₃P)₄Pd, Na₂CO₃ (aq.), 4-(4,4,5,5-tetramethyl-1,3,2dioxaborolan-2-yl)-1*H*-pyrrolo[*2,3-c*]pyridine, dioxane, 95 °C.

Scheme 4^a

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^{*a*} Reagents: (a) 1,2-dibromoethane, 50% NaOH (aq), tetraoctylammonium bromide, toluene, 60 °C; (b) (*R*)-3-methylmorpholine, DIPEA, 1,4-dioxane, 80 °C; (c) NaO₄W·2H₂O, Bu₄NHSO₄, EtOAc, H₂O₂, 0 °C \rightarrow rt; (d) compound **19**: 1*H*-benzo[*d*]imidazol-2-amine, Cs₂CO₃, DMA, 110 °C, microwave; compound **21**: N-methyl-1*H*-benzo[*d*]imidazol-2-amine, Cs₂CO₃, DMA, 90 °C; (e) DMAP, Ac₂O, 90 °C.

Scheme 5^a



^{*a*} Reagents: (a) Substituted 2-nitroaniline, $Pd(OAc)_2$, xantphos, Cs_2CO_3 , 1,4-dioxane, 80 °C, microwave; (b) Zn, AcOH, rt or In, NH₄Cl (aq.), EtOH, reflux; (c) Cyanogen bromide, MeOH, rt.

Scheme 6^{*a*}



^{*a*} Reagents: (a) NaSMe, DMF, rt; (b) NaIO₄, EtOAc, MeOH, H₂O, rt; (c) 1,2-dibromoethane, 50% NaOH (aq.), tetraoctylammonium bromide, 2-Me-THF, 60 °C; (d) X-Phos 2nd gen. precatalyst, 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1*H*-pyrrolo[2,3-*b*]pyridine, Cs₂CO₃, 1,4-dioxane:H₂O (4:1), 90 °C; (e) trifluoroacetamide, iodobenzene diacetate, Rh(OAc)₂ dimer, MgO, *iso*-propylacetate, 80 °C; then 7M NH₃ in MeOH, rt; (f) NaOH (50% aq.), 1,2dibromoethane, tetra-octylammonium bromide, mTHF, rt; (g) compound **33**: (Ph₃P)₂PdCl₂, 2M Na₂CO₃, 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1-tosyl-1*H*-pyrrolo[2,3*b*]pyridine, DME:H₂O (4:1), 90 °C then 2M NaOH (aq.), 50 °C; compound **34**: 1*H*-pyrrolo[2,3*b*]pyridin-4-ylboronic acid, (Ph₃P)₂PdCl₂, 2M Na₂CO₃, DME:H₂O (4:1), 90 °C; compound **35** and **36**: Cs₂CO₃, N-methyl-1*H*-benzo[*d*]imidazol-2-amine, DMA, 80 °C.

DISCUSSION

Compound **1** is a potent ATR inhibitor with excellent kinase selectivity and free exposure and is a useful tool compound to explore ATR pharmacology *in vivo*.¹⁶ However, compound **1** was also found to be a time-dependent inhibitor of CYP3A4 and to suffer from low aqueous

solubility. Inhibition of CYP3A4, particularly mechanism based inhibition, is a concern for clinical DDI,²²⁻²⁴ whereas low solubility limits the maximum absorbable dose (D_{abs}).²⁵ These properties increase risk of failure in clinical development and compound **1** was therefore not considered a suitable candidate for clinical-enabling studies. Eliminating CYP3A4 TDI and achieving high aqueous solubility at the same time maintaining high ATR potency, excellent specificity and the attractive pharmacokinetic properties exhibited by **1**, were the key medicinal-chemistry design goals in the optimization phase.

CYP TDI has been observed as a common feature in kinase inhibitors with CYP3A4 being the most commonly inhibited isoform.²⁶ CYP TDI is associated with the formation of covalent (or reversible-covalent) adducts to heme or protein following metabolic activation.^{23, 24} In addition to the risk of DDIs, formation of reactive metabolites is a causative factor for idiosyncratic drug toxicity.^{23, 27} The risk and impact of clinical DDI will be determined by overall drug disposition, dose, regimen and target patient population. While compound **1** clearly demonstrated TDI of CYP3A4 when incubated at 10 μM in human microsomes,¹⁶ we did not fully characterize this activity or model in detail the human PK and predicted clinical dose to understand the magnitude of the expected clinical DDI. We anticipated ATR inhibitors would be combined with cytotoxic and targeted drugs in the clinic. As DDI arising from inhibition of CYP3A4 would complicate co-dosing of such agents and many likely comedicants,²⁸ we set out to remove this undesirable activity.

The indole and morpholine groups were thought particularly vulnerable to metabolic activation. The susceptibility of cyclic tertiary amines such as morpholine to α -carbon

oxidation, generating reactive iminium ion intermediates is well described.²⁹⁻³¹ Indoles are known to undergo ring hydroxylation particularly at C-5 and/or C-6 positions;³² these species could arise via reactive epoxides and could also lead to quinone-like reactive intermediates following additional bioactivation. Oxidation of indole has also been observed at C-2 or C-3, presumably via the corresponding epoxide, and further oxidation and/or oxidative ring cleavage can lead to anthranilic acid products. The putative metabolic vulnerability of morpholine and indole presented us with a potentially insoluble problem as our previous work clearly demonstrated the importance of both groups to ATR potency.¹⁶

Reactive metabolites formed from bioactivation are generally electrophilic in character and highly unstable. Trapping experiments can be used to detect and characterize metabolites whereby a compound is incubated in human liver microsomal preparations and any reactive metabolites generated are trapped by specific added nucleophiles. Orthogonal nucleophiles are used to trap the different electrophilic species arising from bioactivation of chemical substrates. The nucleophiles commonly used are glutathione (GSH), a soft nucleophile efficient at trapping soft electrophiles such as epoxides, cyanide to trap iminium species arising from oxidation of tertiary amines and methoxyamine to effectively trap aldehyde products as Schiff bases.³³ These screens provide valuable mechanistic information to support rational medicinal chemistry design. When 1 was incubated with human liver microsomes, adducts with GSH but not cyanide were detected. Whilst the mechanisms of bioactivation and TDI may not necessarily overlay, the formation of GSH adducts implicates the indole group, likely through ring oxidation.

We had systematically explored the structure activity relationships (SAR) in the morpholino-pyrimidine pharmacophore, varying each of the substituents on the pyrimidine core. These compounds now allowed facile investigation into the molecular features in **1** responsible for CYP3A4 TDI, independent of ATR potency.

Table 1. Morpholine and C-2 heterocycle CYP3A4 TDI SAR



Compound	R1	R2	ATR IC ₅₀ (μM) ^a	ATR cell IC ₅₀ (µM) ^b	LogD _{7.4} ^c	CYP3A4 % TDI, 10 μM ^d
1		NH	0.005	0.061*	2.5#	50
5	[-N_0	I NH	0.008	0.29	2.3	38
6		I NH	0.007	0.11	2.6	53
7	[0	I NH	0.018	0.27	2.9	49
8	[-N_0	F NH	0.015	0.13	ND	61
9	[-N_0	NH O NH	0.14	16.5	1.6	<9
10	[-N_0	N=NH	0.18	6.2	2.2	<11



^{*a*}Uncertainty (95% confidence) for pIC₅₀ measurements is 0.38 (2.4-fold) based on an average of two repeat occasions per compound. ^{*b*} Uncertainty (95% confidence) for pIC₅₀ measurements is 0.7 (4.6-fold) based on at least two repeat measurements per compound (median number of repeats across tables is 3). The data presented is from a variation of the ATR cell assay described in Ref. ¹⁶(see experimental section). A good correlation is observed between ATR cell assay versions; N=41 compounds from morpholino-pyrimidine series, correlation: 0.94, mean difference: -0.084 pIC50. 'Compound 1 IC₅₀ = 0.050 µM in the earlier version reported.¹⁶ ^{*c*} LogD_{7.4} assay: lipophilicity was determined using the 'shake-flask' method. Plated aliquots of sample are dried down and octanol and water are added. Sample content of the phases is determined by LC/MS/MS after stirring, equilibration, and separation of phases by centrifuge. Full description of the assay can be found in the Supplementary Materials. An excellent correlation is observed between versions of the LogD_{7.4} methods; N=23

 compounds from the morpholino-pyrimidine series, correlation: 0.96, mean difference: -0.075. #Compound 1 LogD_{7.4} = 2.7 using the earlier methodology (Ref. ¹⁶). ND = not determined. ^d Mean value (N≥2) unless otherwise stated. Compounds were pre-incubated at 10 μ M with human liver microsomes (1 mg/mL) with and without NADPH (5 mM) for 30 min at 37 °C followed by 15 min incubation with 10 μ M midazolam; analysis of 1-hydroxymidazolam was performed using liquid chromatography-tandem mass spectrometry.³⁴ No activity detected vs. control for 1A2, 2C19, 2C9, and 2D6. [†] Result for compound **12** was just above background level in test 1 and below background level (<11%) in test 2; result for test 1 shown.

The unsubstituted and 3(R)-methyl substituted morpholines (compounds 5 and 1) respectively) both show clear CYP3A4 TDI activity when incubated at 10 µM in human liver microsomes using a standard liquid chromatography-tandem mass spectrometric endpoint (Table 1).³⁴ Introducing structural architecture to eliminate reactive moieties resulting from oxidation on the morpholine ring, for example by modification of the methylene adjacent to the oxygen atom using the bridged morpholine **6** or removal of the nitrogen heteroatom by substitution of morpholine for the 3,6-dihydro-2*H*-pyran 7, did not eliminate CYP3A4 TDI activity compared with morpholine. The dihydropyran and bridged morpholines, both of which have been described as morpholine isosteres in mTOR inhibitor series,^{35, 36} are equivalent in ATR potency to unsubstituted morpholine but at the price of higher lipophilicity. The impact of blocking and/or deactivating substituents on the indole ring was investigated. Simple ring substituents that retain ATR potency, for example the 6-fluoro indole 8, did not reduce CYP3A4 TDI. In contrast, addition of a polar and deactivating group such as acetamido 9 and replacement of the indole, for example with benzimidazole 10, which has equivalent lipophilicity to the indole 5, led to undetectable CYP3A4 TDI activity. In both cases

ATR potency was also significantly reduced but these results supported the notion that the indole ring was the likely key contributor to CYP3A4 TDI. We discovered the 7-azaindole (1H-pyrrolo[2,3-b]pyridine) 11 retains the ATR potency of the indole while effectively eliminating CYP3A4 TDI. In addition, through a wider campaign to identify ATR-active indole isosteres, the 2-amino-M-substituted benzimidazole 12 was found to have no or weak detectable 3A4 TDI. This variant was also found to possess superior cellular potency without increasing lipophilicity compared with the indole 5. Unknown to us at the time, Safina *et al.*³⁷ had developed a series of PI3K δ -specific morpholinopyrimidines substituted with 4-indole that were also found to be potent CYP3A4 TDIs and further determined that the 4-indole group was associated with this activity. In contrast to our own findings, replacement of the indole with 7-azaindole in the PI3K δ series did not attenuate CYP3A4 TDI. The apparent conflicting results are a reminder that metabolic activation is complex, and SAR may not be simplified to contributions of individual functional groups. It is whole-molecule structure and properties that determine metabolic fate. However, in our efforts towards identifying an ATR inhibitor candidate for human studies, both the 7-azaindole 11 and 2-aminobenzimidazole 12 became productive leads for optimization.

In the 7-azaindole series, the SAR was found to mirror that determined for indole with cellular potency increasing for the corresponding 3-(R) methyl morpholine **13** (Table 2). Pleasingly we could not detect CYP3A4 TDI activity for this compound. Measured lipophilicity for **13** is unchanged in comparison to **1** and unsurprisingly solubility remains low.

Table 2. Azaindole SAR



Compound	R1	R2	ΑΤR IC ₅₀ (μΜ) ^a	ATR cell $IC_{50} (\mu M)^b$	LogD _{7.4}	LLE ^c	Solubility pH7.4 (µM) ^d	CYP3A4 % TDI, 10 μM
1	⊢N_O	NH	0.005	0.061	2.5	4.7	10	50
13		I NH	0.005	0.10	2.6	4.4	8	<10
14	[-N_0	NH N	0.005	0.085	1.8	5.3	130	<15
15	⊢N_O	K NH	0.002	0.012	2.1	5.8	108	<20
16	⊢N_O ∵	K NH	ND	0.36	2.2	4.2	165	ND
17		K NH	ND	0.086	2.2	4.9	>1510	ND
18	E-N-O	К NH	ND	0.26	2.1	4.5	98	ND

^{*a,b*} For assay uncertainty, see footnotes in Table 1 ^{*c*} Lipophilicity ligand efficiency (LLE): ATR cell $pIC_{50} - LogD_{7.4}$. ^{*d*} Solid material was agitated in 0.1 M pH 7.4 phosphate buffer for 24 h, double centrifuged, and the supernatant analyzed for compound concentration by LC–UV–MS. Crystallinity assessed by polarized light microscopy of remaining solid. Full description is provided in Supplementary Materials. Lowest solubility given in the table where multiple measurements were taken: N≥3 for compounds 1, 13, 15; N=1 for compounds 14, 16, 17, 18. ND: not determined.

Substitution with the 6-azaindole (1*H*-pyrrolo[2,3-*c*]pyridine) isomer surprisingly resulted in compounds with much improved cellular potency compared with the corresponding indole and 7-azaindole. Moreover the 6-azaindole series combines improved potency with reduced lipophilicity and improved aqueous solubility (compounds 14 – 18, Table 2; note: the very high solubility result for compound 17 is most likely an outlier that we speculate is due to low crystallinity) while also having no detectable CYP3A4 TDI. It is interesting to observe a seemingly small structural change delivering such a significant effect on both potency and physicochemical properties.³⁸ Directionality of the hydrogen bond acceptor nitrogen in the azaindole isomers is changed relative to the morpholine hinge binder offering the potential to impact key interactions to the protein. Recent structures of ATR-mimicking PI3K mutants,³⁹ and an ATR cryo-EM,⁴⁰ structure suggest that the difference in potency between the azaindoles isomers can be explained by their interaction with Asp2335, potentially mediated by a water molecule. 6-Azaindole is also considerably more basic than 7-azaindole (pKa 7.9 and 4.6 respectively⁴¹) resulting in an expected change in ionization state at physiological pH and concomitant effects on LogD_{7.4} and solubility. The unsubstituted morpholine 7-azaindole 14 demonstrated excellent ATR cellular potency (IC₅₀ <100 nM) and again the established morpholine SAR translated, with a significant increase in potency observed when 3-(R)methyl morpholine was employed over the 3-(S) methyl isomer (compounds 15 and 16 respectively). The bridged morpholines, compounds 17 and 18, retained potency compared to morpholine 14 and the reduced lipophilicity provided by the 6-azaindole group led to compounds with a good overall balance of properties. It is noticeable from the compounds

shown in Table 2 that the hitherto excellent ATR enzyme to cell correlation breaks down. We had high confidence in the cellular assay measuring inhibition of CHK1 phosphorylation, a direct substrate of ATR, in response to a DNA-damage stimulus. Presumably, with the more potent compounds, the IC₅₀ is below the detection threshold (tight binding limit) of the enzyme assay. Therefore, at this stage of the optimization phase the cellular assay was primarily used to drive chemistry in conjunction with lipophilicity and physicochemical driven properties of aqueous solubility, permeability and metabolic stability. The 3-(*R*)-methyl morpholine 6-azaindole **15** stood out in combining excellent cellular potency and moderate lipophilicity (LLE = 5.8) with improved aqueous solubility and no measureable CYP3A4 TDI and represented a significant advancement over the indole lead **1**.

The benzimidazole head group utilized in compound **12**, substituted at the 2-position with amino or alkyl substituents on a morpholino-pyrimidine core was described in a series of antitumor agents with PI3K activity from Zenyaku Kogyo.⁴² We discovered the novel methylsulfonylmethyl morpholino-pyrimidine **12** to possess excellent ATR potency (Table 1) albeit with some class 1 PI3K inhibitory activity (e.g. PI3K α IC₅₀ = 0.24 µM). Substitution on the morpholine hinge binder has been shown to affect PI3K activity,^{16, 36} thus the 3(*R*)-methyl morpholine 2-aminobenzimidazole (compound **19**, Table 3) became a key compound to make that was subsequently shown to possess greatly improved ATR potency and selectivity over PI3K α (ATR cell IC₅₀ = 0.015 µM, PI3K α IC₅₀ = 9 µM).

Table 3. Benzimidazole C2 SAR



Compound	R1	ATR IC ₅₀ (μM) ^a	ATR cell IC ₅₀ (µM) ^b	LogD _{7.4}	LLE	Solubility pH7.4 (μM) ^c	CYP3A4 % TDI, 10 μM
19	Н	0.001	0.015	2.5	5.3	20	20^{\dagger}
20	Ac	0.014	0.017	1.8	6.0	7	<17
21	Me	0.002	0.008	2.8	5.3	49	<18

^{*a,b*} For assay uncertainty, see footnotes in Table 1 ^{*c*} Lowest solubility result is given in the table where multiple measurements were taken: N=2 for compound **21**; N=1 for compounds **19**, **20**. [†] Result for compound **19** was just above background level in test 1 and below background level (<17%) in test 2; result shown for test 1.

The potency improvement relative to indole is achieved without increasing lipophilicity; compound **19** shows borderline CYP3A4 TDI and aqueous solubility is not significantly improved compared to indole **1**. Substitution on the amino group with acetyl (compound **20**) or methyl (compound **21**) were well tolerated with no loss of cellular potency and in the case of the *N*-acetyl **20**, led to higher LLE. Moreover, we did not detect CYP3A4 TDI for either of the substituted 2-amino-benzimidazole compounds. While a relatively modest improvement in aqueous solubility was perhaps evident, particularly for *N*-methyl **21**, aqueous solubility was not robustly improved compared with indole.

Substitution on the benzimidazole aryl ring was explored for effects on potency and to address a theoretical concern that the naked aryl ring might be susceptible to metabolic

instability. The result of systematic substitution on the aryl ring of the benzimidazole group with fluorine is shown in Table 4.

Table 4. Benzimidazole ring SAR



Compound	R1	ATR IC ₅₀ (μM) ^a	ATR cell $IC_{50} (\mu M)^b$
19	Н	0.001	0.015
22	4- F	0.010	0.13
23	5-F	0.002	0.035
24	6-F	0.008	0.12
25	7 - F	0.090	5.1
26	4-Cl	0.017	5.3
27	4-OMe	0.27	5.3
28	5-Cl	0.021	0.87
29	5-CN	4.0	>30
30	5-OMe	0.55	15

^{*a,b*} See footnotes a,b in Table 1 for assay uncertainty; repeat measurements: N=2 for compound **19**; N=1 for compounds **22** – **30**.

A 5-F substituent (compound **23**) retains the same level of ATR potency compared with unsubstituted benzimidazole **19**. While 4-F and 6-F (compounds **22** and **24** respectively) retain a degree of ATR potency the 7-F analogue (compound **25**) is significantly less potent. A broader range of substituents in positions 4- and 5- (exemplified by compounds **26** – **30**, Table 4) were

then made to probe into the protein further but for all these examples reduced ATR potency was seen.

In the solid state, methylsulfone compounds such as 1 display a centrosymmetric methylsulfone to methylsulfone contact (in addition to ring-ring stacking and a hydrogen bonding network between indole N-H and sulfone oxygen) associated with high melting points and low solubility.¹⁶ Our SAR study of the indole series demonstrated that specific changes around the methyl sulfonyl moiety (e.g. addition of charged substituents) could improve physicochemical properties, but these changes ultimately led to reduced potency and/or attenuation of other fundamental properties such as permeability leading to low oral exposure. However, we had yet to attempt specific changes to the sulfone group, for example replacement with sulfoxide or sulfoximine whereby one of the oxygen atoms is replaced with nitrogen. We hypothesized that such changes could disrupt the observed solid-state contacts albeit at the complication of introducing additional chirality and with uncertain impact on target potency.

Table 5. Sulfoxide and sulfoximine SAR



 Compound
 R1
 R2
 ATR
 ATR cell
 LogD_{7.4}
 LLE
 Solubility
 CYP3A4 %

_				$IC_{50} (\mu M)^a$	IC ₅₀ (µM) ^b			pH7.4 (μM) ^c	TDI, 10 μM
	13	°,°, ~ ^{\$} _	NH NH	0.005	0.10	2.6	4.4	8	<10
	31	° S∑]	I NH	0.016	0.18	2.5	4.2	240	ND
	32	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	I NH	0.011	0.090	2.5	4.6	198	ND
	2	HN O S.,	I NH	0.004	0.074	1.9	5.2	661	<20
	33		I NH	ND	0.24	1.8	4.8	179	<11
	34	HN O	NH NH	ND	0.014	1.4	6.5	>2120	ND
	35	HŅ O • ^{\$}]	HN N	0.006	0.009	2.1	6.0	>780	<20
	36		HN HN	ND	0.023	2.1	5.5	933	ND

^{*a,b*} See footnotes in Table 1 for assay uncertainty. ND = not determined. ^{*c*} Lowest solubility result is given in the table where multiple measurements were taken: N \ge 2 for compound 13, 2, 33, 35; N=1 for compounds 31, 32, 34, 36.

The diastereomeric sulfoxides in the 7-azaindole series, compounds **31** and **32** (Table 5), were found to retain ATR potency compared with the corresponding sulfone **13** with no change in measured lipophilicity. These sulfoxides were also found to have high aqueous solubility though it proved challenging to generate stable crystalline forms and therefore the apparent improvement compared with the corresponding sulfones was treated with caution. The sulfoxides are highly permeable and have high exposure from oral doses in rodents. However, the sulfoxide group is prone to oxidation *in vivo* and a significant level of the corresponding sulfone was observed in rodent PK studies; for this reason, despite other properties being generally attractive, we discounted the sulfoxides from further progression. The sulfoximines (compounds 2, 33 - 36 Table 5) achieved both of the key aims we set at the start of the lead optimization campaign. The *R*-stereochemistry of the sulfoximine 2 was initially inferred from the stereochemistry of the sulfoxide precursor, with imination to the sulfoximine proceeding with retention of configuration, and later confirmed from an X-ray structure (Figure 1). Interestingly, hydrogen bonds are observed only between the azaindole substituents of adjacent molecules for 2, whereas the indole NH donor in the structure of 1 does not form a hydrogen bond. The centrosymmetric packing characteristic for methylsulfones as seen for 1 is prevented in the structure of 2 by introduction of the sulfoximine (Figure 1).



Journal of Medicinal Chemistry

Figure 1. Small-molecule crystal structures of **1** and **2**. Short contacts of sulfonomethyl (lefthand panel) and azaindole (right-hand panel) moieties are indicated by dotted lines.

The *R*-isomer of the sulfoximine group showed slight but consistent greater potency than the corresponding S-isomer (compare 2 with 33, Table 5) and similar magnitude compared with the corresponding sulfone. The sulfoximines also benefit from a significant reduction in lipophilicity and greatly improved aqueous solubility compared to the sulfones. Data from molecular matched pairs clearly show consistent retention of ATR cellular potency for the *R*isomer, concomitant reduction in lipophilicity and improved solubility of sulfoximine analogues compared with the corresponding sulfones (Figure 2). While the sulfoximine shows a different solid-state structure, the melting point of sulfoximine 2 was measured at 222 °C and is similar to sulfone 1 (204 °C). Therefore, the observed improvement in solubility appears to be driven mainly through reduced lipophilicity, though we speculate that solvation and the removal of the methylsulfones' centrosymmetric organization may contribute additional effects which are difficult to quantify. Replacement of methylsulfones has been shown to improve solubility in other contexts,⁴³ and sulfoximines may have wider utility in such cases. A further lipophilicity-related benefit of the sulfoximine group was realized in reduced hERG activity and was particularly advantageous for the 2-N-methylamino benzimidazole analogues 35 and 36 (Figure 2). Several reviews on the properties of the sulfoximine moiety in drug

discovery have since appeared elsewhere which observe the same broad trends as described

here.44-46



Figure 2. Sulfone and sulfoximine matched-pairs. Data for a) ATR cell pIC50; b) LogD7.4; c) hERG pIC50 and d) aqueous solubility. Numbers refer to the compounds shown in Tables 2 – 5.

Compounds that addressed the dual risks of low solubility and CYP3A4 TDI activity were further characterized as a shortlist to identify a candidate molecule for clinical-enabling studies. Physicochemical and ADME properties of compounds **2**, **15**, **21** and **35** are presented in Table 6.

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Table 6. Physicochemical and ADME characterization for preclinical candidate shortlist

compared with lead 1

	1	2	15	21	35
LogD _{7.4}	2.5	1.9	2.1	2.8	2.1
Solubility pH7.4 (µM) ^a	10	661	108	49	>780
PPB mouse, hu (%free) ^b	14, 9.2	54, 26	48, 26	4.4, 2.2	15, 4.3
СҮРЗА4 IC ₅₀ (µМ) ^с	>10	>10	>10	>10	>10
CYP3A4 %TDI, 10 μM	50	<20	<20	<18	<20
hERG IC ₅₀ (μ M) ^d	50	166	15†	4.6	15
Caco-2 P_{app} A-B (pH6.5, 7.4) ^e	23, 37	6.8, 12	8, -	-, 53	-, 8
Rat, hu CL _{int} ^f	25, <3	<3.5, <3	10, <3	25, <3	11, <3
Rat AUC (µM.hr) ^g	1.0	0.8	0.08	0.4	0.35

^{*a*} Lowest solubility result is given in the table where multiple measurements were taken: N=5 for compound **1**, N=3 for compound **15**, N=2 for compound **21**, N=7 for compound **2**; N=2 for compound **35**. ^{*b*} Measured in 10% plasma; % free calculated for 100% plasma assuming a single-site binding model. ^{*c*} >10 µM against 1A2, 2C19, 2C9, and 2D6. ^{*d*} Average activity against the human ether-a-go-go-related gene (hERG) encoded potassium channel was determined using automated whole-cell electrophysiology.⁴⁷ ^{*e*} Median A to B P_{app} (1 × 10⁻⁶ cm/s), 10 µM compound concentration. ^{*f*} Median CL_{int}: intrinsic clearance from hepatocytes (µL/min per 1 × 10⁶ cells, 1 µM compound concentration). ^{*g*} Plasma exposure; data normalised to 1 µmol/kg; compounds were dosed orally to male Han–Wistar rats at either 4 (compounds **1** and **15**) or 10 (compounds **21**, **2** and **35**) µmol/kg formulated in propylene glycol; compounds **1**, **2**, **15**, **35** were dosed as solutions and compound **21** as a suspension. [†] The most potent of 3 measurements is shown.

All compounds have equivalent or improved cellular potency combined in most cases with reduced lipophilicity relative to the lead **1**. The sulfoximines in particular have very high aqueous solubility, high unbound fraction and none of this set of compounds exhibited CYP3A4 reversible inhibitory or TDI activity against the five major human P450 isoforms. The benzimidazole sulfone **21** has the highest lipophilicity and hERG potency. Employment of the

sulfoximine group effectively reduces lipophilicity and the hERG potency (compare sulfoximine **35** with the sulfone **21**) but without impact on ATR cellular potency. All compounds exhibited high metabolic stability in human hepatocytes. The 6-azaindole sulfone (compound **15**) and sulfoximine compounds in general tend to have lower permeability compared with indole and sulfones respectively, presumably as a result of more hydrophilic character and, in the case of the sulfoximines, the introduction of an additional strong hydrogen bond donor. However, permeability of all compounds measured in Caco-2 cells was high and each of the compounds had favorable free exposure relative to potency in rats. The 6-azaindole sulfone **15**, has the lowest total AUC. However, this is negated by a high free fraction and high cellular potency. In comparison, the sulfoximine **2** has high total exposure combined with high unbound fraction that results in very high unbound exposure.

Kinase biochemical screening at a concentration of 1 μ M suggests a high level of selectivity for these optimized morpholinopyrimidine ATR inhibitors. Kinome selectivity is depicted in Figure 3 for sulfoximine **2** (the graphs for 2-aminobenzimidazole sulfone **21** and 6-azaindole sulfone **15** in comparison to the aminopyrazine **3** can be found in the Supplementary Materials Figure S1). Compound **2** showed excellent selectivity, with inhibition of ~60% for PIK3C2G and CLK4 (at 1 μ M), and <50% inhibition for the remaining 407 kinases tested (data supplied in Supplementary Materials). The PI3K and PIKK-family specificity for the morpholinopyrimidine ATR inhibitors is shown in Table 7.



Figure 3. Kinome selectivity depiction for compound **2**. Inhibition data (%) shown for a compound test concentration of 1 μ M. PI3K isoforms indicated as α , β , γ , δ . Data in Supplementary Materials, Table T1.

Table 7. PI3K, PIKK-family selectivity and growth inhibitory potency for preclinical candidate

shortlist compared with lead 1

	1	2	15	21	35
ATR Cell IC ₅₀ (μM) ^a	0.061	0.074	0.012	0.008	0.009
mTOR IC ₅₀ (μ M) ^b	0.038	0.37	0.034	0.052	0.15
mTOR cell: AKT pS473 IC ₅₀ (µM) ^c	2.4	>23	1.4	4.0	9.8
mTOR cell: p70S6K S235/236 IC ₅₀ (μM) ^d	0.72	5.7	0.27	1.5	3.1
PI3Kα cell IC ₅₀ (μM) ^e	>30	>30	>30	.30	>30
ATM cell IC ₅₀ (μM) ^f	>30	>30	>30	>30	>30
DNA-PK cell IC ₅₀ (µM) ^g	>30	>30	>30	>30	>30
LoVo GI_{50} (μ M) ^h	0.20	0.44	0.056	0.10	0.25
HT29 $GI_{50} \ (\mu M)^h$	0.97	2.6	0.31	0.55	1.2

^{*a*} See footnote of Table 1. ^{*b*} Standard error of mean (SEM) pIC_{50} measurement is ≤ 0.13 . ^{*c*} Inhibition of AKT pSer473 in MDA-MB-468 cells. ^{*d*} Inhibition of p70S6K pSer235/236 in MDAMB-468 cells. ^{*e*} Inhibition of pAKT T308 in BT-474 cells. ^{*f*} Inhibition of ATM Ser1981 in HT-29 cells following IR treatment. ^{*g*} Inhibition of DNA-PK pSer2056 in HT-29 cells following IR treatment. ^{*b*} MTS (tetrazolium dye) assay with 72 h continuous exposure to compounds.

The 2-aminobenzimidazole **21** and 6-azaindole sulfone **15** also display excellent selectivity over all kinase classes but have a degree of promiscuity versus lipid kinases. It is interesting to compare kinase selectivity with the aminopyrazine **3** which exhibits a wholly different profile, something that is not unexpected given that it belongs to a different structural class (see Supplementary Materials Figure S1). The aminopyrazine **3** has been reported to have high ATR

Page 33 of 73

Journal of Medicinal Chemistry

specificity particularly over the closely related mTOR.⁴⁸ The morpholinopyrimidine inhibitors were also screened specifically for inhibition of the related targets mTOR, PI3Ka, ATM and DNA-PK. These compounds all show moderate inhibitory potency, relative to ATR, against mTOR in an enzyme assay and some activity in cell assays reading out mTORC1 (inhibition of AKT pSer473) and mTORC2 (inhibition of p70S6K pSer235/236). This is unsurprising as the PI3K and PIKK kinases are known to have similar binding sites, with some compounds such as NVP-BEZ235,⁶ exhibiting a promiscuous pan-PI3K and -PIKK profile. In terms of protein sequence similarity, mTOR and ATM are nearest neighbors of ATR, yet, a margin of activity was observed in all cases including mTOR, relative to inhibition of ATR-dependent kinase signaling. Moreover, no activity could be detected for PI3K α , ATM or DNA-PK in cell-based systems. These combined data suggest the optimized compounds are unlikely to have activity against other PI3K/PIKK signaling pathways at relevant doses. LoVo are MRE11A-mutant (MRE11A is key component of the ATM signaling and DNA double-strand break (DSB) repair pathway) colorectal adenocarcinoma cells which are sensitive to ATR inhibitors.¹⁶ Compound 1 was shown to induce S-phase arrest, an increase in yH2AX over time and caspase-3 activation and cell death.⁴⁹ HT29 colorectal adenocarcinoma cells are classified as MRE11A and ATMproficient expressing high levels of total ATM protein without an ATM pathway defect and therefore expected to be relatively insensitive to selective ATR inhibition. As can be seen in Table 7, the morpholinopyrimidine ATR inhibitors show greater growth inhibition in LoVo compared with HT29 in support of this general hypothesis. Across a broader cell panel, ATR inhibitors from structurally orthogonal series show an inhibition profile that is distinct from

PI3K and PIKK-family inhibitors, further supporting a cellular mode of action arising from selective ATR inhibition (Figure 4).



Figure 4. Colon and gastric tumor cell line responses for ATR inhibitors **1**, **2**, **3** compared with: NVPBEZ-235 (labelled pPI3/mTORi)⁶ (mTOR, PI3K, ATR, ATM, DNA-PK), AZD8186⁵⁰ (labelled PI3K β + δ) and AZD8835⁵¹ (labelled PI3K α + δ). Data shown is normalised as pGI50 minus mean pGI50 across the panel to correct for the influence of absolute potency. Hierarchical clustering of profiles is shown on the right.

The compounds were next characterized for tumor growth inhibition (TGI) *in vivo*. Compounds were first administered at their maximum well-tolerated daily dose by oral gavage to female nude mice bearing human LoVo colorectal adenocarcinoma xenografts. Mouse tolerance of the morpholinopyrimidine ATR inhibitors was found to be variable. Indole sulfone 1 and the sulfoximines 2 and 35 were well tolerated at 50 mg/kg once daily (QD) whereas the 6-azaindole sulfone 15 and 2-methylaminobenzimidazole sulfone 21 were

tolerated at a maximum daily dose of 25 mg/kg QD. From a mechanistic standpoint and as a monotherapy, we expected continuous exposure would be required to drive efficacy.¹⁶ A broad relationship can be seen between the observed tolerance, efficacy and compound exposure in mouse relative to potency.

Table 8. Monotherapy *in vivo* tumor growth inhibition (TGI) in human LoVo colorectal adenocarcinoma xenografts.

Compound	Dose (mg/kg)	Schedule	TGI ^a (p-value T-test vs. vehicle control)
	50	QD, 20 d	88 (<i>p</i> <0.0005
1	50	QD, 14 d	67 (<i>p</i> <0.0005)
1	50	QD, 13 d	77 (<i>p</i> <0.0005)
	25	BD, 13 d	78 (<i>p</i> <0.0005)
	75	QD, 11 d	>100 (p<0.0005)
	50	QD, 14 d	72 (<i>p</i> <0.0005)
	50	QD, 13 d	81 (<i>p</i> <0.0005)
	50	QD, 15 d	89(p < 0.0005)
2	25	QD, 15 d	38(p < 0.049)
	25	QD, 11 d	45 (p<0.002)
	10	QD, 15 d	7 (p ns)
	25	BD, 13 d	96 (<i>p</i> <0.0005)
	12.5	BD, 11 d	45 (<i>p</i> <0.0002)
	25	QD, 20 d	60 (<i>p</i> <0.0005)
15	25	QD, 13 d	59 (p < 0.0005)
	12.5	BD, 13 d	62 (<i>p</i> <0.0005)
•			
21	25	QD, 20 d	67 (<i>p</i> <0.0005)
35	50	QD, 14 d	27 (<i>p</i> <0.05)

^{*a*} Female nude mice bearing established human LoVo xenografts were dosed orally with compound at the indicated dose and schedule. ns = not significant
Table 9. Plasma concentration at 8 hours following multiple doses in nude mouse at compound maximum well tolerated dose.

Compound	Dose (mg/kg)	Dosing frequency	Mean 8 h plasma concentration, μM (Std Dev)²	Free plasma LoVo multiple
1	50	QD	3.5 (1.2)	2.5
1	25	BD	1.8 (0.62)	1
	75	QD	2.6 (2.0)	3
2	50	QD	2.2 (0.52)	3
	25	BD	0.74 (0.58)	1
15	25	QD	0.78 (0.62)	6.5
15	12.5	BD	0.22 (0.20)	2
21	25	QD	8.3 (1.3)	3.5
35	50	QD	0.94 (0.39)	0.5

^{*a*} Female nude mice bearing established human LoVo colorectal adenocarcinoma xenografts were dosed orally with compounds either once-daily (QD) [compound 1: averaged data from 3 {N=10, mean=4.2 μ M (Std Dev=1.3)}, 4 {N=10, 3.1 (1.1)} and 14 {N=10, 3.2 (0.77)} consecutive doses; compound 2: 75 mg/kg: averaged data from 4 doses, 50 mg/kg: averaged data from 4 {N=5, 2.1 (0.65)}, and 14 {N=10, 2.3 (0.49)}, consecutive doses; compound 15: 4 consecutive doses (N=10 independent samples); compound 21: 4 consecutive doses (N=5 independent samples); compound 35: 14 consecutive doses (N=10 independent samples)] or twice-daily for 8 consecutive doses.



Figure 5. Exposure of ATR inhibitors 1, 2, 15, 21, and 35 is correlated to tumor growth inhibition (TGI). The observed plasma concentrations following a single dose of each compound were multiplied by the compound specific *in vitro* measured free fraction and divided by *in vitro* GI₅₀ to give fold free concentration above GI₅₀. The time above *in vitro* LoVo GI₅₀ is plotted against LoVo xenograft tumor growth inhibition *in vivo*. A logarithmic trendline (Log.(All)) best-fit curve is shown for all compounds.

All compounds other than compound **35**, achieved TGI >50% in the LoVo model when dosed once daily over the course of the study (Table 8). The comparative lack of efficacy for compound **35** can be explained by the lowest free plasma concentration multiple over LoVo

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GI₅₀ measured at 8 h of all the compounds tested (Table 9). The 7-azaindole sulfoximine 2 showed the greatest TGI, equivalent or greater at 50 mg/kg across multiple experiments, to the indole sulfone lead 1 and this is associated with free plasma concentration at 8 h in excess of the LoVo GI₅₀. The total plasma concentration of the 6-azaindolesulfone 15 was found to be the lowest of the compounds tested but the combination of high unbound fraction with high potency results in a high free plasma multiple over LoVo GI₅₀. In contrast, benzimidazole sulfone 21 shows high total drug plasma concentration and high potency but has a relatively high bound fraction, particularly in comparison to compounds 2 and 15, and this leads to a free plasma multiple over LoVo GI₅₀ in a similar range to the other compounds. Twice-daily dosing was investigated for compounds 1, 2 and 15 in an attempt to achieve longer exposure and drive a greater tumor response. When the maximum well-tolerated daily single dose was split (dosed 8 hour apart), neither compound 15 or 1 showed greater anti-tumor activity. For indole sulfone 1, this may be explained by a relatively flat PK profile in the mouse negating the impact of twice daily dosing (BD).¹⁶ The 6-azaindole sulfone **15** has a relatively short half-life in mouse and we expected BD dosing would lead to greater efficacy. However, the efficacy achieved for compound 15 dosed BD was indistinguishable from the higher single dose. In contrast, 7azaindole sulfoximine 2 achieved near complete TGI in the LoVo xenograft model when administered at a dose of 25 mg/kg BD. The time each day that free concentrations in plasma were above the *in vitro* LoVo GI₅₀ was estimated using the observed plasma concentrations following a single dose of each compound. This duration exhibits a saturating relationship with tumor growth inhibition (Figure 5). The relationship is consistent across compounds

Page 39 of 73

Journal of Medicinal Chemistry

regardless of differences in pharmacokinetic properties, including different terminal half-lives in the mouse. This analysis broadly correlates with the cover seen in earlier experiments, with 15 showing the largest difference. A clear dose-response could be demonstrated for 2 in LoVo (Table 8 and Figure 6, top graph) and this was compared with HT29 (Figure 6, bottom graph). Compound 2 delivers significant TGI in LoVo at doses as low as 25 mg/kg QD or 12.5 mg/kg BD. A dose of 75 mg/kg QD leads to regression in the LoVo model albeit tolerability is borderline with 4 of 10 animals in the group terminated in accord with study protocol due to bodyweight loss greater than 15%. In the remaining six animals a maximum bodyweight loss of 9% was observed. Therefore 75 mg/kg, while formerly tolerated, was not considered to be a well-tolerated dose. The observed in vitro sensitivity (Table 7) translated in vivo with no significant anti-tumor efficacy observed for 2 in HT29 using doses and schedules which are highly active in LoVo xenografts (Figure 6, bottom graph). yH2AX is a sensitive marker for DNA damage and a useful marker to study ATR inhibition. Increases in yH2AX reflect the time- dependent accumulation of collapsed replication forks, which only occur in actively replicating cells during S-phase of the cell cycle), and replication-associated DSBs.^{52, 53} In LoVo xenografts, the magnitude and maintenance of γ H2AX over 24 hours is obtained in a dosedependent manner after repeat daily dosing with the sulfoximine 2 (Figure 7) and this is associated with the greater anti-tumor effect observed for this compound. We observe an indirect relationship between plasma PK and tumor PD based on yH2AX induction, with signals being sustained beyond 24 hours despite plasma concentrations predicted to be below detectable levels at this timepoint (LOQ 0.09 µM data not shown). While we need sufficient

levels and duration of cover to induce DNA breaks and γ H2AX, once these have formed it may take many hours for the damage to be repaired, or cells to die, and the γ H2AX signal to dissipate. Persistence of γ H2AX signal over time (after the damage insult) is indicative of unrepaired DSBs and/or DNA repair inhibition and is observed even after breaks have been repaired.⁵²



Figure 6. *In vivo* tumor growth inhibition (TGI) for compound **2**. Top graph: female nude mice bearing established human LoVo (MRE11A mutant/ATM deficient) colorectal adenocarcinoma xenografts were dosed orally with either vehicle (\blacklozenge) or **2** at 10 mg/kg once

daily (×, day 22 TGI = 7%, p= ns), 25 mg/kg once daily (•, day 22 TGI = 38%, p<0.049), 50 mg/kg once daily (\triangle , day 22 TGI = 89%, p<0.0005). Bottom graph: female nude mice bearing established HT-29 (MRE11A wild type/ATM-proficient) colorectal adenocarcinoma xenografts were dosed orally with either vehicle (•) or 2 at 25 mg/kg twice daily (•, day 27 TGI = 5%, p= ns), 50 mg/kg once daily (\triangle , day 27 TGI = 10%, p= ns) or 75 mg/kg once daily (\square , day 27 TGI = 30%, p<0.005); ns = not significant.



Figure 7. Compound **2** γ H2AX DNA-damage biomarker pharmacodynamics in established LoVo tumor xenografts in female nude mice. Mice were dosed with either vehicle or **2** at 10 mg/kg, 25 mg/kg, 50 mg/kg or 75 mg/kg once daily for 4 consecutive days before tissue sampling at 8 hours or 24 hours after the 4th dose (day 4). Data are presented as average %

γH2AX positive tissue per total tumor area counted ± Standard Deviation (n=4 independent mouse/tumors per point).

Compound **2** has highly attractive physicochemical properties with a low biopharmaceutical risk profile (Table 10). For completeness we examined the methylmorpholine stereoisomers of compounds 2 and 33 and prepared the 3(S)-methylmorpholine matched pairs (compounds 37 and 38 respectively, Table S2 Supplementary Materials) and these were found to have significantly reduced potency as would be predicted from the described SAR. A stable crystalline form of 2 possesses high solubility in aqueous and biorelevant media such as Simulated Gastric Fluid (SGF) and Fasted State Simulated Intestinal Fluid (FaSSIF) (Table 10). Pharmacokinetic evaluation of **2** shows this compound has low to moderate clearance, moderate volume of distribution and good bioavailability in rodent and dog. Physiologically based pharmacokinetic (PBPK) modelling,^{54, 55} using *in vitro* and *in vivo* data across preclinical species, predicts 2 to have low human clearance (average clearance of 1.98 mL/min/kg) and estimated volume of distribution of 1 L/kg leading to a terminal half-life of 6 hours in man. In *silico* modelling of human absorption,⁵⁶ predicts compound **2** will have high bioavailability (~80%) and a maximum absorbable dose $D_{abs}^{25, 57}$ of ~2,500 mg. The preclinical TGI data for compound 2 and related compounds suggests that the duration of free drug exposure is correlated with monotherapy efficacy. A PK/PD model was created linking the preclinical pharmacokinetics of **2** with the DNA-damage biomarker γ H2AX and tumor growth inhibition

Page 43 of 73

in the LoVo xenograft model. Simulations using the monotherapy LoVo GI_{50} as the target unbound trough concentration led to a predicted efficacious dose in human⁵⁸ of 200 mg (unbound $C_{max} = 0.93 \ \mu$ M) dosed BD assuming continuous exposure is required across the full dose interval. While it is likely that target drug concentration and/or duration will be lower when ATR inhibitors are dosed in combination, dose prediction based on monotherapy pharmacodynamic modulation and efficacy provides a reasonable basis to estimate the efficacious dose in man. The monotherapy human dose estimate is well below the calculated D_{abs} , and therefore **2** has low risk of requiring development of an enabling formulation for use in the clinic, whether used alone or in combination. The profound anti-tumor activity and associated pharmacodynamics combined with favorable dose estimate and pharmaceutical properties led to the sulfoximine **2** being selected for IND-enabling studies and subsequently this compound became AZD6738.

Table 10. Physicochemical properties, preclinical PK and predicted human PK and dose of 2

	2
Estimated p <i>K</i> a	3.8 (B1) 2.2 (B2)
Solubility SGF pH2.5, FaSSIF pH6.5 (mg/mL)	26.7, 0.49
Rat PKª: Cl, %LBF / Vd _{ss} , L/kg / bioavailability, %	22 / 3.1 / 67
Dog PK ^b : Cl, %LBF / Vd _{ss} , L/kg / bioavailability, %	10 / 1 / 100
Hu CL _{int} ^c	2.0
Predicted hu PK: Cl, %LBF / Vd _{ss} , L/kg / bioavailability, % / T1/2, h	11 / 1.0 / 80 / 6

Maximum absorbable dose $(D_{abs})^d$	2492
Predicted hu dose (mg) ^e	200

^a Rat plasma PK in male Han Wistar rats, dosed at 8.55 (IV) and 20.3 (PO) μ mol/kg. ^b Dog blood PK in male and female Beagle dogs, dosed at 10.5 (IV) and 15.1 (PO) μ mol/kg. ^c Estimate of median intrinsic clearance obtained from hepatocyte data using intercept method⁵⁹ (μ L/min per 1 x 10⁶ cells, 1 μ M compound **2**. ^d Maximum absorbable dose (D_{abs}).^{25 e} Predicted human dose assuming trough concentration = 0.43 μ M dosed BD.

Compound 2 (AZD6738) has progressed into human clinical trials and is being assessed in multiple Phase I/II studies as a monotherapy and in combination with carboplatin (NCT02264678), paclitaxel (NCT02630199), radiotherapy (NCT02223923), and the novel NCT03330847), acalabrutinib olaparib (NCT03462342, (NCT03328273) agents and durvalumab (NCT03334617). Plasma pharmacokinetics in man showed rapid absorption of compound **2**, with peak plasma concentration at \sim 1.5 hours post-dose, and a biphasic decline with an elimination half-life of 11.0 hours; and despite \sim 45% variability in clearance, there was dose proportionality to at least 320 mg QD.^{60, 61} Plasma pharmacokinetics aligned well with target preclinical free drug exposure and duration, with cover over ATR cell IC₉₀ obtained in \geq 50% of patients with a 80-320 mg daily dose range (QD). The observed levels are expected to drive significant ATR inhibition, with cover increased in a dose dependent manner up to ATR IC₉₆.⁶² Preclinical work has demonstrated further potential for combination therapy with e.g. olaparib⁶³. Early data from dose escalation studies with combination partners olaparib and durvalumab was presented recently⁶¹ showing good tolerance for compound **2** in combination

dose escalation, and preliminary signals of anti-tumor activity, with no evidence of drug-druginteractions (DDI); in 44 patients treated at various doses of compound **2** (60 mg through 240 mg QD) in combination with olaparib, 1 RECIST complete response (CR) and 6 partial responses (PR; 1 unconfirmed) were observed in patients with BRCA1 or BRCA2 mutations independent of ATM status, in advanced breast (3 patients), and 1 each of ovarian, prostate, pancreatic and ampullary cancer; and in 25 patients treated at various doses (80 mg through 240 mg BD) in combination with durvalumab, 1 RECIST CR and 3 PRs, in patients with advanced NSCLC (3 patients) and HNSCC (1 patient), independent of tumor PD-L1 expression were observed.⁶¹

CONCLUSIONS

ATR plays a key role in DNA-damage repair, and only recently highly potent and selective compounds have entered clinical assessment. Here we report the discovery of **2** (AZD6738), an inhibitor of ATR currently being examined in Phase I/II clinical trials. Starting from a lead compound containing a sulfone group and substitution with a sulfoximine moiety led to **2**, a compound with excellent solubility, good cell potency and high selectivity across the kinome.

The promising preclinical data package for compound **2** combined with moderate predicted human dose strongly supported its selection as a clinical candidate. First dose of compound **2** in patients was achieved in 2013. Pharmacokinetics in man was found to be dose-dependent with efficacious exposure in alignment with the preclinical prediction together with promising early signs of clinical efficacy.^{8, 61} AZD6738 is currently in multiple Phase II studies as a monotherapy and in combination with carboplatin (NCT02264678), paclitaxel (NCT02630199), radiotherapy (NCT02223923), and the novel agents olaparib (NCT03462342, NCT03330847), acalabrutinib (NCT03328273) and durvalumab (NCT03334617).

EXPERIMENTAL SECTION

General Synthetic Methods. All experiments were carried out under an inert atmosphere and at room temperature (rt) unless otherwise stated. Microwave reactions were performed using either the Biotage initiator or CEM Explorer. Workup procedures were carried out using traditional phase separating techniques or by using strong cation exchange (SCX) chromatography using Isolute SPE flash SCX-2 column (International Sorbent Technology Limited, Mid Glamorgan, U.K.). When necessary, organic solutions were dried over anhydrous MgSO₄ or Na₂SO₄. Flash column chromatography (FCC) was performed on Merck Kieselgel silica (article 9385) or on Silicycle cartridges (40–63 µm silica, 4–330 g weight) or on GraceResolv cartridges (4-120 g) either manually or automated using an Isco Combi Flash Companion system. Preparative reverse phase HPLC (RP HPLC) was performed on C18 reversed-phase silica, for example, on a Waters "Xterra" or "XBridge" preparative reversedphase column (5 µm silica, 19 mm diameter, 100 mm length) or on a Phenomenex "Gemini" or "AXIA" preparative reversed-phase column (5 µm silica, 110A, 21.1 mm diameter, 100 mm length) using decreasingly polar mixtures as eluent, for example, containing 0.1% formic acid or 1% aqueous ammonium hydroxide (d = 0.88) as solvent A and acetonitrile as solvent B. The

Journal of Medicinal Chemistry

following preperative chiral HPLC methods were used; in general a flow rate of between 10-350 mL/min and detection was by UV absorbance at a typical wavelength of 254 nm. A sample concentration of about 1-100 mg/mL was used in a suitable solvent mixture such as MeOH, EtOH or iPA optionally mixed with iso-hexane or heptane with an injection volume of between 0.5-100 mL and run time of between 10-150 min and a typical oven temperature of 25-35 °C. The following analytical chiral HPLC methods were used; in general a flow rate of 1 mL/min and detection was by UV absorbance at a typical wavelength of 254 nm. A sample concentration of about 1 mg/mL was used in a suitable solvent such as EtOH with an injection volume of about 10 μ l and run time of between 10-60 min and typical oven temperature of 25-35 °C. The following preperative SFC (supercritical fluid chromotography) methods were used; in general a flow rate of about 70 mL/min and detection was by UV absorbance at a typical wavelength of 254 nm. A sample concentration of about 100 mg/mL was used in a suitable solvent such as MeOH with an injection volume of about 0.5 mL and run time of between 10-150 min and typical oven temperature of 25-35 °C. Intermediates were not necessarily fully purified, but their structures and purity were assessed by TLC, NMR, HPLC, and mass spectral techniques and are consistent with the proposed structures. The purities of compounds for biological testing were assessed by NMR, HPLC, and mass spectral techniques and are consistent with the proposed structures; purity was ≥95%. Electrospray mass spectral data were obtained using a Waters ZMD or Waters ZQ LC/mass spectrometer acquiring both positive and negative ion data, and generally, only ions relating to the parent structure are reported. Unless otherwise stated, ¹H NMR spectra were obtained using a Bruker DRX400 operating at

400 MHz in DMSO-d6 or CDCl₃. Chemical shifts are reported as δ values (ppm) downfield from internal TMS in appropriate organic solutions. Peak multiplicities are expressed as follows: s, singlet; d, doublet; dd, doublet of doublet; t, triplet; br s, broad singlet; m, multiplet. Analytical HPLC was performed on C18 reverse-phase silica, on a Phenomenex "Gemini" preparative reversed-phase column (5 µm silica, 110A, 2 mm diameter, 50 mm length) using decreasingly polar mixtures as eluent, for example, decreasingly polar mixtures of water (containing 0.1% formic acid or 0.1% ammonia) as solvent A and acetonitrile as solvent B or MeOH/MeCN, 3:1, with a flow rate of about 1 mL/min, and detection was by electrospray mass spectrometry and by UV absorbance at a wavelength of 254 nm. Accurate mass spectra were recorded on a Themo LTQ-FT in +ve ion mode with a Thermo Accela pump and Surveyor PDA+ with a CTC autosampler, and the results agreed with the theoretical values to within 4 ppm. Combustion analyses (C, H, N) were performed with a Carlo Erba EA1108 analyzer, and the results agreed with the theoretical values to within $\pm 0.5\%$. Water was measured by the Karl Fischer method using a Mettler DL 18. Where the synthesis of an intermediate or reagent is not described it has either been described in the literature previously or is available from commercial sources.

4-{4-[(3*R*)-3-Methylmorpholin-4-yl]-6-[1-((*R*)-*S*-

methylsulfonimidoyl)cyclopropyl]pyrimidin-2-yl}-1*H*-pyrrolo[2,3-*b*]pyridine (2).
Bis(triphenylphosphine)palladium(II) chloride (27.5 g, 39.14 mmol) was added to a suspension of **R-75** (259 g, 782.87 mmol), 1*H*-pyrrolo[2,3-*b*]pyridin-4-ylboronic acid (152 g, 939.45 mmol)

and sodium carbonate (2M aq.) (1174 mL, 2348.61 mmol) in degassed DME:H2O (4:1) (2740
mL) and stirred to 90 °C for 60 min. A further portion of 1 <i>H</i> -pyrrolo[2,3- <i>b</i>]pyridin-4-ylboronic
acid (15.2 g, 93.95 mmol) was added and the reaction stirred for another 60 min. The reaction
mixture was diluted with EtOAc (200 mL) and washed with water (200 mL). The organic layer
was dried over MgSO ₄ , filtered and then concentrated <i>in vacuo</i> . The crude product was
purified by flash chromatography on silica, eluting with a gradient of 0 to 10 % MeOH in
EtOAc. Pure fractions were evaporated to dryness to afford a cream foam. MTBE (2500 mL)
was added and the mixture stirred at room temperature for 3 days before the solid was isolated
by filtration to afford 2 (139 g, 42%) as a white crystalline solid. ¹ H NMR (400 MHz, DMSO-
d6): 1.19 (3H, d), 1.29 - 1.50 (3H, m), 1.61 - 1.72 (1H, m), 3.01 (3H, s), 3.22 (1H, d), 3.43 (1H,
td), 3.58 (1H, dd), 3.68 - 3.76 (2H, m), 3.87 - 3.96 (1H, m), 4.17 (1H, d), 4.60 (1H, s), 6.98 (1H,
s), 7.20 (1H, dd), 7.55 - 7.58 (1H, m), 7.92 (1H, d), 8.60 (1H, d), 11.67 (1H, s). ¹³ C NMR (176
MHz, DMSO-d6) 11.29, 12.22, 13.39, 38.92, 41.14, 46.48, 47.81, 65.97, 70.19, 101.54, 102.82,
114.58, 117.71, 127.21, 136.70, 142.21, 150.12, 161.88, 162.63, 163.20. HRMS-ESI <i>m</i> / <i>z</i>
413.17529 [MH ⁺]; $C_{20}H_{24}N_6O_2S$ requires 413.1760. Chiral HPLC: (HP1100 system 4, 5 μ m
Chiralpak AS-H (250 mm \times 4.6 mm) column, eluting with <i>iso</i> -hexane/EtOH/MeOH/TEA
50/25/25/0.1) <i>R</i> _f = 8.252, >99%. Anal. Found (% w/w): C, 58.36; H, 5.87; N, 20.20; S, 7.55; H ₂ O,
<0.14. C ₂₀ H ₂₄ N ₆ O ₂ S requires C, 58.23; H, 5.86; N, 20.37; S, 7.77.

4-{4-[(3*R*)-3-Methylmorpholin-4-yl]-6-[1-((*R*)-*S*-

methylsulfonimidoyl)cyclopropyl]pyrimidin-2-yl}-1*H*-pyrrolo[2,3-*b*]pyridine (2), and 4-{4-

[(3 <i>R</i>)-3-Methylmorpholin-4-yl]-6-[1-((<i>S</i>)- <i>S</i> -methylsulfonimidoyl)cyclopropyl]pyrimidin-2-
yl}-1 <i>H</i> -pyrrolo[2,3- <i>b</i>]pyridine (33). Dichlorobis(triphenylphosphine)palladium(II) (0.073 g,
0.10 mmol) was added in one portion to R/S-75 (1.383 g, 4.18 mmol), 2M sodium carbonate
(aq.) (2.508 mL, 5.02 mmol) and 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)-1-tosyl-1 <i>H</i> -
pyrrolo[2,3 <i>b</i>]pyridine (1.665 g, 4.18 mmol) in DME:water 4:1 (100 mL) under nitrogen. The
reaction mixture was stirred at 90 °C for 6 hours. The reaction mixture was concentrated and
diluted with EtOAc (400 mL), and washed sequentially with water (300 mL) and saturated
brine (75 mL). The organic layer was dried over MgSO ₄ , filtered and evaporated onto silica gel
(30 g). The resulting powder was purified by flash chromatography on silica, eluting with a
gradient of 0 to 5% MeOH in DCM. Pure fractions were evaporated to dryness to afford $(3R)$ -3-
methyl-4-(6-(1-(<i>S</i> -methylsulfonimidoyl)cyclopropyl)-2-(1-tosyl-1 <i>H</i> -pyrrolo[2,3- <i>b</i>]pyridine-
4-yl)pyrimidin-4-yl)morpholine (2.174 g, 92%). ¹ H NMR (400 MHz, CDC1 ₃): 1.37 (3H, d), 1.56
(2H, m), 1.83 (2H, q), 2.37 (4H, s), 3.16 (3H, s), 3.36 (1H, td), 3.60 (1H, td), 3.74 (1H, dd), 3.85
(1H, d), 4.01 - 4.19 (2H, m), 4.49 (1H, s), 6.95 (1H, d), 7.28 (2H, d, obscured by CDCl ₃ peak),
7.44 (1H, t), 7.82 (1H, d), 8.02 - 8.11 (3H, m), 8.52 (1H, d). MS-ESI <i>m</i> / <i>z</i> 567 [MH ⁺].

(3*R*)-3-Methyl-4-(6-(1-(*S*-methylsulfonimidoyl)cyclopropyl)-2-(1-tosyl-1*H*-pyrrolo[2,3-

b]pyridine-4-yl)pyrimidin-4-yl)morpholine (1.67 g, 2.95 mmol) was dissolved in DME:water 4:1 (60 mL) and heated to 50 °C. 2M Sodium hydroxide (aq.) (2.58 mL, 5.16 mmol) was then added and heating continued for 18 hours. The reaction mixture was acidified with 2M HCI (~2 mL) to pH5. The reaction mixture was evaporated to dryness and the residue dissolved in

EtOAc (250 mL) and washed with water (200 mL). The organic layer was dried over MgSO₄, filtered and evaporated onto silica gel (10 g). The resulting powder was purified by flash chromatography on silica, eluting with a gradient of 0 to 7% MeOH in DCM. Pure fractions were evaporated and the residue was purified by preparative chiral chromatography on a Merck 50mm, 20µm ChiralCel OI column, eluting isocratically with isohexane/EtOH/MeOH/TEA (50/25/25/0.1) as eluent. The fractions containing the desired compound were evaporated to dryness to afford 2 (0.538 g, 44%) as the first eluting compound. ¹H NMR (400 MHz, DMSO-d6): 1.29 (3H, d), 1.51 (3H, m), 1.70 - 1.82 (1H, m), 3.11 (3H, s), 3.28 (1H, m, obscured by water peak), 3.48 - 3.60 (1H, m), 3.68 (1H, dd), 3.75 - 3.87 (2H, m), 4.02 (1H, dd), 4.19 (1H, d), 4.60 (1H, s), 7.01 (1H, s), 7.23 (1H, dd), 7.51 - 7.67 (1H, m), 7.95 (1H, d), 8.34 (1H, d), 11.76 (1H, s). MS-ESI *m/z* 413 [MH⁺]. Chiral HPLC: (HP1100 System 4, 5µm Chiralcel OJ-H (250 mm x 4.6 mm) column eluting with iso-hexane/EtOH/MeOH/TEA (50/25/25/0.1) Rf, 9.013 >99%; and **33** (0.441 g, 36%) as the second eluting compound. ¹H NMR (400 MHz, DMSO-*d*⁶) 1.28 (3H, d), 1.40 - 1.58 (3H, m), 1.70 - 1.80 (1H, m), 3.10 (3H, d), 3.23 - 3.27 (1H, m), 3.51 (1H, dt), 3.66 (1H, dd), 3.80 (2H, d), 4.01 (1H, dd), 4.21 (1H, d), 4.56 (1H, s), 6.99 (1H, s), 7.22 (1H, dd), 7.54 - 7.61 (1H, m), 7.94 (1H, d), 8.33 (1H, d), 11.75 (1H, s). MS-ESI *m*/*z* 413 [MH⁺]. Chiral HPLC: (HP1100 System 4, 5 μm Chiralcel OJ-H (250 mm x 4.6 mm) column eluting with *iso*-hexane/EtOH/MeOH/TEA (50/25/25/0.1) Rf, 15.685 >99%.

(*R*)-4-(2-Chloro-6-(methylthiomethyl)pyrimidin-4-yl)-3-methylmorpholine (71). Compound 70 (see Ref.¹⁶) (17.0 g, 48.1 mmol) was dissolved in DMF (150 mL), to this was

added sodium methanethiolate (3.4 g, 48.1 mmol) and the reaction was stirred for 1 hour at 25 °C. The reaction mixture was quenched with water (50 mL) and then extracted with Et_2O (3 x 50 mL). The combined organic layers were dried over MgSO₄, filtered and then evaporated. The residue was purified by flash chromatography on silica, eluting with a gradient of 50 to 100% EtOAc in *iso*-hexane. Pure fractions were evaporated to afford **71** (12.63 g, 96%). ¹H NMR (400 MHz, DMSO-d6): 1.20 (3H, d), 2.07 (3H, s), 3.11 - 3.26 (1H, m), 3.44 (1H, td), 3.53 (2H, s), 3.59 (1H, dd), 3.71 (1H, d), 3.92 (1H, dd), 3.98 (1H, br s), 4.33 (1H, s), 6.77 (1H, s). MS-ESI *m*/*z* 274 [MH⁺].

(3*R*)-4-(2-Chloro-6-[(methylsulfinyl)methyl]-4-pyrimidinyl)-3-methylmorpholine (R/S-72). Sodium meta-periodate (2.87 g, 13.44 mmol) was added in one portion to 71 (3.68 g, 13.44 mmol) in water (10 mL), EtOAc (20 mL) and MeOH (10 mL). The resulting solution was stirred at 20 °C for 16 hours. The reaction mixture was diluted with DCM (60 mL) and then filtered. The DCM layer was separated and the aqueous layer washed with DCM (3 x 40 mL). The organic layers were combined, dried over MgSO₄, filtered and then evaporated. The residue was purified by flash chromatography on silica, eluting with a gradient of 0 to 7% MeOH in DCM. Pure fractions were evaporated to afford **R/S-72** (2.72 g, 70%). ¹H NMR (400 MHz, DMSO-d6): 1.22 (3H, d), 2.64 (3H, d), 3.14 - 3.26 (1H, m), 3.45 (1H, td), 3.59 (1H, dd), 3.73 (1H, d), 3.88 - 3.96 (2H, m), 4.00 (1H, d), 4.07 (1H, dt), 4.33 (1H, s), 6.81 (1H, s). MS-ESI *m/z* 290 [MH⁺].

(*R*)-4-(2-Chloro-6-((*S*)-methylsulfinylmethyl)pyrimidin-4-yl)-3-methylmorpholine (S-72) and (*R*)-4-(2-chloro-6-((*R*)-methylsulfinylmethyl)pyrimidin-4-yl)-3-methylmorpholine (R-72). Compound R/S-72 (2.7 g, 9.32 mmol) was purified by preparative chiral chromatography on a Merck 100 mm 20 μ m Chiralpak AD column, eluting isocratically with a 50:50:0.1 mixture of *iso*-hexane:EtOH:TEA as eluent. The fractions containing product were evaporated to afford S-72 (1.38 g, 51%) as the first eluting compound. ¹H NMR (CDCl₃): 1.29 (3H, dd), 2.56 (3H, s), 3.15 - 3.33 (1H, m), 3.46 (1H, tt), 3.55 - 3.83 (3H, m), 3.85 - 4.06 (3H, m), 4.31 (1H, s), 6.37 (1H, s). Chiral HPLC: (HP1100 System 6, 20 μ m Chiralpak AD (250 mm x 4.6 mm) column eluting with *iso*-hexane/EtOH/TEA 50/50/0.1) Rf, 7.197 > 99% and R-72 (1.27 g, 47 %) as the second eluting compound. ¹H NMR (400 MHz, CDCl₃): 1.28 (3H, d), 2.58 (3H, s), 3.26 (1H, td), 3.48 (1H, td), 3.62 (1H, dt), 3.77 (2H, dd), 3.88 - 4.13 (3H, m), 4.28 (1H, s), 6.37 (1H, s). Chiral HPLC: (HP1100 System 6, 20 μ m Chiralpak AD (250 mm x 4.6 mm) column eluting with *iso*hexane/EtOH/TEA 50/50/0.1) Rf, 16.897 > 99%.

Compound **R-72** was prepared on a large scale as follows:

Sodium meta-periodate (960 g, 4488.24 mmol) was added portionwise to **71** (1024 g, 3740.20 mmol) in water (3000 mL), EtOAc (6000 mL) and MeOH (3000 mL). The resulting solution was stirred at 17 °C for 16 hours. The reaction was diluted with DCM (18000 mL) and water (3000 mL). The DCM layer was separated and the water layer washed with DCM (3x 5000 mL). The organics were combined and dried over MgSO₄ and sodium bisulphite, filtered and evaporated. The crude product was purified by flash chromatography on silica, eluting with a

gradient of 10 to 50% MeOH in EtOAc. Pure fractions were evaporated to dryness to afford **R/S-72** (935 g, 86%).

Compound **R/S-72** (1933 g, 6670.58 mmol) was purified by preparative chiral chromatography on a Merck 100 mm 20 μ m Chiralpak AD column, eluting isocratically with a 50:50:0.1 mixture of *iso*-hexane:EtOH:TEA as eluent. The fractions containing product were evaporated to afford **R-72** (886.4 g, 46%) as the second eluting compound. ¹H NMR (400 MHz, CDCl₃, 30 °C): 1.28 (3H, d), 2.58 (3H, s), 3.26 (1H, td), 3.48 (1H, td), 3.62 (1H, dt), 3.77 (2H, dd), 3.88 - 4.13 (3H, m), 4.28 (1H, s), 6.37 (1H, s). MS-ESI *m/z* 290, 292 [MH⁺]. Chiral HPLC: (20 μ m Chiralpak AD (250 mm x 4.6 mm) column eluting with *iso*-hexane/EtOH/TEA 50/50/0.1) Rf, 16.897 > 99%.

N-[({2-Chloro-6-[(3*R*)-3-methylmorpholin-4-yl]pyrimidin-4-yl}methyl)(methyl)oxido-λ6sulfanylidene]-2,2,2-trifluoroacetamide (R/S-74). Iodobenzene diacetate (6.54 g, 20.29 mmol) was added to R/S-72 (5.88 g, 20.29 mmol), 2,2,2-trifluoroacetamide (4.59 g, 40.58 mmol), magnesium oxide (3.27 g, 81.16 mmol) and rhodium(II) acetate dimer (0.224 g, 0.51 mmol) in DCM (169 mL) under air. The resulting suspension was stirred at room temperature for 3 days. Further 2,2,2-trifluoroacetamide (1.15 g, 10.15 mmol), magnesium oxide (0.818 g, 20.29 mmol), rhodium(II) acetate dimer (0.056 g, 0.13 mmol) and iodobenzene diacetate (1.64 g, 5.07 mmol) were added and the suspension was stirred at room temperature for a further 24 hours. The reaction mixture was filtered and silica gel (3 g) was added to the filtrate and then the mixture was evaporated. The resulting powder was purified by flash chromatography on silica, eluting with a gradient of 20 to 50% EtOAc in *iso*-hexane. Fractions containing product were Page 55 of 73

Journal of Medicinal Chemistry

evaporated and the residue was triturated with *iso*-hexane/MTBE to give a solid which was collected by filtration and dried under vacuum to afford **R/S-74** (6.64 g, 82%). ¹H NMR (400 MHz, CDC1₃): 1.33 (3H, d), 3.28 (1H, dd), 3.43 (3H, d), 3.46 - 3.59 (1H, m), 3.62 - 3.71 (1H, m), 3.79 (1H, d), 3.90 - 4.50 (2H, br s), 4.21 (1H, s), 4.66 (1H, dd), 4.86 (1H, dd), 6.50 (1H, d). MS-ESI *m/z* 401, 403 [MH⁺].

N-[({2-Chloro-6-[(3*R*)-3-methylmorpholin-4-yl]pyrimidin-4-yl]methyl)(*R*-methyl)oxido- λ 6-sulfanylidene]-2,2,2-trifluoroacetamide (R-74). To a stirring solution of R-72 (285 g, 983.50 mmol) in DCM (2810 mL) was added magnesium oxide (159 g, 3934.00 mmol), iodobenzene diacetate (475 g, 1475.25 mmol), rhodium(II) acetate dimer (10.87 g, 24.59 mmol) and 2,2,2-trifluoroacetamide (222 g, 1967.00 mmol). The mixture was maintained at room temperature under air for 16 hours. The resultant light brown suspension was removed by filtration. The filtrate was evaporated then purified by flash chromatography on silica, eluting with a gradient of 30 to 100% EtOAc in heptane. Pure fractions were evaporated to afford **R-74** (306 g, 77%). ¹H NMR (400 MHz, CDCl₃, 30 °C): 1.33 (3H, d), 3.25 - 3.38 (1H, m), 3.42 (3H, s), 3.51 (1H, td), 3.66 (1H, dd), 3.79 (1H, d), 4.01 (1H, dd), 4.31 (1H, s), 4.64 (1H, d), 4.77 - 4.87 (1H, m), 6.48 (1H, d). MS-ESI *m/z* 401, 403 [MH⁺].

(3R)-4-(2-Chloro-6-(1-(S-methylsulfonimidoyl)cyclopropyl)pyrimidin-4-yl)-3-

methylmorpholine (R/S-75). Sodium hydroxide (50% aq.) (217 mL, 4059.84 mmol) was added to R/S-74 (27.12 g, 67.66 mmol), 1,2-dibromoethane (23.3 mL, 270.66 mmol) and

tetraoctylammonium bromide (3.70 g, 6.77 mmol) in methyl THF (1000 mL) at 20 °C under nitrogen. The resulting mixture was stirred at 20 °C for 24 hours. Further 1,2-dibromoethane (23.3 mL, 270.66 mmol) was added and the mixture was stirred at 20 °C for a further 24 hours. The reaction mixture was diluted with mTHF (1000 mL) and the aqueous layer separated. The organic layer was diluted further with EtOAc (1000 mL) and washed with water (1500 mL). The organic layer was dried over MgSO₄, filtered and then evaporated. The residue was purified by flash chromatography on silica, eluting with a gradient of 0 to 5% MeOH in DCM. Pure fractions were evaporated to afford **R/S-75** (14.80 g, 66%). ¹H NMR (400 MHz, DMSO-d6): 1.21 (3H, d), 1.39 (3H, m), 1.62 - 1.71 (1H, m), 3.01 (3H, s), 3.43 (1H, tt), 3.58 (1H, dd), 3.72 (1H, d), 3.82 (1H, d), 3.93 (1H, dd), 4.01 (1H, s), 4.38 (1H, s), 6.96 (1H, d). MS-ESI m/z 331, 333 [MH⁺].

(3*R*)-4-(2-Chloro-6-(1-((*R*)-*S*-methylsulfonimidoyl)cyclopropyl)pyrimidin-4-yl)-3-

methylmorpholine (R-75). To a solution of **R-74** (476 g, 1188 mmol), 1,2-dibromoethane (1023 mL, 11876 mmol) and tetra-octylammonium bromide (64.9 g, 118.76 mmol) in mTHF (4750 mL) was added sodium hydroxide (3170 mL, 59380.59 mmol) over 10 min. The mixture was stirred for 4 hours at room temperature. The organic layer was separated, washed with brine (2000 mL) and dried over MgSO₄ then filtered. The filtrate was evaporated then purified by flash chromatography on silica, eluting with a gradient of 0 to 5% MeOH in EtOAc. Pure fractions were evaporated to afford **R-75** (259 g, 66%). ¹H NMR (400 MHz, DMSO-d6): 1.14 – 1.21 (m, 4 H) 1.22 – 1.32 (m, 7 H) 1.34 – 1.47 (m, 3 H) 1.56 (br. s., 1 H) 1.59 – 1.71 (m, 1 H) 3.00

(d, *J*=1.27 Hz, 3 H) 3.09 - 3.23 (m, 2 H) 3.29 (s, 2 H) 3.42 (td, *J*=11.85, 2.91 Hz, 1 H) 3.57 (dd, *J*=11.53, 3.17 Hz, 1 H) 3.70 (d, *J*=11.41 Hz, 1 H) 3.81 (s, 1 H) 3.92 (dd, *J*=11.53, 3.68 Hz, 1 H) 3.95 - 4.13 (m, 1 H) 4.37 (br s, 1 H) 6.95 (s, 1 H). MS-ESI *m/z* 331, 333 [MH⁺].

Biological Evaluation. IC₅₀ values reported are geometric mean values of at least two independent measurements unless otherwise stated.

Cell line studies. Cells lines were grown in RPMI-1640 media, 10% FCS, 2 mmol/L glutamine at 37 °C/5% CO₂ unless indicated otherwise. All cell lines were authenticated via the AstraZeneca (AZ) Cell Bank using DNA fingerprinting short tandem repeat (STR) assays. All revived cells were used within 20 passages, and cultured for less than 6 months.

ATR Kinase Assay. As described in Foote et al.¹⁶

ATR Cell (pChk1) Assay in HT29 Tumor Cells. Assay is a variation on that described in Foote *et al.*,¹⁶ with the use of alterative primary pCHK1 S345 antibodies (Cell Signaling Technologies) and secondary AlexaFluor-488 antibody (Molecular Probes).

In Vivo Studies. All animal experiments were conducted in full accordance with the U.K. Home Office Animal (Scientific Procedures) Act 1986. Anti-tumor studies: Female Swiss nu/nu mice (AstraZeneca, U.K.) were housed in negative pressure isolators (PFI Systems Ltd., Oxon, U.K.). LoVo tumor xenografts were established in 8- to 12-week-old mice by injecting 1×10^7 tumor cells subcutaneously (100 µL in serum free medium) on the left dorsal flank. Animals were randomized into treatment groups when tumors became palpable. Compound **2** was administered orally. Tumors were measured up to three times per week with calipers. Tumor volumes were calculated and the data plotted using the geometric mean for each group versus time.

yH2AX Immunohistochemistry pharmacodynamic analysis (PD): 4 tumors were used per point. Sections of tissue 4 µm were cut on a microtome (Thermo Fisher Scientific) and mounted on electrostatically charged glass slides (Thermo Fisher Scientific). Sections were de-waxed in xylene, passed through graded alcohols and rehydrated in water. Heat mediated antigen retrieval was performed using a RHS2 microwave (Milestone) at 110 °C for two min in pH 9.0 Target Retrieval Solution (cat. #S2367, Dako UK Ltd). Immunohistochemistry was then performed at room temperature on a Lab Vision Autostainer 480 (Lab Vision). 0.05 M Tris buffered saline with 0.1% Tween 20 (TBST), pH 7.6 was used for both the reagent and wash buffers. Sections were treated with 10% hydrogen peroxide in TBST for 10 min, washed and incubated with Background Blocker (cat. #MP-966-P500, A.Menarini Diagnostics Ltd) for 20 min. Sections were incubated with anti-phospho-Ser139 (y)H2AX (cat. #2577, Cell Signalling Technology) for 1 hours before washing and incubating with X-Cell Polymer HRP for fifteen min before being visualised by incubation with the chromogen, diaminbenzidine (DAB) (cat. #MP-, A.Menarini Diagnostics Ltd), for 10 min. Sections were washed for 10 min in H₂O and counterstained with Carrazzis haematoxylin (Clin-Tech), dehydrated though graded IMS, cleared in xylene and mounted using glass coverslips and Histomount (RA Lamb). Slides were scanned and digitized using an Scanscope XT slide scanner (Aperio) at x20 magnification. Once scanned, all digital images representing whole tissue sections, were evaluated for image quality. Immunohistochemistry staining was then quantified using the Spectrum Analysis

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algorithm package and Image Scope viewing software (Aperio). Data are presented as % γ H2AX positive tissue from total tissue area counted, through quantification of pixel counts for γ H2AX positive c (brown) and Hematoxylin only tissue (blue) staining. Only viable tumor was included in the analysis (necrotic area were excluded). Four independent mouse tumors were analysed per time point.

ANCILLARY INFORMATION

Supporting information

Figure S1: Kinome tree graphs for **3**, **21**, **15**. Table S1: kinase inhibition data for **2**. Table S2: ATR activity data for **37**, **38**. Scheme S1: synthetic route to **37**, **38**. Crystallisation methods for compound **2**. Experimental details and data for compounds **5** – **32**, **34** – **38**, **40** – **45**, **50** – **69**, **R/S-73**, **77** – **79**. Crystal formation method of **R-72**. Figure S2: X-ray structure of **R-72**. Aqueous solubility, LogD_{7.4} methods.

Molecular formula strings are available in a separate file.

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Author Contributions

The manuscript was written through contributions of all authors. All authors were employees of AstraZeneca at the time of the described work, and have given approval to the final version of the manuscript.

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Abbreviations. The following abbreviations are used: ACN, acetonitrile; API, active pharmaceutical ingredient; ATM, Ataxia telangiectasia mutated; ATR, Ataxia telangiectasia mutated and rad3 related; AUC, area under the curve; BD, *bis in die* – twice daily; CLint, intrinsic clearance; CYP3A4, cyptochrome P450 3A4 isoform; D_{abs}, maximum absorbable dose; DCM, dichloromethane; DDI, drug-drug interaction; DDR, DNA-damage response; DIPEA, di*iso*propylethylamine; DMA, *N*,*N*-dimethylacetamide; DMAP, 4-dimethylaminopyridine; DME, 1,2-dimethoxyethane; DMF, *N*,*N*-dimethylformamide; DMSO, dimethylsulfoxide; DNA-PK, DNA-dependent protein kinase; DSB, double strand breaks; Et₂O, diethylether;

EtOAc, ethyl acetate; EtOH, ethanol; γ -H2AX, variant histone H2A, phosphorylated; GSH, glutathione; hERG, human ether-a-go-go-related gene; IND, Investigational New Drug; IPA, 2-propanol; LCMS, liquid chromatography mass spectrometer; IR, ionizing radiation; MeOH, methanol; mTHF, 2-methyltetrahydrofuran; MTBE, methyl *tert*-butylether; mTOR, mammalian target of Rapamycin; PBPK, physiologically based pharmacokinetics; PK, pharmacokinetics; PIKK, phosphatidylinositol 3-kinase; RSR, replication stress response; SAR, structure activity relationship; SOC, standard of care; TDI, time-dependent inhibitor; TFA, trifluoroacetic acid; THF, tetrahydrofuran; TGI, tumor growth inhibition; Vd_{ss}, volume of distribution at steady state; QD, *quaque die* – once daily.

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TABLE OF CONTENTS GRAPHIC

1; AZ20 ATR cell IC_{50} = 0.060 μ M Low aqueous solubility CYP3A4 TDI

In vivo tool compound

2; AZD6738 ATR cell IC_{50} = 0.074µM High aqueous solubility No CYP3A4 TDI

Clinical candidate