

First Total Synthesis of Staurosporine and *ent*-Staurosporine

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Staurosporine was first isolated by Omura and colleagues from *Streptomyces staurosporeus*.¹ Its configuration was recently revealed to be that shown in structure 1.² Previously, structure 2, now known to be *ent*-staurosporine, had been assigned to staurosporine.³ (See Chart 1.)

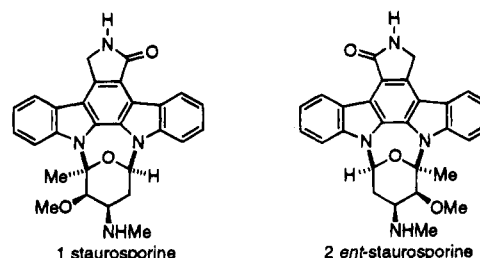
Interest in staurosporine at the biological level springs from its nanomolar inhibition of protein kinase C (PKC).⁴ Given the central role of protein phosphorylation in orchestrating the cell cycle, powerful inhibitors of PKC might be useful as antiproliferative agents.⁵ Needless to say, high levels of tissue specificity in the delivery of a PKC inhibitor would be critical for its development as a drug.

The most obvious chemical challenge posed by a total synthesis of 1^{6,7} is the fashioning of glycosidic bonds to each of the indolic nitrogens. Provision must also be made for the introduction of properly configured methoxy and methylamino groups. A further complication is the daunting issue of regioselectivity in the formation of a lactam from a nearly symmetric imide precursor (cf. 25). Herein we report the first total synthesis of staurosporine (1) and *ent*-staurosporine (2).

Our strategy called for containment of the future *N*-methyl and *O*-methyl moieties in a cyclic framework which would provide chemical protection and offer stereochemical guidance for the intermolecular indole glycosylation. Specifically, we identified oxazolidinone glycal 10 to serve as the glycosyl donor and bis(indolyl) maleimide 5 to function as the aglycon acceptor. (See Scheme 1.)

Aglycon 5 was synthesized from benzyloxymethyl (BOM) dibromomaleimide 3 in the modular fashion shown. Triisopropylsilyl-L-glucal 6 (TIPS-L-glucal) was converted to its bis(trichloroacetimidate) and thence to oxazoline 7 by an apparent

Chart 1



vinyllogous Schmidt glycosylation.⁸ The oxazolidinone, fashioned from the derived 8, was protected as its BOM derivative 9. The TIPS protecting group was cleaved, and a *p*-methoxybenzyl ether (PMB) was installed (see compound 10).

The stage was now set for the intermolecular indole glycosylation. Previous experience had demonstrated that the 1,2 anhydro-sugar coupling method represents, to date, the best available tactic for the glycosylation of complex indoles with complex donors.⁹ Accordingly, glycal 10 was oxidized with 2,2-dimethyldioxirane. The mixture of epoxides (11 and 12) was treated with the sodium salt of 5.¹⁰ Indole glycoside 13 was obtained in 47% yield.¹¹ In the next step, the C₂' hydroxyl function was removed by Barton deoxygenation, yielding 14.¹²

Deprotection of the C₆' PMB and the indolic SEM groups provided seco-system 15. Photolytic oxidative cyclization led to 16.¹³ The *exo*-glycal, needed for the critical intramolecular glycosylation, was fashioned by iodination of 16 followed by elimination. Treatment of 17 with potassium *tert*-butoxide and iodine indeed yielded 18, and after treatment with tri-*n*-butyltin hydride and deprotection of the BOM groups, compound 20 was in hand. (See Scheme 2.)

A BOC group was installed specifically on the oxazolidinone ring (see compound 21). This group was to play a crucial role. Its presence would facilitate disconnection of the oxazolidinone.¹⁴ When opening had been accomplished the BOC group would guard against dimethylation of the amine. To shield the imide sector during operations which would generate the *N*-methyl and *O*-methyl functions, compound 21 was converted into 22. Treatment of 22 with cesium carbonate in methanol gave 23.¹⁵ It was here that the plan for introduction of the *O*-methyl and single *N*-methyl groups was implemented and 24 was secured. Subsequent deprotection was accomplished as shown to provide 7-oxostaurosporine (25), identical with an authentic sample.¹⁶

A protocol was developed to convert the 7-oxo compound 25 to staurosporine itself.¹⁷ It started with a precedented reduction with sodium borohydride.¹⁸ Not easily achieved or precedented was the capacity to deoxygenate the carbanolamide linkage. However, this was eventually smoothly accomplished via the action of benzeneselenol. When this two step sequence was carried out on 25, there was obtained a 1:1 mixture of

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(1) Omura, S.; Iwai, Y.; Hirano, A.; Nakagawa, A.; Awaya, J.; Tsuchiya, H.; Takahashi, Y.; Matsuma, R. *J. Antibiot.* **1977**, *30*, 275.

(2) Funato, N.; Takayanagi, H.; Konda, Y.; Toda, Y.; Harigaya, Y. *Tetrahedron Lett.* **1994**, *35*, 1251.

(3) Furusaki, A.; Hashiba, N.; Matsumoto, T. *J. Chem. Soc., Chem. Commun.* **1978**, 800.

(4) See: Gribble, G.; Berthel, S. *Studies in Natural Products Chemistry*, Vol. 12; Elsevier Science Publishers: New York, 1993; pp 365-409 and references provided therein.

(5) Bradshaw, D.; Hill, C. H.; Nixon, J. S.; Wilkinson, S. E. *Agents Actions* **1993**, *38*, 135.

(6) For previous total synthetic efforts including our construction of the core of staurosporine, see: (a) Link, J. T.; Gallant, M.; Danishefsky, S. J. *J. Am. Chem. Soc.* **1993**, *115*, 3782. (b) Winterfeldt, E. In *Heterocycles in Bio-Organic Chemistry*; Bergman, J., Ed.; The Royal Society of Chemistry: London, 1991; pp 18-27. (c) Weinreb, S. M. *Heterocycles* **1984**, *21*, 309. (d) Joyce, R.; Gainor, J.; Weinreb, S. M. *J. Org. Chem.* **1987**, *52*, 1177. For other related efforts, see: (e) McCombie, S. W.; Bishop, R. W.; Carr, D.; Dobek, E.; Kirkup, M. P.; Kirschmeier, P.; Lin, S. I.; Petrin, J.; Rosinski, K.; Shankar, B. B.; Wilson, O. *Bioorg. Med. Chem. Lett.* **1993**, *3*, 1537.

(7) Selected aglycon synthetic studies: (a) Magnus, P. D.; Exon, C.; Sear, N. *Tetrahedron* **1983**, *39*, 3725. (b) Hughes, I.; Nolan, W. P.; Raphael, R. A. *J. Chem. Soc., Perkin Trans. 1* **1990**, 2475. (c) Moody, C. J.; Rahimtoola, K. F. *J. Org. Chem.* **1992**, *57*, 2105. (d) Bruning, J.; Hache, T.; Winterfeldt, E. *Synthesis* **1994**, 25.

(8) Schmidt, R. R. *Angew. Chem., Int. Ed. Engl.* **1986**, *25*, 212.

(9) Secoaglycons such as 5 are much more potent glycosyl acceptors than are compounds in which the indolocarbazole ring is in place. Gallant, M.; Link, J. T.; Danishefsky, S. J. *J. Org. Chem.* **1993**, *58*, 343.

(10) The epoxidation yielded a 2.5:1 ratio of α -epoxide 11: β -epoxide 12 as determined by ¹H-NMR and by the isolation of the methyl glycosides obtained from their methanolysis and acetylation.

(11) The minor epoxide 12 was a less effective donor than 11. Thus, the 5.5:1 ratio of indole glycosides masks the less selective oxidation.

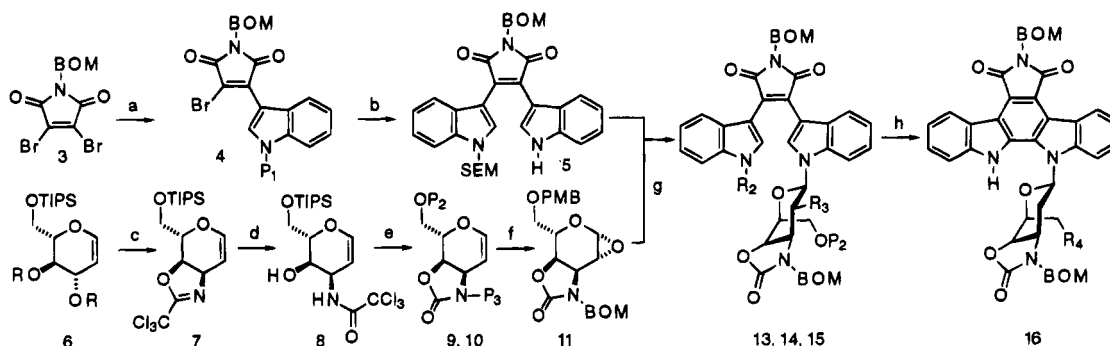
(12) (a) Barton, D. H. R.; Jasberenyi, J. C. *Tetrahedron Lett.* **1989**, *30*, 2619. (b) Gervay, J.; Danishefsky, S. J. *J. Org. Chem.* **1991**, *56*, 5448.

(13) Liu, L.; Yang, B.; Katz, T. J.; Poindexter, M. K. *J. Org. Chem.* **1991**, *56*, 3769 and references contained therein.

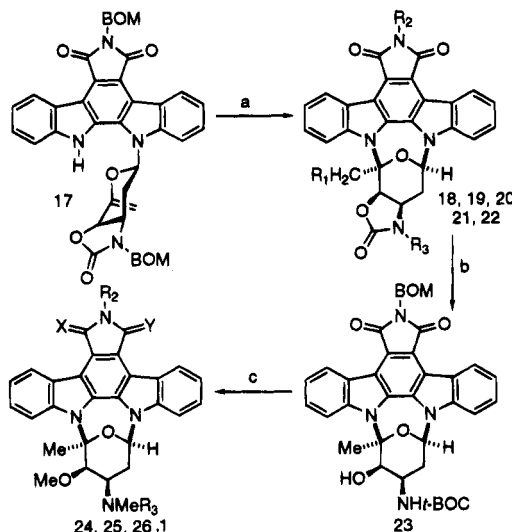
(14) (a) Danishefsky, S.; Berman, E.; Clizbe, L. A.; Hiram, M. *J. Am. Chem. Soc.* **1979**, *101*, 4385. (b) Flynn, D. L.; Zelle, R. E.; Grieco, P. A. *J. Org. Chem.* **1983**, *48*, 2424.

(15) Kuneida, T.; Ishizuka, T. *Tetrahedron Lett.* **1987**, *28*, 4185.

(16) Synthetic material was identical with a natural sample by TLC, ¹H-NMR, ¹³C-NMR, IR, high-resolution mass spectrometry, and rotation.

Scheme 1^a

^a (a) (i) Indole Grignard, PhH, 0 °C → room temperature, overnight, 82% ($P_1 = H$). (ii) NaH, THF, room temperature, then SEMCl, 91% ($P_1 = SEM$). (b) Indole Grignard, PhH, 0 °C → room temperature, overnight, 75%. (c) NaH, CH₂Cl₂, 0 °C, then Cl₃CCN, 0 °C → room temperature, (R = CNHCCl₃), then BF₃·OEt₂, -78 °C, 78%. (d) cat. TsOH, H₂O, pyr, 80 °C, 80%. (e) (i) NaH, CH₂Cl₂, 0 °C → room temperature, 92% of 9 ($P_2 = TIPS$, $P_3 = H$). (ii) NaH, DMF, then BOMCl, 40 °C, 65% ($P_2 = TIPS$, $P_3 = BOM$) and 22% of 9. (iii) TBAF, THF, 0 °C, 95% ($P_2 = H$, $P_3 = BOM$). (iv) NaH, DMF, 0 °C → room temperature, then PMBCl, 0 °C → room temperature, 92% of 10 ($P_2 = PMB$, $P_3 = BOM$). (f) Dimethyldioxirane, CH₂Cl₂, 0 °C, 100% of α-epoxide 11 and β-epoxide 12. (g) (i) 5, NaH, THF, room temperature, then 11 and 12, room temperature → reflux, 47% of 13 ($P_2 = PMB$, $R_2 = SEM$, $R_3 = OH$). (ii) Thiophosgene, DMAP, pyr, CH₂Cl₂, reflux, then C₆F₅OH, reflux, 79% ($P_2 = PMB$, $R_2 = SEM$, $R_3 = OCSOC_6F_5$). (iii) *n*-Bu₃SnH, AIBN, PhH, reflux, 74% 14 ($P_2 = PMB$, $R_2 = SEM$, $R_3 = H$). (iv) DDQ, CH₂Cl₂, H₂O, 0 °C → room temperature, 97% ($P_2 = H$, $R_2 = SEM$, $R_3 = H$). (v) TBAF, THF, reflux, 91% of 15 ($P_2 = H$, $R_2 = H$, $R_3 = H$). (h) (i) *hν*, cat. I₂, air, PhH, room temperature, 73% of 16 ($R_4 = OH$). (ii) I₂, P(Ph)₃, imidazole, CH₂Cl₂, 0 °C → room temperature, 84% ($R_4 = I$).

Scheme 2^a

^a 16 ($R_4 = I$), THF, DBU, room temperature, 89% of 17. (a) (i) *t*-BuOK, I₂, THF, MeOH, room temperature, 65% of 18 ($R_1 = I$, R_2 , $R_3 = BOM$) and 15% of recovered 17. (ii) *n*-Bu₃SnH, AIBN, PhH, reflux, 99% of 19 ($R_1 = H$, R_2 , $R_3 = BOM$). (iii) H₂, Pd(OH)₂, EtOAc, MeOH, room temperature, then NaOMe in MeOH, 92% of 20 (R_1 , R_2 , $R_3 = H$). (iv) (BOC)₂O, THF, cat. DMAP, room temperature, 81% of 21 (R_1 , $R_2 = H$, $R_3 = BOC$). (v) NaH, DMF, room temperature, then BOMCl, 82% of 22 ($R_1 = H$, $R_2 = BOM$, $R_3 = BOC$). (b) Cs₂CO₃, MeOH, room temperature, 93%. (c) (i) NaH, (CH₃)₂SO₄, THF, DMF, room temperature, 86% of 24 ($X, Y = O$, $R_2 = BOM$, $R_3 = BOC$). (ii) H₂, Pd(OH)₂, EtOAc, MeOH, room temperature, then NaOMe in MeOH, 84% ($X, Y = O$, $R_2 = H$, $R_3 = BOC$). (iii) TFA, CH₂Cl₂, room temperature, 97% of 25 ($X, Y = O$, R_2 , $R_3 = H$). (iv) NaBH₄, EtOH, room temperature, workup ($X, Y = O$, OH, R_2 , $R_3 = H$), then PhSeH, cat. TsOH, CH₂Cl₂, room temperature, 39% of 1 ($X = H_2$, $Y = O$, R_2 , $R_3 = H$), 39% of 26 ($X = O$, $Y = H_2$, R_2 , $R_3 = H$), and 15% of 25.

isostaurosporine (26) and *staurosporine* (1). After separation, homogeneous fully synthetic *staurosporine* (1) was isolated.¹⁹ The total synthesis of *staurosporine* (1) has thus been completed.

During the late stages of the synthesis, the absolute configuration of natural *staurosporine* was revealed to be that shown in structure 1 rather than 2. Operating under the prevailing misconception, we had initially carried out the same steps described herein with *D*-glucal as our starting material. This identical sequence led to *ent*-*staurosporine* (2) and to *ent iso* compound 27 (the enantiomer of 26).²⁰ The biological evaluation of these compounds, which is currently underway, should provide unique insights into the interactive roles of the carbohydrate, indolocarbazole, and lactam (or imide) domains in this extraordinary class of PKC inhibitors.

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Supplementary Material Available: Selected analytical data for compounds 13, 18, 25, 26, and 1 (3 pages). This material is contained in many libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.

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(17) We have in fact achieved promising levels of selectivity when the imide reduction is carried out on substrates in which the *staurosporine* core has not been completed. However, those regioselectively produced lactams have not yet been interfaced with intermediates which have gone through to *staurosporine*. We note that the chemistry practiced here, where the protected imide is maintained until virtually the end of the synthesis, works much more smoothly than when it is applied to systems in which the lactam is in place early. See: Link, J. T.; Danishefsky, S. J. *Tetrahedron Lett.* **1994**, 49, 9135.

(18) Caravatti, G.; Meyer, T.; Fredenhagen, A.; Trinks, U.; Mett, H.; Fabbro, D. *Bioorg. Med. Chem. Lett.* **1994**, 4, 399.

(19) *Isostaurosporine* (26) and *staurosporine* (1) were separated by preparative thin layer chromatography on silica gel (19:1 CH₂Cl₂/MeOH). Synthetic *staurosporine* (1) was identical with a natural sample by TLC, ¹H-NMR, IR, high-resolution mass spectrometry, and optical rotation.

(20) Synthetic *ent*-7-oxo*staurosporine* and *ent*-*staurosporine* were identical with natural samples by TLC, ¹H-NMR, ¹³C-NMR, IR, high-resolution mass spectrometry, and magnitude (opposite in sign) of optical rotation.