7 8

9 10

11 12

13 14

Enantioselective Synthesis of Carbo- and Heterocycles Through a CuH-Catalyzed Hydroalkylation Approach

Yi-Ming Wang, Nicholas C. Bruno, Ángel L. Placeres, Shaolin Zhu, and Stephen L. Buchwald*

Department of Chemistry, Massachusetts Institute of Technology, Cambridge, Massachusetts 02139, United States

Supporting Information Placeholder

ABSTRACT: The enantioselective, intramolecular hydroalkylation of halide-tethered styrenes has been achieved through a copper hydride-catalyzed process. This approach allowed for the synthesis of enantioenriched cyclobutanes, cyclopentanes, indanes, and six-membered N- and O-heterocycles. This protocol was applied to the synthesis of the commercial serotonin reuptake inhibitor (–)-paroxetine.

The formation of carbon-carbon (C–C) bonds has long been recognized as a central process in organic synthesis and remains a fundamental objective in the field. In particular, the construction of C–C bonds between *sp*³ centers represents a strategically important approach for the introduction of stereochemical information. A diverse array of catalytic and stoichiometric approaches have been developed for the construction of $C(sp^3)-C(sp^3)$ bonds with high stereoselectivity.¹ However, the majority of these methods rely on the presence of nearby reactive functional groups in the target compound. The formation of stereochemically welldefined sp^3-sp^3 C-C bonds in the absence of such functional groups remains a formidable synthetic challenge.²



Figure 1. Hydroalkylation as a strategy for the construction of unfunctionalized $C(sp^3)-C(sp^3)$ bonds.

The Corey–Posner–Whitesides–House reaction³ between an organocuprate and alkyl halide serves as a prototypical method for the generation of bonds between unfunctionalized saturated carbon atoms. Subsequent work has shown that organocopper species generated under catalytic conditions^{3C,4} similarly react with alkyl halides to form C–C bonds.

Our group's recent work on copper hydride (CuH) catalyzed enantioselective alkene hydroamination⁵ suggested that hydrocupration of an olefin could provide a general approach for the formation of an enantioenriched organocopper species under catalytic conditions (Figure 1). In this context, we posited that an olefin bearing an alkyl (pseudo)halide tether (II, Figure 2) would undergo a formal intramolecular hydroalkylation in the presence of a catalytically generated L*CuH species (I) to furnish enantioenriched, cyclized products.⁶ Although the analogous borocupration/ring closure previously been disclosed,^{4a-d} sequence has competititive reduction of alkyl (pseudo)halides by complexes⁷ renders CuH the proposed hydrocupration/ring closure process nontrivial to Nevertheless, we felt that if suitable execute. conditions could be identified, this strategy would constitute a flexible approach for the synthesis of a variety of 4-, 5-, and 6-membered rings, which are featured prominently in biologically active natural products and pharmaceuticals (Figure 2).⁸ Herein, we report the implementation of this strategy for the enantioselective synthesis of several classes of compounds, including substituted cyclobutanes, cyclopentanes, indanes, and saturated heterocycles.

60

1



Figure 2. (A) Representative saturated 4-, 5-, and 6-membered rings found in pharmaceuticals and natural products. (B) Proposed catalytic cycle for the CuH-catalyzed enantioselective hydroalkylation.

We began our study by examining the reactivity of a homoallylic methanesulfonate in the presence of a DTBM-SEGPHOS-based copper catalyst with diethoxymethylsilane as the hydride source (Table 1). At the outset, we anticipated that the copper (pseudo)halide species (IV, Figure 2) generated upon C-C bond formation would be reluctant to undergo transmetalation with the hydrosilane. We envisioned that the use of an alkoxide base would result in the formation of an intermediate copper alkoxide species (V),⁹ which would more readily transmetalate with hydrosilane to regenerate copper hydride I. Thus during preliminary investigations, a range of alkoxide bases were evaluated. Among them, lithium methoxide was found to be uniquely effective in promoting the desired transformation, providing the cyclobutane product in excellent enantioselectivity, albeit in low yield (entry 3). No desired cyclization product was observed in the absence of base, or when other common alkoxide bases were employed (entries 1, 2, and 4).¹⁰

The effect of the leaving group was then probed in this hydroalkylation protocol. In the presence of additional LiOMe (4.0 eq), bromide was found to be superior to methanesulfonate as the leaving group, affording the desired cyclobutane product in moderate yield and still excellent enantioselectivity (entry 7 vs. entries 3 and 8). Our initial choice of ligand, DTBM-SEGPHOS, was found to be superior to other chiral bisphosphine ligands explored in terms of reactivity (entries 9-11). Finally, the use of dimethoxymethylsilane as the stoichiometric hydride source, as well as further optimization of concentration and temperature, allowed the desired cyclobutane product to be obtained in high yield and excellent enantioselectivity (entry 12).

Under these optimized conditions, we explored the substrate scope of this copper hydride-catalyzed process. Substrates bearing electron-poor aryl substituents reacted efficiently to provide the desired cyclobutane product (2c, 2g, 2h). Replacement of the methyl substituent with a larger ethyl group was also well-tolerated (2b). Substrates containing several functional groups, including the *t*-butyl (2g) and methyl (**2h**) esters, as well as a *cis*-dialkyl olefin (**2e**), were also suitable substrates for the transformation. Under these conditions, an alcohol substrate was rapidly transformed to the corresponding silyl ether, which underwent subsequent intramolecular hydroalkylation to furnish the desired cyclization product (2f). Substrates containing heterocycles, including a dibenzofuran (2i), a benzothiophene (2j), and a pyrrole (2k) also proved competent in the cyclization. However, substrates containing an electron-rich aryl substituents tended to provide cyclization product in lower yields, as exemplified by the preparation of 3,4,5-trimethoxyphenyl-substituted cyclobutane 2d." Nevertheless, enantioselectivities were uniformly high ($\geq 97\%$ ee) in all cases examined, with only one diastereomer (>20:1 dr) observed by ¹H NMR of the crude material. To demonstrate the scalability of the procedure, we performed the synthesis of 2g on a 4.0 mmol scale (1.30 g starting material). The reaction proceeded at this scale without any deleterious effect on yield or enantioselectivity (78% yield, 98% ee).¹²

Table 1. Optimization of Reaction Conditions

		THF (1.0 M), 50 °C, 36 h		Me ^v
entry	х	L*	base (equiv)	% yie l d ^a (% ee ^b)
Effect of ba	se			
1	OMs	L1	KOt-Bu (2.0)	0
2	OMs	L1	LiOt-Bu (2.0)	0
3	OMs	L1	LiOMe (2.0)	17 (96)
4	OMs	L1	NaOMe (2.0)	0
Effect of lea	aving group			
5	OTs	L1	LiOMe (2.0)	9
6	Br	L1	LiOMe (2.0)	16
7	Br	L1	LiOMe (4.0)	70 (97)
8	OMs	L1	LiOMe (4.0)	14
Effect of lig	and			
9	Br	L2	LiOMe (4.0)	16
10	Br	L3	LiOMe (4.0)	8
11	Br	L4	LiOMe (4.0)	5
12	Br	L1	LiOMe (4.0) ^c	92 ^d (99)
12 NMR yields chiral HPLC, 4 (Ar = 3,5-t	Br with 1,3,5-trimeth ²⁴ .0 equiv (MeO) t-Bu ₂ -4-OMe-C ₁	L1 noxybenzene as in 2MeSiH at 2.0 M i 3H ₂)	LiOMe (4.0)° nternal standard, 0.1 mm in THF, 55 °C, #83% iso	92 ^d (99) nol scale, ^b ee determined lated yield, 0.5 mmol sca Ph

Next, we set out to explore the enantioselective synthesis of other ring systems using this approach. Subjection of a homologated substrate to the reac-

tion conditions provided the corresponding cyclopentane product 2l in similarly high yield and enantioselectivity. Notably, a highly electron-poor substrate bearing a 4-trifluoromethyl-3-pyridyl substituent could be used (2m), and a trifluoromethylthio group was also tolerated (2n). As a further extension of this process, we explored the use of substrates with heteroatoms in the tether for the preparation of enantioenriched, saturated heterocycles. Using this approach, 3,4-disubstituted N-sulfonyl piperidines 20 and 2p, as well as tetrahydropyran 2r, could be prepared in moderate to good yield and excellent enantioselectivity. An aryl chloride functional group was tolerated (2p), affording a product that is potentially amenable to subsequent derivatization. The connectivity and stereochemistry of 20 was determined by single crystal X-ray diffraction (Figure 3). As expected, the relative stereochemistry (trans) was consistent with syn-hydrocupration. The absolute stereochemistry was consistent with the sense of stereoinduction observed for the DTBM-SEGPHOS-CuH catalyzed hydroamination.^{5a}



^aYields reported are average isolated yields of two runs (0.5 mmol scale). See the Supporting Information for detailed experimental procedures.

Finally, we sought to expand the hydroalkylation process to the synthesis of 1-alkylindanes through the use of mono- and disubstituted styrenes substrates bearing an alkyl bromide tethered at the *or*-*tho* position. Indeed, subjection of **3a-3c** to hydroal-kylation conditions afforded 1-alkylindane products with high synthetic efficiency, though enantioselec-tivities were somewhat diminished compared to trisubstituted alkene substrates. We note that when a substrate containing both an alkyl chloride and bromide was used, complete selectivity for cyclization at the alkyl bromide was observed while the alkyl chloride remained inert under reaction conditions (**4c**).



Figure 3. X-ray crystallographic structure of 20 (ellipsoids at 50% probability).

To demonstrate the potential applicability of this process to the synthesis of biologically active molecules, we prepared the commercial selective serotonin reuptake inhibitor (–)-paroxetine in seven steps from known starting materials. The key hydroalkylation step on bromide **1r** proceeded smoothly to afford the desired cyclization product in moderate yield and excellent enantioselectivity as one diastereomer. Subsequent deprotection of the sulfonyl group furnished the target compound (Scheme 1). Spectroscopic and optical rotation data matched literature values and further confirmed the stereochemical assignment based on X-ray diffraction (see the Supporting Information).

Scheme 1. Synthesis of (-)-paroxetine



In summary, an intramolecular, enantioselective hydroalkylation of bromide-tethered styrenes was achieved through a copper-catalyzed process. Crucial to the success of this process was the use of lithium methoxide to facilitate regeneration of the proposed copper hydride species and the use of DTBM-SEGPHOS as the supporting ligand. The method presented here was amenable to gram-scale synthesis and could be applied to the synthesis of the pharmaceutical product paroxetine. Importantly, the hydroalkylation process proved general to the synthesis of several scaffolds, including cyclobutanes, cyclopentanes, indanes, and saturated 6-membered heterocycles, all with complete diastereoselectivity and good to excellent levels of enantioselectivity. Efforts toward the development of an intermolecular hydroalkylation reaction are currently under way.

ASSOCIATED CONTENT

Supporting Information. Experimental procedures, characterization data for new compounds, and crystallographic data for **20** (CIF). This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

* <u>sbuchwal@mit.edu</u>

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENT

Research reported in this publication was supported by the National Institutes of Health under award no. GM46059. The content is solely the responsibility of the authors and does not necessarily represent the official views of the National Institutes of Health. Y.-M.W. thanks the National Institutes of Health for a postdoctoral fellowship (GM112218). Á.L.P. gratefully acknowledges the MIT Summer Research Program. We thank Dr. Peter Mueller for X-Ray crystallographic data and Drs. Michael Pirnot and Aaron Sather for their advice on the preparation of this manuscript.

REFERENCES

1. For selected reviews on the enantioselective catalytic formation of $C(sp^3)$ — $C(sp^3)$ bonds, see: (a) Hoveyda, A. H.; Morken, J. P. Angew. Chem. Int. Ed. Engl. 1996, 35, 1262. (b) Kobayashi, S.; Ishitani, H. Chem. Rev. 1999, 99, 1069. (c) Fürstner, A. Chem. Rev. 1999, 99, 991. (d) Johnson, J. S.; Evans, D. A. Acc. Chem. Res. 2000, 33, 325. (e) Mikami, K.; Terada, M.; Matsuzawa, H. Angew. Chem. Int. Ed. 2002, 41, 3554. (f) Davies, H. M. L.; Beckwith, R. E. J. 2003, 103, 2861. (g) Ooi, T.; Maruoka, K. Angew. Chem. Int. Ed. 2007, 46, 4223. (h) Gaunt, M. J. Johansson, C. C. C.; McNally, A.; Vo, N. T. Drug Discovery Today 2007, 12, 8. (i) Shibasaki, M.; Kanai, M. Chem. Rev. 2008, 108, 2853. (j) Trost, B. M.; Brindle, C. S. Chem. Soc. Rev. 2010, 39, 1600. (k) Terada, M. Bull. Chem. Soc. Jpn. 2010, 83, 101. (l) Hassan, A.; Krische, M. J. Org. Process Res. Dev. 2011, 15, 1236. (m) Negishi, E.-i. Arkivoc 2011, viii, 34. (m) Watson, I. D. G.; Toste, F. D. Chem. Sci. 2012, 3, 2899. (n) Yang, L.; Huang, H. Catal. Sci. Technol. 2012, 2, 1099. (0) Schaffer, U.; Mahrwald, R. Chem. Eur. J. 2013, 19, 14346. (p) Tsubogo, T.; Ishiwata, T.; Kobayashi, S. Angew. Chem. Int. Ed. 2013, 52, 6590. (q) Šebesta, R. ChemCatChem 2013, 5, 1069. (r) Tasker, S. Z.; Standley, E. A.; Jamison, T. F. Nature 2014, 509, 299. For selected methods based on stoichiometric auxiliaries, see: (s) Meyers, A. I. Acc. Chem. Res. 1978, 11, 275. (t) Evans, D. A.; Bartoli, J.; Shih, T. L. J. Am. Chem. Soc. 1981, 103, 2127. (u) Myers, A. G.; Yang, B. H.; Chen, H.; McKinstry, L.; Kopecky, D. J.; Gleason, J. L. J. Am. Chem. Soc. 1997, 119, 6496. (v) Ellman, J. A.; Owens, T. D.; Tang, T. P. Acc. Chem. Res. 2002, 35, 984. (w) Job, A.; Janeck, C. F.; Bettray, W.; Peters, R.; Enders, D. *Tetrahedron* 2002, 58, 2253. For an overview of enantioselective C—C bond forming reactions applied to complex molecule synthesis, see: (x) Corey, E. J.; Kürti, L. *Enantioselective Chemical Synthesis*; Academic Press: San Diego, 2013.

- For examples of the enantioselective, catalytic generation of C(sp³)—C(sp³) bonds remote from reactive functional groups, see: (a) Pino, P.; Cioni, P.; Wei, J. J. Am. Chem. Soc. 1987, 109, 6189. (b) Long, J.; Du, H.; Li, K.; Shi, Y. Tetrahedron Lett. 2005, 46, 2737. (c) Arp, F. O.; Fu, G. C. J. Am. Chem. Soc. 2005, 127, 10482. (d) Saito, B.; Fu, G. C. J. Am. Chem. Soc. 2008, 130, 6694.
- (a) Corey, E. J.; Posner, G. H. J. Am. Chem. Soc. 1967, 89, 3911. (b) Whitesides, G. M.; Fischer, W. P.; San Filippo, J.; Bashe, R. W.; House, H. O. J. Am. Chem. Soc. 1969, 91, 4871. For the copper-catalyzed coupling of alkyl halides and Grignard reagents, see: (c) Fouquet, G.; Schlosser, M. Angew. Chem. Int. Ed. Engl. 1974, 13, 82.
- For the construction of C(sp³)—C(sp³) bonds through Cu-catalyzed alkene carboboration, see: (a) Ito, H.; Kosaka, Y.; Nonoyama, K.; Sasaki, Y.; Sawamura, M. Angew. Chem. Int. Ed. 2008, 47, 7424. (b) Ito, H.; Toyoda, T.; Sawamura, M. J. Am. Chem. Soc. 2010, 132, 5990. (c) Zhong, C.; Kunii, S.; Kosaka, Y.; Sawamura, M.; Ito, H. J. Am. Chem. Soc. 2010, 132, 11440. (d) Kubota, K.; Yamamoto, E.; Ito, H. J. Am. Chem. Soc. 2013, 135, 2635. (e) Yoshida, H.; Kageyuki, I.; Takaki, K. Org. Lett. 2013, 15, 952.
- (a) Zhu, S.; Niljianskul, N.; Buchwald, S. L. J. Am. Chem. Soc. 2013, 135, 15746. (b) Zhu, S.; Buchwald, S. L. J. Am. Chem. Soc. 2014, 136, 15913. (c) Niljianskul, N.; Zhu, S.; Buchwald, S. L. Angew. Chem. Int. Ed. 2015, 54, 1638. (d). Shi, S.; Buchwald, S. L. Nat. Chem. 2015, 7, 38. For the independent development of a related system, see: (e) Miki, Y.; Hirano, K.; Satoh, T.; Miura, M. Angew. Chem. Int. Ed. 2013, 52, 10830. (f) Miki, Y.; Hirano, K.; Satoh, T.; Miura, M. Org. Lett. 2014, 16, 1498. For a CuH-catalyzed C—C bond forming reaction resulting in the enantioselective formation of indolines, see: (g) Ascic, E.; Buchwald, S. L. J. Am. Chem. Soc. 2015, 137, 4666.
- 6 Cu-catalyzed diene hydroalkylation: (a) Iwasaki, T.; Shimizu, R.; Imanishi, R.; Kuniyasu, H.; Kambe, N. Angew. Chem. Int. Ed. 2015, 54, 9347. Cu-catalyzed hydroalkylation of alkynes: (b) Uehling, M. R.; Suess, A. M.; Lalic, G. J. Am. Chem. Soc. 2015, 137, 1424. (c) Suess, A. M.; Uehling, M. R.; Kaminsky, W.; Lalic, G. J. Am. Chem. Soc. 2015, 137, 7747. Pd-catalyzed intermolecular hydroalkylation: (d). Urkalan, K. B.; Sigman, M. S. J. Am. Chem. Soc. 2009, 131, 18042. (e) DeLuca, R. J.; Sigman, M. S. Org. Lett. 2013, 15, 92. Co-catalyzed hydroalkylation of [60]fullerene: (f) Lu, S.; Jin, T.; Bao, M.; Yamamoto, Y. J. Am. Chem. Soc. 2011, 133, 12842. (g) Lin, X.; Qing, F.-L. Org. Lett. 2013, 15, 4478. Sc-catalyzed hydromethylation of olefins: (h) Sadow, A. D.; Tilley, T. D. J. Am. Chem. Soc. 2003, 125, 7971. (i) Fontaine, F.-G.; Tilley, T. D. Organometallics 2005, 24, 4340. Lewis acid catalyzed hydromethylation: (j) Parnes, Z. N.; Bolestova, G. I.; Akhrem, I. S.; Vol'pin, M. E.; Kursanov, D. N. J. Chem. Soc., Chem. Commun. 1980, 748. For a two-step one-pot alkene hydromethylation, see (k) Dao, H.T.; Michaudel, Q.; Maxwell, B.D.; Baran, P.S. J. Am. Chem. Soc. 2015, 137, 8046.
- 7. Dang, H.; Cox, N.; Lalic, G. Angew. Chem. Int. Ed. 2014, 53, 752.
- For a review on cyclobutane/cyclopentane natural products and pharmaceuticals, see: (a) Sergeiko, A.; Poroikov, V.V.; Hanus, L.O.; Dembitsky, V.M. Open Med. Chem. J.

, *2*, 26. (b) Heasley, B. *Curr. Org. Chem.* **2014**, *18*, 641. For a perspective on saturated ring systems in drugs, see: (c) Taylor, R.D.; MacCoss, M.; Lawson, A.D.G. *J. Med. Chem.* **2014**, *57*, 5845.

- 9. Tsuda, T.; Hashimoto, T.; Saegusa, T. J. Am. Chem. Soc. 1972, 94, 658.
- The use of other bases were found to be ineffective, including LiOEt, LiOPh, NaOPh, NaO(2,6-t-Bu₂C₆H₃), NaOBz, LiOBz, all of which provided <5% yield of the desired cyclization product.
- 11. We attribute the poorer reactivity of electron-rich substrates to a more difficult hydrocupration step. Reduction of the alkyl bromide to a terminal methyl group was observed as the major side reaction.
- 12. Preliminary experiments indicate that the corresponding (*Z*)-configured homoallylic bromides were unreactive in the present catalyst system.

TOC graphic:

