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Synthesis of chalcones bearing 2-hydroperoxy-3-methyl-3-butenyl or 2-hydroxy-3-methyl-3-butenyl group from prenylated chalcones

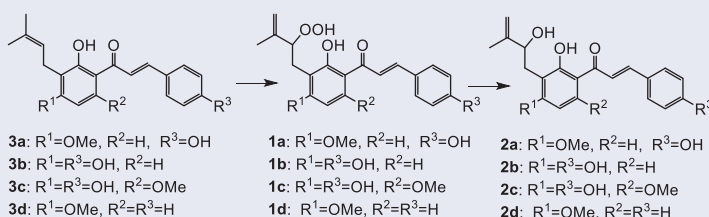
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ABSTRACT

Chalcones bearing 2-hydroperoxy-3-methyl-3-butenyl or 2-hydroxy-3-methyl-3-butenyl groups, such as xanthoangelol E (**1a**), xanthoangelol D (**2a**), psorachalcone A (**2b**), xanthohumol D (**2c**), and related derivatives were first synthesized by using the ene reaction of prenylated chalcones with singlet oxygen.

GRAPHICAL ABSTRACT



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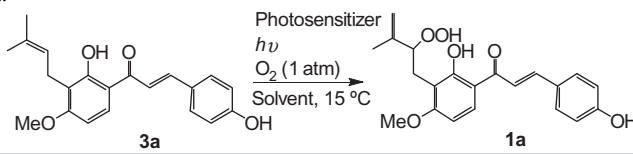
Chalcones bearing 2-hydroperoxy-3-methyl-3-butenyl or 2-hydroxy-3-methyl-3-butenyl groups, such as xanthoangelol E (**1a**), xanthoangelol D (**2a**), psorachalcone A (**2b**), and xanthohumol D (**2c**) were isolated from extracts of plants as biologically active compounds (Supplementary Figure 1). Compounds **1a** and **2a** were isolated from *Angelica keiskei*.^[1] The hydroperoxide **1a** showed to inhibit the production of thromboxane B₂ and 12-hydroxy-5,8,10-heptadecatrienoic acid from exogenous arachidonic acid in platelets^[2] and to inhibit phenylephrine-induced vasoconstriction through endothelium-dependent production of EDRF/NO and/or thorough the reduction of the [Ca²⁺]_i induced by phenylephrine.^[3] The alcohol **2a** was reported to inhibit endothelin-1 production through the suppression of nuclear factor-κB,^[4] to inhibit inflammatory-induced plasminogen activator inhibitor production,^[5] and to inhibit influenza virus neuraminidase.^[6] The alcohol **2b** was reported to be isolated from *Maclura tinctoria*^[7] and *Psoralea corylifolia*.^[8] The alcohol **2c** was isolated from *Humulus lupulus*,^[9] exhibited potent inhibitory activity of nitric oxide production.^[10] Recently, F. Sang et al. reported the synthesis of chalcones bearing 2-hydroxy-3-methyl-3-butenyl groups, such as xanthoangelol D (**2a**), angusticornin A, and paratocarpin E with ene reaction using

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Table 1. Photooxygenation of **3a** under various conditions and lifetime of $^1\Delta_g$ in solvent.



Entry	Conditions Photosensitizer	Solvent	Time (h)	Yield (%) ^a	Lifetime of $^1\Delta_g$ (μs) ^b
1	TPP	CH ₂ Cl ₂	0.5	40	82.9
2	Rose bengal	MeOH	10	38	10.4
3	Methylene blue	MeOH	10	32	10.4
4	Methylene blue	2-PrOH	10	44	22.1
5	Methylene blue	Acetone	1	49	50.5
6	Methylene blue	CH ₃ CN	1	57	58.3

^aIsolated yield.^bThese data are taken from reference 14.

tetraphenylporphyrin (TPP) as the photosensitizer followed by reduction with triphenylphosphine.^[11] However, no work has been done on the synthesis of chalcones bearing 2-hydroperoxy-3-methyl-3-butenyl groups. In this paper, we report the synthesis of these chalcone **1a**, **2a-c** and their related derivatives from prenylated chalcones.

Prenylated chalcones, such as 4-hydroxyderricin (**3a**), isobavachalcone (**3b**), derricin (**3d**), were prepared according to our previous report.^[12] Xanthohumol (**3c**) was synthesized according to the literature.^[13] Helesbeux et al. reported the ene reaction of *ortho*-prenylphenol derivatives with singlet oxygen ($^1\Delta_g$) using tetraphenylporphyrin (TPP) as the photosensitizer followed by a reduction with PPh₃ to give *ortho*-(2-hydroxy-3-methyl-3-butenyl)phenol derivatives.^[14] Although the intermediately produced unstable hydroperoxide was detected by $^1\text{H-NMR}$, it was not isolated. The ene reaction of $^1\Delta_g$ with 4-hydroxyderricin (**3a**) was first attempted in the presence of TPP as the photosensitizer (Table 1, entry 1). Xanthoangelol E (**1a**) with unstable hydroperoxy moiety was first synthesized by this method in 40% yield, along with inseparable by-products. However, compound **1a** was difficult to separate from tpp by SiO₂ column chromatography. Rodgers was reported the natural lifetime of $^1\Delta_g$ in many solvent systems.^[15] Table 1 shows the photooxygenation of **3a** under various conditions and lifetime of $^1\Delta_g$ in each solvents. Although the reaction of **3a** in MeOH using rose bengal (RB) or methylene blue (MB) as photosensitizer were slow, the yield of **1a** was moderate. MB was the easiest to separate from **1a** by SiO₂ column chromatography. Entries 4–6 shows the effect of solvent on the photooxygenation using MB. In the case of using MB as photosensitizer, the reaction time on the photooxygenation of **3a** is dependent on lifetime of $^1\Delta_g$ in each solvent. The yield of **1a** increase with an increase in the lifetime of $^1\Delta_g$. The reaction conditions in entry 6 gave the best result.

The conditions of the photooxygenation using MB in CH₃CN was applied to several prenylated chalcones (Table 2). The photooxygenation of **1b–1d** gave the corresponding chalcones bearing 2-hydroperoxy-3-methyl-3-butenyl group in moderate yields (entries 2–4).

Next, we tried the one-pot synthesis of chalcones bearing 2-hydroxy-3-methyl-3-butenyl groups from the prenylated chalcones. Table 3 shows the photooxygenation of prenylated chalcones in the presence of MB in CH₃CN followed by reduction with

Table 2. Photooxygenation of prenylated chalcones (**3a-d**) in the presence of methylene blue as photosensitizer.

3a: R¹=OMe, R²=H, R³=OH
3b: R¹=R³=OH, R²=H
3c: R¹=R³=OH, R²=OMe
3d: R¹=OMe, R²=R³=H

1a: R¹=OMe, R²=H, R³=OH
1b: R¹=R³=OH, R²=H
1c: R¹=R³=OH, R²=OMe
1d: R¹=OMe, R²=R³=H

Entry	Substrate	Product	Yield (%)
1	3a	1a	57
2	3b	1b	42
3	3c	1c	12
4	3d	1d	37

Bold values represent compound numbers.

Table 3. Synthesis chalcones bearing 2-hydroxy-3-methyl-3-butenyl group (**2a-d**) from prenylated chalcones (**3a-d**).

3a: R¹=OMe, R²=H, R³=OH
3b: R¹=R³=OH, R²=H
3c: R¹=R³=OH, R²=OMe
3d: R¹=OMe, R²=R³=H

2a: R¹=OMe, R²=H, R³=OH
2b: R¹=R³=OH, R²=H
2c: R¹=R³=OH, R²=OMe
2d: R¹=OMe, R²=R³=H

Entry	Substrate	Product	Yield (%)
1	3a	2a	56
2	3b	2b	47
3	3c	2c	39
4	3d	2d	52

Bold values represent compound numbers.

trimethylphosphite. Thus chalcones bearing 2-hydroxy-3-methyl-3-butenyl group such as xanthoangelol D (**2a**), psorachalcone A (**2b**), and xanthohumol D (**2c**) could be synthesized from corresponding prenylated chalcones in 56, 47, and 39% yield, respectively (entries 1–3).

In summary, the present method allowed for the synthesis of chalcones bearing 2-hydroperoxy-3-methyl-3-butenyl or 2-hydroxy-3-methyl-3-butenyl groups, such as xanthoangelol E (**1a**), xanthoangelol D (**2a**), psorachalcone A (**2b**), xanthohumol D (**2c**), and related derivatives from prenylated chalcones.

General procedure for the synthesis of chalcones bearing 2-hydroperoxy-3-methyl-3-butenyl (1a–d)

Oxygen was bubbled through a CH₃CN solution (30 cm³) of prenylated chalcone (30 mg) and methylene blue (3 mg) as photosensitizer. The reaction mixture was water-cooled at 15 °C and irradiated with a halogen lamp (500 W) for 1 h. The reaction was terminated when the substrate was completely consumed by checking on TLC. After removing the solvent under reduced pressure, the residue was purified by silica gel column chromatography with hexane-ethyl acetate.

General procedure for the synthesis of chalcones bearing 2-hydroxy-3-methyl-3-butenyl (2a–d)

Oxygen was bubbled through a CH₃CN solution (30 cm³) of prenylated chalcone (30 mg) and methylene blue (3 mg) as photosensitizer. The reaction mixture was water-cooled at 15 °C and irradiated with a halogen lamp (500 W) for 1 h. The reaction was terminated when the substrate was completely consumed by checking on TLC. To this reaction solution was added 1.3 equiv. of trimethylphosphite at 0 °C, and the mixture was stirred for 3 h. After removing the solvent under reduced pressure, the residue was purified by silica gel column chromatography with hexane-ethyl acetate.

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