

Tetrahedron Letters 40 (1999) 4701-4702

First Total Synthesis of Tetrasubstituted Tetrahydrofuran Lignan, (-)-Virgatusin

Hidemi Yoda,* Masato Mizutani, and Kunihiko Takabe

Department of Molecular Science, Faculty of Engineering, Shizuoka University, Hamamatsu 432-8561, Japan

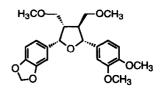
Received 23 March 1999; accepted 23 April 1999

Abstract: The first asymmetric total synthesis of (-)-virgatusin, a new furanolignan, isolated from *phyllanthus virgatus*, was accomplished in a stereoselective manner by nucleophilic addition of organolithium reagent to the functionalized lactone elaborated from dihydroxyacetone dimer followed by asymmetric deoxygenation of the hemiketal intermediate. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: (-)-Virgatusin, Furanolignan, Trisubstituted Lactone, Deoxygenation, Dihydroxyacetone.

Natural lignans display a wide variety of constitution based on phenolic and O-heterocyclic substructures, and an equally wide range of biological activities such as antitumor activity, platelet-activating factor (PAF) antagonists, and inhibitory effects on microsomal monooxygenases in insects.¹ The diverse array of these potentially useful characteristics make them inviting targets for synthesis.² In this connection we have also recently reported the total synthesis of two dibenzylbutyrolactone lignans, (-)-hinokinin^{3a} and (-)-enterolactone,^{3b} employing different synthetic strategies, however, a major sub-group is comprized of tri- and

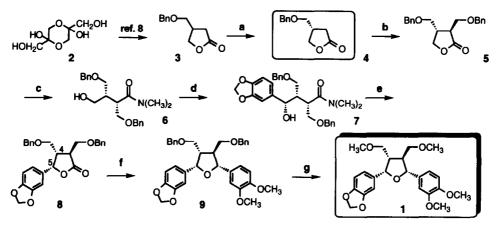
tetrasubstituted tetrahydrofuran groups. Since the synthesis of this type of compounds poses interesting and often unsolved problems of stereocontrol, very few synthetic strategies for the furanolignans have been reported.⁴ Herein we wish to describe the first stereoselective total synthesis of (-)-virgatusin (1) based on the asymmetric Lewis acid-promoted deoxygenation. 1 first isolated in 1996 by Chen et al.⁵ is a new furanolignan with four substituents in the furan ring and is





expected as a herbal drug to inhibit the endogenous DNA polymerase of hepatitis B virus (HBV).6

As shown in Scheme 1, the homochiral benzyllactone 4,⁷ an important building block for the terpenoid synthesis,^{7a} was easily prepared in an enantiomerically pure form starting from dihydroxyacetone dimer 2^8 through diastereomer separation with (R)-(+)- α -methybenzylamine. Hydroxymethylation of 4 with paraformaldehyde followed by benzylation afforded the dibenzyllactone 5 in 96.4% d.e..⁹ After aminolysis of 5 with (CH₃)₂NH, amide 6, thus obtained, was successively subjected to Swern oxidation followed by nucleophilic addition of 3,4-(methylenedioxy)phenylmagnesium bromide *in situ*, leading to the amide alcohol 7 predominantly (80:20 isolated diastereomer ratio)¹⁰ explained in terms of the Cram's non-chelation transition model. This was then cyclized under acidic conditions to give the key trisubstituted lactone 8. Careful treatment of 8 with 3,4-dimethoxyphenyllithium reagent at -78 °C provided the labile hemiketal intermediate, which was readily effected by TiCl₄-induced deoxygenation with Et₃SiH¹¹ at low temperature to lead cleanly to the tetrasubstituted furanolignan derivative 9 as a single stereoisomer in 80% yield from 8 with the desired



Scheme 1. Reagents and conditions: (a) 1, (R)-(+)- α -methylbenzylamine, MeOH, 60 °C; 2, p-TsOH, benzene, 50 °C; 21% (2 steps; diastereomer separation followed by cyclization); (b) 1, LiHMDS, HMPA, (CH₂O)_n, THF, -78~-20 °C; 35%; 2, Ag₂O, BnBr, cat. Bu₄NI; (c) Me₂NH, -20~0°C; 46% (2 steps); (d) 1, (COCI)₂, DMSO, THF then Et₃N, -78~-45 °C; 2, 3,4-(methylenedioxy)phenylmagnesium bromide, THF, 0 °C; 55% (2 steps); (e) p-TsOH, benzene, 50 °C; 87%; (f) 1, 3,4-dimethoxyphenyllithium, Et₂O, -78 °C; 2, Et₃SiH, TiCl₄, CH₂Cl₂, -78 °C; 80% (2 steps); (g) 1, Pd (black), 4.4% HCOOH-MeOH; 93%; 2, NaH, CH₃I, THF; 80%.

configuration.¹² Accompanying formation of the other stereoisomer was not observed in this reaction. Finally, 9 was methylated effectively with NaH-CH₃I after deprotection of the benzyl groups to complete the total synthesis of (-)-virgatusin (1), $[\alpha]_D^{25}$ -12.5 (c 0.51, CH₂Cl₂) [natural 1, $[\alpha]_D^{25}$ -12.7 (c 0.5, CH₂Cl₂)⁵]. The spectral data of the synthetically produced 1 (viscous oil) were completely identical with those of the reported natural product.⁵

In summary, this work constitutes the first synthesis of the natural furanolignan, (-)-virgatusin, and verifies the structure proposed in the literature for this compound.

Acknowledgment:: This work was supported in part by a Grant-in Aid for Scientific Research from the Ministry of Education, Science, Sports, and Culture of Japan.

References and notes

- 1. (a) Ward, R. S. Tetrahedron, 1990, 46, 5029~5041. (b) Ward, R. S. Nat. Prod. Rep. 1993, 10, 1-28 and references cited therein.
- For recent examples: (a) Gaboury, J. A.; Sibi, M. P. J. Org. Chem. 1993, 58, 2173-2180. (b) van Oeveren, A.; Jansen, J. F. G. A.; Feringa, B. L. J. Org. Chem. 1994, 59, 5999-6007. (c) Bode, J. W.; Doyle, M. P.; Protopopova, M. N.; Zhou, Q.-L. J. Org. Chem. 1996, 61, 9146-9155.
- (a) Yoda, H.; Naito, S.; Takabe, K.; Tanaka, N.; Hosoya, K. Tetrahedron Lett. 1990, 31, 7623~7626. (b) Yoda, H.; Kitayama, H.; Katagiri, T.; Takabe, K. Tetrahedron 1992, 48, 3313~3322.
- (a) Stevens, D. R.; Whiting, D. A. J. Chem. Soc., Perkin Trans. 1 1992, 633-637 (b) Mitra, J.; Mitra, A. K. J. Chem. Soc., Perkin Trans. 1 1992, 1285-1286 (c) Maiti, G.; Adhikari, S.; Roy, S. C. J. Chem. Soc., Perkin Trans. 1 1995, 927-929.
- 5. Huang, Y.-L.; Chen, C.-C.; Hsu, F.-L.; Chen, C.-F. J. Nat. Prod. 1996, 59, 520-521.
- 6. Thyagarajan, S. P.; Subramanian, S.; Thirunalasundari, T.; Venkateswaran, P. S.; Blumberg, B. S. Lancet 1988 1, 764~766.
- (a) Takabe, K.; Tanaka, M; Sugimoto, M.; Yamada, T.; Yoda, H. Tetrahedron: Asymmetry 1992, 3, 1385~1386. (b) Mazzini, C.; Lebreton, J.; Alphand, V.; Furstoss, R. J. Org. Chem, 1997, 62, 5215~5218.
- 8 Yoda, H.; Mizutani, M.; Takabe, K. Synlett 1998, 855-856.
- 9. The d.e. was determined by chiral HPLC using Daicel chiralpak AD after derivatization to the amide 6.
- Stereochemistry of the new stereocentre was assigned based on our previous results,⁸ and observed chemical shift (C₅-H) and coupling constant (J_{4,5} = 8.4 Hz) after lactonization to 8 according to the following reference; Marino, J. P.; de la Pradilla, R. F. Tetrahedron Lett. 1985, 26, 5381-5384.
- 11. Yoda, H.; Mizutani, M.; Takabe, K. Heterocycles 1998, 48, 679~686.
- 12. The absolute configuration of the generated stereogenic centre was determined unambiguously based on its spectral data of synthetic (-)-1.