

Bicyclic core estrogens as full antagonists: synthesis, biological evaluation and structure–activity relationships of estrogen receptor ligands based on bridged oxabicyclic core arylsulfonamides†Manghong Zhu,^{‡a} Chen Zhang,^{‡a} Jerome C. Nwachukwu,^b Sathish Srinivasan,^b Valerie Cavett,^b Yangfan Zheng,^a Kathryn E. Carlson,^c Chune Dong,^a John A. Katzenellenbogen,^{*c} Kendall W. Nettles^b and Hai-Bing Zhou^{*a}

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Compounds that block estrogen action through the estrogen receptor (ER) or downregulate ER levels are useful for the treatment of breast cancer and endocrine disorders. In our search for structurally novel estrogens having three-dimensional core scaffolds, we found some compounds with a 7-oxabicyclo[2.2.1]heptene core that bound well to the ERs. The best of these compounds, a phenyl sulfonate ester (termed OBHS for oxabicycloheptene sulfonate), was a partial antagonist on both ER α and ER β . Although OBHS bears no structural resemblance to other estrogen antagonists, it appears to achieve its partial antagonist character by stabilizing a novel conformation of the ER that involves a significant distortion of helix-11. To enhance the antagonist properties of these oxabicyclo[2.2.1]heptane core ligands, we expanded the functional diversity of OBHS by replacing the sulfonate with secondary or tertiary sulfonamides (–SO₂NR–), isoelectronic and potentially isostructural molecular replacements. An array of 16 OBHS sulfonamide analogues were prepared through a Diels–Alder reaction of a 3,4-diarylfuran using various *N*-aryl vinyl sulfonamide dienophiles. While the more polar secondary sulfonamides were weak ligands, certain of the tertiary sulfonamides had very good ER binding affinity. In HepG2 cell reporter gene assays, the sulfonamides had moderate potency, but they showed lower intrinsic transcriptional activity on ER α than the selective estrogen receptor modulator (SERM) hydroxytamoxifen or OBHS, and they were inverse agonists on ER β . Thus, the behaviour of these OBH-sulfonamides more closely mirrors the activity of full antagonists like the drug fulvestrant (ICI 182 780), and their greater antagonist biocharacter appears to arise from an accentuated distortion of helix-11.

Introduction

The estrogen receptors, ER α and ER β , regulate a diverse set of physiological and pathological processes and are well established

pharmaceutical targets.^{1,2} In contrast to the pan-agonist activity of 17 β -estradiol (E₂, Fig. 1), Selective Estrogen Receptor Modulators (SERMs) display agonist activity in certain tissues but antagonist activity in others,^{3,4} and some SERMs, such as tamoxifen and raloxifene, are used for the treatment of breast cancer or for menopausal hormone replacement.^{5–7} Because activity profiles of these SERM are not ideal and resistance to their effectiveness as antitumor agents can develop with time, there has been interest in finding new SERMs that might prove more effective as hormonal or therapeutic agents.^{6,7} ER ligands that reduce the *level* (as well as the *activity*) of ER are termed Selective Estrogen Receptor Downregulators (SERDs),^{8,9} and because they actually lower ER levels, SERDs are distinct from SERMs. SERDs such as fulvestrant (Fig. 1, ICI 182 780) are showing promise in the treatment of metastatic breast cancer, because they can inhibit the growth of tamoxifen-resistant breast cancer cells.¹⁰ Fulvestrant, however, has a poor oral bioavailability; so, there is also a need for improved SERDs.

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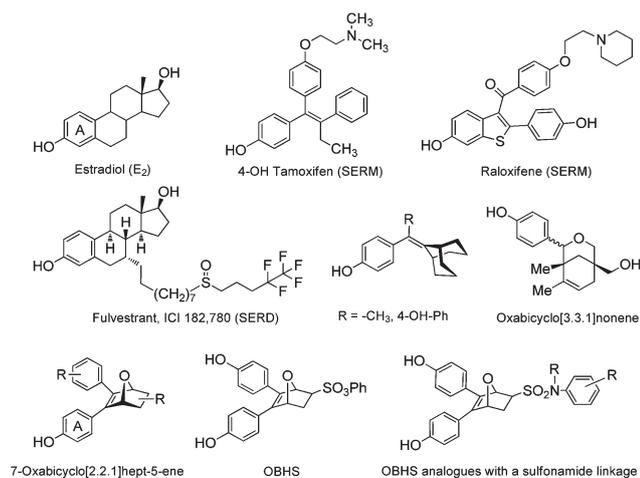
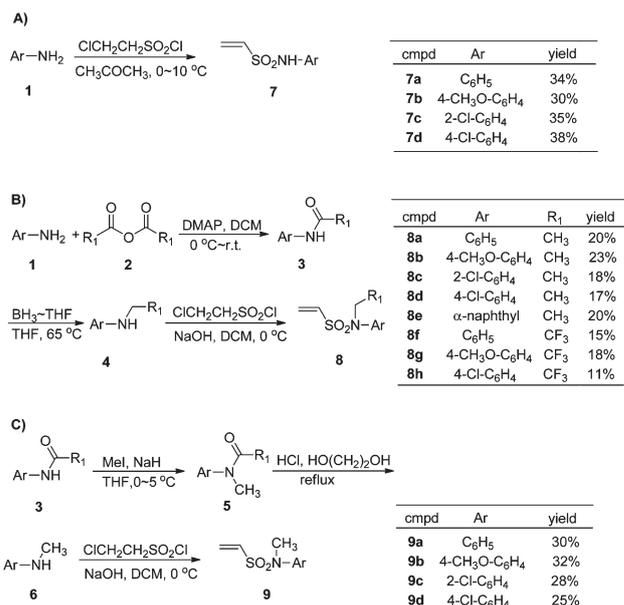


Fig. 1 The structure of estradiol, 4-OH tamoxifen, fulvestrant (ICI 182 780) and representative three-dimensional ER ligands, OBHS and title compounds.

In a new approach to develop novel SERMs and SERDs, we prepared compounds having a more three-dimensional central hydrophobic core topology than is typically found in steroidal and non-steroidal ER ligands. This design was inspired by structural studies of ligand complexes with ER that reveal ample unoccupied space above and below the mean plane of E₂, particularly near the middle of this molecule.^{11,12} A number of structural motifs, such as the bridged bicyclo[3.3.1]nonene core systems (Fig. 1), have been explored by us¹³ and others^{14,15} to probe this extra space and exploit the flexibility of the ligand-binding pocket (LBP).

Recently, we evaluated three-dimensional ER ligands based on a different bridged oxabicyclo[2.2.1]heptene core (Fig. 1).¹⁶ The best compound, *exo*-5,6-bis-(4-hydroxyphenyl)-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonic acid phenyl ester (OBHS), exhibited relative binding affinity (RBA) values of 9.3% and 1.7% for ER α and ER β , respectively (RBA[estradiol] = 100%). OBHS also profiled as a partial antagonist on both ER subtypes, even though it was structurally unlike typical ER antagonists. Through our recent structural studies of this compound¹⁷ as well as other members of this series,¹⁸ it became evident that the high ER binding affinity of OBHS relies on its two 4-hydroxyphenyl substituents, one which mimics the A-ring of estradiol, the other which projects into a subpocket that lies in the 11 β direction with respect to estradiol in the complex with ER α . Of greater interest, however, was the fact that the large phenyl sulfonate moiety, which is too long and extended to fit into the ligand-binding pocket found in crystal structures of typical steroidal and non-steroidal estrogens and SERMs, was readily accommodated by a reorganization of the peptide backbone and residues at the end of helix-11. These changes provided sufficient volume to accommodate this large and extended group, without encountering a marked penalty in ligand binding affinity. This distortion of helix-11 was presumed to result in the overall partial antagonist character of OBHS-type ligands by an indirect—rather than a direct—mechanism through which helix-12 becomes displaced from its agonist conformation because of the dislocation of the C-terminus of helix-11, rather than by direct ligand contact with helix-12.¹⁹



Scheme 1 The synthesis of various dienophiles 7–9. (Yields given are for the final sulfonation step; yields for the other steps are given in the ESI†.)

We were intrigued by the role that the phenyl sulfonate moiety played in engendering the partial antagonist activity of OBHS ligands,¹⁷ and because of our interest in finding ER ligands that have more complete antagonist, even SERD activity, we have, in this study, queried the structure of OBHS at this very position by substituting the sulfonate with secondary and tertiary sulfonamide (–SO₂NR–) linkages, keeping the remainder of the 7-oxabicyclo[2.2.1]hept-5-ene skeleton intact. While the secondary phenyl sulphonamides are isostructural to the phenyl sulfonate, the tertiary sulphonamides introduce an additional substituent that, as will be seen, is of consequence. We prepared a set of 16 OBHS sulfonamides (11a–q) by an efficient Diels–Alder approach,¹⁶ and we evaluated them for ER binding affinity, for transcriptional activity in a relevant cell culture assay system, and by computational modelling for structural analysis. In the process, we have identified some oxabicyclo[2.2.1]heptane sulphonamides that are more complete antagonists than even hydroxytamoxifen, having biological character more like that of fulvestrant (ICI 182 780). Computational modelling indicates that this more complete antagonist biocharacter arises from a more accentuated distortion of helix-11 by the sulphonamides than was effected by the phenylsulfonate of OBHS.

Results

Synthesis

The target sulfonamides were prepared by Diels–Alder cycloaddition¹⁶ of 3,4-bis(4-hydroxyphenyl)furan **10** with various sulfonamide dienophiles (7a–d, 8a–h and 9a–d, see ESI† for details),^{20,21} readily obtained from anilines (Scheme 1). Dienophiles with a secondary sulfonamide –SO₂NH– system (7a–d) were prepared in a single-step by the reaction of 2-chloroethanesulfonyl chloride (1.2 equiv.) with the aniline **1** (Scheme 1A);²²

tertiary sulfonamides (*N*-substituents CH₃, CH₂CH₃, CH₂CF₃) were synthesized as shown (Scheme 1B and C). Acylation of anilines **1** with acetic anhydride or trifluoroacetic anhydride gave compounds **3a–h**,^{23,24} and compounds **4** were obtained by borane reduction of compounds **3**, under optimized conditions.^{25–27} The compounds **6** were prepared from **3** in two steps.²⁸ In the first step, amide **3** was methylated by methyl iodide in the presence of sodium hydride in THF to afford **5**. In the second step, deacylation of **5** in 10% HCl (0.25 mL for 1 mmol amide) and glycol (0.75 mL for 1 mmol amide) with refluxing for 24 h gave the compounds **6**. Reaction of compounds **1**, **4** and **6** with 2-chloroethanesulfonyl chloride gave dieneophiles **7–9**.

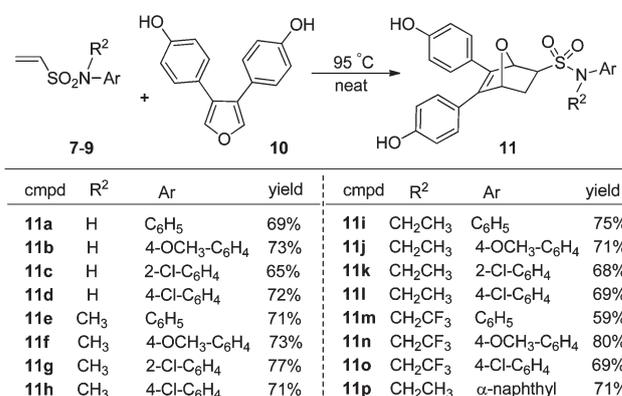
As was the case with the sulfonamides, the Diels–Alder cycloaddition with furan **10** proceeded well at 95 °C without solvent or catalysts, giving isolated yields of 60–80% (Scheme 2).¹⁶ As noted in the earlier sulfonate series, the products are almost exclusively *exo* diastereomers (see ESI† for ¹H NMR assignments of *exo* **11p**); apparently, the high rate and ready reversibility of this reaction results in the predominant formation of the product of thermodynamic control. All of the products are racemates.¹⁶

Binding affinities

ER binding affinities, determined radiometrically,²⁹ are expressed as relative binding affinity (RBA) values, with estradiol = 100 (Table 1). The ER binding affinities depend on the nature of substituents on both the sulfonamide nitrogen and

phenyl group. The RBA values of the secondary sulfonamide –SO₂NH– compounds (**11a–d**) are all rather low (entries 1–4), but all of the tertiary sulfonamides (**11e–o**) bound much better. The *N*-methyl compounds **11e–h**, in particular, showed moderate to high binding affinities (entries 5–8), with compound **11g** being best (7.17% and 1.59% for ERα and ERβ, respectively). Affinity decreased with increasing alkyl chain length, however, with the trifluoroethyl compounds (**11m–o**) having binding affinities similar to or somewhat lower than those of the ethyl compounds (**11i–l**).

Substituents on the pendant phenyl group of the tertiary sulfonamides also affected binding affinity and selectivity, with most



Scheme 2 Diels–Alder reaction of 3,4-dihydroxyphenyl **10** with dieneophiles **7–9**.

Table 1 Relative binding affinities (RBAs) of the compounds **11a–p** for ERα and ERβ^a

Entry	Cmpd	R ²	Ar	RBA ERα	RBA ERβ	α/β ratio
1	11a	H	C ₆ H ₅	0.164 ± 0.02	0.047 ± 0.005	3.49
2	11b	H	4-CH ₃ O-C ₆ H ₄	0.010 ± 0.001	0.007 ± 0.001	1.43
3	11c	H	2-Cl-C ₆ H ₄	0.160 ± 0.01	0.193 ± 0.03	0.83
4	11d	H	4-Cl-C ₆ H ₄	0.094 ± 0.02	0.123 ± 0.03	0.76
5	11e	CH ₃	C ₆ H ₅	2.87 ± 0.19	0.748 ± 0.15	3.84
6	11f	CH ₃	4-CH ₃ O-C ₆ H ₄	0.57 ± 0.06	0.22 ± 0.06	2.64
7	11g	CH ₃	2-Cl-C ₆ H ₄	7.17 ± 1.3	1.59 ± 0.09	4.51
8	11h	CH ₃	4-Cl-C ₆ H ₄	1.06 ± 0.18	0.584 ± 0.050	1.82
9	11i	CH ₂ CH ₃	C ₆ H ₅	1.38 ± 0.10	0.370 ± 0.01	3.73
10	11j	CH ₂ CH ₃	4-CH ₃ O-C ₆ H ₄	0.404 ± 0.04	0.038 ± 0.003	10.6
11	11k	CH ₂ CH ₃	2-Cl-C ₆ H ₄	0.797 ± 0.13	0.146 ± 0.04	5.46
12	11l	CH ₂ CH ₃	4-Cl-C ₆ H ₄	1.05 ± 0.24	0.399 ± 0.07	2.63
13	11m	CH ₂ CF ₃	C ₆ H ₅	0.861 ± 0.03	0.138 ± 0.01	6.24
14	11n	CH ₂ CF ₃	4-CH ₃ O-C ₆ H ₄	0.617 ± 0.08	0.064 ± 0.02	9.64
15	11o	CH ₂ CF ₃	4-Cl-C ₆ H ₄	0.379 ± 0.04	0.125 ± 0.03	3.03
16	11p	CH ₂ CH ₃	α-naphthyl	0.974 ± 0.09	0.109 ± 0.02	8.94

RBA ERα	RBA ERβ	α/β ratio
9.28 ± 0.64	1.71 ± 0.24	5.47 OBHS
0.32 ± 0.10	0.10 ± 0.02	3.3
19.0 ± 4.6	1.76 ± 0.13	10.8
0.78 ± 0.002	0.284 ± 0.003	2.75

^a Relative binding affinity (RBA) values, determined by radiometric assays, are expressed as IC₅₀ estradiol/IC₅₀ compound × 100 ± the range or standard deviation (RBA, estradiol = 100%). The K_d for estradiol is 0.2 nM (ERα) and 0.5 nM (ERβ).

Table 2 Effects of compounds **11a–p** on ER-mediated transcription

Compound ^e	Agonist mode ^a		Antagonist mode ^b			
	ER α		ER α		ER β	
	EC ₅₀ (μ M)	Eff (%E ₂)	IC ₅₀ (μ M)	Eff (%E ₂) ^c	IC ₅₀ (μ M)	Eff (%E ₂) ^c
11a	0.10	78 \pm 1	—	83 \pm 5	0.37	21 \pm 6
11c	0.20	81 \pm 4	—	89 \pm 4	0.10	-5 \pm 3
11d	—	42 \pm 3	—	65 \pm 3	0.10	-15 \pm 1
11e	—	31 \pm 3	0.74	56 \pm 2	0.01	-25 \pm 3
11g	0.008	33 \pm 1	0.45	41 \pm 1	0.12	-24 \pm 3
11i	—	11 \pm 1	0.35	10 \pm 2	0.33	-10 \pm 2
11j	—	3 \pm 1	0.72	3 \pm 1	0.37	-4 \pm 4
11k	—	13 \pm 1	0.75	20 \pm 4	—	3 \pm 3
11l	—	1 \pm 1	0.43	7 \pm 1	0.16	-12 \pm 1
11m	—	9 \pm 2	0.19	3 \pm 1	2.14	0 \pm 2
11n^d	—	0 \pm 0	<i>n.d.</i>	-12 \pm 0	<i>n.d.</i>	45 \pm 5
11o	—	3 \pm 3	0.93	3 \pm 2	4.08	7 \pm 2
11p	0.086	24 \pm 0	6.49	23 \pm 2	2.36	31 \pm 2
Fulvestrant	—	1 \pm 1	0.00033	-7 \pm 1	0.00058	-23 \pm 0
4-OH TAM	0.0011	35 \pm 3	0.0030	35 \pm 3	0.00063	-20 \pm 2
OBHS	0.028	57 \pm 3	0.014	70 \pm 12	0.16	-2 \pm 5

^a Transcriptional activity in HepG2 cells transfected with 3X-ERE-driven luciferase reporter and ER α or ER β expression vectors treated in triplicate with doses of the compounds (up to 10⁻⁵ M). Average efficacy (mean \pm s.e.m.) is shown as a percentage of 10⁻⁷ M E₂. ^b IC₅₀ and average efficacy (mean \pm s.e.m.) determined in the presence of 10⁻⁸ M E₂ on ER α or ER β . ^c ERs have considerable basal activity in HepG2 cells; compounds with inverse agonist activity are given negative efficacy values. ^d A single dose (10⁻⁵ M) of **11n** was tested. ^e Compounds **11b**, **11f**, and **11h** were not assayed.

ligands showing moderate to good binding affinity; those with no substitution (**11a**, **11e**, **11i** and **11m**) were overall the best, with the exception of the *ortho*-chloro phenyl analogue **11g**. The *para* chloro and methoxyl phenyl analogues usually gave lower binding affinities; nevertheless, the sulfonamide with the bulkiest substituent (α -naphthyl) still bound well. The tertiary sulfonamides showed 2.5–11-fold affinity preferences for ER α . It is notable that the affinities of the *N*-methyl sulfonamides (**11e–h**) were comparable with those of the corresponding sulfonates, on which we have recently reported (Table 1, right, data in square brackets),¹⁷ the best sulfonamide (**11g**) being comparable to the original OBHS compound, 7.2% and 1.6% (**11g**) vs. 9.3% and 1.7% (OBHS), for ER α and ER β , respectively.¹⁶

Transcriptional activity

While SERMs generally profile as antagonists in breast cancer cells, they display considerable gene activation in other cell lines that correlates more accurately with their uterotrophic activity *in vivo*.³⁰ For this reason, we profiled these compounds in human hepatocarcinoma (HepG2) cells, using expression plasmids for full-length human ER α or ER β and an estrogen-responsive luciferase reporter gene.³¹ Compounds were assayed alone for direct activation (agonism) and in the presence of 10 nM E₂ for antagonism. Since none of the compounds activated ER β , we report only ER β antagonist data. Agonist and antagonist potencies are given as EC₅₀ and IC₅₀ values, respectively. The intrinsic activity, noted as the % efficacy (%Eff) compared to 100 nM E₂, is given in all cases. (Note, when efficacy is low, it is difficult to determine EC₅₀ values for agonists, and when efficacy is high, it is difficult to determine IC₅₀ values for antagonists.)

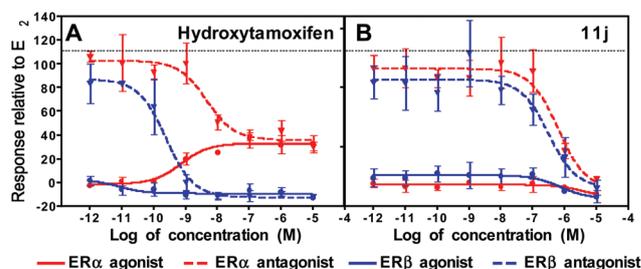


Fig. 2 Luciferase activity was measured in HepG2 cells transfected with 3X-ERE-driven luciferase reporter and expression vectors encoding ER α or ER β , and treated in triplicate with increasing doses (up to 10⁻⁵ M) of **A**. Hydroxytamoxifen; **B**. Compound **11j**. The average efficacy (mean \pm s.e.m.) is shown as a percentage of 10⁻⁷ M 17 β -estradiol (E₂). For the antagonist mode the average efficacy (mean \pm s.e.m.) of the compounds was determined in the presence of 10⁻⁸ M E₂ and is shown as a percentage of 10⁻⁸ M E₂.

In these cells, the SERD, fulvestrant (ICI 182 780), acts as a full antagonist on both receptors (Table 2), while 4-OH tamoxifen (an active tamoxifen metabolite) displays the expected ER α -selective partial-activation profile (Table 2, Fig. 2A). While OBHS is a full antagonist in HEC-1 cells,¹⁶ in HepG2 cells it profiles with greater agonist efficacy than 4-OH tamoxifen. Because ER α and especially ER β have considerable basal activity in HepG2 cells, compounds can show inverse agonist activity; those with intrinsic activity less than that of the apo-ER are reported with negative efficacy values.

With ER α there is a clear trend where agonist activity decreased with an increase in the *N*-alkyl chain length (**11a** > **11e** > **11i**). Overall, the *N*-ethyl and trifluoroethyl compounds have very low efficacy, with those having a 2-chlorophenyl

substitution (**11c**, **11g**, **11k**) showing greater agonist activity than those with substitutions at the 4-position. Also, the naphthyl derivative (**11p**) induced nearly full agonist activity on ER α , despite the bulky *N*-ethyl sulfonamide substitution, highlighting that both substituents on nitrogen are important. Shown in Fig. 2B is a representative assay curve for a low-efficacy compound (**11j**), which has an intrinsic activity on both ER α and ER β as low as that for fulvestrant (Table 2).

On ER β , most of the compounds profiled as inverse agonists, similar to fulvestrant. The inability to fully displace E₂ for a few compounds may reflect their low affinity, as they did not activate ER β on their own.

Discussion

In this work, we have expanded the chemical diversity of OBHS-type ligands by substitution of the sulfonate moiety with a sulfonamide (–SO₂NR–) linkage, a molecular replacement that is isoelectronic and potentially isostructural. A small array of 16 OBHS analogues (**11a–p**) were prepared in moderate to good yield, through a Diels–Alder reaction of a 3,4-diarylfuran with various dienophiles of *N*-aryl vinyl sulfonamides under neat, mild conditions, without catalysts.

The structure–activity relationships that emerge for these OBHS sulfonamide analogues from binding and cell-based activity assays reveal that the affinities depend on the nature of the substituents on both the nitrogen atom and the phenyl group of the sulfonamides; the highest affinity being observed in tertiary sulfonamides having small *N*-alkyl groups and binding enhancing groups on the phenyl group.

Even though the secondary sulfonamides are more isostructural to the sulfonates than are the tertiary sulphonamides, their lower affinity is not surprising. The N–H group in the secondary sulphonamides presents a hydrogen bond donor in a region of the receptor that has no available hydrogen bond acceptor. It is well known that introduction of strong hydrogen bonding groups, both donors and acceptors, in the “middle” of ER ligands generally results in significantly reduced binding affinities.^{32,33} The tertiary sulphonamides, however, have masked this hydrogen bond donor with an alkyl group, and some of these compounds demonstrated affinities and ER α

selectivity comparable to those of the parent sulfonate systems, OBHS and its analogues.

The second substituent, which can be added in the sulfonamide—but not the sulfonate—system, not only improves binding affinity, but also leads to enhanced antagonist activity, giving compounds that are full antagonists of both ER subtypes, similar to the drug fulvestrant (ICI 182 780), by a novel mechanism.

Structural analysis of the origin of the enhanced antagonist character of OBH-sulfonamides

E₂ supports transcriptional activation of ERs by stabilizing helix 12 in a conformation forming one side of a binding groove for transcriptional coactivator proteins (Fig. 3A), while its high affinity derives both from its relatively flat shape and hydrogen bonding on both ends of the ligand: helix-3/E353, R394 and helix 11/H524 in ER α .¹¹ SERMs and full ER antagonists have typically been developed by starting with an agonist, and then adding a bulky side group that physically relocates helix-12 out of this position by direct displacement (Fig. 3B), thus blocking the recruitment of transcriptional coactivator proteins, such as histone acetyl transferases (HATs).¹² The residual agonist activity seen with tamoxifen is due to an allosteric activation of a second coactivator-binding site on the ER N-terminus, termed AF-1, via an unknown mechanism.

Previously, we reported crystal structures of much smaller oxabicyclic core derivatives, such as the diethyl ester ODE (pdb.2QH6)¹⁸, as well as, more recently, that for OBHS itself.¹⁷ These structures revealed a novel mechanism of antagonism without the use of the bulky side chain traditionally found in SERMs: ligand-induced repositioning of helix-11 indirectly modulates helix-12 positioning and receptor activity, a process we had previously termed “passive antagonism”,¹⁹ but might more appropriately be called “indirect antagonism”.

The diaryl oxabicyclic core ligands achieve this indirect antagonism in a unique way: one of the phenols mimics the role of the A-ring phenol of E₂¹¹ or one of the phenols in diethylstilbestrol,¹² engaging in strong hydrogen bonds with E353 and R394 and a structured water in ER α . The second phenol of OBHS makes a distinct hydrogen bonding interaction involving helix-3/

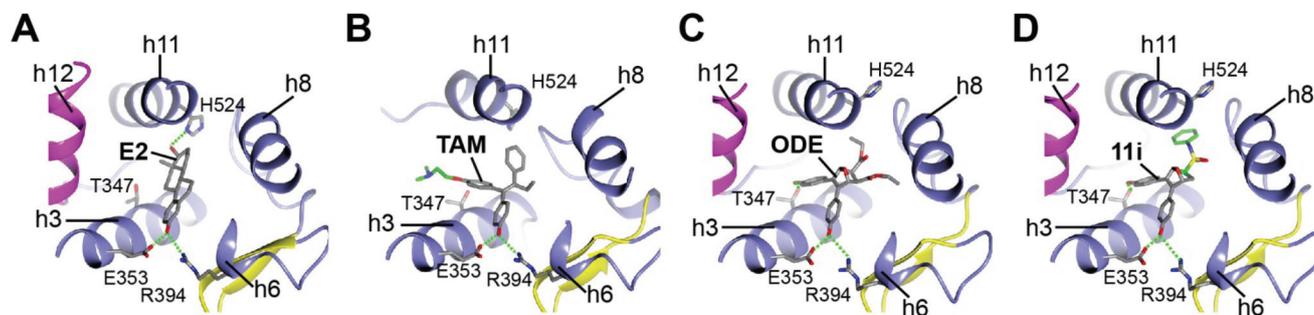


Fig. 3 Modeling of ER α bound to oxabicyclic heptane sulfonamide (OBH-sulfonamide). ER α bound to **A.** E₂ (pdb.1ERE); **B.** 4-Hydroxytamoxifen; **C.** ODE (pdb.2QH6),¹⁸ which has the same oxabicyclic core as the sulfonamides. **D.** Model of the sulfonamide **11i** based on the ODE structure. The model was constructed by substitution of the ODE ester functions with the *N*-ethyl phenyl sulfonamide; the strong hydrogen bonding patterns of the phenols maintained the position of the ligand core. Accommodation of the aryl sulfonamide would require a shift in helix 11, predicted to disrupt the helix 11–helix 12 interface and block agonist activity by an indirect antagonism mechanism.¹⁹

T347. This latter interaction is energetically favourable, because deletion of the second phenolic OH or its etherification greatly reduces ER binding affinity.¹⁶ While this second phenol points in the E₂ 11β direction, as do the third aryl groups in the SERMs, hydroxytamoxifen and raloxifene, it is not long enough to interact directly with helix-12 and displace it, as do these SERMs. Consequently, OBHS does not have a hydrogen bonding interaction to constrain helix 11 by interaction with H524 (Fig. 3C). The crystal structures show that the large, non-polar phenyl sulfonate group in OBHS,¹⁷ and to a lesser extent the ethyl carboxylate group in the smaller analogues,¹⁸ make strong steric clashes with helix-11, displacing H524 and repositioning helix-11 in a manner that indirectly modulates helix-12 positioning and reduces receptor agonist activity, which is the essence of indirect antagonism.¹⁹

Molecular modelling indicates that the aryl sulfonamide substituents could be accommodated in a manner similar to that of aryl sulfonate groups in the OBHS analogues (Fig. 3D). The additional bulk of the tertiary sulphonamides, however, would be expected to accentuate this clash with helix-11, which would result in a greater reduction in agonistic efficacy than that shown by the sulfonates. This is illustrated with the *N*-ethyl sulfonamide **11i**, which is a low efficacy compound that is an analogue of both ODE and OBHS.

Thus, it appears that the enhanced indirect antagonism of the OBH-sulfonamides compared to the OBH-sulfonates, can account for their ability to achieve full antagonism, without the need for a bulky side chain that disrupts helix-12 directly. While it is possible that the scaffold flips to allow the sulfonamide to exit towards helix 12, this would require accommodation of the L-shaped configuration of the two phenols within the pocket, which up to now has never been seen with an ER ligand, as well as the loss of the key hydrogen bonding interaction with T347. Thus, the oxabicyclic heptane sulfonamides may represent a novel binding epitope to generate full antagonists on both ERα and ERβ.

Conclusion

Oxabicyclic heptane sulfonamides appear to represent a novel binding epitope that can generate full ERα and ERβ antagonists with intrinsic activity as low as that of fulvestrant (and possibly also SERD activity). This is an issue that we are exploring further.

Experimental section

Materials and methods

Unless otherwise noted, reagents and materials were obtained from commercial suppliers and were used without further purification. Tetrahydrofuran and toluene were dried over Na and distilled prior to use. Dichloromethane was dried over CaH₂ and distilled prior to use. Glassware was oven-dried, assembled while hot, and cooled under an inert atmosphere. Unless otherwise noted, all reactions were conducted in an inert atmosphere. Reaction progress was monitored using analytical thin-layer chromatography (TLC). Visualization was achieved by UV light (254 nm). ¹H NMR and ¹³C NMR spectra were obtained on

Bruker Biospin AV400 (400 MHz) instrument. The chemical shifts are reported in ppm and are referenced to either tetramethylsilane or the solvent. Mass spectra were recorded under electron impact conditions at 70 eV. Melting points were obtained on SGW X-4 melting point apparatus and are uncorrected. Flash chromatography was performed with silica gel (0.040–0.063 mm) packing.

General procedure for the synthesis of dienophiles 7–9

The synthesis of dienophiles **7**. 2-Chloroethanesulfonyl chloride (1.2 equiv.) was added slowly to a solution of aniline (0.5 equiv.) in acetone at 0 °C. The mixture was stirred overnight at 0–10 °C, then evaporated *in vacuo*. The residue was dissolved into a mixture of CH₂Cl₂ (25 mL) and water (25 mL), then extracted with CH₂Cl₂ (3 × 30 mL). The combined organic layer was washed with saturated NaCl, dried over Na₂SO₄, filtered and evaporated *in vacuo*. The product **7** was purified by column chromatography.

The synthesis of dienophiles **8** and **9**

A solution of *N*-substituted anilines (**4** or **6**, 1.0 equiv.) in methylene chloride (10 mL) and water (10 mL) was stirred at 0 °C, and 25% NaOH (2 mL for 1 mmol 2-chloroethanesulfonyl chloride) and 2-chloroethanesulfonyl chloride (1.2 equiv.) were added simultaneously and slowly under 0 °C, keeping the pH between 8.5 and 9.5. After 12 h, the mixture was extracted with CH₂Cl₂ (3 × 30 mL). The combined organic phase was washed with saturated NaCl and dried with Na₂SO₄, filtered, evaporated *in vacuo*, and purified by flash chromatography on silica gel to give the dienophiles **8** and **9**.

General procedure for the Diels–Alder reaction of dienophiles 7–9 and 10

3,4-Diphenol furans **10** (1.0 equiv.) and dienophiles **7–9** (1.2 equiv.) were placed in a round flask, and the mixture was stirred under Ar₂ atmosphere at 95 °C for 24 h. The crude product was purified by flash chromatography (EtOAc–petroleum ether = 1 : 3).

5,6-Bis(4-hydroxyphenyl)-*N*-phenyl-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11a). Yellow solid (69% yield; mp 94–96 °C); ¹H NMR (400 MHz, acetone-d₆) δ 8.82 (s, 1H), 8.68 (s, 1H), 7.38 (d, *J* = 8.4 Hz, 2H), 7.34–7.27 (m, 2H), 7.20–7.08 (m, 3H), 7.02 (d, *J* = 8.5 Hz, 2H), 6.77 (d, *J* = 8.7 Hz, 4H), 6.73 (d, *J* = 8.7 Hz, 2H), 5.48 (d, *J* = 1.0 Hz, 1H), 5.31 (dd, *J* = 4.4, 1.1 Hz, 1H), 3.49 (dd, *J* = 8.4, 4.5 Hz, 1H), 2.30 (dt, *J* = 12.0, 4.5 Hz, 1H), 1.98 (dd, *J* = 12.4, 8.0 Hz, 1H); ¹³C NMR (101 MHz, acetone-d₆) δ 158.26, 158.17, 139.34, 138.27, 130.18, 129.66, 129.46, 125.26, 125.08, 124.53, 121.56, 121.49, 116.43, 116.37, 85.03, 83.62, 68.10, 26.18; HRMS (ESI) calcd for C₂₄H₂₁NO₅SNa, 458.1031 (M + Na⁺); found, 458.10327.

5,6-Bis(4-hydroxyphenyl)-*N*-(4-methoxyphenyl)-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11b). Yellow solid (73% yield; mp 96–98 °C); ¹H NMR (400 MHz, acetone-d₆) δ 8.70 (s, 1H), 8.65 (s, 1H), 7.25 (d, *J* = 8.9 Hz, 2H), 7.16 (d, *J* = 8.6 Hz, 2H),

7.07 (d, $J = 8.6$ Hz, 2H), 6.83 (d, $J = 9.0$ Hz, 2H), 6.77 (dd, $J = 8.5, 1.7$ Hz, 4H), 5.48 (s, 1H), 5.30 (d, $J = 4.3$ Hz, 1H), 3.75 (s, 3H), 3.39 (dd, $J = 8.4, 4.5$ Hz, 1H), 2.28 (dt, $J = 12.1, 4.4$ Hz, 1H), 1.69 (dd, $J = 11.5, 4.8$ Hz, 1H); ^{13}C NMR (101 MHz, acetone- d_6) δ 158.24, 158.21, 141.84, 138.34, 131.58, 129.77, 129.50, 125.21, 125.15, 124.87, 124.78, 116.49, 116.39, 115.22, 85.10, 83.59, 61.80, 55.74, 20.93; HRMS (ESI) calcd for $\text{C}_{25}\text{H}_{23}\text{NO}_6\text{SNa}$, 488.1135 (M + Na $^+$); found, 488.11383.

***N*-(2-Chlorophenyl)-5,6-bis(4-hydroxyphenyl)-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11c).** Yellow solid (65% yield; mp 93–95 °C); ^1H NMR (400 MHz, acetone- d_6) δ 8.70 (s, 1H), 8.38 (s, 1H), 7.70 (d, $J = 9.6$ Hz, 1H), 7.45 (d, $J = 9.5$ Hz, 1H), 7.36–7.29 (m, 1H), 7.19 (d, $J = 6.6$ Hz, 3H), 7.09 (d, $J = 6.7$ Hz, 2H), 6.77 (dd, $J = 8.7, 7.1$ Hz, 4H), 5.49 (s, br, 1H), 5.34 (dd, $J = 4.4, 1.1$ Hz, 1H), 3.57 (dd, $J = 8.0, 4.1$ Hz, 1H), 2.36 (dt, $J = 12.0, 4.4$ Hz, 1H), 1.98–1.96 (m, 1H); ^{13}C NMR (101 MHz, acetone- d_6) δ 158.29, 158.18, 142.12, 138.26, 135.54, 135.46, 130.71, 129.66, 129.57, 128.77, 127.26, 125.39, 125.31, 124.47, 116.45, 116.37, 85.15, 83.65, 63.80, 18.85; HRMS (ESI) calcd for $\text{C}_{24}\text{H}_{20}\text{NO}_5\text{SClNa}$, 492.0656 (M + Na $^+$); found, 492.06429.

***N*-(4-Chlorophenyl)-5,6-bis(4-hydroxyphenyl)-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11d).** Yellow solid (72% yield; mp 121–123 °C); ^1H NMR (400 MHz, acetone- d_6) δ 8.96 (s, 1H), 8.72 (s, 1H), 7.40–7.37 (m, 2H), 7.34–7.31 (m, 2H), 7.17 (d, $J = 8.7$ Hz, 2H), 7.04 (d, $J = 8.6$ Hz, 2H), 6.76 (dd, $J = 8.7, 6.0$ Hz, 4H), 5.47 (d, $J = 1.0$ Hz, 1H), 5.31 (dd, $J = 4.4, 1.1$ Hz, 1H), 3.52 (dd, $J = 8.4, 4.4$ Hz, 1H), 2.28 (dt, $J = 12.0, 4.5$ Hz, 1H), 2.02 (dd, $J = 8.7, 3.3$ Hz, 1H); ^{13}C NMR (101 MHz, acetone- d_6) δ 158.27, 158.17, 141.95, 138.18, 130.10, 130.01, 129.64, 129.51, 125.04, 124.50, 123.09, 123.02, 116.43, 116.36, 85.03, 83.62, 62.65, 18.85; HRMS calcd for $\text{C}_{24}\text{H}_{20}\text{NO}_5\text{SClNa}$, 492.0643 (M + Na $^+$); found, 492.06429.

5,6-Bis(4-hydroxyphenyl)-*N*-methyl-*N*-phenyl-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11e). Yellow solid (71% yield; mp 86–88 °C); ^1H NMR (400 MHz, acetone- d_6) δ 8.71 (s, 1H), 8.67 (s, 1H), 7.44 (d, $J = 8.0$ Hz, 2H), 7.35 (dd, $J = 8.0, 3.0$ Hz, 2H), 7.27 (d, $J = 7.3$ Hz, 1H), 7.20–7.13 (m, 4H), 6.79 (dd, $J = 10.9, 8.6$ Hz, 4H), 5.45 (s, 1H), 5.29 (d, $J = 4.3$ Hz, 1H), 3.57 (dd, $J = 8.1, 4.0$ Hz, 1H), 3.39 (s, 3H), 2.43–2.27 (m, 1H), 1.97 (dd, $J = 8.2, 3.7$ Hz, 1H); ^{13}C NMR (101 MHz, acetone- d_6) δ 158.31, 158.25, 141.88, 138.27, 130.40, 129.80, 129.77, 129.49, 127.66, 127.37, 125.17, 124.59, 116.50, 116.34, 85.24, 83.61, 61.71, 39.18, 18.84; HRMS (ESI) calcd for $\text{C}_{25}\text{H}_{23}\text{NO}_5\text{SNa}$, 472.1191 (M + Na $^+$); found, 472.11892.

5,6-Bis(4-hydroxyphenyl)-*N*-(4-methoxyphenyl)-*N*-methyl-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11f). Yellow solid (73% yield; 89–91 °C); ^1H NMR (400 MHz, acetone- d_6) δ 8.70 (s, 1H), 8.64 (s, 1H), 7.32 (d, $J = 8.9$ Hz, 2H), 7.17 (dd, $J = 8.6, 3.4$ Hz, 4H), 6.86 (d, $J = 9.0$ Hz, 2H), 6.82 (d, $J = 8.6$ Hz, 2H), 6.78 (d, $J = 8.6$ Hz, 2H), 5.45 (s, br, 1H), 5.30 (d, $J = 3.9$ Hz, 1H), 3.78 (s, 3H), 3.53 (dd, $J = 8.3, 4.4$ Hz, 1H), 3.33 (s, 3H), 2.16 (dt, $J = 11.8, 4.4$ Hz, 1H), 2.01 (dd, $J = 8.8, 3.1$ Hz, 1H); ^{13}C NMR (101 MHz, acetone- d_6) δ 159.49, 158.31, 158.22, 141.85, 138.30, 135.61, 129.89, 129.48, 129.20, 125.22,

124.64, 116.58, 116.41, 114.94, 85.31, 83.63, 60.67, 55.79, 32.66, 20.93; HRMS (ESI) calcd for $\text{C}_{26}\text{H}_{25}\text{NO}_6\text{SNa}$, 502.1282 (M + Na $^+$); found, 502.12948.

***N*-(2-Chlorophenyl)-5,6-bis(4-hydroxyphenyl)-*N*-methyl-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11g).** Yellow solid (77% yield; mp 97–99 °C); ^1H NMR (400 MHz, acetone- d_6) δ 8.70 (s, 1H), 8.67 (s, 1H), 7.52 (ddd, $J = 7.2, 6.3, 4.3$ Hz, 2H), 7.36 (ddd, $J = 6.4, 3.4, 2.2$ Hz, 2H), 7.22 (dd, $J = 8.7, 3.3$ Hz, 4H), 6.81 (dd, $J = 8.6, 7.2$ Hz, 4H), 5.56 (s, br, 1H), 5.36 (dd, $J = 4.4, 1.1$ Hz, 1H), 3.68 (dd, $J = 8.3, 4.4$ Hz, 1H), 3.29 (s, 3H), 2.36 (dt, $J = 11.1, 4.1$ Hz, 1H), 2.20 (dd, $J = 11.9, 8.3$ Hz, 1H); ^{13}C NMR (101 MHz, acetone- d_6) δ 158.29, 158.21, 140.11, 138.46, 134.96, 132.74, 131.33, 130.64, 130.39, 129.91, 129.49, 128.88, 125.26, 124.61, 116.52, 116.36, 85.42, 83.68, 60.62, 39.33, 18.85; HRMS (ESI) calcd for $\text{C}_{25}\text{H}_{22}\text{NO}_5\text{SClNa}$, 506.0787 (M + Na $^+$); found, 506.07994.

***N*-(4-Chlorophenyl)-5,6-bis(4-hydroxyphenyl)-*N*-methyl-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11h).** Yellow solid (71% yield; mp 86–88 °C); ^1H NMR (400 MHz, acetone- d_6) δ = 8.70 (s, 1H), 8.65 (s, 1H), 7.46 (d, $J = 8.8, 2\text{H}$), 7.36 (d, $J = 8.7, 2\text{H}$), 7.17 (t, $J = 8.6, 4\text{H}$), 6.80 (dd, $J = 12.7, 8.6, 4\text{H}$), 5.46 (s, 1H), 5.30 (d, $J = 4.2, 1\text{H}$), 3.60 (dd, $J = 8.3, 4.5, 1\text{H}$), 3.39 (s, 3H), 2.42–2.31 (m, 1H), 2.27 (dd, $J = 8.6, 4.7, 1\text{H}$); ^{13}C NMR (101 MHz, acetone- d_6) δ 158.32, 158.26, 142.00, 138.14, 132.70, 130.43, 130.31, 129.81, 129.52, 128.83, 125.10, 124.54, 116.55, 116.39, 85.24, 83.67, 60.56, 39.15, 20.94; HRMS (ESI) calcd for $\text{C}_{25}\text{H}_{22}\text{NO}_5\text{SClNa}$, 506.0793 (M + Na $^+$); found, 506.07994.

***N*-Ethyl-5,6-bis(4-hydroxyphenyl)-*N*-phenyl-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11i).** Yellow solid (75% yield; mp 102–104 °C); ^1H NMR (400 MHz, acetone- d_6) δ 8.72 (s, 1H), 8.67 (s, 1H), 7.37–7.33 (m, 4H), 7.33–7.30 (m, 1H), 7.19 (dd, $J = 8.6, 1.7$ Hz, 4H), 6.80 (dd, $J = 14.2, 8.6$ Hz, 4H), 5.47 (d, $J = 1.1$ Hz, 1H), 5.31 (dd, $J = 4.4, 1.1$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, 2H), 3.51 (dd, $J = 8.3, 4.5$ Hz, 1H), 2.17 (dt, $J = 11.9, 4.5$ Hz, 1H), 2.05–2.00 (m, 1H), 1.03 (t, $J = 7.1$ Hz, 3H); ^{13}C NMR (101 MHz, acetone- d_6) δ 158.35, 158.24, 140.32, 138.33, 130.40, 130.10, 129.91, 129.89, 129.44, 128.40, 125.21, 124.57, 116.56, 116.37, 85.30, 83.63, 62.59, 47.10, 18.86, 14.94; HRMS (ESI) calcd for $\text{C}_{26}\text{H}_{25}\text{NO}_5\text{SNa}$, 486.1334 (M + Na $^+$); found, 486.13457.

***N*-Ethyl-5,6-bis(4-hydroxyphenyl)-*N*-(4-methoxyphenyl)-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11j).** Yellow solid (71% yield; mp 92–93 °C); ^1H NMR (400 MHz, acetone- d_6) δ 8.85 (s, 1H), 8.79 (s, 1H), 7.25 (d, $J = 7.5$ Hz, 2H), 7.19 (d, $J = 8.6$ Hz, 4H), 6.89–6.76 (m, 6H), 5.46 (s, 1H), 5.31 (d, $J = 4.1$ Hz, 1H), 3.79 (s, 3H), 3.76 (q, $J = 7.1$ Hz, 1H), 3.47 (dd, $J = 7.7, 4.1$ Hz, 1H), 2.20 (dt, $J = 8.9, 3.8$ Hz, 1H), 1.72 (dd, $J = 13.8, 7.0$ Hz, 1H), 1.03 (t, $J = 7.1$ Hz, 3H); ^{13}C NMR (101 MHz, acetone- d_6) δ 159.90, 158.22, 141.83, 138.43, 132.65, 131.99, 131.51, 129.97, 129.39, 125.28, 124.64, 116.56, 116.36, 114.95, 85.34, 83.59, 62.29, 55.74, 47.29, 31.29, 14.90; HRMS (ESI) calcd for $\text{C}_{27}\text{H}_{27}\text{NO}_6\text{SNa}$, 516.1450 (M + Na $^+$); found, 516.14513.

***N*-(2-Chlorophenyl)-*N*-ethyl-5,6-bis(4-hydroxyphenyl)-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11k).** Yellow solid

(68% yield; mp 92–94 °C); ^1H NMR (400 MHz, acetone- d_6) δ 8.78 (s, 1H), 8.75 (s, 1H), 7.51 (dd, $J = 7.5, 1.7$ Hz, 2H), 7.38 (dd, $J = 9.1, 4.1$ Hz, 2H), 7.21 (d, $J = 6.4$ Hz, 4H), 6.80 (t, $J = 8.3$ Hz, 4H), 5.53 (s, 1H), 5.34 (d, $J = 4.1$ Hz, 1H), 3.77 (dd, $J = 22.3, 14.8$ Hz, 2H), 3.62 (dd, $J = 8.1, 4.4$ Hz, 1H), 2.30–2.24 (m, 1H), 2.20 (d, $J = 9.2$ Hz, 1H), 1.06 (t, $J = 7.1$ Hz, 3H); ^{13}C NMR (101 MHz, acetone- d_6) δ 158.39, 158.29, 142.09, 139.71, 138.50, 137.38, 134.30, 131.36, 130.69, 129.93, 129.42, 128.56, 125.26, 124.59, 116.55, 116.38, 85.50, 83.65, 64.10, 47.05, 32.66, 14.580; HRMS (ESI) calcd for $\text{C}_{26}\text{H}_{24}\text{NO}_5\text{SClNa}$, 520.0960 ($\text{M} + \text{Na}^+$); found, 520.09559.

***N*-(4-Chlorophenyl)-*N*-ethyl-5,6-bis(4-hydroxyphenyl)-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11i).** Yellow solid (69% yield; mp 107–109 °C); ^1H NMR (400 MHz, acetone- d_6) δ 8.70 (s, 1H), 8.65 (s, 1H), 7.38 (d, $J = 5.5$ Hz, 4H), 7.19 (dd, $J = 8.6, 1.7$ Hz, 4H), 6.82 (d, $J = 8.6$ Hz, 2H), 6.78 (d, $J = 8.7$ Hz, 2H), 5.48 (d, $J = 1.1$ Hz, 1H), 5.32 (dd, $J = 4.3, 1.0$ Hz, 1H), 3.84 (q, $J = 7.1$ Hz, 2H), 3.53 (dd, $J = 8.3, 4.5$ Hz, 1H), 2.27 (dd, $J = 13.0, 7.1$ Hz, 1H), 2.15 (dt, $J = 11.9, 4.4$ Hz, 1H), 1.04 (t, $J = 7.1$ Hz, 3H); ^{13}C NMR (101 MHz, acetone- d_6) δ 158.34, 158.24, 141.87, 139.21, 138.24, 133.54, 131.59, 129.92, 129.90, 129.47, 125.17, 124.57, 116.57, 116.39, 85.29, 83.65, 62.90, 47.06, 20.94, 14.85; HRMS (ESI) calcd for $\text{C}_{26}\text{H}_{24}\text{NO}_5\text{SClNa}$, 520.0953 ($\text{M} + \text{Na}^+$); found, 520.09559.

5,6-Bis(4-hydroxyphenyl)-*N*-phenyl-*N*-(2,2,2-trifluoroethyl)-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11m). Yellow solid (59% yield; mp 91–93 °C); ^1H NMR (400 MHz, acetone- d_6) δ 8.76 (s, 1H), 8.70 (s, 1H), 7.48–7.45 (m, 2H), 7.36 (dd, $J = 5.2, 2.0$ Hz, 3H), 7.19 (d, $J = 3.5$ Hz, 2H), 7.17 (d, $J = 3.5$ Hz, 2H), 6.82 (d, $J = 8.6$ Hz, 2H), 6.78 (d, $J = 8.6$ Hz, 2H), 5.53 (d, $J = 1.0$ Hz, 1H), 5.33 (dd, $J = 4.3, 1.0$ Hz, 1H), 4.59 (q, $J = 8.7$ Hz, 2H), 3.58 (dd, $J = 8.3, 4.5$ Hz, 1H), 2.50–2.40 (m, 1H), 2.14 (dd, $J = 8.2, 3.8$ Hz, 1H); ^{13}C NMR (101 MHz, acetone- d_6) δ 158.47, 158.32, 142.10, 140.54, 138.07, 130.19, 130.04, 129.85, 129.35, 129.18, 125.03, 124.36, 116.56, 116.36, 85.20, 83.59, 62.99, 60.59, 30.62; HRMS (ESI) calcd for $\text{C}_{26}\text{H}_{22}\text{NO}_5\text{-SF}_3\text{Na}$, 540.1076 ($\text{M} + \text{Na}^+$); found, 540.10630.

5,6-Bis(4-hydroxyphenyl)-*N*-(4-methoxyphenyl)-*N*-(2,2,2-trifluoroethyl)-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11n). Yellow solid (80% yield; mp 117–119 °C); ^1H NMR (400 MHz, acetone- d_6) δ 8.80 (s, 1H), 8.73 (s, 1H), 7.35–7.32 (m, 2H), 7.19 (dd, $J = 8.6, 6.6$ Hz, 4H), 6.85 (t, $J = 9.1$ Hz, 4H), 6.78 (d, $J = 8.5$ Hz, 2H), 5.52 (d, $J = 1.1$ Hz, 1H), 5.33 (dd, $J = 4.3, 0.9$ Hz, 1H), 4.51 (q, $J = 8.7$ Hz, 2H), 3.79 (s, 3H), 3.55–3.53 (m, 1H), 2.50–2.41 (m, 1H), 2.17 (dd, $J = 8.2, 3.8$ Hz, 1H); ^{13}C NMR (101 MHz, acetone- d_6) δ 160.37, 158.45, 158.27, 142.10, 138.13, 132.78, 131.29, 130.16, 129.33, 125.11, 124.45, 116.59, 116.37, 115.21, 85.25, 83.57, 62.62, 60.59, 55.79, 31.37; HRMS (ESI) calcd for $\text{C}_{27}\text{H}_{24}\text{NO}_6\text{SF}_3\text{Na}$, 570.1173 ($\text{M} + \text{Na}^+$); found, 570.11687.

***N*-(4-Chlorophenyl)-5,6-bis(4-hydroxyphenyl)-*N*-(2,2,2-trifluoroethyl)-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11o).** Yellow solid (69% yield; mp 93–95 °C); ^1H NMR (400 MHz, acetone- d_6) δ 8.80 (s, 1H), 8.73 (s, 1H), 7.33 (d, $J = 9.0$ Hz, 2H), 7.19 (dd, $J = 8.6, 6.6$ Hz, 4H), 6.85 (t, $J = 9.1$ Hz, 4H), 6.78 (d, $J = 8.5$ Hz, 2H), 5.52 (d, $J = 1.1$ Hz, 1H), 5.33 (dd, $J = 4.3, 0.9$ Hz,

1H), 4.51 (q, $J = 8.7$ Hz, 2H), 3.56–3.53 (m, 2H), 2.50–2.41 (m, 1H), 2.17 (dd, $J = 8.2, 3.8$ Hz, 1H); ^{13}C NMR (101 MHz, acetone- d_6) δ 158.41, 158.37, 142.14, 139.40, 137.98, 134.43, 131.46, 130.24, 130.02, 129.37, 124.95, 124.35, 116.56, 116.37, 85.22, 83.61, 63.38, 60.59, 31.37; HRMS (ESI) calcd for $\text{C}_{26}\text{H}_{21}\text{NO}_5\text{SF}_3\text{ClNa}$, 574.0665 ($\text{M} + \text{Na}^+$); found, 574.06733.

***N*-Ethyl-5,6-bis(4-hydroxyphenyl)-*N*-(naphthalen-2-yl)-7-oxabicyclo[2.2.1]hept-5-ene-2-sulfonamide (11p).** Yellow solid (71% yield; mp 165–167 °C); ^1H NMR (400 MHz, acetone- d_6) δ 8.80 (d, $J = 8.2$ Hz, 1H), 8.32 (d, $J = 8.2$ Hz, 1H), 8.19 (d, $J = 8.5$ Hz, 1H), 7.62 (t, $J = 7.2$ Hz, 1H), 7.53 (t, $J = 7.1$ Hz, 1H), 7.15 (d, $J = 8.7$ Hz, 2H), 6.82–6.68 (m, 6H), 6.58 (d, $J = 8.7$ Hz, 2H), 5.32 (d, $J = 5.2$ Hz, 1H), 5.27 (s, br, 1H), 3.66 (dd, $J = 8.3, 4.6$ Hz, 1H), 3.47 (q, $J = 7.2$ Hz, 2H), 2.41 (dt, $J = 11.8, 4.5$ Hz, 1H), 1.89 (dd, $J = 11.8, 8.4$ Hz, 1H), 1.39 (t, $J = 7.2$ Hz, 3H); ^{13}C NMR (101 MHz, acetone- d_6) δ 158.28, 157.92, 150.59, 141.99, 138.45, 134.76, 131.66, 129.88, 129.02, 128.84, 125.95, 125.53, 125.11, 124.59, 123.95, 122.89, 120.67, 116.41, 116.31, 116.22, 101.72, 85.01, 83.83, 65.83, 38.91, 14.28; HRMS (ESI) calcd for $\text{C}_{30}\text{H}_{27}\text{NO}_5\text{SNa}$, 536.1513 ($\text{M} + \text{Na}^+$); found, 536.15022.

Estrogen receptor binding assays

Relative binding affinities were determined by a competitive radiometric binding assay, as previously described,²⁹ using 2 nM [^3H]estradiol as tracer (Perkin Elmer, Waltham, MA) and purified full-length human ER α and ER β (PanVera/Invitrogen, Carlsbad, CA). Incubations were for 18–24 h at 0 °C. Then the receptor–ligand complexes were absorbed onto hydroxyapatite (BioRad, Hercules, CA), and unbound ligand was washed away. The binding affinities are expressed as relative binding affinity (RBA) values, with the RBA of estradiol set to 100. The values given are the average \pm range or SD of two or more independent determinations. Estradiol binds to ER α with a K_d of 0.2 nM and to ER β with a K_d of 0.5 nM.

Luciferase reporter gene assays

HepG2 cells cultured in Dulbecco's minimum essential medium (DMEM) (Cellgro by Mediatech, Inc., Manassas, VA) supplemented with 10% fetal bovine serum (FBS) (Hyclone by Thermo Scientific, South Logan, UT), and 1% non-essential amino acids (Cellgro), Penicillin–Streptomycin–Neomycin antibiotic mixture and Glutamax (Gibco by Invitrogen Corp., Carlsbad, CA), were maintained at 37 °C and 5% CO_2 . Cells were transfected with 10 μg of 3X ERE-luciferase reporter plus 1.6 μg of ER α or ER β expression vector per 10 cm dish using FugeneHD reagent (Roche Applied Sciences, Indianapolis, IN). The next day, the cells were transferred to phenol red-free growth media supplemented with 10% charcoal-dextran sulfate-stripped FBS at a density of 20 000 cells per well, incubated in 384-well plates overnight at 37 °C and 5% CO_2 , and stimulated with various concentrations of compounds in triplicate. Luciferase activity was measured after 24 h using BriteLite reagent (Perkin-Elmer Inc., Shelton, CT) according to manufacturer's protocol.

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