



Synthesis of poly-substituted tetrahydropyridines from Baylis–Hillman adducts modified with *N*-allylamino group via radical cyclization

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ARTICLE INFO

Article history:

Received 19 February 2009

Accepted 27 February 2009

Available online 6 March 2009

Keywords:

Radical cyclization

Tetrahydropyridines

Baylis–Hillman adducts

ABSTRACT

An expedient method was developed for the synthesis of 1,4,5,6-tetrahydropyridines by radical cyclization protocol involving consecutive 1,5-hydrogen transfer and double bond isomerization process starting from Baylis–Hillman adducts.

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Radical cyclizations have been used for the synthesis of various cyclic compounds. In some instances initial radical species underwent 1,*n*-H transfer to form another radical before cyclization reaction.^{1,2} Among the 1,*n*-H transfers, 1,5- and 1,6-H transfers are the most common.^{1,2} Very recently we observed an interesting radical cyclization procedure for the synthesis of tricyclic lactam derivatives involving 1,5-H transfer and concomitant isomerization.^{1a}

Suitably substituted dihydro- and tetrahydropyridine derivatives have been regarded as important synthetic intermediates for the synthesis of various important compounds.³ Especially, 1,4,5,6-tetrahydropyridine derivative **A** has been used for the synthesis of antidepressant drug paroxetine (**B**)⁴ and many paroxetine-like PSSRIs (phenylpiperidine selective Serotonin reuptake inhibitors) including femoxetine (**B'**).^{4,5} In addition, many 1,2,5,6-tetrahydropyridine derivatives **C** have been reported as renin inhibitors (Fig. 1).⁵ During our recent studies on the chemical transformations of Baylis–Hillman adducts,^{6,7} we imagined that we could synthesize tetrahydropyridine skeleton^{4,5} via the radical cyclization of **4a–e** as in Scheme 1. *N*-Tosyl-*N*-allyl derivatives **4a–e** were prepared in good to moderate yields from the acetates of Baylis–Hillman adducts **1a** and **1b** as summarized in Scheme 2 in two steps.^{7,8}

With the substrates **4a–e** we examined the radical cyclization reaction under the conditions of *n*-Bu₃SnH (1.2 equiv)/AIBN in refluxing benzene.⁷ As expected, we obtained 1,4,5,6-tetrahydropyridines **5a–e** in good to moderate yields (56–82%) in short time (Table 1).⁹ Compounds having two stereogenic centers including **5a**, **5c**, and **5d** were isolated as their *syn/anti* diastereomeric mix-

ture in a ratio of 3:2–4:1. However, we could not separate each isomer in their pure form. The mechanism for the formation of **5** could be regarded as in Scheme 1: (i) 1,5-hydrogen transfer from the initial radical (**I**) to form (**II**),^{1a} isomerization to more stable benzylic radical (**III**), the following cyclization in a 6-*exo-trig* mode to (**IV**), and final hydrogen radical abstraction to **5**.

However, the reaction of *N*-benzyl derivative **4f** showed low yield of product **5f** (38%) under the same conditions (1.2 equiv of *n*-Bu₃SnH) due to the formation of many intractable side products. Fortunately, we obtained **5f** in an improved yield (76%) when we used *n*-Bu₃SnH in excess amounts (2.5 equiv) presumably due to rapid hydrogen abstraction of the corresponding radical intermediate (**IV**) from *n*-Bu₃SnH (Scheme 3). Similarly, the reaction of *N*-phenyl derivative **4g** showed similar pattern. When we used 1.2–1.5 equiv of *n*-Bu₃SnH, pyrrolidine derivative **6** was formed in appreciable amounts with low yield of **5g**. The desired compound **5g** was isolated in moderate yield (50%) with 2.5 equiv of *n*-Bu₃SnH, together with pyrrolidine derivative **6** in 21% yield as a *syn/anti* (1:1) mixture (Scheme 4). The formation of **6** could be explained as sequential hydrostannylation at the allyl group¹⁰ and following radical cyclization in a 5-*exo-trig* mode. The reduction of bromophenyl moiety might occur independently with excess Bu₃SnH. Compound **6** was also prepared from **4g'** under the same conditions (2.5 equiv of *n*-Bu₃SnH) in good yield (77%). We obtained compounds **5h** (31%) and **7** (24%) from the reaction of *N*-methallyl derivative **4h**, similarly. In the reaction, two minor compounds **8** (12%) and **9** (28%) were isolated together as side products and we did not confirm the geometry of double bond Scheme 5.

In summary, we disclosed an efficient synthetic way for 1,4,5,6-tetrahydropyridines by radical cyclization protocol involving

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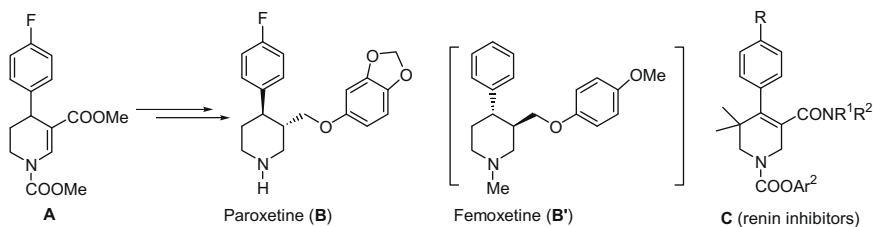
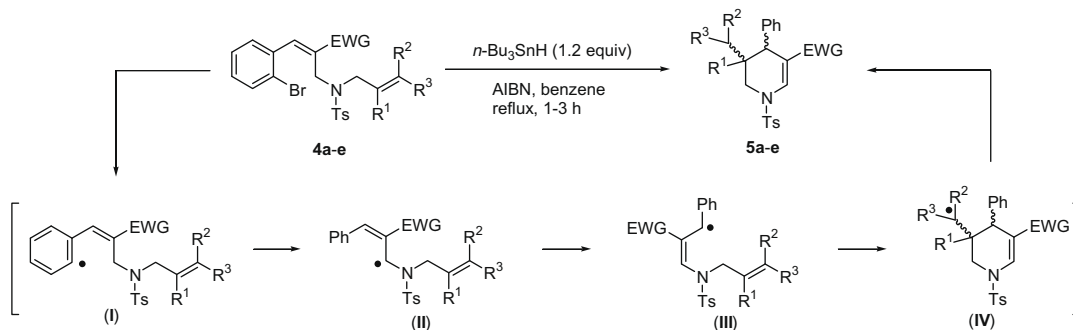


Figure 1.



Scheme 1.

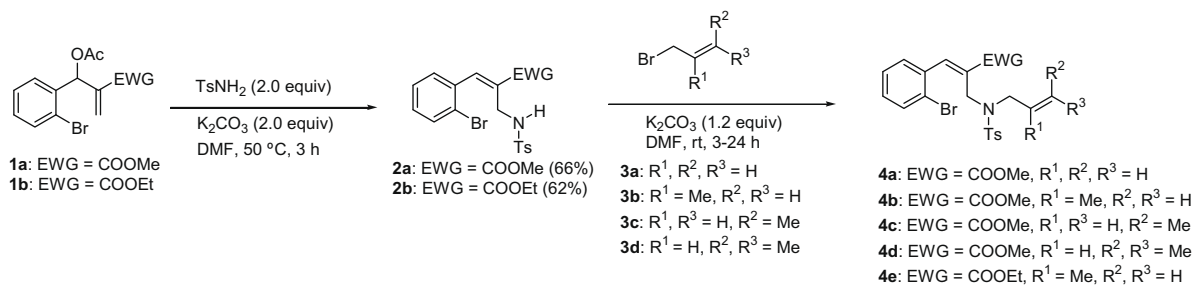
Table 1
Synthesis of tetrahydropyridine derivatives

Entry	Conditions ^a	Substrate 4 (%)	Conditions ^b (h)	Product 5 (%)
1	2a + 3a , 6 h	4a (80)	2	5a (72, <i>syn/anti</i> = 2:3) ^c
2	2a + 3b , 4 h	4b (93)	1	5b (70)
3	2a + 3c , 3 h	4c (80)	2	5c (82, <i>syn/anti</i> = 2:3) ^c
4	2a + 3d , 24 h	4d (87)	2	5d (80, <i>syn/anti</i> = 1:4) ^c
5	2b + 3b , 4 h	4e (92)	3	5e (56)

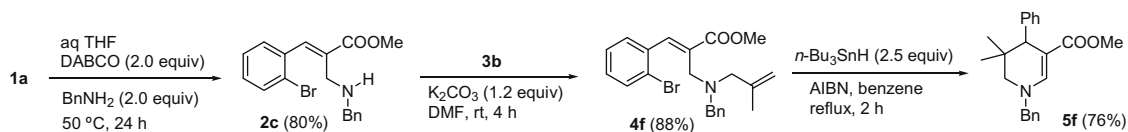
^a Conditions: Compound **2** (1.0 mmol), compound **3** (1.5 mmol), K_2CO_3 (1.2 equiv), DMF, rt.

^b Conditions: Substrate **4** (0.5 mmol), $n\text{-Bu}_3\text{SnH}$ (0.6 mmol), AIBN (cat) benzene, reflux.

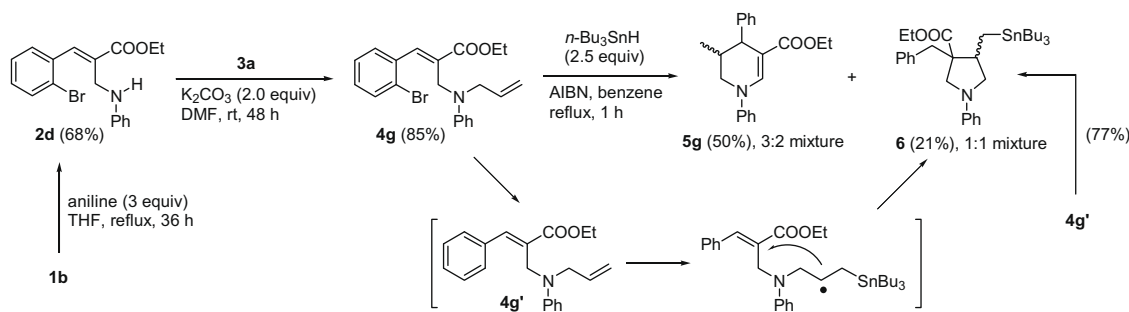
^c The ratio of *syn/anti* was determined in ^1H NMR and is arbitrary.



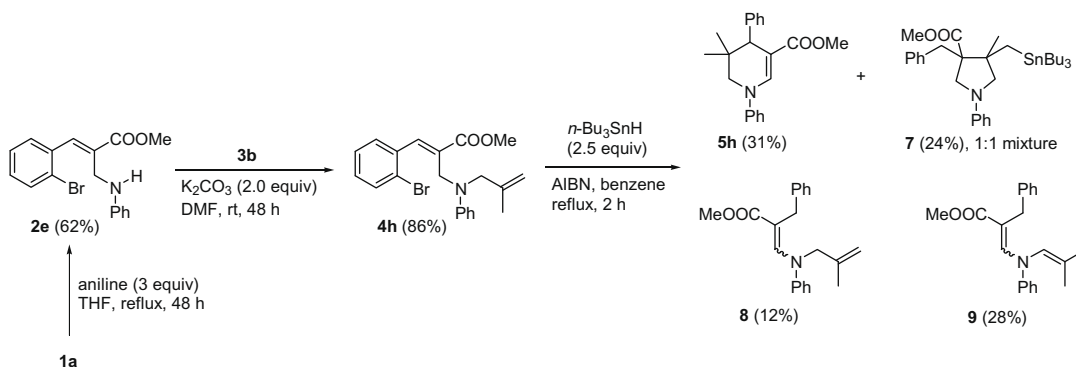
Scheme 2.



Scheme 3.



Scheme 4.



Scheme 5.

consecutive 1,5-hydrogen transfer and double bond isomerization process. Applications of this methodology are currently underway for the synthesis of paroxetine derivatives having 5-alkyls.

Acknowledgments

This work was supported by the Korea Research Foundation Grant funded by the Korean Government (MOEHRD, KRF-2008-313-C00487). Spectroscopic data were obtained from the Korea Basic Science Institute, Gwangju branch.

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solid, mp 110–112 °C; IR (KBr) 1709, 1633, 1168, 1096 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 0.59 (s, 3H), 0.88 (s, 3H), 2.49 (s, 3H), 2.80 (d, $J = 12.0$ Hz, 1H), 3.18 (d, $J = 12.0$ Hz, 1H), 3.37 (s, 1H), 3.61 (s, 3H), 6.70 (d, $J = 5.7$ Hz, 2H), 7.03–7.15 (m, 3H), 7.40 (d, $J = 8.1$ Hz, 2H), 7.79 (d, $J = 8.1$ Hz, 2H), 8.16 (s, 1H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 21.64, 25.25, 27.03, 31.96, 48.00, 49.38, 51.43, 110.10, 126.62, 127.33, 127.76, 128.66, 130.09, 134.07, 134.44, 141.23, 144.69, 167.04; ESIMS m/z 400 ($\text{M}^+ + 1$). Anal. Calcd for $\text{C}_{22}\text{H}_{25}\text{NO}_4\text{S}$: C, 66.14; H, 6.31; N, 3.51. Found: C, 66.37; H, 6.13; N, 3.45. Compound **5f**: 76%; white solid, mp 123–124 °C; IR (KBr) 1682, 1620, 1167, 1144 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 0.54 (s, 3H), 0.99 (s, 3H), 2.36 (d, $J = 12.3$ Hz, 1H), 2.85 (d, $J = 12.3$ Hz, 1H), 3.35 (s, 1H), 3.56 (s, 3H), 4.37 (d, $J = 15.3$ Hz, 1H), 4.44 (d, $J = 15.3$ Hz, 1H), 7.02 (d, $J = 6.6$ Hz, 2H), 7.10–7.41 (m, 8H), 7.77 (s, 1H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 25.51, 28.07, 31.52, 47.85, 50.51, 52.41, 60.11, 97.51, 125.91, 127.46, 127.90, 127.98, 128.71, 128.78, 136.59, 144.32, 144.55, 168.89; ESIMS m/z 336 ($\text{M}^+ + 1$). Anal. Calcd for $\text{C}_{22}\text{H}_{25}\text{NO}_2$: C, 78.77; H, 7.51; N, 4.18. Found: C, 78.89; H, 7.76;

N, 4.05. Compound **6** (separated pure isomer, but *syn/anti* was not determined): 21%; colorless oil; IR (film) 1726, 1598, 1507, 1374 cm^{-1} ; ^1H NMR (CDCl_3 , 300 MHz) δ 0.85–0.92 (m, 15H), 1.02–1.07 (m, 2H), 1.20 (t, $J = 7.2$ Hz, 3H), 1.24–1.55 (m, 12H), 2.46–2.53 (m, 1H), 2.74 (d, $J = 13.8$ Hz, 1H), 2.94 (dd, $J = 8.7$ and 6.3 Hz, 1H), 3.34 (d, $J = 13.8$ Hz, 1H), 3.41 (d, $J = 9.9$ Hz, 1H), 3.58 (d, $J = 9.9$ Hz, 1H), 3.56–3.62 (m, 1H), 4.09–4.20 (m, 2H), 6.50 (d, $J = 7.5$ Hz, 2H), 6.66 (t, $J = 7.5$ Hz, 1H), 7.12–7.29 (m, 7H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 9.20, 9.31, 13.69, 14.30, 27.41, 29.19, 41.24, 45.70, 51.87, 54.42, 59.34, 60.44, 111.17, 115.52, 126.66, 128.26, 129.12, 129.92, 137.67, 147.39, 173.31; ESIMS m/z 613 ($\text{M}^+ + 1$).
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