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Glucosylation of flavonoids and flavonoid glycosides by mutant dextransucrase from *Lactobacillus reuteri* TMW 1.106

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#### 1 ABSTRACT

2 Flavonoids are commonly abundant, plant-derived polyphenolic compounds which are 3 responsible for color, taste, and antioxidant properties of certain plant based food. 4 Glucosylation by glucansucrases or other glycosyltransferases/glycoside hydrolases has been 5 described to be a promising approach to modify stability, solubility, bioavailability, and taste 6 profile of flavonoids and related compounds. In this study, we modified and applied a 7 recombinant dextransucrase from Lactobacillus reuteri TMW 1.106 to glucosylate various 8 flavonoids and flavonoid glycosides. The glucoconjugates were subsequently isolated and 9 characterized by using two-dimensional NMR spectroscopy. Efficient glucosylation was 10 achieved for quercetin and its glycosides quercetin-3-O-β-glucoside and rutin. Significant 11 portions of  $\alpha$ -glucose conjugates were also obtained for epigallocatechin gallate, 12 dihydromyricetin, and cyanidine-3-*O*-β-glucoside, whereas glucosylation efficiency was low for naringin and neohesperidin dihydrochalcone. Most of the flavonoids with a catechol or 13 14 pyrogallol function at the B-ring were predominantly glucosylated at position O4'. However, 15 glycosyl substituents such as  $\beta$ -glucose, rutinose, or neohesperidose were glucosylated at varying positions. Therefore, mutant dextransucrase from L. reuteri TMW 1.106 can be applied 16 17 for versatile structural modification of flavonoids.

#### 18 KEYWORDS

glucansucrase, modification, NMR spectroscopy, acceptor reaction, structural characterization,
purification.

#### 21 ABBREVIATIONS

L. = Lactobacillus, HSQC = Heteronuclear Single Quantum Coherence, COSY = H,H Correlated Spectroscopy, H2BC = Heteronuclear Two-bond Correlation, TOCSY = Total
 Correlated Spectroscopy, HMBC = Heteronuclear Multiple Bond Correlation.

## 25 1 INTRODUCTION

26 Flavonoids are a very heterogeneous group of polyphenolic secondary plant metabolites which 27 are widely distributed among different plants. Consumption of polyphenol-rich foods was 28 associated with beneficial health effects which were discussed to be related to the antioxidant 29 activity of flavonoids [1]. In addition, some flavonoids also affect the sensory properties of plant based food. For example, anthocyanins are responsible for the color of many fruits, 30 31 whereas some flavonoids such as the flavanone naringin contribute to the bitter taste of some 32 citrus fruits [2]. Furthermore, the flavonoid derivative neohesperidin dihydrochalcone is added 33 to food products as a sweetener or flavor enhancer [3]. In nature, flavonoids are often 34 glycosylated at varying positions such as position O3 (if present) or position O7 and O5 [4]. It 35 has already been demonstrated that glycosylation significantly influences the solubility, 36 stability, bioavailability, and antioxidant properties [4-7]. Therefore, enzymatic in vitro 37 glycosylation has also been investigated as an approach to modify the described properties. 38 Besides cyclodextrin glucanotransferases,  $\alpha$ -glucosidases, and maltogenic amylases [5, 7-13], 39 glucansucrases have already been applied to glucosylate selected flavonoids [14-20]. These 40 enzymes belong to the GH70 family and normally catalyze the conversion/incorporation of 41 sucrose into polymeric  $\alpha$ -glucans, but by selecting appropriate reaction conditions, oligosaccharides or other hydroxyl group containing compounds can also be conjugated with 42  $\alpha$ -glucose [21]. However, the above mentioned studies focused on glucosylation of non-43 glycosylated flavonoids such as luteolin, quercetin, myricetin, dihydromyricetin, and 44 45 epigallocatechin gallate. In addition, only glucansucrases from Leuconostoc mesenteroides 46 have been used for flavonoid glucosylation. However, recombinant glucansucrases from L. 47 *reuteri* have been successfully applied for glucosylation of steviol glycosides or simple phenols 48 such as catechol [22-26]. Furthermore, site directed mutagenesis was applied to improve 49 glucosylation of these substrates and other non-carbohydrate molecules [23, 24, 26].

50 In previous studies, it was demonstrated that dextransucrase from Lactobacillus (L.) reuteri 51 TMW 1.106 synthesizes *O*4-branched dextrans and thus produces a rather rare dextran type [27, 28]. Furthermore, we observed a high expression level and thus a comparably high enzyme 52 53 activity after cloning of the encoding gene into a pET-based expression vector and heterologous expression of the recombinant sucrase in E. coli. Consequently, this recombinant 54 55 dextransucrase is a promising candidate for the modification of polyphenolic compounds. 56 Therefore, the aim of this study was to analyze the structures of  $\alpha$ -glucose conjugates formed 57 by the incubation of different flavonoids, flavonoid glycosides, and flavonoid derivatives with 58 dextransucrase from L. reuteri TMW 1.106.

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#### 2 **RESULTS AND DISCUSSION**

Several flavonoid substrates which showed a high structural heterogeneity of both the flavonoid 61 residues as well as the attached sugars were used for the glucosylation experiments (Fig. S1). 62 Besides the unglucosylated flavonole quercetin, quercetin-3-O- $\beta$ -glucoside and rutin (which 63 corresponds to quercetin-3-O-rutinoside) were also used as substrates. Furthermore, the plant 64 pigment cvanidin-3-*O*-β-glucoside was isolated from blackberry and used for the glucosylation 65 experiments. With regards to taste-active flavonoids/flavonoid derivatives, naringin and 66 neohesperidin dihydrochalcone were used as substrates. These compounds carry a 67 68 neohesperidose residue at position 07, thus they show a different type and location of the sugar side chain than the previously mentioned substrates. In addition, dihydromyricetin 69 (ampelopsin) and epigallocatechin gallate were used. In contrast to all other substrates, these 70 71 non-glycosylated flavonoids both carry a pyrogallol group at the B-ring (and, in case of 72 epigallocatechin gallate, at position O3 in form of gallic acid). Therefore, the substrates used in 73 this study exhibit different flavonoid types, different substitution patterns of the flavonoid B-74 ring, and differences in the type and position of the flavonoid-bound sugar units.

75 In preliminary experiments, wildtype dextransucrase from L. reuteri TMW 1.106 was 76 successfully applied for glucosylation of some flavonoid substrates, but a rather low glucosylation efficiency was achieved. Devlamynck, te Poele, Meng, van Leeuwen and 77 78 Dijkhuizen [26] demonstrated that modification of certain amino acids in the +1 and +2acceptor binding site of glucansucrase Gtf180 from L. reuteri 180 eventually led to an increased 79 glucosylation of non-carbohydrate molecules by suppressing synthesis of polymeric  $\alpha$ -glucans. 80 Therefore, we aligned the amino acid sequence of L. reuteri TMW 1.106 dextransucrase and 81 82 Gtf180 to identify the leucine residue equivalent to L981 in Gtf180 (Fig. S2). Subsequently, we changed this leucine residue to alanine by using site-directed mutagenesis (yielding L. reuteri 83 84 TMW 1.106 L242A). The mutant L. reuteri TMW 1.106 dextransucrase showed a clearly increased glucosylation activity and was thus used for all further experiments. To evaluate 85 incubation conditions which allow for the isolation of conjugate amounts suitable for structure 86 87 elucidation by NMR spectroscopy, different amounts of substrate, sucrose, and enzyme were 88 tested in a small-scale. It was concluded that the highest amount of conjugates can be obtained 89 by using the highest acceptor concentration possible (with regards to solubility), 500 mM sucrose, and 100 mU mutant L. reuteri TMW 1.106 dextransucrase / mL substrate solution. 90 91 Therefore, these conditions were used for the glucosylation of all substrates.

92 To obtain information on the portion and the molecular masses of the conjugates formed, the 93 incubation mixtures were analyzed by HPLC-DAD/MS. As a first approximation, conjugation 94 efficiency can be derived from the UV/Vis absorption signal intensity. However, changes in the 95 UV/Vis response due to conjugation need to be considered. The conjugates were named after the substrate (prefix) and the number of added  $\alpha$ -glucosyl moieties (suffix, derived from the 96 97 m/z of the individual conjugates). For unambiguous structural elucidation, conjugates were isolated by preparative HPLC and characterized by using several one- and two-dimensional 98 99 NMR experiments. Although the proton spectra yield valuable information on the flavonoid

100 derived protons, sugar-derived protons can often not be resolved due to signal overlap. However, a Heteronuclear Single Quantum Coherence (HSQC) experiment mostly allows for 101 the resolution of all signals, because they are spread into the <sup>13</sup>C dimension. To identify 102 downfield shifted HSQC signals due to a-glucose substitution, spectra of the untreated 103 104 compounds were also recorded. Identification and assignment of the individual HSQC signals 105 was achieved by using H,H-Correlated Spectroscopy (COSY), Heteronuclear Two-bond 106 Correlation (H2BC), Total Correlated Spectroscopy (TOCSY), and HSQC-TOCSY 107 experiments. Eventually, unambiguous information on interglycosidic linkages and on 108 substitution of the flavonoid residues was obtained from the Heteronuclear Multiple Bond Correlation (HMBC) spectra. The <sup>1</sup>H and <sup>13</sup>C chemical shifts of all flavonoids and their 109 110 conjugates are provided in the supplementary data (Tab. S1-S8).

# **3.1** Quercetin and quercetin glycosides



Fig. 1: HPLC-DAD/MS chromatograms of incubations of quercetin (Q, final concentration: 3
mM, subfigure A), quercetin-3-*O*-β-glucoside (Q3G, final concentration: 5 mM, subfigure B),
and rutin (R, final concentration: 3 mM, subfigure C) with 100 mU / mL mutant *L. reuteri*TMW 1.106 dextransucrase and 500 mM sucrose. The structures of the conjugates (subfigure
D) were determined by NMR spectroscopy.

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120 One major monoglucosylated conjugate (Q-G1) was obtained after incubation of quercetin with 121 mutant L. reuteri TMW 1.106 dextransucrase (Fig. 1A). According to the peak areas, this conjugate was present in about equal amounts than quercetin which indicates a high conjugation 122 123 efficiency (assuming an at least roughly comparable UV response). In accordance with its 124 molecular mass, signals of one  $\alpha$ -glucose were detected in the HSQC spectrum of the isolated 125 compound. This sugar unit was identified by its characteristic <sup>1</sup>H and <sup>13</sup>C chemical shifts and the <sup>3</sup>J<sub>HH</sub> coupling constant of the anomeric proton. With regards to the quercetin unit, a 126 127 downfield shift of C5'H5' and C6'H6' was observed compared to unmodified quercetin. 128 Furthermore, H1 of glucose as well as H2', H5', and H6' of quercetin showed an HMBC 129 correlation at 146.5 ppm. Because of this <sup>13</sup>C chemical shift and because of the absence of 130 another HMBC signal between 145 and 150 ppm, these results show that the B-ring is 131 substituted at position O4' (Fig. 1D). This conclusion was based on previous NMR data on 132 quercetin α-glucosides [17]. As reported by Moon, Lee, Jhon, Jun, Kang, Sim, Choi, Moon and Kim [17], substitution at position O4' leads to almost identical <sup>13</sup>C chemical shifts of C3' and 133 134 C4' (146.4 and 146.9 ppm), whereas substitution at position O3' results in two distinct <sup>13</sup>C 135 chemical shifts for C3' (144.8 ppm) and C4' (149.8 ppm). This is also in good agreement with 136 NMR data from our laboratory for glucosylated phenolic acids with a catechol function. 137 Notably, the HPLC chromatogram also contained very low portions of another 138 monoglucosylated product (retention time 43 min) which most likely corresponds to the O3' 139 substituted conjugate. However, from the very low abundance of this conjugate, it can 140 concluded that position O4' of quercetin is highly preferred. In previous reports, glucansucrases 141 from Leuconostoc mesenteroides B-1299CB and Leuconostoc mesenteroides NRRL B-512F 142 were also reported to prefer position O4' over position O3', but compared to our results, higher 143 portions of O3' glucosylated quercetin were detected [14, 17]. Furthermore, Malbert, Moulis,

144 Brison, Morel, Andre and Remaud-Simeon [20] obtained several mono- and oligoglucosylated 145 products by using a branching sucrase from Leuconostoc mesenteroides B-1299. However, the 146 selective glucosylation of quercetin at position O4' observed for mutant L. reuteri TMW 1.106 147 dextransucrase may be advantageous for some applications. The glucosylation efficiencies described for the previously used wildtype glucansucrases showed broad variation from 4 % 148 [14] to 25 % [17] and 50 % [20]. In addition, some mutants of the branching sucrase yielded 149 glucosylation efficiencies of even about 90 % [20]. Therefore, mutant L. reuteri TMW 1.106 150 151 dextransucrase provides a high glucosylation efficiency (about 50 % based on the peak areas) compared to some other enzymes, but a higher efficiency may be achieved by creating and 152 153 screening more mutants.

154 Conjugation of quercetin-3-*O*-β-glucoside also resulted in one major product (Q3G-G1a in Fig
155 1B) which was isolated from the reaction mixture. In addition, it was possible to purify one
156 diglucosylated product (Q3G-G2) and two other monoglucosylated conjugates (Q3G-G1b and
157 Q3G-G1c).

Just as for Q-G1, comparison of the main product Q3G-G1a with unmodified quercetin-3-O- $\beta$ glucoside showed that the HSQC signals derived from the B-ring of quercetin showed significantly different chemical shifts. Furthermore, the chemical shifts of the additional  $\alpha$ glucose unit were in good agreement with the chemical shifts of the  $\alpha$ -glucose unit in Q-G1, suggesting a similar chemical environment. Eventually, the attachment of  $\alpha$ -glucose to position O4' of the B-ring was confirmed by an HMBC correlation between H1 of glucose and C4' of quercetin at 5.41 / 147.0 ppm.

For the two monoglucosylated conjugates Q3G-G1b and Q3G-G1c, quercetin derived HSQC signals were not significantly altered, suggesting substitution of the  $\beta$ -glucose unit. After assignment of the HSQC signals by using COSY/TOCSY spectra, characteristic chemical shifts of the  $\beta$ -glucose derived signals were used to assign the substitution position of the  $\beta$ -glucose

169 unit (Fig. S3). Q3G-G1b showed a significantly downfield shifted C3H3 signal (3.44 / 84.8 170 ppm), whereas Q3G-G1c showed a downfield shifted C4H4 signal (3.38 / 79.1 ppm). Therefore, 171 these two monoglucosylated conjugates result from substitution of  $\beta$ -glucose at position *O*3 and 172 *O*4, respectively.

173 As expected, the diglucosylated conjugate Q3G-G2 showed two sets of  $\alpha$ -glucose derived 174 signals (Fig. 2). From the characteristically shifted HSQC signals of quercetin and the chemical 175 shifts of one  $\alpha$ -glucose unit, it was concluded that quercetin is substituted at position O4' of the B-ring. The attachment of the corresponding glucose unit to position O4' of quercetin was also 176 177 confirmed by an HMBC correlation between H1 of glucose and C4' of quercetin. The chemical 178 shifts of the  $\beta$ -glucose unit and the remaining  $\alpha$ -glucose unit were in good agreement with the chemical shifts observed for  $\alpha/\beta$ -glucose in Q3G-G1b. Only the C1H1 and C2H2 signals of  $\beta$ -179 180 glucose were slightly shifted which is most likely a result of the B-ring substitution. However, 181 the attachment of the second  $\alpha$ -glucose to position O3 of  $\beta$ -glucose was clearly demonstrated 182 by the diagnostic C3H3 signal of  $\beta$ -glucose (Fig. 2) and an HMBC correlation between H1 of 183  $\alpha$ -glucose and C3 of  $\beta$ -glucose.



186Fig. 2: HSQC spectrum and structure of the diglucosylated conjugate of quercetin-3-O-β-187glucoside (Q3G-G2). Signals belonging to the two α-glucose residues are colored in red and188blue. The diagnostic, downfield shifted C3H3 signal of the β-glucose unit is encircled in orange.

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190 The predominant abundance of Q3G-G1a which is glucosylated at position O4' of the B-ring 191 demonstrates that the preference for this position is not altered by the  $\beta$ -glucosyl residue at 192 position O3 of quercetin. Furthermore, the other three monoglucosylated conjugates show that 193 glucosylation can appear at different positions of the  $\beta$ -glucosyl residue. In addition, already 194 glucosylated compounds can be further substituted with  $\alpha$ -glucose which results in the 195 formation of Q3G-G2. From the peak intensities in the chromatogram, a comparably high 196 glucosylation efficiency can be derived, because more than half of quercetin-3-O-β-glucoside 197 is glucosylated.

198 When rutin was incubated with mutant *L. reuteri* TMW 1.106 dextransucrase, only two 199 monoglucosylated conjugates were detected. For R-G1a, the rutinose derived HSQC signals 200 were in good agreement with unsubstituted rutin. Furthermore, <sup>1</sup>H and <sup>13</sup>C chemical shifts of 201 quercetin as well as the additional  $\alpha$ -glucose unit were in good agreement with the chemical 202 shifts obtained for Q3G-G1a. Therefore, R-G1a is glucosylated at position *O*4' which was also 203 confirmed by an HMBC correlation between H1 of glucose and C4' of quercetin.

For R-G1b, the quercetin signals were unmodified compared to the standard. In addition, assignment of the rutinose HSQC signals demonstrated that the chemical shifts of  $\beta$ -glucose were not significantly altered. However,  $\alpha$ -rhamnose showed a clearly downfield shifted signal at 3.22 / 81.8 ppm. This signal was identified as C4H4 by using the COSY spectrum as well as an HMBC correlation between C4 and the characteristically located H6 of  $\alpha$ -rhamnose (Fig. 3). From an HMBC correlation between C4 of the  $\alpha$ -rhamnose unit and H1 of the  $\alpha$ -glucose unit, it was also unambiguously demonstrated that the  $\alpha$ -glucose unit is attached to this position.

These results demonstrate that the addition of a disaccharide to position *O*3 of quercetin does not impede glucosylation at position *O*4', but prevents glucosylation of the  $\beta$ -glucose unit. Furthermore, the glucosylation efficiency seems to be lowered, as the major part of rutin remained unmodified under the conditions used. However, glucosylation of the rhamnosyl residue at position *O*4 is an interesting novel feature of glucansucrases.



Fig. 3: HMBC spectrum and structure of the monoglucosylated conjugate of rutin (R-G1b). Diagnostic signals which helped to identify C4 of  $\alpha$ -rhamnose and demonstrate the attachment of  $\alpha$ -glucose to position *O*4 of  $\alpha$ -rhamnose are encircled in red and blue.

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#### 222 **3.2** Cyanidin-3-glucoside

223 The chromatogram obtained for the cyanidin-3-*O*-β-glucoside incubation mixture contained 224 one additional, earlier eluting peak (Fig. 4).

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Fig. 4: HPLC-DAD/MS chromatogram of the incubation of cyanidin-3-*O*-β-glucoside (Cy3G,
final concentration: 8 mM) with 100 mU / mL mutant *L. reuteri* TMW 1.106 dextransucrase
and 500 mM sucrose. The structure of the monoglucosylated conjugate Cy3G-G1 is shown in
Fig. 5.

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The compound (Cy3G-G1) showed a quasimolecular ion at m/z 611 which demonstrates monoglucosylation of cyanidin-3-O- $\beta$ -glucoside. For structural elucidation, the conjugate was isolated by preparative HPLC and NMR spectra were recorded in methanol-d4 with 0.1 % DCl. This solvent was used because it provides complete solubilization and well resolved spectra of the flavylium cation form. As expected, the HSQC spectrum contained signals typical for an  $\alpha$ glucose in addition to the signals of cyanidin-3-O- $\beta$ -glucoside. Besides the slightly shifted C4H4 signal, the HSQC signals of cyanidin showed identical chemical shifts than for

239 unmodified cyanidin-3-*O*-β-glucoside. Thus it can be concluded that the β-glucose residue is 240 glucosylated. After assigning the HSQC signals of the β-glucose unit by using COSY/TOCSY 241 experiments, it was observed that the C6H6 signals showed a characteristic downfield shift (Fig 242 5).

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Fig. 5: HSQC spectrum and structure of the monoglucosylated conjugate of cyanidin-3-O- $\beta$ glucoside (Cy3G-G1). The diagnostic, downfield shifted C6H6 signals of the  $\beta$ -glucose unit are encircled in blue.

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These downfield shifted signals strongly suggest substitution at position *O*6 of the  $\beta$ -glucose unit, which was eventually confirmed by an HMBC correlation between C6 of the  $\beta$ -glucose and H1 of the  $\alpha$ -glucose. Therefore, cyanidin-3-*O*- $\beta$ -glucoside shows a significantly different product spectrum than the structurally related quercetin-3-*O*- $\beta$ -glucoside which was

- 253 glucosylated at position *O*4' of the B-ring and at positions *O*3 and *O*4 of  $\beta$ -glucose. Most likely, 254 the structural differences between the two compounds result in a different orientation of 255 cyanidin-3-*O*- $\beta$ -glucoside in the active site of the enzyme.
- 256

257 **3.3** Naringin and neohesperidin dihydrochalcone



Fig. 6: HPLC-DAD/MS chromatograms of incubations of naringin (Na, final concentration: 30 mM, subfigure A) and neohesperidin dihydrochalcone (Ne, final concentration: 50 mM, subfigure B) with 100 mU / mL mutant *L. reuteri* TMW 1.106 dextransucrase and 500 mM sucrose. The peaks marked with an asterisk in subfigure B were also present in the untreated neohesperidin dihydrochalcone preparation (see text). The structures of the conjugates (subfigure C) were determined by NMR spectroscopy.

265 Two monoglucosylated conjugates were detected in the incubation mixture of naringin with 266 mutant L. reuteri TMW 1.106 dextransucrase (Fig. 6A). However, the peak intensity suggested that the conjugates are lowly abundant and that glucosylation efficiency is rather low. 267 268 Nevertheless, it was possible to isolate and characterize the two conjugates. The HSQC 269 spectrum of Na-G1a vielded unmodified neohesperidose signals and the anomeric signal of the  $\alpha$ -glucose showed a comparably high <sup>1</sup>H chemical shift. This was already observed for 270 271 quercetin (glycoside) conjugates which are glucosylated at position O4' of the B-ring. Because 272 the C2'H2', C3'H3', C5'H5', C6'H6' of naringin were also shifted downfield, it can be concluded that naringin is glucosylated at the hydroxyl group of the B-ring. The attachment of 273  $\alpha$ -glucose to the B-ring was also confirmed by an HMBC correlation between H1 of  $\alpha$ -glucose 274 and C4' of naringin (Fig. S4). 275

The second monoglucosylated conjugate Na-G1b did not show any signal drift for the HSQC signals derived from the aglycone and the  $\alpha$ -rhamnose unit. However, just as for the conjugate of cyanidin-3-*O*- $\beta$ -glucoside, C6H6 signals of the naringin-bound  $\beta$ -glucose unit showed a characteristic <sup>13</sup>C downfield shift to 66 ppm (Fig. S5). Therefore it can be concluded that the additional  $\alpha$ -glucose unit is bound to position *O*6 of  $\beta$ -glucose which was also confirmed by a corresponding HMBC correlation.

282 HPLC-DAD/MS and NMR spectroscopic investigations revealed that the neohesperidin 283 dihydrochalcone preparation contained some residual amounts of two flavonoids which were 284 identified as neohesperidin and diosmin-7-neohesperidoside (peaks marked with an asterisk in 285 Fig. 6B). The monoglucosylated neohesperidin dihydrochalcone conjugate formed by mutant 286 L. reuteri TMW 1.106 dextransucrase eluted near the two flavonoids and was therefore only 287 obtained in a mixture with neohesperidin. However, it was possible to assign all HSQC signals 288 corresponding to the aglycone (hesperetin) and the unsubstituted neohesperidose by using 289 COSY and TOCSY experiments. With regards to the monoglucosylated neohesperidin

290 dihydrochalcone conjugate, the chemical shifts of the HSQC signals derived from the aglycone 291 were comparable to the chemical shifts obtained for unmodified neohesperidin 292 dihydrochalcone. In the anomeric region of the HSQC spectrum, two anomeric signals were 293 detected at 5.10/100.1 ppm and 5.14/97.0 ppm besides the additional signals of a carbohydrate-294 bound α-glucose unit. The carbon shifts of these two signals were similar to unmodified neohesperidose, therefore the signals can be assigned to  $\beta$ -glucose and  $\alpha$ -rhamnose. The 295 assignment of the remaining HSQC signals of these two sugar units showed that the C4H4 296 297 signal of the  $\alpha$ -rhamnose unit was clearly shifted downfield in the <sup>13</sup>C dimension. Furthermore, an HMBC correlation between H1 of  $\alpha$ -glucose and C4 of  $\alpha$ -rhamnose was found. Therefore, 298 it was possible to conclude that the  $\alpha$ -glucose unit is bound to position O4 of the  $\alpha$ -rhamnose 299 300 unit.

The structures of the naringin and neohesperidin dihydrochalcone conjugates demonstrate that 301 302 mutant L. reuteri TMW 1.106 dextransucrase is also able to glucosylate flavonoid-bound 303 neohesperidose units. However, naringin was glucosylated at the  $\beta$ -glucose unit and neohesperidin dihydrochalcone was glucosylated at the  $\alpha$ -rhamnose unit. These results suggest 304 305 that the aglycone has a significant influence on the glucosylation position. This may be caused 306 by different orientations of the acceptor substrate in the active site of the dextransucrases. 307 However, just as rutin (R-G1b), neohesperidin dihydrochalcone was glucosylated at position *O*4 of the  $\alpha$ -rhamnose unit. Thus, the position to which  $\alpha$ -rhamnose is attached seems to be of 308 309 secondary importance and conjugation of  $\alpha$ -rhamnose preferably occurs at position O4. 310 However, glucosylation of α-rhamnose was only observed for substrates which lack substitution 311 of the  $\beta$ -glucose unit. Therefore, it is likely that the corresponding conjugates are only formed 312 if the  $\beta$ -glucose units are not in a suitable position in the active site. Furthermore, Na-G1a, 313 which is substituted at position O4' of naringin, is a rather uncommon reaction product because

- 314 previous studies suggested that glucansucrases prefer vicinal phenolic hydroxyl groups for315 glucosylation [14, 29].
- 316
- 317 **3.4 Epigallocatechin gallate and dihydromyricetin**



Fig. 7: HPLC-DAD/MS chromatograms of incubations of dihydromyricetin (DHM, final concentration: 30 mM, subfigure A) and epigallocatechin gallate (EGCG, final concentration: 20 mM, subfigure B) with 100 mU / mL mutant *L. reuteri* TMW 1.106 dextransucrase and 500 mM sucrose. The structures of the conjugates (subfigure C) were determined by NMR spectroscopy.

In the incubation mixture of dihydromyricetin, a monoglucosylated conjugate was largely predominant besides low amounts of a diglucosylated conjugate (Fig. 7A). Preparative HPLC allowed for the isolation of both compounds, but both fractions still contained residual unmodified dihydromyricetin. However, signals derived from the additional  $\alpha$ -glucose units were clearly assignable because of their different chemical shift.

The HSQC spectrum of the monoglucosylated conjugate DHM-G1 contained the characteristic a-glucose signals as well as one downfield shifted signal in the aromatic region which was identified as the C2'H2' and C6'H6' signals of the conjugated dihydromyricetin. Because glucosylation at position O3' / O5' would result in two distinct signals, these results suggest that dihydromyricetin is glucosylated at position O4'. This was also confirmed by an HMBC correlation between C4' of the conjugated dihydromyricetin and H1 of  $\alpha$ -glucose.

For DHM-G2, the chemical shifts of conjugated dihydromyricetin were identical to DHM-G1, 336 337 thus this conjugate is also substituted at position O4'. As expected for a diglucosylated conjugate, two anomeric signals were detected (Fig. S6). The anomeric signal of the 338 dihydromyricetin-bound glucosyl unit was identified based on its chemical shift and an HMBC 339 correlation between H1 and C4' of dihydromyricetin. From the assignment of the remaining 340 HSQC signals, it was demonstrated that the C4H4 signal of the dihydromyricetin-bound  $\alpha$ -341 glucose unit was significantly downfield shifted in the <sup>13</sup>C dimension (69.1 to 78.6 ppm). In 342 343 addition, the chemical shifts of the second glucosyl unit were in good agreement with glycoside-344 bound, terminal  $\alpha$ -glucose. Therefore, DHM-G2 contains a maltosyl unit bound to position O4' 345 of the B-ring.

Incubation of epigallocatechin gallate with mutant *L. reuteri* TMW 1.106 dextransucrase yielded two monoglucosylated conjugates which eluted near to each other and also near to unmodified epigallocatechin gallate (Fig. 7B). As a consequence, the two conjugates were only isolated in a mixture with each other and unconjugated epigallocatechin gallate. Thus, the

350 HSQC spectrum contained two close-by anomeric  $\alpha$ -glucose signals (Fig. 8A). However, the 351 five non-anomeric HSQC signals of the two  $\alpha$ -glucose units were identical. The aromatic region 352 of the spectrum was dominated by two signals (6.82/108.4 ppm and 6.40/105.2 ppm) which can 353 be assigned to the unconjugated B-ring (C2'H2' / C6'H6') and unconjugated gallic acid 354 (C2''H2'' / C6''H6'') of epigallocatechin gallate. Besides these signals (and the C6H6 / C8H8 signals), one additional signal at 6.48/105.2 ppm was detected. This signal showed a 355 356 comparable <sup>13</sup>C chemical shift than C2'H2'/C6'H6' of the unconjugated B-ring which suggests 357 that one conjugate is substituted at position *O*4'. This was confirmed by an HMBC correlation 358 between the H1 of the glucose unit at 4.83 ppm and the carbon with a chemical shift of 133.3 359 ppm (red bar in Fig 9B, assigned as C4'). The other  $\alpha$ -glucose showed an HMBC correlation 360 between the anomeric proton (5.06 ppm) and a carbon at 137.7 ppm (blue bar in Fig. 8B) which 361 corresponds to C4'' of the ester-bound gallic acid. Therefore, an  $\alpha$ -glucose unit is bound to the 362 gallic acid residue, but glucosylation does not impact the proton shift of the substituted gallic 363 acid. This phenomenon was also previously observed by Moon, Kim, Lee, Jin, Kim and Kim 364 [16].



366

**Fig. 8.** A: HSQC spectrum and structures of the two monoglucosylated conjugates of epigallocatechin gallate (EGCG-G1a and EGCG-G1b) which were only obtained in a mixture. Signals belonging to the two different  $\alpha$ -glucose residues are colored in red and blue. B: HMBC spectrum of the conjugate mixture. Diagnostic correlations between H1 of the  $\alpha$ -glucose units and C4' of the B-ring and C4'' of gallic acid are marked by red and blue bars, respectively.

373 The incubation of dihydromyricetin and epigallocatechin gallate with mutant L. reuteri TMW 374 1.106 dextransucrase demonstrates that flavonoids with a pyrogallol function (located at the B-375 ring or ester-bound gallic acid) are also efficiently glucosylated by this enzyme. From the 376 structures of the conjugates it can also be derived that pyrogallol functions are preferably 377 glucosylated at position O4 as it was also observed for flavonoids with catechol functions. 378 Maltosyl-substituted conjugates were not detected for all substrates but dihydromyricetin, 379 therefore, the formation of this conjugate type is most likely also dependent on the structure of the aglycone. Previous studies on the glucosylation of dihydromyricetin and epigallocatechin 380

381 gallate by glucansucrase from *Leuconostoc mesenteroides* B-1299CB also showed preferred 382 glucosylation at position *O*4 of the B-ring and gallic acid [16, 18]. In contrast to the previous 383 studies, oligoglucosylated conjugates or conjugates which are additionally glucosylated at 384 position *O*7 of the flavonoid were not observed in our study. This may be a result of the 385 dextransucrase specificity and or the mutations introduced. However, these results once again 386 demonstrate that mutant *L. reuteri* TMW 1.106 dextransucrase is suitable for effective and 387 specific monoglucosylation without creating high amounts of other products.

388

#### 389 3 CONCLUSION

390 The results of our study demonstrate that mutant L. reuteri TMW 1.106 dextransucrase is highly 391 suitable for the synthesis of different flavonoid glucosides. For substrates such as quercetin and 392 its glycosides, a high glucosylation efficiency was achieved, whereas the incubation of e.g. 393 neohesperidin dihydrochalcone and naringin only yielded low portions of glucoconjugates. The 394 different substrates were not only glucosylated with varying efficiencies but also at different 395 positions. While glucosylation of the phenolic hydroxyl groups of the flavonoid B-ring occurred 396 almost exclusively at position  $O4^{\circ}$ , flavonoid-bound  $\beta$ -glucosyl units were conjugated at 397 position 03, 04, and 06 depending on the substrate. Furthermore, it was possible to 398 demonstrate that flavonoids with a rutinose and a neohesperidose residue can be glucosylated 399 either at position O6 of the glucose unit or at position O4 of the  $\alpha$ -rhamnose unit. Overall, 400 glucosylation by mutant L. reuteri TMW 1.106 dextransucrase is highly dependent both on the 401 flavonoid residue and the glycosyl residue. Future studies could evaluate how stability, 402 antioxidant properties, bioavailability, and taste of the conjugates are affected by glucosylation.

#### 404 **4 EXPERIMENTAL**

#### 405 **4.1 Materials**

406 Quercetin ( $\geq$  98 %), quercetin-3-*O*- $\beta$ -glucoside ( $\geq$  98 %), rutin ( $\geq$  97 %), naringin ( $\geq$  98 %), 407 neohesperidin dihydrochalcone ( $\geq 98$  %), epigallocatechin gallate ( $\geq 98$  %), and (if not stated 408 otherwise) all other chemicals were purchased from VWR (Radnor, PA, USA), Sigma Aldrich 409 (Schnelldorf, Germany), Carl Roth (Karlsruhe, Germany), or Carbosynth (Newbury, UK). 410 Cyanidin-3-*O*-β-glucoside was extracted from freeze-dried blackberry powder by using acidic 411 methanol. Anthocyanins were concentrated by solid phase extraction as described previously 412 [30] and purified by preparative HPLC (see below). Dihydromyricetin was obtained from a 413 commercial dietary supplement. pLIC-SGC1 was a gift from Nicola Burgess-Brown (Addgene 414 plasmid # 39187).

## 415 **4.2** Molecular cloning and site-directed mutagenesis

416 The gene encoding the previously described dextransucrase from *L. reuteri* TMW 1.106 [27] 417 was amplified from genomic DNA by using a Phusion High-Fidelity PCR kit (Thermo Fisher 418 Scientific, Waltham, MA, USA) and the primers listed in Table S9. The amplified and purified 419 gene was subsequently cloned into the pLIC-SGC1 vector by using a ligation independent 420 cloning approach [31-33]. Site directed mutagenesis of L. reuteri dextransucrase was conducted 421 by using a QuikChange Lightning Multi Site-Directed Mutagenesis Kit (Agilent Technologies, 422 Santa Clara, CA, USA) and the mutagenic primer listed in Table S9. After purification, the 423 sequence of the mutagenic plasmid was confirmed by Sanger sequencing (Eurofins GATC 424 Biotech, Konstanz, Germany).

#### 425 **4.3** Heterologous expression

For heterologous expression of the mutated dextransucrase gene, the plasmid was transformed
into OneShot BL21 Star (DE3) cells by heat shock. After transferring one colony into 5 mL of

428 LB medium (100 µg ampicillin / mL), cells were grown at 37 °C and 225 rpm for 6 h. The 429 culture was used to inoculate 200 mL of fresh LB medium (100 µg ampicillin / mL) which was 430 further incubated at 37 °C and 225 rpm for 3 h. After induction of protein expression with 431 isopropyl-β-D-thiogalactopyranoside (final concentration: 0.1 mM), the culture was incubated 432 overnight at room temperature and 225 rpm. Cells were harvested by centrifugation and resuspended in binding buffer (50 mM sodium phosphate, 300 mM NaCl, 10 mM imidazole, pH 433 434 7.5). Lysis of the re-suspended cells was carried out by sonication (amplitude 50 %, 3x 20 s 435 pulse, 59.9 s pause) with a Sonifier W-250 D (Branson, Dabury, CT, USA) and cell debris were removed by centrifugation (30 min, 4 °C, 20000 x g). For purification of the recombinant 436 437 dextransucrase, the clear supernatant was transferred to a HisPur Ni-NTA resin (Thermo Fisher 438 Scientific; pre-equilibrated with binding buffer) and incubated at 4 °C for 1 h. After washing 439 with 4 volumes of binding buffer, recombinant proteins were eluted with elution buffer (50 mM 440 sodium phosphate, 300 mM NaCl, 100 mM imidazole, pH 7.5) which resulted in a typical yield 441 of 100 mg dextransucrase / L culture. Hydrolytic activity of the recombinant dextransucrase 442 was determined by analyzing the release of para-nitrophenol from 35 mM para-nitrophenyl-a-443 glucoside (25 mM sodium acetate buffer, supplemented with 1 mM CaCl<sub>2</sub>, pH 5.5) in a microplate reader (Infinite M200 PRO, Tecan, Männedorf, Switzerland) at 400 nm. A standard 444 curve was used to calculate the volume activity (1 µmol of released *para*-nitrophenol / min) per 445 446 μL.

#### 447 4.4 Acceptor reactions

For glucosylation, quercetin, quercetin-3-*O*-β-glucoside, rutin, cyanidin-3-*O*-β-glucoside,
naringin, neohesperidin dihydrochalcone, epigallocatechin gallate, and dihydromyricetin were
dissolved in DMSO and four volumes of 25 mM sodium acetate buffer (1 mM CaCl<sub>2</sub>, pH 5.5)
were added. This procedure allowed for solubilization of comparably high amounts of these
substrates (quercetin: 3 mM, quercetin-3-*O*-β-glucoside: 5 mM, rutin: 3 mM, cyanidin-3-*O*-β-

glucoside: 8 mM, naringin: 30 mM, neohesperidin dihydrochalcone: 50 mM, epigallocatechin
gallate: 20 mM, dihydromyricetin: 30 mM). After the addition of sucrose (final concentration:
500 mM), 100 mU of *L. reuteri* TMW 1.106 L242A dextransucrase / mL substrate solution
were added and the mixture was incubated for 24 h at 30 °C. The enzyme was inactivated by
heating to 95 °C for 5 min and the mixture was used for further analysis and fractionation.

458 **4.5** Glucoconjugate analysis and purification

459 For chromatographic analysis of the glucoconjugates, the glucosylation mixtures were diluted and analyzed on a Surveyor HPLC-DAD system coupled to an LXQ linear ion trap MS<sup>n</sup> system 460 (Thermo Fisher Scientific). A Kinetex C18 column (100 mm x 4.6 mm i.d., 2.6 µm particle 461 462 size; Phenomenex, Torrance, CA, USA) and the following gradient composed of water with 0.1 463 % formic acid (eluent A) and acetonitrile with 0.1 % formic acid (eluent B) were used at 0.5 mL/min and 30 °C: 0-5 min, isocratic 3 % B; 5-25 min, linear to 8 % B; 25-45 min, linear to 464 465 28 % B; 45-55 min, linear to 60 % B; 55-56 min, linear to 100 % B; 56-62 min, isocratic 100 % B; 62-64 min, linear to 3 % B; 64-70 min, isocratic 3 % B. For the analysis of cyanidin-3-O-466 β-glucoside, 0.5 % formic acid was added to both eluents. Flavonoids were detected at 280 nm 467 (neohesperidin dihydrochalcone, epigallocatechin gallate, and dihydromyricetin), 320 nm 468 469 (quercetin, quercetin-3-O-\beta-glucoside, rutin, and naringin), and 510 nm (cyanidin-3-O-βglucoside). For mass spectrometric analysis of the unmodified and conjugated flavonoids, 470 471 electrospray ionization was used in positive (cyanidin-3-O-β-glucoside) and negative (all other 472 compounds) mode. Prior to purification with preparative HPLC, conjugated flavonoids were 473 enriched by using solid phase extraction on a Chromabond C18 (Macherey Nagel, Düren, 474 Germany; preconditioned with methanol and water) stationary phase. The glucosylation 475 mixtures were loaded onto the column, the column was washed with water, and conjugates were 476 eluted by using methanol. Fractionation of the conjugates was conducted on a preparative 477 HPLC-UV system (LC-8A pumps, SPD-20A UV detector, Shimadzu, Kyoto, Japan) equipped

with a Luna C18 column (250 mm × 10 mm i.d., 5 µm particle size, Phenomenex). The same conditions as described above were used at 8 mL/min and room temperature. In some cases, the gradient was truncated to: 0-5 min, linear from 3 % B to 8 % B; 5-25 min linear to 28 % B; 25-35 min, linear to 40 % B; 35-36 min, linear to 100 % B; 36-41 min, isocratic 100 % B; 41-42 min, linear to 3 % B, 42-47 min, isocratic 3 % B. Detection of the conjugates was carried out at 510 nm (cyanidin-3-O-β-glucoside) and 280 nm (all other compounds). The resulting fractions were evaporated and freeze-dried.

#### 485 **4.6** NMR spectroscopy

NMR spectroscopic analysis of the standard compounds and conjugates was carried out on an
Ascend 500 MHz spectrometer (Bruker, Rheinstetten, Germany) equipped with a Prodigy
cryoprobe. Cyanidin-3-*O*-β-glucoside was dissolved in methanol-*d*4 whereas all other
conjugates were dissolved in DMSO-*d*6. Chemical shifts were referenced to the solvent residual
peak (methanol-*d*4: 3.31 / 49.00 ppm; DMSO-*d*6: 2.50 / 39.52 ppm) according to Gottlieb,
Kotlyar and Nudelman [34]. Standard Bruker parameter sets were used to record proton, HSQC,
H2BC, COSY, (HSQC-)TOCSY, and HMBC spectra.

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- 620

# 1 HIGHLIGHTS

- Multiple flavonoids were glucosylated by using a mutant *L. reuteri* dextransucrase
- The B-ring of different flavonoids was preferably glucosylated at position *O*4
- Flavonoid-bound glucose units were glucosylated at varying positions
- 5 Rutinose and neohesperidose units were glucosylated at position *O*4 of rhamnose
- Glucosylation positions and extent greatly depend on the structure of the substrate

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