

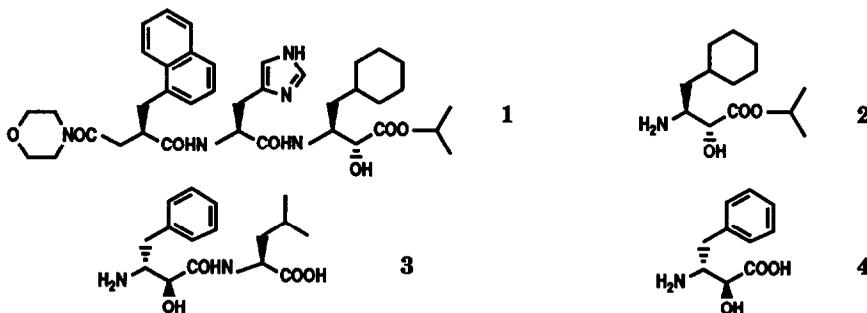
A NOVEL SYNTHESIS OF THE (2*R*,3*S*)- AND (2*S*,3*R*)-3-AMINO-2-HYDROXYCARBOXYLIC ACID DERIVATIVES, THE KEY COMPONENTS OF A RENIN INHIBITOR AND BESTATIN, FROM METHYL (*R*)- AND (*S*)-MANDELATE

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Abstract: The title synthesis could be accomplished by featuring the [2+2]-cycloaddition reaction of a chiral imine with benzyloxyketene, alcoholysis of the formed 2-azetidinone derivative, and reductive removal of the mandelate-derived benzylic oxygen by way of a 2-oxazolidone derivative.

Optically active 3-amino-2-hydroxycarboxylic acid derivatives are often involved in medicinally important compounds as their key components. Thus, one of the promising renin inhibitor (1)²⁾ bears isopropyl (2*R*,3*S*)-3-amino-4-cyclohexyl-2-hydroxybutyrate (2)^{2,3)} as its C-terminal moiety, and bestatin(3), the famous immunological response modifier,⁴⁾ consists of (2*S*,3*R*)-3-amino-2-hydroxy-4-phenylbutyric acid (4)^{3d,5)} and (*S*)-leucine.

We wish to report here a novel synthesis of these antipodal compounds (2 and 4) starting from methyl (*R*)- and (*S*)-mandelate (5 and ent-5). The explored synthetic scheme features the [2+2]-cycloaddition reaction of a chiral imine with benzyloxyketene, alcoholysis of the formed 2-azetidinone derivative, and reductive removal of the mandelate-derived benzylic oxygen by way of a 2-oxazolidone derivative. It was previously disclosed that the [2+2]-cycloaddition reaction of chiral imine having an asymmetric center on the nitrogen atom of C=N bond with benzyloxyketene was unrewarding for the preparation of 2 because of its low diastereoselectivity.^{3c)} However, we have now found that a chiral imine derived from 5 or ent-5, which bears an asymmetric center on the carbon atom of C=N bond, can react with benzyloxyketene in a highly diastereoselective manner and the 2-azetidinone derivative produced as a major addition product can be ingeniously elaborated to 2 or 4.



As shown in Scheme 1, the synthesis of 2 commences with protection of the hydroxy group of 5 with *t*-butyldimethylsilyl (TBDMS) or *t*-butyl (*t*-Bu) group.⁶⁾ These protective groups were employed since they can be readily removed under the conditions for acidic alcoholysis of a 2-azetidinone derivative (*vide infra*). Reduction of the protected esters (6 and 7) with diisobutylaluminum hydride (DIBAL) smoothly produced the corresponding aldehydes (8 and 9).⁶⁾ Two sorts of the aldehydes (8

Table 1 Chemical Yields of Imine Formation (8 or 9→10), [2+2]-Cycloaddition (10→11 and 12), Alcoholysis (11→13 or 14), 2-Oxazolidone Formation (13→15 or 14→16), and Hydrogenolysis (15 or 16→17)^{a)}

	R ²	R ³	Yield (%)				
			8 or 9→10	10→11 and 12 (11:12)	11→13 or 14 ^{d)}	13→15 or 14→16 ^{d)}	15 or 16→17
a	TBDMS	DAM	100	88 (10:1) ^{b)}	84 (13) ^{e)}	90 (15)	94
b	TBDMS	Bn	100	59 (12:1) ^{b)}	86 (14) ^{e)}	64 (16)	81
c	<i>t</i> -Bu	DAM	100	77 (9:1) ^{c)}	59 (13) ^{f)}	—	—
d	<i>t</i> -Bu	Bn	98	62 (15:1) ^{c)}	69 (14) ^{f)}	—	—

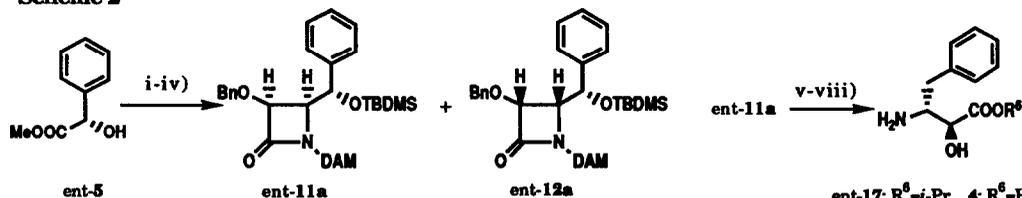
a) The reaction conditions are given in the footnotes of Scheme 1. b) Determined by weighing separated 11 and 12. c) Determined by the ¹H-NMR spectrum of the mixture of 11 and 12. Separation of 11 and 12 could not be achieved. d) Numbers in parentheses indicate the compound numbers. e) A pure sample of 11 was used for the reaction. f) A mixture of 11 and 12 was directly used for the reaction. The yield was calculated based on the total amount of the mixture.

remove the benzylic hydroxyl groups derived from 5, two sorts of the isopropyl esters (13 and 14) were converted to the 2-oxazolidone derivatives (15 and 16)⁶⁾ by treating with trichloromethyl chloroformate (phosgen dimer) in the presence of pyridine. Hydrogenolyses of 15 and 16 over 10% Pd on charcoal afforded same isopropyl (2*R*,3*S*)-3-amino-2-hydroxy-4-phenylbutyrate (17).⁶⁾ The convergent syntheses of 17 could nicely correlate the stereochemistries of 11*a*-*d*. Further catalytic reduction of 17 over 5% Rh on alumina furnished optically pure 2, mp 86-86.5°C and $[\alpha]_D^{20}$ -22.0° (c=1.18, CHCl₃) [*lit.*,^{3c)} mp 86-87°C and $[\alpha]_D^{20}$ -22.0° (c=1.08, CHCl₃)].

With completion of the synthesis of 2 from 5, the explored synthetic route was next applied to ent-5 to prepare 4. Thus, as shown in Scheme 2, a mixture of ent-11*a* and ent-12*a* (8:1) could be prepared similarly from ent-5 in 4 steps. A combination of TBDMS and DAM groups was employed to obtain a higher yield in the [2+2]-cycloaddition reaction. Elaboration of ent-11*a* to ent-17 according to the same procedure as described above gave ent-17. Acidic hydrolysis of ent-17 followed by treatment with an ion exchange resin, afforded 4, mp 235-237°C (decomp.) and $[\alpha]_D^{20}$ +29.9° (c=0.214, 1M HCl) [*lit.*,⁴⁾ mp 219-221°C and $[\alpha]_D^{20}$ +27.9° (c=0.717, 1M HCl)].¹²⁾

As mentioned above, we have succeeded in exploring a novel synthetic route to antipodal 2 and 4. The best combined yields (overall 8 steps) of 2 and 4 from 5 and ent-5 by way of 11*a* and ent-11*a* can be calculated as 44% and 30%, respectively. Taking into account high diastereoselectivity observed for the [2+2]-cycloaddition reaction, expeditious elaboration of 11 or ent-11 to 2 or 4, and use of com-

Scheme 2

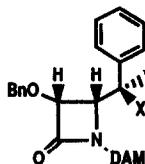
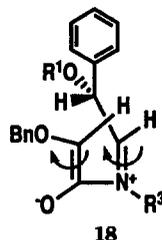


a) For reaction conditions (steps i-viii), see footnotes of Scheme 1. i) 93% ii) 80% iii) 100% iv) 90% (ent-11*a*:ent-12*a* = 8:1) v) 70% vi) 93% vii) 92% viii) 6M HCl, 100°C, 4h, then ion exchange resin (AG 50 X W2, H⁺-form), 85%

mercially available **5** or ent-**5** as a starting material, the overall process may have potential as one of the most reliable synthetic methods for preparing the (2*R*,3*S*)- and (2*S*,3*R*)-3-amino-2-hydroxycarboxylic acid derivatives such as **2** and **4**.

References and Notes

- 1) Present Address: Department of Chemistry, Faculty of Science, Kyushu University, Fukuoka 812, Japan.
- 2) a) K. Iizuka, T. Kamijo, T. Kubota, K. Akahane, H. Umeyama, and Y. Kiso, *Japan Kokai Tokkyo Koho*, JP 62-234071 (1987). b) K. Iizuka, T. Kamijo, H. Harada, K. Akahane, T. Kubota, H. Umeyama, and Y. Kiso, *J. Chem. Soc., Chem. Commun.*, 1989, 1678.
- 3) For other syntheses of **2**, see, a) T. Kamijo, H. Harada, A. Tsubaki, T. Yamaguchi, A. Iyobe, K. Iizuka, and Y. Kiso, *Japan Kokai Tokkyo Koho*, JP 1-172365 (1989). b) H. Harada, A. Tsubaki, T. Kamijo, K. Iizuka, and Y. Kiso, *Chem. Pharm. Bull.*, **37**, 2570 (1989). c) Y. Ito, T. Kamijo, H. Harada, and S. Terashima, *Heterocycles*, **30**, 299 (1990). d) F. Matsuda, T. Matsumoto, M. Ohsaki, Y. Ito, and S. Terashima, *Chemistry Lett.*, submitted for publication.
- 4) H. Suda, T. Takita, T. Aoyagi, and H. Umezawa, *J. Antibiot.*, **29**, 100 (1976).
- 5) For previous syntheses of **4**, see, a) H. Suda, T. Takita, T. Aoyagi, and H. Umezawa, *J. Antibiot.*, **29**, 600 (1976). b) R. Nishizawa, T. Saino, T. Takita, H. Suda, T. Aoyagi, and H. Umezawa, *J. Med. Chem.*, **20**, 510 (1977). c) R. Nishizawa, T. Saino, M. Suzuki, T. Fujii, T. Shirai, T. Aoyagi, and H. Umezawa, *J. Antibiot.*, **36**, 695 (1983).
- 6) Representative physical data of the synthetic intermediates are as follows. **6**: mp 40-41°C, $[\alpha]_D^{20}$ -50.0° (c=1.04, CHCl₃); **8**: oil; **10a**: caramel; **11a**: caramel, $[\alpha]_D^{20}$ +32.4° (c=1.11, CHCl₃); **12a**: mp 96-97°C, $[\alpha]_D^{20}$ +2.5° (c=1.22, CHCl₃); **13**: oil; **15**: oil, $[\alpha]_D^{20}$ +99.7° (c=1.62, CHCl₃); **17**: mp 112.5-113°C, $[\alpha]_D^{20}$ -32.7° (c=1.04, CHCl₃).
- 7) The precise reaction mechanism which may rationalize the preferential formations of **11a-d** is quite ambiguous. However, as previously suggested for the similar [2+2]-cycloaddition reaction,⁸⁾ the observed results may be accounted for by initial formation of the zwitter-ionic intermediates (**18**) from **10a-d** and benzyloxyketene and subsequent conrotatory ring closure to the indicated direction under an influence of the adjacent chiral center.
- 8) a) H. G. Holden, "Chemistry and Biochemistry of β -Lactam Antibiotics," Vol. 2, ed. R. B. Morin and M. Gorman, Academic Press, Inc., New York-London, p. 114 (1982). b) C. Hubschwerlen and G. Schmidt, *Helv. Chim. Acta*, **66**, 2206 (1983).
- 9) The coupling constants of the C₃- and C₄-protons in the 2-azetidinone rings of **11a-d** and **12a-d** were found to be 4.9-5.3 Hz. Since C_{3,4}-*trans*-2-azetidinone derivatives regularly exhibit J_{3,4} = -2.5 Hz,¹⁰⁾ these spectral characteristics clearly suggest that both **11a-d** and **12a-d** bears the C_{3,4}-*cis* stereochemistries.
- 10) Y. Ito, Y. Kobayashi, T. Kawabata, M. Takase, and S. Terashima, *Tetrahedron*, **45**, 5767 (1989).
- 11) That both **11a** and **12a** has the C_{3,4}-*cis* stereochemistry was further ascertained by the following chemical correlation. Thus, desilylation of **11a** [Bu₄NF, THF, 99%] followed by Swern oxidation of the alcohol (**19**) [(COCl)₂-DMSO-Et₃N, CH₂Cl₂, 100%] and reduction of the ketone (**20**) [NaBH₄, MeOH-CH₂Cl₂, 87%] gave a mixture of the diastereomeric alcohols (**19** and **21**, 19:21=1:10). The predominantly produced alcohol (**21**), $[\alpha]_D^{20}$ +26.1° (c=1.31, CHCl₃), was found to be enantiomeric to the alcohol, $[\alpha]_D^{20}$ -26.9° (c=1.20, CHCl₃), independently prepared by desilylation of **12a** [Bu₄NF, THF, 71%].
- 12) The hydrochloride of **4** showed mp 190-192°C (decomp.) [*lit.*,^{3d)} mp 190°C (decomp.)].



	X	Y
11a :	OTBDMS	H
19 :	OH	H
20 :	—O—	
21 :	H	OH