



Palladium-catalyzed highly regioselective 2-arylation of 2,*x*-dibromopyridines and its application in the efficient synthesis of a 17 β -HSD1 inhibitor

Qizhong Zhou^{a,b,*}, Bin Zhang^a, Liangjun Su^a, Tiansheng Jiang^c, Renner Chen^{a,*},
Tieqi Du^a, Yuyuan Ye^a, Jianfen Shen^a, Guoliang Dai^a, Deman Han^a, Huajiang Jiang^{a,*}

^a Department of Chemistry, Taizhou University, Taizhou 318000, China

^b Department of Chemistry, Fudan University, Shanghai 200433, China

^c College of Natural Sciences and Mathematics, University of Houston, Houston, TX 77204-5008, USA

ARTICLE INFO

Article history:

Received 20 August 2013

Received in revised form 30 September 2013

Accepted 14 October 2013

Available online 26 October 2013

ABSTRACT

2,3- and 2,5-Dibromopyridines reacted with arylboronic acids, catalyzed by Pd(OAc)₂/PPh₃ in the presence of K₂CO₃ in CH₃CN/MeOH (2:1) at 50 °C for 24 h, to afford 2-arylpyridines in good to high yields, while 2,4-dibromopyridine reacted with arylboronic acid pinacol esters, catalyzed by Pd(OAc)₂/PPh₃ in the presence of KOH in CH₃CN at 70 °C for 24 h, to afford 2-arylpyridines in good to high yields. To expand this methodology, a 17 β -HSD1 inhibitor was synthesized in good yield.

© 2013 Elsevier Ltd. All rights reserved.

1. Introduction

Pyridines are significant components of many drugs and natural products.^{1–3} 2-Arylpyridines are widely used as substrates for C–H activation.⁴ Substituted 2-arylpyridines are found in a wide range of fluorescent probes,⁵ organic conjugated materials,⁶ metal complexing ligands,⁷ pharmaceuticals,⁸ and natural products.⁹ (Fig. 1). Substituted 2-arylpyridines can be prepared by selective Suzuki coupling from 2,*x*-dihalopyridines, such as 2,5-,¹⁰ 2,4-,¹¹ or 2,3-dibromopyridines,¹² and arylboronic acids followed by other reactions, but the yields are usually low and high temperature may be required.^{12b} Previously, we have reported the copper-catalyzed highly regioselective 2-aryloxylation of 2,*x*-dihalopyridines.¹³ Herein, we describe a highly regioselective and efficient palladium-catalyzed arylation of 2,*x*-dibromopyridines and its application in the efficient synthesis of a 17 β -HSD1 inhibitor.

2. Results and discussion

First, a set of experiments was performed using 2,5-dibromopyridine and phenylboronic acid as model substrates in the presence of different ligands or bases in different temperatures to explore the optimized reaction conditions (Table 1). In MeOH with PPh₃ as ligand and K₂CO₃ as base, the reaction afforded **3a**, **A**, and **B** in 73%, 4%, and 23% isolated yields, respectively (entry 3),

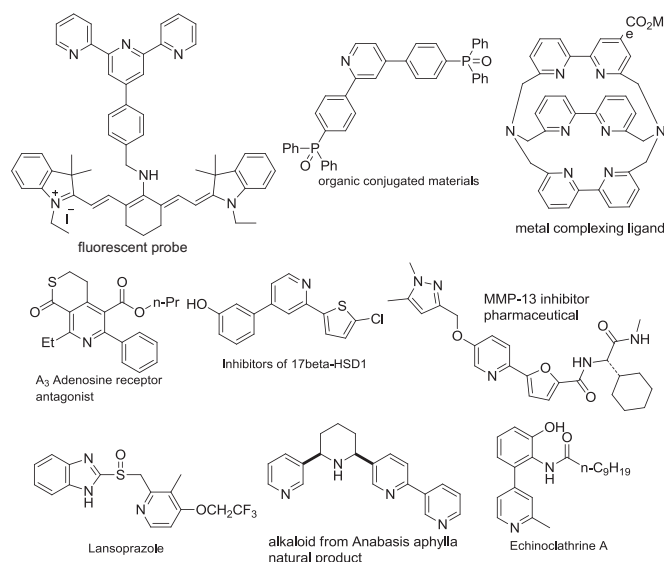


Fig. 1. The substituted 2-pyridyl motif is found in many important small molecules.

* Corresponding authors. Tel.: +86 576 88660177; e-mail addresses: qizhong-chou@yahoo.com (Q. Zhou), cre@tzc.edu.cn (R. Chen), jhj@tzc.edu.cn (H. Jiang).

while in MeCN a better selectivity was achieved (entry 4). Encouraged by these results, the reaction was investigated in a mixture of acetonitrile and alcohol. In MeCN/EtOH (2:1) at 50 °C, the three products were obtained in 93%, 2%, and 0% yields, respectively (entry 19), and in MeCN/MeOH (2:1), the yield of **3a** could be further improved to 97% (entry 22). We thus established the optimized

Table 1
Palladium-catalyzed 2-arylation of 2,5-dibromopyridine with phenylboronic acid: parametric study^{a,b}

Reaction scheme showing the 2-arylation of 2,5-dibromopyridine (**1**) with phenylboronic acid (**2**) to form 2-phenyl-5-bromopyridine (**3a**) and 2-phenyl-2,5-dibromopyridine (**A**) and 2-phenyl-3,5-dibromopyridine (**B**). The reaction conditions are Pd(OAc)₂ (5 mol %), ligand (10 mol %), base (2 eq), N₂, T, 24 h.

Entry	Ligand	Base	Solvent	T (°C)	Yield (%) (3a / A / B)
1 ^c	PPh ₃	K ₂ CO ₃	ClCH ₂ CH ₂ Cl	Reflux	30:12:10
2 ^c	PPh ₃	K ₂ CO ₃	THF	Reflux	38:7:36
3 ^c	PPh ₃	K ₂ CO ₃	MeOH	Reflux	73:4:23
4 ^c	PPh ₃	K ₂ CO ₃	CH ₃ CN	Reflux	41:8:0
5 ^c	PPh ₃	K ₂ CO ₃	Toluene	Reflux	43:8:5
6 ^c	PPh ₃	K ₂ CO ₃	DME	Reflux	50:10:24
7 ^c	PPh ₃	K ₂ CO ₃	Dioxane	Reflux	0:0:0
8 ^c	PPh ₃	K ₂ CO ₃	DMAc	110	40:9:18
9 ^c	PPh ₃	K ₂ CO ₃	DMSO	110	26:7:8
10 ^c	PPh ₃	K ₂ CO ₃	DMF	110	41:13:12
11		K ₂ CO ₃	MeOH	30	54:22:12
12	PPh ₃	K ₂ CO ₃	MeOH	30	67:8:18
13	Sphos	K ₂ CO ₃	MeOH	30	15:5:50
14	Xphos	K ₂ CO ₃	MeOH	30	24:13:10
15	BINAP	K ₂ CO ₃	MeOH	30	53:9:3
16	Xantphos	K ₂ CO ₃	MeOH	30	0:0:0
17	PPh ₃	K ₂ CO ₃	CH ₃ CN/EtOH (1:1)	30	75:8:3
18	PPh ₃	K ₂ CO ₃	CH ₃ CN/EtOH (2:1)	40	80:0:0
19	PPh ₃	K ₂ CO ₃	CH ₃ CN/EtOH ^d	50	93:2:0
20 ^e	PPh ₃	K ₂ CO ₃	CH ₃ CN/EtOH ^d	50	85:3:0
21	PPh ₃	K ₂ CO ₃	CH ₃ CN/MeOH ^d	50	91:1:7
22 ^f	PPh₃	K₂CO₃	CH₃CN/MeOH^d	50	97:2:0
23	PPh ₃	K ₂ CO ₃	CH ₃ CN/ <i>n</i> -BuOH ^d	50	85:2:9
24	PPh ₃	K ₂ CO ₃	CH ₃ CN/ <i>i</i> -BuOH ^d	50	65:3:1
25	PPh ₃	KOH	CH ₃ CN/MeOH ^d	50	73:13:12
26	PPh ₃	K ₃ PO ₄	CH ₃ CN/MeOH ^d	50	86:12:0
27	PPh ₃	Cs ₂ CO ₃	CH ₃ CN/MeOH ^d	50	90:5:5

The bold signifies the best reaction condition.

^a Reaction conditions: **1** (0.5 mmol), **2** (0.6 mmol), Pd(OAc)₂ (5 mol %), ligand (10 mol %), base (1 mmol), 24 h.

^b Isolated yield.

^c Compounds **1** (0.6 mmol), **2** (0.5 mmol), Pd(OAc)₂ (5 mol %), ligand (10 mol %), base (1 mmol), 24 h.

^d CH₃CN/alcohol (2:1).

^e Compounds **1** (0.5 mmol), **2** (0.6 mmol), Pd(OAc)₂ (2.5 mol %), ligand (5 mol %), base (1 mmol), 24 h.

^f Compounds **1** (0.5 mmol), **2** (0.55 mmol), Pd(OAc)₂ (5 mol %), PPh₃ (10 mol %), K₂CO₃ (1 mmol), 24 h.

conditions for this reaction: Pd(OAc)₂/PPh₃ as the catalysis system, K₂CO₃ as the base, CH₃CN/MeOH (2:1) as the solvent and at 50 °C.

With the optimized reaction condition available, the scope of the reaction was then investigated (Table 2). It can be found that phenylboronic acids bearing electron-rich groups generally afforded products (**3a–3e**, **3i–3m**, **3r–3t**) in good to high yields (80–97%). 4-Fluorophenylboronic acid also gave rise to good to high yields (82–90%, **3f** and **3n**). Even 4-methoxyphenylboronic acid reacted with 2,6-dibromopyridine to give a good yield (73%, **3x**). Furthermore, substituted 2,3-dichloropyridines also reacted with phenylboronic acid to give good yields (82–88%, **3u,3v**). In contrast, electron-withdrawing CF₃- or NO₂-substituted phenylboronic acid gave low yields (21–50%, **3g**, **3o,3p**) and 4-pyridylboronic acid did not afford any products (entries 8 and 17). Clearly, the new methodology has the advantages of higher regioselectivity, higher yields, more generality, simpler and cheaper catalyst system.^{10,12}

The reaction of 2,4-dibromopyridine and phenylboronic acid only gave a 68% yield of the product (**3w**). To further improve the yield of the reaction, phenylboronic acid pinacol ester was used as the coupling partner (Table 3).¹⁴ After an extensive investigation of the reaction conditions, we found that with KOH as base, the reaction could produce the required product selectively in 16% yield in MeCN at 30 °C (entry 3). Further screening of the ligands revealed that PPh₃ is still a suitable ligand leading to the highest yield of **3w**

(89%) and the best regioselectivity at 70 °C. The yields for the side products **C** and **D** were as low as 6% and 2%, respectively (entry 14).

After the optimized reaction condition was established, the reactions of 2,4-dibromopyridine with other phenylboronic acid pinacol esters were investigated and the results are shown in Table 4. Again, the pinacol esters bearing electron-donating groups afforded good to high yields (81–95%, entries 1–8), while the nitro-substituted substrate only gave the coupling product in 31% yield (entry 9) and 2-cyano-5-pyridylboronic acid pinacol ester did not give rise to the required product.

Why the regioselectivity is so high for the arylation of 2,3-, 2,5-, and 2,4-dibromopyridines? We can explain that the 2-bromo of 2,*x*-dibromopyridines is more reactive than the 4-bromo, 5-bromo, and 3-bromo of 2,*x*-dibromopyridines.^{11a,13,15}

The high regioselectivity of the new method made it possible to conveniently conduct the step-by-step coupling reactions for 2,*x*-dibromopyridines. To demonstrate this, we studied the Suzuki, Sonogashira, Ullmann, and Ullmann-typed coupling reactions of **3w** with different substrates (Scheme 1), which all afforded the coupling products **6–9** in excellent yields.

The method can also be expanded to prepare compound **11**, a 17β-HSD1 inhibitor (Scheme 2). 2,4-Dibromopyridine reacted with 2-chlorothiophene pinacboronate under the standard condition to give compound **10** in 82% yield and its coupling with 3-hydroxyphenylboronic acid produced **11** in 90% yield. The total yield (74%) is substantially higher than the route previously reported (12%).^{8b}

3. Conclusion

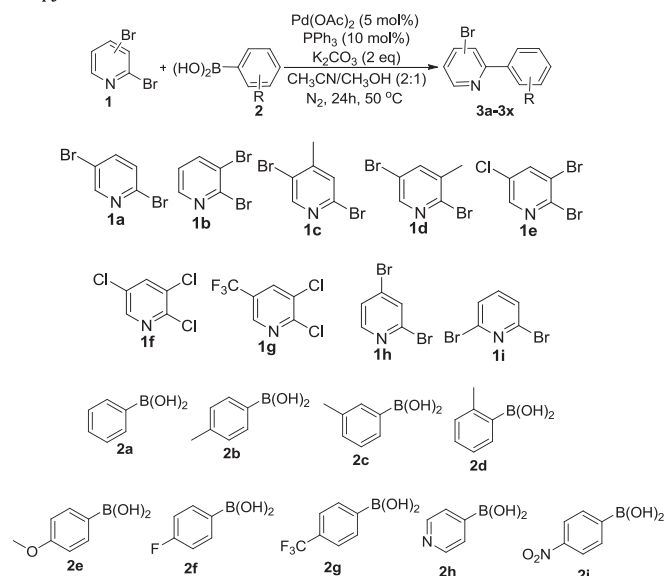
In summary, we have described a highly regioselective approach to bromo-2-arylpyridines, which involves palladium-catalyzed direct 2-arylation of 2,5-, 2,3-, and 2,4-dibromopyridines in high yields. The practicality of the new method has been demonstrated by using it to synthesize a 17β-HSD1 inhibitor. Step-by-step coupling reaction of other dihalopyridines is being investigated and the results will be reported in due course.

4. Experimental section

4.1. General

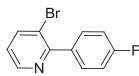
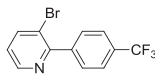
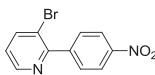
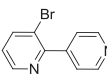
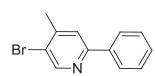
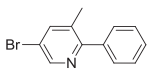
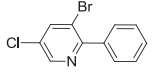
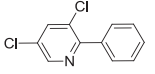
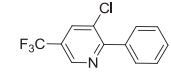
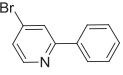
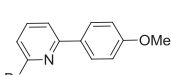
Typical procedure for the 2-arylation of 2,5- and 2,3-dibromopyridines: 2,5-dibromopyridine (0.12 g, 0.50 mmol), phenylboronic acid (67 mg, 0.55 mmol), K₂CO₃ (0.14 g, 1.0 mmol), Pd(OAc)₂ (11 mg, 5 mol %), PPh₃ (26 mg, 10 mol %) were dissolved in CH₃CN/CH₃OH (2:1, 6 mL). The solution was stirred at 50 °C under nitrogen atmosphere for 24 h and then cooled and the solid was filtered off. The filtrate was then concentrated and the resulting crude product was dissolved in CH₂Cl₂ (10 mL). The solution was washed with water (10 mL×3) and brine (10 mL), and dried over sodium sulfate. Upon removal of the solvent with a rotavapor, the resulting residue was subjected to column chromatography (petroleum ether/AcOEt, 400:1) to give the desired product **3a** (114 mg, 97%) as a white solid.

Typical procedure for the 2-arylation of 2,4-dibromopyridine: 2,4-dibromopyridine (0.12 g, 0.50 mmol), phenylboronic acid pinacol ester (0.11 g, 0.55 mmol), KOH (56 mg, 1.0 mmol), Pd(OAc)₂ (11 mg, 5 mol %), PPh₃ (52 mg, 20 mol %) were dissolved in CH₃CN (6 mL). The reaction was stirred at 70 °C under nitrogen atmosphere for 24 h and then cooled. The solid was filtrated off and the filtrate was concentrated. The crude product was then dissolved in CH₂Cl₂ (10 mL) and the solution was washed with water (10 mL×3) and brine (10 mL), and dried over sodium sulfate. Upon evaporation, the resulting residue was subjected to column

Table 2Scope of Pd-catalyzed 2-arylation of 2,x-dibromopyridines^{a,b}

Entry	1	2	Products	Yield (%)
1	1a	2a	3a	97
2	1a	2b	3b	91
3	1a	2c	3c	91
4	1a	2d	3d	94
5	1a	2e	3e	85
6	1a	2f	3f	90
7	1a	2g	3g	50
8	1a	2h	3h	0
9	1b	2a	3i	95
10	1b	2b	3j	83
11	1b	2c	3k	94
12	1b	2d	3l	93
13	1b	2e	3m	80

Table 2 (continued)

Entry	1	2	Products	Yield (%)
14	1b	2f	 3n	82
15	1b	2g	 3o	46
16	1b	2i	 3p	21
17	1b	2h	 3q	0
18	1c	2a	 3r	82
19	1d	2a	 3s	88
20	1e	2a	 3t	90
21	1f	2a	 3u	86
22	1g	2a	 3v	96
23	1h	2a	 3w	68
24	1i	2e	 3x	73

^a Reaction condition: **1** (0.5 mmol), **2** (0.55 mmol), Pd(OAc)₂ (5 mol %), PPh₃ (10 mol %), K₂CO₃ (1 mmol) in CH₃CN/CH₃OH (2:1, 6 mL) at 50 °C under N₂ for 24 h.

^b Isolated yield.

chromatography (petroleum ether/AcOEt, 400:1) to give **3w** (104 mg, 89%) as a colorless liquid.

4.1.1. 5-Bromo-2-phenylpyridine (3a). Compound **3a**: white solid, mp 74–75 °C; ¹H NMR (CDCl₃, 400 MHz): 8.74 (d, *J*=1.6 Hz, 1H), 7.95–7.97 (m, 2H), 7.87 (dd, *J*₁=2.4 Hz, *J*₂=2.8 Hz, 1H), 7.62 (d, *J*=8.4 Hz, 1H), 7.43–7.49 (m, 3H); ¹³C NMR (CDCl₃, 100 MHz): 156.1, 150.9, 139.7, 138.3, 129.7, 129.2, 127.1, 122.0, 119.6; MS (ESI): 234.1 [M+H]⁺; HRMS (ESI) [M+H]⁺ calcd for C₁₁H₉BrN: 233.99129, found: 233.99171.

4.1.2. 2-Bromo-5-phenylpyridine (A). Compound **A**: white solid, mp 76–77 °C; ¹H NMR (CDCl₃, 400 MHz): 8.57 (d, *J*=2.4 Hz, 1H), 7.71 (dd, *J*₁=2.4 Hz, *J*₂=2.4 Hz, 1H), 7.39–7.54 (m, 6H); ¹³C NMR (CDCl₃, 100 MHz): 148.7, 141.1, 137.2, 136.7, 136.2, 129.5, 128.7, 128.2, 127.2; MS (EI): 233.0 (M⁺); HRMS (EI) calcd for C₁₁H₈BrN: 232.9840, found: 232.9846.

4.1.3. 2,5-Diphenylpyridine (B). Compound **B**: white solid, mp 173–174 °C; ¹H NMR (CDCl₃, 400 MHz): 8.93 (d, *J*=2.0 Hz, 1H), 8.04 (d, *J*=7.2 Hz, 2H), 7.91 (dd, *J*₁=2.0 Hz, *J*₂=2.0 Hz, 1H), 7.77 (d, *J*=8.0 Hz, 1H), 7.61 (d, *J*=7.2 Hz, 2H), 7.37–7.49 (m, 6H); ¹³C NMR (CDCl₃, 100 MHz): 156.4, 148.3, 139.2, 137.9, 135.3, 135.1, 129.3,

129.2, 129.0, 128.3, 127.2, 127.1, 120.6; MS (EI): 231.0 (M⁺); HRMS (EI) calcd for C₁₇H₁₃N: 231.1048, found: 231.1053.

4.1.4. 5-Bromo-2-(4-methylphenyl)pyridine (3b). Compound **3b**: white solid, mp 109–110 °C; ¹H NMR (CDCl₃, 400 MHz): 8.70 (d, *J*=1.6 Hz, 1H), 7.90–7.86 (m, 3H), 7.57 (d, *J*=8.4 Hz, 1H), 7.26 (d, *J*=8.0 Hz, 2H), 2.39 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): 156.1, 150.8, 139.7, 139.4, 135.7, 129.8, 126.9, 121.5, 119.1, 21.5; MS (ESI): 248.0 [M+H]⁺; HRMS (ESI) [M+H]⁺ calcd for C₁₂H₁₁BrN: 248.0069, found: 248.0067.

4.1.5. 5-Bromo-2-(3-methylphenyl)pyridine (3c). Compound **3c**: white solid, mp 40–41 °C; ¹H NMR (CDCl₃, 400 MHz): 8.72 (d, *J*=2.4 Hz, 1H), 7.70–7.84 (m, 3H), 7.58 (dd, *J*₁=2.4 Hz, *J*₂=2.4 Hz, 1H), 7.34 (t, *J*=7.6 Hz, 1H), 7.23 (d, *J*=7.6 Hz, 1H), 2.42 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): 156.3, 150.7, 139.6, 138.4, 138.3, 130.4, 129.0, 127.7, 124.2, 122.0, 119.5, 21.8; MS (ESI): 248.1 [M+H]⁺; HRMS (ESI) [M+H]⁺ calcd for C₁₂H₁₁BrN: 248.0069, found: 248.0058.

4.1.6. 5-Bromo-2-(2-methylphenyl)pyridine (3d). Compound **3d**: yellow liquid; ¹H NMR (CDCl₃, 400 MHz): 8.75 (d, *J*=2.4 Hz, 1H), 7.85 (dd, *J*₁=2.8 Hz, *J*₂=2.4 Hz, 1H), 7.24–7.37 (m, 5H), 2.35 (s, 3H); ¹³C

Table 3
Pd-catalyzed 2-arylation of 2,4-dibromopyridine: parametric study^{a,b}

Entry	Catalyst	Ligand	T/°C	Yield (%) (3w / C / D)
1	Pd(PPh ₃) ₄		30	14:0:0
2	Pd(dba) ₃	PPh ₃	30	25:4:0
3	Pd(OAc) ₂	PPh ₃	30	16:0:0
4	Pd(dppe) ₂ Cl ₂	PPh ₃	30	28:14:7
5	Pd(TFA) ₂	PPh ₃	30	37:13:9
6	Pd(OAc) ₂	1038-95-5	Reflux	27:18:0
7	Pd(OAc) ₂	855-38-9	Reflux	44:20:0
8	Pd(OAc) ₂	37943-90-1	Reflux	65:2:9
9	Pd(OAc) ₂	5518-53-5	Reflux	56:7:10
10	Pd(OAc) ₂	18437-78-0	Reflux	40:5:0
11	Pd(OAc) ₂	13406-29-6	Reflux	56:4:8
12	Pd(OAc) ₂	1259-35-4	Reflux	0:0:0
13	Pd(OAc) ₂	PPh ₃	Reflux	88:6:2
14	Pd(OAc) ₂	PPh ₃	70	89:6:2
15	Pd(OAc) ₂	PPh ₃	65	77:13:0
16	Pd(OAc) ₂	PPh ₃	60	38:7:0

The bold signifies the best reaction condition.

^a Compounds **1h** (0.5 mmol), **4a** (0.55 mmol), [Pd] (5 mol %), ligand (20 mol %), KOH (1 mmol), CH₃CN, 24 h.

^b Isolated yield. 1038-95-5: Tri-*p*-tolylphosphine; 855-38-9: Tris(4-methoxyphenyl)phosphine; 37943-90-1: diphenyl-2-pyridylphosphine; 5518-53-5: Tri(2-furyl) phosphine; 18437-78-0: Tris(*p*-fluorophenyl) phosphine; 13406-29-6: Tris(*p*-trifluoromethylphenyl) phosphine; 1259-35-4: Tris(pentafluorophenyl) phosphine.

NMR (CDCl₃, 100 MHz): 158.6, 150.3, 139.2, 139.1, 136.0, 131.1, 129.7, 128.9, 126.2, 125.6, 119.2, 20.5; MS (ESI): 248.0 [M+H]⁺; HRMS (ESI) [M+H]⁺ calcd for C₁₂H₁₁BrN: 248.0069, found: 248.0058.

4.1.7. 5-Bromo-2-(4-methoxyphenyl)pyridine (3e). Compound **3e**: white solid, mp 136–137 °C; ¹H NMR (CDCl₃, 400 MHz): 8.69 (d, *J*=2.0 Hz, 1H), 7.92 (d, *J*=8.4 Hz, 2H), 7.82 (dd, *J*₁=2.4 Hz, *J*₂=2.0 Hz, 1H), 7.55 (d, *J*=8.4 Hz, 1H), 6.98 (d, *J*=8.8 Hz, 2H), 3.86 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): 161.1, 155.7, 150.5, 139.7, 130.8, 128.4, 121.2, 118.7, 114.5, 55.7; MS (ESI): 264.0 [M+H]⁺; HRMS (ESI) [M+H]⁺ calcd for C₁₂H₁₁BrNO: 264.0019, found: 264.0012.

4.1.8. 5-Bromo-2-(4-fluorophenyl)pyridine (3f). Compound **3f**: white solid, mp 73–74 °C; ¹H NMR (CDCl₃, 400 MHz): 8.71 (d, *J*=2.4 Hz, 1H), 7.92–7.96 (m, 2H), 7.84 (dd, *J*₁=2.4 Hz, *J*₂=2.0 Hz, 1H), 7.55 (d, *J*=8.4 Hz, 1H), 7.14 (t, *J*=8.8 Hz, 2H); ¹³C NMR (CDCl₃, 100 MHz): 164.0 (d, *J*=247.6 Hz), 155.1, 150.9, 139.6, 134.6 (d, *J*=3.2 Hz), 128.9 (d, *J*=9.2 Hz), 121.5, 119.5, 116.1 (d, *J*=20.9 Hz); MS (ESI): 252.0 [M+H]⁺; HRMS (ESI) [M+H]⁺ calcd for C₁₁H₈BrFN: 251.9819, found: 251.9817.

4.1.9. 5-Bromo-2-(4-trifluoromethylphenyl)pyridine (3g). Compound **3g**: white solid, mp 109–110 °C; ¹H NMR (CDCl₃, 400 MHz): 8.76 (d, *J*=1.6 Hz, 1H), 8.07 (d, *J*=8.0 Hz, 2H), 7.90 (dd, *J*₁=2.0 Hz, *J*₂=2.4 Hz, 1H), 7.68 (dd, *J*₁=8.4 Hz, *J*₂=8.4 Hz, 3H); ¹³C NMR (CDCl₃, 100 MHz): 154.5, 151.3, 141.8, 139.8, 131.5 (q, *J*=33.2 Hz), 127.3, 126.1 (q, *J*=3.6 Hz), 124.4 (d, *J*=270.4 Hz), 122.1, 120.6; MS (EI): 303.0 (M⁺); HRMS (EI) calcd for C₁₂H₇BrF₃N: 300.9714, found: 300.9711.

4.1.10. 3-Bromo-2-phenylpyridine (3i). Compound **3i**: yellow liquid; ¹H NMR (CDCl₃, 400 MHz): 8.62 (dd, *J*₁=1.2 Hz, *J*₂=1.2 Hz, 1H), 7.98 (dd, *J*₁=1.6 Hz, *J*₂=1.6 Hz, 1H), 7.68 (dd, *J*₁=1.6 Hz, *J*₂=1.2 Hz, 2H), 7.40–7.48 (m, 3H), 7.12 (dd, *J*₁=4.8 Hz, *J*₂=4.8 Hz, 1H); ¹³C NMR (CDCl₃, 100 MHz): 158.3, 148.2, 141.6, 139.6, 129.5, 129.0, 128.2,

Table 4
Scope of Pd-catalyzed 2-arylation of 2,4-dibromopyridine^{a,b}

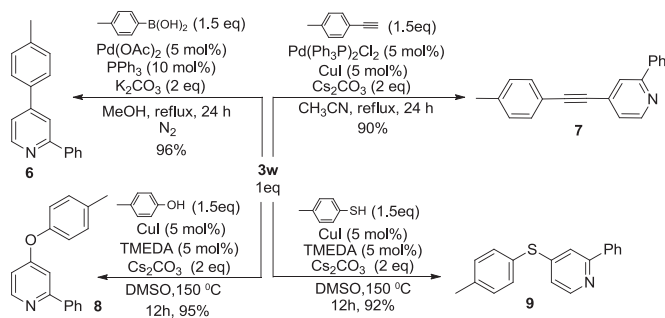
Entry	4	Products	Yield (%)
1	4a	3w	89
2	4b	5a	91
3	4c	5b	86
4	4d	5c	95
5	4e	5d	84
6	4f	5e	93
7	4g	5f	89
8	4h	5g	81
9	4i	5h	31
10	4j	5i	0

^a Compound **1h** (0.5 mmol), compound **4** (0.55 mmol), Pd(OAc)₂ (5 mol %), PPh₃ (20 mol %), KOH (1 mmol), CH₃CN, 24 h.

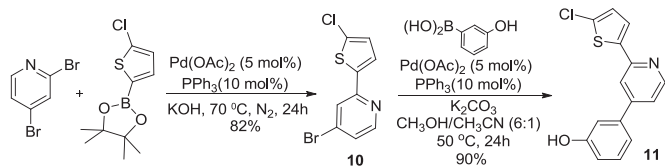
^b Isolated yield.

123.5, 122.1; MS (EI): 233.0 (M⁺); HRMS (EI) calcd for C₁₁H₈BrN: 232.9840, found: 232.9844.

4.1.11. 3-Bromo-2-(4-methylphenyl)pyridine (3j). Compound **3j**: yellow liquid; ¹H NMR (CDCl₃, 400 MHz): 8.60 (dd, *J*₁=1.2 Hz, *J*₂=1.6 Hz, 1H), 7.96 (dd, *J*₁=1.6 Hz, *J*₂=1.6 Hz, 1H), 7.59 (d, *J*=8.0 Hz,



Scheme 1. Application of the 4-bromo-2-phenylpyridine to the other reactions.



Scheme 2. Application of the 2,4-dibromopyridine to the inhibitor of 17β-HSD1 (**11**).

2H), 7.26 (d, $J=7.6$ Hz, 2H), 7.08–7.11 (m, 1H); 2.41 (s, 3H); ^{13}C NMR (CDCl_3 , 100 MHz): 158.4, 148.2, 141.6, 139.0, 136.8, 129.5, 128.9, 123.3, 120.1, 21.6; MS (EI): 247.0 (M^+); HRMS (EI) calcd for $\text{C}_{12}\text{H}_{10}\text{BrN}$: 246.9997, found: 246.9998.

4.1.12. 3-Bromo-2-(3-methylphenyl)pyridine (3k). Compound **3k**: yellow liquid; ^1H NMR (CDCl_3 , 400 MHz): 8.61 (dd, $J_1=1.2$ Hz, $J_2=1.2$ Hz, 1H), 7.96 (dd, $J_1=1.2$ Hz, $J_2=1.2$ Hz, 1H), 7.48 (d, $J=6.8$ Hz, 2H), 7.34 (t, $J=7.6$ Hz, 1H), 7.24 (d, $J=8.0$ Hz, 1H), 7.11 (dd, $J_1=4.8$ Hz, $J_2=4.8$ Hz, 1H), 2.42 (s, 3H); ^{13}C NMR (CDCl_3 , 100 MHz): 158.5, 148.1, 141.5, 139.6, 137.9, 130.1, 129.7, 128.0, 126.6, 123.4, 120.1, 21.7; MS (EI): 247.0 (M^+); HRMS (EI) calcd for $\text{C}_{12}\text{H}_{10}\text{BrN}$: 246.9997, found: 247.0001.

4.1.13. 3-Bromo-2-(2-methylphenyl)pyridine (3l). Compound **3l**: yellow liquid; ^1H NMR (CDCl_3 , 400 MHz): 8.62 (dd, $J_1=1.2$ Hz, $J_2=1.2$ Hz, 1H), 7.98 (dd, $J_1=0.8$ Hz, $J_2=1.2$ Hz, 1H), 7.15–7.35 (m, 5H), 2.14 (s, 3H); ^{13}C NMR (CDCl_3 , 100 MHz): 159.8, 148.0, 140.7, 139.9, 135.9, 130.4, 128.9, 128.8, 125.8, 123.7, 121.6, 19.6; MS (EI): 247.0 (M^+); HRMS (EI) calcd for $\text{C}_{12}\text{H}_{10}\text{BrN}$: 246.9997, found: 246.9996.

4.1.14. 3-Bromo-2-(4-methoxyphenyl)pyridine (3m). Compound **3m**: yellow solid, mp 61–62 °C; ^1H NMR (CDCl_3 , 400 MHz): 8.59 (dd, $J_1=1.2$ Hz, $J_2=1.2$ Hz, 1H), 7.96 (dd, $J_1=1.2$ Hz, $J_2=1.6$ Hz, 1H), 7.66–7.69 (m, 2H), 7.08 (dd, $J_1=4.8$ Hz, $J_2=4.8$ Hz, 1H), 6.98 (d, $J=8.4$ Hz, 2H), 3.85 (s, 3H); ^{13}C NMR (CDCl_3 , 100 MHz): 160.3, 157.9, 148.1, 141.7, 132.1, 131.0, 123.0, 119.9, 113.6, 55.5; MS (EI): 263.0 (M^+); HRMS (EI) calcd for $\text{C}_{12}\text{H}_{10}\text{BrNO}$: 262.9946, found: 262.9950.

4.1.15. 3-Bromo-2-(4-fluorophenyl)pyridine (3n). Compound **3n**: white solid, mp 90–91 °C; ^1H NMR (CDCl_3 , 400 MHz): 8.61 (dd, $J_1=1.2$ Hz, $J_2=1.2$ Hz, 1H), 7.98 (dd, $J_1=1.2$ Hz, $J_2=1.2$ Hz, 1H), 7.67–7.71 (m, 2H), 7.11–7.17 (m, 3H); ^{13}C NMR (CDCl_3 , 100 MHz): 163.2 (d, $J=247.3$ Hz), 157.3, 148.3, 141.7, 135.7 (d, $J=3.2$ Hz), 131.6 (d, $J=8.2$ Hz), 123.6, 120.0, 115.2 (d, $J=22.0$ Hz); MS (EI): 251.0 (M^+); HRMS (EI) calcd for $\text{C}_{11}\text{H}_7\text{BrNF}$: 250.9746, found: 250.9744.

4.1.16. 3-Bromo-2-(4-trifluoromethylphenyl)pyridine (3o). Compound **3o**: white solid, mp 51–52 °C; ^1H NMR (CDCl_3 , 400 MHz): 8.64 (dd, $J_1=1.2$ Hz, $J_2=1.2$ Hz, 1H), 8.02 (dd, $J_1=1.2$ Hz, $J_2=1.2$ Hz, 1H), 7.68 (dd, $J_1=8.0$ Hz, $J_2=8.4$ Hz, 4H), 7.19 (dd, $J_1=4.4$ Hz, $J_2=4.4$ Hz, 1H), ^{13}C NMR (CDCl_3 , 100 MHz): 157.0, 148.4, 143.1, 141.9, 131.0 (q, $J=32.4$ Hz), 130.1, 125.3 (q, $J=3.6$ Hz), 124.4 (d, $J=270.4$ Hz), 124.2, 120.0; MS (EI): 301.0 (M^+); HRMS (EI) calcd for $\text{C}_{12}\text{H}_7\text{BrF}_3\text{N}$: 300.9714, found: 300.9716.

4.1.17. 3-Bromo-2-(4-nitrophenyl)pyridine (3p). Compound **3p**: white solid, mp 148–149 °C; ^1H NMR (CDCl_3 , 400 MHz): 8.67 (d,

$J=4.4$ Hz, 1H), 8.32 (d, $J=6.8$ Hz, 2H), 8.04 (d, $J=8.4$ Hz, 1H), 7.88 (d, $J=7.6$ Hz, 2H), 7.23–7.27 (m, 1H); ^{13}C NMR (CDCl_3 , 100 MHz): 156.1, 148.7, 148.1, 145.8, 141.9, 130.8, 124.6, 123.5, 120.0, MS (EI): 278 (M^+); HRMS (EI) calcd for $\text{C}_{11}\text{H}_7\text{BrN}_2\text{O}_2$: 277.9691, found: 277.9690.

4.1.18. 5-Bromo-4-methyl-2-phenylpyridine (3r). Compound **3r**: colorless liquid; ^1H NMR (CDCl_3 , 400 MHz): 8.68 (s, 1H), 7.93–7.95 (s, 2H), 7.57 (s, 1H), 7.38–7.47 (m, 3H), 2.43 (s, 3H); ^{13}C NMR (CDCl_3 , 100 MHz): 156.3, 151.2, 147.7, 138.4, 129.4, 129.0, 127.0, 122.9, 122.3, 22.7; MS (EI): 249.1 (M^+); HRMS (EI) calcd for $\text{C}_{12}\text{H}_{10}\text{BrN}$: 246.9997, found: 246.9999.

4.1.19. 5-Bromo-3-methyl-2-phenylpyridine (3s). Compound **3s**: colorless liquid; ^1H NMR (CDCl_3 , 400 MHz): 8.58 (d, $J=2.0$ Hz, 1H), 7.73 (d, $J=2.4$ Hz, 1H), 7.39–7.50 (m, 5H), 2.34 (s, 3H); ^{13}C NMR (CDCl_3 , 100 MHz): 157.4, 148.1, 140.9, 139.7, 132.9, 129.0, 128.5, 119.1, 20.1; MS (EI): 248.1 (M^+); HRMS (EI) calcd for $\text{C}_{12}\text{H}_9\text{BrN}$: 245.9918, found: 245.9921.

4.1.20. 5-Chloro-3-bromo-2-phenylpyridine (3t). Compound **3t**: white solid, mp 60–61 °C; ^1H NMR (CDCl_3 , 400 MHz): 8.58 (d, $J=1.6$ Hz, 1H), 8.00 (d, $J=2.0$ Hz, 1H), 7.64–7.67 (m, 2H), 7.42–7.48 (m, 3H); ^{13}C NMR (CDCl_3 , 100 MHz): 156.6, 147.1, 140.7, 138.7, 130.7, 129.5, 129.3, 128.3, 119.6; MS (EI): 268.9 (M^+); HRMS (EI) calcd for $\text{C}_{11}\text{H}_7\text{BrClN}$: 266.9450, found: 266.9453.

4.1.21. 3,5-Dichloro-2-phenylpyridine (3u). Compound **3u**: white solid, mp 50–51 °C; ^1H NMR (CDCl_3 , 400 MHz): 8.55 (d, $J=2.4$ Hz, 1H), 7.80 (d, $J=2.0$ Hz, 1H), 7.68–7.71 (m, 2H), 7.42–7.48 (m, 3H); ^{13}C NMR (CDCl_3 , 100 MHz): 154.9, 146.7, 137.7, 137.3, 130.7, 130.4, 129.5, 129.4, 128.4; MS (EI): 223.0 (M^+); HRMS (EI) calcd for $\text{C}_{11}\text{H}_7\text{Cl}_2\text{N}$: 222.9956, found: 222.9954.

4.1.22. 3-Chloro-5-trifluoromethyl-2-phenylpyridine (3v). Compound **3v**: colorless liquid; ^1H NMR (CDCl_3 , 400 MHz): 8.85 (s, 1H), 8.04 (d, $J=1.6$ Hz, 1H), 7.74–7.77 (m, 2H), 7.48–7.52 (m, 3H); ^{13}C NMR (CDCl_3 , 100 MHz): 160.2, 144.5 (m), 137.2 (d, $J=2.2$ Hz), 135.6, 130.5, 130.0, 129.7, 128.5, 126.4 (q, $J=13.2$ Hz), 123.0 (d, $J=261.9$ Hz); MS (EI): 257.1 (M^+); HRMS (EI) calcd for $\text{C}_{12}\text{H}_7\text{ClF}_3\text{N}$: 257.0219, found: 257.0214.

4.1.23. 4-Bromo-2-phenylpyridine (3w). Compound **3w**: colorless liquid; ^1H NMR (CDCl_3 , 400 MHz): 8.48 (d, $J=5.2$ Hz, 1H), 7.87–7.96 (m, 3H), 7.36–7.48 (m, 4H); ^{13}C NMR (CDCl_3 , 100 MHz): 159.0, 150.5, 138.1, 133.8, 129.9, 129.1, 127.2, 125.5, 124.1; MS (EI): 233.1 (M^+); HRMS (EI) calcd for $\text{C}_{11}\text{H}_8\text{BrN}$: 232.9840, found: 232.9843.

4.1.24. 6-Bromo-2-(4-methoxyphenyl)pyridine (3x). Compound **3x**: white solid, mp 113–114 °C; ^1H NMR (CDCl_3 , 400 MHz): 7.91–7.94 (m, 2H), 7.50–7.59 (m, 2H), 7.31–7.33 (m, 1H), 6.94–6.97 (m, 2H), 3.84 (s, 3H); ^{13}C NMR (CDCl_3 , 100 MHz): 161.2, 158.5, 142.3, 139.1, 130.5, 128.6, 125.7, 118.4, 114.4, 55.6; MS (EI): 263.0 (M^+); HRMS (EI) calcd for $\text{C}_{12}\text{H}_{10}\text{BrNO}$: 262.9946, found: 262.9951.

4.1.25. 2-Bromo-4-phenylpyridine (C). Compound **C**: white solid, mp 62–63 °C; ^1H NMR (CDCl_3 , 400 MHz): 8.40 (d, $J=5.2$ Hz, 1H), 7.69 (d, $J=0.8$ Hz, 1H), 7.59 (dd, $J_1=2.0$ Hz, $J_2=1.6$ Hz, 2H), 7.44–7.51 (m, 4H); ^{13}C NMR (CDCl_3 , 100 MHz): 151.5, 150.6, 143.2, 137.0, 129.9, 129.5, 127.3, 126.1, 121.1; MS (EI): 233.0 (M^+); HRMS (EI) calcd for $\text{C}_{11}\text{H}_8\text{BrN}$: 232.9840, found: 232.9842.

4.1.26. 2,4-Diphenylpyridine (D). Compound **D**: yellow liquid; ^1H NMR (CDCl_3 , 400 MHz): 8.71 (d, $J=5.6$ Hz, 1H), 8.04 (d, $J=7.2$ Hz, 2H), 7.90 (s, 1H), 7.65–7.90 (m, 2H), 7.39–7.49 (m, 7H); ^{13}C NMR (CDCl_3 , 100 MHz): 158.3, 150.3, 149.5, 139.7, 138.8, 129.3 (d,

$J=9.0$ Hz), 129.0, 127.3 (d, $J=3.1$ Hz), 120.5, 119.0; MS (EI): 230.0 (M^+); HRMS (EI) calcd for $C_{17}H_{13}N$: 231.1048, found: 231.1045.

4.1.27. 4-Bromo-2-(4-methylphenyl)pyridine (5a). Compound **5a**: white solid, mp 60–61 °C; 1H NMR ($CDCl_3$, 400 MHz): 8.46 (d, $J=5.2$ Hz, 1H), 7.84–7.86 (m, 3H), 7.25–7.34 (m, 3H), 2.40 (s, 3H); ^{13}C NMR ($CDCl_3$, 100 MHz): 159.0, 150.4, 140.0, 135.3, 133.7, 129.8, 127.1, 125.2, 123.8, 21.5; MS (EI): 249.1 (M^+); HRMS (EI) calcd for $C_{12}H_{10}BrN$: 246.9997, found: 246.9999.

4.1.28. 4-Bromo-2-(3-methylphenyl)pyridine (5b). Compound **5b**: yellow liquid; 1H NMR ($CDCl_3$, 400 MHz): 8.48 (d, $J=5.2$ Hz, 1H), 7.87 (d, $J=1.6$ Hz, 1H), 7.79 (s, 1H), 7.72 (d, $J=7.6$ Hz, 1H), 7.33–7.37 (m, 2H), 7.24 (d, $J=7.6$ Hz, 1H), 2.42 (s, 3H); ^{13}C NMR ($CDCl_3$, 100 MHz): 159.2, 150.4, 138.8, 138.1, 133.8, 130.7, 129.0, 127.9, 125.4, 124.4, 124.2, 21.7; MS (EI): 247.1 (M^+); HRMS (EI) calcd for $C_{12}H_{10}BrN$: 246.9997, found: 246.9995.

4.1.29. 4-Bromo-2-(2-methylphenyl)pyridine (5c). Compound **5c**: colorless liquid; 1H NMR ($CDCl_3$, 400 MHz): 8.50 (d, $J=5.2$ Hz, 1H), 7.59 (d, $J=1.6$ Hz, 1H), 7.25–7.43 (m, 5H), 2.36 (s, 3H); ^{13}C NMR ($CDCl_3$, 100 MHz): 161.5, 150.0, 139.2, 136.1, 133.2, 131.1, 129.8, 129.1, 127.7, 126.2, 125.2, 20.5; MS (EI): 248.1 (M^+); HRMS (EI) calcd for $C_{12}H_9BrN$: 245.9918, found: 245.9920.

4.1.30. 4-Bromo-2-(4-ethylphenyl)pyridine (5d). Compound **5d**: colorless liquid; 1H NMR ($CDCl_3$, 400 MHz): 8.47 (d, $J=5.6$ Hz, 1H), 7.86–7.89 (m, 3H), 7.25–7.35 (m, 3H), 2.70 (q, $J=8.0$ Hz, 2H), 1.26 (t, $J=7.2$ Hz, 3H); ^{13}C NMR ($CDCl_3$, 100 MHz): 159.2, 150.5, 146.3, 135.7, 133.6, 128.7, 127.2, 125.2, 123.8, 28.9, 15.7; MS (EI): 261 (M^+); HRMS (EI) calcd for $C_{13}H_{12}BrN$: 261.0153, found: 261.0155.

4.1.31. 4-Bromo-2-(4-methoxyphenyl)pyridine (5e). Compound **5e**: yellow liquid; 1H NMR ($CDCl_3$, 400 MHz): 8.44 (d, $J=5.2$ Hz, 1H), 7.91 (d, $J=8.4$ Hz, 2H), 7.82 (s, 1H), 7.30–7.32 (m, 1H), 6.98 (d, $J=8.4$ Hz, 2H), 3.84 (s, 3H); ^{13}C NMR ($CDCl_3$, 100 MHz): 161.2, 158.7, 150.4, 133.6, 130.8, 128.6, 124.7, 123.3, 114.5, 55.6; MS (EI): 263.1 (M^+); HRMS (EI) calcd for $C_{12}H_{10}BrNO$: 262.9946, found: 262.9950.

4.1.32. 4-Bromo-2-(4-phenylphenyl)pyridine (5f). Compound **5f**: white solid, mp 105–106 °C; 1H NMR ($CDCl_3$, 400 MHz): 8.52 (d, $J=4.8$ Hz, 1H), 8.04–8.06 (m, 2H), 7.94 (d, $J=1.6$ Hz, 1H), 7.64–7.73 (m, 4H), 7.38–7.49 (m, 4H); ^{13}C NMR ($CDCl_3$, 100 MHz): 158.7, 150.7, 142.7, 140.6, 137.1, 133.8, 129.1, 128.0, 127.8, 127.7, 127.4, 125.5, 124.0; MS (EI): 309.0 (M^+); HRMS (EI) calcd for $C_{17}H_{12}BrN$: 309.0153, found: 309.0151.

4.1.33. 4-Bromo-2-(3-phenylphenyl)pyridine (5g). Compound **5g**: colorless liquid; 1H NMR ($CDCl_3$, 400 MHz): 8.52 (d, $J=5.2$ Hz, 1H), 8.19 (t, $J=1.6$ Hz, 1H), 7.92–7.95 (m, 2H), 7.66 (d, $J=7.2$ Hz, 3H), 7.24–7.56 (m, 5H); ^{13}C NMR ($CDCl_3$, 100 MHz): 159.1, 150.7, 142.2, 141.1, 138.8, 133.8, 129.6, 129.1, 128.7, 127.8, 127.5, 126.2, 126.1, 125.7, 124.3; MS (EI): 309 (M^+); HRMS (EI) calcd for $C_{17}H_{12}BrN$: 309.0148, found: 309.0150.

4.1.34. 4-Bromo-2-(4-nitrophenyl)pyridine (5h). Compound **5h**: yellow solid, mp 168–169 °C; 1H NMR ($DMSO-d_6$, 400 MHz): 8.63 (d, $J=5.2$ Hz, 1H), 8.32–8.44 (m, 5H), 7.76 (d, $J=3.6$ Hz, 1H); ^{13}C NMR ($DMSO-d_6$, 100 MHz): 155.2, 151.0, 148.0, 143.2, 133.6, 128.1, 126.9, 124.5, 123.9; MS (EI): 278.0 (M^+); HRMS (EI) calcd for $C_{11}H_7BrN_2O_2$: 277.9691, found: 277.9690.

4.1.35. 2-Phenyl-4-(4-methylphenyl)pyridine (6). Compound **6**: white solid, mp 95–96 °C; 1H NMR ($CDCl_3$, 400 MHz): 8.71 (d, $J=5.6$ Hz, 1H), 8.04 (d, $J=7.6$ Hz, 2H), 7.91 (d, $J=0.8$ Hz, 1H), 7.31–7.59 (m, 6H), 7.26 (d, $J=20$ Hz, 2H), 2.41 (s, 3H); ^{13}C NMR ($CDCl_3$,

100 MHz): 158.2, 150.1, 149.6, 139.6, 139.5, 135.7, 130.1, 129.3, 129.0, 127.3, 127.2, 120.3, 118.8, 21.5; MS (EI): 245.2 (M^+); HRMS (EI) calcd for $C_{18}H_{15}N$: 245.1204, found: 245.1209.

4.1.36. 2-Phenyl-4-(4-methylphenylethynyl)pyridine (7). Compound **7**: white solid, mp 121–122 °C; 1H NMR ($CDCl_3$, 400 MHz): 8.66 (d, $J=5.2$ Hz, 1H), 8.01 (d, $J=7.2$ Hz, 2H), 7.82 (s, 1H), 7.42–7.50 (m, 5H), 7.31 (d, $J=4.8$ Hz, 1H), 7.18 (d, $J=8.4$ Hz, 2H), 2.38 (s, 3H); ^{13}C NMR ($CDCl_3$, 100 MHz): 157.6, 149.6, 139.9, 138.8, 133.0, 132.1, 129.6 (d), 129.1, 127.2, 124.1, 122.8, 119.3, 94.7, 86.7, 21.9; MS (EI): 269.2 (M^+); HRMS (EI) calcd for $C_{20}H_{15}N$: 269.1204, found: 269.1205.

4.1.37. 2-Phenyl-4-(4-methylphenoxy)pyridine (8). Compound **8**: yellow liquid; 1H NMR ($CDCl_3$, 400 MHz): 8.51 (d, $J=6.0$ Hz, 1H), 7.91 (d, $J=7.2$ Hz, 2H), 7.36–7.45 (m, 3H), 7.20–7.26 (m, 3H), 7.01 (d, $J=8.4$ Hz, 2H), 6.74 (dd, $J_1=2.4$ Hz, $J_2=2.8$ Hz), 2.37 (s, 3H); ^{13}C NMR ($CDCl_3$, 100 MHz): 166.1, 159.8, 152.1, 151.3, 139.4, 135.2, 130.9, 129.3, 128.9, 127.2, 120.8, 110.7, 109.1, 21.1; MS (EI): 260.2 (M^+); HRMS (EI) calcd for $C_{18}H_{15}NO$: 261.1154, found: 261.1156.

4.1.38. 2-Phenyl-4-(p-tolylthio)pyridine (9).¹ Compound **9**: white solid, mp 84–85 °C; 1H NMR ($CDCl_3$, 400 MHz): 8.40 (d, $J=5.6$ Hz, 1H), 7.86 (d, $J=7.2$ Hz, 2H), 7.36–7.47 (m, 6H), 7.26 (d, $J=8.0$ Hz, 2H), 6.83 (dd, $J_1=1.2$ Hz, $J_2=1.2$ Hz, 1H), 2.41 (s, 3H); ^{13}C NMR ($CDCl_3$, 100 MHz): 157.5, 151.9, 149.4, 140.3, 139.2, 135.5, 131.0, 129.3, 128.9, 127.2, 126.1, 119.3, 117.8, 21.6; MS (EI): 276.1 (M^+); HRMS (EI) calcd for $C_{18}H_{15}NS$: 277.0925, found: 277.0930.

4.1.39. 4-Bromo-2-(5-chlorothiophen-2-yl)pyridine (10).¹ Compound **10**: yellow solid, mp 69–70 °C; 1H NMR ($CDCl_3$, 400 MHz): 8.32 (d, $J=5.2$ Hz, 1H), 7.70 (d, $J=1.6$ Hz, 1H), 7.28–7.31 (m, 2H), 6.91 (d, $J=4.0$ Hz, 1H); ^{13}C NMR ($CDCl_3$, 100 MHz): 153.2, 150.5, 142.1, 133.8, 133.6, 127.6, 125.5, 124.6, 121.5; MS (EI): 274.9 (M^+); HRMS (EI) calcd for $C_9H_5BrClNS$: 272.9015, found: 272.9014.

4.1.40. 4-(3-Hydroxyphenyl)-2-(5-chlorothiophen-2-yl)pyridine (11).^{8b} Compound **11**: yellow solid, mp 190–191 °C; 1H NMR ($CDCl_3$, 400 MHz): 9.70 (s, 1H), 8.54 (d, $J=5.2$ Hz, 1H), 8.18 (s, 1H), 1H NMR 7.91 (d, $J=4.4$ Hz, 1H), 7.54 (dd, $J_1=1.2$ Hz, $J_2=1.2$ Hz, 1H), 7.20–7.37 (m, 4H), 6.91–6.93 (m, 1H); ^{13}C NMR ($CDCl_3$, 100 MHz): 158.9, 152.4, 150.7, 149.4, 144.6, 139.2, 131.4, 131.0, 129.1, 126.2, 121.1, 118.6, 117.3, 116.3, 114.7; MS (EI): 287.0 (M^+); HRMS (EI) calcd for $C_{15}H_{10}ClNOS$: 287.0172, found: 287.0171.

Acknowledgements

This work was supported by the Natural Science Foundation of China (21172166, 21203135, and 21375092), China Postdoctor Foundation (2013M541456), The Science and Technology Plan Project of Zhejiang Province (2011C37055), and Taizhou Science Project (121ZD04).

Supplementary data

Supplementary data related to this article can be found at <http://dx.doi.org/10.1016/j.tet.2013.10.065>.

References and notes

- Dong, H.; Latka, R. T.; Driver, T. G. *Org. Lett.* **2011**, *13*, 2726–2729.
- Mousseau, J. J.; Bull, J. A.; Charette, A. B. *Angew. Chem., Int. Ed.* **2010**, *49*, 1115–1118.
- Yuan, C.; Chang, C.; Siegel, D. J. *Org. Chem.* **2013**, *78*, 5647–5668.
- See some recent examples of C–H activation of 2-arylpyridines: (a) Zhou, B.; Chen, H.; Wang, C. *J. Am. Chem. Soc.* **2013**, *135*, 1264–1267; (b) Hofmann, N.; Ackermann, L. *J. Am. Chem. Soc.* **2013**, *135*, 5877–5884; (c) Brasse, M.; Campora, J.; Ellman, J. A.; Bergman, R. G. *J. Am. Chem. Soc.* **2013**, *135*, 6427–6430; (d) Pan, F.; Lei, Z.; Wang, H.; Li, H.; Sun, J.; Shi, Z. *Angew. Chem., Int. Ed.* **2013**, *52*,

- 2063–2067; (e) Yoshino, T.; Ikemoto, H.; Matsunaga, S.; Kanai, M. *Angew. Chem., Int. Ed.* **2013**, 52, 2207–2211; (f) Li, X.; Yu, S.; Wang, F.; Wan, B.; Yu, X. *Angew. Chem., Int. Ed.* **2013**, 52, 2577–2580; (g) Aiso, H.; Kochi, T.; Mutsutani, H.; Tanabe, T.; Nishiyama, S.; Kakiuchi, F. *J. Org. Chem.* **2012**, 77, 7718–7724.
5. Tang, B.; Yu, F.; Li, P.; Tong, L.; Duan, X.; Xie, T.; Wang, X. *J. Am. Chem. Soc.* **2009**, 131, 3016–3023.
 6. Durben, S.; Baumgartner, T. *Inorg. Chem.* **2011**, 50, 6823–6836.
 7. (a) Chi, C.; Chiang, C.; Liu, S.; Yueh, H.; Chen, C.; Chen, C. *J. Mater. Chem.* **2009**, 19, 5561–5571; (b) Whittell, G. R.; Manners, I. *Adv. Mater.* **2007**, 19, 3439–3468; (c) Yamaguchi, Y.; Kobayashi, S.; Miyamura, S.; Okamoto, Y.; Wakamiya, T.; Matsubara, Y.; Yoshida, Z. *Angew. Chem.* **2004**, 116, 370–373; *Angew. Chem., Int. Ed.* **2004**, 43, 366–369.
 8. (a) Sha, F.; Wu, L.; Huang, X. *J. Org. Chem.* **2012**, 77, 3754–3765; (b) Oster, A.; Klein, T.; Werth, R.; Kruchten, P.; Bey, E.; Negri, M.; Marchais-Oberwinkler, S.; Frotscher, M.; Hartmann, R. W. *Bioorg. Med. Chem.* **2010**, 18, 3494–3505; (c) Heim-Riether, A.; Taylor, S. J.; Liang, S.; Donghong, A. G.; Xiong, Z.; Michael, A. E.; Collins, B. K.; Farmer, B. T.; Haverty, K.; Hill-Drzewi, M.; Junker, H.-D.; Margarit, S. M.; Moss, N.; Neumann, T.; Proudfoot, J. R.; Keenan, L. S.; Sekul, R.; Zhang, Q.; Li, J.; Farrow, N. A. *Bioorg. Med. Chem. Lett.* **2009**, 19, 5321–5324; (d) Sun, L.; Yang, Z.; Guo, H.; Pei, W. *Chin. J. Org. Chem.* **2012**, 32, 624–626.
 9. (a) Aida, W.; Ohtsuki, T.; Li, X.; Ishibashi, M. *Tetrahedron* **2009**, 65, 369–373; (b) Kitamura, A.; Tanaka, J.; Ohtani, I. I.; Higa, T. *Tetrahedron* **1999**, 55, 2487–2492.
 10. (a) Hudson, Z. M.; Sun, C.; Helander, M. G.; Amarne, H.; Lu, Z.; Wang, S. *Adv. Funct. Mater.* **2010**, 20, 3426–3439; (b) So, K. H.; Kim, R.; Park, H.; Kan, I.; Thangaraju, K.; Park, Y. S.; Kim, J. J.; Kwon, S.; Kim, Y. *Dyes Pigm.* **2011**, 92, 603–609; (c) Rao, Y.; Amarne, H.; Zhao, S.; McCormick, T. M.; Martić, S.; Sun, Y.; Wang, R.; Wang, S. *J. Am. Chem. Soc.* **2008**, 130, 12898–12900; (d) Ismail, M. A.; Arafa, R. K.; Brun, R.; Wenzler, T.; Miao, Y.; Wilson, W. D.; Generaux, C.; Bridges, A.; Hall, J. E.; Boykin, D. W. *J. Med. Chem.* **2006**, 49, 5324–5332; (e) Handy, S. T.; Wilson, T.; Muth, A. *J. Org. Chem.* **2007**, 72, 8496–8500; (f) Wong, K.; Hwu, T.; Balaiah, A.; Chao, T.; Fang, F.; Lee, C.; Peng, Y. *Org. Lett.* **2006**, 8, 1415–1418; (g) Burzicki, G.; Voisin-Chiret, A. S.; Sopková-de Oliveira Santos, J.; Rault, S. *Tetrahedron* **2009**, 65, 5413–5417.
 11. (a) Sicre, C.; Alonso-Gómez, J.; Cid, M. M. *Tetrahedron* **2006**, 62, 11063–11072; (b) Blackaby, W. P.; Atack, J. R.; Bromidge, F.; Castro, J. L.; Goodacre, S. C.; Hallett, D. J.; Lewis, R. T.; Marshall, G. R.; Pike, A.; Smith, A. J.; Street, L. J.; Tattersall, D. F. D.; Wafford, K. A. *Bioorg. Med. Chem. Lett.* **2006**, 16, 1175–1179.
 12. (a) Ding, M.; He, F.; Poss, M. A.; Rigat, K. L.; Wang, Y.; Roberts, S. B.; Qiu, D.; Fridell, R. A.; Gao, M.; Gentles, R. G. *Org. Biomol. Chem.* **2011**, 9, 6654–6662; (b) Dong, C.; Liu, T.; Hu, Q. *Synlett* **2009**, 1081–1086.
 13. Zhou, Q.; Zhang, B.; Du, T.; Gu, H.; Ye, Y.; Jiang, H.; Chen, R. *Tetrahedron* **2013**, 69, 327–333.
 14. Furuya, T.; Benitez, D.; Tkatchouk, E.; Strom, A. E.; Tang, P.; Goddard, W. A., III; Ritter, T. *J. Am. Chem. Soc.* **2010**, 132, 3793–3807.
 15. Garcia, Y.; Schoenebeck, F.; Legault, C. Y.; Merlic, C. A.; Houk, K. N. *J. Am. Chem. Soc.* **2009**, 131, 6632–6639.