

Radical mediated stereoselective synthesis of (4*R*,8*R*)-4,8-dimethyldecanal, an aggregation pheromone of *Tribolium* flour beetles

Yoko Kameda and Hajime Nagano*

Department of Chemistry, Faculty of Science, Ochanomizu University, Otsuka, Bunkyo-ku, Tokyo 112-8610, Japan

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Abstract—(4*R*,8*R*)-4,8-Dimethyldecanal, a common aggregation pheromone of *Tribolium* flour beetles, has been synthesized from (*R*)-2,3-*O*-isopropylidene-glyceraldehyde in 11 steps and 7% overall yield. The key step in the synthesis is the highly diastereoselective chelation-controlled radical reaction of ethyl (4*S*,5*R*)-4-benzyloxy-5,6-(isopropylidenedioxy)-2-methylenehexanoate with ethyl (*R*)-5-iodo-3-methylpentanoate performed in the presence of 7 equiv of MgBr₂·OEt₂.

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1. Introduction

The 1,5-*syn*-dimethylalkyl motif is ubiquitous in many natural products such as tocopherols, several insect pheromones, and membrane lipids of archaeobacteria. The stereoselective construction of the structural motif is therefore of particular interest^{1–3} and we intend to apply the chelation-controlled radical reaction of γ -benzyloxy- α -methylene-carboxylic acid ester **I** yielding *syn*-adduct **III**, recently developed in our laboratory,⁴ to the synthesis of these natural products (Scheme 1). The highly *syn*-selective addition of alkyl iodide R²I to **I** is referred to the H-atom transfer to the outside face of radical center in the sharply folded seven-membered chelate intermediate **II**.^{4c–e}

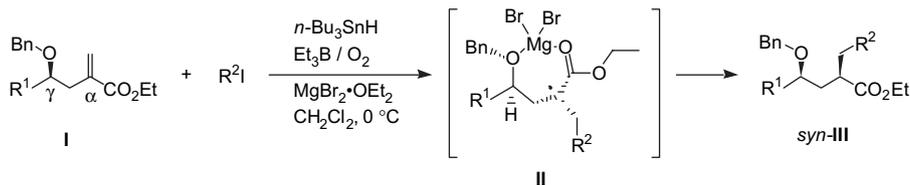
We now report the radical mediated stereoselective synthesis of (4*R*,8*R*)-4,8-dimethyldecanal (**1**), a common aggregation pheromone produced by the male flour beetles of *Tribolium castaneum*, *Tribolium confusum*, *Tribolium freemani*, and *Tribolium madens* (Coleoptera: Tenebrionidae).^{5,6} The pheromone possessing a 1,5-dimethylalkyl motif would be

synthesized by using the radical addition of alkyl iodide **IV** to optically active γ -benzyloxy- α -methylene-carboxylic acid ester **I** followed by the reduction of the ethoxycarbonyl group to a methyl group (Scheme 2).

The first synthesis of (4*R*,8*R*)-4,8-dimethyldecanal (**1**) and three other stereoisomers from (*R*)-citronellol and (*R*)-citronellic acid has established that the absolute configuration of the pheromone is (4*R*,8*R*).^{7,8} Since the identification of the structure, several stereoselective syntheses of the pheromone **1** have been reported.⁹

2. Results and discussion

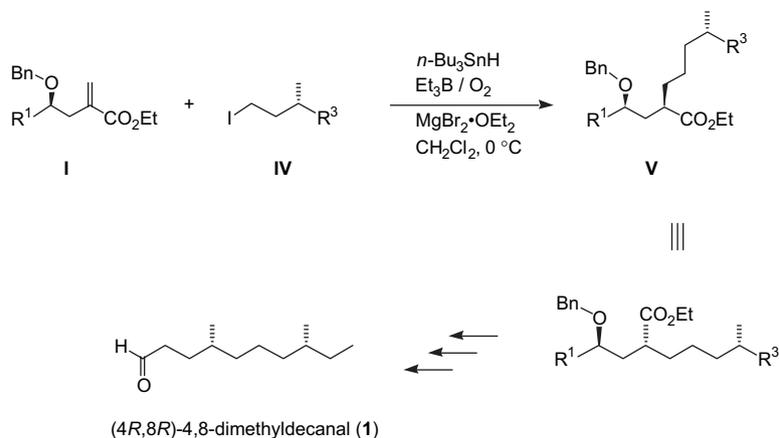
Scheme 3 shows the synthetic route of (4*R*,8*R*)-4,8-dimethyldecanal (**1**) starting from (*R*)-2,3-*O*-isopropylidene-glyceraldehyde (**2**), which is prepared from 1,2:5,6-di-*O*-isopropylidene-D-mannitol.¹⁰ The Reformatsky reaction of aldehyde **2** with ethyl 2-(bromomethyl)propanoate (**3**) gave an inseparable diastereomeric mixture of γ -hydroxy



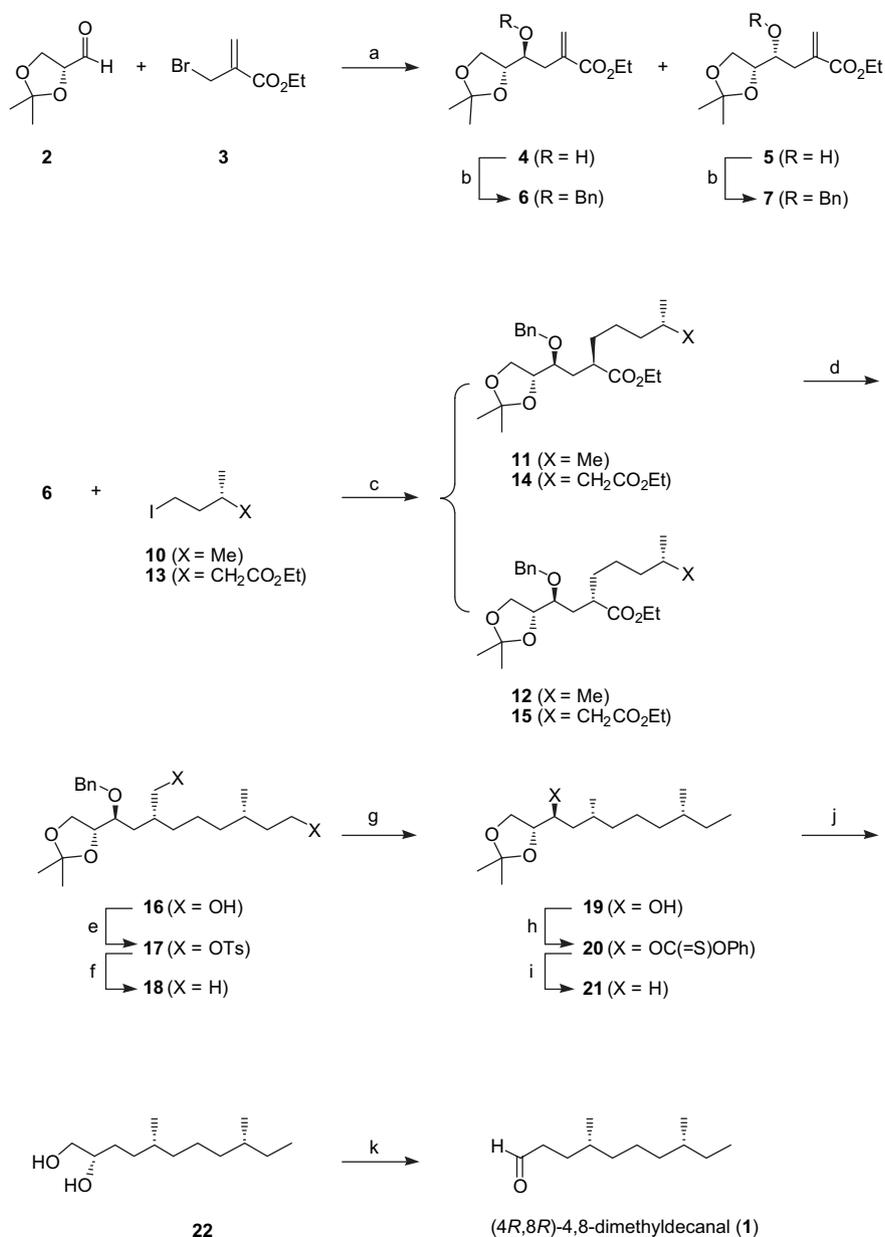
Scheme 1. Chelation-controlled diastereoselective radical reactions of α -methylene- γ -oxycarboxylic acid esters **I** with alkyl iodides R²I yielding *syn*-adducts **III**.

Keywords: *Tribolium* flour beetle; Pheromone; (4*R*,8*R*)-4,8-Dimethyldecanal; Radical reaction; 1,3-Asymmetric induction.

* Corresponding author. Fax: +81 3 5978 5715; e-mail: nagano@cc.ocha.ac.jp

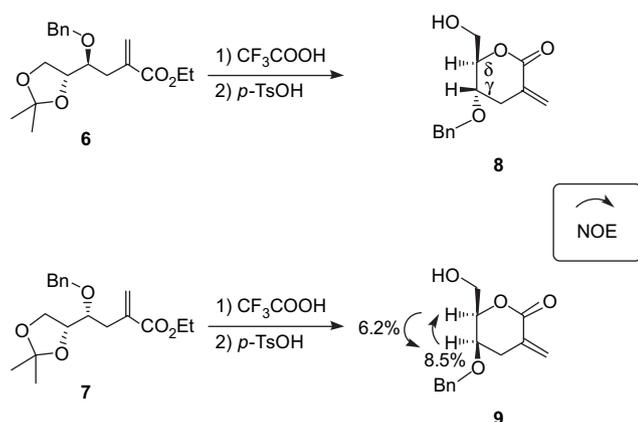


Scheme 2. Synthetic plan of (4R,8R)-4,8-dimethyldecanal (1).



Scheme 3. Synthetic route of (4R,8R)-4,8-dimethyldecanal (1). Reagents: (a) Zn, THF, aq NH₄Cl; (b) PhCH₂Br, Ag₂O, toluene; (c) $n\text{-Bu}_3\text{SnH}$, Et_3B , $\text{MgBr}_2 \cdot \text{OEt}_2$, CH_2Cl_2 ; (d) LiAlH_4 , diethyl ether; (e) $p\text{-TsCl}$, pyridine, CH_2Cl_2 ; (f) LiAlH_4 , diethyl ether; (g) H₂, Pd-C, ethanol; (h) PhOC(=S)Cl, pyridine, CH_2Cl_2 ; (i) $n\text{-Bu}_3\text{SnH}$, AIBN, toluene, 85°C ; (j) 1 mol dm⁻³ HCl, THF-H₂O (1:1); (k) NaIO₄, aq CH₃CN.

esters **4** and **5** in a ratio of 2.8:1.¹¹ Treatment of the mixture with benzyl bromide and silver oxide gave a mixture of benzyl ethers **6** and **7**, which was easily separated by silica gel column chromatography to give **6** in 26% yield from 1,2:5,6-di-*O*-isopropylidene-*D*-mannitol. To assign unambiguously the stereochemistry of the newly formed chiral center in the Reformatsky reaction,¹¹ the benzyl ethers **6** and **7** were transformed into δ -lactones **8** and **9**, respectively, and their NOE experiments were performed (Scheme 4). For the lactone **9**, NOE enhancements of γ -H (6.2%) and δ -H (8.5%) were observed by irradiating δ -H and γ -H, respectively, while for the lactone **8**, irradiation of γ -H and δ -H did not enhance the signals of their vicinal methine protons. The NOE difference spectra of the δ -lactones **8** and **9** thus established the stereochemistry of **6** and **7** as 4,5-*anti* and 4,5-*syn*, respectively.



Scheme 4. Determination of the configurations of **6** and **7**.

The radical reaction of the major diastereomer **6** with 1-iodo-3-methylbutane (**10**) was at first performed under the reaction conditions used in our previous work⁴ [iodide **10** (3 equiv), *n*-Bu₃SnH (2 equiv), Et₃B (1 equiv), MgBr₂·OEt₂ (3 equiv) in CH₂Cl₂ at 0 °C] to give the 2,4-*syn*-adduct **11** and 2,4-*anti*-adduct **12** in 63% yield and 14:1 diastereomer ratio. In order to ameliorate the stereoselectivity, the radical reaction was then performed using 7 equiv of MgBr₂·OEt₂. The diastereoselectivity and yield were improved to **11**/**12**>50:1 and 73%, respectively.

In our previous work,^{4c} we confirmed the seven-membered chelate ring formation of the starting material **I** (R¹=Ph) by the complexation experiment with 3 equiv of MgBr₂·OEt₂ in CDCl₃. The large difference in chemical shift increments $\Delta\delta=[\delta_{\text{H}}(\text{substrate}+\text{MgBr}_2\cdot\text{OEt}_2)-\delta_{\text{H}}(\text{substrate})]$ by adding the Lewis acid between the diastereotopic β -methylene protons suggests the formation of bidentate complexation. However, in the complexation experiment of **6** with 3 equiv of MgBr₂·OEt₂, only slight chemical shift increments were observed. The results of the complexation experiment of **6** with 7 equiv of MgBr₂·OEt₂ are shown in Figure 1. The large $\Delta\delta$ values suggest that the addition of 7 equiv of the Lewis acid is required to achieve the chelate ring formation and the highly diastereoselective radical addition reaction.¹²

The radical addition of ethyl (*R*)-5-iodo-3-methylpentanoate (**13**)¹³ to **6**, i.e., the key step in the synthesis of (4*R*,8*R*)-4,8-

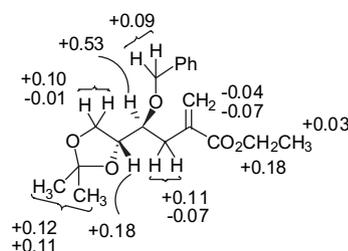
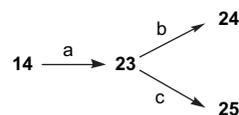
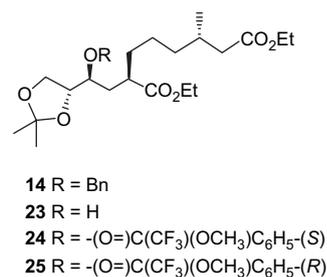


Figure 1. $\Delta\delta$ values (ppm) for the substrate **6**. $\Delta\delta_{\text{H}}=\delta_{\text{H}}(\text{substrate } \mathbf{6}+\text{MgBr}_2\cdot\text{OEt}_2)-\delta_{\text{H}}(\text{substrate } \mathbf{6})$. The $\delta_{\text{H}}(\text{substrate } \mathbf{6}+\text{MgBr}_2\cdot\text{OEt}_2)$ value was obtained after sonication of **6** with 7 equiv of MgBr₂·OEt₂ in CDCl₃.

dimethyldecanal (**1**), was carried out using 7 equiv of MgBr₂·OEt₂. The desired 2,4-*syn*-adduct **14** was obtained in 71% yield and >96% de. The ¹H NMR and ¹³C NMR spectra of α -methoxy- α -(trifluoromethyl)phenylacetates **24** and **25**, which were transformed from **14** via the corresponding alcohol **23**, did not show the epimerization of aldehyde **2** during the Reformatsky reaction of **2** with **3** (Scheme 5).



Scheme 5. Reagents: (a) H₂, Pd-C, ethanol; (b) (*R*)-(-)- α -methoxy- α -(trifluoromethyl)phenylacetyl chloride, 4-dimethylaminopyridine, pyridine; (c) (*S*)-(+)- α -methoxy- α -(trifluoromethyl)phenylacetyl chloride, 4-dimethylaminopyridine, pyridine.

The diester **14** was then reduced with lithium aluminium hydride to give the corresponding diol **16** in 89% yield. The tosylation of the diol with *p*-toluenesulfonyl chloride followed by the reduction with lithium aluminium hydride gave compound **18** in 76% yield. The hydrogenolysis of **18** over Pd-C gave alcohol **19** quantitatively. The alcohol was then treated with phenyl chloroformate to give phenoxythiocarbonyl ester **20**, which was then reduced under radical conditions using AIBN and *n*-Bu₃SnH to give compound **21** in 71% two-step yield. The acid catalyzed hydrolysis of **21** followed by the oxidative cleavage of the resulting diol **22** with sodium periodate gave (4*R*,8*R*)-4,8-dimethyldecanal (**1**), as an oil, [α]_D²⁵ -5.7 (*c* 1.0, CHCl₃) (lit.^{7b} [α]_D^{22.5} -7.3 (*c* 2.04, CHCl₃)), in 85% two-step yield. The IR, ¹H NMR, ¹³C NMR, and MS spectral data of the synthetic aldehyde **1** were identical with those of (4*R*,8*R*)- and (4*S*,8*S*)-4,8-dimethyldecanals reported in the literatures.^{7,9,14}

We have thus synthesized (4*R*,8*R*)-4,8-dimethyldecanal (**1**) from (*R*)-2,3-*O*-isopropylidene-glyceraldehyde (**2**) in 11 steps and 7% overall yield.

3. Conclusion

We have synthesized (4*R*,8*R*)-4,8-dimethyldecanal (**1**) from (*R*)-2,3-*O*-isopropylidene-glyceraldehyde (**2**), easily prepared from *D*-mannitol, in 11 steps and 7% overall yield. The key step in the synthesis is the highly diastereoselective chelation-controlled radical reaction of γ -benzyloxy- α -methylene-carboxylic acid ester **6** and ethyl (*R*)-5-iodo-3-methylpentanoate (**13**) performed in the presence of 7 equiv of $\text{MgBr}_2 \cdot \text{OEt}_2$.

4. Experimental

4.1. General

^1H NMR spectra were recorded on a JEOL GSX-400 (400 MHz) spectrometer with CDCl_3 as the solvent and tetramethylsilane as an internal standard. ^{13}C NMR spectra were recorded on the instrument operating at 100.5 MHz with CDCl_3 as the solvent and internal standard (δ 77.0). IR spectra were taken on a SIMADZU FTIR-8700 spectrometer. Mass spectra (EI^+) were obtained on a JEOL JMS-700 mass spectrometer. Precoated Merck Kieselgel 60 F_{254} and Kanto silica gel 60 (spherical neutral) were used for thin layer chromatography and column chromatography, respectively.

4.1.1. Ethyl (4*S*,5*R*)- and (4*R*,5*R*)-4-hydroxy-5,6-(isopropylidenedioxy)-2-methylenehexanoates (**4**) and (**5**).

(*R*)-2,3-*O*-isopropylidene-glyceraldehyde (**2**) was prepared from 1,2:5,6-di-*O*-isopropylidene-*D*-mannitol (2.87 g, 10.9 mmol) following the reported procedures.¹⁰ To a solution of the aldehyde **2** in THF (36 cm^3) were added ethyl 2-(bromomethyl)propeonate (**3**) (5.95 g, 30.8 mmol), saturated aqueous NH_4Cl (57 cm^3), and activated zinc powder (3.34 g, 51.3 mmol) and the mixture was stirred at 0 °C for 3 h. The product was extracted with ethyl acetate and the extract was washed with saturated brine and then dried over anhydrous sodium sulfate. The solvent was evaporated in vacuo and the residue was purified by column chromatography on silica gel [eluent: hexane–ethyl acetate (8:1)] to give an inseparable mixture of **4** and **5** (3.38 g, 63% yield from 1,2:5,6-di-*O*-isopropylidene-*D*-mannitol; **4**/**5**=2.8:1) as an oil. MS m/z 229 ($\text{M}^+ - \text{Me}$, 68%), 143 (94), 123 (71), 101 (100); HRMS calcd for $\text{C}_{11}\text{H}_{17}\text{O}_5$ [$\text{M}^+ - \text{Me}$] 229.1076, found 229.1082. Major diastereomer **4**. ^1H NMR δ 1.32 (3H, t, $J=7.2$ Hz, CH_3), 1.36 (3H, s, CH_3), 1.43 (3H, s, CH_3), 2.37 (1H, ddd, $J=14.2$, 8.4, 0.8 Hz, $\text{CHHC}=\text{CH}_2$), 2.70 (1H, ddd, $J=14.2$, 3.6, 0.8 Hz, $\text{CHHC}=\text{CH}_2$), 2.86 (1H, d, $J=3.6$ Hz, OH), 3.75–4.10 (4H, m, $\text{CH}(\text{O})\text{CHOH}$, $\text{CO}_2\text{CH}_2\text{CH}_3$), 4.21 (1H, dd, $J=14.2$, 7.0 Hz, CHHO), 4.25 (1H, dd, $J=14.2$, 7.0 Hz, CHHO), 5.74 (1H, d, $J=1.6$ Hz, $\text{C}=\text{CHH}$), 6.29 (1H, d, $J=1.2$ Hz, $\text{C}=\text{CHH}$); ^{13}C NMR δ 14.20, 25.30, 26.67, 36.15, 61.19, 65.94, 71.28, 78.10, 109.12, 128.00, 136.92, 167.89. Minor diastereomer **5**. ^1H NMR δ 1.31 (3H, t, $J=7.2$ Hz, CH_3), 1.37 (3H, s, CH_3), 1.44 (3H, s, CH_3), 2.41–2.49 (2H, m, $\text{CH}_2\text{C}=\text{CH}_2$), 2.86 (1H, d, $J=3.6$ Hz, OH), 3.75–4.10 (4H, m, $\text{CH}(\text{O})\text{CHOH}$, $\text{CO}_2\text{CH}_2\text{CH}_3$), 4.20 (1H, dd, $J=14.2$, 6.8 Hz, CHHO), 4.23 (1H, dd, $J=14.2$, 6.8 Hz, CHHO), 5.72 (1H, d, $J=1.6$ Hz, $\text{C}=\text{CHH}$), 6.28 (1H, d, $J=1.2$ Hz, $\text{C}=\text{CHH}$); ^{13}C NMR

δ 14.22, 25.30, 26.58, 36.77, 60.93, 66.00, 70.62, 78.40, 109.36, 127.69, 136.64, 167.08.

4.1.2. Ethyl (4*S*,5*R*)- and (4*R*,5*R*)-4-benzyloxy-5,6-(isopropylidenedioxy)-2-methylenehexanoates (**6**) and (**7**).

To a solution of alcohols **4** and **5** (29.2 mg, 0.12 mmol) in toluene (2 cm^3) were added benzyl bromide (0.05 cm^3 , 0.36 mmol) and freshly prepared silver oxide (84.5 mg, 0.36 mmol). The mixture was stirred at room temperature for 48 h and then filtered through a pad of Celite. The filtrate was evaporated in vacuo and the residue was purified by column chromatography on silica gel [eluent: hexane–ethyl acetate (20:1)] to give **6** (14 mg) and **7** (8.3 mg), and **6** (12.3 mg). Major diastereomer **6**. $[\alpha]_D^{23} +22.0$ (c 1.95, CHCl_3); ^1H NMR δ 1.28 (3H, t, $J=7.0$ Hz, CH_3), 1.35 (3H, s, CH_3), 1.43 (3H, s, CH_3), 2.46 (1H, dd, $J=14.2$, 7.8 Hz, $\text{CHHC}=\text{CH}_2$), 2.72 (1H, ddd, $J=14.2$, 4.4, 0.8 Hz, $\text{CHHC}=\text{CH}_2$), 3.73 (1H, m, CH-O), 3.90 (1H, dd, $J=7.8$, 6.0 Hz, CHH-O), 4.03 (1H, d, $J=7.8$, 6.4 Hz, CHH-O), 4.12 (1H, dd, $J=10.5$, 6.0 Hz, CH-O), 4.16 (2H, q, $J=7.0$ Hz, $\text{CH}_2\text{-O}$), 4.59 (2H, s, PhCH_2O), 5.68 (1H, d, $J=1.2$ Hz, $\text{C}=\text{CHH}$), 6.23 (1H, d, $J=1.2$ Hz, $\text{C}=\text{CHH}$), 7.28–7.32 (5H, m, Ph); ^{13}C NMR δ 14.24, 25.36, 26.56, 34.25, 60.72, 66.00, 72.98, 77.42, 77.82, 109.12, 127.50, 127.63, 127.78, 128.17, 136.91, 138.16, 166.91; MS m/z 319 ($\text{M}^+ - \text{Me}$, 26%), 233 (95), 101 (58), 91 (100); HRMS calcd for $\text{C}_{18}\text{H}_{23}\text{O}_5$ [$\text{M}^+ - \text{Me}$] 319.1545, found 319.1552. Minor diastereomer **7**. ^1H NMR δ 1.28 (3H, t, $J=7.2$ Hz, CH_3), 1.37 (3H, s, CH_3), 1.45 (3H, s, CH_3), 2.42–2.52 (2H, m, $\text{CH}_2\text{C}=\text{CH}_2$), 3.63 (1H, m, CH-O), 3.79 (1H, dd, $J=8.0$, 7.3 Hz, CHH-O), 3.99 (1H, d, $J=8.0$, 6.5 Hz, CHH-O), 4.13–4.23 (3H, m, CH-O , $\text{CH}_2\text{-O}$), 4.59 (1H, d, $J=11.7$ Hz, PhCHH-O), 4.70 (1H, d, $J=11.7$ Hz, PhCHH-O), 5.69 (1H, d, $J=1.2$ Hz, $\text{C}=\text{CHH}$), 6.23 (1H, d, $J=1.6$ Hz, $\text{C}=\text{CHH}$), 7.26–7.36 (5H, m, Ph); ^{13}C NMR δ 14.24, 25.47, 26.53, 34.09, 60.72, 65.88, 72.93, 77.95, 78.05, 109.30, 127.41, 127.83, 127.88, 128.11, 136.64, 138.38, 166.80. MS m/z 319 ($\text{M}^+ - \text{Me}$, 25%), 233 (94), 101 (54), 91 (100); HRMS calcd for $\text{C}_{18}\text{H}_{23}\text{O}_5$ [$\text{M}^+ - \text{Me}$] 319.1545, found 319.1597.

4.1.3. (5*S*,6*R*)-5-Benzyloxy-6-hydroxymethyl-3-methylenetetrahydropyran-2-one (**8**).

To a solution of **6** (18.2 mg, 0.054 mmol) in $\text{THF-H}_2\text{O}$ (4:1, 1 cm^3) was added trifluoroacetic acid at 0 °C. The solution was stirred at room temperature overnight and then the product was extracted with ethyl acetate and the extract was washed with saturated brine and dried over anhydrous sodium sulfate. Concentration of the solution gave a crude product, which was used without further purification. To a solution of the crude product in benzene (1 cm^3) was added catalytic amount of *p*-toluenesulfonic acid. The mixture was stirred at room temperature overnight. The solvent was evaporated in vacuo and the residue was purified by column chromatography on silica gel [eluent: hexane–ethyl acetate (2:1)] to give **8** (6.9 mg, 51%) as an oil. ^1H NMR δ 2.66 (1H, m, CHHCHOBn), 3.03 (1H, m, CHHCHOBn), 3.85 (1H, m, CHOBn), 3.86 (1H, dd, $J=12.3$, 3.4 Hz, CHHOH), 3.93 (1H, dd, $J=12.3$, 3.4 Hz, CHHOH), 4.34 (1H, dt, $J=7.8$, 3.4 Hz, CHCH_2OH), 4.60 (1H, d, $J=11.5$ Hz, PhCHH), 4.68 (1H, d, $J=11.5$ Hz, PhCHH), 5.65 (1H, m, $\text{C}=\text{CHH}$), 6.46 (1H, m, $\text{C}=\text{CHH}$), 7.30–7.39 (5H, m, Ph); ^{13}C NMR δ 33.16, 61.74, 70.14, 71.50, 82.14, 127.72, 128.06,

128.52, 130.18, 131.19, 137.20, 164.53; MS m/z 248 (M^+ , 14%), 124 (15), 97 (41), 91 (100); HRMS calcd for $C_{14}H_{16}O_4$ [M^+] 248.1049, found 248.1055.

4.1.4. (5R,6R)-5-Benzyloxy-6-hydroxymethyl-3-methylenetetrahydropyran-2-one (9). To a solution of **7** (12.9 mg, 0.039 mmol) in THF–H₂O (4:1, 1 cm³) was added trifluoroacetic acid at 0 °C. The solution was stirred at room temperature overnight and then the product was extracted with ethyl acetate and the extract was washed with saturated brine and dried over anhydrous sodium sulfate. Evaporation of the solvent in vacuo gave the product, which was used without further purification. To a solution of the product in benzene (1 cm³) was added catalytic amount of *p*-toluenesulfonic acid. The mixture was stirred at room temperature overnight. The solvent was evaporated in vacuo and the residue was purified by column chromatography on silica gel [eluent: hexane–ethyl acetate (2:1)] to give **9** (3.5 mg, 37%) as an oil. ¹H NMR δ 2.65 (1H, m, CHHCHOBN), 3.07 (1H, dd, $J=16.4$, 3.9 Hz, CHHCHOBN), 3.78 (1H, dd, $J=11.9$, 5.1 Hz, CHHOH), 3.93 (1H, m, CHOBN), 3.99 (1H, dd, $J=11.9$, 6.6 Hz, CHHOH), 4.42 (1H, d, $J=12.1$ Hz, PhCHH), 4.45 (1H, m, CHCH₂OH), 4.66 (1H, d, $J=12.1$ Hz, PhCHH), 5.63 (1H, m, C=CHH), 6.52 (1H, m, C=CHH), 7.28–7.38 (5H, m, Ph); ¹³C NMR δ 31.88, 62.42, 69.33, 70.22, 82.00, 127.69, 128.09, 128.53, 130.31, 130.57, 137.03, 164.16; MS m/z 248 (M^+ , 7%), 124 (18), 97 (27), 91 (100); HRMS calcd for $C_{14}H_{16}O_4$ [M^+] 248.1049, found 248.1058.

4.1.5. Ethyl (2R)-2-[(2S,3R)-2-benzyloxy-3,4-(isopropylidenedioxy)butyl]-6-methyl-heptanoate (11). To a solution of α -methylene ester **6** (21.5 mg, 0.064 mmol) in dry CH₂Cl₂ (1.3 cm³) was added MgBr₂·OEt₂ (117 mg, 0.45 mmol) under N₂ and the mixture was stirred at room temperature for 15 min. To the suspension cooled to 0 °C were added 1-iodo-3-methylbutane (**10**) (0.026 cm³, 0.19 mmol), *n*-Bu₃SnH (0.036 cm³, 0.13 mmol), and Et₃B (0.064 cm³, 1.0 mol dm⁻³ in hexane). The mixture was stirred at 0 °C for 6 h. KF and water were added and the mixture was stirred at room temperature overnight. After filtration through a pad of Florisil, the filtrate was evaporated in vacuo and the residue was purified by column chromatography on silica gel [2 g; eluent: hexane then hexane–ethyl acetate (30:1)] to give a mixture of **11** and **12** (19 mg, 73% yield; **11/12**>50:1) as an oil. ¹H NMR δ 0.85 (6H, d, $J=6.3$ Hz, 2×CH₃), 1.22 (3H, t, $J=7.0$ Hz, CH₃), 1.13–1.65 (8H, m, CHHCH(CH₂)₃CH(CH₃)₂), 1.35 (3H, s, CH₃), 1.42 (3H, s, CH₃), 1.95 (1H, ddd, $J=14.2$, 11.3, 3.0 Hz, CHH), 2.68 (1H, m, CHC=O), 3.52 (1H, m, CH–O), 3.88 (1H, dd, $J=8.0$, 7.6 Hz, CHH–O), 4.00–4.14 (4H, m, CHH–O, CH–O, CH₂–O), 4.55 (1H, d, $J=11.0$ Hz, PhCHH), 4.65 (1H, d, $J=11.0$ Hz, PhCHH), 7.26–7.34 (5H, m, Ph); ¹³C NMR δ 14.39, 22.57, 22.66, 25.02, 25.33, 26.59, 27.84, 33.72, 33.98, 38.77, 41.54, 60.13, 66.06, 73.36, 77.62, 77.73, 109.04, 127.54, 127.85, 128.23, 138.34, 176.09; MS m/z 391 (M^+ –Me, 13%), 305 (76), 101 (32), 91 (100); HRMS calcd for $C_{23}H_{35}O_5$ [M^+ –Me] 391.2485, found 391.2504.

4.1.6. Diethyl (2R,6S)-2-[(2S,3R)-2-benzyloxy-3,4-(isopropylidenedioxy)butyl]-6-methyl-1,8-octanedioate (14). The radical reaction of **6** (83 mg, 0.25 mmol) with iodide

13 (210 mg, 0.78 mmol) in the presence of MgBr₂·OEt₂ (453 mg, 1.75 mmol) as described above gave **14** (85 mg, 71% yield; 96% de) as an oil. $[\alpha]_D^{23}$ –3.0 (*c* 2.0, CHCl₃); ¹H NMR δ 0.91 (3H, d, $J=6.3$ Hz, CH₃), 1.22 (3H, t, $J=7.3$ Hz, CH₃), 1.25 (3H, t, $J=7.3$ Hz, CH₃), 1.35 (3H, s, CH₃), 1.42 (3H, s, CH₃), 1.13–1.62 (7H, m), 1.92–1.99 (2H, m, CH₂C=O), 2.07 (1H, dd, $J=14.7$, 7.8 Hz, CHH), 2.26 (1H, dd, $J=14.7$, 5.8 Hz, CHH), 2.65 (1H, m, CHC=O), 3.52 (1H, m, CHOBN), 3.86 (1H, dd, $J=8.0$, 6.5 Hz, CHH–O), 4.00–4.16 (6H, m, CH–O, CHH–O, 2×CH₂–O), 4.55 (1H, d, $J=11.0$ Hz, PhCHH), 4.64 (1H, d, $J=11.0$ Hz, PhCHH), 7.27–7.34 (5H, m, Ph); ¹³C NMR δ 14.41, 19.75, 24.64, 25.36, 26.65, 30.30, 33.59, 33.95, 36.57, 41.49, 41.87, 60.13, 60.20, 66.08, 73.34, 77.59, 77.68, 109.02, 127.51, 127.80, 128.19, 138.25, 172.94, 175.85; MS m/z 463 (M^+ –Me, 22%), 377 (95), 285 (46), 101 (45), 91 (100); HRMS calcd for $C_{26}H_{39}O_7$ [M^+ –Me] 463.2695, found 463.2740.

4.1.7. (2R,6S)-2-[(2S,3R)-2-benzyloxy-3,4-(isopropylidenedioxy)butyl]-6-methyloctane-1,8-diol (16). To a solution of diester **14** (248 mg, 0.52 mmol) in dry diethyl ether (13 cm³) was added lithium aluminium hydride (3.3 equiv) at 0 °C and the mixture was stirred at room temperature for 12 h. To the mixture cooled to 0 °C were added dropwise water, then 15% aqueous sodium hydroxide. After filtration through a pad of Celite, the filtrate was dried over anhydrous sodium sulfate. The solvent was evaporated in vacuo and the residue was purified by column chromatography on silica gel [eluent: hexane–ethyl acetate (2:1)] to give diol **16** (181 mg, 89%) as an oil. $[\alpha]_D^{23}$ –0.52 (*c* 2.2, CHCl₃); ¹H NMR δ 0.88 (3H, d, $J=6.3$ Hz, CH₃), 1.36 (3H, s, CH₃), 1.11–1.40 (7H, m, (CH₂)₃CH), 1.43 (3H, s, CH₃), 1.44–1.64 (4H, m, 2×CH₂), 1.73 (1H, m, CH), 3.42 (1H, dd, $J=11.2$, 6.4 Hz, CH–O), 3.51 (1H, dd, $J=11.2$, 4.4 Hz, CH–O), 3.63–3.72 (3H, m, CH–O, CH₂–O), 3.86–4.18 (3H, m, CH–O, CH₂–O), 4.61 (1H, d, $J=11.3$ Hz, PhCHH), 4.71 (1H, d, $J=11.3$ Hz, PhCHH), 7.26–7.37 (5H, m, Ph); ¹³C NMR δ 19.71, 24.26, 25.28, 26.55, 29.36, 32.26, 33.54, 37.18, 39.91, 61.10, 65.63, 66.31, 72.80, 78.05, 109.02, 127.79, 127.96, 128.38, 137.87; MS m/z 379 (M^+ –Me, 4%), 185 (91), 101 (24), 91 (100); HRMS calcd for $C_{22}H_{35}O_5$ [M^+ –Me] 379.2485, found 379.2516.

4.1.8. (2R,3S,5R,9S)-3-Benzyloxy-1,2-(isopropylidenedioxy)-9-methyl-11-(*p*-toluenesulfonyloxy)-5-(*p*-toluenesulfonyloxymethyl)undecane (17). To a solution of diol **16** (181 mg, 0.46 mmol) in dry CH₂Cl₂ (7 cm³) were added pyridine (0.3 cm³) and *p*-toluenesulfonyl chloride (879 mg, 4.6 mmol) at 0 °C. The solution was stirred at room temperature overnight. After dilution with water, the product was extracted with chloroform. The extract was washed with water, saturated sodium hydrogencarbonate solution, and saturated brine, and then dried over anhydrous sodium sulfate. The solvent was evaporated in vacuo and the residue was purified by column chromatography on silica gel [eluent: hexane–ethyl acetate (6:1)] to give tosylate **17** (294 mg, 91%) as an oil. $[\alpha]_D^{23}$ –7.5 (*c* 2.1, CHCl₃); ¹H NMR δ 0.74 (3H, d, $J=6.1$ Hz, CH₃), 0.97–1.48 (10H, m), 1.34 (3H, s, CH₃), 1.41 (3H, s, CH₃), 1.60 (1H, m, CH), 1.80 (1H, m, CH), 2.44 (3H, s, CH₃), 2.45 (3H, s, CH₃), 3.49 (1H, m, CH–O), 3.77 (1H, dd, $J=9.8$, 4.9 Hz, CH–O), 3.83 (2H, m, CH–O), 3.97–4.11 (4H, m, CH–O), 4.45 (1H,

d, $J=11.7$ Hz, PhCHH), 4.67 (1H, d, $J=11.7$ Hz, PhCHH), 7.25–7.38 (9H, m, Ar), 7.73 (2H, d, $J=8.3$ Hz, Ar), 7.78 (2H, d, $J=8.3$ Hz, Ar); ^{13}C NMR δ 19.01, 21.69, 23.74, 25.23, 26.45, 29.10, 31.78, 32.44, 33.91, 35.67, 36.61, 65.70, 68.88, 71.92, 72.88, 76.06, 78.11, 108.98, 127.66, 127.77, 127.78, 127.84, 128.29, 129.75, 132.75, 133.01, 138.13, 144.58, 144.66.

4.1.9. (2R,3S,5R,9R)-3-Benzoyloxy-1,2-(isopropylidenedioxy)-5,9-dimethylundecane (18). To a solution of tosylate **17** (134.3 mg, 0.19 mmol) in dry diethyl ether (10 cm³) was added lithium aluminium hydride (3.3 equiv) at 0 °C. The mixture was stirred at room temperature for 12 h. To the mixture cooled to 0 °C were added dropwise water and then 15% aqueous sodium hydroxide. After filtration through a pad of Celite, the filtrate was dried over anhydrous sodium sulfate. The solvent was evaporated in vacuo and the residue was purified by column chromatography on silica gel [eluent: hexane–ethyl acetate (40:1)] to give **18** (57.3 mg, 83%) as an oil. $[\alpha]_{\text{D}}^{23}$ –6.6 (*c* 1.8, CHCl₃); ^1H NMR δ 0.83 (6H, d, $J=6.4$ Hz, 2×CH₃), 0.85 (3H, t, $J=7.3$ Hz, CH₃), 1.03–1.66 (12H, m, 5×CH₂, 2×CH), 1.37 (3H, s, CH₃), 1.43 (3H, s, CH₃), 3.70 (1H, dt, $J=10.0$, 3.4 Hz, CHOBn), 3.94 (1H, dd, $J=7.8$, 7.3 Hz, CHH–O), 4.02 (1H, dd, $J=7.8$, 6.3 Hz, CHH–O), 4.10 (1H, m, CH–O), 4.59 (1H, d, $J=11.2$ Hz, PhCHH), 4.77 (1H, d, $J=11.2$ Hz, PhCHH), 7.27–7.34 (5H, m, Ph); ^{13}C NMR δ 11.47, 19.31, 19.41, 24.42, 25.43, 26.53, 29.12, 29.49, 34.42, 36.87, 38.25, 39.43, 65.59, 73.41, 76.62, 79.08, 108.84, 127.47, 127.74, 128.23, 138.66; MS m/z 347 (M^+ –Me, 16%), 261 (23), 101 (73), 91 (100); HRMS calcd for C₂₂H₃₅O₃ [M^+ –Me] 347.2587, found 347.2578.

4.1.10. (2R,3S,5R,9R)-1,2-(Isopropylidenedioxy)-5,9-dimethylundecan-3-ol (19). A solution of **18** (47.2 mg, 0.13 mmol) in dry ethanol (5 cm³) was stirred with Pd–C (35.5 mg, 0.032 mmol) under hydrogen atmosphere at room temperature for 12 h. The mixture was filtered through a pad of Celite and the filtrate was concentrated in vacuo to give **19** in quantitative yield. $[\alpha]_{\text{D}}^{23}$ –0.73 (*c* 1.86, CHCl₃); ^1H NMR δ 0.84 (3H, d, $J=6.4$ Hz, CH₃), 0.85 (3H, t, $J=7.3$ Hz, CH₃), 0.90 (3H, d, $J=6.8$ Hz, CH₃), 1.05–1.69 (12H, m, 5×CH₂, 2×CH), 1.37 (3H, s, CH₃), 1.43 (3H, s, CH₃), 1.92 (1H, br s, OH), 3.87–4.02 (4H, m, CH₂–O, CH–O, CHOH); ^{13}C NMR δ 11.47, 19.09, 19.30, 24.43, 25.37, 26.52, 29.18, 29.48, 34.42, 36.88, 38.24, 39.70, 64.44, 68.29, 79.15, 108.83; MS m/z 257 (M^+ –Me, 100%), 229 (18), 101 (13); HRMS calcd for C₁₅H₂₉O₃ [M^+ –Me] 257.2117, found 257.2121.

4.1.11. (2R,3S,5R,9R)-3-Phenoxythiocarbonyloxy-1,2-(isopropylidenedioxy)-5,9-dimethylundecane (20). To a solution of alcohol **19** (55.5 mg, 0.20 mmol) in dry CH₂Cl₂ (3 cm³) were added dry pyridine (0.050 cm³, 0.60 mmol) and *O*-phenyl chlorothionoformate (0.032 cm³, 0.23 mmol) at 0 °C. The solution was stirred at room temperature for 3 h. The reaction mixture was then washed with water, saturated sodium hydrogencarbonate solution, and saturated brine, and then dried over anhydrous sodium sulfate. The solvent was evaporated in vacuo and the residue was purified by column chromatography on silica gel [eluent: hexane–ethyl acetate (60:1)] to give phenoxythiocarbonyl ester **20** (62.1 mg, 75%) as an oil. $[\alpha]_{\text{D}}^{23}$ –18.7

(*c* 1.74, CHCl₃); ^1H NMR δ 0.85 (3H, d, $J=6.4$ Hz, CH₃), 0.86 (3H, d, $J=6.4$ Hz, CH₃), 0.98 (3H, d, $J=6.8$ Hz, CH₃), 1.22–1.43 (10H, m, 4×CH₂, 2×CH), 1.39 (3H, s, CH₃), 1.48 (3H, s, CH₃), 1.62 (1H, m, CH), 1.84 (1H, ddd, $J=14.1$, 10.0, 3.9 Hz, CH), 3.94 (1H, dd, $J=8.3$, 6.4 Hz, CH–O), 4.12 (1H, m, CH–O), 4.33 (1H, m, CH–O), 5.69 (1H, m, PhOC(=S)OCH), 7.09–7.44 (5H, m, Ph); ^{13}C NMR δ 11.49, 19.32, 19.78, 24.37, 25.32, 26.32, 29.03, 29.51, 34.42, 36.83, 37.36, 37.86, 65.55, 76.91, 82.48, 109.86, 121.87, 126.47, 129.41, 153.28, 195.09; MS m/z 393 (M^+ –Me, 10%), 254 (54), 239 (56), 197 (32), 179 (47), 149 (48), 127 (100), 123 (45), 101 (55), 97 (47), 69 (95); HRMS calcd for C₂₂H₃₃O₄S [M^+ –Me] 393.2099, found 393.2053.

4.1.12. (2S,5R,9R)-1,2-(Isopropylidenedioxy)-5,9-dimethylundecane (21). To a solution of **20** (23.8 mg, 0.059 mmol) in dry toluene (2 cm³) were added *n*-Bu₃SnH (0.027 cm³, 0.12 mmol) and AIBN (12.8 mg, 0.064 mmol). The mixture was stirred at 85 °C for 3 h. KF and water were then added and the reaction mixture was stirred at room temperature overnight. After filtration through a pad of Florisil, the filtrate was evaporated in vacuo and the residue was purified by column chromatography on silica gel to give **21** (14.1 mg, 95%) as an oil. $[\alpha]_{\text{D}}^{23}$ +10.8 (*c* 1.95, CHCl₃); ^1H NMR δ 0.84 (3H, d, $J=6.9$ Hz, CH₃), 0.85 (3H, t, $J=7.6$ Hz, CH₃), 0.87 (3H, d, $J=6.3$ Hz, CH₃), 1.08–1.45 (13H, m, 5×CH₂, 3×CH), 1.36 (3H, s, CH₃), 1.41 (3H, s, CH₃), 1.68 (1H, m, CH), 3.52 (1H, m, CH–O), 4.02–4.07 (2H, m, CH₂–O); ^{13}C NMR δ 11.48, 19.32, 19.67, 24.47, 25.81, 27.02, 29.50, 31.17, 32.82, 32.90, 34.44, 36.95, 37.19, 69.58, 76.49, 108.52; MS m/z 241 (M^+ –Me, 100%), 101 (9); HRMS calcd for C₁₅H₂₉O₂ [M^+ –Me] 241.2246, found 241.2218.

4.1.13. (2S,5R,9R)-5,9-Dimethyl-1,2-undecanediol (22). To a solution of **21** (32.1 mg, 0.12 mmol) in THF–H₂O (1:1, 2 cm³) was added 1 mol dm^{–3} HCl at 0 °C. The solution was stirred at room temperature overnight. The reaction mixture was then extracted with ethyl acetate. The extract was washed with saturated brine, and then dried over anhydrous sodium sulfate. The solvent was evaporated in vacuo and the residue was purified by column chromatography on silica gel [eluent: hexane–ethyl acetate (3:1)] to give **22** (24.0 mg, 89%) as an oil. $[\alpha]_{\text{D}}^{23}$ –3.4 (*c* 1.0, CHCl₃); ^1H NMR δ 0.84 (3H, d, $J=5.9$ Hz, CH₃), 0.85 (3H, t, $J=7$ Hz, CH₃), 0.87 (3H, d, $J=6.8$ Hz, CH₃), 1.06–1.45 (14H, m, 6×CH₂, 2×CH), 2.02 (1H, br s, OH), 2.13 (1H, br s, OH), 3.45 (1H, m, CHOH), 3.66–3.68 (2H, m, CH₂OH); ^{13}C NMR δ 11.48, 19.32, 19.63, 24.51, 29.50, 30.75, 32.73, 32.87, 34.44, 36.96, 37.34, 66.89, 72.66; MS m/z 185 (M^+ –CH₂OH, 100%), 111 (43), 97 (64), 83 (56), 69 (50); HRMS calcd for C₁₂H₂₅O [M^+ –CH₂OH] 185.1905, found 185.1854.

4.1.14. (4R,8R)-4,8-Dimethyldecanal (1). To a solution of **22** (24.0 mg, 0.11 mmol) in aqueous acetonitrile (60%, 2 cm³) was added NaIO₄ (18.4 mg, 0.079 mmol) at 0 °C and the mixture was stirred at 0 °C for 1 h. The reaction mixture was then filtered through a pad of Celite and extracted with chloroform. The extract was washed with water and saturated brine, and then dried over anhydrous sodium sulfate. The solvent was evaporated in vacuo and the residue was purified by column chromatography on silica gel

[eluent: hexane–ethyl acetate (60:1)] to give (4*R*,8*R*)-4,8-dimethyldecanal (**1**) (18.5 mg, 91%) as an oil. $[\alpha]_D^{23}$ -5.6 (c 1.1, CHCl₃); IR (neat) 2961, 2929, 2714, 1712, 1463, 1379 cm⁻¹; ¹H NMR δ 0.84 (3H, d, $J=6.3$ Hz, CH₃), 0.85 (3H, d, $J=7.2$ Hz, CH₃), 0.88 (3H, d, $J=5.6$ Hz, CH₃), 1.04–1.14 (12H, m, 5×CH₂, 2×CH), 2.40–2.46 (2H, m, CH₂CH=O), 9.78 (1H, t, $J=1.2$ Hz, CH=O); ¹³C NMR δ 11.47, 19.30, 19.45, 24.43, 28.92, 29.48, 32.46, 34.43, 36.88, 37.07, 41.77, 202.95; MS m/z 184 (M⁺, 1%), 140 (67), 111 (42), 85 (54), 81 (57), 70 (90), 57 (100); HRMS calcd for C₁₂H₂₄O [M⁺] 184.1827, found 184.1802.

4.1.15. α -Methoxy- α -(trifluoromethyl)phenylacetate **24**.

¹H NMR δ 0.88 (3H, d, $J=6.3$ Hz, CH₃), 1.00–1.40 (4H, m), 1.25 (3H, t, $J=7.3$ Hz, CH₃), 1.29 (3H, t, $J=7.0$ Hz, CH₃), 1.33 (3H, s, CH₃), 1.41 (3H, s, CH₃), 1.46–1.60 (1H, m), 1.65 (1H, ddd, $J=14.1, 10.7, 3.4$ Hz), 1.89 (1H, m), 1.99 (1H, ddd, $J=14.1, 11.9, 2.4$ Hz), 2.05 (1H, dd, $J=15.1, 7.8$ Hz), 2.15–2.25 (1H, m), 2.23 (1H, dd, $J=14.6, 5.9$ Hz), 3.59 (3H, s, OCH₃), 3.76 (1H, dd, $J=8.3, 6.4$ Hz, CHH–O), 3.99 (1H, dd, $J=8.3, 6.4$ Hz, CHH–O), 4.12 (2H, q, $J=7.3$ Hz, CH₂–O), 4.17 (2H, q, $J=7.0$ Hz, CH₂–O), 4.22 (1H, m, CH–O), 5.13 (1H, m, CH–O), 7.39–7.43 (3H, m, Ph), 7.57–7.60 (2H, m, Ph); ¹³C NMR δ 14.30, 14.34, 19.61, 24.34, 24.94, 26.21, 30.14, 32.28, 33.18, 36.26, 41.02, 41.78, 55.54, 60.15, 60.59, 65.58, 74.30, 76.23, 77.20, 109.75, 127.36, 128.31, 129.58, 132.03, 165.90, 173.01, 175.06.

4.1.16. α -Methoxy- α -(trifluoromethyl)phenylacetate **25**.

¹H NMR δ 0.91 (3H, d, $J=6.4$ Hz, CH₃), 1.13 (1H, m), 1.25 (3H, t, $J=7.0$ Hz, CH₃), 1.20–1.35 (3H, m), 1.27 (3H, t, $J=67.0$ Hz, CH₃), 1.30 (3H, s, CH₃), 1.36 (3H, s, CH₃), 1.43 (1H, m), 1.56–1.64 (1H, m), 1.69 (1H, ddd, $J=14.1, 10.2, 3.9$ Hz), 1.92 (1H, m), 2.03–2.13 (2H, m), 2.25 (1H, dd, $J=15.0, 6.0$ Hz), 2.39 (1H, m), 3.69 (1H, dd, $J=8.5, 6.9$ Hz, CHH–O), 3.52 (3H, s, OCH₃), 3.91 (1H, dd, $J=8.5, 6.4$ Hz, CHH–O), 4.10 (1H, m, CH–O), 4.12 (2H, q, $J=6.4$ Hz, CH₂–O), 4.16 (2H, q, $J=7.0$ Hz, CH₂–O), 5.14 (1H, m, CH–O), 7.39–7.43 (3H, m, Ph), 7.57–7.60 (2H, m, Ph); ¹³C NMR δ 14.29, 14.35, 19.67, 24.43, 25.17, 26.26, 30.18, 33.08, 33.27, 36.37, 41.26, 41.81, 55.36, 60.16, 60.63, 65.93, 74.34, 76.35, 109.69, 127.55, 128.26, 128.38, 129.66, 131.62, 165.90, 173.01, 174.96.

References and notes

- Bell, S.; Wüstenberg, B.; Kaiser, S.; Menges, F.; Netscher, T.; Pfaltz, A. *Science* **2006**, *311*, 642–644.
- (a) Mori, K. *Tetrahedron* **1989**, *45*, 3233–3298; (b) Mori, K. *The Total Synthesis of Natural Products*; ApSimon, J., Ed.; Wiley: New York, NY, 1992; Vol. 9, pp 1–534.
- (a) Boucher, Y.; Kamekura, M.; Doolittle, W. F. *Mol. Microbiol.* **2004**, *52*, 515–527; (b) Eguchi, T. *Yuki Goseki Kagaku Kyokaishi* **2005**, *63*, 1069–1079.
- (a) Nagano, H.; Toi, S.; Yajima, T. *Synlett* **1999**, 53–54; (b) Nagano, H.; Matsuda, M.; Yajima, T. *J. Chem. Soc., Perkin Trans. 1* **2001**, 174–182; (c) Nagano, H.; Toi, S.; Hirasawa, T.; Matsuda, M.; Hirasawa, S.; Yajima, T. *J. Chem. Soc., Perkin Trans. 1* **2002**, 2525–2538; (d) Nagano, H.; Ohkouchi, H.; Yajima, T. *Tetrahedron* **2003**, *59*, 3649–3663; (e) Yajima, T.; Okada, K.; Nagano, H. *Tetrahedron* **2004**, *60*, 5683–5693.
- For the isolation of (4*R*,8*R*)-4,8-dimethyldecanal (**1**), see: (a) Suzuki, T. *Agric. Biol. Chem.* **1980**, *44*, 2519–2520; (b) Suzuki, T. *Agric. Biol. Chem.* **1981**, *45*, 1357–1363; (c) Suzuki, T.; Nakakita, H.; Kuwahara, Y. *Appl. Entomol. Zool.* **1987**, *22*, 340–347; (d) Arnaud, L.; Lognay, G.; Verscheure, M.; Leenaers, L.; Gaspar, C.; Haubruge, E. *J. Chem. Ecol.* **2002**, *28*, 523–532.
- For the biosynthesis of (4*R*,8*R*)-4,8-dimethyldecanal (**1**), see: Kim, J.; Matsuyama, S.; Suzuki, T. *J. Chem. Ecol.* **2005**, *31*, 1381–1400.
- For the synthesis of four possible stereoisomers of 4,8-dimethyldecanal and the absolute configuration of the pheromone, see: (a) Suzuki, T.; Mori, K. *Appl. Entomol. Zool.* **1983**, *18*, 134–136; (b) Mori, K.; Kuwahara, S.; Ueda, H. *Tetrahedron* **1983**, *39*, 2439–2444; (c) Levinson, H. Z.; Mori, K. *Naturwissenschaften* **1983**, *70*, 190–192.
- The natural pheromone of *T. castaneum* is suggested to be an 8:2 mixture of (4*R*,8*R*)- and (4*R*,8*S*)-4,8-dimethyldecanals because the mixture is about 10 times more active than (4*R*,8*R*)-isomer alone. Suzuki, T.; Kozaki, J.; Sugawara, R.; Mori, K. *Appl. Entomol. Zool.* **1984**, *19*, 15–20.
- For other syntheses of (4*R*,8*R*)-4,8-dimethyldecanal (**1**), see: (a) Mori, K.; Kato, M.; Kuwahara, S. *Liebigs Ann. Chem.* **1985**, 861–865; (b) Nguyen, Kong Hao; Ceskis, B.; Mavrov, M. V.; Moiseenkov, A. M.; Serebryakov, E. P. *Zh. Org. Khim.* **1987**, *23*, 498–503; (c) Nguyen, Kong Hao; Mavrov, M. V.; Serebryakov, E. P. *Zh. Org. Khim.* **1987**, *23*, 1649–1653; (d) Ceskis, B.; Lebedeva, K. V.; Moiseenkov, A. M. *Izv. Akad. Nauk SSSR, Ser. Khim.* **1988**, 865–871; (e) Mavrov, M. V.; Nguyen, Cong Hao; Serebryakov, E. P. *Bioorg. Khim.* **1989**, *15*, 123–126; (f) Fuganti, C.; Grasselli, P.; Servi, S.; Hoegberg, H. E. *J. Chem. Soc., Perkin Trans. 1* **1988**, 3061–3065; (g) Chênevert, R.; Desjardins, M. *J. Org. Chem.* **1996**, *61*, 1219–1222; (h) Sankaranarayanan, S.; Sharma, A.; Chattopadhyay, S. *Tetrahedron: Asymmetry* **2002**, *13*, 1373–1378.
- Schmid, C. R.; Bryant, J. D.; Dowlatzedah, M.; Phillips, J. L.; Prather, D. E.; Schantz, R. D.; Sear, N. L.; Vianco, C. S. *J. Org. Chem.* **1991**, *56*, 4056–4058.
- (a) Chattopadhyay, A.; Salaskar, A. *Synthesis* **2000**, 561–564; (b) Bhalay, G.; Clough, S.; McLaren, L.; Sutherland, A.; Willis, C. L. *J. Chem. Soc., Perkin Trans. 1* **2000**, 901–910.
- Férmeier, S.; Griep-Raming, J.; Hayen, A.; Metzger, J. O. *Chem.—Eur. J.* **2005**, *11*, 5545–5554.
- (a) Beifuss, U.; Tietze, M. *Tetrahedron Lett.* **2000**, *41*, 9759–9763; (b) Kolb, M.; Barth, J. *Synth. Commun.* **1981**, *11*, 763–967.
- Zarbin, P. H. G.; Cruz, W. de O.; Ferreira, J. T. B. *J. Braz. Chem. Soc.* **1998**, *9*, 511–513.