

ORIGINAL ARTICLE

Cytosine-type nucleosides from marine-derived *Streptomyces rochei* 06CM016

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Rocheicoside A (3), a nucleoside analog possessing a novel 5-(hydroxymethyl)-5-methylimidazolidin-4-one substructure, was isolated from marine-derived actinomycete *Streptomyces rochei* 06CM016, together with a new (4) and three known compounds. Structures of the new metabolites were elucidated by one-dimensional (¹H and ¹³C) and 2D NMR (COSY, HMQC and HMBC) and HR-TOF-MS analyses. All the metabolites exhibited significant antimicrobial activity. A plausible mechanism was proposed for compound 3's formation from amicetin.

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INTRODUCTION

Antibiotics are natural compounds produced by microorganisms as secondary metabolites to kill or inhibit other microorganisms. As their first discovery in the middle of the twentieth century, they had an important role in the treatment of infectious diseases. Up to now, pharmaceutical industry has primarily targeted drugs from soil organisms, and of the antibiotics in clinical use, most are of bacterial or fungal origin. Among the bacteria, Actinobacteria are particularly noteworthy, as they produce antibiotics such as streptomycin, neomycin, chloramphenicol, chlortetracycline and erythromycin.^{1,2} Members of Actinomycetes, especially *Streptomyces*, are fruitful organisms, producing ~80% of the antibiotics.^{3,4}

The acquired resistance of pathogen microorganisms due to their high mutation rates, toxic effects of some known antibiotics on human health and the existence of naturally resistant bacteria require the discovery of novel antibiotics.^{5,6} As antibiotic discovery efforts result in identified secondary metabolite frameworks with known mode of actions, recently, the researchers have started to focus on microorganisms living in extreme conditions, namely ocean and ocean bottoms, to discover novel chemical entities.^{2,7,8} It is thought that these novel compounds in general are produced as response to stress conditions.⁹ This research line led to the discovery of many new microbial species producing bioactive compounds.^{10,11} Recent studies have revealed that unusual marine Actinomycetes are good candidates for the discovery of new natural products.¹²

As part of our ongoing studies,¹³ the aim of this study was to discover new antimicrobial molecules from marine Actinomycetes living in unexplored areas of Mediterranean Sea.

RESULTS AND DISCUSSION

The isolation process performed using the sediment sample resulted in acquiring a bioactive mesophilic actinomycete 06CM016. The isolate showed antibacterial activity against *Escherichia coli* (17 mm) and MRSA (22 mm), and antifungal activity against *Candida albicans* (37 mm). Sequencing using primers 27F and 1492R revealed that the isolate was a member of *Streptomyces* genus. When the molecular identification evaluated with the phenotypical results (data not shown), the isolate was identified as *Streptomyces rochei*. After large-scale fermentation, 1.92 g of EtOAc extract was acquired. Bioassay-guided fractionation using the extract led to the isolation of five compounds (1–5), of which 1 (plicacetin), 2 (norplicacetin) and 5 were identified as cytosine-type nucleoside derivatives, by comparison of their spectral data with those previously reported (Figure 1).^{14–17}

The high resolution time of flight mass spectrum (HR-TOF-MS) of 3 [$[\alpha]_D^{25} = 5.7$ (c 0.0035, MeOH)] gave molecular ions of m/z 631.3123 [(+) mode] and 629.2387 [(-) mode], attributed to $[M+H]^+$ and $[M-H]^-$, respectively. Together with the ¹³C NMR data, accurate mass data supported a molecular formula of C₃₀H₄₂N₆O₉. Initial inspection of the one-dimensional NMR spectra (¹H, ¹³C and DEPT) also suggested a cytosine derivative, similar with compounds 1 and 2.

¹H NMR spectrum of 3 exhibited a tertiary and two secondary up-field methyl groups (δ 1.10, 1.16 and 1.27, respectively), a tertiary lower field methyl group (δ 2.38, 6H) attributable to a N(CH₃)₂ unit, a *para*-disubstituted aromatic ring (δ 7.79, 2H, d, $J = 8.8$ Hz; δ 8.07, 2H, d, $J = 8.8$ Hz) and two vinyl protons (δ 7.34, 1H, d, $J = 7.6$ Hz; δ 8.16, 1H, d, $J = 7.6$ Hz; Table 1). The chemical shifts of the latter protons are typical of the CH(5) and CH(6) resonances of a cytosine unit.^{18–20} On the basis of the HMQC connectivity, this assumption was substantiated, which indicated characteristic C-5 and C-6 signals at 97.1 and 145.7 p.p.m., respectively.

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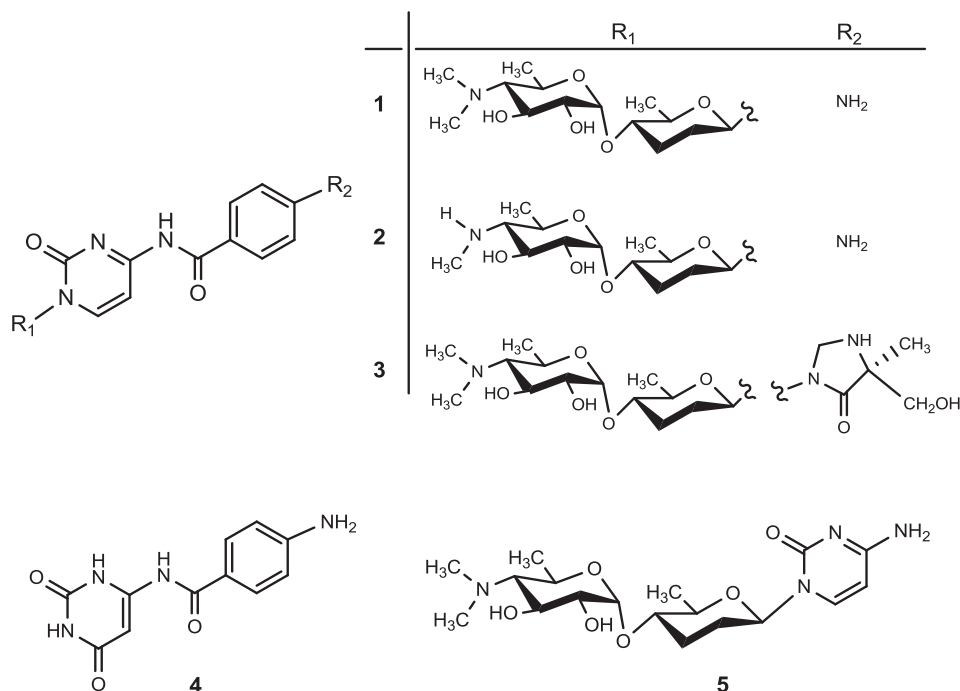


Figure 1 Structures of 1–5.

In addition, ¹H NMR spectrum of **3** displayed typical signals for two sugar units, namely amicitose and amosamine as in **1** and **2**.^{16–20} The two major spin systems of the disaccharidic part were deduced from the COSY spectrum starting from the well resolved anomeric protons (Figure 2). The H-1' resonance of amicitose (δ 5.73) showed cross peaks with two methylene protons (H-2': δ 1.73 and 2.07), which, in turn, coupled with H₂-3' (δ 1.50 and 2.29). H₂-3' exhibited correlation with H-4' (δ 3.36), whereas the latter proton showed cross peaks with H-5' (δ 3.65), which coupled with an up-field methyl signal at δ 1.16 (H₃-6'). In a similar manner, the spin system of amosamine moiety could be traced from the low-field H-1'' (δ 4.81) to H₃-6'' (1.27; Table 1). The use of HMQC spectrum of **3** allowed the assignment of the carbon signals ascribable to the saccharides, which also verified the presence of inner amicitose and terminal amosamine moieties. The absolute configurations of these deoxy sugars were suggested to be D based on the same biosynthetic origin as in **1**, **2** and **5**. This assumption was substantiated by carrying out acidic hydrolysis on compound **3**.

The interfragment relationship deduced from the HMBC spectrum helped us to locate the O- and N-glycosidic linkages, shown in Figure 2. Thus, the anomeric proton of the amosamine unit (δ 4.81) displayed a long-range correlation to C-4' of amicitose (δ 73.6), whereas the anomeric carbon of the latter sugar showed a ³J C–H heteronuclear correlation with H-6 of cytosine moiety (δ 8.16). Moreover, the methyl signal observed at δ 2.38 (6H) was readily assigned to C-4''-[N(CH₃)₂] on the basis of a long-range connectivity to C-4'' (δ 70.5).

The combined use of COSY, HMQC and HMBC spectra also permitted the complete assignment of the aromatic ring (Table 1, Figure 2) as *p*-substituted benzoyl extending from cytosine moiety. The proton chemical shift of H-11(13) was notably deshielded compared with those in compound **1** (δ 7.79 and 6.57, respectively), implying an electron-withdrawing group at *p*-position instead of an

electron donating amino group of *p*-amino benzoic acid, possibly an amide functionality.

After exclusion of the above-mentioned parts, the remaining side chain of the molecule had the formula C₅H₉N₂O₂. On the basis of the comprehensive examination of one-dimensional and 2D NMR data of **3**, and the comparison of these data with those of related metabolites, well-known cytosine-type antibiotics, namely amicitin, bamicitin and oxamicitin,^{16,19,20–22} it was inferred that a framework derived from α-methylserine was connected to the benzoyl unit via an amide bond. Four of the left over carbons at δ 176.9, 66.2, 65.6, 19.3 (C-16 (s), C-19 (t), C-17 (s), and C-20 (q), respectively) in the ¹³C NMR spectrum, a singlet methyl at δ 1.10 (H₃-20) and two methylene protons at δ 3.32 and 3.62 (each d, *J* = 5.6 Hz, H₂-19) in the ¹H NMR spectrum were evident for the integration of the aforementioned amino-acid moiety. Furthermore, in the HMBC spectrum, long-range correlations between C-16/C-17 and H₃-20/H₂-19 substantiated our assumption. The remaining NMR signals indicated an isolated methylene group (H₂-21: δ 4.73, 4.85, each d, *J* = 7.6 Hz; C-21: δ 63.1, t), secured by the COSY and HMQC spectra. One of the two unsaturation numbers calculated for the side chain was readily accounted for the amide carbonyl at δ 176.9, suggestive of a cyclization. Thus, incorporation of the remaining carbon should afford a ring system. In this case, there are two possibilities, either –N–CH₂–N– or –N–CH₂–O– bridges. In case of the latter (oxazolidine ring), C-21 was supposed to be resonated around 80–85 p.p.m. in the ¹³C NMR spectrum.^{23–25} However, the observed carbon data (δ 63.1) was entirely compatible with –N–CH₂–N– connection,²⁶ referring a 5-(hydroxymethyl)-5-methylimidazolidin-4-one skeleton. The stereochemistry of C-17 was assigned by a combination of 2D NOESY data and molecular modeling studies performed on **3**. In regard to the absolute configuration of C-17, the minimum energy conformations of *R* and *S* stereoisomers were calculated with a three-dimensional computer-generated model. In the case of *R*-configuration at C-17 (*D*-serine derivative), CH₃-20 is required to show NOE interactions

Table 1 ^{13}C and ^1H NMR data of **3** and **4** recorded in DMSO-d_6 (100 and 400 MHz, respectively)

C/H	δ_{C} (3) (p.p.m.)	δ_{H} (3) (p.p.m.), J (Hz)	δ_{C} (4) (p.p.m.)	δ_{H} (4) (p.p.m.), J (Hz)
1	—	—	—	—
2	154.3	—	151.5	—
3	—	—	—	—
4	163.6	—	151.3	—
5	97.1	7.34 d (7.6)	84.7	6.06 s
6	145.7	8.16 d (7.6)	165.4	—
7	—	—	—	—
8	167.2	—	166.5	—
9	128.3	—	119.2	—
10	130.1	8.07 d (8.8)	130.5	7.80 d (8.4)
11	117.6	7.79 d (8.8)	113.1	6.59 d (8.4)
12	142.4	—	153.8	—
13	117.6	7.79 d (8.8)	113.1	6.59 d (8.4)
14	130.1	8.07 d (8.8)	130.5	7.80 d (8.4)
15	—	—	—	6.00 br s
16	176.9	—	—	—
17	65.6	—	—	—
18	—	—	—	—
19	66.2	3.32 d, 3.62 d (5.6)	—	—
20	19.3	1.10 s	—	—
21	63.1	4.73, 4.85, each d (7.6)	—	—
1'	82.6	5.73 dd (1.6, 10.4)	—	—
2'	29.2	1.73 m, 2.07m	—	—
3'	27.0	1.50 m, 2.29 d (10.8)	—	—
4'	73.6	3.36 m	—	—
5'	76.9	3.65 m	—	—
6'	19.6	1.16 d (6.0)	—	—
1''	95.2	4.81 d (3.6)	—	—
2''	73.4	3.27 dd (3.6, 9.6)	—	—
3''	68.7	3.77 t (10.0)	—	—
4''	70.5	2.00 m	—	—
4''-N(CH₃)₂	42.6	2.38 s	—	—
5''	66.2	3.63 m	—	—
6''	18.9	1.27 d (6.0)	—	—

with CH_2 -21 due to their close proximity (2.64 Å) above the rigid imidazolidin-4-one ring system, whereas the *S*-configuration (l-serine) is predicted to provide a correlation between CH_2 -19 and CH_2 -21 (2.58 Å) (calculated CH_3 -20/ CH_2 -21 distance = 4.07 Å). On the basis of the correlation between CH_3 -20 and CH_2 -21 observed in the 2D NOESY spectrum, the *R*-configuration of C-17 was concluded (see Supplementary Information).

A detailed literature search revealed that compound **3** was the first nucleoside derivative possessing imidazolidin-4-one framework. This prompted us to engage in the biosynthetic means of this exceptional ring system. Although the structure of amicetin, the originator of **3**, was established in 1962, characterization of its biosynthesis was reported in 2012 by Zhang *et al.*¹⁶ In that study, first, the amicetin biosynthesis gene cluster from *S. vinaceusdrappus* NRRL 2363 was cloned, followed by heterologous expression in *S. lividans* TK64, which resulted in the production of amicetin and its analogs, confirming the

identity of the gene cluster. Total of 21 genes, including **3** for regulation and transportation, 10 for disaccharide biosynthesis and 8 for the formation of the amicetin skeleton by the linkage of cytosine, *p*-amino benzoic acid and the terminal α -methylserine moieties, were described putatively in amicetin biosynthesis by *in silico* sequence analysis. In the case of **3**, after biosynthesis of amicetin, an additional enzymatic step is required for a carbon transfer followed by a ring formation between N-15 and N-18 atoms. A comprehensive search pointed to $\text{N}^5, \text{N}^{10}$ -methylene tetrahydrofolate (THF), an important metabolite for the synthesis of purines, thymidine, methionine, choline and lipids as the major source of C1 units in biosynthesis. Conversion of THF to 5,10-methylene-THF is one of the well-known enzymatic processes: the serine hydroxymethyltransferase, a pyridoxal 5'-phosphate-dependent enzyme, converts serine to glycine, transferring a methyl group to THF, thus producing 5,10-methylene-THF (Figure 3).^{26–28} With respect to the additional carbon's incorporation (C-21), a plausible mechanism is proposed (Figure 4), which involves a serine hydroxymethyltransferase homolog in generation of the imidazolidine ring.^{27–29}

To the best of our knowledge, such a methylene insertion between the *p*-amino of *p*-amino benzoic acid moiety and the nitrogen atom of D-serine residue is encountered for the first time in secondary metabolites.

In addition to **3**, another new metabolite (**4**) with a simple structural core was isolated. The HR-TOF-MS of **4** provided $[\text{M} - \text{H}]^-$ and $[\text{M} + \text{H}]^+$ at m/z 245.0593 and 247.0864, respectively, indicating the molecular formula $\text{C}_{11}\text{H}_{10}\text{N}_4\text{O}_3$. Inspection of ^1H and ^{13}C NMR spectra of **4** suggested the presence of a 6-monosubstituted uracil residue instead of the cytosine moiety established in compounds **1–3** and **5**.^{28–30} The substituted part was deduced to be *p*-amino benzamide as in the other isolates. The assignment of the ^1H and ^{13}C NMR signals of **4** was secured by COSY, HMQC and HMBC experiments (Table 1). On the basis of the spectral evidence, the structure of **4** was established as shown in Figure 1.

Amicetin-type nucleoside analogs are not common in nature, and about 10 derivatives have been identified so far, namely bamicitin,¹⁴ plicacetin,¹⁴ norplicacetin,¹⁵ oxamicetin,²¹ SF2457,³¹ oxyplicacetin³² and cytosaminomycins A to D.³³ Amicetin and its analogs have been described as potent antibiotics against microorganisms, including archaea, bacteria and eukarya. These compounds exhibit their activity via peptidyl transferase inhibition blocking protein biosynthesis.³⁴ Compounds **1–5** showed stronger activity than the positive controls against the test organisms with MIC values ranging from 4 to $16 \mu\text{g ml}^{-1}$ (Table 2), substantiating antibiotic property of cytosine-type nucleosides towards microorganisms from different origins.

MATERIALS AND METHODS

Sampling station and isolation

A marine sediment sample, collected from Kaş, Turkey at a depth of 35 m by Scuba diving, was used for isolation of marine actinomycetes. The isolation procedure on the sediment was performed according to the method of Maldonado *et al.*³⁵ M1 medium (10 g soluble starch, 4 g yeast extract, 2 g peptone and 15 g agar, 1000 ml seawater) was used for marine actinomycete isolation.³⁶

Antimicrobial activity

Before the molecular characterization, a preliminary fermentation was performed with broth version of the same medium at 28 °C for 14 days at 150 r.p.m. to confirm antimicrobial activity. Sequential 25-ml aliquots of the fermentation broth were removed from the culture every 4 days from day 2 to 14, and activities of the ethyl acetate extracts were monitored using disc diffusion method on Mueller-Hinton agar against four different bacteria

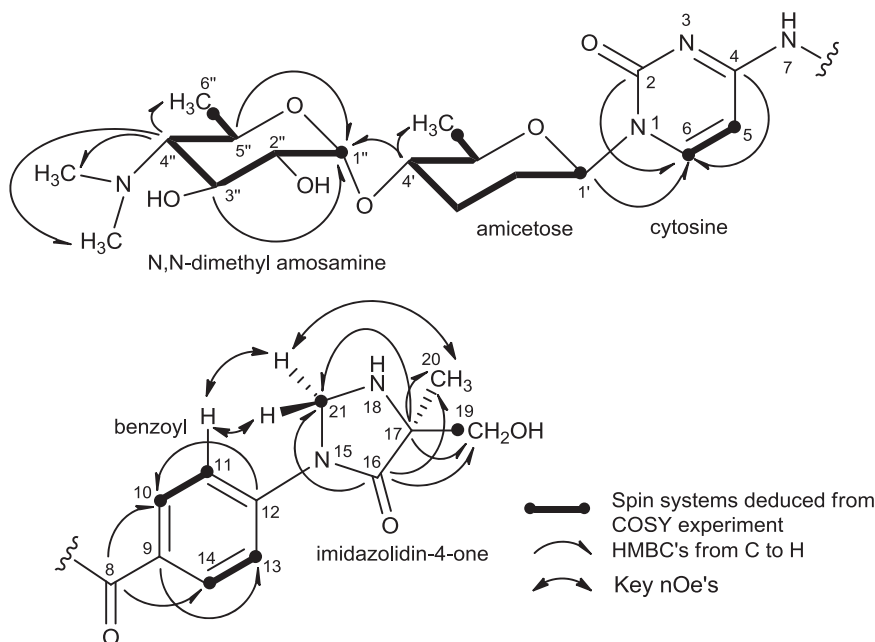


Figure 2 Spin systems and key long-range correlations of **3**.

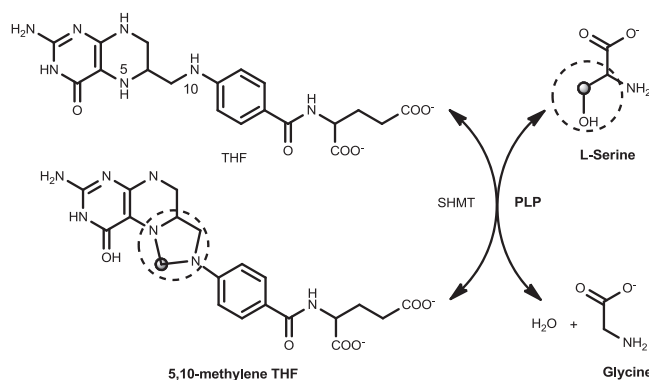


Figure 3 SHMT catalyzed 5,10-methylene-THF formation. PLP, pyridoxal 5'-phosphate; SHMT, serine hydroxymethyltransferase; THF, tetrahydrofolate.

(*Enterococcus faecium* DSM 13590: vancomycin resistant; *E. coli* O157:H7 RSKK 234: streptomycin, sulfisoxazole and tetracycline resistant; *Staphylococcus aureus*: methicillin resistant; *Pseudomonas aeruginosa* ATCC 27853) and a yeast species (*C. albicans* DSM 5817).

Molecular identification

DNA isolation was carried out by a modified method of Liu *et al.*³⁷ The isolate was identified using sequencing of the 16S ribosomal DNA region.³⁸ Primers for amplification of the 16S ribosomal RNA region were 27F (5'-AGA GTT TGA TCC TGG CTC AG-3') and 1492R (5'-TAC GGC TAC CTT GTT ACG ACT T-3'). The amplification reaction was performed with Corbett Cool Gradient Palm Cycler CGI-96 (Concorde, Sydney, NSW, Australia) under the following conditions as described elsewhere.¹³ Sequence analysis of 16S ribosomal RNA region was carried out on an ABI 3130XL automated sequencer (Applied Biosystems, Foster City, CA, USA). The sequence (JX912315.1) obtained was compared pairwise using a BLASTN search and aligned with the sequences of related species retrieved from GenBank, NCBI.

Large-scale fermentation

For large-scale fermentation, a loopful culture of *S. rochei* 06CM016 was inoculated in a flask containing 50 ml of M1 medium, and it was incubated at

28 °C for 3 days on a rotary shaker at 150 r.p.m. Five milliliter of the culture medium was transferred into 200 ml of M1 liquid medium in 1-l flasks, and further incubation was carried out at 28 °C for 10 days. In total, 56 l of fermentation liquid was obtained, and after removal of the cells by centrifugation, the broth was gradually extracted with EtOAc. After evaporation at 40 °C *in vacuo*, EtOAc extract was acquired (1.92 g; 0.034 g l⁻¹ yield).

Thin-layer chromatography

To compare chemical compositions of fractions exhibiting activity and to determine the required solvent systems for chromatographical separations, silica gel (aluminum foil, prepared plate, Kieselgel 60 F254, 0.2 mm, Merck, Art.5554, Merck KGaA, Darmstadt, Germany) TLC analyses were performed. Solvent systems consisting of CHCl₃:MeOH (9:1), CHCl₃:MeOH:H₂O (85:15:0.5, 80:20:2, 70:30:3 and 61:32:7), EtOAc:MeOH:H₂O (100:17.5:13.5) and Hexane:EtOAc:MeOH (10:10:2 and 10:10:3) were used.

Bioassay-guided fractionation

Silica gel (Kieselgel 60, 70–230 mesh, Merck), reversed-phase silica gel (RP-C18, Sephalryte 40 µm) and Sephadex LH-20 as stationary phases on preparative thin layer, open column and high-performance flash chromatography (Biotage, Uppsala, Sweden) systems (see isolation scheme in the Supplementary Material) were used for purification studies.

Acidic hydrolysis of compound 1

Compound **3** (3.1 mg) was refluxed in 1 M HCl for 4 h. After removal of the solvent, the residue was partitioned between CHCl₃ and H₂O. The H₂O solution was neutralized with 5% NaOH, and desalted on Sephadex LH-20 column by passing MeOH. The sugar fraction was concentrated *in vacuo* and purified on preparative TLC using CHCl₃:Me₂CO (1:1) to afford two saccharides, providing positive color reaction to anisaldehyde-H₂SO₄ (**3a**: 0.3 mg; **3b**: 0.4 mg). Positive optical rotation data were observed for both sugar moieties (D-amicetose: [α]_D²⁵ = +1.6 (c 0.0023, Me₂CO); D-amosamine [α]_D²⁵ = +1.3 (c 0.0024, Me₂CO)).

Broth microdilution method

MICs of the purified compounds were performed by the microdilution method described by CLSI.³⁹

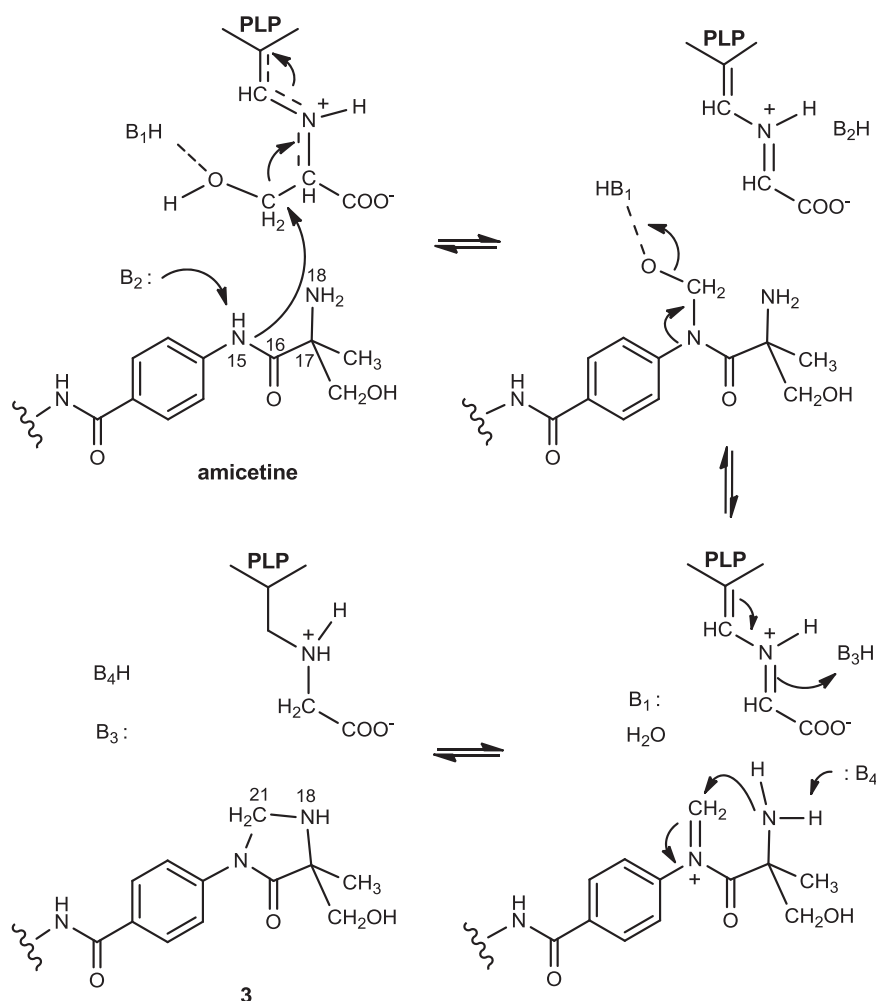


Figure 4 A plausible mechanism for the transformation of amicetin into compound **3**. PLP, pyridoxal 5'-phosphate.

Table 2 MIC values of compounds 1–5 and positive controls against the test organisms

Compounds	MIC ($\mu\text{g ml}^{-1}$)		
	<i>E. coli</i> O157:H7 RSKK 234	<i>MRSA</i> DSM 11729	<i>C. albicans</i> DSM 5817
1	8	4	8
2	16	8	4
3	16	8	4
4	16	16	8
5	16	16	8
Lincomycin	64	—	—
Nalidixicacid	4	32	—
Azithromycin	16	—	—
Cefaclor	8	32	—
Ceftriaxone	—	—	—
Clindamycin	16	4	—
Tetracycline	8	4	—
Ampicillin sulbactam	32	—	—
Amphotericin B	—	—	0.5
Fluconazole	—	—	8

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