Carbohydrate Research 357 (2012) 47-52

Contents lists available at SciVerse ScienceDirect

Carbohydrate Research

journal homepage: www.elsevier.com/locate/carres

Benzoylated ethyl 1-thioglycosides: direct preparation from per-O-benzoylated sugars

Deepak Sail, Pavol Kováč*

NIDDK, LBC, National Institutes of Health, Bethesda, MD 20892-0815, USA

ARTICLE INFO

Article history: Received 16 April 2012 Received in revised form 14 May 2012 Accepted 15 May 2012 Available online 24 May 2012

Keywords: β-Selective 1-O-benzoylation Thioglycosylation Thioglycosidation

ABSTRACT

p-Glucose, lactose, maltose, and melibiose were benzoylated with Bz₂O–Et₃N reagent to give fully benzoylated β products. Under the same conditions, p-mannose produced a mixture where the β-benzoate predominated. Treatment of the foregoing compounds with EtSH at slightly elevated temperature (50–60 °C) in the presence of BF₃·Et₂O as a promoter gave the corresponding ethyl 1-thio glycosides in high yields. The α-products predominated in all cases in the anomeric mixtures formed. Individual products of all reactions were isolated by chromatography, they were obtained in analytically pure state, and were fully characterized by ¹H and ¹³C NMR data and physical constants.

Published by Elsevier Ltd.

1. Introduction

Because of their long shelf-life, 1-thioglycosides are very popular glycosyl donors.^{1,2} While synthesis of per-O-acetylated 1-thioglycosides is fairly simple,^{3–6} their benzoylated counterparts are normally prepared indirectly from acetylated thioglycosides, by sequential deacetylation and benzoylation.^{7–12} Alternatively, fully benzoylated thioglycosides were prepared from the corresponding glycosyl thiocyanates¹³ or iodides.¹⁴ Cao et al.¹⁵ prepared ethyl 2,3,4,6-tetra-*O*-benzoyl-1-thio- β -D-galactopyranoside (**13**) from the corresponding penta-O-benzoate by treatment with EtSH in presence of SnCl₄.

Following reports on the use of 1-O- β -acetates of common sugars as glycosyl donors in trifluoromethanesulfonate (TMSOTf)-catalyzed glycosylation,^{16,17} we have extended the method to the use of the corresponding 1-O- β -benzoyl derivatives.¹⁸ More than a decade later, Gallo-Rodriguez et al. reported similar approach in their oligo-saccharide synthesis.¹⁹ Here, we describe a similar approach, but to glycosylation of EtSH, which allows direct preparation of 1-thiogly-cosides from fully-O-benzoylated saccharides using BF₃·Et₂O as a promoter (Scheme 1). To our knowledge, such conversion has not been described in accessible literature.

2. Results and discussion

Transfer of the acetyl group during glycosylation reactions from 2-O-acetylated glycosyl donors to the free hydroxyl group of glycosyl acceptors is a well established undesirable side reaction, which often occurs during Lewis acid mediated glycosylations.^{16,17,20} For example, formation of as much as ~30% of methyl 6-O-acetyl-2,3,4-tri-O-benzoyl- β -D-galactopyranoside was observed when methyl 2,3,4-tri-O-benzoyl- β -D-galactopyranoside was glycosylated with 1,2,3,4,6-penta-O-acetyl- β -D-galactopyranose in the presence of TMSOTf.¹⁸ The significance of 2-O-benzoylated glycosyl donors in glycoside or oligosaccharide synthesis lies in the fact that the aforementioned transesterification does not take place when such donors are used.^{18,21} Also, benzoylated glycosyl donors are often stable at base-deficient conditions, which are required for isomerization of the intermediate orthoesters into glycosides.²¹ In view of the above, developing a more straightforward protocol for making benzoylated 1-thio-glycosides than through the corresponding acetates is desirable.

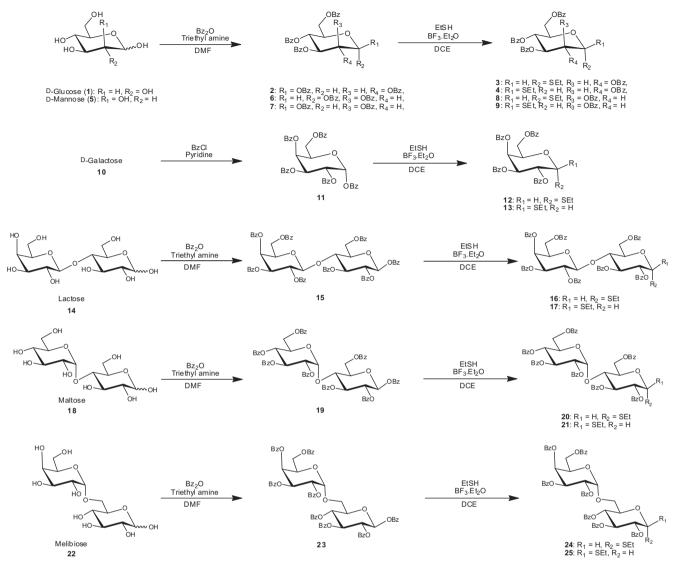
Knowing that 1,2-trans-O-acetates are normally more reactive than their 1,2-*cis* counterparts,^{1,2} we searched for a simple preparation of this class of benzoyl derivatives, and have found it in the work by Luo²² et al. who showed one example of using benzoic anhydride-triethylamine (Bz₂O-Et₃N) reagent in stereoselective 1-O-benzoylation. When treated in this way, D-glucose (1), lactose (14), maltose (18), and melibiose (22) produced virtually only β -anomers. The product formed from mannose (**5**) was an anomeric mixture of per-O-benzoates where the β-compound predominated (Table 1). There was no advantage to using the Bz₂O-Et₃N reagent for benzoylation of D-galactose (10). Several products were formed, one major of which co-chromatographed with the authentic sample^{22,23} of 1,2,3,4,6-penta-O-benzoyl- α -D-galactopyranose (11).^{23,24} It has been reported that high proportion of furanoses is formed during benzovlation of D-galactose at both elevated¹⁹ and sub-ambient²⁵ temperatures. The products were difficult to





^{*} Corresponding author. Tel.: +1 301 496 3569; fax: +1 30 480 5703. *E-mail address:* kpn@helix.nih.gov (P. Kováč).

^{0008-6215/\$ -} see front matter Published by Elsevier Ltd. http://dx.doi.org/10.1016/j.carres.2012.05.012



Scheme 1. Benzoylation of D-glucose, D-mannose, D-galactose, lactose, maltose, and melibiose, and thioglycosidation of their per-O-benzoyl derivatives.

Table 1Benzoylation of saccharides with benzoic anhydride: product composition and yields

Starting material	Product(s) formed	Combined yield %	α/β ratio ^a
D-Glucose (1)	2	91	β Only
D-Mannose (5)	6, 7	83	1/1.5
Lactose (14)	15	82	β Only
Maltose (18)	19	74	β Only
Melibiose (22)	23	75	β Only

^a Determined after separation by chromatography.

separate; hence accurate yield of individual compounds formed could not be determined. As we later discovered during this work, when thioglycosidation was conducted at elevated temperature 1,2-*cis*-O-benzoates served our purpose equally well, which agreed with Xu et al.'s²⁶ similar observation concerning the corresponding penta-O-acetyl derivatives. Thus, α -D-galactopyranose **11**^{22,23} was used in the conversion into the corresponding 1-thio-glycosides **12** and **13**.

Reactions of O-benzoyl derivatives with EtSH in the presence of BF_3 - Et_2O as a promoter were examined next. The reactions at room temperature were impractically slow. At moderately elevated temperature, mixtures of anomers were formed in all cases here

Table 2

Glycosidation of per-O-benzoates with EtSH: combined yields and product composition

Starting material	Products formed	Combined yield %	α/β ratio
2	3, 4	91	2 ^a
6	8, 9	93	8.7 ^a
7	8, 9	89	7.9 ^a
11	12, 13	80	2.7 ^b
15	16, 17	76	1.6 ^b
19	20, 21	90	2.8 ^b
23	24, 25	87	3.8 ^b

^a Ratio determined after separation by chromatography.

^b Ratio in the crude product determined by NMR.

examined (Table 2), with ethyl 1-thio- α -glycosides predominating. Anomerization of initially formed kinetic products in the presence of Lewis acid cannot be excluded. Ethyl 1-thio- β -glycosides of p-galactose and melibiose, (**13**) and (**25**), respectively, were found to co-chromatograph with the starting benzoates. This precluded monitoring of the progress of the reaction by thin-layer chromatography. In these situations small-scale experiments were conducted in NMR tubes using dichloroethane- d_4 as solvent, and the reaction times required for complete conversion (see Section 3) of benzoates **11** and **23** were determined by ¹H NMR spectroscopy, by monitoring the disappearance of the low-field signals for H-1 of the starting materials. All products were fully characterized and produced NMR spectra consistent with the expected structures.

3. Experimental

3.1. General methods

Optical rotations were measured at ambient temperature for solution in CHCl₃ with a digital Jasco automatic polarimeter. Model P-2000. Melting points were measured on a Kofler hot stage. All reactions were monitored by thin-layer chromatography (TLC) on Silica Gel 60 coated glass slides. Column chromatography was performed by elution from prepacked columns of silica gel (Varian, Inc.) with the Isolera Flash Chromatograph (Biotage). Nuclear Magnetic Resonance (NMR) spectra were measured at 400 MHz (¹H) and 100 MHz $({}^{13}C)$ or at 600 MHz $({}^{1}H)$ and 150 MHz $({}^{13}C)$ with Bruker Avance spectrometers. Assignments of NMR signals were made by homonuclear and heteronuclear 2-dimensional correlation spectroscopy, run with the software supplied with the spectrometers. When reporting assignments of NMR signals of disaccharides, sugar residues are serially numbered, beginning with the one bearing the aglycon, and are identified by a Roman numeral superscript in listings of signal assignments. Liquid Chromatography-Electron Spray-Ionization Mass Spectrometry (ESI-MS) was performed with a Waters LCT Premier spectrometer. Reactions were carried out in closed flasks. Rubber septa used to close reaction flasks containing organic solvents were protected with a thin Teflon[™] sheet (Laboratory Supplies Co., Inc., Hicksville, NY), to avoid leaching. Solutions in organic solvents were dried with anhydrous Na₂SO₄, and concentrated at 40 °C/2 kPa.

3.2. General procedures for benzoylation

The free sugar (2–5 mmol) was dissolved in DMF (10 mL/ mmol), Et₃N (4 equiv/OH) was added followed by addition of Bz₂O (2 equiv/OH), and the mixture was stirred at conditions specified below. When the conversion was complete (TLC), excess of MeOH was added and the mixture was stirred overnight at room temperature. After concentration, a solution of the residue in CH₂Cl₂ was washed with saturated solution of sodium bicarbonate. The organic phase was dried, concentrated, and the crude mixture was chromatographed, to give pure substances and, occasionally, mixed fractions. For combined yields, see Table 1.

3.3. General procedure for thioglycosidation

 $BF_3 \cdot Et_2O$ was added to a solution of EtSH and sugar per-O-benzoate in 1,2-dichloroethane, and the mixture was stirred until all starting material was consumed (TLC). For amounts of solvent, reagents, and reaction conditions, see individual preparations (below). Et_3N (excess of over the molar amount of $BF_3 \cdot Et_2O$) was added, the mixture was concentrated, and a solution of the residue in CH_2Cl_2 was washed with saturated solution of sodium bicarbonate. After concentration of the organic phase, chromatography of the material in the residue gave pure substances or mixtures of anomers. For combined yields, see Table 2.

3.3.1. 1,2,3,4,6-Penta-O-benzoyl-β-D-glucopyranose (2)

Reaction conditions: 16 h, rt; flash chromatography, 4:1 hexane-acetone; yield, starting from **1** (0.9 g, 5.0 mmol), 3.2 g (91%); $R_{\rm f}$ = 0.3 (4:1 hexane-acetone); Mp 182–185 °C (MeOH); [α]_D +24.4 (*c* 1.0, CHCl₃); lit.²⁷ Mp 189–192 °C (AcOH); [α]_D +24.2 (*c* 2.6, CHCl₃); ¹H NMR (600 MHz, CDCl₃): δ 6.30 (d, 1H, $J_{1,2}$ = 8.1 Hz, H-1), 6.05 (t, 1H, J = 9.5 Hz, H-3), 5.90–5.81 (m, 2H, H-2, H-4), 4.66 (dd, 1H, $J_{6a,6b} = 12.3$, $J_{5,6a} = 3.0$ Hz, H-6a), 4.52 (dd, 1H, $J_{5,6b} = 4.8$ Hz, H-6b), 4.41 (ddd, 1H, $J_{4,5} = 9.9$ Hz, H-5); ¹³C NMR (150 MHz, CDCl₃) δ 92.65 (C-1), 73.12 (C-5), 72.76 (C-3), 70.78 (C-2), 68.99 (C-4), 62.62 (C-6); ESI-HRMS calc. for C₄₁H₃₆NO₁₁ [M+NH₄]⁺: 718.2288; Found: 718.2292. Anal. Calcd For C₄₁H₃₂O₁₁: C, 70.28; H, 4.60. Found: C, 70.46; H, 4.65.

3.3.2. Ethyl 2,3,4,6-tetra-O-benzoyl-1-thio- α - (3) and β -D-glucopyranoside (4)

Reaction conditions: EtSH (0.45 mL, 6.0 mmol), $BF_3 \cdot Et_2O$ (0.554 mL, 4.4 mmol), 1 h, 50 °C; flash chromatography, $30:1 \rightarrow 20:1$ toluene–EtOAc; yield, starting from **2** (2.80 g, 4.0 mmol), 1.53 g (60%) of **3** and 0.80 g (31%) of **4**.

Compound **3**: $R_f = 0.7$ (15:1 toluene–EtOAc); Mp 137–138 °C (CH₂Cl₂–MeOH); $[\alpha]_D$ +104.2 (*c* 1.0, CHCl₃); lit.⁷ Mp 135 °C (toluene–pet. ether); $[\alpha]_D$ +104 (*c* 2.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 6.08 (t, 1H, J = 9.9 Hz, H-3), 5.95 (d, 1H, $J_{1,2} = 5.8$ Hz, H-1), 5.68 (t, 1H, J = 9.8 Hz, H-4), 5.51 (dd, 1H, $J_{2,3} = 10.2$ Hz, H-2), 4.88 (ddd, 1H, $J_{4,5} = 10.1$, $J_{5,6b} = 5.4$, $J_{5,6a} = 2.8$ Hz, H-5), 4.60 (dd, 1H, $J_{6a,6b} = 12.2$ Hz, H-6a), 4.52 (dd, 1H, $J_{5,6b} = 5.5$ Hz, H-6b), 2.70–2.55 (m, 2H, CH₂), 1.25 (t, 3H, J = 7.4 Hz, CH₃); ¹³C NMR (150 MHz, CDCl₃) δ 81.98 (C-1), 71.63 (C-2), 70.87 (C-3), 69.49 (C-4), 68.08 (C-5), 63.02 (C-6), 24.24 (CH₂), 14.61 (CH₃); ESI-HRMS calcd for C₃₆H₃₂O₉S [M+NH₄]⁺: 658.2111; found: 658.2123; Anal. Calcd for C₃₆H₃₂O₉S: C, 67.49; H, 5.03. Found: C, 67.71; H, 5.03.

Compound **4**: The solid obtained after freeze-drying of a solution in benzene could not be crystallized. $R_f = 0.6$ (15:1 toluene-EtOAc); $[\alpha]_D + 29.1$ (c 1.0, CHCl₃); lit.⁷ Mp 108–109 °C (ether–petroleum ether); $[\alpha]_D + 27$ (c 1.8, CHCl₃); ¹H NMR (600 MHz, CDCl₃): δ 5.93 (t, 1H, J = 9.5 Hz, H-3), 5.67 (t, 1H, J = 9.8 Hz, H-4), 5.57 (t, 1H, J = 9.7 Hz, H-2), 4.87 (d, 1H, $J_{1,2} = 10.0$ Hz, H-1), 4.63 (dd, 1H, $J_{6a,6b} = 12.2$, $J_{5,6a} = 3.1$ Hz, H-6a), 4.50 (dd, 1H, $J_{5,6b} = 5.5$ Hz, H-6b), 4.18 (ddd, 1H, $J_{4,5} = 10.0$ Hz, H-5), 2.84–2.69 (m, 2H, SCH₂CH₃), 1.26 (t, 3H, J = 7.4 Hz, SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃): δ 83.90 (C-1), 76.25 (C-5), 74.05 (C-3), 70.54 (C-2), 69.57 (C-4), 63.30 (C-6), 24.38 (SCH₂CH₃), 14.90 (SCH₂CH₃); ESI-HRMS calcd for C₃₆H₃₆NO₉S [M+NH₄]*: 658.2111; found: 658.2114. Anal. Calcd for C₃₆H₃₂O₉S: C, 67.49; H, 5.03. Found: C, 67.56; H, 5.01.

3.3.3. 1,2,3,4,6-Penta-O-benzoyl-α- (6) and β-D-mannopyranose (7)

Compound **5**: Reaction conditions: 40 h, rt; flash chromatography, $50:1 \rightarrow 30:1$ toluene–EtOAc; yield, starting from **5** (0.180 g, 1.0 mmol), 0.234 g (33%) of **6** and 0.342 g (50%) of **7**.

Compound **6**: $R_f = 0.5$ (19:1 toluene–EtOAc); Mp 150.5–152 °C (MeOH); $[\alpha]_D -20.2$ (*c* 1.0, CHCl₃); lit.²⁷ Mp 152–153 °C; $[\alpha]_D -18.6$ (*c* 1.6, CHCl₃); ¹H NMR (400 MHz, CDCl₃): δ 6.63 (d, 1H, $J_{1,2} = 2.0$ Hz, H-1), 6.28 (t, 1H, J = 10.2 Hz, H-4), 6.07 (dd, 1H, $J_{3,4} = 10.2$, $J_{2,3} = 3.3$ Hz, H-3), 5.91 (dd, 1H, H-2), 4.70 (dd, 1H, $J_{6a,6b} = 12.2$, $J_{5,6a} = 2.5$ Hz, H-6a), 4.57 (dt, 1H, $J_{4,5} = 10.0$ Hz, H-5), 4.50 (dd, 1H, $J_{5,6b} = 3.7$ Hz, H-6b); ¹³C NMR (100 MHz, CDCl₃) δ 91.37 (C-1, $J_{C-1,H-1} = 178.8$ Hz), 71.18 (C-5), 69.99 (C-3), 69.43 (C-2), 66.18 (C-4), 62.34 (C-6); ESI-HRMS calcd for C₄₁H₃₆NO₁₁ [M+NH₄]⁺: 718.2288; found: 718.2274. Anal. Calcd for C₄₁H₃₂O₁₁: C, 70.28; H, 4.60. Found: C, 70.56; H, 4.80.

Compound **7**: $R_f = 0.4$ (19:1 toluene–EtOAc); Mp 160–161.5 °C (MeOH); $[\alpha]_D - 84.2$ (*c* 1.0, CHCl₃); lit.²⁸ Mp 161–161.5 °C (EtOH); $[\alpha]_D - 80.7$ (CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 6.44 (d, 1H, $J_{1,2} = 1.2$ Hz, H-1), 6.18 (t, 1H, J = 9.8 Hz, H-4), 6.11 (dd, 1H, $J_{2,3} = 3.2$ Hz, H-2), 5.81 (dd, 1H, $J_{3,4} = 9.9$ Hz, H-3), 4.76 (dd, 1H, $J_{6a,6b} = 12.3, J_{5,6a} = 2.8$ Hz, H-6a), 4.56 (dd, 1H, $J_{5,6b} = 4.3$ Hz, H-6b), 4.38 (ddd, 1H, $J_{4.5} = 9.7$ Hz, H-5); ¹³C NMR (150 MHz, CDCl₃) δ 91.19 (C-1, $J_{C-1,H-1} = 163.3$ Hz), 73.28 (C-5), 71.51 (C-3), 69.36 (C-2), 69.32 (C-4), 62.61 (C-6); ESI-HRMS calcd for C₄₁H₃₆NO₁₁

$$\label{eq:constraint} \begin{split} \left[M\!+\!NH_4\right]^+\!\!: 718.2288; \mbox{ found: } 718.2292. \mbox{ Anal. Calcd for } C_{41}H_{32}O_{11} \!: \\ C, 70.28; \mbox{ H}, 4.60. \mbox{ Found: } C, 70.49; \mbox{ H}, 4.69. \end{split}$$

3.3.4. Ethyl 2,3,4,6-tetra-O-benzoyl-1-thio- α - (8) and β -D-mannopyranoside (9)

Reaction conditions: EtSH (1.5 mL, 2.0 mmol), BF₃·Et₂O (0.23 mL, 1.8 mmol), 3 h (starting from **6**), 5 h (starting from **7**), 50 °C; flash chromatography, $50:1 \rightarrow 25:1$ toluene–EtOAc.

Compound **8**: yield, starting from **6** (0.700 g, 1 mmol), 0.536 g (84%), starting from **7** (0.700 g, 1 mmol), 0.504 g (79%); $R_f = 0.7$ (15:1 toluene–EtOAc); Mp 120–121.5 °C (EtOH); $[\alpha]_D - 14.0$ (*c* 1.0, CHCl₃); lit.²⁹: Mp 124–125 °C (EtOH); $[\alpha]_D - 15.8$ (*c* 0.8, CHCl₃). ¹H NMR (600 MHz, CDCl₃) δ 6.17 (t, 1H, *J* = 10.0 Hz, H-4), 5.89 (dd, 1H, *J*_{3,4} = 9.9, *J*_{2,3} = 3.2 Hz, H-3), 5.87 (dd, 1H, *J*_{1,2} = 1.3 Hz, H-2), 5.59 (br d, 1H, H-1), 4.85 (ddd, 1H, *J*_{4,5} = 10.1, *J*_{5,6b} = 4.5, *J*_{5,6a} = 2.6 Hz, H-5), 4.69 (dd, 1H, *J*_{6a,6b} = 12.2 Hz, H-6a), 4.54 (dd, 1H, H-6b), 2.95–2.58 (m, 2H, SCH₂CH₃), 1.35 (t, 3H, SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃) δ 82.29 (C-1, *J*_{C-1,H-1} = 168.3 Hz), 72.08 (C-2), 70.48 (C-3), 69.19 (C-5), 66.99 (C-4), 62.82 (C-6), 25.57 (SCH₂CH₃), 1.481 (SCH₂CH₃); ESI-HRMS calcd for C₃₆H₃₆No₉S [M+NH₄]⁺: 658.2111; found: 658.2126. Anal. Calcd for C₃₆H₃₂O₉S: C, 67.49; H, 5.03. Found: C, 67.70; H, 5.09.

Compound **9**: Amorphous solid after freeze-drying of a solution in benzene; yield, starting from **6** (0.700 g, 1 mmol), 0.062 g (10%); yield, starting from **7** (0.700 g, 1 mmol), 0.061 g (10%); $R_f = 0.5$ (15:1 toluene–EtOAc); $[\alpha]_D -150.4$ (*c* 1.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 6.02 (dd, 1H, $J_{2,3} = 3.3$, $J_{1,2} = 0.9$ Hz, H-2), 6.00 (t, 1H, J = 10.0 Hz, H-4), 5.67 (dd, 1H, $J_{3,4} = 10.1$ Hz, H-3), 5.11 (br d, 1H, H-1), 4.71 (dd, 1H, $J_{6a,6b} = 12.1$, $J_{5,6a} = 2.9$ Hz, H-6a), 4.54 (dd, 1H, $J_{5,6b} = 5.4$ Hz, H-6b), 4.20–4.18 (m, 1H, H-5), 2.83–2.74 (m, 2H, SCH₂CH₃), 1.31 (t, 3H, J = 7.4 Hz, SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃) δ 82.84 (C-1, $J_{C-1,H-1} = 152.2$ Hz), 76.53 (C-5), 72.77 (C-3), 71.35 (C-2), 66.85 (C-4), 63.43 (C-6), 25.86 (SCH₂CH₃), 14.97 (SCH₂CH₃); ESI-HRMS calcd for C₃₆H₃₆NO₉S [M+NH₄]*: 658.2111; found: 658.2119; Anal. Calcd for C₃₆H₃₂O₉S: C, 67.49; H, 5.03. Found: C, 67.19; H, 4.99.

3.3.5. 1,2,3,4,6-Penta-O-benzoyl-α-D-galactopyranose (11)

This compound was prepared as described.²⁴ Mp 161.5–162.5 °C (EtOH); $[\alpha]_D$ +193.9 (*c* 1.0, CHCl₃); lit.²⁴ Mp 158–159 °C (MeOH); $[\alpha]_D$ +187.0 (*c* 4.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ : 6.96 (d, 1H, $J_{1,2} = 3.7$ Hz, H-1), 6.20 (br d, 1H, H-4), 6.13 (dd, 1H, $J_{2,3} = 10.7$, $J_{3,4} = 3.4$ Hz, H-3), 6.03 (dd, 1H, H-2), 4.84 (t, 1H, J = 6.8 Hz, H-5), 4.64 (dd, 1H, $J_{6a,6b} = 11.4$, $J_{5,6a} = 6.4$ Hz, H-6a), 4.43 (dd, 1H, $J_{5,6b} = 6.9$ Hz, H-6b); ¹³C NMR (150 MHz, CDCl₃): δ 90.65 (C-1), 69.41 (C-5), 68.50 (C-3), 68.43 (C-4), 67.66 (C-2), 61.80 (C-6). ESI-HRMS calcd for C₄₁H₃₆NO₁₁ [M+NH₄]⁺: 718.2288; found: 718.2258; Anal. Calcd for C₄₁H₃₂O₁₁: C, 70.28; H, 4.60. Found: 70.14; H, 4.70.

3.3.6. Ethyl 2,3,4,6-tetra-O-benzoyl-1-thio- α - (12) and β -D-galactopyranoside (13)

Reaction conditions: EtSH (0.375 mL, 5.0 mmol), $BF_3 \cdot Et_2O$ (0.504 mL, 4.0 mmol), 16 h (optimum time determined by NMR spectroscopy), 60 °C; flash chromatography, 5:1 hexane–acetone; yield, starting from **11** (1.40 g, 2.0 mmol), 0.422 g of pure **12** (33%), 0.317 g of pure **13** (25%). Unresolved mixture of **12** and **13** (0.282 g, 22%) was also obtained.

Compound **12**: Amorphous solid after freeze-drying of a solution in benzene; $R_{\rm f} = 0.4$ (3:1 hexane–acetone); $[\alpha]_{\rm D} + 166.6$ (*c* 1.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 6.09–5.99 (m, 2H, H-1, H-4), 5.92–5.82 (m, 2H, H-2, H-3), 5.03 (dd, 1H, $J_{5,6a} = 7.0, J_{5,6b} = 5.6$ Hz, H-5), 4.62 (dd, 1H, $J_{6a,6b} = 11.5$ Hz, H-6a), 4.44 (dd, 1H, H-6b), 2.74–2.49 (m, 2H, SCH₂CH₃), 1.25 (m, 3H, SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃) δ 81.22 ($J_{C-1,H-1} = 168.5$ Hz, C-1), 68.05, 67.96, 67.90 (C-2, C-3, C-4), 66.22 (C-5), 61.60 (C-6), 23.03 (SCH₂CH₃), 13.57 (SCH₂CH₃); ESI-HRMS calcd for C₃₆H₃₆NO₉S [M+NH₄]⁺:

658.2111; found: 658.2111; Anal. Calcd for C₃₆H₃₂O₉S: C, 67.49; H, 5.03. Found: C, 67.39; H, 5.00.

Compound **13**; Amorphous solid after freeze-drying of a solution in benzene; $R_f = 0.3$ (3:1 hexane–acetone); $[\alpha]_D +98.4$ (*c* 1.0, CHCl₃); lit.³⁰ $[\alpha]_D +106$ (*c* 1.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ : 6.04 (br d, 1H, H-4), 5.84 (t, 1H, *J* = 9.9 Hz, H-2), 5.65 (dd, 1H, $J_{2,3} = 9.9$, $J_{3,4} = 3.5$ Hz, H-3), 4.88 (d, 1H, $J_{1,2} = 10.0$ Hz, H-1), 4.67 (dd, 1H, $J_{6a,6b} = 11.2$, $J_{5,6a} = 6.5$ Hz, H-6a), 4.41 (dd, 1H, $J_{5,6b} = 6.5$ Hz, H-6b), 4.36 (t, 1H, J = 6.6 Hz, H-5), 2.95–2.72 (m, 2H, SCH₂CH₃), 1.32 (t, 3H, SCH₂SCH₃); ¹³C NMR (150 MHz, CDCl₃) δ : 84.31 ($J_{C-1,H-1} = 152.5$ Hz, C-1), 75.03 (C-5), 72.70 (C-3), 68.36 (C-4), 68.19 (C-2), 62.60 (C-6), 24.53 (SCH₂CH₃), 14.98 (SCH₂CH₃); ESI-HRMS calcd for C₃₆H₃₆NO₉S [M+NH₄]⁺: 658.2111; found: 658.2111; Anal. Calcd for C₃₆H₃₂O₉S: C, 67.49; H, 5.03. Found: C, 67.73; H, 5.02.

3.3.7. 1,2,3,6-Tetra-O-benzoyl-4-O-(2,3,4,6-tetra-O-benzoyl-β-Dgalactopyranosyl)-β-D-glucopyranose (Octa-O-benzoyl-βlactose, 15)

Reaction conditions: 40 h, 50 °C; flash chromatography, $30:1 \rightarrow$ 20:1 toluene-EtOAc; yield, starting from lactose monohydrate $(1.44 \text{ g}, 4.0 \text{ mmol}), 3.85 \text{ g} (82\%); R_{f} = 0.5 (10:1 \text{ toluene}-\text{EtOAc});$ $[\alpha]_{\rm D}$ +45.7 (*c* 1.0, CHCl₃); lit.³¹ Mp 140–142 °C (acetone–MeOH); $[\alpha]_{\rm D}$ +38.1 (c 0.9, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 6.15 (d, 1H, $J_{1,2}$ = 8.1 Hz, H-1'), 5.96 (t, 1H, J = 9.3 Hz, H-3), 5.80 (dd, 1H, $J_{2,3} = 9.6$ Hz, H-2), 5.78–5.72 (m, 2H, H-2',4'), 5.39 (dd, 1H, $J_{2',3'}$ = 10.4, $J_{3',4'}$ = 3.4 Hz, H-3'), 4.90 (d, 1H, $J_{1',2'}$ = 7.9 Hz, H-1'), 4.60 (dd, 1H, $J_{6a,6b} = 12.4$, $J_{5,6a} = 1.9$ Hz, H-6a), 4.54 (dd, 1H, $J_{5.6b}$ = 3.9 Hz, H-6b), 4.41 (t, 1H, J = 9.4 Hz, H-4), 4.08 (ddd, 1H, $J_{4,5} = 9.9$ Hz, H-5), 3.90 (t, 1H, J = 6.8 Hz, H-5'), 3.78 (dd, 1H, $J_{6a',6b'} = 11.3$, $J_{5',6a'} = 6.4$ Hz, H-6a'), 3.71 (dd, 1H, $J_{5',6b'} = 7.1$ Hz, H-6b'); 13 C NMR (150 MHz, CDCl₃) δ 101.00 (C-1'), 92.53 (C-1), 75.48 (C-4), 73.75 (C-5), 72.76 (C-3), 71.69 (C-3'), 71.32 (C-5'), 70.61 (C-2), 69.73, 67.39 (C-2',4'), 62.04 (C-6), 60.92 (C-6'); ESI-HRMS calcd for C₆₈H₅₈NO₁₉ [M+NH₄]⁺: 1192.3603; found: 1192.3623; Anal. Calcd for C₆₈H₅₄O₁₉: C, 69.50; H, 4.63. Found: C, 69.53; H, 4.61.

3.3.8. Ethyl 2,3,6-tri-O-benzoyl-4-O-(2,3,4,6-tetra-O-benzoyl- β p-galactopyranosyl)-1-thio- α - (16) and β -p-glucopyranoside (17) (Hepta-O-benzoyl-1-thio- α - and β -lactoside)

Reaction conditions: EtSH (0.15 mL, 2.0 mmol), $BF_3 \cdot Et_2O$ (0.23 mL, 1.8 mmol), 2 h, 50 °C; chromatography, 50:1 \rightarrow 25:1 toluene–EtOAc; yield, starting from **15** (1.175 g, 1.0 mmol), 1.024 g (76%) of mixture of **16** and **17**. Compound **16** crystallized from DCM–MeOH.

Compound **16**: *R*_f = 0.5 (19:1 toluene–EtOAc); Mp 221–222.5 °C (CH₂Cl₂–MeOH); $[\alpha]_D$ +91.3 (*c* 1.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 5.95 (t, 1H, J = 9.6 Hz, H-3), 5.82 (d, 1H, J_{1,2} = 5.9 Hz, H-1), 5.75 (br d, partially overlapped, H-4'), 5.73 (dd, partially overlapped, $J_{2',3'}$ = 10.2, $J_{1',2'}$ = 8.0 Hz), 5.43 (dd, 1H, $J_{2,3}$ = 10.2 Hz, H-2), 5.38 (dd, 1H, $J_{3',4'}$ = 3.4 Hz, H-3'), 4.92 (d, 1H, H-1'), 4.61– 4.52 (m, 3H, H-5, H-6a', H-6b'), 4.21 (t, 1H, J = 9.5 Hz, H-4), 3.91 (t, 1H, J = 6.7 Hz, H-5'), 3.85 (d, 1H, $J_{6a',6b'} = 11.3$, $J_{5',6a'} = 6.4$ Hz, H-6a'), 3.76 (dd, 1H, J_{5',6b'} = 7.1 Hz, H-6b'), 2.61–2.45 (m, 2H, SCH₂CH₃), 1.18 (t, J = 7.4 Hz, 3H, SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃) δ 101.13 $(J_{C-1',H-1'} = 160.1 \text{ Hz}, C-1')$, 81.91 $(J_{C-1,H-1} = 168.2 \text{ Hz}, C-1)$, 76.35 (C-4), 71.95 (C-3'), 71.49 (C-2), 71.30 (C-5'), 70.87 (C-3), 69.88 (C-2'), 68.78 (C-5), 67.45 (C-4'), 62.53 (C-6), 61.05 (C-6'), 24.23 (SCH₂CH₃), 14.55 (SCH₂CH₃); ESI-HRMS calcd for C₆₃H₅₈ NO17S [M+NH4]+: 1132.3425; found: 1132.3457; Anal. Calcd for C₆₃H₅₄O₁₇S: C, 67.85; H, 4.88. Found: C, 67.94; H, 4.90.

The mother liquor contained predominantly ethyl 2,3,6-tri-O-benzoyl-4-O-(2,3,4,6-tetra-O-benzoyl- β -D-galactopyranosyl)-1-t hio- β -D-glucopyranoside **17**, and a solvent suitable for separation by chromatography could not be found; $R_f = 0.5$ (19:1 toluene–EtOAc); amorphous solid; lit.³⁰ amorphous solid; ¹H NMR (600 MHz, CDCl₃) δ 5.82 (t, 1H, *J* = 9.3 Hz, H-3), 5.73 (m, partially overlapped, H-4'), 5.72 (dd, partially overlapped, $J_{2',3'}$ = 10.3, $J_{1'2'}$ = 7.9 Hz, H-2'), 5.50 (t, 1H, *J* = 9.7 Hz, H-2), 5.37 (dd, 1H, $J_{3',4'}$ = 3.4 Hz, H-3'), 4.87 (d, 1H, H-1'), 4.73 (d, 1H, $J_{1,2}$ = 10.0 Hz, H-1), 4.60 (dd, 1H, $J_{6a,6b}$ = 12.2, $J_{5,6a}$ = 1.8 Hz, H-6a), 4.49 (dd, 1H, $J_{5,6b}$ = 4.7 Hz, H-6b), 4.24 (t, 1H, *J* = 9.5 Hz, H-4), 3.90 (t, 1H, *J* = 6.8 Hz, H-5'), 3.86 (ddd, 1H, $J_{4,5}$ = 10.0 Hz, H-5), 3.72 (m, 2H, H-6a', H-6b'), 2.75–2.61 (m, 2H, SCH₂CH₃), 1.20 (t, *J* = 7.5 Hz, 3H, SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃) δ 100.92 ($J_{C-1',H-1'}$ = 161.4 Hz, C-1'), 83.74 ($J_{C-1,H-1}$ = 153.5 Hz, C-1), 77.00 (C-5), 75.92 (C-4), 73.97 (C-3), 71.75 (C-3'), 71.31 (C-5'), 70.48 (C-2), 69.82 (C-2'), 67.44 (C-4'), 62.65 (C-6), 60.98 (C-6'), 24.43 (SCH₂CH₃), 14.85 (SCH₂CH₃); ESI-HRMS calcd for C₆₃H₅₈NO₁₇S [M+NH₄]⁺: 1132.3425; found: 1132.3441; Anal. Calcd for C₆₃H₅₄ O₁₇S: C, 67.85; H, 4.88. Found: C, 67.82; H, 4.80.

3.3.9. 1,2,3,6-tetra-O-benzoyl-4-O-(2,3,4,6-tetra-O-benzoyl-α-D-glucopyranosyl)-β-D-glucopyranose (Octa-O-benzoyl-β-maltose, 19)

Reaction conditions: 40 h, 50 °C; flash chromatography, $50:1 \rightarrow 25:1$ toluene-EtOAc; yield, starting from 18 (1.71 g, 5.0 mmol), 4.32 g (74%); $R_f = 0.5$ (15:1 toluene-EtOAc); Mp 193-195 °C (MeOH); [α]_D +62.0 (*c* 1.0, CHCl₃); lit.³² Mp 190–192 °C (acetone–MeOH); $[\alpha]_{D}$ +68.2 (*c* 1.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 6.27 (d, 1H, $I_{1,2}$ = 7.4 Hz, H-1), 6.11 (t, 1H, I = 10.0 Hz, H-3'), 5.85 (t, 1H, J = 8.6 Hz, H-3), 5.79 (d, 1H, $J_{1'2'} = 3.9$ Hz, H-1'), 5.68 (t, 1H, J = 9.8 Hz, H-4'), 5.64 (dd, 1H, $J_{2,3} = 8.7$ Hz, H-2), 5.26 (dd, 1H, $J_{2',3'}$ = 10.5 Hz, H-2'), 4.90 (dd, 1H, $J_{6a,6b}$ = 12.3, J_{5,6a} = 2.5 Hz, H-6a), 4.78 (dd, 1H, J_{5,6b} = 3.8 Hz, H-6b), 4.67 (t, 1H, J = 9.0 Hz, H-4), 4.44 (dt, 1H, $J_{4',5'} = 10.1$ Hz, H-5'), 4.41 – 4.35 (m, 2H, H-6a', H-5), 4.21 (dd, 1H, J_{6a',6b'} = 12.3, J_{5',6b'} = 3.5 Hz, H-6b'); ¹³C NMR (150 MHz, CDCl₃) δ 96.37 (C-1'), 92.13 (C-1), 74.81 (C-3), 73.47 (C-5), 72.60 (C-4), 71.09 (C-2), 70.91 (C-2'), 69.80 (C-3'), 69.14 (C-5'), 68.94 (C-4'), 63.09 (C-6), 62.34 (C-6'); ESI-HRMS calcd for C₆₈H₅₈NO₁₉ [M+NH₄]⁺: 1192.3603; found: 1192.3596; Anal. Calcd for C₆₈H₅₄O₁₉: C, 69.50; H, 4.63. Found: C, 69.39; H, 4.80.

3.3.10. Ethyl 2,3,6-tri-O-benzoyl-4-O-(2,3,4,6-tetra-O-benzoyl- α p-glucopyranosyl)-1-thio- α - (20) and β -p-glucopyranoside (21) (Ethyl hepta-O-benzoyl-1-thio- α - and β -maltoside)

Reaction conditions: EtSH (0.34 mL, 4.5 mmol), $BF_3 \cdot Et_2O$ (0.42 mL, 3.3 mmol), 3 h, 50 °C; chromatography, 50:1 toluene–EtOAc; yield, starting from **19** (3.525 g, 3.0 mmol), 3.02 g (90%) of mixture of **20** and **21**. Only small amounts of pure **20** and **21** could be isolated by flash chromatography.

Compound 20: Amorphous solid after freeze-drying of a solution in benzene; $R_{\rm f} = 0.5$ (19:1 toluene–EtOAc); $[\alpha]_{\rm D}$ +112.2 (*c* 1.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 6.11 (dd, 1H, $J_{2',3'}$ = 10.5, *J*_{3',4'} = 9.6 Hz, H-3'), 5.96 (dd, 1H, *J*_{2,3} = 10.0, *J*_{3,4}= 8.9 Hz, H-3), 5.82 (d, 1H, $J_{1,2}$ = 5.6 Hz, H-1), 5.77 (d, 1H, $J_{1',2'}$ = 3.9 Hz, H-1'), 5.68 (t, 1H, J = 9.8 Hz, H-4'), 5.29 (dd, partially overlapped, H-2'), 5.28 (dd, partially overlapped, H-2), 4.87 (dd, 1H, $J_{6a,6b} = 11.9$, J_{5.6a} = 2.1 Hz, H-6a), 4.83–4.80 (m, 1H, H-5), 4.78 (dd, 1H, $J_{5,6b}$ = 4.5 Hz, H-6b), 4.50 (dt, 1H, $J_{4',5'}$ = 10.1, $J_{5',6a',b'}$ = 3.4 Hz, H-5'), 4.45 (dd, 1H, $J_{6a',6b'}$ = 12.3, $J_{5',6a'}$ = 3.1 Hz, H-6a'), 4.42 (t, J = 9.3 Hz, H-4), 4.31 (dd, 1H, $J_{5',6b'} = 3.7$ Hz, H-6b'), 2.65–2.55 (m, 2H, SCH₂CH₃), 1.25 (t, 3H, J = 7.4 Hz, SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃) δ 96.72 ($J_{C-1',H-1'}$ = 176.2 Hz, C-1'), 81.78 ($J_{C-1,H-1}$ = 168.3 Hz, C-1), 73.90 (C-4), 72.74 (C-3), 72.03 (C-2), 70.83 (C-2'), 69.84 (C-3'), 69.17 (C-5'), 69.08 (C-4'), 68.62 (C-5), 63.54 (C-6), 62.44 (C-6'), 24.33 (SCH₂CH₃), 14.69 (SCH₂CH₃); ESI-HRMS calcd for C₆₃H₅₈NO₁₇S [M+NH₄]⁺: 1132.3425; found: 1132.3400; Anal. Calcd for C₆₃H₅₄O₁₇S: C, 67.85; H, 4.88. Found: C, 67.87; H, 4.80.

Compound **21**: $R_{\rm f}$ = 0.5 (19:1 toluene–EtOAc); Mp 146–148 °C (MeOH); [α]_D +69.6 (*c* 1.0, CHCl₃); lit.³³, mode of preparation and characterization data were not disclosed; ¹H NMR (600 MHz, CDCl₃) δ 6.08 (dd, 1H, $J_{2',3'}$ = 10.5, $J_{3',4'}$ = 9.6 Hz, H-3'), 5.80 (t, 1H,

J = 9.2 Hz, H-3), 5.74 (d, 1H, $J_{1',2'}$ = 3.9 Hz, H-1'), 5.65 (t, 1H, *J* = 9.8 Hz, H-4'), 5.36 (t, 1H, *J* = 9.6 Hz, H-2), 5.24 (dd, 1H, H-2'), 4.93 (dd, 1H, $J_{6a,6b}$ = 12.1, $J_{5,6a}$ = 2.4 Hz, H-6a), 4.81 (d, 1H, $J_{1,2}$ = 9.8 Hz, H-1), 4.74 (dd, 1H, $J_{5,6b}$ = 4.4 Hz, H-6b), 4.49 (t, 1H, *J* = 9.3 Hz, H-4), 4.45 (dt, partially overlapped, $J_{4',5'}$ = 10.1, $J_{5',6a',b'}$ = 3.5 Hz, H-5'), 4.40 (dd, 1H, $J_{6a',6b'}$ = 12.2 Hz, $J_{5',6a'}$ = 3.1 Hz, H-6a'), 4.26 (dd, 1H, $J_{5',6b'}$ = 3.8 Hz, H-6b'), 4.10 (ddd, 1H, $J_{4,5}$ = 9.6 Hz, H-5), 2.86–2.58 (m, 2H, SCH₂CH₃), 1.22 (t, 3H, *J* = 7.5 Hz, SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃) δ 96.3 ($J_{C-1',H-1'}$ = 175.7 Hz, C-1'), 83.5 ($J_{C-1,H-1}$ = 154.7 Hz, C-1), 76.76 (C-5), 76.11 (C-3), 73.03 (C-4), 70.98 (C-2), 70.87 (C-2'), 69.81 (C-3'), 69.12 (C-5'), 69.04 (C-4'), 63.62 (C-6), 62.48 (C-6'), 24.3 (SCH₂CH₃), 14.9 (SCH₂CH₃); ESI-HRMS calcd for C₆₃H₅₈NO₁₇S [M+NH₄]⁺: 1132.3425; found: 1132.3401; Anal. Calcd for C₆₃H₅₄O₁₇S: C, 67.85; H, 4.88. Found: C, 67.93; H, 4.76.

3.3.11. 1,2,3,4-tetra-O-benzoyl-6-O-(2,3,4,6-tetra-O-benzoyl- α -D-galactopyranosyl)- β -D-glucopyranose (Octa-O-benzoyl- β melibiose, 23)

Reaction conditions: 16 h, 50 °C; flash chromatography, 50:1→20:1 toluene-EtOAc; yield, starting from melibiose monohydrate (1.08 g, 3.0 mmol), 2.53 g (75%); $R_f = 0.5$ (15:1 toluene– EtOAc); Mp 165–168 °C (MeOH); [α]_D +141.3 (*c* 1.0, CHCl₃); lit.³⁴ (anomeric configuration not specified; in the absence of NMR data, judging by physical constants, the compound reported previously is likely the α anomer): Mp 111–112 °C (EtOH), $[\alpha]_D$ +174 (*c* 4.55, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 6.15 (d, 1H, $J_{1,2}$ = 8.2 Hz, H-1), 6.12 (dd, 1H, $J_{2',3'}$ = 10.7, $J_{3',4'}$ = 3.5 Hz, H-3'), 6.02–5.99 (m, partially overlapped, H-4'), 5.98 (t, partially overlapped, J 9.7 Hz, H-3), 5.71 (dd, 1H, $J_{1',2'}$ = 3.7 Hz, H-2'), 5.70 (t, 1 H, J = 9.8 Hz, H-4), 5.61 (dd, 1H, $J_{2,3}$ = 9.8 Hz, H-2), 5.43 (d, 1H, H-1'), 4.65 (br t, 1H, H-5'), 4.36 (dd, 1H, $J_{6a',6b'}$ = 11.5, $J_{5',6a'}$ = 7.7 Hz, H-6'a'), 4.23 (ddd, 1H, $J_{4,5} = 10.1, J_{5,6a} = 5.1, J_{5,6b} = 1.8$ Hz, H-5), 4.09 (dd, 1H, $J_{5',6b'} = 4.9$ Hz, H-6b'), 4.02 (dd, 1H, J_{6a,6b} = 11.4 Hz, H-6a), 3.83 (dd, 1H, H-6b); ¹³C NMR (150 MHz, CDCl₃) δ 96.57 (C-1'), 92.92 (C-1), 74.20 (C-5), 72.87 (C-3), 70.85 (C-2), 69.35 (C-4'), 68.89 (C-2'), 68.59 (C-4), 68.55 (C-3'), 67.14 (C-5'), 65.68 (C-6), 62.86 (C-6'); ESI-HRMS calcd for C₆₈H₅₈NO₁₉ [M+NH₄]⁺: 1192.3603; found: 1192.3628; Anal. Calcd for C₆₈H₅₄O₁₉: C, 69.50; H, 4.63. Found: C, 69.23; H, 4.89.

3.3.12. Ethyl 2,3,4-tri-O-benzoyl-6-O-(2,3,4,6-tetra-O-benzoyl- α -D-galactopyranosyl)-1-thio- α - (24) and β -D-glucopyranoside (25) (Ethyl hepta-O-benzoyl-1-thio- α - and β -melibioside)

Reaction conditions: EtSH (0.19 mL, 2.5 mmol), $BF_3 \cdot Et_2O$ (0.25 mL, 2.0 mmol), 16 h, 60 °C (optimum reaction time determined by NMR spectroscopy); chromatography, 50:1 \rightarrow 30:1 toluene–EtOAc; yield, starting from **23** (1.175 g, 1.0 mmol), 0.320 g (29%) of **24**, 0.159 g (14%) of **25**. Unresolved mixture of **24** and **25** (0.495 g, 44%) was also obtained.

Compound 24: Amorphous solid after freeze-drying of a solution in benzene; $R_f = 0.6$ (12:1 toluene–EtOAc); $[\alpha]_D$ +165.0 (*c* 1.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 6.05 (br d, 1H, $J_{3',4'}$ = 3.4 Hz, H-4′), 6.01 (dd, *J*_{2′,3′} = 10.5, Hz, H-3′), 5.98 (t, 1H, *J* = 9.9 Hz, H-3), 5.81 (d, 1H, $J_{1,2}$ = 5.9 Hz, H-1), 5.71 (dd, 1H, $J_{1',2'}$ = 3.7 Hz, H-2'), 5.48 (d, 1H, H-1'), 5.35 (t, 1H, J = 9.9 Hz, H-4), 5.05 (dd, 1H, $J_{2,3} = 10.1$ Hz, H-2), 4.72 (br dd, 1H, H-5'), 4.69 (ddd, 1H, $J_{4,5} = 10.1, J_{5,6a} = 6.5, J_{5,6b} = 1.9$ Hz, H-5), 4.51 (dd, 1H, $J_{6a',6b'} = 11.5$, J_{5',6a'} = 4.8 Hz, H-6a'), 4.43 (dd, 1H, J_{5',6b'} = 7.7 Hz, H-6b'), 4.04 (dd, 1H, $J_{6a,6b}$ = 11.0 Hz, H-6a), 3.65 (dd, 1H, H-6b), 2.76–2.58 (m, 2H, SCH₂CH₃), 1.35 (t, 3H, J = 7.4 Hz, SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃) δ 96.25 ($J_{C-1',H-1'}$ = 173.7 Hz, C-1'), 81.10 ($J_{C-1,H-1}$ = 170.0 Hz, C-1), 71.57 (C-2), 70.84 (C-3), 69.30 (C-4'), 69.17 (C-4), 69.08 (C-2'), 68.71 (C-5), 68.42 (C-3'), 67.18 (C-5'), 65.98 (C-6), 62.84 (C-6'), 23.93 (SCH₂CH₃), 14.45 (SCH₂CH₃); ESI-HRMS calcd for C₆₃H₅₈NO₁₇S [M+NH₄]⁺: 1132.3425; found: 1132.3422; Anal. Calcd for C₆₃H₅₄O₁₇S: C, 67.85; H, 4.88. Found: C, 67.82; H, 5.02.

Compound **25**: *R*_f = 0.5 (19:1 toluene–EtOAc); Mp 157–159 °C; $[\alpha]_{D}$ +128.7 (c 1.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 6.06 (dd, 1H, $J_{2',3'} = 10.6$, $J_{3',4'} = 3.4$ Hz, H-3'), 6.03 (dd, 1H, $J_{4,5} = 1.2$ Hz, H-4'), 5.85 (t, 1H, I = 9.6 Hz, H-3), 5.73 (dd, 1H, $I_{1',2'} = 3.7$ Hz, H-2'), 5.59 (t, 1H, J=9.6 Hz, H-4), 5.49 (d, 1H, H-1'), 5.33 (t, 1H, J = 9.7 Hz, H-2), 4.74 (d, 1H, J_{1,2} = 10.0 Hz, H-1), 4.66 (br t, 1H, H-5'), 4.47 (dd, 1H, $J_{6a',6b'}$ = 11.4, $J_{5',6a'}$ = 7.2 Hz, H-6a'), 4.31 (dd, 1H, $_{5',6b'}$ = 5.6 Hz, H-6b'), 4.06-3.98 (m, 2H, H-5, H-6a), 3.80-3.76 (m, 1H, H-6b), 2.64 (q, 2H, J = 7.4 Hz, SCH₂CH₃), 1.16 (t, 3H, J = 7.4 Hz, SCH₂CH₃); ¹³C NMR (150 MHz, CDCl₃) δ 96.57 ($J_{C-1',H-1'}$ = 173.7 Hz, C-1'), 83.64 (J_{C-1.H-1} = 153.0 Hz, C-1), 77.42 (C-5), 74.23 (C-3), 70.41 (C-2), 69.19 (C-4'), 69.02 (C-2'), 68.98 (C-4), 68.40 (C-3'), 67.15 (C-5'), 66.29 (C-6), 62.60 (C-6'), 23.84 (SCH2CH3), 14.69 (SCH_2CH_3) ; ESI-HRMS calcd for $C_{63}H_{58}NO_{17}S$ $[M+NH_4]^+$: 1132.3425; found: 1132.3397; Anal. Calcd for C₆₃H₅₄O₁₇S: C, 67.85: H. 4.88. Found: C. 67.96: H. 5.15.

Acknowledgement

This research was supported by the Intramural Research Program of the NIH, NIDDK.

References

- Norberg, T. Glycosylation Properties and Reactivity of Thioglycosides, Sulfoxides and Other S-glycosides: Current Scope and Future Prospects. In Modern Methods in Carbohydrate Synthesis; Khan, S. H., O'Neill, R. A., Eds.; Harwood Academic Publishers: Amsterdam, 1996; pp 82–106.
- 2. Garegg, P. Adv. Carbohydr. Chem. Biochem. 1997, 52, 179-205.
- 3. Ferrier, R. J.; Furneaux, R. H. Methods Carbohydr. Chem. 1980, 8, 251–253.
- 4. Takeo, K.; Maki, K.; Wada, Y.; Kitamura, S. Carbohydr. Res. 1993, 245, 81–96.
- 5. Kramer, S.; Nolting, B.; Ott, A. J.; Vogel, C. J. Carbohydr. Chem. 2000, 19, 891–921.
- Koike, K.; Sugimoto, M.; Sato, S.; Ito, Y.; Nakahara, Y.; Ogawa, T. Carbohydr. Res. 1987, 163, 189–208.

- 7. Weygand, F.; Ziemann, H. Ann. 1962, 657, 179-198.
- Veeneman, G. H.; van Leeuven, S. H.; van Boom, J. H. Tetrahedron Lett. 1990, 31, 1331–1334.
- 9. Dasgupta, F.; Garegg, P. J. Acta Chem. Scand. 1989, 43, 471-475.
- 10. Balavoine, G.; Berteina, S.; Gref, A.; Fischer, J.-c.; Lubineau, A. J. Carbohydr. Chem. **1995**, 14, 1217–1236.
- 11. Zhang, Z.; Ollmann, I. R.; Ye, X.-S.; Wischnat, R.; Baasov, T.; Wong, C.-H. J. Am. Chem. Soc. **1999**, 121, 734–753.
- 12. Biessen, E. A. L.; Beuting, D. M.; Roelen, H. C. P. F.; van de Marel, G. A.; van Boom, J. H.; van Berkelf, T. J. C. *J. Med. Chem.* **1995**, 38.
- 13. Pakulski, Z.; Pierozynski, D.; Zamojski, A. *Tetrahedron* **1994**, *50*, 2975–2992.
- 14. Valerio, S.; Iadonisi, A.; Adinolfi, M.; Ravida, A. J. Org. Chem. 2007, 72, 6097-6106.
- 15. Cao, S.; Hernandez-Mateo, F.; Roy, R. J. Carbohydr. Chem. 1998, 17, 609-631.
- 16. Ogawa, T.; Beppu, K.; Nakabayashi, S. Carbohydr. Res. 1981, 93, C6-C9.
- 17. Paulsen, H.; Paal, M. Carbohydr. Res. 1984, 135, 53-69.
- 18. Kováč, P. Carbohydr. Res. 1986, 153, 237-251.
- Gallo-Rodriguez, C.; Varela, O.; de Lederkremer, R. M. Carbohydr. Res. 1998, 305, 163–170.
- 20. Ziegler, T.; Kováč, P.; Glaudemans, C. P. J. Liebigs Ann. Chem. 1990, 613-615.
- 21. Garegg, P. J.; Norberg, T. Acta Chem. Scand. 1979, B33, 116-118.
- Luo, S.-Y.; Kulkarni, S. S.; Chou, C.-H.; Liao, W.-M.; Hung, S.-C. J. Org. Chem. 2006, 71, 1226–1229.
- 23. Wolfrom, M. L.; Christman, C. C. J. Am. Chem. Soc. 1936, 58, 39-43.
- 24. Deferrari, J.; Deulofeu, V. J. Org. Chem. 1952, 17, 1097-1101.
- Kováč, P.; Glaudemans, C. P. J.; Guo, W.; Wong, T. C. Carbohydr. Res. 1985, 140, 299–311.
- 26. Xu, C.; Liu, H.; Li, X. Carbohydr. Res. 2011, 346, 1149–1153.
- Ness, R. K.; Fletcher, H. G.; Hudson, C. S. J. Am. Chem. Soc. 1950, 72, 2200–2205.
 Fischer, E.; Oetker, R. Chem. Ber. 1914, 46, 4029–4040.
- Sarbajna, S.; Roy, N. Indian J. Chem., Sect. B: Org. Chem. Incl. Med. Chem. 1998, 37B, 252–256.
- Dondoni, A.; Marra, A.; Scherrmann, M.-C.; Casnati, A.; Sansone, F.; Ungaro, R. Chem. Eur. J. 1997, 3, 1774–1782.
- 31. Vazquez, I. M.; Thiel, I. M. E.; Deferrari, J. O. Carbohydr. Res. **1973**, *26*, 351–356. 32. Thiel, I. M. E.; Deferrari, J. O.; Cadenas, R. A. Justus Liebigs Ann. Chem. **1969**, 723,
- 192-197. 192-197. V. M. S., Delettari, J. C., Cadenas, K. A. Jastas Liebigs Anni. Chem. 1909, 725
- Zhang, Q.; Ma, X.; Ward, A.; Hong, W.-X.; Jaakola, V.-P.; Stevens, R. C.; Finn, M. G.; Chang, G. Angew. Chem., Int. Ed. 2007, 46, 7023–7025.
- 34. Lerner, L. J. Org. Chem. 1967, 32, 3663-3665.