STUDIES ON CYCLOADDITION REACTIONS OF 1,3-DIPHENYL-2-AZAALLYL LITHIUM AND ETHYL(BENZYLIDENEAMINO)ACETATE ANION WITH α-OXOKETENE DITHIOACETALS

Maliakel P. Balu, Hiriyakkanavar Ila^{*} and Hiriyakkanavar Junjappa^{*} Department of Chemistry, North-Eastern Hill University Shillong 793 003, Meghalava, India

(Received in UK 14 June 1990)

Abstract: The *a*-oxoketene dithioacetals undergo anionic [1,3] cycloaddition with 1,3-diphenyl-2-azaallyllithium (<u>1</u>) to give either pyrroles or spiropyrrolines with two exceptions. Lithium bromide/triethylamine induced cycloaddition of ethyl (benzylideneamino)acetate (<u>17</u>) with acyclic *a*-oxoketene dithioacetals afforded either pyrrolidine, pyrrole or the corresponding 3-benzylideneamino-pyran-2-one derivatives depending on the reaction conditions and the structural variations in the *a*-oxoketene dithioacetals. Nitroketene dithioacetal also underwent anionic [1,3] cycloaddition with either <u>1</u> or <u>17</u> under described mechanisms for the formation of various products are suggested.

As a part of our programmed studies on aromatic¹ and heteroaromatic annelation through α -oxoketene dithioacetals, we had recently reported the reactions of 1-azaallyl anions such as 2-lithioacetonitrile^{2a} and 2picolyllithium^{2b} with these intermediates, initially to afford the corresponding carbinol acetals which on Lewis acid assisted cyclization yielded the respective pyridine and quinolizinium fluoroborate derivatives in high yields. Apparently, the 1-azaallyl anions have exhibited their propensity of 1,2-addition mode to the α -oxoketene dithioacetals, known for their behaviour as reagents for directed aldol-condensation³. In contrast, the pioneering work of Kauffmann⁴ has shown that 2-azaallyl anions like 1,3-diphenyl-2-azaallyllithium (1) undergo anionic [1,3]-cycloadditions with unsaturated compounds containing CC,CN,NN and CS double bonds as well as those containing CC and CN triple bonds, although their reactions with carbonyl compounds were shown to yield only the open-chain carbinol adducts. Similarly, conjugated carbonyl compounds were reported⁴ to be unsuitable for anionic [1,3]-cycloaddition owing to the pronounced nucleophilic character of the anionic reagent 1. However, subsequently it has been demonstrated that the α -cyano- β -phenyl cinnamate^{5a} and the phenylpropiolic acid amide^{5b} undergo cycloaddition on CC double and triple bonds with $\underline{1}$ to afford pyrrolidine and pyrrole derivatives respectively. These studies clearly demonstrate that the 2-azaallyl anions behave more like lithio 1,3-dipolar species rather than mere nucleophilic reagents even towards carbonyl activated double bonds 6 . It has been demonstrated 4 that the heteroallyl anions best suited for anionic [1,3]-cycloadditions are the 2-azaallyl anions where a relatively electronegative nitrogen is located at the center of the heteroallyl systems. The negative charge in these systems is largely located at the terminal carbon atoms and during the cycloaddition reactions the charge migrates to the more electronegative

central nitrogen atom in the five membered cycloadducts thus making them efficient 1,3-dipolar systems⁴. We therefore became interested in the reaction of these anions with α -oxoketene dithioacetals, whether they would occur via [3+2] cycloaddition to β,β -bis(methylthio) double bond or by intial aldol type nucleophilic addition to the carbonyl group. The two azaallyl anions investigated have indeed behaved as 1,3-anionic dipolar systems with a few exceptions, undergoing [3+2] cycloadditions with mercapto double bond to yield the respective pyrrolidine or the corresponding pyrrole derivatives in good yields. We have investigated the reaction of 1,3-diphenyl-2-azaallyllithium (<u>1</u>) and ethyl (benzylidene amino)acetate anion <u>2</u> with α -oxoketene dithioacetals and our results are reported herein.



Cycoladdition Reactions of 1,3-Diphenyl-2-Azaallyllithium with α -Oxoketene Dithioacetals

Kauffmann and coworkers⁴ have shown that the 2-azaallyl anion <u>1</u> generated by deprotonation of N-benzylidenebenzylamine 3 with lithium diisopropyl amide undergoes [1,3] anionic cycloaddition with conjugated and activated double bonds in a highly stereoselective and stereospecific manner to afford the corresponding pyrrolidine derivatives. The reaction of 1 with a-oxoketene dithioacetals under identical reaction conditions was therefore examined. In a typical experiment, when the dithioacetal 4a was reacted with $\underline{1}$ at -78°C, work-up of the reaction mixture yielded a product (79%) which was characterized as 3-benzoy1-2,5-dipheny1pyrrole (5a) on the basis of spectral and analytical data (Scheme 1). Apparently the reaction involves an intial [3+2] cycloaddition of <u>4a</u> and <u>1</u> to afford the corresponding pyrrolidine lithium 6a which on successive prototropic shift followed by methylthic group elimination affords 5a. The cycloaddition of other acyclic oxoketene dithioacetals 4b-g and the carbomethoxyacetal 4h also gave the corresponding 3-acyl (5b-g) and 3-carbomethoxy (5h) pyrroles in 71-78% overall yields (Scheme 1). Similarly when the cyclic oxoketene dithioacetal <u>7a</u> derived from α -tetralone was treated with <u>1</u>, the corresponding spiropyrroline 8a was formed (75%) by elimination of methylthiolithium from an unstable cycloadduct 9. The structure of 8a was confirmed with the help of spectral and analytical data. The observed long range coupling (2Hz) between H_A and H_B protons confirms the *cis* stereochemistry in accordance with the earlier observations⁷. The other cyclic oxoketene dithioacetals 7b and 7c derived from benzothiepinone and 5-pyrazolone also underwent [3+2] cycloaddition with $\underline{1}$ to yield the corresponding spiropyrrolines <u>8b</u> and <u>8c</u> in 68% and 71% yields respectively (Scheme 2).



Interestingly the cyclic oxoketene dithioacetal <u>10</u> derived from cyclohexanone did not react in the expected manner to yield the corresponding spiropyrroline cycloadduct and the product thus obtained was characterized as the N-substituted isoindole derivative <u>12a</u> (79%), on the basis of spectral and analytical data (Scheme 3). The structure of <u>12a</u> was further confirmed by its mercuric chloride catalyzed hydrolytic cleavage to the isoindole <u>13a</u> (Scheme 3). Similarly <u>11</u> from cycloheptanone reacted with <u>1</u> to give the corresponding 3.4 annelated pyrrole <u>12b</u> (76%), which on treatment with mercuric chloride afforded the corresponding dealkylated derivative <u>13b</u> in 68% yield (Scheme 3). The probable mechanism for the formation of <u>12a-b</u> from <u>10</u> and <u>11</u> is described in Scheme 3. Apparently <u>1</u> reacted in the 1,2-fashion followed by nitrogen lone pair assisted intramolecular cyclization to give zwitterionic intermediate <u>15</u>, which on 1,3-methylthio shift affords the corresponding stable pyrrole derivatives 12a-b (Scheme 3).



Lithium Bromide/Triethylamine Induced Cycloaddition of Ethyl (Benzylidene~ amino) acetate to α -Oxoketene Dithioacetals

The deprotonation studies of N-alkylidene-2-amino esters have been investigated by several groups⁸. Generally the deprotonation is achieved by sodium or potassium alkoxides or Triton B either in protic (MeOH/EtOH) or aprotic solvents. The resulting azaallyl anions are trapped by electron deficient olefins to yield the corresponding Michael adducts in most of the cases. However Grigg and co-workers have reported in some cases, competetive formation of both Michael adducts and stereoselective cycloadducts and proposed a mechanism involving sequential Michael adduct mediated cyclization⁹. The same group was the first to demonstrate the facile thermal tautomerism in imines of α -amino esters to N-protonated azomethine ylides which were shown to undergo cycloaddition with a variety of dipolarophiles¹⁰. Attempts to generate the N-metalated azomethine ylides by deprotonation of ethyl (benzylideneamino) acetate 17 with strong bases such as LDA, n-BuLi, NaH and EtMgBr and to trap them by conjugated carbonyl compounds resulted in complex reaction mixture. However, subsequently it was shown¹¹ that a combination of metal salt (Ag,Li,Zn,Ti) and triethylamine in polar aprotic solvent affects rapid regio and stereospecific cycloaddition of imines of α -amino esters to a range of dipolarophiles. The nature of the reactive species in these Lewis acid catalyzed cycloadditions is not clear, though an equilibrium involving metallo 1,3dipoles and N-protonated azomethine ylide has been suggested^{11c,11e}.

The addition mode of these anionic species from N-benzylideneaminoacetate with α -oxoketene dithioacetals was therefore examined under the above reaction conditions. When <u>4a</u> was refluxed with <u>17</u> in toluene, the envisaged cycloaddition of N-protonated azomethine ylide was not observed and the starting materials were recovered unchanged. Similarly the formation of N-lithiated azomethine ylide could not be observed by deprotonation of 17 by treatment with either LDA or n-BuLi in the presence of 4a and the reaction mixture resulted only in an intractable tar. However when 17 was treated with lithium bromide and triethylamine in the presence of 4a in THF at room temperature the reaction mixture after work-up gave the expected pyrrolidine cycloadduct <u>18a</u> in high yield as a single stereo and regioisomer (Scheme 4). The structure of <u>18a</u> was established by its analytical and spectral data. The stereochemistry at 2,4 and 5 carbon atoms was assigned on the basis of vicinal coupling constants measured from molecular models as well as through the known stereochemical mode of their cycloadditions^{8c}. However in an another experiment, when 17 and 4a were refluxed in the presence of lithium bromide and triethylamine in THF under a N₂ atmosphere a different product was isolated which was characterized as 3-(benzylideneamino)pyran-2-one 19a (81%) (Scheme 4). The intermediacy of <u>18a</u> in the formation <u>19a</u> was established by treating <u>18a</u> with LiBr/Et₃N in refluxing THF (16 hr), when the full conversion of 18a to 19a was observed. Similarly the α -(2-furoyl)ketene dithioacetal 4c reacted with 17 under similar reaction conditions to afford <u>18b</u> or <u>19b</u> in 69% and 61% yields respectively. However the cycloadditions of α -acetylketene dithioacetal 4g with 2 under the described temperature conditions gave only the corresponding pyran-2-one (19c) and no trace of the corresponding 4acetylpyrrolidine-2-carboxylate was observed (Scheme 4).



The possible mechanism for the formation of highly stereo and regioselective pyrrolidines <u>18</u> and the corresponding pyran-2-ones <u>19</u> is described in the Scheme 5 which is similar to that proposed by Tsuge and cowokers^{8c}. The pyrrolidine <u>18</u> can be formed either through path A involving a concerted cycloaddition of the syn N-lithiated azomethine ylide <u>2A</u> in endo fashion through a chelated intermediate <u>20</u>. Alternatively the lithium enolate <u>2B</u> may undergo tandem Michael imine addition as depicted in <u>21</u> and <u>23</u> (path B). The lithiated pyrrolidine <u>22</u> suffers rapid bond cleavage at higher temperature leading to N-lithiated betaine intermediate <u>24</u> that follows intramolecular enollactonization and loss of methylthio group to afford the thermodynamically more stable pyran-2-one. The formation of betaine <u>24</u> from <u>22</u> appears to proceed due to steric crowding in <u>22</u> as

well as by the anion stabilization offered by aryl and acyl group¹². The exclusive formation of only the pyran-2-one <u>19c</u> from α -acetylketene dithioacetal <u>4g</u> both at room temperature and under refluxing conditions may be rationalized in terms of higher stability of the betaine intermediate <u>24</u> due to the increased anion stabilizing ability of the acetyl group.



When the α -(2-thienoy1) (4d) and α -(3-pyridinoy1) (4e) ketene dithioacetals were reacted with 2 under identical reaction conditions, starting materials along with intractable tar were recovered. However, when the same reaction mixtures were refluxed for prolonged time (30 hr) in the absence of nitrogen, the products isolated were found to be pyrrole-2-carboxylates 25b (69%) and 25c (79%) respectively (Scheme 6). Similarly 4c and 2 under indentical prolonged reaction conditions, yielded the corresponding 3-(2furoy1)pyrrole-2-carboxylate 25a in 73% yield. Obviously the intially formed lithiated pyrrolidines 22 undergo a facile elimination of methylthiolithium followed by dehydrogenation of the resulting pyrrolines 26 to afford the corresponding pyrroles 25a-c. The structures of these pyrroles were fully established by their analytical and spectral data.



As an extension of these studies the cycloaddition of nitroketene S,Sacetal $\underline{27}$ with 2-azaallyl anion $\underline{1}$ and the enolate anion $\underline{2}$ was also examined. Thus $\underline{27}$ underwent cycloaddition with $\underline{3}$ in the presence of LDA at -78°C to afford the corresponding pyrrolidine $\underline{28}$ as single stereoisomer in 64% yield (Scheme 7). the corresponding 3-nitro-2,5-diphenyl pyrrole however could not be isolated even after prolonged heating of the reaction mixture. High stereoselectivity and the stability of the cycloadduct $\underline{28}$ indicates the participation of chelated *endo* intermediate <u>30</u> through syn conformation of azaallyl anion <u>1</u>. The cylcoaddition of <u>27</u> with <u>17</u> at room temperature in the presence of LiBr/Et₃N did not proceed in the expected manner and no clear-cut product could be isolated. However when the same reaction mixture was refluxed in the presence of a N₂ atmosphere, the corresponding 3-nitropyrrole-2-carboxylate <u>31</u> was isolated in 56% yield (Scheme 7). The formation of <u>31</u> could be explained through successive elimination of methylthio groups as described earlier.



In conclusion, the 2-azaallyl anions behave as 1,3-anionic dipolar species towards α -oxoketene dithioacetals through cycloaddition to the mercapto double bond rather than nucleophilic addition to the carbonyl group with a few exceptions. The formation of fully aromatized pyrroles from open-chain oxoketene dithioacetals and <u>1</u> (or <u>2</u>) clearly qualifies these intermediates as corresponding acylacetylene equivalents.

EXPERIMENTAL

Meling points were determined on a Thomas Hoover capillary melting point apparatus and are uncorrected. IR spectra were run as KBr discs on a Perkin-Elmer 297 spectrophotometer. ¹H NMR spectra were recorded on a Varian EM-390 (90 MHz) spectrometer in deuteriochloroform with tetramethylsilane as internal standard unless otherwise stated. Chemical shifts are expressed as δ (ppm) downfield from TMS. ¹³C NMR spectra were recorded on a Brucker WM-400 spectrometer. Mass spectra were obtained using a Jeol JMS D-300 spectrometer. Elemental analysis were carried out on Heraeus CHN-O-RAPID instrument. All α -oxoketene dithioacetals were prepared according to the general procedure reported earlier¹³. Ethyl (benzylideneamino)acetate was prepared as reported¹³.

General Procedure for the Generation and Reaction of 1,3-Diphenyl-2-Azaallyllithium with a-Oxoketene Dithioacetals:

To a stirred solution of lithium diisopropylamide [(11 mmol) freshly prepared from n-butyllithium (11 mmol) and diisopropylamine (1.10gm, 11

mmol) in dry THF (20 ml)], N-benzylidenebenzylamine (1.95gm, 10 mmol) in THF (5 ml) was added slowly at -78°C under nitrogen atmosphere. The lithiation was indicated by the appearance of reddish brown colour. The solution was stirred for 15 min at -78°C followed by addition of oxoketene dithioacetal (10 mmol) in THF (30 ml) and further stirring for 1 hr at -78°C. The temperature was raised to ambient (30 min) and stirring was further continued for 15 hr. The reaction mixture was poured into satd. NH₄Cl solution and the organic layer was separated, the aqueous layer extracted with ether (2x50 ml) and the combined organic layer was washed with water (100 ml), dried (Na₂SO₄) and evaporated to give viscous residue which were purified by column chromatography over silica gel using hexane-EtOAc as eluent (5:1) to give pyrroles 5a-h, spiropyrrolines 8a-c and fused pyrroles 12a, b which were further purified by crystallization from EtOAc/hexane.

Benzoyl-2,5-diphenylpyrrole (5a) colourless crystals; 79%; m.p. 168-169°C; IR ν_{max} 3200,1595,1572 cm⁻¹; ¹H NMR (CDCl₃): δ 6.75 (d, J=1.5Hz, 1H,H-4); 7.62-7.81 (m,13H,arom); 7.59-7.83 (m,2H,arom); 11.93 (brs,1H,NH,exchangeable with D₂O); (Found: C, 85.71,H,5.21;N,4.62.Calc. for C₂₃H₁₇NO: C,85.42; H, 5.30; N, 4.36%); m/z 323 (M⁺, 100%); 246 (72); 218 (7).

2,5-Diphenyl-3-(2-furoyl)pyrrole (5c) yellow crystals; 76%; m.p. 162-163°C; IR y_{max} 3240, 1602, 1578 cm⁻¹; ¹H NMR (CDCl₃) δ 6.31 (dd, J=3 and 1.5 Hz, 1H, H-4' furyl); 6.79-7.56 (m, 13H, arom); 9.60 (brs, 1H, NH, exchangeable with D₂O); (Found: C, 80.68; H, 4.58; N, 4.23; Calc. for $C_{23H_{15}NO_{2}}$: C, 80.49; H, 4.83; N, 4.47%); m/z 313 (M⁺, 100%); 296(17).

2.5-Dipheny1-3-(2-thienoy1)pyrrole (5d) yellow crystals; 78%; m.p. 145-146°C; $IR\mathcal{D}_{max}$ 3200,1600,1588 cm⁻¹; ¹H NMR (CDCl₃) δ 6.74-7.58 (m, 14H, arom); 9.51(brs,1H,NH,exchangeable with D₂O); (Found:C,76.69;H, 4.38; N, 4.18;Calc.for C₂₁H₁₅NOS:C,76.57;H,4.59;N,4.25%);m/z 329 (M⁺, 100%).

 $\begin{array}{l} 2,5-\text{Diphenyl-}3-(3-\text{pyridinoyl})\text{pyrrole(5e)} \quad \text{light yellow crystals;} & 81\%;\text{m.p.}266-267^\circ\text{C} \quad \text{IR} \ensuremath{\mathcal{Y}_{\text{max}}}3150,1645,1600,1582 \quad \text{cm}^{-1};^1\text{H} \quad \text{NMR}(\text{CDCl}_3) \\ \delta 6.73(d,J=1.5\text{Hz},1\text{H},\text{H}-4); \\ 6.94-7.99(m,12\text{H},\text{arom and pyridyl}); & 38(d,J=6\text{Hz},1\text{H},\text{H}-6'\text{pyridyl}); & 8.70(s,1\text{H},\text{H}-2'\text{pyridyl}); & 10.61(\text{brs},1\text{H},\text{NH},\text{exchangeable with D}_2\text{O}); \quad (\text{Found C},81.23;\text{H},4.82,\text{N},\\ 8.73.\text{Calc.for C}_{22}\text{H}_16\text{N}_2\text{O}:\text{C},81.46;\text{H},4.97;\text{N},8.64\%); \\ m/z & 324(\text{M}^+,100\%); & 247(22). \end{array}$

2.5-Diphenyl-3-(2-naphthoyl)pyrrole (5f) yellow crystals; 84%; m.p. 181°C, $IR\mathcal{V}_{max}$ 3255,1601,1570 cm⁻¹; ¹ NMR (CDCl₃) δ 6.81 (d,J=1.5Hz,1H, H-4); 7.02-7.98 (m,16H,arom); 8.31 (s,1H,H-1'naphthyl); 9.30 (brs,1H, NH, exchangeable with D₂O); (Found: C, 86.58; H, 5.24; N, 3.86. Calc. for C₂₇H₁₉NO: C, 86.84; H, 5.13; N, 3.75%); m/z 373 (M⁺,100%); 344(5).

3-Acetyl-2,5-diphenylpyrrole (5g) colourless crystals; 74%; m.p. 169°C; IR y_{max} 3240,1605,1455 cm⁻¹; ¹H NMR (CDCl₃) δ 2.25 (s, 3H, CH₃); 6.90 (d,

J=1.5Hz, 1H, H-4); 7.19-7.61 (m,10H,arom); 9.08 (brs,1H,NH,exchangeable with D_2O); (Found: C,82.54; H,5.82; N,5.23.Calc. for $C_{18}H_{15}NO$: C,82.73; H,5.79; N,5.36%); m/z 261 (M⁺,70%);246(100);217(16).

 $\begin{array}{l} {\rm Spiro[(3,4-dihydronaphthalene-1-oxo)-2(1H),3'-(2',5'-diphenyl-4'-methyl-thio-5'-pyrroline)]} & (8a) colourless crystals; 74%, m.p.163°C. IRJ_{max} 1670, 1605, 1595 cm^{-1}; ^1H NMR (CDCl_3): \delta 1.36-1.59 (m.2H,CH_2); 1.90 (s, 3H, SCH_3); 2.54-3.21 (m.2H,CH_2); 4.45(d,J=2Hz,1H,H_A); 6.22 (d, J=2Hz, 1H, H_B); 7.00-7.51 (m.11H,arom); 8.05-8.25(m.3H,arom); (Found:C.78.26,H,5.89;N,3.68. Calc.for C_{26}H_{23}ONS: C.78.55; H,5.83; N,3.52%); m/z 397 (M⁺,25%); 350(100). \end{array}$

 $\begin{array}{l} {\rm Spiro[(2,3-dihydro-8-methyl-5-oxo-[1]-benzothiepin)-4(5H),3'-(2',5'-diphenyl 4'-methylthio-5'-pyrroline)] (8b) colourless crystals; 68%; m.p. 173-174°C. IR <math>{\rm J}_{max}$ 1659,1616,1600 cm⁻¹; ¹H NMR (CDCl₃): 1.82 (s,3H,CH₃); 2.34 (s, 3H, SCH₃); 2.48-2.95 (m,4H,-CH₂-); 5.05 (d, J=2Hz,1H,H_A) 5.90 (d, J=2Hz, 1H, H_B); 7.05-7.88 (m,12H,arom); 8.09 (d,J=9 Hz, H-6); (Found: C,72.94; H, 5.74; N, 3.28. Calc.for C₂₇H₂₅ONS₂ C,73.10; H,5.68; N, 3.16%); m/z 443 (M⁺,14%); 397 (88). \end{array}

 $\begin{array}{l} {\rm Spiro[3-(4-methoxyphenyl)-1-phenyl-5-pyrazolone)-4,3'-(2',5'-diphenyl-4'-methylthio-5'-pyrroline)]} (8c) yellow crystals; 71%; IR <math display="inline">\mathcal{Y}_{max}$ 1710, 1625, 1608, 1598 cm⁻¹; ¹H NMR (CDCl₃) & 2.18 (s,3H,SCH₃); 3.69 (s,3H, OCH₃); 5.00 (d, J=2Hz,1H,H_A); 6.02 (d,J=2Hz,H_B); 6.56-6.74 (m,2H,arom); 7.02-7.65 (m,13H,arom); 8.01-8.29 (m,4H,arom); (Found:C,74.38; H,5.16; N,8.32. Calc. for C₃₂H₂₇O₂N₃S: C, 74.25,H,5.26; N,8.12%). \end{array}

1-(Methylthio)-2-N(α -methylthiobenzyl)-3-phenyl-4,5,6,7-tetrahydroisoindole (12a) colourless crystals; 77%; m.p. 155°C, IR ν_{max} 1605,1571,1520 cm⁻¹; ¹H NMR (CDCl₃) δ 1.62-1.82 (m,4H,-CH₂-); 1.90 (s,3H,SCH₃); 2.15 (s, 3H, SCH₃); 2.50-2.74 (m,4H,-CH₂-); 6.38 (s,1H,H-3,ArCH); 7.21-7.50 (m,10H,arom). ¹³C NMR (CDCl₃) δ 14.27,20.03 (SCH₃); 22.21,23.24,23.43,24.35 (CH₂); 64.10 (C₆H₅CH); 110.42(C-7a);112.52 (C-3a);126.64,127.64,127.77,128.21,128.49, 129.25 (CH,arom);130.93,132.09(C-,Ar);137.60,138.87 (C-1,C-3);(Found: C,72.53;H,6.82; N,3.43. Calc.for C₂₃H₂₅NS₂: C,72.78; H,6.64; N,3.69%); m/z 379 (M⁺,96%); 332(40).

Mercuric Chloride Catalyzed Hydrolytic Cleavage of <u>12a</u> and <u>12b</u>: General Procedure: A suspension of N-substituted isoindole <u>12a</u> or <u>12b(1 mmol)</u> and HgCl₂ (0.27g, 1 mmol) in dry THF (10 ml) was stirred at room temperature

for 6 hr. The reaction mixture was filtered through a G-4 sintered funnel to remove traces of mercuric chloride and the filtrate diluted with chloroform (25 ml) washed with saturated sodium bicarbonate solution and evaporated to give the crude products <u>13a</u> or <u>13b</u>, which were purified by column chromatography over silica gel using hexane-EtOAc (20:1) as eluent.

1-(Methylthio)-3-phenyl-4,5,6,7-tetrahydroiosindole (13a) colourless crystals (EtOAc-hexane);73%;m.p.91-92°C;IR)_{max} 3320,1600,1585,1512 cm⁻¹; ¹H NMR (CDCl₃) δ 1.71-1.92 (m,4H,CH₂); 2.19 (s,3H,SCH₃); 2.48-2.72 (m,4H,CH₂); 7.21-7.54 (m,3H,arom); 7.61-7.82 (m,2H,arom); 8.01 (brs,1H,NH exchangeable with D₂O); (Found: C,74.12; H,7.28; N,5.54. Calc. for C₁₅H₁₇NS: C, 74.03, H,7.04; N,5.76%); m/z 243 (M⁺,100%) 228(41).

2~(Methylthio)-8-phenyl-4,5,6,7-tetrahydro-3H-cyclohepta[c]pyrrole (13b) colourless crystals (EtOAc-hexane); 68%, m.p. 90-91°C; IR)_{max} 3325, 1605, 1598 cm⁻¹, ¹H NMR (CCl₄) δ 1.45-1.82 (m,6H,CH₂); 1.99 (s, 3H, SCH₃); 2.38-2.79 (m,4H,CH₂); 7.01-7.35 (m,3H,arom); 7.42-7.64 (m,2H,arom); 7.74 (brs, 1H,NH exchangeble with D₂O); (Found: C,74.40; H,7.53; N,5.23. Calc. for C₁₆H₁₉NS: C,74.66; H,7.44; N,5.44%); m/z 257 (M⁺,100%); 242(25).

Lithium Bromide/Triethylamine Induced Cycloaddition of Ethyl(benzylideneamino)acetate with α-Oxoketene Dithioacetals: General Procedure: Το a solution of 17 (2.10g, 11 mmol) and the appropriate oxoketene dithioacetal (4a, 4c, 4g) (10 mmol) in dry THF (25 ml) was added lithium bromide (1.30g, 15 mmol) in THF (10 ml) and triethylamine (1.21g, 12 mmol) in THF (5 ml) with the help of a syringe. The mixture was stirred at room temperature for 14-16 hr under nitrogen (checked by TLC) and poured into concentrated aqueous ammonium chloride (50 ml), extracted with chloroform (50mlx3), dried over sodium sulphate and evaporated in vaccuo. The residue was chromatographed over silica gel by using hexane-EtOAc (5:1) as eluent to give the corresponding pyrrolidines, 18a,b and pyran-2-one 19c. When the same reaction mixture from <u>4a</u> and <u>4c</u>, after an initial 3 hr stirring at room temperature was refluxed for 16 hr at 70°C under an efficient atmosphere of nitrogen, work-up as described gave the pyran-2-one 19a and 19b. Pyrroles 25a, b, c were obtained when the same reaction mixture from 4c,d,e after an initial 3 hr stirring at room temperature was refluxed for 30 hr at 70°C in the absence of nitrogen.

Ethyl-4-benzoyl-3,3-bis(methylthio)-5-phenylpyrrolidine-2-carboxylate (18a) colourless crystals (EtOAc-hexane); 72%, m.p. 154°C; IR) max 3300,1750, 1680, 1600 cm⁻¹; ¹H NMR (CDCl₃) δ 1.39 (t,J=7Hz, 3H, CH₃); 1.98 (s, 3H,SCH₃); 2.21 (s,3H,SCH₃); 2.59-2.71 (brs,1H,NH, exchangeable with D₂O) 4.31 (q,J=7Hz,2H,OCH₂); 4.68 (d,J=7Hz,1H,H-4);4.96 (brs,1H,H-2); 5.41 (brd, J=7Hz,1H,H-5); 6.98-7.69 (m,10H,arom); ¹³C NMR (CDCl₃) 12.95 (CH₃); 13.81, 14.04 (SCH₃); 61.29 (OCH₂); 61.78, 63.78, 70.13 (CH,C-2,C-4,C-5); 70.69 (C-3); 126.93, 127.26, 127.63, 128.21, 132.61 (<u>CH</u>-phenyl) 139.60, 139.61 (quaternary Ar) 170.81 (<u>C</u>-OEt); 197.44 (C₆H₅C-); (Found: C, 63.39; H,6.01, N,3.22. Calc. for C₂₂H₂₅NO₃S₂: C,63.58; H,6.06; N,3.37%); m/z 368 (M⁺-47, 100%); 225(59).

Ethyl-4-(2-furoyl)-3,3-bis(methylthio)-5-phenylpyrrolidine-2-carboxylate (18b) colourless crystals (EtOAc-hexane); 69%; m.p. 135-136°C; IRV_{max} 3340, 1723, 1660, 1565 cm⁻¹; ¹H NMR (CDCl₃) δ 1.49 (t, J=7Hz,3H,CH₃); 2.02 (s, 3H,SCH₃); 2.21 (s,3H,SCH₃); 3.05-3.20 (brs,1H,NH); 4.25 (q,J=7Hz,2H,OCH₂); 4.49 (d,J=7Hz,1H,H-4); 4.89 (brs,1H,H-3); 5.38 (d,J=6.5Hz,1H,H-5); 6.48 (dd, J=3 and 1.5Hz, H-4' furyl); 6.87-7.49 (m,7H,arom and furyl); (Found:C,59.12; H, 5.61; N,3.48. Calc. for C₂₀H₂₃NO₄S₂: C,59.23; H,5.72; N,3.45%); ,m/z 358 (M⁺-47,14%); 312(4).

3-Benzylideneamino-4-(methylthio)-6-phenylpyran-2-one (19a) yellow crystals (EtOAc-hexane); 69%; m.p. 164-165°C; IR \mathcal{D}_{max} 1704,1688,1618,1575 cm⁻¹; ¹H NMR (CDCl₃) δ 2.46 (s,3H,SCH₃); 6.68 (s,1H,H-5); 7.30-7.59 (m,6H,arom); 7.78-8.02 (m,4H,arom); 9.49 (s,1H,CH=N-); (Found:C,71.18; H,4.79; N,4.46. Calc.for C₁₉H₁₅NO₂S: C,71.00; H,4.70; N,4.36%); m/z 321 (M⁺,61%); 306(100).

3-Benzylideneamino-6-(2-furyl)-4-(methylthio)pyran-2-one (19b) yellow crystals (EtOAc-hexane); 61%; m.p. $119-120^{\circ}C$; IR)_{max} 1705,1692,1625,1562 cm⁻¹; ¹H NMR (CDCl₃) δ 2.56 (s,3H,SCH₃); 6.58 (dd,J=3.0 and 1.5Hz, H-4'furyl); 6.73 (s,1H,H-5); 7.10 (d,J=3Hz,H-3'furyl); 7.31-7.53 (m,4H,arom); 7.81-8.04 (m,2H,arom); 8.52 (s,1H,C₆H₅<u>CH</u>=N-); (Found: C,65.32; H,4.38; N,4.68. Calc. for C₁₇H₁₃NO₃S: C,65.58; H,4.21; N,4.68%); m/z 311 (M⁺, 3%); 296(4); 214(22).

3-Benzylideneamino-6-methyl-4-(methylthio)pyran-2-one (19c) yellow crystals (EtOAc-hexane); 81%; m.p. 139-140°C; IR)_{max} 1703,1690,1622,1598 cm⁻¹; ¹H NMR (CDCl₃) δ 2.28 (s,3H,CH₃); 2.43 (s,3H,SCH₃); 6.12 (s,1H,H-5); 7.31-7.56 (m,3H,arom); 7.74-8.08 (m,2H,arom); 9.53 (s,1H,C₆H₅C<u>H</u>=N-); ¹³C NMR (CDCl₃) 14.19 (CH₃); 19.78 (SCH₃); 101.22 (C-5); 122.95 (C-3);127.67,128.68,130.91 (CH,arom) 137.50 (quaternary Ar); 155.40 (C-4);156.16 (C-6);157.93 (C=0); 159.59 (C<u>H</u>=N-); (Found: C,64.80; H,5.26; N,5.59.Calc.for C₁₄H₁₃NO₂S: C, 64.84; H,5.05; N,5.40%); m/z 259 (M⁺ 60%); 244(100), 216(22).

Bthyl-4-(2-furoyl)-3-(methylthio)-5-phenylpyrrole-2-carboxylate (25a) colourless crystals (EtOAc-hexane); 73%; m.p.117-118°C; $IR\mathcal{Y}_{max}$ 3310,1680, 1668,1565 cm⁻¹; ¹H NMR (CDCl₃) δ 1.38 (t,J=7Hz,3H,CH₃); 2.35 (s,3H,SCH₃); 4.30 (q,J=7Hz,2H,OCH₂); 6.36 (dd,J=3 and 1.5Hz,1H,H-4'furyl); 6.91 (d, J=3Hz,1H,H-3'furyl); 714-7.52 (m,6H,arom and 5'furyl); 10.01 (brs,1H,NH exchangeable with D₂O); (Found: C,64.02;H,4.96; N,3.83. Calc.for C₁₉H₁₇NO₄S: C,64.21; H,4.82; N,3.94%); m/z 356 (M⁺+1,100%); 309(17); 280(18).

Bthyl 3-(methylthio)-5-phenyl-4-(2-thienoyl)pyrrole-2-carboxylate (25b) colourless crystals (EtOAc-hexane); 69%; m.p. 128-129°C; IR)_{max} 3260,1680, 1620,1550 cm⁻¹; ¹H NMR (CDCl₃) δ 1.35 (t,J=7Hz,3H,CH₃); 2.45 (s, 3H, SCH₃); 4.31 (q,J=7Hz,2H,OCH₂); 7.05-7.49 (m,6H,arom and thienyl); 7.50-7.98 (m,2H,arom); 10.11 (brs,1H,NH, exchangeable with D₂O) (Found: C,61.48; H,4.58; N,3.71; Calc.for C₁₉H₁₇NO₃S₂: C,61.43;H,4.60;N,3.77%).

Ethyl 3-(methylthio)-5-phenyl-4-(3-pyridinoyl)pyrrole-2-carboxylate (25c) colourless cyrstals(EtOAc-hexane);79%; m.p.141-142°C;IR) max 3410,1718,1640 cm⁻¹;¹H NMR(CDCl₃) δ 1.32 (t,J=7Hz,3H,CH₃);2.31(s,3H,SCH₃);4.31(q,J=7Hz,2H, OCH₂);7.21-7.49 (m,6H,arom and H-5'pyridyl); 8.08 (dt,J=8 and 1.5Hz,1H,H-4' pyridyl);8.65(dd,J=6 and 1.5Hz,1H,H-6'pyridyl);8.98(d,J=1.5Hz,H-2'pyridyl); 10.61(brs,1H,NH,exchangeable with D₂O); (Found:C,65.32;H,4.91,N,7.72;Calc. for C₂₀H₁₈N₂O₃S:C,65.55;H,4.95;N,7.65%);m/z 366 (M⁺,100%); 320(34).

3,3-Bis (methylthio)-2,5-diphenyl-4-nitropyrrolidine (28) colourless crystals (EtOAc-hexane); 64%; m.p. 118°C; IR)_{max} 3342,1600 cm⁻¹; ¹H NMR (CDCl₃) δ 1.26 (s,3H,SCH₃); 2.08 (s,3H,SCH₃); 2.39 (brs,1H, exchangeable with D₂O); 4.80 (brs,1H,H-2); 4.69 (d,J=7Hz,1H,H-4); 4.92 (brd,J=7Hz, 1H, H-5); 7.12-7.75 (m,10H,arom); (Found: C,59.81; H,5.48; N,3.63. Calc. for C₁₈H₂₀N₂O₂S₂: C, 59.97; H,5.59; N,3.89%); m/z 360 (M⁺,4%), 266(37).

Ethyl 3-nitro-2-phenylpyrrole-5-carboxylate (31) colourless crystals (EtOAc-hexane); 56%; m.p. 128-129°C; IR) max 3250,1700,1580,1510 cm⁻¹; ¹H NMR (CDCl₃) δ 1.22 (t,J=7Hz,3H,CH₃); 4.12 (q,J=7Hz,2H,OCH₂); 7.08-8.02 (m,6H,H-4 and arom); 10.35 (brs,1H,NH,exchangeable with D₂O); (Found: C, 59.68; H,4.59; N,10.73. Calc.for C₁₃H₁₂N₂O₄: C,59.99; H,4.65; N,10.84%).

Acknowledgement: M.P.B. thanks CSIR, New Delhi for S.R.F. Financial assistance under COSSIST programme by UGC, New Delhi is also acknowledged.

REFERENCES AND NOTES

- (a) Singh,G.; Ila, H.; Junjappa, H. <u>Tetrahedron Lett.</u> 1984, <u>25</u>, 5095-5098; (b) Balu, M.P.; Singh,G.; Ila, H.; Junjappa,H. <u>Tetrahedron Lett.</u> 1986 <u>27</u>,117-120; (c) Gupta, A.K.; Ila,H.; Junjappa,H. <u>Tetrahedron Lett.</u> 1987, <u>23</u>, 1459-1462; (d) Balu,M.P.; Pooranchand,D; Ila,H.; Junjappa, H. <u>Tetrahedron Lett.</u> 1988, <u>29</u>, 501-504; (e) Datta,A.; Ila,H.; Junjappa, H. <u>Tetrahedron Lett.</u> 1988, <u>29</u>, 497-500.
- (a)Gupta,A.K.;Ila,H.;Junjappa,H.<u>Tetrahedron Lett.</u>1988,<u>29</u>,6633-6636; (b) Balu, M.P.;Ila,H.;Junjappa,H.<u>Tetrahedron Lett.</u>1987,<u>28</u>, 3023-3026.
- 3. Wittig, G.; Reif, H. Angew. Chem. Int. Ed. Engl. 1968, 7 7-14.
- (a) Kauffmann, Th. <u>Angew.Chem.Int.Ed.Engl.</u> 1974, <u>13</u>, 627-639; (b) Kauffmann, Th.; Ahlers, H.; Echsler, K-J.; Schultz, H.; Tilhard, H.J. <u>Chem.Ber.</u> 1985; <u>118</u>, 4496-4506.
- 5. (a) Sinbandhit, S.; Hamelin, J. <u>J. Chem. Soc. Chem. Comm.</u> 1977, 768-769; (b) Vo Quang, L.; Gaessler, H.; Vo Quang, Y. <u>Angew. Chem. Int. Ed. Engl.</u> 1981, <u>20</u>, 880-881.
- 6. Grigg.R.; Kemp,J.; <u>Tetrahedron Lett.</u> 1978, 2823-2826.
- 7. Tsuge, O.; Veno, K. Heterocycles 1982, 19, 1411-1414.
- (a) Barr,D.A.; Grigg,R., Gunaratne,H.Q.N.; Kemp.J; McMeckum,P.; Sridharan,V. <u>Tetrahedron</u> 1988, <u>44</u>, 557; (b) Grigg,R.; Mongkolaussavaratana,T.<u>J.Chem.Soc.Perkin</u> <u>Trans</u> <u>1</u>, 1988,541-544; (c)Tsuge,O.;Kanemasa,S.; Yoshioka,M. <u>J.Org.Chem.</u> 1988, <u>53</u>, 1384-1391 and references therein.
- 9. (a) Grigg,R.; Kemp,J.; Maone,J.; Tangthongkum,A. <u>J. Chem. Soc. Chem. Comm.</u> 1980, 648-650; (b) Grigg.R.; Kemp,J.; Malone, J.R.; Rajviroongit,S.; Tangthoughkum,A. <u>Tetrahedron</u> 1988, <u>44</u>, 5361-5374.
- 10. (a) Grigg, R.; Kemp, J; Sheldrick, G. Tratter, J. <u>J.C.S.Chem.Comm.</u> 1978, 109-111; (b) Grigg, R. <u>Chem.Soc.Rev.</u> 1987, <u>16</u>, 89-121; (c) Tsuge, O.; Kanemasa, S. <u>Adv.Heterocycl.Chem.</u> 1989, <u>4</u>5, 231-349.
- 11. (a) Grigg,R.; Gunaratne, H.Q.N. J.C.S.Chem.Comm. 1982, 384-386; (b) Grigg.R.; Gunaratne,H.Q.N.; Sridharan.V. <u>Tetrahedron</u> 1987, <u>43</u>, 5887-5899; (c) Amornvaska, K.; Barr,D.; Donegan,G.; Grigg.R.; Ratnanukul,P.; Sridharan, V. <u>Tetrahedron</u> 1989, <u>45</u>, 4649-4668; (d) Barr,D.A.; Grigg,R.; Sridharan,V. <u>Tetrahedron Lett.</u> 1989, <u>30</u>, 4727-4730; (e) Barr, D.A.; Donegan,G.; Grigg,R. J.Chem.Soc.Perkin.Trans I 1989, 1550-1551.
- 12. Tsuge, O.; Kaneamasa, S.; Takenaka, S. B<u>ull.Chem.Soc.Japan</u> **1985**, 58, 3320-3336.
- 13. Chauhan, S.M.S.; Junjappa, H. <u>Tetrahedron</u> 1970, 32, 1779-1787.