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Transient Directing Group Enabled Pd-catalyzed γ -C(sp³)-H Oxygenation of Alkyl Amines

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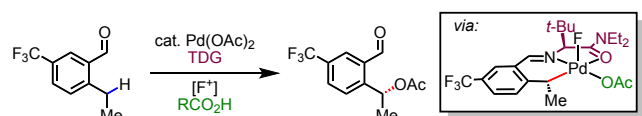
ABSTRACT: We report a general protocol for γ -C(sp³)-H acyloxylation and alkoxylation of free amines using 2-hydroxynicotinaldehyde as the transient directing group. In the presence of an electrophilic fluorinating bystanding oxidant and acetic acid, a wide range of aliphatic amines could be oxygenated selectively at the γ -methyl positions. A vast variety of aryl, heteroaryl, and aliphatic acids could also be successfully coupled under this C–O bond formation reaction to afford amine containing esters. Switching the nucleophile from acids to alcohols enables alkoxylation of free amines. Importantly, natural products and drug molecules such as ibuprofen, isozepac, fenbufen, and lithocholic acid are all compatible coupling partners. Notably, synthesis of these mono-protected amino alcohols from free amino alcohols using conventional selective protection are not always feasible.

KEYWORDS: Transient Directing Group, C(sp³)-H Activation, Palladium, C(sp³)-H Oxygenation, Bystanding Oxidant, Amine

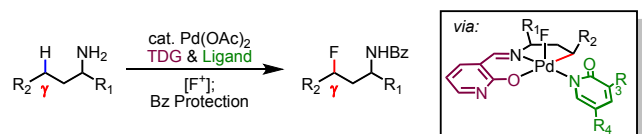
One of the major challenges in directed C–H activation is to enable the use of native substrates without covalently installing external directing groups. While the combination of weak coordination from substrates and ligand acceleration has been successful in enabling a wide range of C–H activation reactions of free carboxylic acids and Weinreb amides,¹ the development of catalytic transient directing groups (TDGs) for aliphatic aldehydes, ketones, and free amines has emerged as a complementary approach.^{2–5} While most of these reactions are limited to (hetero)arylations,^{3–5} efforts to develop other C–X, and C–O bond forming reactions have recently afforded new advances.⁶ Using an electrophilic [F⁺] reagent as either the bystanding oxidant or the fluorinating agent, we have recently developed selective C–O and C–F bond formation reactions of 2-alkylbenzaldehydes via a novel amino amide transient directing group (**Scheme 1**).^{6a} Inspired by this development, we have subsequently developed the first example of TDG enabled γ -C(sp³)-H fluorination of free amines using similar [F⁺] reagents (**Scheme 2**).^{6b} Based on this particular investigation and our previous studies of using [F⁺] as bystanding oxidant for making C–C, C–N or C–O bonds,⁷ we decided to explore the possibility of γ -oxidation of free amines using transient directing groups.

Considering the synthetic utility, we began to develop γ -acyloxylation with carboxylic acids and γ -etherification with alcohols which could afford of mono-protected amino alcohols that are not trivial to make. Although γ -C(sp³)-H acetoxylation of amines have been reported before, most of these reactions require the installation of exogenous directing groups.^{8–9} A single example of free amine directed γ -C(sp³)-H acetoxylation is limited to amines containing α -quaternary centers (**Scheme 3**).¹⁰ Herein, we report a transient directing group strategy for γ -C(sp³)-H oxygenation of broad range of free amines using N-fluoro-2,4,6-trimethylpyridinium tetrafluoroborate as the bystanding oxidant (**Scheme 4**). Advantageous to traditional γ -

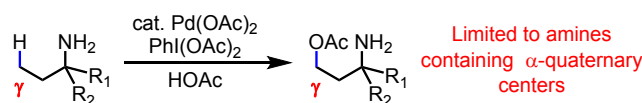
amino alcohol derivative synthesis which requires protection and deprotection of the amino group, our protocol features a one-step coupling between a wide range of aliphatic amines and a vast variety of acids and alcohols via selective C–O bond formation at the unactivated γ positions.



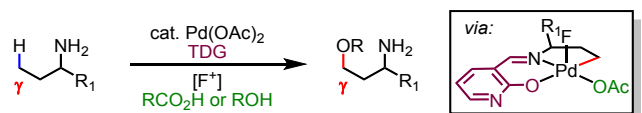
Scheme 1. TDG Enabled Acetoxylation of 2-Alkylbenzaldehydes



Scheme 2. Pyridone Assisted γ -C(sp³)-H Fluorination of Free Amines



Scheme 3. Free Amine Directed γ -C(sp³)-H Acetoxylation



Scheme 4. This Work: TDG Enabled γ -C(sp³)-H Acyloxylation and Alkoxylation of Free Amines

Table 1. Initial Studies and Reaction Optimization of γ -C(sp³)-H Acetoxylation of Amines^{a,b}

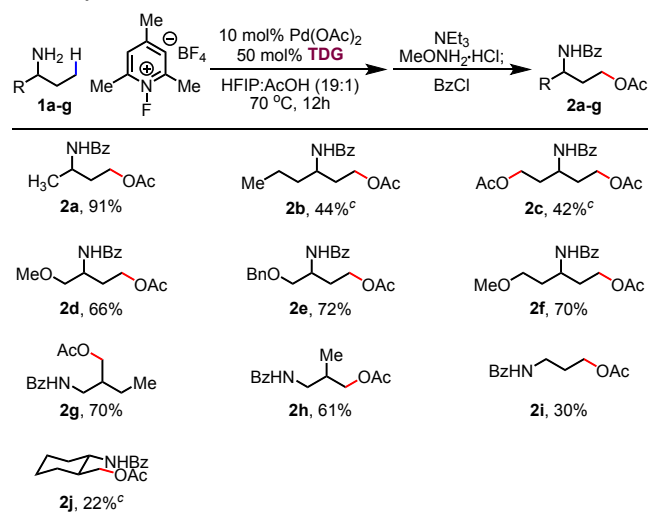
Entry	Oxidants	Reaction Conditions	Yield (%)
1	1.50 eq. N-fluoro-2,4,6-trimethylpyridinium tetrafluoroborate	20 mol% TDG, HFIP:AcOH (19:1)	68 (63) ^c
2	1.50 eq. N-fluoro-2,4,6-trimethylpyridinium tetrafluoroborate	No TDG, HFIP:AcOH (19:1)	<10
3	1.50 eq. PhI(OAc) ₂	20 mol% TDG, HFIP:AcOH (19:1)	17
4	1.50 eq. H ₂ O ₂	20 mol% TDG, HFIP:AcOH (19:1)	15
5	1.50 eq. AcOOtBu	20 mol% TDG, HFIP:AcOH (19:1)	21
6	1.50 eq. TBHP (5M in decane)	20 mol% TDG, HFIP:AcOH (19:1)	53
7	1.50 eq. K ₂ S ₂ O ₈	20 mol% TDG, HFIP:AcOH (19:1)	36
8	2.0 eq. TBHP (5M in decane)	50 mol% TDG, HFIP:AcOH (19:1)	68 (67) ^c
9	2.0 eq. K ₂ S ₂ O ₈ + 10 mol% (nBu) ₄ NOAc	30 mol% TDG, HFIP:AcOH (2:1)	66
10	2.0 eq. N-fluoro-2,4,6-trimethylpyridinium tetrafluoroborate	30 mol% TDG, HFIP:AcOH (19:1)	89
11	2.0 eq. N-fluoro-2,4,6-trimethylpyridinium tetrafluoroborate	30 mol% TDG, HFIP:AcOH (2:1)	89
12	2.0 eq. N-fluoro-2,4,6-trimethylpyridinium tetrafluoroborate	50 mol% TDG, HFIP:AcOH (19:1)	94 (91)^c

^aExperiments were performed with substrate **1a** (0.10 mmol), Pd(OAc)₂ (0.01 mmol), TDG, oxidants, HFIP:AcOH (0.5 mL), 70 °C, 12 h. ^bYields were determined by ¹H NMR analysis of the crude product using 1,1,2,2-tetrachloroethane as the internal standard. ^cIsolated yields.

We began our experimental efforts by testing γ -acetoxylation of *sec*-butylamine (**Table 1**). Under similar reaction conditions as our previous γ -methyl fluorination conditions in a 19:1 mixture HFIP and AcOH,^{6b} we were pleased to find the desired acetoxyated product in 68% NMR yield when N-fluoro-2,4,6-trimethylpyridinium tetrafluoroborate was used as the [F⁺] bystander oxidant and 2-hydroxynicotinaldehyde as the TDG (**Table 1**, entry 1). Only less than 10% of the product was detected in the absence of TDG, thus demonstrating the importance of TDG in this acetoxylation reaction. (entry 2). Since we had already established moderate reactivity with pyridinium salt, we wondered whether other cheaper and more accessible oxidants could also promote this reaction. When we changed the oxidant to PhI(OAc)₂ the yield lowered to only 17% (entry 3). Upon evaluation of various peroxide-based oxidants, we discovered that although H₂O₂ and AcOOtBu gave inferior yields of the product in 15% and 21% respectively (entry 4-5), TBHP was an effective oxidant for this reaction, providing the product in 53% yield (entry 6). K₂S₂O₈ was also identified as another potentially effective oxidant for this reaction (entry 7). Unfortunately, despite extensive reaction optimization, the reaction could only be improved to around 70% yield using TBHP or K₂S₂O₈ as the oxidant (entry 8-9). Gratifyingly, the reaction could be further optimized to above 90% isolated yield by increasing the loading of the TDG with 2.0 eq. of N-fluoro-2,4,6-trimethylpyridinium tetrafluoroborate (entry 10-12).

With the optimized conditions in hand, we next investigated the amine scope of this acetoxylation reaction (**Table 2**). For ease of analysis and separation, the acetoxyated products were isolated in the form of the Bz-protected amines. While *sec*-butylamine could be acetoxyated in 91% isolated yield (**2a**), the yield lowered to 44% when 3-aminoheptane was employed as the substrate (**2b**). For 3-aminopentane, the diacetoxyated product was obtained as the sole product in 42% yield (**2c**). Various oxygen containing substrates could also be acetoxyated in moderate yields (**2d-f**). Simple aliphatic amines without α -alkyl substituents such as 2-methylbutylamine,

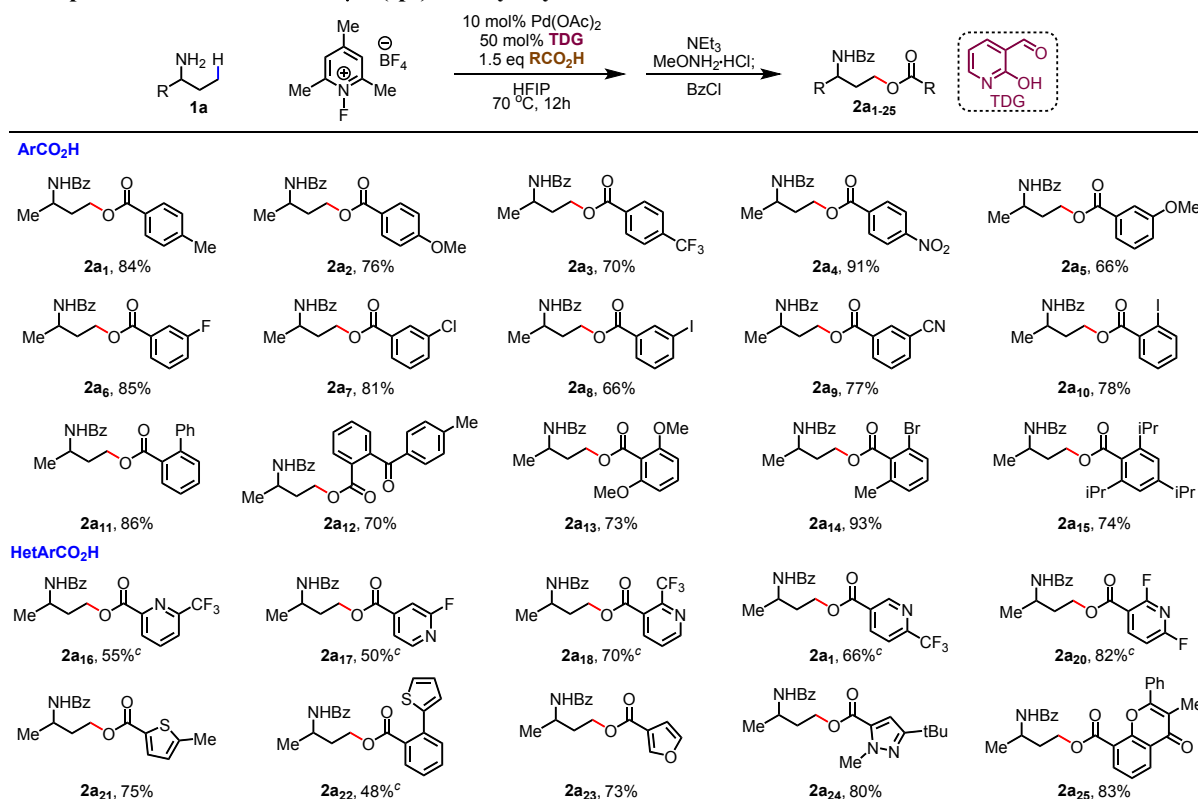
isobutylamine and even propylamine were also compatible substrates for this reaction, affording the corresponding products in 30–70% yields (**2g-i**). Acetoxylation of *trans*-2-methyl-cyclohexylamine could afford the product in 22% yield, whereas the *cis* isomer was completely unreactive (**2j**). Since AcOH was a good oxygen source for this protocol, we wondered whether other acids could also be coupled similarly for this reaction. We were pleased to find that γ -C(sp³)-H oxygenation of **1a** with a vast variety of aryl acids proceeded smoothly with good to excellent yields (**Table 3**). Notably, no acetoxylation product was observed in the presence of a different carboxylic acid without AcOH in pure HFIP. Various benzoic acids containing *para* and *meta* substituents, including methyl, methoxy, trifluoromethyl, nitro, cyano, and halogens, were well tolerated regardless of their electronic properties (**2a₁₋₉**). *Ortho*-substituted benzoic acids containing iodo or phenyl groups provided the corresponding products in good to excellent yields (**2a₁₀₋₁₂**). Sterically demand-

Table 2. Scope of Alkyl Amines for γ -C(sp³)-H Acetoxylation^{a,b}

^aExperiments were performed with substrate **1a** (0.10 mmol), Pd(OAc)₂ (0.01 mmol), **TDG** (0.05 mmol), N-fluoro 2,4,6-methylpyridinium

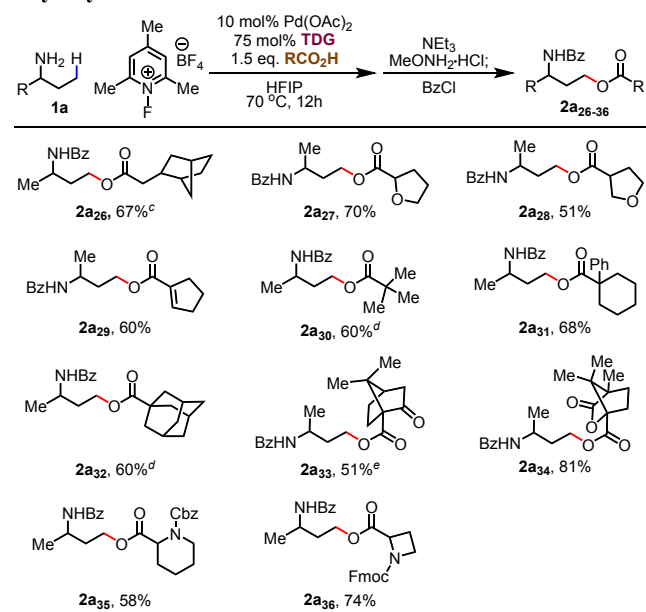
tetrafluoroborate (0.20 mmol), HFIP:AcOH (19:1) (0.5 mL), 70 °C, 16 h. ^bIsolated yields. ^cReaction performed with 0.75 eq of **TDG**.

Table 3. Scope of Aromatic Acids for γ -C(sp³)-H Acyloxylation of Free Amines^{a,b}



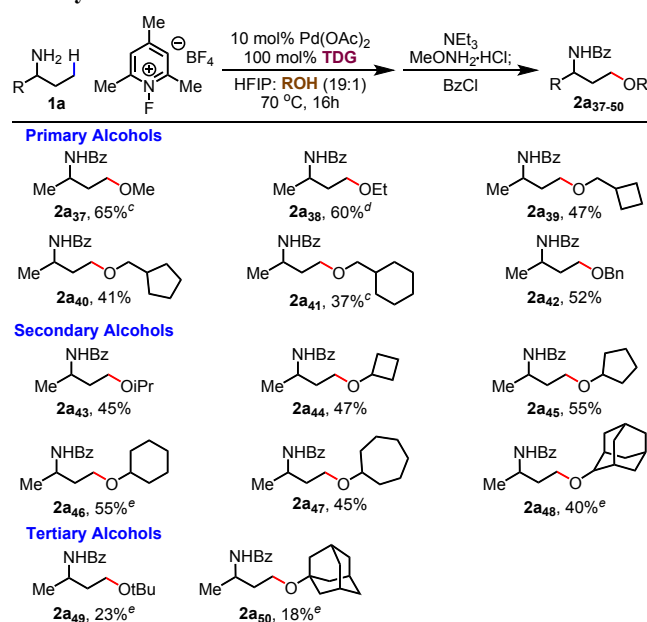
^aExperiments were performed with substrate **1a** (0.10 mmol), Pd(OAc)₂ (0.01 mmol), **TDG** (0.05 mmol), N-fluoro 2,4,6-methylpyridinium tetrafluoroborate (0.20 mmol), HFIP (0.5 mL), 70 °C, 16 h. ^bIsolated yields. ^cReaction performed with 0.75 eq of **TDG**.

Table 4. Scope of Aliphatic Acids for γ -C(sp³)-H Acyloxylation of Free Amines^{a,b}



^aExperiments were performed with substrate **1a** (0.10 mmol), Pd(OAc)₂ (0.01 mmol), **TDG** (0.075 mmol), N-fluoro 2,4,6-methylpyridinium tetrafluoroborate (0.20 mmol), HFIP (0.5 mL), 70 °C, 16 h. ^bIsolated yields. ^cReaction performed with 0.5 eq of **TDG**. ^dReaction performed with 2.0 eq of acid. ^eReaction performed with 1.0 eq of **TDG**.

Table 5. Scope of Aliphatic Alcohols for γ -C(sp³)-H Alkoxylation of Free Amines^{a,b}



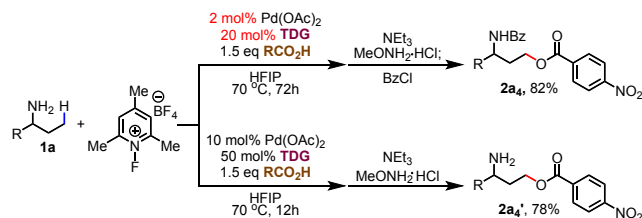
^aExperiments were performed with substrate **1a** (0.10 mmol), Pd(OAc)₂ (0.01 mmol), **TDG** (0.1 mmol), N-fluoro 2,4,6-methylpyridinium tetrafluoroborate (0.20 mmol), HFIP:ROH (19:1) (0.5 mL), 70 °C, 16 h. ^bIsolated yields. ^cReaction performed with 0.75 eq **TDG** (0.075 mmol) and 0.5 eq of 5-chloro-3-nitropyridin-2-ol as the pyridone ligand. ^dReaction

performed with 0.5 eq of **TDG**. ^cReaction performed with 1.0 eq **TDG** (0.1 mmol) and 0.5 eq of 5-chloro-3-nitropyridin-2-ol as the pyridone ligand.

ing substrates such as 2,6-dimethoxy, 2-bromo-6-methyl, and even 2,4,6-triisopropyl benzoic acids were also well tolerated (**2a₁₃₋₁₅**). Importantly, heteroaryl acids containing pyridines with different substitutions such as fluoro- and trifluoromethyl groups are all compatible coupling partners, providing 50–82% yields (**2a₁₆₋₂₀**). Coupling with other heteroaryl acids containing thiophene, furan, chromene, and even pyrazole groups were successfully accomplished with this protocol (**2a₂₁₋₂₅**). Notably, higher TDG loadings are used for certain heteroaryl acids presumably to prevent catalyst poisoning by the heteroaryl acids.

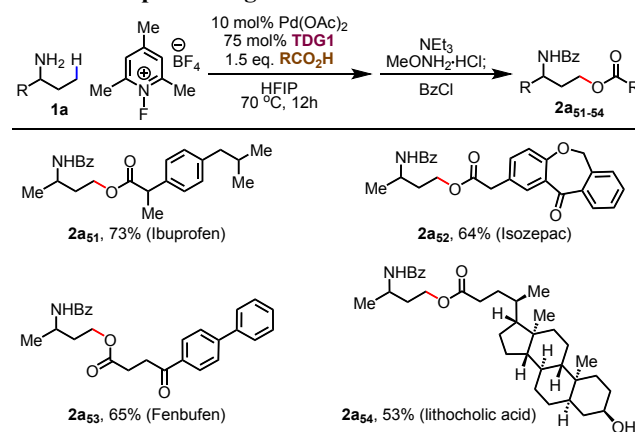
The scope of various aliphatic acids was investigated next (**Table 4**). Primary aliphatic acid such as 2-norbornaneacetic acid afforded the corresponding ester product in 67% isolated yield (**2a₂₆**). Secondary aliphatic acids containing oxygenated functionalities and unsaturation were also tolerated in moderate yields (**2a₂₇₋₂₉**). Sterically demanding tertiary acids containing tert-butyl, phenyl, and adamantyl groups were all coupled successfully in 60–68% isolated yields (**2a₃₀₋₃₂**). Aliphatic acids from natural products such as ketopinic acid and camphanic acid afforded the corresponding products in 51% and 81% respectively (**2a₃₃₋₃₄**). Lastly, N-protected amino acids could also be coupled under this protocol in good efficiencies (**2a₃₅₋₃₆**).

Since the reaction was compatible with a vast range of different carboxylic acids, we next wondered whether other oxygen containing nucleophiles such as alcohols could also serve as effective coupling partners for this reaction. We were pleased to find that γ -C(sp³)-H oxygenation of **1a** with a range of aliphatic alcohols could also be accomplished, albeit in lower efficiencies (**Table 5**). Primary alcohols such as methanol, ethanol, cyclobutylmethanol, cyclopentylmethanol, cyclohexylmethanol, and benzyl alcohol all underwent the desired alkoxylation reaction in 37–65% yields (**2a₃₇₋₄₂**). Secondary alcohols such as isopropanol, cyclobutanol, cyclopentanol, cyclohexanol, cycloheptanol, and even 2-adamantanol were also effective nucleophiles, affording the hindered ethers in 40–55% isolated yields (**2a₄₃₋₄₈**). Remarkably, even tertiary alcohols such as *t*BuOH, and 1-adamantanol were competent nucleophiles to afford the highly hindered ethers in 23% and 18% isolated yields respectively with the aid of an electron-deficient pyridone ligand (**2a₄₉₋₅₀**). The ability to perform C–H oxygenation with secondary and tertiary alcohols provides a straight-forward method for the synthesis of hindered aminoethers, which is difficult to achieve via conventional approaches. Furthermore, we have also evaluated other less reactive amines (**1c-d**, **1g-h**, **1i**) with methanol as the coupling partner. The NMR yields of **1c** and **1d** were similar to that of the model substrate **1a**. Roughly 40% NMR yield of the alkoxyated product was detected for **1g** and **1h**. Simple propylamine (**1i**) with methanol only gave 10–15% NMR yields.



Scheme 5. Catalyst Loading and Synthetic Applications

Table 6. Scope of Drug Derivatives^{a,b}



^aExperiments were performed with substrate **1a** (0.10 mmol), Pd(OAc)₂ (0.01 mmol), **TDG** (0.075 mmol), N-fluoro-2,4,6-methylpyridinium tetrafluoroborate (0.20 mmol), HFIP (0.5 mL), 70 °C, 16 h. ^bIsolated yields.

To demonstrate the synthetic utility of this γ -C(sp³)-H oxygenation reaction, we were pleased to find that the catalyst and TDG loading could be lowered to 2 mol% and 20 mol% respectively, thus rendering our protocol highly efficient (**Scheme 5**). In addition, the C–H oxygenated free amine product could also be isolated in 78% yield. More importantly, natural products and drug molecules such as ibuprofen, isozepam, fenbufen, and lithocholic acid were all well tolerated (**Table 6**), thus demonstrating this protocol's potential to perform late-stage functionalization and provide facile access to amine containing drug analogs.

In summary, we have developed a general method for γ -C(sp³)-H oxygenation of free aliphatic amines using 2-hydroxynicotinaldehyde as the transient directing group and N-fluoro-2,4,6-trimethylpyridinium tetrafluoroborate as the [F⁺] bystander oxidant. This reaction features a wide amine scope with good functional group tolerance. A vast range of aryl, heteroaryl, and aliphatic acids, as well as primary, secondary, and even tertiary alcohols are all compatible nucleophiles in this γ -C(sp³)-H oxygenation reaction. Compared to traditional γ -amino alcohol derivative synthesis which require protection and deprotection of the amino group, our reaction provides facile access to these compounds and even hindered ethers in a single step. More importantly, natural products and drug molecules are also competent coupling partners, thus demonstrating this protocol's potential for late-stage functionalization to access amine containing drug analogs. Efforts in designing new TDG and ligands to enable coupling of other nucleophiles are currently underway in light of these developments.

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Supporting Information. Experimental procedures and spectral data for all new compounds and full computational data (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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