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The Total Synthesis of (+)-Petasin and (+)-Isopetasin

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Petasin 1 and isopetasin 2, anti-inflammatory and analgesic components of the butterbur (*Petasites hybridus*), were prepared for the first time in a short and enantioselective synthesis. The key steps were a Michael addition/alkylation $(7 \rightarrow 8)$, an enamine alkylation to give 9 and a cyclization with N-phenyliminoketenylidenetriphenylphosphorane 11 $(10 \rightarrow 15)$.

Extracts of the butterbur (*Petasites hybridus*) have been used for centuries as a cure for inflammations and spasms.¹ Their therapeutic applications were described in the first century by Dioskurides, the physician of Claudius and Nero, in his 'De materia medica'. The major physiological active ingredients were identified² in the 1950s as the eremopilane sesquiterpenes (+)-petasin (1) and (+)-isopetasin (2). In the 1960s degradation and ORD-experiments confirmed the absolute configuration.³

Scheme 1

In recent years several highly interesting pharmacological properties have been attributed to the *Petasites* extracts.⁴ Besides analgetic and anti-inflammatory activities the extract exhibits a high cyto- and ulcusprotective activity and a high antiproliferation activity against human T24 bladder carcinoma cells in vitro.⁵ Studies of our group in cooperation with K. Brune⁶ showed that 1 and 2 also have a strong influence on the biosynthesis of leucotrienes. For further examinations larger amounts of these compounds were necessary. Because of the presence of a variety of similar sesquiterpenes in the extracts, the isolation of sufficient amounts of pure 1 and 2 from plant material was not successful. This prompted us to attempt a practical synthesis of enantiomerically pure 1 and 2.⁷

The only eremopilane found in the butterbur, that has been synthesized so far, is racemic (\pm) -isopetasol 18,8 the corresponding alcohol to isopetasin 2. Our synthesis starts with the readily available (1S,4R)-(-)-1-acetoxycyclohex-2-en-4-ol (4). Compound 4 has been synthesized via an enantioselective cleavage of a *meso* diester, prepared from cyclohexa-1,3-diene (3), by Kazlauskas and Bäckvall⁹ in > 97% ee. We performed this synthesis on a scale of over 100 mmols of 4 with an optimized total yield of 49%.

The alcohol 4 was protected as the *tert*-butyldimethylsilyl ether 5 (83%). The acetate group was cleaved with

PBQ = p-benzoquinone; CCL = lipase of *Candida cylandrea* Scheme 2

MeOH/ K_2 CO₃ (86%) to give **6**. For the determination of the enantiomeric purity of **6** the alcohol was esterified with (S)-3,3,3-trifluoro-2-methoxy-2-phenylpropionic acid [(S)-MTPA]. The ¹H NMR and ¹⁹F NMR spectra showed an enantiomeric eccess of > 90%. Collins oxidation of the allylic alcohol **6** gave (4R)-(+)-tert-(butyldimethylsiloxy)cyclohex-2-en-1-one (+)-7 (82%). Danishefsky has shown that the enantiomer (-)-7 is a versatile building block, which was used in several natural product syntheses. ¹⁰

The success of the subsequent Michael addition/alkylation was highly dependent upon the solvent and the temperature. For the initial dimethylcuprate addition diethyl ether was shown to be a superior solvent. Addition of a polar cosolvent was essential for the subsequent methylation of the intermediate enolate. The carcinogenic hexamethylphosphoramide (HMPA) could be successfully substituted by *N*-methylpyrrolidin-2-one (NMP)¹¹ with identical yields of **8** (70%). ¹H NMR spectroscopy of the crude product showed a 9:1 epimeric mixture in the 2-position. No 3-epimer could be detected. The relative configuration at C-3 and C-4 of **8** was assigned in analogy to the known antiselective cuprate additions to (-)-**7**¹⁰ and by subsequent transformation to known compounds.

Under a variety of conditions that we have examined to alkylate the thermodynamically more stable enolate of 8, we observed significant amounts of polyalkylation and low regio- and diastereoselectivity (2:1-3:1). Alkylations of imines of several achiral amines like benzylamine or cyclohexylamine with methyl acrylate were also not selective. In contrast, the imine with (R)-(+)-1-phenylethylamine that is formed, alkylated with methyl acrylate and cleaved in a one-pot procedure according to Pfau,12 gives 9 in a regio- and diastereoselective manner (selectivity 8:1; yield for the one-pot procedure 77%). The relative orientation of the stereocenters has been proven after the cyclization to 15 and is in accordance with the model of the enamine alkylation proposed by Pfau.¹² Cleavage of the ester with LiOH in water/THF gave the crystalline, diastereomerically pure oxo acid 10 (91%).

We have previously described the cyclization of γ - and δ -oxo acids with N-phenyliminoketenylidenetriphenyl-

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Scheme 3

phosphorane 1113 to cyclic enones,14 e.g. anti-Bredt alkenes. We describe now a one-pot procedure for this reaction. We treated 10 with 11 in toluene at 80°C, then added ethanol and heated the reaction mixture to reflux to give directly enone 15 in 67% yield. In this reaction the acid 10 reacts with 11 to form the bis-acyl ylid 12, which undergoes further reaction with the alcohol to form the acyl ylid 14 and a urethane 13. The acyl ylid 14 cyclizes in situ to form enone 15.

Irradiation at the ¹H NMR resonance frequencies of the two methyl groups of 15 showed a positive nuclear Overhauser Effect (NOE) for the carbinol proton. This indicates a cis-configuration of these groups as depicted in Scheme 4. Compound 15 has been used before as an intermediate in the synthesis of sporogene-AO 1 and related natural products. 8f In this synthesis 15 was prepared via a different route in 17 steps, compared to 11 steps starting from cyclohexadiene in our aproach (or 7 steps from known 4).

The zinc chloride catalyzed aldol reaction of the lithium enolate of 15 with acetone gave the alcohol 16 (90%) as a single diastereomer. The relative configuration has been assigned in analogy to the aldol reaction of similar compounds^{8a} and by subsequent transformation to (+)-petasin 1. Elimination of the tertiary alcohol of 16 with methanesulfonyl chloride and triethylamine resulted in a 3: 1 mixture of products with a terminal and an internal

Scheme 4

double bond. This mixture was directly isomerized with acidic alumina to the more stable isomer with the internal double bond, which is the TBDMS-ether of isopetasol 17 (75%). Cleavage of the TBDMS ether with aqueous HF/acetonitrile provided isopetasol 18 (83%).

Until recently¹⁵ no successful esterifications of alcohols with (Z)-2-methylbut-2-enoic acid (angelic acid) have been described because of the facile isomerisation of angelic esters under nucleophilic esterification conditions to the more stable tiglic esters. Activation of the acid with trichlorobenzoyl chloride¹⁶ and reaction with 18 gave isopetasin 2 (83%). In the ¹H NMR of the crude reaction mixture no isomeric ester could be detected.

Scheme 5

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The enantioselective total synthesis of isopetasin 2 has therefore been performed in 11 steps and 9.1% overall yield starting from 4.

Cleavage of the silvl ether of 16 with HF/acetonitrile (94%) gave 20. The secondary alcohol group of 20 was selectively esterified with the mixed anhydride 19. Due to the slow simultaneous elimination of the tertiary alcohol under the reaction conditions the reaction was stopped after ca. 75% conversion. Isopetasin 2 (20%), 20 (25%) and the monoester **21** (39%, 52% based on recovered 20) were isolated by chromatography on silica gel. Elimination of the tertiary alcohol of 21 with trifluoromethanesulfonic anhydride and diisopropylethylamine gave, as determined by gas chromatography, a 6:1 ratio of petasin 1 and isopetasin 2 in the crude product mixture. Petasin 1 isomerized slowly during workup and purification to isopetasin 2. Isopetasin 2 (17%), **21** (32%) and petasin **1** (48%, 62% based on reovered 21) were finally isolated by careful flash chromatography on silica gel.

The overall yield of petasin 1 was 4.3% over 11 steps starting from 4. The spectroscopical and physical properties of synthetic petasin 1, isopetasin 2, and isopetasol 18 were in agreement with those reported for the natural products.

All melting and boiling points are uncorrected. NMR spectra were recorded on a JEOL JNM GX 400 FT spectrometer. IR spectra were recorded on a Beckmann Acculab A8 spectrophotometer. Optical rotations were measured on a Schmidt & Haensch Polartronic E polarimeter. Mass spectra were recorded on a Varian MAT 311 spectrometer at 70 eV. Column chromatography was carried out on columns packed with Merck Kieselgel 60 (Art. Nr. 7734). For all new compounds satisfactory microanalyses (C,H \pm 0.4%) were obtained from a Hareaus Mikromat C-H-N. All reactions were carried out under N_2 . Solvents were purified according to the standard procedures.

Acetic Acid (1S,4R)-(-)-4-Hydroxycyclohex-2-enyl Ester⁹ (4):

The literature procedures have been scaled up without major changes. The workup on a larger scale of the palladium catalyzed oxidation of cyclohexadiene and the enzymatic ester hydrolysis proved to be difficult because of the formation of emulsions. The general

problem of the formation of emulsions could be solved by dilution of the crude reaction mixture with the same amount of $\rm Et_2O$ and slow addition of the mixture to half the volume of stirred $\rm Na_2SO_4$. The resulting slurry was washed five times with $\rm Et_2O$. The combined organic phases were extracted with sat. $\rm NaHCO_3$ and brine, dried $\rm (Na_2SO_4)$ and concentrated in vacuo to give the crude products. Further purification has been performed according to the literature procedures.

Acetic Acid (1R,4S)-(-)-(tert-Butyldimethylsiloxy)cyclohex-2-enyl Ester (5):

To a solution of 4 (16.2 g, 104 mmol) in DMF (70 mL), imidazole (17.9 g, 259 mmol) and TBDMSCl (18.8 g, 124 mmol) were added. The solution was stirred at 37 °C for 15 h. After cooling to r.t. the solution was added to $\rm H_2O$ (400 mL) and the aqueous phase was extracted with $\rm Et_2O$ (3 × 400 mL). The collected organic phases were washed with brine (2 × 100 mL), dried (Na₂SO₄) and concentrated in vacuo. The residue was distilled in a Kugelrohr aparatus (0.05 mbar, bath temperature 68–70 °C) to give 5 as a colorless oil; yield 23.3 g (82 %); $[\alpha]_D^{20}$ –35.3 (c = 1.6, CHCl₃).

IR (film): v = 2940, 2900, 2840, 1730, 1240, 1085 cm⁻¹.

¹H NMR (CDCl₃, TMS): δ = 5.83 (dd, 1 H, J = 3.6, 10.4 Hz), 5.71 (dd, 1 H, J = 3.7, 10.4 Hz), 5.12 (m, 1 H), 4.14 (m, 1 H), 2.01 (s, 3 H), 169–1.89 (m, 4 H), 0.87 (s, 9 H), 0.05 (s, 6 H).

 $^{13}\text{C NMR (CDCl}_3): \delta = 170.7, 136.4, 126.2, 67.0, 66.3, 28.4, 25.8, 25.3, 21.3, 18.1, <math display="inline">-4.7.$

MS: m/z (%) = 270 (M⁺, 1), 213 (4), 117 (100), 75 (60).

(1R,4S)-(+)-4-(tert-Butyldimethylsiloxy)cyclohex-2-enol (6):

Compound 5 (23.3 g, 86.2 mmol) was dissolved in MeOH (50 mL) then added to a suspension of K_2CO_3 (35.7 g, 258 mmol) in MeOH (300 mL) and stirred for 2 h at r.t. The suspension was filtered, the filtrate washed with MeOH (3 × 20 mL) and the solution concentrated in vacuo. The residue was dissolved in Et₂O (400 mL) washed with sat. NH₄Cl (2 × 200 mL), brine (2 × 200 mL), dried (Na₂SO₄) and concentrated in vacuo. The residue was chromatographed on silica gel (500 g) with hexanes/EtOAc (6:1 \rightarrow 1:1) to give 6 as a colorless oil; yield: 17.0 g (86%); bp (0.05 mbar, Kugelrohr bath temperature) 72–75°C; [α]_D²⁰ + 25.4 (c = 1.3, CHCl₃). IR (film): v = 3300, 2920, 2900, 2850, 1250, 1080, 1020 cm⁻¹.

 $^{1}\mathrm{H}$ NMR (CDCl₃, TMS): $\delta = 5.74$ (dd, 1 H, J = 3.1, 9.9 Hz), 5.66 (dd, 1 H, J = 2.5, 9.9 Hz), 4.07 (m, 1 H), 4.03 (m, 1 H), 2.54 (br, 1 H), 1.63 - 1.76 (m, 4 H), 0.84 (s, 9 H), 0.02 (s, 6 H).

¹³C NMR (CDCl₃): δ = 133.6, 130.7, 66.2, 64.6, 28.3, 28.2, 25.8, 18.1, 4.7.

MS: $m/z(\%) = 228 \text{ (M}^+, 1), 213 (2), 171 (95), 75 (100).$

(4R)-(+)-tert-Butyldimethylsiloxy)cyclohex-2-enone (7):

CrO₃ (43.6 g, 436 mmol) was added to a stirred solution of pyridine (68.8 g, 871 mmol) in CH₂Cl₂ (1 L) and the red-brown solution was stirred for 30 min at r.t. Compound 6 (16.6 g, 72.6 mmol) in CH₂Cl₂ (100 mL) was added and the resulting suspension was stirred at r.t. for 15 h. To avoid the formation of an emulsion the mixture was added to stirred Na₂SO₄ (400 g) and the flask was washed with additional CH₂Cl₂ (2 × 100 mL) which was added to the slurry. After stirring for 30 min the CH₂Cl₂ was decanted and the Na₂SO₄ was washed with CH₂Cl₂ (4 × 100 mL). The collected organic phases were washed with sat. NaHCO₃ (2 × 100 mL), 2% H₂SO₄ (2 × 100 mL), sat. NaHCO₃ (100 mL), brine (2 × 100 mL) and dried (Na₂SO₄). Concentration in vacuo and chromatography (300 g silica gel, 6:1 hexanes/EtOAc) gave 7 as a colorless oil; yield 13.4 g (81%); bp (0.05 mbar, Kugelrohr bath temperature) 60–65°C; [α]²⁰ +107.1 (c = 1.3, CHCl₃).

IR (film): v = 3000, 2940, 2840, 1675, 1375, 1250 cm⁻¹.

¹H NMR (CDCl₃, TMS): δ = 6.83 (ddd, 1 H, J = 10.2, 2.4, 1.7 Hz), 5.92 (ddd. 1H, J = 10.2, 2.0, 1.0 Hz), 4.53 (m, 1 H), 2.58 (m, 1 H), 2.35 (ddd, 1 H, J = 16.7, 12.4, 4.7 Hz), 2.21 (m, 1 H), 2.00 (m, 1 H), 0.92 (s, 9 H), 0.13, 0.12 (2s, 6 H).

MS: $m/z(\%) = 226 \text{ (M}^+, 1), 169 (100), 151 (26).$

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(2RS,3R,4R)-(-)-4-(tert-Butyldimethylsiloxy)-2,3-dimethyl-cyclohexanone (8):

To a suspension of CuI (7.07 g, 37.1 mmol) in Et₂O (150 mL) at 0°C MeLi (46.4 mL, 1.6 M in Et₂O, 74.2 mmol) was added. The resulting colorless solution was stirred for 1 h at 0°C and cooled with a MeOH/dry-ice bath to -60° C internal temperature. To this solution via a transfer needle a solution of 7 (8.00 g, 35.3 mmol), precooled to 0°C, in Et₂O (50 mL) was added, so that the internal temperature did not rise above -55° C (ca. 30 min). The flask was placed into a second ice-bath, precooled to -35 °C, and the resulting yellow suspension was stirred for 100 min at an internal temperature of -35 to $-30\,^{\circ}\text{C}$. The flask was cooled to $-70\,^{\circ}\text{C}$ and MeI (17.6 mL, 40.1 g, 282 mmol) in N-methylpyrrolidinone (50 mL) was added rapidly (5 min) via a transfer needle to the intensely stirred suspension. During the addition, the internal temperature rose to -30° C and this temperature ($-30 \text{ to } -35^{\circ}$ C) was maintained for 3 h. During this time the yellow precipitate dissolved and a white precipitate formed. After the reaction was complete (TLC), sat. NH₄Cl/NH₃(1:1)-buffer (20 mL) was added, the suspension was warmed to r.t. and diluted with hexanes/EtOAc (1:1, 500 mL). Extraction with NH₄Cl/NH₃ (1:1, 2×200 mL), H₂O (2×50 mL) and brine (2 × 100 mL), drying (Na₂SO₄) and concentration in vacuo gave an oil that was chromatographed over silica gel (200 g; 10:1 hexanes/EtOAc) to give 8 (9:1 mixture of C-2-epimers according to GC) as a colorless oil. On storage the minor product slowly epimerized to the major; yield 6.41 g (71 %); bp (0.05 mbar, Kugelrohr heating bath) 65-70 °C, $[\alpha]_D^{20}$ (Major) -35.8 (c=1.06, CHCl₂).

IR (major, film): $v = 1940, 1920, 1850, 1840, 1705, 1500, 1245, 1100, 1060, 830, 765 \, \mathrm{cm}^{-1}.$

¹H NMR (major, CDCl₃/TMS): δ = 3.64 (m, 1 H), 2.31-2.46 (m, 2H), 2.03-2.16 (m, 2H), 1.72 (m,1H), 1.48 (m, 1H), 1.07 (d, 3H, J = 6.7 Hz), 1.06 (d, 3H, J = 6.7 Hz), 0.90 (s, 9H), 0.10 (s, 3H), 0.08 (s, 3H).

¹³C NMR (major, CDCl₃); δ = 211,6, 74.2, 48.3, 47.1, 38.4, 34.1, 25.8, 18.0, 17.3, 12.3, -4.2, -4.7.

MS: $m/z(\%) = 256 \text{ (M}^+, 1), 241 (5), 199 (100), 75 (63).$

(1R,2R,3R)-(-)-3-[3-(tert-Butyldimethylsiloxy)-1,2-dimethyl-6-oxocyclohexyl]propionic Acid Methyl Ester (9):

Compound **8** (9:1 epimeric mixture, 1.80 g, 70.02 mmol) was dissolved in toluene (20 mL) and (R)-(+)-1-phenylethylamine (1.09 mL, 8.42 mmol) was added. The solution was heated to reflux under azeotropic removal of $\rm H_2O$ for 3 days. The toluene was evaporated with an oil pump, methyl acrylate (6.30 mL, 6.02 g, 70.2 mmol) was added and the solution was heated for 3 days to 80 °C. The solution was evaporated with an oil pump, MeOH (5 mL) and 20 % HOAc (3 mL) were added and the solution was stirred at r.t. for 48 h. Et₂O (2 mL) was added and the organic phase was extracted with 2 % $\rm H_2SO_4$ (2 × 50 mL), sat. NaHCO₃ (2 × 20 mL), brine (2 × 20 mL) and dried (Na₂SO₄). Concentration in vacuo and chromatography on silica gel (hexanes/EtOAc 6:1) gave 9 as a colorless oil that contained less than 5 % (GC, ¹H NMR resonance of the singlet at 1.06 ppm) of the C-2-epimer; yield 1.86 g (77 %); [α]_D²⁰ –19.5 (c = 2.2, CHCl₃).

IR (film): v = 2920, 2850, 1725, 1695, 1450, 1430, 1245, 1060, 825, 775 cm⁻¹.

¹H NMR (CDCl₃, TMS): δ = 4.08 (m, 1 H), 3.62 (s, 3 H), 1.60–2.60 (m, 8 H), 0.98 (s, 3 H), 0.91 (d, 3 H, J = 7.2 Hz), 0.87 (s, 9 H), 0.06 (s, 3 H), 0.05 (s, 3 H).

¹³C NMR (CDCl₃): δ = 213.8, 174.1, 71.6, 51.5, 50.5, 44.5, 35.5, 33.4, 30.5, 29.2, 25.8, 20.1, 18.0, 12.6, -4.3, -4.8.

MS: $m/z(\%) = 342 \text{ (M}^+, 1), 327 \text{ (4), 311 (20), 285 (100)}.$

(1R,2R,3R)-(-)-3-[tert-Butyldimethylsiloxy)-1,2-dimethyl-6-oxocyclohexyl] Propionic Acid (10):

Compound 9 (1.80 g, 5.26 mmol) was dissolved in THF (20 mL), LiOH solution (0.5 M in $\rm H_2O$, 20 mL) was added and the solution was stirred at r.t. for 15 h. The solution was concentrated in vacuo and $\rm Et_2O$ (100 mL) and 2 % $\rm H_2SO_4$ (50 mL) were added. Extraction, washing of the organic phase with brine (2 × 20 mL), drying

 $(\mathrm{Na_2SO_4})$, concentration in vacuo and chromatography over a short column (30 g silica gel, $\mathrm{CH_2Cl_2/MeOH}\ 20:1$) gave 10 as colorless crystals. Compound 10 was recrystallized from hexanes, whereby the small contamination with the minor C-2-epimer remained in solution; yield: 1.58 g (91 %); mp 71–73 °C; $[\alpha]_D^{20}-23.3$ (c=5.6, $\mathrm{CCl_3}$).

IR (CCl₄): $\nu = 2920$, 2850, 1700, 1460, 1250, 1090, 1070, 830, 770 cm cm⁻¹.

¹H NMR (CDCl₃, TMS): $\delta = 3.75$ (m, 1 H), 1.60–2.60 (m, 8 H), 0.99 (s, 3 H), 0.91 (d, 3 H, J = 7.2 Hz), 0.87 (s, 9 H), 0.06 (s, 3 H), 0.05 (s, 3 H).

¹³C NMR (CDCl₃): δ = 214.1, 179.8, 71.6, 50.5, 44.5, 35.5, 33.5, 30.5, 29.2, 24.8, 20.1, 18.0, 12.5, -4.3, -4.8.

MS: $m/z(\%) = 328 \text{ (M}^+, 1)$, 313 (1), 271 (45), 179 (95), 75 (100).

(4aR,5R,6R)-(+)-6-(tert-Butyldimethylsiloxy)-4a,5-dimethyl-4,4a,5,6,7,8-hexahydro-3H-naphthalen-2-one (15):

Compounds 10 (2.10 g, 6.39 mmol) and 11 (2.90 g, 7.68 mmol) were dissolved in toluene (60 mL) and heated to 80 °C for 5 h. EtOH (3 mL) was added and the solution was refluxed for 15 h. Et₂O (200 mL) was added, the organic phase was extracted with brine (2 × 20 mL), dried (Na₂SO₄), concentrated in vacuo and filtered with Et₂O through a short silica gel column (10 g). After concentration the residue was chromatographed (150 g silica gel, hexanes/EtOAc 6:1) to give 15 as an oil that slowly crystallized. It was recrystallized from hexanes; yield 1.33 g (67 %); mp 30–35 °C; [α]_D²⁰ + 76.3 (c = 1.14, CHCl₃); lit. ^{8f}: mp 36–37 °C, [α]_D^{24.5} + 88.7 (c = 1.1, CHCl₃).

IR (film): v = 1665, 1620, 1255, 1075, 840, 780 cm⁻¹.

¹H NMR (CHCl₃, TMS): δ = 5.72 (d, 1 H, J = 1.0 Hz), 3.55 (dt, 1 H, J = 4.3, 10.2 Hz), 2.24–2.39 (m, 4 H), 1.99–2.04 (m, 2 H), 1.66–1.71 (m, 2 H), 1.47 (m, 1 H), 1.09 (s, 3 H), 0.96 (d, 3 H, 6.7 Hz), 0.86 (s, 9 H), 0.06 (s, 3 H), 0.05 (s, 3 H).

¹³C NMR (CHCl₃): δ = 199.5, 169.3, 124.1, 71.8, 50.0, 39.2, 35.9, 35.7, 33.4, 31.3, 25.8, 18.0, 17.3, 11.1, -4.0, -4.7.

MS: $m/z(\%) = 308 \text{ (M}^+, 1)$, 293 (6), 251 (100), 159 (55), 75 (82).

(3S,4aR,5R,6R)-(+)-6-(tert-Butyldimethylsiloxy)-3-(1-hydroxy-1-methylethyl)-4a,5-dimethyl-4,4a,5,6,7,8-hexahydro-3*H*-naphthalen-2-one (16):

Hexamethyldisilazane (6.77 mL), 5.18 g, 32.1 mmol) was dissolved in THF (60 mL), cooled to $-60\,^{\circ}$ C, BuLi (12.6 mL, 2.5 M in hexanes, 31.5 mmol) was added and the solution was stirred for 30 min. The solution was cooled to $-78\,^{\circ}$ C and 15 (1.94 g, 6.29 mmol) in THF (20 mL) was added slowly via a transfer needle (ca. 15 min). After stirring for 1 h ZnCl₂ (3.15 mL, 2.2 M in CH₂Cl₂, 6.93 mmol) was added. After 10 min, acetone (3.70 mL, 2.93 g, 50.4 mmol) was added and the reaction was quenched after 15 min by the addition of 5% tartraric acid (20 mL) and warming to r.t. Hexanes/Et₂O (1:1, 200 mL) was added, the organic phase was extracted with 5% tartraric acid (2 × 20 mL), brine (2 × 10 mL), dried (Na₂SO₄) and concentrated in vacuo. The residue was chromatographed (50 g silica gel, hexanes/EtOAc 3:2) to give 16 as a colorless oil; yield 2.08 g (90 %); [α]_D²⁰ + 74.3 (c = 1.1, CHCl₃).

IR (film): v = 3460, 2940, 2929, 2850, 1645, 1460, 1350, 1250, 1070, 830 cm^{-1} .

 1 H NMR (CDCl₃, TMS): δ = 5.65 (d, 1 H, J = 1.7 Hz), 5.10 (br s, 1 H), 3.50 (m, 1 H), 2,22–2.43 (m, 4 H), 1.95–2.00 (m, 2 H), 1.18–1.45 (m, 2 H), 1.15 (s, 6 H), 1.07 (s, 3 H), 0.93 (d, 3 H, J = 6.6 Hz), 0.83 (s, 9 H), 0.03 (s, 3 H), 0.02 (s, 3 H).

¹³C NMR (CDCl₃): δ = 203.0, 169.9, 124.6, 72.4, 71.5, 50.5, 50.4, 39.7, 38.9, 35.4, 30.8, 28.2, 25.7, 24.5, 17.9, 17.1, 10.9, -4.1, -4.8. MS: m/z(%) = 366 (M⁺, 1), 351 (7), 309 (68), 251 (67), 176 (87), 75 (100).

(+)-Isopetasol-(tert-butyldimethylsilyl) Ether (17):

Compound 16 (900 mg, 2.45 mmol) was dissolved in CH₂Cl₂ (15 mL), cooled to 0°C and Et₃N (4.27 mL, 3.10 g, 30.6 mmol) and methanesulfonyl chloride (1.14 mL, 1.68 g, 14.7 mmol) were added. After stirring for 2 h the white suspension was diluted with CH₂Cl₂

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(50 mL), extracted with H₂O (2 × 10 mL), 5% H₂SO₄ (2 × 10 mL), brine (2 × 10 mL) and dried (Na₂SO₄). The solution was filtered through a column of acidic alumina (WOELM, 20 g), concentrated in vacuo and chromatographed (50 g silica gel, hexanes/EtOAc 6:1) to give 17 as a colorless oil; yield: 638 mg (75%); [α]_D²⁰ + 29.9 (c = 5.5, CHCl₃).

IR (film): v = 2940, 2920, 2840, 1660, 1620, 1600, 1300, 1070, 830, 770 cm⁻¹.

¹H NMR (CDCl₃, TMS): δ = 5.68 (d, 1 H, J = 1.5 Hz), 3.47 (m, 1 H), 2.84 (d, 1 H, J = 13.7 Hz), 1.94–2.32 (m, 3 H), 2.03 (d, 3 H, 1.8 Hz), 1.79 (s, 3 H), 1.34–1.45 (m, 2 H), 0.96 (d, 3 H, J = 7.0 Hz), 0.92 (s, 3 H), 0.84 (s, 9 H), 0.03 (s, 3 H), 0.02 (s, 3 H).

¹³C NMR (CDCl₃): δ = 191.8, 166.5, 142.7, 127.4, 126.2, 71.9, 49.3, 42.0, 41.3, 35.7, 30.4, 25.8, 22.5, 22.0, 18.0, 17.2, 11.2, -4.1, -4.8. MS: m/z(%) = 348 (M⁺, 34), 291 (100), 216 (38), 161 (71), 75 (50).

(+)-Isopetasol (18):

Compound 17 (420 mg, 1.20 mmol) was dissolved in MeCN (20 mL), then ca. 50 % HF (2 mL) was added and the solution was stirred at r.t. for 2 h. Et₂O (100 mL) was added and the organic phase was extracted with sat. NaHCO₃ (2 × 20 mL), brine (2 × 10 mL), dried (Na₂SO₄) and concentrated in vacuo. Chromatography (20 g silica gel, hexanes/EtOAc 1:1) gave 18 as colorless needles. It was recrystallized from Et₂O/hexanes; yield 232 mg (83%); mp 116–118 °C; [α]_D²⁰ +108.3 (c = 1.3, CHCl₃); lit. ¹⁷ mp 122–125 °C; [α]_D +116.6 (c = 1.04, CHCl₃).

IR (CCl₄): v = 3400, 2960, 2930, 2870, 1655, 1620, 1440, 1300, 1225, 1040, 885 cm⁻¹.

¹H NMR (CDCl₃, TMS): δ = 5.70 (d, 1 H, J = 1.5 Hz), 3.51 (m, 1 H), 2.85 (d, 1 H, J = 13.8 Hz), 2.05–2.37 (m, 3 H), 2.02 (d, 3 H, J = 1.7 Hz), 1.79 (d, 3 H, J = 1.0 Hz), 1.30–1.45 (m, 2 H), 1.05 (d, 3 H, J = 6.6 Hz), 0.92 (s, 3 H).

 13 C NMR (CDCl₃): $\delta = 191.8, 166.3, 143.1, 127.2, 126.3, 71.0, 49.0, 42.0, 41.1, 35.2, 30.4, 22.5, 22.0, 17.2, 10.7.$

MS: m/z(%) = 234 (M⁺, 87), 201 (33), 161 (100).

(+)-Isopetasin (2):

Angelic acid (256 mg, 2.56 mmol) was dissolved in toluene (4 mL) and 2,4,6-trichlorobenzoyl chloride (0.400 mL, 625 mg, 2.56 mmol) and Et₃N (0.357 mL, 259 mg, 2.56 mmol) were added. The resulting white suspension was stirred at r.t. for 2 h and 18 (300 mg, 1.28 mmol) dissolved in toluene (4 mL) was added. After heating to 80 °C for 24 h Et₂O (20 mL) was added, the crude suspension filtered through silica gel (5 g, Et₂O) and the solution was concentrated in vacuo. The residue was chromatographed (40 g, silica gel, hexanes/EtOAc 10:1) to give 2 as colorless crystals that were recrystallized from Et₂O/hexanes; yield: 336 mg (83 %); mp 82–84 °C, $[\alpha]_D^{20} + 23.8$ (c = 0.8, CHCl₃); lit.¹⁷ mp 89–91 °C, $[\alpha]_D^{20} + 28.6$ (no concentration stated, CHCl₃).

IR (CCl₄): v = 2970, 2930, 1705, 1655, 1230, 1160, 1040 cm⁻¹.

¹H NMR (CDCl₃, TMS): $\delta = 6.08$ (m, 1 H), 5.78 (d, 1 H, J = 1.8 Hz), 4.93 (m, 1 H), 2.94 (d, 1 H, J = 13.8 Hz), 2.51 (m, 1 H), 2.36 (m, 1 H), 2.23 (m, 1 H), 2.20 (d, 1 H, J = 13.5 Hz), 2.10 (d, 3 H, J = 1.8 Hz), 2.00 (dd, 3 H, J = 7.3, 1.5 Hz), 1.90 (s, 3 H), 1.86 (d, 3 H, J = 1.2 Hz), 1.70 (m, 1 H), 1.51 (m, 1 H), 1.06 (s, 3 H), 1.01 (d, 3 H, J = 6.7 Hz).

 $^{13}\mathrm{C}$ NMR (CDCl₃): $\delta = 191.4,\ 167.5,\ 165.0,\ 143.2,\ 137.8,\ 127.8,\ 127.0,\ 126.6,\ 73.1,\ 46.0,\ 42.1,\ 41.0,\ 31.6,\ 30.0,\ 22.4,\ 22.0,\ 20.4,\ 17.0,\ 15.6,\ 10.7.$

MS: m/z (%) = 316 (83), 216 (54), 201 (31), 161 (100), 147 (12), 83 (22), 55 (44).

(3S,4aR,5R,6R)-(+)-6-Hydroxy-3-(1-hydroxy-1-methylethyl)-4a,5-dimethyl-4,4a,5,6,7,8-hexahydro-3H-naphthalen-2-one (20):

Compound 16 (500 mg, 1.36 mmol) was dissolved in MeCN (20 mL), 50 % HF (2 mL) was added and the solution was stirred at r.t. for 2 h. Et₂O (100 mL) was added and the organic phase was extracted with sat. NaHCO₃ (2×20 mL), brine (2×10 mL), dried (Na₂SO₄) and concentrated in vacuo. Chromatography (20 g silica gel, hexanes/EtOAc 1:2) gave 20. It was recrystallized from Et₂O/

hexanes to give **20** as colorless crystals; yield: 323 mg (94%); mp 104-106 °C; [α]_D²⁰ + 100.0 (c=1.3, CHCl₃).

IR (CCl₄): $\nu = 3400$, 2960, 2920, 2860, 1650, 1640, 1375, 1230, $1030~{\rm cm}^{-1}$.

 $^{1}\mathrm{H}$ NMR (CDCl $_{3}$, TMS): $\delta=5.71$ (d, 1 H, J=1.8 Hz), 5.10 (br s, 1 H), 3.58 (m, 1 H), 2.31–2.48 (m, 4 H), 2.00–2.15 (m, 2 H), 1.69 (br s, 1 H), 1.25–1.49 (m, 2 H), 1.19 (s, 6 H), 1.11 (s, 3 H), 1.06 (d, 3 H, J=6.7 Hz).

 $^{13}\mathrm{C}$ NMR (CDCl₃): $\delta = 203.1,\,169.4,\,124.9,\,72.5,\,70.8,\,50.6,\,50.1,\,39.8,\,38.8,\,35.0,\,30.8,\,28.2,\,24.6,\,17.1,\,10.5.$

MS: m/z(%) = 252 (M⁺, 1), 237 (6), 234 (7), 194 (57), 176 (32), 121 (100).

2-Methylbut-2-enoic Acid (1R,2R,7S,8aR)-7-(1-Hydroxy-1-methylethyl)-1,8a-dimethyl-6-oxo-1,2,3,4,6,7,8,8a-octahydronaphthalen-2-vl Ester (21):

Angelic acid (159 mg, 1.59 mmol) was dissolved in toluene (4 mL) and 2,4,6-trichlorobenzoyl chloride (0.248 mL, 387 mg, 1.59 mmol) and Et₃N (0.221 mL, 160 mg, 1.59 mmol) were added. The resulting white suspension was stirred at r.t. for 2 h and **20** (200 mg, 0.793 mmol) dissolved in toluene (4 mL) was added. After heating to 80 °C for 24 h Et₂O (20 mL) was added, the crude suspension filtered over silica gel (5 g, Et₂O) and the solution was concentrated in vacuo. The residue was chromatographed (40 g silica gel, hexanes/ EtOAc 6:1) to give isopetasin **2** (50 mg, 20 %) starting material **20** (50 mg) and the product **21** as colorless crystals, that were recrystallized from Et₂O/hexanes; yield 104 mg (39 %, 52 % based on recovered **20**); mp 83-85 °C, [α]₂⁰ -42.1 (c = 1.6, CHCl₃).

IR (CCl₄): v = 3450, 2960, 2940, 2870, 1700, 1645, 1450, 1375, 1230, 1155, 990 cm⁻¹.

¹H NMR (CDCl₃, TMS): $\delta = 6.05$ (m, 1 H), 5.71 (d, 1 H, J = 1.8 Hz), 5.05 (br s, 1 H), 4.90 (m, 1 H), 2.33–2.56 (m, 4 H), 2.00–2.23 (m, 2 H), 1.95 (d, 3 H), 1.85 (s, 3 H), 1.20–1.64 (m, 2 H), 1.19 (s, 3 H), 1.17 (s, 3 H), 0.96 (d, 3 H, J = 6.7 Hz).

 $^{13}\mathrm{C}\,\mathrm{NMR}$ (CDCl₃): $\delta=203.2,\ 168.4,\ 167.5,\ 138.2,\ 127.8,\ 125.0,\ 72.7,\ 72.3,\ 50.8,\ 50.6,\ 40.0,\ 38.8,\ 37.4,\ 30.5,\ 28.2,\ 24.5,\ 20.5,\ 17.0,\ 15.7,\ 10.5.$

MS: m/z% = 334 (M⁺, 1), 319 (4), 316 (7), 219 (6), 176 (100), 121 (80).

(+)-Petasin (1):

Compound 21 (120 mg, 0.359 mmol) was dissolved in CH₂Cl₂ (5 mL), cooled to $-78\,^{\circ}$ C and diisopropylethylamine (0.188 mL), 140 mg, 1.08 mmol) and trifluoromethanesulfonic anhydride (0.091 mL, 152 mg, 0.539 mmol) were added. After stirring for 60 min sat. NH₄Cl (2 mL) was added and the suspension was warmed to r.t. CH₂Cl₂ (50 mL) was added and the organic phase was extracted with sat. NH₄Cl (2 × 10 mL), brine (2 × 5 mL), dried (Na₂SO₄) and concentrated in vacuo. Chromatography (20 g silica gel, hexanes/EtOAc 10:1) gave isopetasin 2 (19 mg, 17%), 21 (36 mg) and product 1 as a colorless oil; yield 55 mg (48%, 69% based on recovered 21); $[\alpha]_D^{20} + 34.1$ (c = 0.7, CHCl₃); lit.²: $[\alpha]_D = 39.6$ (no concentration, temperature and solvent stated). IR (film): v = 2940, 2870, 1700, 1665, 1625, 1455, 1380, 1230, 1160,

IR (film): $\nu = 2940, 2870, 1700, 1665, 1625, 1455, 1380, 1230, 1160 1045, 985, 890, 850 cm⁻¹.$

¹H NMR (CDCl₃, TMS): $\delta = 6.08$ (m, 1 H), 5.79 (d, 1 H, J = 2.2 Hz), 5.00 (s, 1 H), 4.96 (m, 1 H), 4.83 (s, 1 H), 3.13 (dd, 1 H, J = 14.8, 4.4 Hz), 2.54 (m, 1 H), 2.39 (m, 1 H), 2.26 (m, 1 H), 2.04 (dd, 1 H, 13.2, 4.4 Hz), 2.00 (dd, 3 H, J = 7.3, 1.5 Hz), 1.94 (d, 1 H, J = 13.2 Hz), 1.89 (s, 3 H), 1.86 (s, 3 H), 1.69 (m, 1 H), 1.52 (m, 1 H), 1.25 (s, 3 H), 0.98 (d, 3 H, J = 6.6 Hz).

 $^{13}\mathrm{C}$ NMR (CDCl₃): $\delta = 198.5,\ 167.6,\ 166.8,\ 143.3,\ 138.1,\ 127.9,\ 124.6,\ 114.4,\ 72.9,\ 50.3,\ 47.2,\ 41.6,\ 40.0,\ 31.6,\ 30.6,\ 20.6,\ 20.0,\ 17.1,\ 15.7,\ 10.5.$

MS: $m/z(\%) = 316 \,(M^+, 27), 216 \,(22), 161 \,(15), 148 \,(100), 105 \,(20).$

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