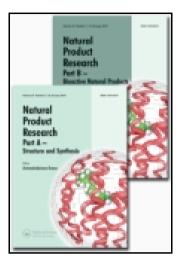
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Chemical constituents and antibacterial activity of Melastoma malabathricum L.

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Chemical constituents and antibacterial activity of *Melastoma* malabathricum L.

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The aqueous methanolic extracts of *Melastoma malabathricum* L. exhibited antibacterial activity when assayed against seven microorganisms by the agar diffusion method. Solvent fractionation afforded active chloroform and ethyl acetate fractions from the leaves and the flowers, respectively. A phytochemical study resulted in the identification of ursolic acid (1), 2α -hydroxyursolic acid (2), asiatic acid (3), β -sitosterol 3-O- β -D-glucopyranoside (4) and the glycolipid glycerol 1,2-dilinolenyl-3-O- β -D-galactopyanoside (5) from the chloroform fraction. Kaempferol (6), kaempferol 3-O- α -L-rhamnopyranoside (7), kaempferol 3-O- β -D-glucopyranoside (8), kaempferol 3-O- β -D-galactopyranoside (9), kaempferol 3-O-(2",6"-di-O-E*p*-coumaryl)- β -*D*-galactopyranoside (10), quercetin (11) and ellagic acid (12) were found in the ethyl acetate fraction. The structures of these compounds were determined by chemical and spectral analyses. Compounds 1-4, the flavonols (6 and 11) and ellagic acid (12) were found to be active against some of the tested microorganisms, while the kaempferol 3-O-glycosides (7-9) did not show any activity, indicating the role of the free 3-OH for antibacterial activity. Addition of p-coumaryl groups results in mild activity for 10 against Staphylococcus aureus and Bacillus cereus. Compounds 2-5, 7 and 9-12 are reported for the first time from *M. malabathricum*. Compound **10** is rare, being reported only once before from a plant, without assignment of the double bond geometry in the *p*-coumaryl moiety.

Keywords: *Melastoma malabathricum*; Sendudok; pentacyclic triterpenoids; acylated kaempferol glycosides; antibacterial activity

1. Introduction

Melastoma malabathricum L. (Melastomataceae, Malay name: Sendudok) is a small shrub widely used in the traditional medicine of Malaysia and Indonesia for treating diarrhoea, dysentery, leucorrhoea, ulcers and wounds (Burkill, 1966; Perry, 1980). The methanolic extract of the aerial parts was reported to have a moderate antiviral activity against HSV-1, a significant activity against Poliovirus and a cytotoxic effect

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 $(IC50 < 25 \,\mu g \,ml^{-1})$ on murine cell lines (Lohezic-Le Devehat, Bakhtiar, Bezivin, Amoros, & Boustie, 2002). The ethanolic extract of the stem bark and leaves was found to be potentially antinociceptive, acting at both the peripheral and central levels in mice (Sulaiman, Somchit, Israf, Ahmad, & Moina, 2004). The flowers were shown to contain antioxidant and cytotoxic flavonoids (Susanti et al., 2007). Previous phytochemical studies have revealed the presence of malvidin 3,5-diglucoside (Lowry, 1976), 4'-methylpeonidin 7- $O-\beta$ -D-glucoside (Mohandoss & Ravindran, 1993), kaempferol (Das & Kotoky, 1988; Mohandoss & Ravindran, 1993; Susanti et al., 2007), kaempferol 3-O- β -D-xyloside (Mohandoss & Ravindran, 1993), kaempferol 3-O-glucoside, kaempferol 3-O-(2",6"-di-O-p-trans-coumaroyl) glucoside and naringenin (Susanti et al., 2007), p-hydroxybenzoic acid and gallic acid (Das & Kotoky, 1988) in the flowers, β -sitosterol and melastomic acid in the roots (Manzoor-I-Khuda, S.A. Chowdhury, Reza, & A.K. Chowdhury, 1981), quercetin $3-O-\alpha-L$ -rhamnosyl- $(1\rightarrow 2)-\beta-D$ -galactoside (Dinda & Saha, 1988), 1-octyl docosanoate, 11-methyl-1-tritriacontanol (Dinda & Saha, 1986) in the aerial parts, 32-methyl-1-tritriacontanol, β -sitosterol, ursolic acid (Das & Kotoky, 1988) and several tannins (Yoshida, Nakata, Hosotani, Nitta, & Okuda, 1992a, 1992b) in the dried leaves. The extensive use of *M. malabathricum* in traditional medicine led us to carry out further phytochemical investigations of this plant, guided by antibacterial activity assay. In this article we report the isolation of 12 compounds from M. malabathricum, some of them possessing antibacterial activity. Nine of the isolated compounds are reported for the first time in this plant.

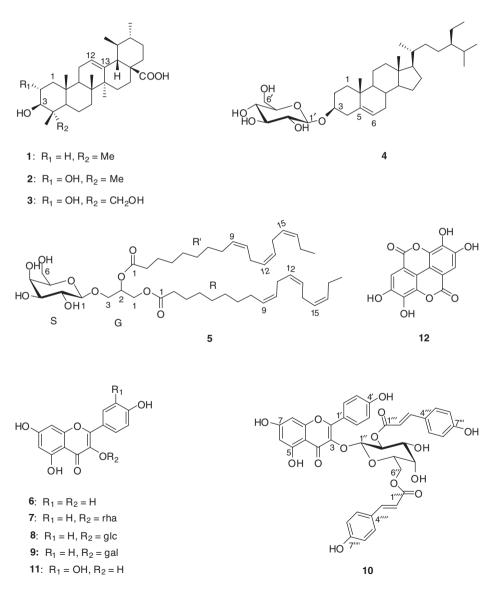
2. Results and discussion

2.1. Extraction and isolation

The aqueous methanolic extracts of the fresh leaves and flowers of *M. malabathricum* L. were evaluated for antibacterial activity by the agar diffusion method. Guided by this activity, solvent fractionation and isolation were carried out by silica gel and Sephadex column chromatography using different solvent systems. The isolated structures were confirmed by comparison of their spectral data (UV, IR, MS, ¹H- and ¹³C-NMR) with literature data (Collins, Pilotti, & Wallis, 1992; Furuya, Orihara, & Hayashi, 1987; Kriwacki & Pitner, 1989; Li, Elsohly, Hufford, & Clark, 1999; Mabry, Markham, & Thomas, 1970; Markham, Ternai, Stanley, Geiger, & Mabry, 1978; Sakakaibara, Kaiya, Fukuda, & Ohki, 1983; Seo, Tomita, & Tori, 1981; Ternai & Markham, 1976; Yamagishi et al., 1988), 2D NMR, chromatographic methods (GC–MS, TLC, PC) and chemical analyses.

The chloroform fraction (fr-2) of the aqueous methanolic extract of the leaves (ext-1) gave ursolic acid (1), 2α -hydroxyursolic acid (2), asiatic acid (3), β -sitosterol 3-O- β -D-glucopyranoside (4) and glycerol 1, 2-dilinolenyl-3-O- β -D-galactopyranoside (5). The ethyl acetate fraction (fr-3) of the aqueous methanolic extract of the flowers (ext-2) afforded kaempferol (6), kaempferol 3-O- α -L-rhamnopyranoside (7), kaempferol 3-O- β -D-glucopyranoside (8), kaempferol 3-O- β -D-galactopyranoside (9), kaempferol 3-O-(2",6"-di-O-E-p-coumaryl)- β -D-galactopyranoside (10), querce-tin (11) and ellagic acid (12). The triterpenoids (2–4), glycolipid (5), flavonoids (7, 9–11) and ellagic acid (12) are reported for the first time from M. malabathricum.

Compound (10) is uncommon, being reported only once before in a related species (Chou & Liao, 1982), but the configuration of the olefinic function of the *p*-coumaryl moiety was not specified (CAS Registry Number: 84461-47-2). In this study the olefinic protons of the two *p*-coumaryl groups were observed as two pairs of doublets at 7.69 and 6.38 ppm (3''', 2'''), and 7.39 and 6.05 ppm (3'''', 2'''') with a coupling constant (*J*) 15.9 Hz (Figure S1). The large value of *J* indicated the *E*-configuration of the *p*-coumaryl moiety (Silverstein, Bassler, & Morrill, 1991).



Scheme 1. Compounds 1 to 12.

2.2. Antibacterial activity

The aqueous methanolic extracts of the leaves (ext-1) and flowers (ext-2) of *M. malabathricum* exhibited antibacterial activity when tested by the agar diffusion method (Table 1). From ext-1 an active precipitate (fr-1) was obtained which was extracted with chloroform to give fraction fr-2. Five compounds (1-5) were isolated from fr-2 of which 1-4 are triterpenoids active against some of the tested microorganisms, while 5 is an inactive glycolipid. Compound 3, at concentrations ten times those of refampicin, showed activity against *Bacillus cereus* comparable to that shown by this reference antibiotic (Table 1). ext-2 was extracted with ethyl acetate to give fraction fr-3 from which seven compounds (6-12) were obtained. Among these are two flavonols (6 and 11) and ellagic acid (12), which are active compounds, and three kaempferol 3-O-glycosides (7-9), which did not show any activity indicating the role of the free 3-OH for antibacterial activity. The seventh compound is the diacylated kaempferol 3-O-glycoside (10). It appears that the addition of two p-coumaryl groups results in a mild activity for 10 against Staphylococcus aureus and B. cereus as shown in Table 1. The presence of p-coumaryl group has been seen before to dramatically increase the dermatitis activity of astragalin (Loggia, Negro, Bianchi, Romussi, & Tubaro, 1989). The results of extracts and pure compounds which possess partial activity are reported within parentheses in Table 1.

3. Experimental

3.1. General

Melting points were determined using a Gallenkamp apparatus. FTIR spectra were recorded on a Perkin–Elmer System 2000 spectrometer. EI–MS and GC–MS were recorded on a Hewlett Packard 5989 A. NMR spectra were recorded in deuterated methanol (Sigma Chem. Co.), chloroform (Aldrich) or pyridine (Merck), by a Bruker Avance 300 or a Bruker Avance 400 spectrometer operated at 300 and 400 MHz for ¹H NMR and 75 and 100 MHz for ¹³C NMR, respectively. UV spectra of the flavonoids were measured in methanol, using a Jasco-V-530 UV/Vis spectro-photometer as described in Mabry et al. (1970). TLC was carried out on pre-coated silica gel 60 F_{250} plates (Merck). Column chromatography was carried out using silica gel (Riedel-deHaen 230-400 mesh, ASTM) and Sephadex LH-20 (Sigma-Aldrich). Paper chromatography was carried on Whatman No. 54. All solid compounds isolated were purified by recrystallisation in acetone (1, 2, 3 and 4), methanol (6, 8, 9 and 12) or acetone-chloroform (7 and 10). Compound 5 was purified by repeated CC.

3.2. Plant material

M. malabathricum was collected from Penang Island, Malaysia. The plant was identified by Professor Lai-Keng Chan, School of Biological Sciences, Universiti Sains Malaysia, where a voucher specimen (No. 10613) was deposited.

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Table 1. Antibacterial activity of extracts and isolated compounds from *M. malabathricum*.

K. pneumoniae $\begin{array}{c} 13.0\\111.0\\09.0\\08.0\\09.0\\-\\-\\111.0\\09.0\\09.0\\09.0\\\end{array}$ 09.0 $\begin{array}{c} 111.5\\ 10.5\\ 09.0\\ 11.6\\ 11.6\\ 10.0\\ 09.0\\ 09.0\\ 111.0\\ 09.0\\ 00.0\\ 10.0\\ 00.5\end{array}$ S. typhi 15.5 13.0 12.0 11.0 13.0 10.0 I Zone of inhibition (diameter, mm, average of duplicate) P. aeruginosa 0.60 $\begin{array}{c} 112.0\\ 111.5\\ 09.5\\ 09.5\\ 112.5\\ 111.5\\ 08.0\\ 08.0\\ \end{array}$ $\begin{array}{c} - \\ - \\ - \\ 11.5 \\ 10.5 \\ - \\ - \\ - \\ - \end{array}$ 10.5 09.0 09.0 09.0 coliЩ I L T 14.0 (27.0) 11.0 (25.0) 08.0 (22.0) 07.0 (19.0) $\begin{array}{c} 10.0 & (24.5) \\ 08.0 & (21.5) \\ 07.0 & (18.5) \end{array}$ -(16.0)-(12.0)-(11.0)cereus $\begin{array}{c} 07.0\\ 07.0\\ 07.0\\ 07.5\\ 07.5\\ 07.5\\ 07.5\\ 07.5\\ 113.5\\ 112.0\\ 110.0\\ 10.5\end{array}$ B. I Ι subtilis $\begin{array}{c} 14.5\\ 111.5\\ 111.0\\ 09.5\\ 09.5\\ 112.5\\ 112.5\\ 08.5\\ 09.5\\ 09.0\\ 08.0\\ 0$ 3.02.0 $1.0 \\ 0.0$ I B. $\begin{array}{c} 15.0 \ (23.5)^{a} \\ 12.5 \ (18.0) \\ 10.0 \ (13.5) \\ 09.0 \ (11.0) \end{array}$ $\begin{array}{c} 10.0 & (25.5) \\ 09.0 & (20.0) \\ 08.0 & (18.0) \\ 07.0 & 07.0 \\ 112.0 & (19.5) \\ 110.5 & 08.0 \\ 08.5 & 08.5 \\ 09.5 & 09.5 \\ 07.5 & 111.0 \\ 111.0 \\ 111.0 \\ 111.0 \\ 0.90 \\ 0.90 \\ 08.0 \\ 08.0 \\ 07.0 \end{array}$ S. aureus I Conc $(mg mL^{-1})$ $\begin{array}{c} 0.5 \\ 0.25 \\ 0.25 \\ 0.5 \\ 0.25 \\ 0.25 \\ 0.25 \\ 0.25 \\ 0.25 \\ 0.25 \end{array}$ Sample ext-1 fr-2 fr-1 2 3 4

(continued)

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Table 1. Continued.

			Ζ	one of inhibitio	n (diameter, mn	Zone of inhibition (diameter, mm, average of duplicate)	cate)	
Sample	Conc $(mg mL^{-1})$	S. aureus	B. subtilis	B. cereus	$E. \ coli$	P. aeruginosa	S. typhi	K. pneumoniae
ext-2	20	15.5 (28.5)	16.5 (24.5)	13.0	19.5	19.5 (24.5)	18.0 (25.0)	14.0
	10		15.0 (22.0)	0.00	18.0		14.0 (20.5)	12.5
	5		11.5 (18.5)	07.0	15.5	16.0(19.5)	12.0 (19.5)	11.5
	2.5	09.5 (15.5)	10.0(14.5)	Ι	14.0	12.5	11.0(16.0)	10.0
fr-3	20	16.0(31.5)	18.0(30.0)	11.0 (22.0)	18.5	17.5 (25.5)	-(29.5)	16.5 (36.0)
	10		16.0(27.0)	10.0(20.0)	15.5		-(25.0)	
	5		15.0 (23.5)	08.0(18.0)	14.0	14.0(20.5)	-(22.0)	
	2.5			07.0(14.5)	12.5	13.0(18.5)	-(19.0)	13.0(24.0)
6	2	13.0	I	12.0	11.0	I	I	
	1	12.5	Ι	12.0	Ι	Ι	Ι	Ι
	0.5	11.0	Ι	10.0	Ι	Ι	Ι	Ι
	0.25	10.0	I	0.00	Ι	Ι	I	Ι
10	7	09.0(11.0)	I	11.0	I	I	I	6
	1		I	10.0	I	Ι	Ι	8
	0.5	07.5(10.0)	Ι	09.5	I	Ι	Ι	Ι
	0.25		I	08.5	Ι	Ι	I	Ι
11	2		18.0	13.0	16.0	14.0	10.5	12.5
	-	11.0(19.0)	16.0	12.0	14.0	12.0	10.0	10.5
	0.5	09.0(16.0)	13.0	10.0	13.0	10.0	0.00	09.5
	0.25	08.0(14.0)	10.0	08.0	11.0	08.0	Ι	0.00
12	2	15.0 (29.5)	25.5	-(21.5)	Ι	22.5	-(31.0)	-(36.0)
	1	15.0(29.0)	24.5	-(20.0)	Ι	21.0	-(29.0)	-(32.0)
	0.5	14.0(28.0)	22.0	-(19.0)	Í	19.5	-(27.5)	-(28.0)
	0.25	13.0 (25.5)	20.0	-(18.0)	Ι	17.5	-(21.5)	-(24.0)
Ref^{c}	0.2		16.0	14.0	15.0	10.0	14.0	14.0
	0.1	28.5	13.5	13.0	12.0	08.5	12.0	12.0
	0.05	27.5	11.5	12.0	11.0	07.5	10.0	10.0
	0.025	26.0	09.5	11.0	10.0	-(07.5)	0.00	0.60
$\operatorname{Con}^{\mathrm{d}}$		I	-(0.00) -	I	-(10.0)	-(07.5)	-(0.00)	-(07.0)
Notes: ^a V _i	Notes: ^a Values within parentheses		tivity. ^b -, no activ	vity. ^c Ref, refere	nce antibiotic (1	for partial activity. ^b -, no activity. ^c Ref, reference antibiotic (refampicin). ^d Con, control (DMSO), partial activity	control (DMSO)), partial activity.

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3.3. Extraction and isolation

Fresh leaves (3 kg) were homogenised in MeOH-H₂O (4:1) at room temperature. The slurry was filtered and the filtrate was concentrated under reduced pressure at a temperature $<40^{\circ}$ C to 1/10 of its original volume, using a rotary evaporator, to give the methanolic extract (ext-1) (193.5 g). When chilled and centrifuged, (ext-1) gave the precipitate (fr-1) (40.2 g). The precipitate (fr-1) was defatted with petroleum ether and extracted with chloroform $(300 \text{ mL} \times 2 \text{ and } 200 \text{ mL} \times 2)$. The chloroform extract (fr-2) (16.1 g) was fractionated by silica gel column chromatography (200 g), eluting chloroform–acetone–methanol (18:6:1), chloroform-acetone-methanol with (10:9:1) and acetone successively. The eluent was collected in 25 mL fractions. Fractions 15-19, which eluted with chloroform-acetone-methanol (18:6:1), yielded ursolic acid (1) (405 mg). Fractions 25–30 yielded 2α -hydroxyursolic acid (2) (96 mg). Fractions 46–50, which eluted with chloroform–acetone–methanol (10:9:1), yielded asiatic acid (3) (42.5 mg) and glycerol 1, 2-dilinolenyl-3-O- β -D-galactopyranoside (5) (1.35 g). Fractions 68–75, which eluted with acetone, yielded β -sitosterol 3-O- β -Dglucopyranoside (4) (111 mg).

The purple-magenta flowers (250 g) were defatted with petroleum ether and extracted with MeOH–H₂O (9:1) at room temperature. The solvent was removed under reduced pressure at a temperature <40°C to give a solid (ext-2) (13.0 g) which was extracted with ethyl acetate. The ethyl acetate extract (fr-3) (1.0 g) was fractionated on a Sephadex LH-20 column (20 g) using methanol. The eluent was collected in 20 ml fractions. Fractions 7–9 were chromatographed on a silica gel column (60 g) using dichloromethane–acetone–methanol (10:10:2), yielding kaempferol (6) (21 mg), kaempferol 3-*O*- α -*L*-rhamnopyranoside (7) (37 mg), kaempferol 3-*O*- β -*D*-glucopyranoside (8) (52 mg) and kaempferol 3-*O*- β -*D*-galactopyranoside (9) (8 mg). Fractions 10–12 precipitated ellagic acid (12) (14 mg) on concentration, while kaempferol 3-*O*-(2", β "-di-*O*-*E*-*p*-coumaryl)- β -*D*-galactopyranoside (10) (41 mg) was recovered from the resulting solution. Fractions 13–15, on passing through a Sephadex LH-20 column eluting with methanol, yielded quercetin (11) (5 mg).

3.4. Spectral data

Kaempferol 3-O-(2",6"-di-O-E-p-coumaryl)- β -D-galactopyranoside (**10**): Pale yellow powder, m.p. 194°C (dec) (CH₃COCH₃–CHCl₃); IR (KBr) ν_{max} : 3391, 1696, 1654, 1630, 1604, 1589, 1513, 1444, 1359, 1259, 1204, 1171, 1080, 982, 830 cm⁻¹; EI-MS m/z (rel. int.): 432 (0.2), 342 (0.7), 311 (2.4), 286 (42.0), 258 (4.6), 257 (5.3), 229 (6.6), 213 (6.2), 164 (35.8), 147 (41.2), 121 (28.3), 91 (19.5), 65 (12.4) and 44 (100); UV λ_{max} nm (MeOH): 268, 299, 313, 353, 366, (MeOH + NaOMe): 274, 310, 366, (MeOH + AlCl₃): 276, 315, 398, (MeOH + AlCl₃/HCl): 278, 317, 392, (MeOH + NaOAc): 275, 299, 314, 370 and (MeOH + NaOAc/H₃BO₃): 268, 301, 317, 365; ¹H NMR (methanol- d_4 , 400 MHz) δ (ppm): 6.05 (d, 1.9, 1H, H-6), 6.21 (d, 1.9, 1H, H-8), 7.93 (d, 8.8, 2H, H-2',6'), 6.86 (d, 8.8, 2H, H-3',5'), 5.50 (d, 8.0, 1H, H-1"), 5.37 (dd, 1.6, 8.0, 1H, H-2"), 3.78-3.84 (m, 1H, H-3"), 3.88 (d, 2.9, 1H, H-4"), 3.78-3.84 (m, 1H, H-5"), 4.37 (dd, 3.2, 11.5, 1H, H-6"), 4.19 (dd, 4.0, 11.5, 1H, H-6"), 6.38 (d, 15.9, 1H, H-2"'), 7.69 (d, 15.9, 1H, H-3"'), 7.39 (d, 15.9, 1H, H-3"'), 7.26 (d, 8.6, 2H, H-5"", 9"") and 6.78 (d, 8.6, 2H, H-6"", 8"'); ¹³C-NMR (methanol- d_4 , 75 MHz)

δ (ppm): 158.4 (C-2), 134.7 (C-3), 179.2 (C-4), 162.9 (C-5), 100.4 (C-6), 167.0 (C-7), 95.0 (C-8), 158.3 (C-9), 105.3 (C-10), 122.7 (C-1'), 132.1 (C-2',6'), 116.3 (C-3',5'), 161.5 (C-4'), 101.1 (C-1''), 74.0 (C-2''), 73.2 9 C-3''), 70.6 (C-4''), 75.0 (C-5''), 64.2 (C-6''), 168.8 (C-1'''), 115.2 (C-2'''), 147.0 (C-3'''), 127.2 (C-4'''), 131.2 (C-5''',9''), 116.8 (C-6''',8'''), 161.3 (C-7'''), 168.8 (C-1'''), 114.5 (C-2'''), 146.7 (C-3'''), 127.0 (C-4'''), 131.2 (C-5''',9'''), 116.8 (C-6''',8''') and 161.3 (C-7''') (see also Supplementary Figures S1–S6).

3.5. Acid hydrolysis of compounds 4 and 5

Four mg of 4 and 10 mg of 5 were separately placed in a 50 mL round bottom flask. Ten millilitres of 2 M HCl in methanol were added and the mixture was refluxed on a water bath (80°C) for three hours, and then cooled and diluted with 5 mL of water. The aglycones were extracted with chloroform (5 mL \times 3) while the aqueous layer gave the monosaccharides after evaporation under reduced pressure. The sugars were identified by comparison with authentic monosaccharides on PC using solvent systems butanol–acetic acid–water (4:1:5, top layer) and butanol–toluene–pyridine–water (5:1:3:3), and silica gel TLC, using butanol–acetic acid–diethyl ether–water (9:6:3:1).

3.6. Acid hydrolysis of the flavonoids 7-10

Three milligrams of each compound were dissolved in 150 μ L MeOH and 150 μ L 2 M H₂SO₄ was added. After letting it stand for 20 min at 100°C, the MeOH was evaporated off and 1 mL of water was added. The aqueous layer was extracted 3 times with 300 μ L Et₂O to afford the aglycone, while the aqueous layer was neutralised with CaCO₃, collected using a Pasteur pipette and concentrated to give the sugars. The sugars were identified by comparison with authentic monosaccharides on silica gel TLC, using the solvent systems *n*-butanol–acetic acid–diethyl ether–water (9:6:3:1) and ethyl acetate–acetic acid–methanol–water (13:4:3:3).

3.7. Trans-esterification of compound 5 and 10

Three to nine milligrams of each compound were placed in a 5 mL round bottom flask. 0.5 mL of 2 M methanolic sodium hydroxide was added and the mixture was refluxed for 5 min on a water bath (80° C). 0.5 mL of BF₃ solution (10% in methanol) was then added and the reflux was continued for 2 min, followed by the addition of 1 mL of heptane and further refluxing for 1 min. The mixture was then cooled, transferred into a small test tube and after addition of a small amount of a saturated solution of sodium chloride, gently swirled and allowed to separate. The organic layer, which contained the methyl ester, was collected by a Pasteur pipette. The methyl esters were identified by comparison of their retention times and mass spectra with those of authentic methyl linolenate and methyl (*E*)-*p*-coumarate by TLC and GC–MS analyses.

3.8. Antibacterial activity

Antibacterial activity was carried out by the agar diffusion method (Barry, 1991). Different concentrations of the test materials were prepared in DMSO and 5 mm diameter wells in the agar were filled with 60 µL of each sample. Refampicin was used as the reference compound. Seven microorganisms were used: *Staphylococcus aureus* ATCC 25923, *Bacillus subtilis* NCTC 8236, *Bacillus cereus* ATCC 10876, *Escherichia coli* ATCC 25922, *Pseudomonas aeruginosa* ATCC 17853, *Salmonella typhi* (lab. strain) and *Klebsiella pneumoniae* (lab. strain). The microorganisms were provided by the Microbiology Laboratory, School of Pharmaceutical Sciences, Universiti Sains Malaysia, Malaysia. The experiments were carried out in duplicate.

Supplementary material

Additional spectral data relating to this article is available online, alongside Figures S1–S6.

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References

- Barry, A.L. (1991). Procedures and theoretical considerations for testing antimicrobial agents in agar media. In V. Lorian (Ed.), *Antibiotics in laboratory medicine* (3rd ed., pp. 1–23). Baltimore: Williams and Wilkins.
- Burkill, I.H. (1966). A dictionary of the economic products of the Malay Peninsula (Vol. 2). Kuala Lumpur: Ministry of Agriculture and Cooperatives.
- Chou, C.J., & Liao, C. (1982). Phytochemical and pharmacological studies on the flower of Melastoma candidum D. Don forma albiflorum J. C. Ou. Guoli Zhongguo Yiyao Yangjiuso Yanjiu Baogao, 69–129. Abstract retrieved from SciFinder (Accession No. 1983:77988).
- Collins, D.J., Pilotti, C.A., & Wallis, A.F.A. (1992). Triterpene acids from some Papua New Guinea Terminalia species. Phytochemistry, 31, 881–884.
- Das, K.K., & Kotoky, J. (1988). A new aliphatic constituent of *Melastoma malabathricum* Linn. Journal of the Indian Chemical Society, 65, 385–386.
- Dinda, B., & Saha, M.K. (1986). Aliphatic compounds from Melastoma malabathricum. Journal of the Indian Chemical Society, 63, 764–766.
- Dinda, B., & Saha, M.K. (1988). A flavonol diglycoside from Melastoma malabathricum. Journal of the Indian Chemical Society, 65, 209–211.
- Furuya, T., Orihara, Y., & Hayashi, C. (1987). Triterpenoids from *Eucalyptus perriniana* cultured cells. *Phytochemistry*, 26, 715–719.
- Kriwacki, R.W., & Pitner, T.P. (1989). Current aspects of practical two-dimensional (2D) nuclear magnetic resonance (NMR) spectroscopy: applications to structure elucidation. *Pharmaceutical Research*, 6, 531–554.

- Li, X., Elsohly, H.N., Hufford, C.D., & Clark, A.M. (1999). NMR in assignments of ellagic acid derivatives. *Magnetic Resonance in Chemistry*, 37, 856–859.
- Loggia, R.D., Negro, P.D., Bianchi, P., Romussi, G., & Tubaro, A. (1989). Topical antiinflammatory activity of some flavonoids from *Quercus ilex* leaves. *Planta Medica*, 55, 109–110.
- Lohezic-Le Devehat, F., Bakhtiar, A., Bezivin, C., Amoros, M., & Boustie, J. (2002). Antiviral and cytotoxic activities of some Indonesian plants. *Fitoterapia*, 73, 400–405.
- Lowry, J.B. (1976). Anthocyanins of the Melastomataceae, Myrtaceae and allied families. *Phytochemistry*, 15, 513–516.
- Mabry, T.J., Markham, K.R., & Thomas, M.B. (1970). The systematic identification of flavonoids. Heidelberg: Springer Verlag.
- Manzoor-I-Khuda, M., Chowdhury, S.A., Reza, T., & Chowdhury, A.K. (1981). Chemical investigation on *Melastoma malabathricum*. Part 1: isolation of melastomic acid and β -sitosterol from the roots. *Journal of Bangladesh Academy of Sciences*, 5, 55–59.
- Markham, K.R., Ternai, B., Stanley, R., Geiger, H., & Mabry, T.J. (1978). ¹³C-NMR studies of flavonoids II. *Tetrahedron*, 34, 1389–1397.
- Mohandoss, S., & Ravindran, P. (1993). Flavonoids from *Melastoma malabathricum*. *Fitoterapia*, 64, 277–278.
- Perry, L.M. (1980). Medicinal plants of east and south-east Asia: Attributed properties and uses. Cambridge, MA: MIT press.
- Sakakaibara, J., Kaiya, T., Fukuda, H., & Ohki, T. (1983). 6β-Hydroxyursolic acid and other triterpenoids of *Enkianthus cernuus*. *Phytochemistry*, 22, 2553–2555.
- Seo, S., Tomita, Y., & Tori, K. (1981). Biosynthesis of oleanene- and ursene-type triterpenes from [4-¹³C] mevalonolactone and [1,2-¹³C₂] acetate in tissue cultures of *Isodon japonicus* Hara. *Journal of the American Chemical Society*, 103, 2075–2080.
- Silverstein, R.M., Bassler, G.C., & Morrill, T.C. (1991). Spectrometric identification of organic compounds (5th ed.). Toronto: John Wiley & Sons, Inc.
- Sulaiman, M.R., Somchit, M.N., Israf, D.A., Ahmad, Z., & Moina, S. (2004). Antinociceptive effect of *Melastoma malabathricum* ethanolic extract in mice. *Fitoterapia*, 75, 667–672.
- Susanti, D., Sirat, H.M., Ahmad, F., Ali, R.M., Aimi, N., & Kitajima, M. (2007). Antioxidant and cytotoxic flavonoids from the flowers of *Melastoma malabathricum* L. *Food Chemistry*, 103, 710–716.
- Ternai, B., & Markham, K.R. (1976). Carbon-13 NMR studies of flavonoids-I, flavones and flavonols. *Tetrahedron*, 32, 565–569.
- Yamagishi, T., Zhang, D., Chang, J., McPhail, D.R., McPhail, A.T., & Lee, K. (1988). The cytotoxic principles of *Hyptis capitata* and the structures of the new triterpenes hyptatic acid-A and -B. *Phytochemistry*, 27, 3213–3216.
- Yoshida, T., Nakata, F., Hosotani, K., Nitta, A., & Okuda, T. (1992a). Dimeric hydrolysable tannins from *Melastoma malabathricum*. *Phytochemistry*, 31, 2829–2833.
- Yoshida, T., Nakata, F., Hosotani, K., Nitta, A., & Okuda, T. (1992b). Tannins and related polyphenols of Melastomataceous plants. V. Three new complex tannins from *Melastoma malabathricum* L. *Chemical Pharmaceutical Bulletin*, 40, 1727–1732.