

Phytochemistry 52 (1999) 1739-1744

PHYTOCHEMISTRY

Cyclopeptide alkaloids from Heisteria nitida

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Received 18 November 1998; received in revised form 10 March 1999; accepted 10 March 1999

Abstract

Integerrenine, and a new cyclopeptide alkaloid, containing the unusual amine oxide function, anorldianine 27-N oxide, stigmasterol, β -sitosterol, lupeol, (+)-catechin, (—)-epicatechin and 4-hydroxy-2-methoxy benzoic acid were isolated from the Ecuadorian medicinal plant *Heisteria nitida* (Engl.), Olacaceae. The structures were determined by UV, IR, NMR and mass spectroscopic investigations and chemical transformations. The cyclopeptide alkaloids have been isolated and characterized for the first time in the family Olacaceae. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: Heisteria nitida; Olacaceae; Cyclopeptide alkaloids; Anorldianine 27-N oxide; Integerrenine

1. Introduction

Heisteria (Jacq.), Olacaceae, is a genus of small trees primarily found in tropical America and West Africa. H. nitida has been used in traditional medicine for treatment of diarrhoea and hepatic infection (Ghia, F., Personal communication, 1992). Various Heisteria species are used by South-American Indians in the treatment of rheumatism, abscesses, headache, throat infections, swellings, nose bleedings, pain in joints and muscles (Kvist & Holm-Nielsen, 1987; Pinkley, 1969; Russo, 1992; Schultes & Raffauf, 1990; Sleumer, 1984; Williams, 1936). H. pallida is reported to have antiphlogistic effect after external application in patients with rheumatism and arthritis (Wiemann, 1990). Two varieties of H. accuminata have been investigated in the rat-paw oedema model where one of the extracts showed a moderate inhibition of carrageenan induced oedema (Ortega, Carretero, Pascual, Villar & Chiriboga, 1996). Scopolamine have been isolated from H. olivae (Cairo Valera, De Budowski, Delle

Monache & Marini-Bettolo, 1977). Triterpenes and proanthocyanidines have been isolated from H. pallida (Gonzalez, Gonzalez, Ferro & Ravelo, 1988; Dirsch, Wiemann & Wagner, 1992; Dirsch, Neszmèlyi & Wagner, 1993). Recently from H. accuminata several acetylenic fatty acids have been isolated and structudetermined rally (Kraus, Neszmèlyi, Holly, Wiedemann, Nenninger, Torssell et al., 1998). H. nitida (Engl.) has not been investigated previously. The aim of this study was to find novel chemical structures with anti-inflammatory or anti-viral activity. Preliminary pharmacological tests of crude extracts from the bark of H. nitida showed moderate or weak anti-inflammatory (PAF, PG tests), anti-viral activity and toxic properties. However, the NMR spectra showed interesting features which directed the study to isolation and characterization of new chemical structures.

Cyclopeptide alkaloids are widespread and occur in several families: Asteraceae, Celastraceae, Euphorbiaceae, Menispermaceae, Pandaceae, Rhamnaceae, Rubiaceae, Sterculiaceae and Urticaceae (Gournelis, Lascaris & Verpoorte, 1997). Anti-fungal, antibacterial and sedative activities of cyclopeptide

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alkaloids have been reported (Gournelis et al., 1997; Tschesche, David, Serbes, Von Radloff, Kaussmann & Eckhardt, 1974; Schmidt, Lieberknecht & Haslinger, 1985; Pandey & Devi, 1990; Blanpin, Païs & Quevauviller, 1963). In the present work cyclopeptide alkaloids have been isolated and characterized for the first time in the family Olacaceae.

2. Results and discussion

The ¹H NMR spectrum of the crude ethyl acetate extract showed some interesting features both in the aromatic, olefinic and aliphatic methyl regions. From the unpolar fractions long chain hydrocarbons, fatty acids and fatty acid methyl esters, stigmasterol, β -sitosterol and lupeol were isolated. From the most polar

Table 1

¹H and ¹³C NMR correlation data of **2**; At 400 MHz (¹H) and 100.6 MHz (¹³C), in CDCl₃ with TMS as int. standard. Chemical shifts are relative to residual CHCl₃ for ¹H (d 7.25) and to the solvent for ¹³C (δ 77.00). The multiplicity of the carbons was assigned from the HMQC and DEPT spectra

Position	Kind of carbon	^{13}C shift δ_C	1 H shift δ_{H} (multi., J_{HH} Hz)	HMBC correlations via ${}^{2,3}J_{CH}$
1	qC=	157.5(s)	_	Н-3
3	ĊH	83.4(d)	5.01 (dd, 8.2; 1.5)	H-18: H-19: H-17
4	СН	53.9(d)	4.78 (br w _{1/2} 18 Hz)	_
5	C=O	170.9(s)	_	H-3; H-7
7	СН	62.9(d)	4.21 (d, 8.2)	H-30; H-31
8	C=O	167.1 (s)	_	H-7; NH-9; H_{α} -30; H_{β} -30
9	NH	-	6.20 (<i>d</i> , 10.2)	_
10	CH=	126.4(d)	6.64 (<i>dd</i> , 10.2; 7.3)	NH-9; H-11
11	CH=	116.3(d)	6.39(d, 7.3)	H-10; NH-9
12	qC==	131.7(s)	_	H-11; H-10
13	CH=	130.6(d)	6.96 (<i>d</i> , 7.3)	_
14	CH=	120.7(d)	7.07 (d, 7.3)	_
15	CH=	121.1(d)	7.29 (<i>d</i> , 8.8)	_
16	CH=	132.1(d)	7.10 (<i>d</i> , 8.8)	_
17	СН	29.0(d)	1.26 (<i>m</i>)	H-18; H-19; H-3
18	CH ₃	15.2(q)	0.99 (<i>d</i> , 6.7)	H-19
19	CH ₃	20.6(q)	1.24 (<i>d</i> , 6.7)	H-3
20	NH	-	$9.64 (br \ s)$	_
21	C=O	165.9 (s)	_	_
22	СН	77.2(d)	4.64 (br d, 10.3)	H-28; H-29
23	CH ₂	36.3 (<i>t</i>)	1.70 (<i>m</i>), 1.98 (<i>m</i>)	H-25; H-26
24	СН	25.7(d)	1.25 (<i>m</i>)	H-25; H-26
25	CH ₃	21.4(q)	0.94 (<i>d</i> , 6.4)	H-26
26	CH ₃	23.9(q)	0.93 (d, 6.4)	_
28	$N(CH_3)$	54.2(q)	3.35(s)	H-29
29	$N(CH_3)$	55.1(q)	3.40(s)	H-28
30	CH_2	28.2(t)	1.47(m), 2.32(m)	H-7
31	CH_2	23.9(t)	1.70 (<i>m</i>), 1.90 (<i>m</i>)	H-7
32	CH ₂	46.9(t)	3.35 (<i>m</i>), 3.86 (<i>m</i>)	H-7



Scheme 1. Most significant mass spectral fragments of compound 2.

fractions two cyclopeptide alkaloids: integerrenine 1 and a new alkaloid, anorldianine 27-N oxide 2, belonging to the same type of tetrameric cyclopeptides were obtained.

The high resolution mass spectrum of the major alkaloid 1 (Scheme 1) in the FAB (positive) mode gave an $[M+H]^+$ ion at m/z 535.3286 corresponding to a molecular formula of C₃₁H₄₂ N₄ O₄. Our UV, IR, ¹H and EI MS spectra of the compound agreed well with previously published data for the cyclopeptide alkaloid integerrenine (Tschesche, Rheingans, Fehlhaber & Legler, 1967; Medina & Spiteller, 1981; Lagarias, Goff, Klein & Rapoport, 1979). It ought to be observed that the ¹H and ¹³C NMR spectra are different in pure CDCl₃ and in CDCl₃-10% CD₃OD as a result of conformational changes of the molecule in different solvents (Haslinger, 1978).

The new peptide alkaloid, $m/z [M+H]^+$ 501, corre-

sponding to the elementary composition $C_{27}H_{40}$ N₄ O₅, was isolated as a semisolid from a more polar fraction. It gave a slightly yellow colour with Dragendorff's reagent. The IR spectrum exhibited and Ar-O-C groups. The UV spectrum showed absorption maxima at 320 and 266 nm typical for a styrylamine chromophore in cyclopeptide alkaloids (Tschesche, David, Uhlendorf & Fehlhaber, 1972). The ¹H, ¹³C NMR, DEPT and HMQC spectra showed the presence of a 14-membered cyclic system consisting of the three amino acids leucine, proline and 3-hydroxyleucine, three carbonyl groups at $\delta_{\rm C}$ 170.9, 167.1 and 165.9 ppm. The olefinic carbon atoms were located at $\delta_{\rm C}$ 116.3 and 126.4 ppm and a *p*-hydroxystyryl amine unit which derives from tyrosine by degradation. In the HMBC spectrum the signal at $\delta_{\rm C}$ 167.1 showed two-bond ¹H-¹³C correlations with $\delta_{\rm H}$ 6.20 of the styrylamino NH-9 and $\delta_{\rm H}$ 4.21 of the proton H-7 of proline. Moreover three-bond correlations to β and β' protons of the amino acid at $\delta_{\rm H}$ 1.47 and 2.32, were also present. For the signal at δ_C 170.9, three-bond correlation peaks to H-7 (δ_H 4.21) and H-3 (δ_H 5.01) were found. More HMBC data are summarized in Table 1. The methyl absorptions at δ 3.35 and 3.40 ppm were puzzling because with reference to the elementary composition it was not possible to accomodate two methoxy groups in the molecule and the N—CH₃ absorptions normally occur in the δ 2.0–2.5 ppm region. Another possibility was that the compound contained the unusual amine oxide function O^- — N^+ (CH₃)₂ with two prochiral N-methyl groups (C-28, 29), $\delta_{\rm H}$ 3.35 and 3.40 ppm, $\delta_{\rm C}$ 54.2 and 55.1 ppm (CDCl₃, 10% CD₃OD). Before we realized that the compound contained this polar amine oxide function we experienced considerable loss of substance by trying to purify the compound on preparative silica gel plates. However, column chromatography on Sephadex LH-20 was successful. The shifts of the C-3, δ 83.4 ppm, C-4, δ 53.9 ppm, and $J_{3,4-H} = 8$ Hz indicated an L-erythro-3-hydroxyleucine structural unit (Gournelis et al., 1997). The vicinal coupling constant of 3,17-H was small, approximately 1–1.5 Hz and the four C—CH₃ groups appeared as doublets at δ 1.24, 0.99, 0.94 and 0.93 ppm. Four methylene groups were observed in the DEPT spectrum, in agreement with the incorporation of one proline and one leucine unit. Hydrolytic cleavage of the alkaloid gave free proline. The studies of the fragmentation pattern in the EI and FAB MS spectra gave conclusive evidence for the structure. Scheme 1 shows the prominent fragments, the formation of which follows earlier proposed fragmentation scheme (Tschesche & Kaussmann, 1975). Of interest is the peak I at m/z = 439, which corresponds to the expected Cope reaction product, M⁺ minus [(CH₃)₂ NOH]. The ion e at m/z 135 revealed the presence of a styrylamine unit, the ion **f** at m/z 189 indicated a 3-hydroxylated leucine unit and the oddelectron ion **i** at m/z 328 and its fragmentation ions **j** and **k** proved that proline was part of the molecule. All together, the data fit the structure **2** for the alkaloid. The deoxy compound anorldianine has previously been isolated from *Canthium anorldianum* (Dongo, Ayafor, Sondengam & Connolly, 1989).

3. Experimental

3.1. General

The ¹H, ¹³C NMR, HMQC (heteronuclear multiple quantum correlation) and HMBC (heteronuclear multiple bond correlation) spectra were recorded at 400/ 100.6 MHz in CDCl₃ on a Bruker DRX 400 instrument with TMS as internal standard. The IR spectra were recorded with a Nicolet MX-S, the UV spectra with a Perkin-Elmer Lambda 2UV/VIS spectrophotometer and the EI MS and positive-ion FAB MS (with glycerol as matrix) spectra were recorded with a JEOL JMS SX/SX102A instrument (JEOL, Japan). The melting points were determined using a Digital Point Apparatus (model Melting IA 8103, Electrothermal Engineering Ltd, Southend-on-Sea, Essex, UK) and are uncorrected. Optical rotation was determined at ambient temperature using a Perkin-Elmer polarimeter 241. TLC was performed on precoated aluminum sheets on silica 60 F₂₅₄, 0.25 mm (Merck, Darmstadt, Germany) and preparative TLC was performed on silica gel (60, PF₂₅₄₊₃₆₀, Merck) glass plates, 20×20 , 0.25 and 0.5 mm (Merck). UV light (245 and 366 mm) and spraying with vanillin-sulfuric acid reagent followed by heating (120°) , was used for detection.

Medium-pressure liquid chromatography (MPLC) was performed using a SEPARO AB MPLC equipment (Baeckström Separo AB, Lidingö, Sweden) (Jirón, 1996). SEPARO Variable length glass columns with an inner diameter of 1.5 or 2.5 cm, packed with silica gel 60, 40-63 µm (Merck) were used. A FMI Lab pump, model QD (Fluid Metering Inc., Oyster Bay, NY, USA) was used at a flow rate of 20-30 ml/ min. Fractions of 9 ml were collected with a Gilson fraction collector model 201. The columns were eluted with continuous gradients running from hexane, over CH₂Cl₂ to MeOH, and H₂O afforded by a SEPARO constant-volume mixing chamber combined with an open reservoir. Initially, the mixing chamber contained 50 ml non-polar solvent and the reservoir the first of 15-20 premixed binary (less polar/more polar solvent) gradient mixtures, of 20-40 ml each, which were successively fed to the reservoir during the separation.

3.2. Plant material

The bark of *Heisteria nitida* (Engl.), was collected by Dr Felipe Ghia in 1992 at the Reserva Biologica, Jatun Sacha, Provincia del Napo, Ecuador. Voucher specimens are deposited in the Herbario Economico, Escuela Politecnica Nacional, EPN, Quito, Ecuador, (G. F. 539) and in the Herbarium of the Department of Systematic Botany, Uppsala University, Sweden.

3.3. Extraction

The plant material was dried at 40° in the dark in a ventilated hood and grounded. The material, 1.1 kg, was extracted exhaustively at room temperature three times with light petroleum (40–60°) with occasional stirring followed by three times with methanol for 8 days each time. The extracts were evaporated in vacuo to give 8.3 and 86.5 g of a gelatinous and an oily material respectively. The methanol extract was partitioned between ethyl acetate and water to give 17.2 g of an ethyl acetate soluble fraction. An insoluble residue (2.8 g) was discarded. The water phase was freeze dried to give 66.6 g of crude material which consisted mainly of carbohydrates. It was not further investigated.

3.4. Isolation and purification

The ethyl acetate fraction, 16 g, was subjected to SEPARO column chromatography on silica gel 60 (30 g) with gradient elution using hexane-CH₂Cl₂ and EtOAc-MeOH. At low solvent polarity stigmasterol, β -sitosterol and lupeol were eluted as identified by ¹H NMR spectroscopy. The CH₂Cl₂-EtOAc fractions contained impure integerrenine **1** mixed with fatty acid material. Increased polarity gave small amounts of impure anorldianine 27-N oxide **2**. It was purified by chromatography on a Sephadex LH-20 column using gradient elution with CHCl₃-MeOH, and EtOAc-MeOH mixtures. At still higher eluent polarity (+)-catechin, (-)-epicatechin and 4-hydroxy-3-methoxy-benzoic acid were obtained.

Stigmasterol was purified by chromatography on prep. TLC plates using CHCl₃–MeOH (99:1) as eluent, 63 mg. It was identified by its ¹H NMR spectrum.

 β -Sitosterol was purified by chromatography on prep. TLC plates using CHCl₃–MeOH (98:2) as eluent, 87 mg. It was identified by its ¹H NMR spectrum.

Lupeol was purified by chromatography on prep. TLC plates using CHCl₃–MeOH (95:5) as eluent, amorphous solid, 57 mg, $[\alpha]_D^{22} + 27.3^\circ$ (CHCl₃; *c* 0.4). ¹H and ¹³C NMR data agreed with lit. data (Connolly & Hill, 1991).

Integerrenine 1 was purified by chromatography on a Sephadex LH-20 column using CH_2Cl_2 as eluent.

Further purification was achieved by dissolving the combined fractions containing integerrenine in 0.1 N hydrochloric acid, basification with solid NaHCO₃ and extraction with chloroform followed by prep. TLC using CHCl₃–MeOH (99:1) as eluent, 240 mg, mp 278° (acetonitrile), lit. [21] mp 274°, $[\alpha]_D^{22} -232°C$ (CHCl₃; *c* 0.2) HRFAB-MS, found: m/z [M+H]⁺, 535.3286. Cal. for [M+H]⁺ C₃₁H₄₃ N₄ O₄, 535.3284; EI-MS and the ¹H NMR data agreed with published data of integerrenine (Tschesche et al., 1967; Medina & Spiteller, 1981; Lagarias et al., 1979).

Anorldianine 27-N oxide **2** was purified by chromatography on a Sephadex LH-20 column using CH₂Cl₂–MeOH (1:1) as eluent. Further purification was achieved by dissolving the combined fractions containing **2a** in 0.1 N hydrochloric acid, basification with solid NaHCO₃ and extraction with chloroform. The yield of **2a** was 45 mg. IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹ 3343, 1675 (secondary amide), 2790 (N—Me), 1623 (conjugated C=C), 1241, 1115 (aryl ether). UV $\lambda_{\text{max}}^{\text{MeOH}}$ (log ϵ): 266 (3.71), 320 (3.31). HRFAB-MS, found: [M+H]⁺, 501.3062, C₂₇H₄₁ N₄ O₅ requires [M+H]⁺, 501.3072; EI-MS *m*/*z* (rel. int.): 500 [M]⁺ (<1), 439 (76), 328 (25), 305 (78), 287 (8), 251 (19), 214 (19), 194 (30), 189 (100), 166 (18), 135 (23), 114 (37), 97 (70), 89 (37), 70 (99), 61 (60), 60 (71), 42 (74). The ¹H and ¹³C NMR data are listed in Table 1.

Hydrolysis of anorldianine 27-N oxide **2a** (ca 200 μ g) was hydrolyzed with 6 M HCl (100 μ l, 100°, 16 h) in a sealed glass tube. After evaporization, esterification was performed with 100 μ l of 1 M methanolic HCl (100°, 30 min). For N-acetylation, MeOH–Ac₂O (4:1) was added to the methyl ester in 100 μ l H₂O at room temp. (45 min). The product in MeOH was analyzed by GC-MS using an HP-5, 25 m fused silica WCOT column, temp. programmed 140° for 3 min–230° for 6 min, the MS of the peak at ret. time 8.2 min was identical to an authentic sample of N-acetylproline methyl ester.

(+)-Catechin was purified by chromatography on a Sephadex LH-20 column using CHCl₃–MeOH (1:9) as eluent, amorphous solid, 46 mg, $[\alpha]_D^{22}$ +11.3° (Me₂CO; *c* 0.2). The ¹H and ¹³C NMR data agreed with lit. data (Morimoto, Nonaka, Nishioka, Ezaki & Takizawa, 1985; Baldé et al., 1991).

(—)-Epicatechin was purified by chromatography on a Sephadex LH-20 column using MeOH as eluent, amorphous solid, 35 mg, $[\alpha]_D^{22} - 12.2^\circ$ (Me₂CO; *c* 0.2). The ¹H and ¹³C NMR data agreed with lit. data (Morimoto et al., 1985; Baldé et al., 1991).

4-Hydroxy-2-methoxybenzoic acid, was purified by chromatography on Sephadex LH-20 using MeOH as eluent followed by prep. TLC using $CHCl_3$ -MeOH- H_2O (3 : 9 : 1) as eluent, 28 mg. It was identified by its ¹H NMR spectrum.

Acknowledgements

We are indebted to the Swedish Institute and Egyptian government for fellowships to H.E. This work was supported by the Swedish Council for Forestry and Agricultural Science and the Danish Natural Science Research Council. We are also very grateful to Dr Felipe Ghia for identifying and collecting the plant material.

References

- Baldé, A. M., Pieters, L. A., Gergely, A., Kolodziej, H., Claeys, M., & Vlietinck, A. J. (1991). *Phytochemistry*, 30, 337.
- Blanpin, O., Païs, M., & Quevauviller, M. A. (1963). Ann. Pharm. Fr., 21, 147.
- Cairo, Valera G., De Budowski, J., Delle, Monache F., & Marini-Bettolo, G. B. (1977). Atti Accad. Naz. Lincei, Cl. Sci. Fis. Mat. Nat. Rend., 62, 363.
- Connolly, J. D., & Hill, R. A. (1991). In (p. 1261). In *Dictionary of terpenoids*, vol. 2. Chapman & Hall and references therein.
- Dirsch, V., Wiemann, W., & Wagner, H. (1992). *Pharm. Pharmacol. Lett.*, 2, 184.
- Dirsch, V., Neszmèlyi, A., & Wagner, H. (1993). *Phytochemistry*, 34, 291.
- Dongo, E., Ayafor, J. F., Sondengam, B. L., & Connolly, J. D. (1989). J. Nat. Prod., 52, 840.
- Gonzalez A.G., Gonzalez C.M., Ferro E.A., Ravelo A.G. (1988) J. Chem. Research (M), 0273.
- Gournelis, D. C., Lascaris, G. G., & Verpoorte, R. (1997). Nat. Prod. Reports, 14, 75 and references therein.
- Haslinger, E. (1978). Tetrahedron, 34, 685.
- Jirón Z. (1996) Approaching optimal conditions for running liquid

adsorption column chromatography using simple computational models, Licentiate thesis, Department of Chemistry, Royal Institute of Technology, Stockholm, Sweden.

- Kraus, C. M., Neszmèlyi, A., Holly, S., Wiedemann, B., Nenninger, A., Torssell, K. B. G., Bohlin, L., & Wagner, H. (1998). J. Nat. Prod., 61, 422.
- Kvist, L. P., & Holm-Nielsen, L. B. (1987). Opera Botanica, 92, 83.
- Lagarias, J. C., Goff, D., Klein, F. K., & Rapoport, H. (1979). J. Nat. Prod., 42, 220.
- Medina E., Spiteller G. (1981) Liebigs Ann. Chem., 1981, 538.
- Morimoto, S., Nonaka, G-I., Nishioka, I., Ezaki, N., & Takizawa, N. (1985). Chem. Pharm. Bull., 33, 2281.
- Ortega, T., Carretero, M. E., Pascual, E., Villar, A. M., & Chiriboga, X. (1996). *Phytother. Res.*, 10, S121.
- Pandey, V. B., & Devi, S. (1990). Planta Med., 56, 649.
- Pinkley, H. W. (1969). Lloydia, 32, 305.
- Russo, E. B. (1992). J. Ethnopharmacol., 36, 193.
- Schmidt, U., Lieberknecht, A., & Haslinger, E. (1985). In A. Brossi, *The alkaloids*, vol. 26 (p. 299). New York: Academic Press.
- Schultes, R. E., & Raffauf, R. F. (1990). In *The healing forest*. *Medicinal and toxic plants of the Nortwest Amazonia* (p. 342). Portland, Oregon: Dioscorides Press.
- Sleumer H.O. (1984) Flora neotropica, Monograph number 38.
- Tschesche, R., Rheingans, J., Fehlhaber, H-W., & Legler, G. (1967). *Chem. Ber.*, 100, 3924.
- Tschesche, R., David, S. T., Uhlendorf, J., & Fehlhaber, H-W. (1972). *Chem. Ber.*, *105*, 3106.
- Tschesche R., David S.T., Serbes R., Von Radloff M., Kaussmann E.V., Eckhardt G. (1974) *Liebigs Ann. Chem.*, 1974, 1915.
- Tschesche, R., & Kaussmann, E. U. (1975). In R. H. F. Manske, *The alkaloids, vol. 15* (p. 165). New York: Academic Press.
- Wiemann W. (1990) German Patent, P 39 05 033. 5, European Patent EP, 0384 308.
- Williams, L. (1936). *Woods of northwestern Peru* Publication 377 of the Field Museum of Natural History, Chicago.