

Anionic O  $\rightarrow$   $\alpha$ - and  $\beta$ -Vinyl Carbamoyl  
Translocation of 2-(*O*-Carbamoyl)  
Stilbenes<sup>†</sup>

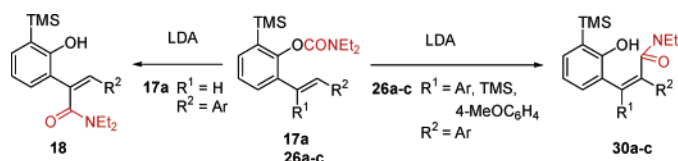
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## ABSTRACT



New anionic oxygen to  $\alpha$ - and  $\beta$ -vinyl carbamoyl migration reactions, 17a and 26a–c  $\rightarrow$  18 and 30a–c, proceed under LDA-mediated conditions leading stereoselectively to highly substituted stilbenes bearing electron-donating and -withdrawing substituents. Compounds 17a and 26a–c are prepared by combination of efficient, directed *ortho* metalation, Sonogashira, and Suzuki–Miyaura cross-coupling procedures.

In continuation of efforts (Figure 1, 1–6) to devise useful carbanion-based methodology for the synthesis of aromatics

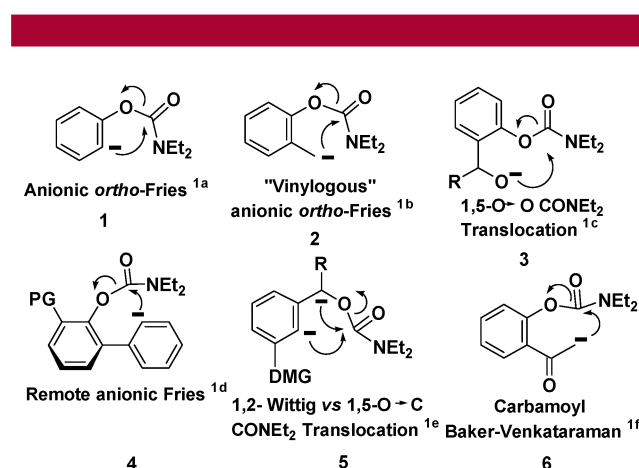
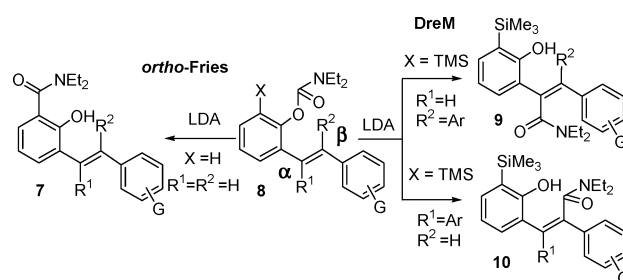


Figure 1.

and heteroaromatics,<sup>1</sup> we have uncovered a stereoselective LDA-mediated O  $\rightarrow$  *ortho*-vinyl carbamoyl rearrangement

<sup>†</sup> This paper is dedicated with respect and appreciation to Al Meyers who pointed the way to *ortho* metalation and provides a rich lore of synthetic organolithium chemistry.

## Scheme 1

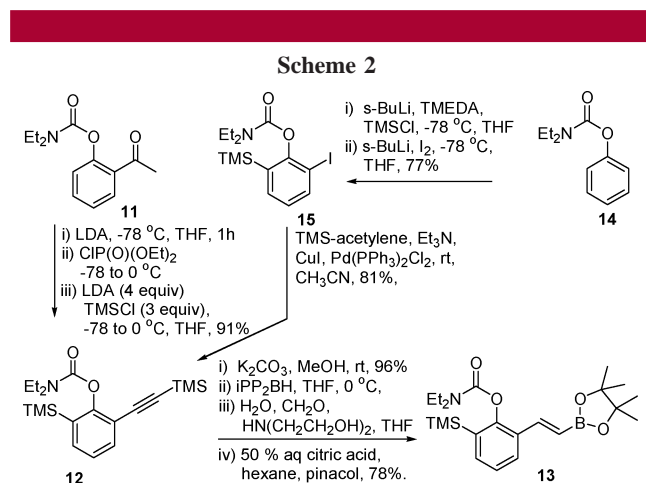


of 2-*O*-carbamoyl stilbenes (Scheme 1). With the obligatory prevention of the anionic Fries rearrangement<sup>1a</sup> (8  $\rightarrow$  7) by silicon protection (8, X = TMS), migration to both  $\alpha$ - (8  $\rightarrow$  9) and  $\beta$ -vinyl (8  $\rightarrow$  10) positions may be induced, depending

(1) (a) *Anionic ortho-Fries*: Sibi, M. P.; Snieckus, V. *J. Org. Chem.* **1983**, *48*, 1935–1937. (b) *Vinylogous anionic ortho-Fries*: Kalinin, A. V.; Miah, M. A. J.; Chattopadhyay, S.; Tsukazaki, M.; Wicki, M.; Nguen, T.; Coelho, A. L.; Kerr, M.; Snieckus, V. *Synlett* **1997**, *7*, 839–841. (c) *1,5-O  $\rightarrow$  O CONEt<sub>2</sub> translocation*: Chauder, B. A.; Kalinin, A. V.; Taylor, N. J.; Snieckus, V. *Angew. Chem., Int. Ed.* **1999**, *38*, 1435–1438. (d) *Remote anionic Fries*: Wang, W.; Snieckus, V. *J. Org. Chem.* **1992**, *57*, 424–426. (e) *1,2 Wittig versus 1,5-O  $\rightarrow$  C CONEt<sub>2</sub> translocation*: Zhang, P.; Gawley, R. E. *J. Org. Chem.* **1993**, *12*, 3222–3223. (f) *Carbamoyl Baker–Venkatarman reaction*: Kalinin, A. V.; da Silva, A. J. M.; Lopes, C. C.; Lopes, R. S. C.; Snieckus, V. *Tetrahedron Lett.* **1998**, *39*, 4995–4998.

on the open site, and thereby provides new heterocycles. In particular, our method allows the synthesis of 3-aryl coumarin and 3,4-diaryl coumarin approaches to diversely substituted stilbenes and, from these, benzofurananone<sup>2</sup> **19** and coumarin<sup>3</sup> **31**, which, in the context of current studies, may be of future biological interest.<sup>4,5</sup> Although mechanistically somewhat ambiguous (vide infra), the carbamoyl migrations are, to the best of our knowledge, unprecedented in aromatic carbanionic chemistry<sup>6</sup> and may be conceptualized to be driven by the complex-induced proximity effect (CIPE).<sup>7</sup>

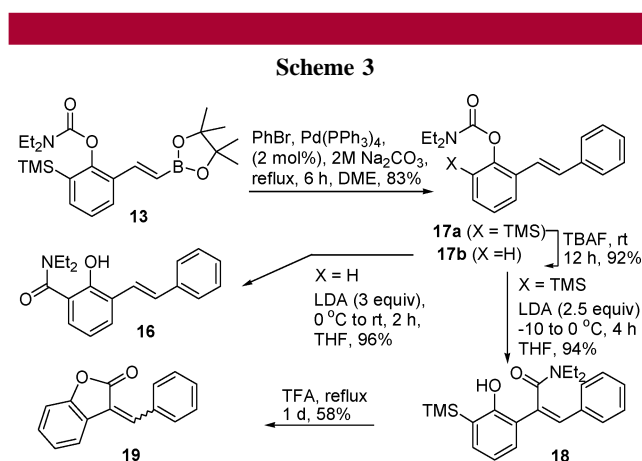
The requisite test substrate **8**, X = TMS for the *O*-carbamoyl  $\rightarrow$   $\alpha$ -vinyl migration was prepared by first constructing the intermediate styryl borolane **13** using a two-pronged approach (Scheme 2). Thus, in a five-step synthetic



operation in one pot and following, in part, the interesting Negishi protocol,<sup>8</sup> treatment of the enolate of the readily

available 2-*O*-carbamoyl acetophenone **11**<sup>9</sup> with diethyl chlorophosphonate at  $-78$  °C followed by LDA (4 equiv) and TMSCl (3 equiv) gave the bis-silylated aryl acetylene **12** in high yield. Selective desilylation followed by hydroboration with *i*(isopropylprenyl)borane (*i*PP<sub>2</sub>BH)<sup>10</sup> gave the vinyl borolane **13**.<sup>11</sup>

In the alternative route, subjection of the *O*-phenyl carbamate **14** to a sequential, one-pot double-directed ortho metalation (DoM)—TMSCl and iodine quench procedure afforded the contiguously substituted iodide **15** (77% yield) which, upon Sonogashira coupling with TMS acetylene furnished the bis-silylated derivative **12** in 81% yield. In view of the high cost of TMS acetylene and the difficulties of scaling-up this route, the procedure starting from **11** was preferred. Suzuki–Miyaura cross coupling of **13** with bromobenzene under standard conditions delivered the requisite 2-*O*-carbamoyl stilbene **17a** in 83% yield (Scheme 3).<sup>12</sup> As expected from general observation,<sup>1a</sup> treatment of



**17b** (Scheme 3), derived by selective desilylation (TBAF/rt) of **17a**, with LDA (3 equiv) resulted in quantitative conversion to the anionic *ortho*-Fries rearrangement product **16**. On the other hand, the TMS-protected derivative **17a**, upon treatment with LDA (2.5 equiv) between  $-10$  and  $0$  °C led smoothly to the formation of the carbamoyl-migrated product **18** in 94% yield. To ensure the position of amide translocation, **18** was cyclized to the known benzofuranone **19** (1:1 *E:Z* mixture).<sup>13</sup> A single-crystal X-ray structure analysis of **18**<sup>14</sup> established stereochemical conservation of (*E*)-stilbene in the transformation. Generalization of the

(2) For naturally occurring benzofuranones (isocoumarins) with topomerase inhibitory effects, see: (a) Suzuki, K.; Yahara, S.; Machata, K.; Uyeda, M. *J. Nat. Prod.* **2001**, *64*, 204–207. For recent synthetic work, see: (b) Burke, A. J.; O'Sullivan, W. I. *Tetrahedron* **1997**, *35*, 2539–2543.

(3) For naturally occurring coumarins, some of which exhibit diverse (proapoptotic, antitumor, transcriptional suppression of HIV promoter) bioactivity, see: Bailly, C.; Bal, C.; Barbier, P.; Combes, S.; Finet, J.-P.; Hildebrand, M.-P.; Peyrot, V.; Watez, N. *J. Med. Chem.* **2003**, *46*, 5437–5444 and references therein.

(4) (a) Guilet, D.; Hélesbeux, J.-J.; Séraphin, D.; Sévenet, T.; Richomme, P.; Bruneton, J. *J. Nat. Prod.* **2001**, *64*, 563–568. (b) Murray, R. D. H.; Méndez, J.; Brown, S. A. *The Natural Coumarins: Occurrence, Chemistry and Biochemistry*; J. Wiley: New York, 1982.

(5) 4-Diarylcoumarins may be considered as restricted tamoxifen derivatives, the latter being an FDA approved drug widely used for treatment of estrogen-dependent breast cancer. For key references and recent synthetic work, see: Yu, D. D.; Forman, B. M. J. *Org. Chem.* **2003**, *68*, 9489–9491. Yus, M.; Ramón, D. J.; Gómez, I. *Tetrahedron* **2003**, *59*, 3219–3225.

(6) Stilbene undergoes reductive dilithiation in the presence of Li metal as discovered by Schlenk in his prognostic contributions to organolithium chemistry: (a) Schlenk, W.; Bergmann, E. *Annalen* **1928**, *483*, 106; *Houben-Weyl 13/1*, 162 ff. Monolithio and 1,1- or 1,2-dilithio stilbene species have been generated mainly by metal–halogen exchange or Li addition to diphenylacetylene: (b) Maercker, A.; Kemmer, M.; Wang, H. C.; Dong, D.-H.; Szwarc, M. *Angew. Chem., Int. Ed.* **1998**, *37*, 2136–2138. (c) Boche, G. *Top. Curr. Chem.* **1988**, *146*, 3–56. Their configurational stability and proton-transfer reactions are highly dependent on solvent and temperature, see: (d) *Houben-Weyl*, **1952**, *13/1*, p 133 and **1989E**, *19d*, pp 176, 483, 498. (e) Maercker, A. In *Sapse, A. M., Schleyer, P. von R. Lithium Chemistry. A Theoretical and Experimental Overview*, Wiley: New York, 1995; p 477.

(7) Whisler, M. C.; MacNeil, S.; Snieckus, V.; Beak, P. *Angew. Chem., Int. Ed.* **2004**, *43*, 2206.

(8) Negishi, E.-i.; King, A. O.; Klima, W. L. *J. Org. Chem.* **1980**, *2*, 193–196.

(9) Prepared in 93% yield from 2-hydroxyacetophenone using NaH, CICONET<sub>2</sub>/DMF.

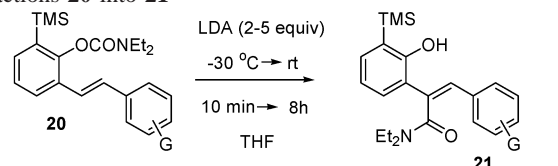
(10) For the general utility of this new hydroboration reagent, see: Kalinin, A.; Scherer, S.; Snieckus, V. *Angew. Chem., Int. Ed.* **2003**, *42*, 3399–3404.

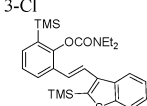
(11) For purification purposes, the intermediate boronic acid was first converted into its crystalline diethanolamine adduct, which was subjected to treatment with pinacol in 50% citric acid/hexane mixture to give analytically pure **13**.

(12) Kalinin, A. V.; Reed, M. A.; Norman, B. H.; Snieckus, V. *J. Org. Chem.* **2003**, *68*, 5992–5999.

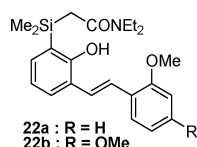
(13) Msaddek, M.; Rammah, M.; Ciamala, K.; Vebrel, J.; Laude, B. *Synthesis* **1997**, 1495–1498.

**Table 1.** Synthetic 2-(*O*-Carbamoyl)stilbene Translocation Reactions **20** into **21**



entry	product	G	ylt %
1	<b>21a</b> <sup>a</sup>	H ( <i>Z</i> -stilbene)	81
2	<b>21b</b> <sup>b</sup>	2-OMe	68
3	<b>21c</b>	3-OMe	54
4	<b>21d</b>	4-OMe	73
5	<b>21e</b> <sup>b</sup>	2,4-OMe	69(88) <sup>c</sup>
6	<b>21f</b> <sup>d</sup>	3-NHBoc	56
7	<b>21g</b> <sup>e</sup>	4-CHO	59
8	<b>21h</b>	4-SO <sub>2</sub> NEt <sub>2</sub>	81
9	<b>21i</b>	4-CO <sub>2</sub> t-Bu	75
10	<b>21j</b>	4-CN	0
11	<b>21k</b>	3-Cl	0
12	<b>21l</b> <sup>f</sup>		48

<sup>a</sup> (*Z*)-Stilbene derivative. <sup>b</sup> In addition, 12 and 15% yields of  $\alpha$ -methylsilyl amides **22a** and **22b**, respectively, were obtained. <sup>c</sup> Corresponding TES derivative of **21e** was used. <sup>d</sup> Substrate **20f** was first treated with TMSCl (1.1 equiv)/LDA (1.1 equiv) at  $-78$  °C, warmed to  $0$  °C, and recooled to  $-78$  °C. <sup>e</sup> Treated with LiNEt<sub>2</sub> at  $-78$  °C and stirred for 30 min, and then the conditions shown were followed. <sup>f</sup> Substrate was sequentially treated with TMSCl (1.5 equiv) and LDA (5 equiv).

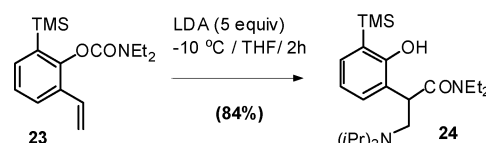


carbamoyl translocation reaction (**20**  $\rightarrow$  **21**, Table 1) deserves comment. Starting substrates were prepared according to the method described (Schemes 2 and 3) using aryl bromides and aryl iodides as coupling partners (See Supporting Information). Interestingly, the (*Z*)-stilbene derivative corresponding to the (*E*)-isomer **17a**, underwent comparatively faster rearrangement to give isomeric carbamoyl-migrated product **21a** in high yield (entry 1). Stilbenes bearing electron-donating OMe groups, **20b–e**, provided the expected products **21b–e** (entries 2–5) but also, in two cases (entries 2 and 5), minor amounts of  $\alpha$ -methylsilyl amides **22a** and **22b**, presumably due, in balance, to an  $\alpha$ -vinyl C–H deacidifying effect of the OMe groups. Following our established circumvention of such byproducts in *O*-biaryl carbamate remote anionic Fries rearrangement,<sup>1d</sup> the *ortho*-triethylsilyl (TES) derivative corresponding to **20e** was prepared and subjected to the LDA conditions to afford cleanly the  $\alpha$ -vinyl carbamoyl-migrated product **21e** in 88%

(14) CCDC 235465 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via the Internet at [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif), by emailing [data\\_request@ccdc.cam.ac.uk](mailto:data_request@ccdc.cam.ac.uk), or by contacting The Cambridge Crystallographic Data Centre, 12, Union Road, Cambridge CB2 1EZ, UK, fax: +44 1223 336033.

yield (entry 5). The *N*-Boc stilbenoid **20f**, upon treatment under the standard excess LDA conditions, led to recovered starting material. Suspecting the possible deactivation effect of the incipient N-anion, we subjected compound **20f** to LDA/TMSCl (1.1:1.1 equiv) followed by LDA (3 equiv) and workup to give the desired product **21f** in modest yield (entry 6). The aldehydic stilbenoid **20g** required initial in situ protection with LiNEt<sub>2</sub> (1 equiv)<sup>15</sup> followed by application of the standard LDA conditions to afford **21g** (entry 7). Turning attention to substrates bearing electron-withdrawing groups, both tertiary sulfonamide and *tert*-butyl ester systems, **20h** and **20i**, showed good behavior to furnish products **21h** and **21i**, respectively (entries 8 and 9), while cyano- and chloro-substituted derivatives **20j** and **20k** showed instability in the presence of LDA even at low temperatures and did not lead to the rearranged products **21j** and **21k**, respectively (entries 10 and 11). For these two cases, attack at the CN function<sup>16</sup> and benzyne formation<sup>17</sup> may be the complicating factors. In accord with previous experience on similar systems,<sup>18</sup> and consistent with *pK<sub>a</sub>* considerations, benzo-thiophene **20l** required prior in situ 2-TMS protection in order to achieve reasonably effective formation of the  $\alpha$ -carbamoyl migration product **21l**. Reaction of CF<sub>3</sub>-substituted aryl, 2-furanyl, 2-thienyl, and 3- and 4-pyridinyl substituted systems led only to overall decomposition. Treatment of **23** (Scheme 4) under the LDA conditions yielded **24**, perhaps

**Scheme 4**



not a surprising result of carbamoyl migration followed by Michael addition or vice versa, which may be viewed as a product of a first step of an intramolecular Baylis–Hillman reaction.<sup>19</sup> To test the *O*-carbamoyl- $\beta$ -vinyl translocation (Scheme 1, **8**  $\rightarrow$  **10**), the  $\alpha$ -phenyl stilbenes **26a–c** were prepared (Scheme 5). Thus, Suzuki–Miyaura cross coupling of the (*Z*)-stilbene borolane, **25a**,<sup>10</sup> with the previously prepared iodide **15** afforded compound **26a**, while analogous couplings of **27**<sup>20</sup> with **28** gave **29** and then, with 4-Me-OC<sub>6</sub>H<sub>4</sub>B(OH)<sub>2</sub>, the corresponding **26c**. Disilylated stilbene **26b** was obtained by the corresponding coupling of **25b** with **15**. All coupling reactions proceeded with complete stereo-selectivity.<sup>21</sup> Treatment with LDA at room temperature led to the  $\beta$ -vinyl migration products **30a–c**, respectively, in

(15) Roschchangar, F.; Brown, J. C.; Cooley, B. E.; Sharp, M. J.; Matsuoka, R. T. *Tetrahedron* **2002**, *58*, 1657–1666.

(16) Krizan, T. D.; Martin, J. C. *J. Am. Chem. Soc.* **1983**, *105*, 6155–6157.

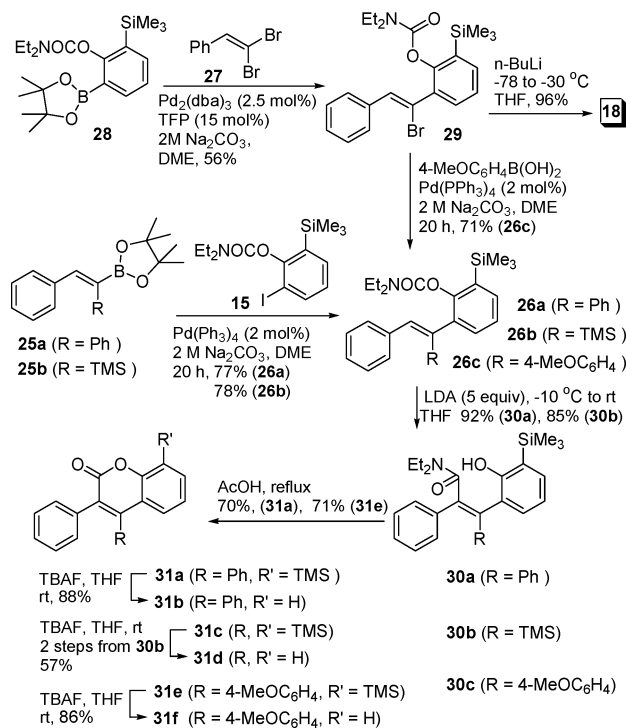
(17) Gohier, F.; Castanet, A.-S.; Mortier, J. *Org. Lett.* **2003**, *5*, 1919–1922.

(18) James, C. Ph.D. Thesis, University of Waterloo, Waterloo, Canada, 1998.

(19) Basavaiah, D.; Rao, A. J.; Satyanarayana, T. *Chem. Rev.* **2003**, *103*, 811–892.

(20) Prepared by the method in: Shen, W. *Synlett* **2000**, 737–739.

Scheme 5



high yields. The disappearance of the vinyl C–H signal in the <sup>1</sup>H NMR discounted the alternative path of remote anionic Fries rearrangement<sup>1a</sup> into the α-phenyl ring. Structural integrity and synthetic utility was shown by conversion, under glacial acetic acid conditions, into compounds **31a**, **31c**, and **31e**, which underwent smooth desilylation to afford the 3-aryl and 3,4-diaryl coumarins **31b**, **31d**, and **31f**, respectively.<sup>22</sup> To obtain evidence for the intermediacy of the α-vinyl lithium species in the *O*-carbamoyl migration, compound **29**, prepared from **28**, was subjected to typical metal–halogen exchange conditions (*n*-BuLi/–78 → –30 °C/THF), which led to the formation of product **18**, shown to be identical with that obtained from the LDA-mediated reaction of **17a**. Although not unequivocal, this result coupled with the observation of β-vinyl migration (**26a–c** → **30a–c**) supports the α- and β-vinyl anionic pathways as reasonable

(21) Stereoselective trans coupling of 1,2-dibromostilbenes is predated; see ref 20 and: Gaukröger, K.; Hadfield, J. A.; Hepworth, L. A.; Lawrence, N. J.; McGown, A. T. *J. Org. Chem.* **2001**, *66*, 8135–8138.

(22) **27d**: Yoneda, E.; Sugioka, T.; Hirao, K.; Zhang, S.-W.; Takahashi, S. *J. Chem. Soc., Perkin Trans. 1* **1998**, *3*, 477–483. **27b**: see ref 3d. **27f**: see Supporting Information.

mechanistic interpretations of the results.<sup>23</sup> Furthermore, the conservation of stereochemistry of the incipient lithio species derived from both **17a** and **26** suggests *O*-carbamate coordinative stabilization, which is unavailable for ortho-unfunctionalized lithiostilbene species and hence may contribute to their configurational instability.<sup>6d,f</sup> Although the conversion of **17b** → **16** implies higher aromaticity over vinyl C–H acidity, comparative *pK<sub>a</sub>* data are not available.<sup>24</sup> A contribution of the CIPE<sup>7</sup> via the *O*-carbamate may also be suggested for the observed anionic rearrangements.

In summary, stereoselective and moderately general oxygen to α- and β-vinyl carbamoyl migration reactions **20** → **21** (Table 1) and **26** → **30** have been demonstrated, which may be of mechanistic interest and which provide new routes to benzofuranones **19** and 3-aryl **31d** and 3,4-diaryl **31b** and **31f** coumarins of current biological interest.<sup>4</sup> Of potential added synthetic value may be the methodologies devised for the efficient construction of precursor aromatics **12** and stilbenes **17**<sup>25</sup> whose contiguous substitution patterns are the trademarks of directed ortho metalation-mediated synthetic strategies. Several aspects of these observations are under further study.

**Acknowledgment.** We are grateful to NSERC Canada for support of our synthetic programs. Dr. Ruiyao Wang provided expert X-ray crystallographic analysis. M.T.C. thanks Pfizer Global Research and Development for a 2004 Pfizer Summer Undergraduate Research Award. Sheldon Lyn, 2003 NSERC Undergraduate Summer Research Award-ee, reproduced a number of experiments in this research with great enthusiasm. Frontier Scientific graciously provided a number of boronic acids.

**Supporting Information Available:** Full spectral data and experimental procedures for the synthesis of compounds **12**, **13**, **15**, **16**, **17a**, **17b**, **18**, **20e**, **21e**, **20i**, **21i**, **23**, **24**, **26a**, and **30a** and X-ray crystallographic data for compound **18**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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(23) Michael addition of LDA followed by carbamoyl transfer and LDA elimination, envisaged only for the α-vinyl rearrangement result; a type of an intramolecular Baylis–Hillman reaction (ref 19) is an alternate, less likely, explanation.

(24) For *O*-aryl carbamate, *pK<sub>a</sub>* = 37.2, see: Fraser, R. R.; Bresse, M.; Mansour, T. S. *J. Am. Chem. Soc.* **1983**, *105*, 7790–7791.

(25) Stereoselective construction of stilbene is of interest in context of natural product, bioactive molecule, and material science substance synthesis; see *inter alia*: Rathore, R.; Deselinicu, M. I.; Burns, C. L. *J. Am. Chem. Soc.* **2002**, *124*, 14832–14833. Jeffery, T.; Ferber, B. *Tetrahedron Lett.* **2003**, *44*, 193–197. Kabalka, G. W.; Wu, Z.; Ju, Y. *Tetrahedron Lett.* **2001**, *42*, 4759–4760.