

Direct Synthesis of Monofunctionalized Indolizine Derivatives Bearing Alkoxymethyl Substituents at C-3 and Their Benzofused Analogues

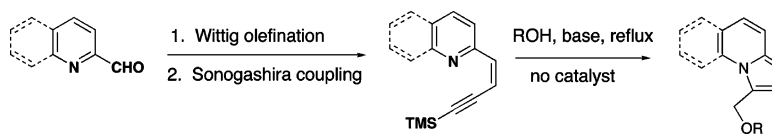
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ABSTRACT



Treatment of (*Z*)-2-pyridine and quinoline silylated vinylacetylenes at reflux with several alcohols in the presence of a suitable inorganic base (KOH, K₂CO₃, CsF, or KF) serendipitously gave 3-alkoxymethylindolizines and the corresponding 1-alkoxymethylpyrrolo [1,2-*a*]quinolines and not the anticipated desilylated vinylacetylene derivatives. A mechanistic possibility for this unexpected chemical transformation is suggested.

Indolizine derivatives (Figure 1) have been the subject of a number of biological and theoretical studies because of their intriguing molecular structures¹ and their therapeutic applications.^{2–5} Among other uses, synthetic and natural indolizine derivatives have considerable potential as new

calcium entry blockers, chemotherapeutics, and cardiovascular agents.^{2–4} In addition, several *O*-containing synthetic

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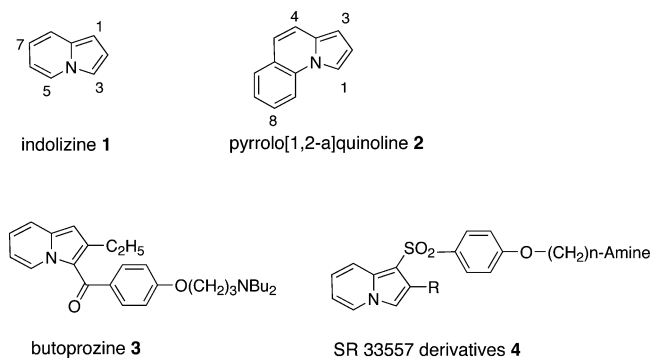


Figure 1. Nuclei of indolizine, benzo[*e*]indolizine, and useful synthetic indolizines.

indolizines have been screened and identified as possessing strong antioxidant effects that prevent the initiation of

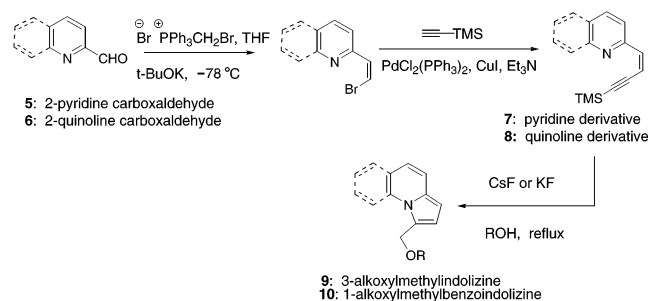
processes that lead to DNA damage.⁵ In this respect, indolizine derivatives represent prime synthetic targets and investigational compounds useful in the development of drugs to treat human diseases such as cancer⁶ and HIV infections.⁷

A number of synthetic approaches to these azabicyclic heterocycles have been reported and reviewed in the literature.^{1,8} Most previous practical syntheses of the indolizine ring system have employed the Scholtz or Tschitschibabin cyclocondensation reactions.^{1,2a,9} In addition to these existing procedures, a number of dipolar cycloaddition reactions of pyridinium ylides with various olefinic and acetylenic precursors appended with electron-withdrawing groups have been developed and proven to be quite valuable in the synthesis of certain indolizine derivatives.^{1,10} However, some of the cycloaddition methods are plagued by low-yielding reactions and regioselectivity problems when unsymmetrical, highly functionalized, and sterically demanding alkynes or olefins are used as substrates.^{1,8–10} Therefore, an alternate method for the preparation of indolizines that allows functional group variation on the indolizine nucleus is highly desirable for structural and biological activity assessments. In this paper, we wish to report an unprecedented cyclization reaction of the silicon-capped (*Z*)-2-pyridine vinylacetylene with various basic alcohol solutions to give 3-alkoxymethyl-substituted indolizines. Likewise, 1-alkoxymethylpyrrolo[1,2-*a*]quinolines (benz[*e*]indolizine) derivatives were successfully obtained from the cyclization of (*Z*)-2-quinoline vinylacetylene.^{11a}

Our laboratory recently described the preparation of

silylated vinylacetylene derivatives **7** and **8** from commercially available 2-pyridine and 2-quinoline carboxaldehydes by means of modified Wittig bromoolefination/Sonogashira coupling procedures.^{11b} The treatment of silylated conjugated acetylenic derivatives with basic alcoholic solutions or fluoride ions has been known to afford the corresponding terminal acetylenes.¹² However, to our surprise, upon the attempted desilylation of (*Z*)-2-pyridine-silylated vinylacetylenes **7** and **8** under standard basic methanol conditions at room temperature, we observed that the desired terminal vinylacetylenes were relatively unstable under the reaction conditions and were converted to 3-methoxymethylindolizine **9a** and 1-methoxymethylpyrrolo[1,2-*a*]quinoline **10b**, respectively (Scheme 1, Tables 1 and 2). The assigned

Scheme 1. Synthesis of 3-Alkoxyethylindolizines and 1-Alkoxyethylbenzo[*e*]indolizines



structures of the products of the reactions are strongly supported by ¹H NMR, ¹³C NMR, and high-resolution mass spectrometry studies. Notably, the ¹H NMR spectra obtained for all of the indolizine and benz[*e*]indolizine products (except the deuterated compound **10a**) listed in Tables 1 and 2 showed a downfield singlet signal between 4.8 and 4.90 ppm consistent with indolizy-type methylene hydrogens bearing an alkoxy group.

Investigations aimed at the optimization of the reaction conditions indicated that the use of KF or CsF along with heating the reaction mixture to reflux in the desired alcohol as solvent resulted in rapid and efficient synthesis of indolizine products. In subsequent reactions, the alcohol was systematically varied to include primary, secondary, tertiary, and cyclic alcohols. In addition, functionalized alcohols such as propargyl, allylic, and benzyl alcohols were used to furnish a library of 3-alkoxyethylindolizines and 1-alkoxyethylbenzo[*e*]indolizine derivatives (Tables 1 and 2). In some cases (Table 1, entries 1–5, and Table 2, entries 1–6), the desired products were simply extracted from the reaction mixture with excellent purity, thereby eliminating the need for further chromatographic purification. It is of interest to note that, when the desilylation reactions were performed at 0 °C (basic methanolic solution or TBAF), the desilylated vinylacetylenes were obtained.^{11b} However, when the reac-

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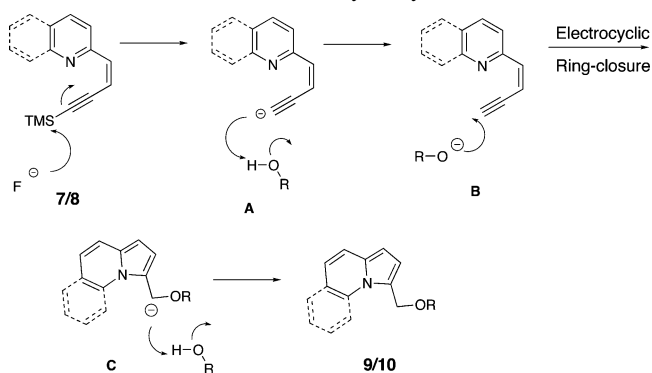
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Table 1. Array of 3-Alkoxyethylindolizines

entry	alcohol	products	base	yield (%)	time (hr)
1	CH ₃ OH		KF	90	1
2	CH ₃ CH ₂ OH		KF	92	1
3			KF	100	0.75
4			KF	94	1.5
5			KF	92	7
6			CsF	37	1
7			CsF	77	2
8			CsF	68	1
9			CsF	82	2

tions were allowed to warm to room temperature, the indolizine products eventually prevailed. It is also worth noting that the treatment of the desilylated vinylacetylenes with various alcohols also yielded the indolizine products in the absence of base or fluoride ions.

The use of high boiling and/or expensive alcohols such as benzyl alcohol and deuterated alcohols as solvent in the transformation was undesirable. Thus, we have demonstrated that such transformations proceeded smoothly using CsF and 10–20 molar equiv of the alcohol in refluxing anhydrous toluene (Table 1, entries 6–9, and Table 2, entries 7–10).

Scheme 2. Proposed Mechanism for Formation of Monosubstituted Alkoxyethylindolizines**Table 2.** Array of 1-Alkoxyethylbenz[e]indolizines

entry	alcohol	products	base	yield (%)	time (hr)
1	CH ₃ OD		KF	87	1
2	CH ₃ OH		KF	95	1
3	CH ₃ CH ₂ OH		KF	96	1
4			KF	100	0.75
5			KF	86	1.5
6			KF	97	7
7			CsF	57	1
8			CsF	51	2
9			CsF	77	1
10			CsF	47	2

The formation of the observed products could be rationalized by postulating an electrocyclic reaction pathway that begins with the formation of the terminal vinylacetylene **A** via protodesilylation of **7** or **8** upon treatment with fluoride ions (Scheme 2). A 1,6 Michael-type conjugate addition of alkoxide at the acetylenic terminus followed by electrocyclic ring-closure involving pyridine nitrogen results in the direct formation of 3-alkoxyethyl-substituted indolizine or benz[e]indolizine derivatives under mild conditions. On the basis of the proposed mechanism, treatment of **8** with KF in refluxing deuterated methanol (CH₃OD) gave the doubly deuterated benzo[e]indolizine **10a** in 87% yield (Table 2).

In terms of synthetic value, the protocol unveiled in this paper has proven to be general and very successful in the preparation of a variety of 3-alkoxyethyl-substituted indolizines and 1-alkoxyethylbenz[e]indolizines in moderate to high yields. These compounds not only constitute a new class of C(3)-hydroxymethyl-protected indolizines and C(1)-benz[e]indolizine analogues, they are also potential building blocks in the synthesis of biologically relevant C(3)-substituted hydroxymethyl indolizidines and C-1 benz[e]indolizidines via suitable protective group removal and traditional hydrogenation methods.

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Supporting Information Available: Experimental procedures and spectra data for the preparation of starting materials, **7**, and **8** in addition to the ^1H NMR and ^{13}C NMR of 3-alkoxymethylindolizines **9a–i** and 1-alkoxymethylben-

zo[e]indolizines **10a–j**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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