

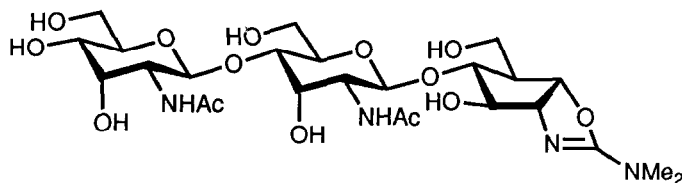
ENANTIOSPECIFIC TOTAL SYNTHESIS OF (-)-ALLOSAMIZOLINE, AN AMINOCYCLITOL MOIETY OF THE INSECT CHITINASE INHIBITOR ALLOSAMIDIN

Masaya Nakata, Seiji Akazawa, Shuji Kitamura, and Kuniaki Tatsuta*

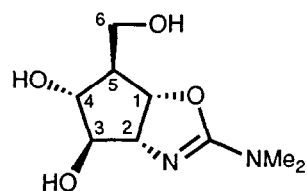
Department of Applied Chemistry, Keio University, Hiyoshi, Kohoku-ku, Yokohama 223, Japan

Summary: Total synthesis of (-)-allosamizoline (2), an aminocyclitol moiety of the insect chitinase inhibitor allosamidin (1) has been enantiospecifically synthesized from D-glucosamine by using an intramolecular cycloaddition of the nitrile oxide to an olefin as a key step.

Allosamidin (1), isolated¹ from the mycelial extract of *Streptomyces* sp. no. 1713, exhibits the inhibitory activity against the chitinases of the silkworm, *Bombyx mori*, *in vitro*, and prevents its larval ecdysis *in vivo*. It has been thought that the chitinase inhibitor would be the good models for insect growth regulators.² Allosamidin (1) consists of 2 equiv. of N-acetyl-D-allosamine, the first example in nature, and 1 equiv. of a new aminocyclitol 2, named allosamizoline.³ The relative configuration of allosamizoline (2) was initially suggested to be the 3,4-cis diol^{3a} and later revised to the 3,4-trans,^{3b} and finally the absolute configuration was elucidated by the exciton chirality method using its 3,4-bis[*p*-(dimethylamino)benzoyl]-6-trityl derivative.^{3c} We wish to describe here the enantiospecific total synthesis⁴ of (-)-allosamizoline (2) by using an intramolecular cycloaddition of the nitrile oxide to an olefin as a key step. By the similar methodology, we have already succeeded in the total syntheses of glycosidase inhibitors, cyclophellitols,⁵ which are the highly oxygenated cyclohexane compounds.⁶

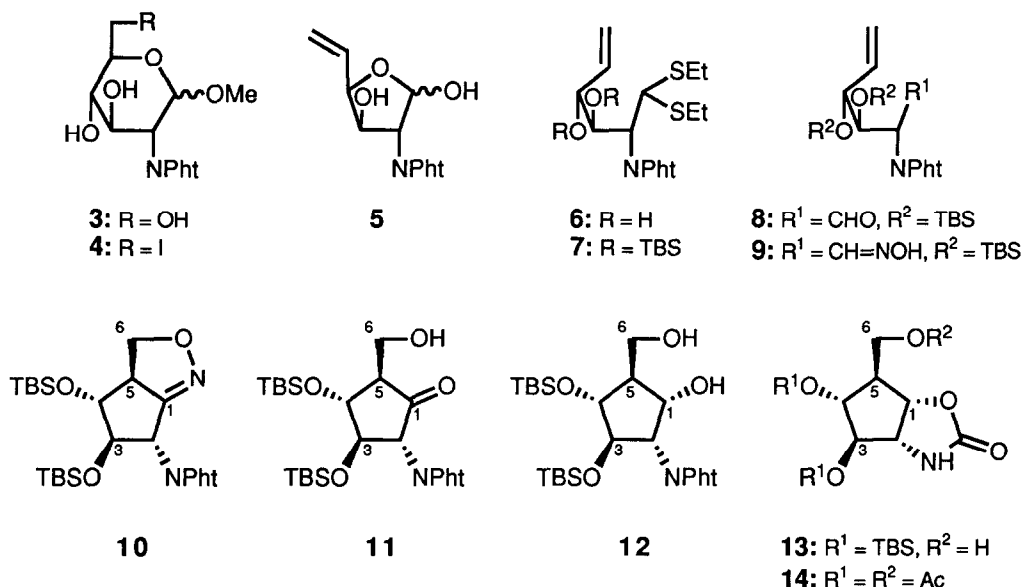


1: Allosamidin



2: Allosamizoline

Our synthesis began with D-glucosamine, which might be the biosynthetic precursor of 2.^{3c} 1,3,4,6-Tetra-O-acetyl-2-deoxy-2-phthalimido-D-glucopyranose, readily prepared from D-glucosamine hydrochloride by the Lemieux procedure,⁷ was treated with 1%



HCl-MeOH at 60°C for 5 d to afford the α and β mixture ($\alpha : \beta = 1 : 3.5$) of methyl glycoside **3**⁸ in 60 % yield from D-glucosamine. Iodination⁹ of **3** with iodine, triphenylphosphine, and imidazole in CH₂Cl₂ at 35°C for 4 d gave the iodide **4**¹⁰ in 90% yield (mp 204 ~ 205°C). Reductive β -elimination of **4** by active zinc¹¹ in THF at 25°C for 1.5 h afforded the 5-enofuranose **5** concomitant with the reductively dehalogenated C-6 product. Treatment of this mixture with ethanethiol and conc. HCl (0°C, 18 h) gave the dithioacetal **6**¹⁰ [61 % from **4**; [α]_D -11° (c 0.77, CHCl₃)] which was silylated with TBSOTf and 2,6-lutidine in CH₂Cl₂ (0°C, 2 h) to give the disilylether **7**¹⁰ [90 %; [α]_D -56° (c 1.04, CHCl₃)]. After dethioacetalization (HgCl₂-CaCO₃, 80 % aq. acetone, 25°C, 12 h) of **7**, the resulting aldehyde **8** was treated with NH₂OH•HCl in pyridine at 25°C for 18 h to afford the oxime **9** (81 % from **7**). Intramolecular cycloaddition⁵ of **9** was best realized by using 0.7 M aq. NaOCl in CH₂Cl₂ at 0°C for 18 h to produce the isoxazoline **10**¹⁰ as a single product [91 %; mp 125.5 ~ 126.5°C; [α]_D -6° (c 0.99, CHCl₃)]. The stereochemistry was assumed considering the results of an intramolecular cycloaddition of glucose-derived nitron to an olefin⁶ and ultimately secured by conversion into **2**.

Isoxazoline opening was a troublesome step. Of a variety of conditions including H₂/Raney Ni-W4/AcOH/aq. dioxane⁵ and H₂/Raney Ni-W4/B(OH)₃/aq. dioxane,¹² ozonolytic cleavage (O₃/O₂, 10 : 1 CH₂Cl₂-MeOH, -78 ~ -30°C, 24 h)¹³ of **10** furnished the β -hydroxy ketone **11**¹⁰ [60 % yield based on the reacted **10**; mp 121.0 ~ 122.5°C; [α]_D -51° (c 0.53, CHCl₃)]. Hydroxyl-directed reduction of **11** with Zn(BH₄)₂ in 1 : 1 THF-ether (0°C, 3 h) afforded a single diol **12**¹⁰ in quantitative yield [mp 102 ~ 103°C; [α]_D +4° (c 0.64, CHCl₃)]. The stereochemistry of this diol **12** was confirmed by its transformation to the cyclic carbamate **13**¹⁰ [60 % overall yield; mp 192 ~ 193°C; [α]_D -7° (c 0.39, CHCl₃)] by de-N-phthaloylation (NH₂NH₂•H₂O, 95 % aq. EtOH, 70°C, 4 h)

and benzyloxycarbonylation (BnOCOCl , Na_2CO_3 , 1 : 2 $\text{H}_2\text{O}-\text{CH}_2\text{Cl}_2$, 0°C , 1.5 h) followed by base treatment (NaH , THF , 25°C , 4 h). Desilylation (1% $\text{HCl}-\text{MeOH}$, 25°C , 2.5 h) of **13** followed by acetylation (Ac_2O , pyridine, 25°C , 18 h) gave the triacetate **14**¹⁰ in quantitative yield $[[\alpha]_D^{25} (c\ 0.41, \text{CHCl}_3)]$. This triacetate **14** was treated with 4 equiv of MeOTf in CH_2Cl_2 ^{4a} at 25°C for 5.5 h and the intermediary iminomethylether was treated with $\text{Me}_2\text{N}\cdot\text{HCl}$ (TEA , CH_2Cl_2 , 25°C , 2 h) in one pot to produce the allosamizoline triacetate. Finally, this was deacetylated with 1M aq. HCl at 50°C for 4.5 h to give (-)-allosamizoline hydrochloride (**2**) in 80 % yield from **14**. The ^1H and ^{13}C NMR spectra and the optical rotation of the synthetic **2** were identical with the authentic data.^{3b} Synthetic studies toward the total synthesis of allosamidin (**1**) are now in progress.

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- 10) All compounds were purified by silica-gel column chromatography and were fully characterized by spectroscopic means and elemental analyses. Optical rotations were measured using a 0.5 dm tube at 25°C. Significant ^1H NMR spectral data (270 MHz, CDCl_3 , δ : TMS = 0 or CHCl_3 = 7.26) are the following. 4: δ (TMS = 0) 3.36 (0.67H, s, α -OMe), 3.47 (2.33H, s, β -OMe), 4.90 (0.22H, d, α -H-1, $J_{1,2}$ = 3.6 Hz), 5.17 (0.78H, d, β -H-1, $J_{1,2}$ = 8.4 Hz). 6: δ (TMS = 0) 1.11 and 1.33 (each 3H, each t, $2\times\text{SCH}_2\text{Me}$, J = 7.6 Hz), 4.65 (1H, d, H-1, $J_{1,2}$ = 11.8 Hz), 5.24 (1H, br d, H-6cis, $J_{5,6}$ = 10.2 Hz), 5.28 (1H, br d, H-6trans, $J_{5,6}$ = 17.0 Hz), 5.85 (1H, ddd, H-5, $J_{4,5}$ = 6.4 Hz). 7: δ (CHCl_3 = 7.26) 0.07, 0.13, 0.14, and 0.21 (each 3H, each s, $4\times\text{SiMe}$), 0.91 and 1.02 (each 9H, each s, $2\times\text{t-Bu}$), 4.64 (1H, ddd, H-6cis, $J_{5,6}$ = 11.4 Hz, J_{gem} = 2.0 Hz), 4.93 (1H, d, H-1, $J_{1,2}$ = 11.0 Hz), 5.12 (1H, ddd, H-6trans, $J_{5,6}$ = 17.8 Hz), 5.89 (1H, ddd, H-5, $J_{4,5}$ = 3.0 Hz). 10: δ (CHCl_3 = 7.26) 3.91 (1H, dd, H-4, $J_{3,4}$ = 9.2 Hz, $J_{4,5}$ = 7.2 Hz), 3.99 (1H, dd, H-6, $J_{5,6}$ = 13.2 Hz, J_{gem} = 7.8 Hz), 4.16 (1H, dddd, H-5, $J_{5,6'}$ = 7.8 Hz, $J_{2,5}$ = 2.0 Hz), 4.69 (1H, dd, H-3, $J_{2,3}$ = 8.0 Hz), 4.78 (1H, dd, H-6'), 5.02 (1H, dd, H-2). 11: δ (CHCl_3 = 7.26) 1.79 (1H, dd, OH, $J_{6,\text{OH}}$ = 4 Hz, $J_{6',\text{OH}}$ = 6.0 Hz), 2.64 (1H, ddd, H-5, $J_{4,5}$ = 10 Hz, $J_{5,6}$ = 4 Hz, $J_{5,6'}$ = 3 Hz), 4.56 (1H, d, H-2, $J_{2,3}$ = 10 Hz). 12: δ (CHCl_3 = 7.26) 1.85 (1H, dd, 6-OH, J = 4 and 6 Hz), 2.39 (1H, m, H-5), 2.89 (1H, d, 1-OH, J = 10 Hz), 4.22 (1H, ddd, H-1, $J_{1,2}$ = 8 Hz, $J_{1,5}$ = 6 Hz), 4.51 (1H, dd, H-2, $J_{2,3}$ = 6 Hz). 13: δ (CHCl_3 = 7.26) 2.49 (1H, br dd, 6-OH), 4.01 (1H, dd, H-2, $J_{1,2}$ = 2 Hz, $J_{2,3}$ = 8 Hz), 5.18 (1H, dd, H-1, $J_{1,5}$ = 8 Hz), 5.35 (1H, br s, NH). 14: δ (TMS = 0) 2.10, 2.11, and 2.13 (each 3H, each s, $3\times\text{OAc}$), 2.63 (1H, dddd, H-5), 3.95 (1H, dd, H-2, $J_{1,2}$ = 9.2 Hz, $J_{2,3}$ = 4.2 Hz), 4.22 (2H, d, $2\times\text{H-6}$, $J_{5,6}$ = $J_{5,6'}$ = 5.0 Hz), 4.76 (1H, dd, H-3, $J_{3,4}$ = 7.8 Hz), 4.87 (1H, dd, H-1, $J_{1,5}$ = 6.0 Hz), 5.25 (1H, dd, H-4, $J_{4,5}$ = 10.0 Hz), 5.77 (1H, br s, NH). 2: δ (D_2O , DOH = 4.80) 2.43 (1H, m, H-5), 3.09 and 3.11 (each 3H, each s, NMe_2), 3.74 (1H, dd, H-6, $J_{5,6}$ = 7 Hz, J_{gem} = 12 Hz), 3.83 (1H, dd, H-4, $J_{3,4}$ = 7 Hz, $J_{4,5}$ = 8 Hz), 3.90 (1H, dd, H-6', $J_{5,6'}$ = 5 Hz), 4.34 (1H, dd, H-2, $J_{1,2}$ = 9 Hz), 5.37 (1H, dd, H-1, $J_{1,5}$ = 5 Hz). ^{13}C NMR spectrum (67.5 MHz, D_2O , dioxane = 67.4) of 2: δ 38.1, 38.3, 52.1, 60.1, 64.3, 75.6, 82.4, 87.4, 161.4.
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