SYNTHESIS OF GLYCOFURANO[2,1-d]IMIDAZOLIDIN-2-ONES

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ABSTRACT

Several 2-(3-arylureido)-2-deoxyglycopyranoses (13–16) have been converted into the corresponding 1-aryl-(1,2-dideoxyglycofurano)[2,1-d]imidazolidin-2-ones (17–20) by acid-catalyzed cyclization. Acetylation of these compounds gave 1-aryl-(per-O-acetyl-1,2-dideoxyglycofurano)[2,1-d]imidazolidin-2-ones (21–24) or 3-acetyl-1-aryl-(per-O-acetyl-1,2-dideoxyglycofurano)[1,2-d]imidazolidin-2-ones (25– 28) depending on the reaction. 3-Ethyl-1-phenyl-(1,2-dideoxy-D-glycero- β -D-taloheptofurano)[2,1-d]imidazolidin-2-one (33) was obtained by the reaction of 2deoxy-2-ethylamino-D-glycero- β -D-talo-heptopyranose with phenyl isocyanate, and converted into the tetra-acetate 34.

INTRODUCTION

The reaction of aminodeoxyaldoses with isothiocyanates has been studied extensively¹⁻³, but the analogous reaction with isocyanates has received less attention. Steudel⁴ proposed the structure 2-hydroxy-1-phenyl-4-D-*arabino*-tetra-hydroxybutylimidazole (1) for the product of reaction of 2-amino-2-deoxy-D-glucose hydrochloride with phenyl isocyanate. Pauly and Ludwig⁵ assigned the structure 4-D-*arabino*-tetrahydroxybutylimidazolin-2-one (2) for the product of condensation of this amino sugar with silver cyanate. Structure **3** was assigned by Heyns and Meinecke⁶ to the product obtained by Steudel, but Micheel and Lengsfeld⁷ obtained this compound in low yield and gave it the structure of 1-phenyl-4,5-(1,2-D-glucopyrano)imidazolidin-2-one (4) when they prepared its triacetate **5**. This structure was used by several authors⁸⁻¹⁰, and Morel⁹ prepared a tetra-acetate to which he assigned structure **6**. The isomeric structure **7** has been synthesized by the reaction of 1-alkyl(aryl)amino-1-deoxy-D-fructose with alkaline or ammonium cyanates¹¹⁻¹³. Structures **8** and **9** were also proposed for one of the degradation products, in an acid medium, of the antitumour antibiotic streptozoto-

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cin¹⁴ (10) and for the antibiotic SF-1993¹⁵ (11), respectively. Later, the glucofurancid structure 12 was established by ¹H-n.m.r. spectroscopy^{16,17} and oxidation with periodate¹⁶, and subsequently confirmed by X-ray crystallography¹⁸. There appear to be no data on the reaction of 2-alkylamino-2-deoxyaldoses with aryl isocyanates.

We now describe the synthesis of several 1-aryl-(1,2-dideoxyglycofurano)-[2,1-d]imidazolidin-2-ones and their acetylated derivatives, and the correction of the structures erroneously assigned in the literature. The first 3-alkyl-1-aryl-(1,2-dideoxyglycofurano)[2,1-d]imidazolidin-2-one is also reported.



RESULTS AND DISCUSSION

Heating solutions of the 2-arylureido-2-deoxy-D-glucopyranoses¹⁹ **13** and **14** in dilute acetic acid gave the corresponding 1-aryl- $(1,2-\text{dideoxy}-\alpha-D-\text{gluco-furano})[2,1-d]$ imidazolidin-2-ones **17** and **18** in high yields. Likewise, the 2-arylureido-2-deoxy-D-glycero-L-gluco-heptopyranoses¹⁹ **15** and **16** gave the 1-aryl- $(1,2-\text{dideoxy}-D-glycero-\alpha-L-gluco-heptofurano)[2,1-d]$ imidazolidin-2-ones **19** and **20**, respectively.

The structures of 17-20 accord with the elemental analyses and spectroscopic

properties, and the structure of **17** has been confirmed by X-ray crystallography¹⁸. The ¹H-n.m.r. signal for NH appears at ~7.5 p.p.m. and the $J_{2',3'}$ value of ~0 Hz shows² the sugar ring to be furanoid (Tables I and II). The ¹³C-n.m.r. spectrum (Table V) confirms these assumptions since C-1' is the most deshielded glycosidic carbon, the signal for C-4' also appears downfield (~79.5 p.p.m.), and C-2' is the most shielded carbon. The signal at ~157 p.p.m., the u.v. absorption at 240 nm, and the i.r. bands at 1700–1660 cm⁻¹, all due to the carbonyl group, do not accord with a tautomeric structure of 2-hydroxyimidazolidine such as 1.

Treatment of **17–20** severally with acetic anhydride and pyridine at <5° gave high yields of the *O*-acetylated derivatives **21–24**, each of which had an i.r. band at ~3300 cm⁻¹ and a signal in the ¹H-n.m.r. spectra at ~6.6 p.p.m. due to the NH group. The chemical shift of the signal for H-1' is practically constant since position 1' is not involved in the acylation; instead, the presence of the 3'-acetate group induces a downfield shift (~1.2 p.p.m.) in the signal for H-3' and, to a lesser extent (~0.2 p.p.m.), in that for H-2'. In the ¹³C-n.m.r. spectra, the 3'-acetate group shifts the C-2' signal upfield (~1.5 p.p.m.; β -effect), and that of C-4' is shifted most (~3.5 p.p.m.) since the β -effect of the 5'-acetate group is added; conversely, the signal for C-3' is shifted downfield (~1.0 p.p.m.; α -effect). The elemental analysis and the weak i.r. band at ~1690 cm⁻¹ demonstrated that **23** crystallised with 1 mol of water^{20,21}.



On acetylation of **21–24** with acetic anhydride–zinc chloride, the heterocyclic nitrogen also reacted, giving the 1-acetyl-3-aryl-(per-*O*-acetyl-1,2-dideoxy-D-glyco-furano)[1,2-*d*]imidazolidin-2-ones **25–28**.

Micheel and Lengsfeld⁷ and García González *et al.*⁸, using the glycopyranoid structure **5**, described a triacetate of **17** but did not determine the structure. However, the physical constants {lit.⁷ m.p. 152°, $[\alpha]_D + 40^\circ$ (ethanol); lit.⁸ m.p. 151°} and the reagent used (acetic anhydride-zinc chloride) indicate the structure to be **25**.

The u.v. spectra of **25–28** are identical to those of **21–24**, but in the i.r. spectra the absorptions characteristic of the acetamido and heterocyclic carbonyl groups coincide. The ¹H-n.m.r. spectra contain three or four singlets ($\delta \sim 2.0$) due to OAc groups and a singlet at $\delta \sim 2.60$ due to the NAc group. The latter chemical shift probably reflects deshielding of the methyl group by the C=O bond of imidazolidin-2-one in the more stable antiperiplanar conformation of the NAc group, analogous to those found in 3-acetyl-2-oxazolidones^{22,23} and in 1-acetyl-3-aryl-(per-*O*-acetyl-1,2-dideoxyglycofurano)[1,2-*d*]imidazolidine-2-thiones². In the ¹³C-n.m.r. spectra, the methyl signal of the NAc group appears at a lower field ($\delta \sim 24$) than that ($\delta \sim 20$) of the OAc groups.

The NAc group deshields (~0.4 p.p.m.) H-2' and H-3' in **25–28** compared to **21–24**, and the signals due to heterocyclic C=O, C-1', and C-3' are shifted to higher field (~6.0, ~3.6, and ~3.0–1.5 p.p.m., respectively) and that of C-2' is shifted downfield (~2.2 p.p.m.). Furthermore, the NAc group has a marked effect on the $[\alpha]_D$ value. Thus, the $[\alpha]_D$ values of **21** and **22** are ~60° greater than those of **25** and **26**, and those of **23** and **24** are ~60° less than those of **27** and **28**.

The structures assigned to 25 and 27 were confirmed² by their synthesis from the thio-analogues 29 and 30. Compounds 29 and 30 were obtained from 31 and 32, the structures of which have been elucidated by X-ray crystallography^{24,25}.

The reaction of 2-deoxy-2-ethylamino-D-glycero- β -D-talo-heptopyranose²⁶ and phenyl isocyanate in an aqueous medium gave 51% of 3-ethyl-1-phenyl-(1,2-dideoxy-D-glycero- β -D-talo-heptofurano)[2,1-d]imidazolidin-2-one (**33**), the structure of which accorded with the elemental analysis and spectroscopic data. Compound **33** consumed 2 mol. equiv. of periodate, indicating that the structure 3-ethyl-1phenyl-4-D-manno-pentahydroxypentylimidazolin-2-one can be discarded, and the formation of 1 mol. of formic acid confirmed the furanoid structure. Treatment of **33** with acetic anhydride-pyridine gave the tetra-acetate **34**, for which the $J_{2',3'}$ value of 5.8 Hz accords with a D-glycero-D-talo configuration, in which H-2',3' are cis²⁶.



EXPERIMENTAL

General methods. — Melting points were determined with a Gallenkamp apparatus and are uncorrected. Optical rotations were measured with a Perkin–Elmer

141 polarimeter. U.v. spectra were recorded with a Pye-Unicam SP8-250 automatic spectrophotometer on solutions in aqueous 96% ethanol, and i.r. spectra (KBr disc) with a Perkin–Elmer 399 spectrophotometer. ¹H-N.m.r. spectra (internal Me₄Si or sodium 2,2-dimethyl-2-silapentane-5-sulfonate) were recorded with a Varian XL-200 (200 MHz), Perkin–Elmer R-32 (90 MHz), or Bruker WP-80-SY (80 MHz) spectrometer. ¹³C-N.m.r. spectra were recorded with a Varian XL-200 (50.2 MHz) or Bruker WP-80-SY (20.15 MHz) spectrometer. P.c. (ascending) was performed on Whatman No. 1 paper, using 1-butanol–pyridine–water (1:1:1) and detection with silver nitrate–sodium hydroxide. T.l.c. was conducted on Silica Gel GF₂₅₄ (Merck), using 3:1 ethyl acetate–ethanol and detection with iodine vapour and u.v. light at 254 nm. The consumption of oxidant and the release of formic acid on periodate oxidation of **33** were determined as described^{27,28}.

1-Phenyl-(1,2-dideoxy- α -D-glucofurano)[2,1-d]imidazolidin-2-one (17). — To a suspension of 2-deoxy-2-(3-phenylureido)-D-glucopyranose¹⁹ (13; 2.3 g, 7.7 mmol) in water (5 mL) was added acetic acid (1 mL), and the mixture was stirred at ~100° for 30 min⁹ and then filtered. The product (1.92 g, 89%) which crystallised on cooling was collected, washed with cold water and acetone, and recrystallised from water to give material having m.p. 214–216°, $[\alpha]_D$ +93°, $[\alpha]_{578}$ +98°, $[\alpha]_{546}$ +113°, $[\alpha]_{436}$ +214.5°, $[\alpha]_{365}$ +395.5° (c 0.5, pyridine); λ_{max} 235 nm (ε_{mM} 15.2); ν_{max} 3400–3300 (OH, NH), 1660 cm⁻¹ (C=O); lit.⁷ m.p. 211°, $[\alpha]_D$ +92° (c 1.25, aqueous 25% *N*,*N*-dimethylformamide). The ¹H- and ¹³C-n.m.r. data are given in Tables I, II, and V.

1-(4-Chlorophenyl)-(1,2-dideoxy-α-D-glucofurano)[2,1-d]imidazolidin-2-one (18). — Obtained, as described for 17, from 2-[3-(4-chlorophenyl)ureido]-2-deoxy-D-glucopyranose^{9,19} (14; 2.0 g, 6.0 mmol) in aqueous 80% ethanol (65 mL) and acetic acid (6 mL), 18 (1.6 g, 85%) had m.p. 217–218° (from ethanol), $[\alpha]_D$ +116.5°, $[\alpha]_{578}$ +123°, $[\alpha]_{546}$ +141.5°, $[\alpha]_{436}$ +268°, $[\alpha]_{365}$ +492.5° (*c* 0.5, pyridine); λ_{max} 243 nm (ε_{mM} 15.3); ν_{max} 3500–3300 (OH, NH), 1670 cm⁻¹ (C=O); lit.⁹ m.p. 221–222°, $[\alpha]_D$ +145.9° (*c* 1, *N*,*N*-dimethylformamide). The ¹H- and ¹³C-n.m.r. data are given in Tables I, II, and V.

1-Phenyl-(1,2-dideoxy-D-glycero- α -L-gluco-*heptofurano)*[2,1-d]*imidazolidin-*2-one (19). — A mixture of 2-deoxy-2-(3-phenylureido)-D-glycero-L-gluco-heptopyranose¹⁹ (15; 1.4 g, 4.3 mmol), water (5 mL), and acetic acid (0.6 mL) was heated at ~100° for 30 min⁹. The solvent was removed under reduced pressure and the residual syrup was crystallised from ethanol. Recrystallisation of the product (1.09 g, 82%) from aqueous 96% ethanol gave 19 as large prisms, m.p. 211–212°, $[\alpha]_D$ -114.5°, $[\alpha]_{578}$ -120.5°, $[\alpha]_{546}$ -139.5°, $[\alpha]_{436}$ -266°, $[\alpha]_{365}$ -496° (c 0.5, pyridine); λ_{max} 236 nm (ϵ_{mM} 13.0); ν_{max} 3540, 3460–3200 (OH, NH), 1705 cm⁻¹ (C=O). The ¹H- and ¹³C-n.m.r. data are given in Tables I, II, and V.

Anal. Calc. for C₁₄H₁₈N₂O₆: C, 54.19; H, 5.85; N, 9.03. Found: C, 54.09; H, 6.06; N, 9.18.

 $1-(4-Chlorophenyl)-(1,2-dideoxy-D-glycero-\alpha-L-gluco-heptofurano)[2,1-d]-imidazolidin-2-one$ (20). — A mixture of 2-[3-(4-chlorophenyl)ureido]-2-deoxy-D-

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17	5.97 d	4.00 d ^c	4.08 dd ^d	\downarrow	.76–3.29 m	Î		7.50 s	7.63-7.05 m	5.30 d	4.74 d	4.491			
18	5.97 d	4.00 d ^c	4.06 dd ^d	₹ 3	.80-3.30 m.			7.62 s	7.65-7.35 de	1 5.33 d	4.75 d	4.50t			
61	5.95 d	4.00 dd	4.06 dd	3.94 dd	3.	86-3.30 m		7.50 s	7.64-7.06 m	5.22 d	4.32 d ^h	4.29 db	4.50 t		
20	5.96 d	4.02 d ^c	$4.07 {\rm m}^{d}$	3.94 dd	3.	87-3.30 m	1	7.61 s	7.67-7.35 de	1 5.24 d	4.34 d ^b	4.30 d ^b	4.501		
33	5.85 d	4.18t		3.97 d	€ 3	48-3.31 m	1		7.70-7.00 m	5.44 d	4.53 n	n (2 H)	4.32 m	3.52 m (1 H)	1.11 t
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TABLE I

Broad singlets. dediteration with $\nu_2 U_{1}$ LYOUDIEL ALLCT deuteration snarp doublet after oc interchangeo. may ^aIn (CD₃)₂SO. ^bAssignments glycero-L-gluco-heptopyranose¹⁹ (16; 2.5 g, 6.7 mmol) in aqueous 70% ethanol (30 mL) and acetic acid (5 mL) was heated at ~100° for 45 min⁹, then concentrated. Treatment of the residual syrup with acetone gave crystalline 20 (1.39 g, 58%). Recrystallisation from water gave material having m.p. 192–194°, $[\alpha]_D$ -66.5°, $[\alpha]_{578}$ -71°, $[\alpha]_{546}$ -82°, $[\alpha]_{436}$ -156°, $[\alpha]_{365}$ -288° (c 1, pyridine); λ_{max} 241 nm (ε_{mM} 18.2); ν_{max} 3560–3100 (OH, NH), 1715, 1680 cm⁻¹ (C=O). The ¹H- and ¹³C- n.m.r. data are given in Tables I, II, and V.

Anal. Calc. for $C_{14}H_{17}ClN_2O_6 \cdot 0.5 H_2O$: C, 47.53; H, 5.13; N, 7.92. Found: C, 47.52; H, 5.19; N, 7.74.

I-Aryl-(per-O-acetyl-1,2-dideoxyglycofurano)[2,1-d]*imidazolidin-2-ones.* — A solution of each 1-aryl-(1,2-dideoxyglycofurano)[2,1-d]*imidazolidin-2-one* (1.5 mmol) in pyridine (4–6 mL) was heated gently and then cooled to 0° , and acetic anhydride was added (3 mL for **17** and **18**, 4.5 mL for **19** and **20**). Each mixture was stored for 10 h at 0° and then poured into ice-water, and the solid product was collected, washed with cold water, and dried *in vacuo*. The following compounds were prepared in this way.

1-Phenyl-(3,5,6-tri-*O*-acetyl-1,2-dideoxy-α-D-glucofurano)[2,1-*d*]imidazolidin-2-one (**21**, 82% from **17**), m.p. 109–110° (from aqueous 96% ethanol), $[\alpha]_D$ +95.5°, $[\alpha]_{578}$ +100°, $[\alpha]_{546}$ +116°, $[\alpha]_{436}$ +210°, $[\alpha]_{365}$ +355.5° (*c* 0.5, chloroform); λ_{max} 235 nm (ε_{mM} 14.3); ν_{max} 3310 (NH), 1745, 1720 (C=O ester), 1705 cm⁻¹ (C=O heterocycle). The ¹H- and ¹³C-n.m.r. data are given in Tables III–V.

Anal. Calc. for C₁₉H₂₂N₂O₈: C, 56.15; H, 5.46; N, 6.89. Found: C, 55.82; H, 5.53; N, 6.79.

1-(4-Chlorophenyl)-(3,5,6-tri-*O*-acetyl-1,2-dideoxy-α-D-glucofurano)[2,1-*d*]imidazolidin-2-one (**22**, 86% from **18**), m.p. 96–98° (from aqueous 96% ethanol), $[\alpha]_{578}$ +99°, $[\alpha]_{546}$ +113.5°, $[\alpha]_{436}$ +206.5°, $[\alpha]_{365}$ +357° (*c* 1, chloroform); λ_{max} 243 nm (ε_{mM} 16.1); ν_{max} 3240 (NH), 1745, 1715 (C=O ester), 1700 cm⁻¹ (C=O heterocycle). The ¹H- and ¹³C-n.m.r. data are given in Tables III–V.

Anal. Calc. for $C_{19}H_{21}ClN_2O_8$: C, 51.77; H, 4.80; N, 6.35. Found: C, 51.50; H, 5.13; N, 6.12.

1-Phenyl-(3,5,6,7-tetra-*O*-acetyl-1,2-dideoxy-D-*glycero-* α -L-*gluco*-heptofurano)[2,1-*d*]imidazolidin-2-one (**23**, 90% from **19**), m.p. 106–108° (from aqueous 50% ethanol), $[\alpha]_D = -69.5^\circ$, $[\alpha]_{578} = -73^\circ$, $[\alpha]_{546} = -84.5^\circ$, $[\alpha]_{436} = -155^\circ$, $[\alpha]_{365} = -272^\circ$ (*c* 0.5, chloroform); λ_{max} 234 nm (ε_{mM} 13.7); ν_{max} 3560–3180 (NH, H₂O), 1740 (C=O ester), 1695 (C=O heterocycle), 1645 cm⁻¹ H₂O. The ¹H- and ¹³C-n.m.r. data are given in Tables III–V.

Anal. Calc. for $C_{22}H_{26}N_2O_{10} \cdot H_2O$: C, 53.22; H, 5.68; N, 5.67. Found: C, 53.47; H, 5.62; N, 5.56.

1-(4-Chlorophenyl)-(3,5,6,7-tetra-*O*-acetyl-1,2-dideoxy-D-glycero-α-L-glucoheptofurano)[2,1-*d*]imidazolidin-2-one (**24**, 71% from **20**), m.p. 76–78° (from aqueous 90% ethanol), $[\alpha]_{578}$ -73.5°, $[\alpha]_{546}$ -76°, $[\alpha]_{436}$ -88.5°, $[\alpha]_{365}$ -287° (*c* 0.5, chloroform); λ_{max} 242 nm (ε_{mM} 17.1); ν_{max} 3440, 3220, 3140 (NH, OH), 1760, 1745 1730 cm⁻¹ (C=O ester and heterocycle). The ¹H- and ¹³C-n.m.r. data are given in Tables III–V.

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¹H-n.m.r. Spin-coupling data^a (Hz) for compounds **17–20** and **33**

Compound	$J_{l',2'}$	$\mathbf{J}_{2',3'}$	$\mathbf{J}_{3',4'}$	J _{4',5'}	$J_{3',OH}$	$J_{S',OH}$	J _{6',0Н}	$J_{\mathcal{T},OH}$	Et	
									\mathbf{J}_{CH_2}	J _{снсн}
17	6.2	0.0	1.8		4.6	5.5	5.6			
18	6.2	0.0^{b}	1.8		4.6	5.5	5.6			
19	6.2	0.0 ^c	2.4	9.5	4.8	3.2	1.8	5.5		
20	6.2	0.0	2.2^d	9.5	4.4	3.6	1.9	5.3		
33	6.0	5.8	8.6	~ 0.0	5.4				-14.1	6.8
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^{*a*}Conditions described in Table I. $^{b}J_{2,NH} \sim 0.8$ Hz, $^{c}J_{2,NH} 0.9$ Hz. ^{d}V alue obtained after addition of D₂O.

TABLE III

¹H-n.m.r. chemical shift data $(\delta)^a$ for compounds **21–28** and **34**

								ĺ							
Com-	,I-H	H-2'	Н-3'	H-4'	H-5'	,9-H	"9-H	Н-7'	Н-7'	ΗN	NAc	Et		Ar	OA_i
punod		į	,					1				NCH_2	CH,		
21 ^b	5.99 d	4.19 d	5.33 d	4.42 dd	5.28 m	4.55 dd	4.16 dđ			6.76 bđ				7.637.16 m (5 H)	2.0
22 th	5.98 d	4.20 d	5.31 d	4.39 dd	5.27 m	4.55 dd	4.16 dd			6.62 bd				7.60–7.29 dd (4 H)	2.1 2.1 2.0
2 3 ⁶	5.99 d	4.18 dd	5.25 d	4.34 dd	5.5	3-5.40 m		4.32 dd	3.94 dd	6.53 d				7.61–7.10 m (5 H)	1.5 2.1 2.0 2.0
24 ^{b,d}	5.95 d	4.20 d	5.24 d	4.33 dd	5.45 dd	5.49 dd		4.33 dd	3.94 dd	6.56 bd				7.58–7.30 dd (4 H)	2 1 1 1 1 1

1.79 s (3 H)							
2.03 s (3 H)							
H) 2.15 s (3 H)	3.04 m (1						
(H) 1.17t(3H) 7.66–7.00m(5H) 2.17s(3H)	3.60 m (1	4.38 dd 4.17 dd	4.25 dd 5.36 dd 5.26 m	t 4.96 dd	4.38 dd-i	5.90 d	Å
1.84 s (3 H)							
2.03 s (3 H)							
2.05 s (3 H)							
7.59–7.36 dd (4 H) 2.13 s (3 H)	2.60 s (3 H)	4.32 dd 3.87 dd	4.15 dd 5.37 dd 5.47 m	5.69 d	4.62 d	5.94 d	28 ⁶
1.80 s (3 H)							
2.00 s (3 H)							
2.04 s (3 H)							
7.70-7.10 m (5 H) 2.12 s (3 H)	2.59 s (3 H)	4.31 dd 3.87 dd	4.17 dd 5.55–5.25 m	5.68 d	4.60 d	5.98 d	27c
1.98 s (3 H)							
2.02 s (3 H)							
7.59–7.35 dd (4 H) 2.12 s (3 H)	2.61 s (3 H)		4.22 dd 5.22 m 4.49 dd 4.14 dd	5.73 d	4.64 d	5.96 d	26 ^b
1.97 s (3 H)							
2.01 s (3 H)							
7.70-7.20 m (5 H) 2.11 s (3 H)	2.61 s (3 H)		4.23 dd 5.22 qd 4.50 dd 4.13 dd	5.73 d	4.63 d	5.98 d	25°

^aIn CDCl₃, ^bAt 200 MHz. ^cAt 80 MHz. ^dOther signals: § 3.72 (q, CH₂) and 1.25 (t, CH₃, J 7.0 Hz) from ethanol of crystallisation.

			,	F	÷	-	_		-	I	Ĺ	
Compound	$\mathbf{J}_{I',2'}$	J _{2',3'}	J _{3',4'}	J4'.5'	J 5',6'	a 5.'0″	6 , 6	0,7	J 6', 7'	1. i c c	E1	
											J _{CH2}	J _{CHCH}
21	6.4	0.0	2.8	9.2	2.2	4.8	-12.4					
52	6.2	0.0	2.6	9.2	2.1	5.1	-12.5					
23	6.4	0.0^{b}	2.8	9.3°	2.0			2.7^{c}	7.0	-11.6		
24	6.4	0.0^{d}	2.8	9.2	2.0				7.1	-11.5		
25	6.6	0.0	2.8	9.4	2.2	4.7	-12.4					
26	0.0	0.0	2.8	9.4	2.2	4.8	-12.2					
27	6.7	0.0	3.0	9.3				4.7	6.9	-11.6		
28	6.6	0.0	3.0	9.4	2.4			4.8	7.1	-11.8		
34	6.6	5.8	9.5	2.1	7.6			2.4	5.0	-12.4	- 14.4	7.0

¹H-n.m.r. spin-coupling data^a (Hz) for compounds 21-28 and 34

TABLE IV

pullou							,	¢							1		VICIN	
pound								5	C-I	C-3,5	C-4	C-2,6	CH_3	со	CH_2	CH_3	CH_3	со
17 a.c	89.76	60.67	74.32	79.47	68.72	63.93		157.34	139.39	128.89	123.06	119.41						
18 4.0	89.51	60.65	74.24	79.54	68.65	63.86		157.09	138.35	128.56	126.33	120.63						
19 a,c	89.69	60.78	74.28	78.86	66.37	70.85	62.61	157.35	139.48	128.62	122.96	119.42						
20a.c	89.46	60.77	74.22	78.94	66.32	70.81	62.57	157.13	138.43	128.51	126.77	120.67						
21 <i>a</i> , <i>d</i>	90.76	59.34	75.30	75.70	67.22	63.05		157.84	137.63	128.77	124.59	120.64	20.59	170.48				
													20.54	169.58				
1000												:	20.48	169.54				
	40.04	40.60	17.01	C8.C/	17.10	CK-70		26./61	136.24	128.80	129.82	121.53	20.58	170.46				
													20.53	169.67				
23 a,d	90.43	59.28	75.21	J.	67.02	69.20	61.70	157.51	137.47	128.77	124.49	120.21	20.82	170.31				
													20.47	169.83				
													20.39	169.53				
													20.15	169.42				
24 a,d,g	90.50	59.49	75.31	75.53	67.17	69.33	61.97	157.52	136.34	128.98	129.93	121.44	20.82	170.56				
													20.68	170.10				
													20.58	169.76				
													20.40	169.60				
25 ^{b,d}	87.16	61.73	73.74	76.25	67.20	63.25		152.00	136.87	129.27	126.46	122.50	20.66	170.31			24.07	168.47
-													20.46	169.72				
26 a.d	86.70	61.26	73.00	75.79	66.57	62.84		151.47	135.02	129.10	131.57	122.99	20.59	170.29			24.08	168.39
													20.50	170.08				
						1							20.44	169.61				
p'aLZ	87.00	61.72	73.52	75.97	67.13	69.76	62.15	151.88	136.82	129.29	126.24	121.88	20.73	170.24			24.07	168.61
													20.55	169.70				
70 <i>a</i> . <i>d</i>	70 70	61 70	12 22	75 40	57 E1	00 07	21 DO	CT 131	00101	10.00	0, 10,		05.02	109.47				
3	00.00	07.10	11.21	01.01	10.00	60.60	06'10	C+.1CI	134.92	CU.421	20.161	10.221	10.02	1/0.27			24.05	168.55
													20.45	CU.U/1				
													01-02 20-22	06.901				
33 a,c	87.78	57.83	72.73¢	81.03	72.18	69.16 ^e	64.42	158.37	139.70	129.60	124.38	120.55			38 70	12.01		
34 a,d	87.14	54.81	72.40	73.82	66.60	69.59	61.71	158.08	137.97	128.71	123.96	119.37	20.50	170.37	38.46	12.11		
													20.42	170.11				
													20.35	169.64				
														169.44				

¹³C-N.M.R. CHEMICAL SHIFT DATA (P.P.M.) FOR COMPOUNDS 17-28, 33 AND 34

TABLE V

Anal. Calc. for C₂₂H₂₅ClN₂O₁₀· EtOH: C, 51.57; H, 5.59; N, 5.01. Found: C, 51.52; H, 5.61; N, 4.99.

I-Acetyl-3-aryl-(per-O-acetyl-1,2-dideoxyglycofurano)[1,2-d]imidazolidin-2ones. — Each 1-aryl-(1,2-dideoxyglycofurano)[2,1-d]imidazolidin-2-one (2.0 mmol) was added to a solution of anhydrous, freshly fused zinc chloride (0.2 g) in acetic anhydride (5.7 mL). The mixture was kept for 48 h at room temperature or for 12 h at 40°, then poured into ice-water. The solid product was collected, washed with cold water, and dried *in vacuo*. The following compounds were prepared in this way.

1-Acetyl-3-phenyl-(3,5,6-tri-*O*-acetyl-1,2-dideoxy-α-D-glucofurano)[1,2-*d*]imidazolidin-2-one (**25**, 90% from **17**), m.p. 153–155° (from ethanol), $[\alpha]_D$ +32.5°, $[\alpha]_{578}$ +34°, $[\alpha]_{546}$ +38.5°, $[\alpha]_{436}$ +65°, $[\alpha]_{365}$ +102.5° (*c* 0.5, chloroform); λ_{max} 238 nm (ε_{mM} 13.3); ν_{max} 1740 (C=O ester), 1700 cm⁻¹ (C=O amide and heterocycle). The ¹H- and ¹³C-n.m.r. data are given in Tables III–V.

Anal. Calc. for $C_{21}H_{24}N_2O_9$: C, 56.25; H, 5.39; N, 6.25. Found: C, 56.19; H, 5.45; N, 6.18.

1-Acetyl-3-(4-chlorophenyl)-(3,5,6-tri-*O*-acetyl-1,2-dideoxy-α-D-glucofurano)[1,2-*d*]imidazolidin-2-one (**26**, 72% from **18**), m.p. 130–132° (from ethanol), $[\alpha]_{\rm D}$ +44°, $[\alpha]_{578}$ +47.5°, $[\alpha]_{546}$ +54°, $[\alpha]_{436}$ +98°, $[\alpha]_{365}$ +168° (*c* 0.5, chloroform); $\lambda_{\rm max}$ 244 nm ($\varepsilon_{\rm mM}$ 17.0); $\nu_{\rm max}$ 1750 (C=O ester), 1690 cm⁻¹ (C=O amide and heterocycle). The ¹H- and ¹³C-n.m.r. data are given in Tables III–V.

Anal. Calc. for $C_{21}H_{23}ClN_2O_9$: C, 52.24; H, 4.80; N, 5.80. Found: C, 52.13; H, 4.77; N, 5.67.

1-Acetyl-3-phenyl-(3,5,6,7-tetra-*O*-acetyl-1,2-dideoxy-D-glycero- α -L-gluco-heptofurano)[1,2-d]imidazolidin-2-one (**27**, 95% from **19**), m.p. 149–150° (from ethanol), $[\alpha]_D -7.5^\circ$ (c 1, chloroform); λ_{max} 239 nm (ε_{mM} 14.9); ν_{max} 1740 (C=O ester), 1690 cm⁻¹ (C=O amide and heterocycle). The ¹H- and ¹³C-n.m.r. data are given in Tables III–V.

Anal. Calc. for C₂₄H₂₈N₂O₁₁: C, 55.38; H, 5.42; N, 5.38. Found: C, 55.73; H, 5.45; N, 5.31.

1-Acetyl-3-(4-chlorophenyl)-(3,5,6,7-tetra-*O*-acetyl-1,2-dideoxy-D-glycero- α -L-gluco-heptofurano)[1,2-d]imidazolidin-2-one (**28**, 88% from **20**), m.p. 160–162° (from ethanol), $[\alpha]_D = 27.5^\circ$, $[\alpha]_{578} = 27.5^\circ$, $[\alpha]_{546} = -32^\circ$, $[\alpha]_{436} = -61.5^\circ$, $[\alpha]_{365} = -118^\circ$ (*c* 0.5, chloroform); λ_{max} 245 nm (ε_{mM} 17.9); ν_{max} 1750 (C=O ester), 1695 cm⁻¹ (C=O amide and heterocycle). The ¹H- and ¹³C-n.m.r. data are given in Tables III–V.

Anal. Calc. for C₂₄H₂₇ClN₂O₁₁: C, 51.95; H, 4.90; N, 5.05. Found: C, 52.21; H, 4.92; N, 4.91.

3-Ethyl-1-phenyl-(1,2-dideoxy-D-glycero- β -D-talo-heptofurano)[2,1-d]imidazolidin-2-one (**33**). — To a solution of 2-deoxy-2-ethylamino-D-glycero- β -D-taloheptopyranose hydrochloride²⁶ (2.74 g, 10.0 mmol) in water (9 mL) was added sodium hydrogencarbonate (0.84 g, 10.0 mmol). The mixture was stirred vigorously, and phenyl isocyanate (1.1 mL, 10.0 mmol) and aqueous 96% ethanol (6 mL) were added. After 3 h, *N*,*N'*-diphenylurea was removed and p.c. of the filtrate showed that a large quantity of amino sugar still remained. The mixture was concentrated to a thick syrup and diluted with ethanol (10 mL), and phenyl isocyanate (1.1 mL) was added. After several hours, the *N*,*N'*-diphenylurea was removed, the filtrate was concentrated, and an aqueous solution of the residue was extracted with ether (2 × 15 mL), then concentrated, and stored for several days at room temperature. A mixture (2.97 g) of **33** and sodium chloride crystallised. Recrystallisation from ethanol gave **33** (1.74 g, 51%). A second recrystallisation from aqueous 96% ethanol gave material having m.p. 184–185°, $[\alpha]_D - 143^\circ$, $[\alpha]_{578} - 150^\circ$, $[\alpha]_{546} - 174^\circ$, $[\alpha]_{436} - 328^\circ$, $[\alpha]_{365} - 600^\circ$ (*c* 0.5, pyridine); λ_{max} 240 nm (ε_{mM} 18.1); ν_{max} 3500–3200 (OH), 1665 cm⁻¹ (C=O heterocycle). The ¹H- and ¹³C-n.m.r. data are given in Tables I, II, and V. Compound **33** consumed 2.1 mol. equiv. of periodate and released 1.0 mol. equiv. of formic acid.

Anal. Calc. for $C_{16}H_{22}N_2O_6$: C, 56.80; H, 6.55; N, 8.28. Found: C, 56.58; H, 6.91; N, 8.44.

3-Ethyl-1-phenyl-(3,5,6,7-tetra-O-acetyl-1,2-dideoxy-D-glycero-β-D-talo-heptofurano)[2,1-d]imidazolidin-2-one (**34**). — Conventional treatment of **33** (0.5 g, 1.5 mmol) with acetic anhydride (3.2 mL) in pyridine (3.0 mL) gave **34** (0.73 g, 97%), m.p. 111–112° (from aqueous 60% ethanol), $[\alpha]_D - 120^\circ$, $[\alpha]_{578} - 126^\circ$, $[\alpha]_{546}$ -144.5°, $[\alpha]_{436} - 259.5^\circ$, $[\alpha]_{365} - 442^\circ$ (c 0.5, chloroform); λ_{max} 236 nm (ε_{mM} 15.8); ν_{max} 1740, 1730 (C=O ester), 1690 cm⁻¹ (C=O heterocycle). The ¹H- and ¹³Cn.m.r. data are given in Tables III–V.

Anal. Calc. for C₂₄H₃₀N₂O₁₀: C, 56.91; H, 5.97; N, 5.53. Found: C, 57.03; H, 5.97; N, 5.62.

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REFERENCES

- 1 F. GARCIA GONZALEZ, J. FERNANDEZ-BOLAÑOS, AND F. J. LOPEZ APARICIO, ACS Symp. Ser., 39 (1976) 207-226.
- 2 M. AVALOS GONZALEZ, J. L. JIMENEZ REQUEJO, J. C. PALACIOS ALBARRAN, M. D. RAMOS MONTERO, AND J. A. GALBIS PEREZ, *Carbohydr. Res.*, 161 (1987) 49-64, and references therein.
- 3 J. A. GALBIS PEREZ, J. L. JIMENEZ REQUEJO, J. C. PALACIOS ALBARRAN, M. AVALOS GONZALEZ. AND J. FERNANDEZ-BOLAÑOS, An. Quim., Ser. C, 82 (1986) 11-17.
- 4 H. STEUDEL, Hoppe-Seyler's Z. Physiol. Chem., 33 (1901) 223; 34 (1902) 352.
- 5 H. PAULY AND E. LUDWIG, Hoppe-Seyler's Z. Physiol. Chem., 121 (1922) 170-176.
- 6 K. HEYNS AND K. H. MEINECKE, Chem. Ber., 86 (1953) 1453-1462.
- 7 F. MICHEEL AND W. LENGSFELD, Chem. Ber., 89 (1956) 1246-1253.
- 8 F. GARCIA GONZALEZ, J. FERNANDEZ-BOLAÑOS, AND A. PANEQUE GUERRERO, Las Ciencias, 29 (1959) 189–197.
- 9 CH. J. MOREL, Helv. Chim. Acta, 44 (1961) 403-412.
- 10 N. D. HEINDEL, D. H. BURNS, T. HONDA, V. R. RISCH, AND L. W. BRADY, Org. Prep. Proced. Int., 7 (1975) 291–296.

- 11 G. HUBER, O. SCHIER, AND J. DRUEY, Helv. Chim. Acta, 43 (1960) 713-717, 1787-1795.
- 12 J. DRUEY AND G. HUBER, Helv. Chim. Acta, 40 (1957) 342-349.
- 13 J. FERNANDEZ-BOLAÑOS AND J. VIGUERA RUBIO, An. Quim., Ser. B, 72 (1976) 991-995.
- 14 R. R. HERR, H. K. JAHNKE, AND A. D. ARGOUDELIS, J. Am. Chem. Soc., 89 (1967) 4808–4809; P. F. WILEY, D. L. MCMICHAEL, J. M. KOERT, AND V. H. WILEY, J. Antibiot., 29 (1976) 1218–1225.
- 15 S. OMOTO, T. SHOMURA, H. SUZUKI, AND S. INOUYE, J. Antibiot., 32 (1979) 436-441.
- 16 H. FRITZ, CH. J. MOREL, AND O. WACKER, Helv. Chim. Acta, 51 (1968) 569-576.
- 17 P. F. WILEY, R. R. HERR, H. K. JAHNKE, C. G. CHIDESTER, S. A. MIZSAK, L. B. SPAULDING, AND A. D. ARGOUDELIS, J. Org. Chem., 44 (1979) 9-16.
- 18 A. CONDE, F. BERNIER, AND R. MARQUEZ, Acta Crystallogr., Sect. B, 36 (1980) 3048-3052.
- 19 M. AVALOS GONZALEZ, P. CINTAS MORENO, J. FUENTES MOTA, I. M. GOMEZ MONTERREY, J. L. JIMENEZ REQUEJO, J. C. PALACIOS ALBARRAN, AND F. REBOLLEDO VICENTE, unpublished results.
- 20 K. NAKAMOTO, Infrared and Raman Spectra of Inorganic and Coordination Compounds, 3rd edn., Wiley, New York, 1978, p. 227.
- 21 K. NAKANISHI AND P. H. SOLOMON, *Infrared Absorption Spectroscopy*, 2nd edn., Holden-Day, San Francisco, 1977, p. 42.
- 22 E. FISCHER, J. Chem. Soc., (1952) 4525-4527.
- 23 C. M. LEE AND W. D. KUMBER, J. Am. Chem. Soc., 83 (1961) 4593-4596; 83 (1961) 4596-4600; 84 (1962) 565-571.
- 24 R. JIMENEZ-GARAY, A. LOPEZ CASTRO, AND R. MARQUEZ, Acta Crystallogr., Sect. B, 32 (1976) 2115-2118; 34 (1978) 1042.
- 25 R. JIMENEZ-GARAY, P. VILLARES, A. LOPEZ CASTRO, AND R. MARQUEZ, Acta Crystallogr., Sect. B, 34 (1978) 184–187.
- 26 J. A. GALBIS PEREZ, J. C. PALACIOS ALBARRAN, J. L. JIMENEZ REQUEJO, M. AVALOS GONZALEZ. AND J. M. FERNANDEZ-BOLAÑOS, *Carbohydr. Res.*, 131 (1984) 71–82.
- 27 F. GARCIA GONZALEZ, J. M. FERNANDEZ-BOLAÑOS. AND M. A. PRADERA DE FUENTES, An. Quim., 70 (1974) 57–59.
- 28 E. L. HIRST AND J. K. N. JONES, J. Chem. Soc., (1949) 1659-1662.