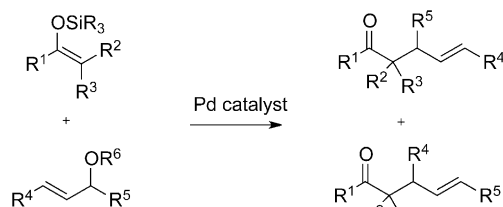


Asymmetric Catalysis

Highly Enantioselective Synthesis of 3-Substituted γ -Butenolides by Palladium-Catalyzed Kinetic Resolution of Unsymmetrical Allyl Acetates**

Bin Mao, Yining Ji, Martín Fañanás-Mastral, Giuseppe Caroli, Auke Meetsma, and Ben L. Feringa*

The Pd-catalyzed asymmetric allylic alkylation (AAA) holds a prominent position among the most versatile methods for C–C bond formation that are widely applied in natural product synthesis.^[1] This transformation typically features broad functional group tolerance and excellent regio- and enantioselectivity.^[2] In particular, Pd-catalyzed kinetic resolutions^[3,4] of symmetrical allylic substrates and dynamic kinetic transformations^[5] have been developed recently, which extends their potential in synthetic chemistry. However, Pd-catalyzed kinetic resolutions of unsymmetrical acyclic allylic substrates with high yield and enantioselectivity are rare.^[6] Furthermore, the enantioselective allylic alkylation of silyl enol ethers is still a challenging reaction.^[7] Graening and Hartwig reported a highly enantioselective alkylation of monosubstituted allylic substrates with silyl enol ethers catalyzed by an Ir complex.^[8] Although excellent results of allylic alkylations have been reported with enolates that are preformed or generated in situ,^[9] to the best of our knowledge, Pd-catalyzed AAA with a nonstabilized silyl enol ether as nucleophile remains an elusive goal (Scheme 1).^[10]

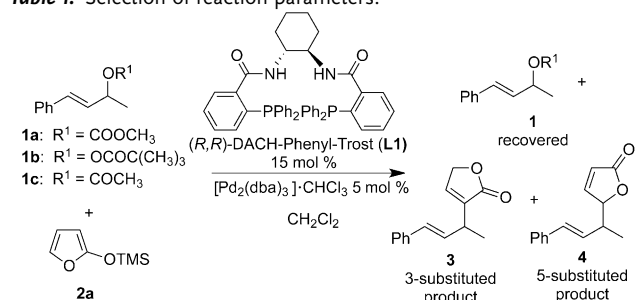


Scheme 1. Pd-catalyzed asymmetric allylic alkylation of unsymmetrical substrates with silyl enol ethers as nucleophiles.

Herein, we present the Pd-catalyzed kinetic resolution of 1,3-disubstituted unsymmetrical allylic substrates with non-stabilized silyl enol ethers as nucleophiles, which provides a highly regio- and enantioselective synthesis of 3-substituted γ -butenolides. The γ -butyrolactone skeleton is present in more than 13000 natural products, which have attracted considerable attention because of the range of important biological activities associated with this class of compounds.^[11] As part of our program to develop catalytic enantioselective methods to access optically active γ -butenolides and lactones,^[12] we envisioned the possibility of using 2-trimethylsilyloxyfuran (TMSOF)^[13] as nucleophile in a Pd-catalyzed AAA reaction.

Initially, we examined the allylic substitution of carbonate **1a** by utilizing TMSOF as nucleophile and a chiral Pd catalyst based on Trost Ligand **L1** (Table 1, entry 1). The 3-substituted product **3** and 5-substituted product **4** were obtained in a 2:1 ratio and in 36% *ee* for product **3** (Table 1, entry 1). Using a lower temperature (0°C), the *ee* value only slightly increased (49% *ee*, Table 1, entry 2). With *tert*-butyl carbonate as the leaving group, the regioselectivity shifted in favor of the formation of the 5-substituted product (**3/4** = 1:2, Table 1,

Table 1: Selection of reaction parameters.



Entry	1	T [°C]	t [h]	Conv. [%] ^[c]	Recovered 1 <i>ee</i> [%] ^[d]	3/4 ^[c]	3 <i>ee</i> [%] ^[d]
1 ^[a]	1a	RT	21	100	–	2:1	36
2 ^[a]	1a	0	25	100	–	2:1	49
3 ^[a]	1b	RT	22	100	–	1:2	60
4 ^[a]	1c	RT	24	80	96(S) ^[e]	5:1	78
5 ^[a]	1c	0	7	53	99(S) ^[e]	> 99:1	99
6 ^[b]	1c	0	7	52	99(S) ^[e]	> 99:1	99

[a] 2 equiv of **2a** used. [b] 1 equiv of **2a** used. [c] The conversion and ratio of regioisomers were determined by GC analysis with *n*-dodecane as the internal standard. [d] Determined by HPLC analysis on a chiral stationary phase. [e] Absolute configuration was assigned by comparison of the sign of optical rotation with published values.

[*] B. Mao, Dr. Y. Ji, Dr. M. Fañanás-Mastral, Dr. G. Caroli, A. Meetsma, Prof. Dr. B. L. Feringa
Stratingh Institute for Chemistry, University of Groningen
Nijenborgh 4, 9747 AG, Groningen (The Netherlands)
E-mail: b.l.feringa@rug.nl

[**] Financial support from The Netherlands Organization for Scientific Research (NWO-CW) is acknowledged. B.M. thanks the China Scholarship Council for financial support (No. 2008618001). Y.J. thanks the MEC for a fellowship. M.F.M. thanks the Spanish Ministry of Science and Innovation (MICINN) for a postdoctoral fellowship. We thank Prof. Dr. A. J. Minnaard for fruitful discussions.

Supporting information for this article is available on the WWW under <http://dx.doi.org/10.1002/anie.201109075>.

entry 3). Remarkably, when the simple allylic acetate **1c** was used, 80% conversion was reached at room temperature and both the regio- and enantioselectivity were significantly enhanced (**3/4** = 5:1, 78% *ee*, Table 1, entry 4).

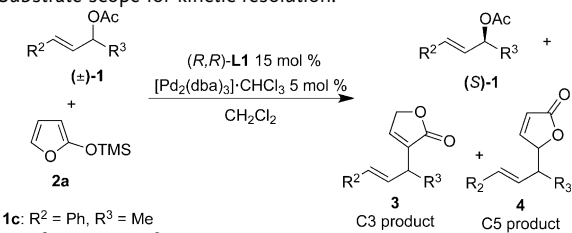
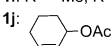
Encouraged by these results, the reaction of allyl acetate **1c** was carried out at 0°C instead of room temperature. In this case, the reaction ceased at 53% conversion but provided enantiomerically pure recovered (*S*)-**1c** through kinetic resolution of the substrate. Furthermore, in this Pd-catalyzed allylic alkylation, **3c** is the only reaction product that is obtained in near-perfect chemo-, regio- (**3/4** > 99:1) and enantioselectivity (99% *ee*). Notably, the reaction could also be carried out with one equivalent of **2a** to give the same results (Table 1, entry 6).^[14]

Remarkably, when the kinetic resolution of **1c** was performed under the optimized conditions shown in Table 1, entry 6, but over a longer reaction time, the transformation virtually ceased at 50% conversion and **1c** was recovered in 99% *ee*, which illustrates a near-perfect selectivity in the kinetic resolution as well (Figure 1). It is also remarkable that the reaction proceeds with complete regioselectivity towards the formation of **3c** (Table 1, entry 5) instead of **4c**, which would be expected based on the common reactivity pattern of TMSOF.^[13] Only very few examples are known in which TMSOF reacts at the C3 position.^[15]

To establish unequivocally the absolute stereochemistry of **3c**, it was converted into the corresponding chromiumtricarbonyl complex by treatment with [Cr(CO)₃(CH₃CN)₃] in THF.^[16] The absolute configuration of **3c** was determined as the *R* configuration by X-ray diffraction analysis on a single crystal of the resulting chromium complex (see the Supporting Information).^[17]

The scope of the reaction was examined under the optimized conditions (Table 1, entry 6) for a range of racemic allylic acetates (Table 2). Generally, most of the products were obtained with excellent *ee* values and very high *S* factors. When unsymmetrical acetates **1d** and **1e** with electron-withdrawing or -donating groups at the *para* position of phenyl ring, respectively, were investigated, excellent regio- and enantioselectivity was maintained (Table 2, entries 2 and 3). The *ortho*-methoxy-substituted substrate **1f** also led to

Table 2: Substrate scope for kinetic resolution.^[a]

									
		1c: R² = Ph, R³ = Me 1d: R² = <i>p</i>-ClC₆H₄, R³ = Me 1e: R² = <i>p</i>-MeOC₆H₄, R³ = Me 1f: R² = <i>o</i>-MeOC₆H₄, R³ = Me 1g: R² = <i>n</i>Pr, R³ = Me 1h: R² = 1-cyclohexyl, R³ = Me 1i: R² = Me, R³ = Me 1j:  1k: R² = Ph, R³ = Ph							
Entry	Conv. ^[b] [%]	1	<i>ee</i> [%] ^[c]	Acetate 1 yield [%] ^[d]	3/4 ^[b]	3	Product 3 <i>ee</i> [%] ^[c]	yield [%] ^[d,e]	<i>S</i> ^[f]
1	52	1c	99(<i>S</i>)	43	> 99:1	3c	99(<i>R</i>) ^[g]	47	116
2	52	1d	99	42	> 99:1	3d	99	47	116
3	53	1e	99	38	> 99:1	3e	99	46	80
4	49	1f	94	44	> 99:1	3f	98	41	> 200
5	48	1g	94	36	22:1	3g	99	39	> 200
6 ^[h]	46	1h	92	44	> 99:1	3h	99	36	87
7	41	1i	91	18	14:1	3i	88	25	25
8	52	1j	99	33	37:1	3j	99	35	116
9	100	1k	—	—	3:1	3k	87	55	—

[a] Reaction conditions: **1/2a**/(*R,R*)-**L1**/[Pd₂(dba)₃]·CHCl₃ (100:100:15:5), **1** in CH₂Cl₂ (0.175 M) at 0°C, 0.5 h for addition of **1** by syringe pump. [b] The conversion and the ratio of regioisomers were determined by GC analysis with *n*-dodecane as the internal standard. [c] Determined by HPLC and GC analyses on chiral stationary phases. [d] Yield of isolated product. [e] No trace of *S*_N2' product was detected. [f] $S = k_{\text{fast}}/k_{\text{slow}} = \ln[(1-C/100)(1-ee/100)]/\ln[(1-C/100)(1+ee/100)]$ (*C* = conversion; *ee* = enantiomeric excess of recovered substrate). [g] The absolute configuration of **3c** was assigned based on the X-ray analysis of a single crystal. [h] The reaction was performed at room temperature with 2 equiv of **2a**.

excellent enantioselectivity both in recovered **1f** and the product **3f** (Table 2, entry 4).

We next turned our attention to dialkyl substituted allylic acetates for the kinetic resolution. Although the recovered allylic acetates had a slightly lower enantiomeric excess than the aryl-substituted allylic acetates, the 3-substituted products **3g** and **3h** were obtained in 99% *ee* and with excellent regioselectivity, both with respect to butenolide and unsymmetrical disubstituted allyl fragments (Table 2, entries 5 and 6). It is important to note that the reaction proceeds very well with both aromatic- and alkyl-substituted unsymmetrical substrates and provides impressive selectivity in both cases.

The allylation of symmetrical, dimethyl-substituted substrate **1i** shows only a slight decrease in enantioselectivity, both for recovered **1i** and the product **3i** (Table 2, entry 7).^[18] Furthermore, treatment of racemic cyclohexenyl acetate **1j** under the optimized conditions gave the corresponding product **3j** almost exclusively with an excellent *ee* value, and **1j** was also recovered in 99% *ee* (Table 2, entry 8). In contrast, racemic 1,3-diphenylallyl acetate **1k** provided the 3-substituted product **3k** with full conversion in 87% *ee* and with poor regioselectivity (Table 2, entry 9). The reaction of

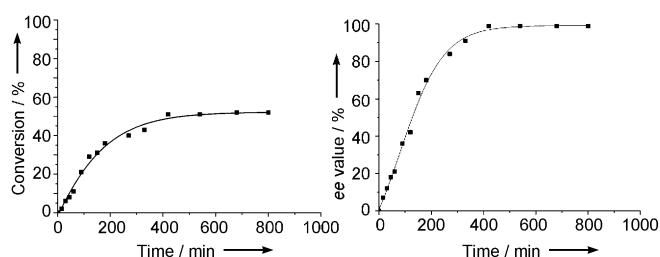
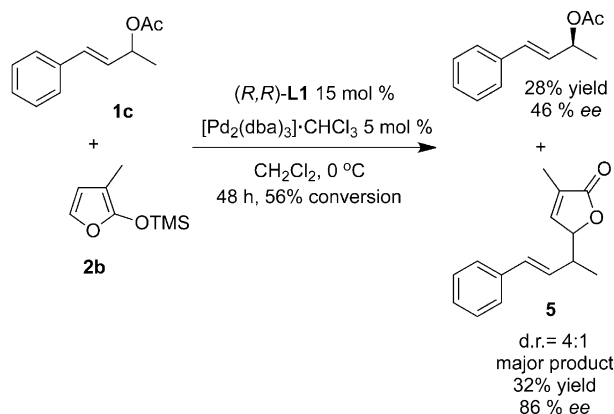


Figure 1. Kinetic resolution of **1c** (see Table 1, entry 6).

this common model substrate for allylic alkylation does not follow the kinetic resolution pathway.^[19]

To gain further insight into the key structural parameters and mechanistic features of the reaction, different methyl-substituted 2-silyloxyfurans were investigated (Schemes 2 and 3). When the 3 position of the furan ring was blocked by a methyl group, the allylation reaction went exclusively through attack at the 5-position, which afforded a 4:1 mixture of diastereomers and the major isomer in 86 % *ee* (Scheme 2).

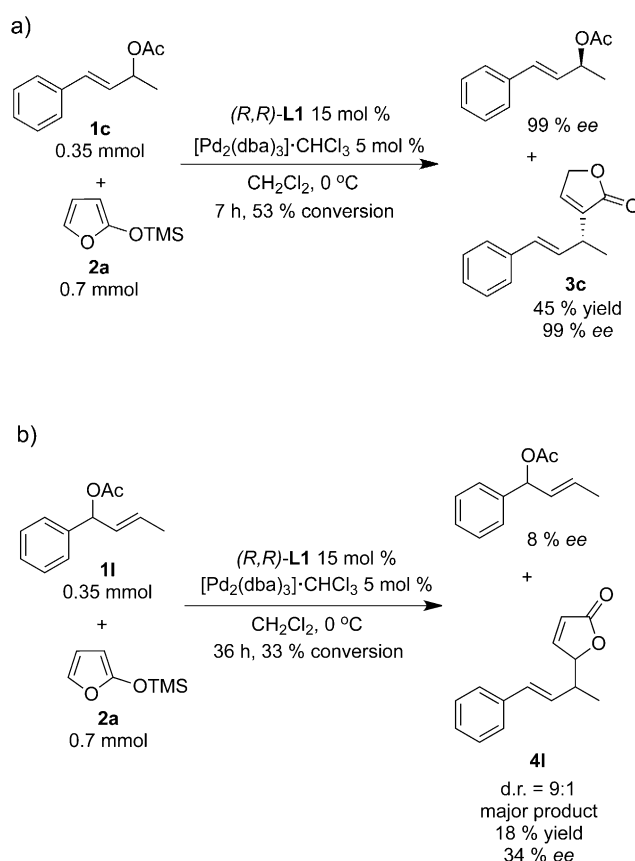


Scheme 2. Allylic alkylation of substituted 2-silyloxyfuran **2b**. TMS = trimethylsilyl.

A moderate enantiomeric excess (46 % *ee*) was obtained for recovered **1c**.

A remarkable shift in selectivity was obtained when we carried out the reaction with allyl acetate **1l**, that is, the regioisomer of **1c** (Scheme 3). Instead of exclusive formation of **3c** in 99 % *ee*, surprisingly, the reaction only provided a 9:1 diastereomeric mixture of **4l**, in which the 3-substituted product was not detected. Furthermore, the recovered starting material was obtained as a nearly racemic mixture.

To further understand the mechanism of the reaction, we carried out model studies and DFT calculations.^[20] Based on the conformation of Pd/(*R,R*)-**L1** reported by Lloyd-Jones and co-workers,^[21] we performed a conformational search on the Pd–olefin complexes of both enantiomers of **1c** (see the Supporting Information). These results indicate that the acetate carbonyl group of enantiomer (*R*)-**1c** forms a hydrogen bond with the amide hydrogen of ligand **L1** on the concave side of catalyst Pd/(*R,R*)-**L1** (Figure 2 a), whereas the same stabilization for enantiomer (*S*)-**1c** is absent (Figure 2 b). The energy difference of about 11 kcal mol^{−1} between these (η^2 -allyl)Pd complexes would explain the result of the kinetic resolution in which the *R* substrate has been completely consumed. A similar result was reported by Lloyd-Jones and coworkers.^[21] In the neutral pre-ionization of η^2 -cyclic ester complexes, they found that the acetate carbonyl group of the *S* enantiomer accepts a hydrogen bond from the amide hydrogen on the concave side of the Pd complex, whereas no corresponding stabilization is available for the *R* enantiomer. This result suggests that only one enantiomer of the acetate can be selectively ionized because of the



Scheme 3. Allylic alkylation of **1c** and **1l** with **2a**.

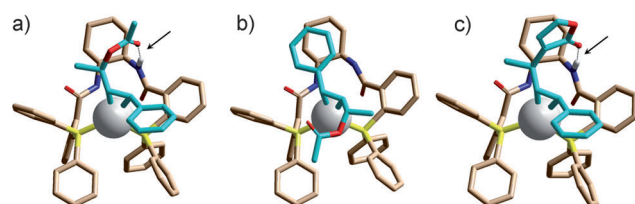
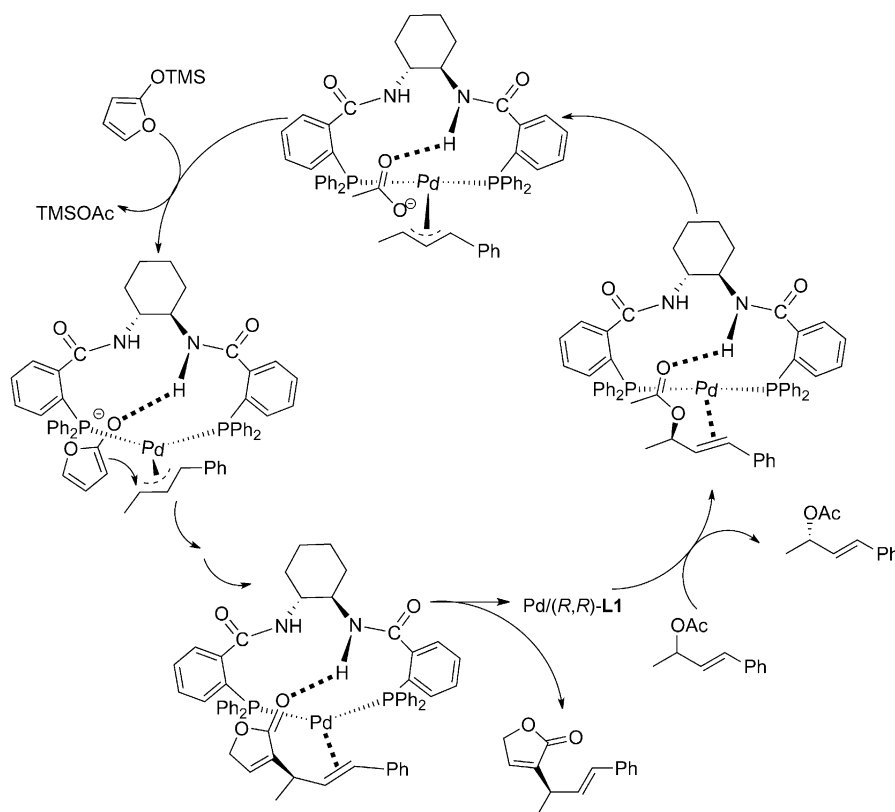


Figure 2. B3LYP structures of a) Pd/(*R,R*)-**L1**/(*R*)-**1c**, b) Pd/(*R,R*)-**L1**/(*S*)-**1c**, and c) the 3-substituted product **3c** coordinated with complex Pd/(*R,R*)-**L1** (arrow indicates hydrogen bond).

stabilization of the leaving acetate anion through hydrogen bonding.

Furthermore, the DFT calculations showed that the most stable product of the reaction is the one that is obtained from allylic alkylation at the 3 position on the furanone ring (which bears a conjugated double bond) with the *R* configuration. We also determined that there is hydrogen bond between the oxygen of the carbonyl group of the furanone product **3c** and the amide hydrogen on the concave side of the Pd complex (Figure 2 c).

It has been described, for the Pd-catalyzed allylic alkylation of malonate derivatives with this particular ligand (*R,R*)-**L1**, that the hydrogen bonding interaction between the enolate oxygen atom and the amide hydrogen atom can direct the enolate carbon atom to the closest enantioface of the η^3 -allyl complex.^[21] We envision that hydrogen bonding (also



Scheme 4. Proposed catalytic cycle.

present in the final product, see Figure 2c) between the oxygen anion of the incoming enolate and the amide hydrogen atom of the catalyst (with which the carbonyl oxygen atom of the acetate leaving group interacts) controls the stereoselectivity of the nucleophilic attack (Scheme 4). That is, the removal of the acetate leaving group and the delivery of the nucleophile proceeds by an identical enantioface selection pathway. The key to the selectivity is a selective acetate–enolate exchange, which suggests that the whole process takes place with net retention of configuration, which is in perfect agreement with the obtained experimental results. This result is in full accordance with the proposal of Lloyd-Jones and co-workers,^[21] in which they identify that the hydrogen-bond interaction of one N–H unit in the Pd-coordinated complex can substantially accelerate both ionization and nucleophilic attack. Moreover, this hydrogen bond pre-oriens the allyl unit for ionization and also directs the orientation of nucleophile delivery.^[22] Most probably, a subsequent olefin isomerization of the butenolide also takes place in the Pd complex, which gives rise to the final product.^[23]

In summary, we have developed a Pd-catalyzed kinetic resolution of 1,3-disubstituted unsymmetrical allylic acetates and a concomitant allylic alkylation by using silyl enol ethers as nucleophiles, to access important 3-substituted γ -butenolides. The reaction proceeds under mild conditions and provides the desired product with excellent chemo-, regio-, and enantioselectivity. Preliminary studies indicate that hydrogen-bonding interactions with the chiral ligand might play a key role in the control of the regio- and enantioselectivity.

Further studies on the unusual reactivity of TMSOF, the mechanistic implications of our findings, and extension of this highly selective catalytic method are ongoing.

Experimental Section

General procedures for Pd-catalyzed kinetic resolution (Table 2): A solution of the ligand (0.0525 mmol, 38 mg) in CH_2Cl_2 (1 mL) was added to a dry Schlenk tube that contained $[\text{Pd}_2(\text{dba})_3]\cdot\text{CHCl}_3$ (0.0175 mmol, 18 mg). After stirring at room temperature for 15 min, a solution of TMSOF (0.35 mmol, 60 μL) in CH_2Cl_2 (0.5 mL) was added dropwise. After the mixture had been stirred for another 15 min, a solution of the allylic acetate (0.35 mmol) and *n*-dodecane (30 μL , internal standard) in CH_2Cl_2 (0.5 mL) were added by syringe pump over 0.5 h at 0°C. The progress of the reaction was monitored by GC and GC-MS. After the completion of the reaction, the reaction solvent was evaporated under vacuum. The crude product was purified by flash chromatography on silica gel by using different mixtures of pentane/ Et_2O as eluents. Toluene/ Et_2O 95:5 was used as the eluent for TLC to distinguish between the regioisomers.

Received: December 22, 2011

Revised: January 17, 2012

Published online: February 16, 2012

Keywords: allyl acetates · γ -butenolides · kinetic resolution · palladium · silyl enol ethers

- [1] For reviews, see: a) B. M. Trost, D. L. Van Vranken, *Chem. Rev.* **1996**, 96, 395–422; b) G. Helmchen, A. Pfaltz, *Acc. Chem. Res.* **2000**, 33, 336–345; c) B. M. Trost, M. L. Crawley, *Chem. Rev.* **2003**, 103, 2921–2943; d) B. M. Trost, *J. Org. Chem.* **2004**, 69, 5813–5837; e) A. Pfaltz, W. J. Drury, *Proc. Natl. Acad. Sci. USA* **2004**, 101, 5723–5726; f) B. M. Trost, M. R. Machacek, A. Aponick, *Acc. Chem. Res.* **2006**, 39, 747–760; g) Z. Lu, S. M. Ma, *Angew. Chem.* **2008**, 120, 264–303; *Angew. Chem. Int. Ed.* **2008**, 47, 258–297.
- [2] For selected examples of Pd-catalyzed AAA, see: a) B. M. Trost, D. L. Van Vranken, *Angew. Chem.* **1992**, 104, 194–196; *Angew. Chem. Int. Ed. Engl.* **1992**, 31, 228–230; b) P. Von Matt, A. Pfaltz, *Angew. Chem.* **1993**, 105, 614–615; *Angew. Chem. Int. Ed. Engl.* **1993**, 32, 566–568; c) H. Steinhagen, M. Reggelin, G. Helmchen, *Angew. Chem.* **1997**, 109, 2199–2202; *Angew. Chem. Int. Ed. Engl.* **1997**, 36, 2108–2110; d) T. Hayashi, M. Kawatsura, Y. Uozumi, *J. Am. Chem. Soc.* **1998**, 120, 1681–1687; e) U. Bremberg, M. Larhed, C. Moberg, A. Hallberg, *J. Org. Chem.* **1999**, 64, 1082–1083; f) S. L. You, X. L. Hou, L. X. Dai, X. Z. Zhu, *Org. Lett.* **2001**, 3, 149–151; g) M. Ogasawara, K. Yoshida, T. Hayashi, *Organometallics* **2001**, 20, 3913–3917; h) T. Hayashi, A. Okada, T. Suzuka, M. Kawatsura, *Org. Lett.* **2003**, 5, 1713–1715; i) N. Oohara, K. Katagiri, T. Imamoto, *Tetrahedron*:

- Asymmetry* **2003**, *14*, 2171–2175; j) G. A. Molander, J. P. Burke, P. J. Carroll, *J. Org. Chem.* **2004**, *69*, 8062–8069; k) J. W. Fallor, J. C. Wilt, J. Parr, *Org. Lett.* **2004**, *6*, 1301–1304; l) H. Yorimitsu, K. Oshima, *Angew. Chem.* **2005**, *117*, 4509–4513; *Angew. Chem. Int. Ed.* **2005**, *44*, 4435–4439; m) B. M. Trost, A. Aponick, *J. Am. Chem. Soc.* **2006**, *128*, 3931–3933; n) R. Shintani, W.-L. Duan, T. Hayashi, *J. Am. Chem. Soc.* **2006**, *128*, 5628–5629; o) T. Imamoto, M. Nishimura, A. Koide, K. Yoshida, *J. Org. Chem.* **2007**, *72*, 7413–7416; p) B. M. Trost, D. A. Thaisrivongs, *J. Am. Chem. Soc.* **2008**, *130*, 14092–14093; q) W. Liu, D. Chen, X. Z. Zhu, X. L. Wan, X. L. Hou, *J. Am. Chem. Soc.* **2009**, *131*, 8734–8735; r) X. Caldentey, M. A. Pericas, *J. Org. Chem.* **2010**, *75*, 2628–2644.
- [3] For general reviews on kinetic resolution, see: a) H. B. Kagan, J. C. Fiaud, *Top. Stereochem.* **1988**, *18*, 249–330; b) G. R. Cook, *Curr. Org. Chem.* **2000**, *4*, 869–885; c) J. M. Keith, J. F. Larrow, E. N. Jacobsen, *Adv. Synth. Catal.* **2001**, *343*, 5–26; d) D. E. J. E. Robinson, S. D. Bull, *Tetrahedron: Asymmetry* **2003**, *14*, 1407–1446; e) E. Vedejs, M. Jure, *Angew. Chem.* **2005**, *117*, 4040–4069; *Angew. Chem. Int. Ed.* **2005**, *44*, 3974–4001.
- [4] For selected examples of Pd-catalyzed kinetic resolution, see: a) S. Ramdeehul, P. Dierkes, R. Aguado, P. C. J. Kamer, P. W. N. M. van Leeuwen, J. A. Osborn, *Angew. Chem.* **1998**, *110*, 3302–3304; *Angew. Chem. Int. Ed.* **1998**, *37*, 3118–3121; b) H. J. Gais, N. Spalthoff, T. Jagusch, M. Frank, G. Raabe, *Tetrahedron Lett.* **2000**, *41*, 3809–3812; c) B. Dominguez, N. S. Hodnett, G. C. Lloyd-Jones, *Angew. Chem.* **2001**, *113*, 4419–4421; *Angew. Chem. Int. Ed.* **2001**, *40*, 4289–4291; d) H. J. Gais, T. Jagusch, N. Spalthoff, F. Gerhards, M. Frank, G. Raabe, *Chem. Eur. J.* **2003**, *9*, 4202–4221; e) B. J. Lüssem, H. J. Gais, *J. Am. Chem. Soc.* **2003**, *125*, 6066–6067; f) B. L. Lei, C. H. Ding, X. F. Yang, X. L. Wan, X. L. Hou, *J. Am. Chem. Soc.* **2009**, *131*, 18250–18251; g) R. K. Rao, G. Sekar, *Tetrahedron: Asymmetry* **2011**, *22*, 948–954.
- [5] For general reviews on dynamic kinetic resolutions, see: a) B. M. Trost, D. R. Fandrick, *Aldrichimica Acta* **2007**, *40*, 59–72. For selected examples of dynamic kinetic resolution, see: b) G. R. Cook, P. S. Shanker, K. Pararajasingham, *Angew. Chem.* **1999**, *111*, 226–228; *Angew. Chem. Int. Ed.* **1999**, *38*, 110–113; c) B. M. Trost, F. D. Toste, *J. Am. Chem. Soc.* **1999**, *121*, 3543–3544; d) B. M. Trost, J. Dudash, E. J. Hembre, *Chem. Eur. J.* **2001**, *7*, 1619–1629; e) B. M. Trost, B. S. Brown, E. J. McEachern, O. Kuhn, *Chem. Eur. J.* **2003**, *9*, 4442–4451; f) A. Parvulescu, D. De Vos, P. Jacobs, *Chem. Commun.* **2005**, 5307–5309; g) M. J. Kim, W. H. Kim, K. Han, Y. K. Choi, J. Park, *Org. Lett.* **2007**, *9*, 1157–1159; h) A. N. Parvulescu, P. A. Jacobs, D. E. De Vos, *Chem. Eur. J.* **2007**, *13*, 2034–2043; i) I. Mangion, N. Strotman, M. Drahl, J. Imbriglio, E. Guidry, *Org. Lett.* **2009**, *11*, 3258–3260.
- [6] For Pd-catalyzed kinetic resolution with unsymmetrical allylic substrates, see: a) Y. K. Choi, J. H. Suh, D. Lee, I. T. Lim, J. Y. Jung, M. J. Kim, *J. Org. Chem.* **1999**, *64*, 8423–8424; b) H. J. Gais, O. Bondarev, R. Hetzer, *Tetrahedron Lett.* **2005**, *46*, 6279–6283.
- [7] The undesired side reactions as well as insufficient regioselectivity and diastereoselectivity have kept the Pd-catalyzed allylic alkylation of silyl enol ethers and silyl ketenes from having been developed further, see: M. Braun, T. Meier, *Synlett* **2006**, 661–676.
- [8] For Ir-catalyzed enantioselective allylation of silyl enol ethers, see: T. Graening, J. F. Hartwig, *J. Am. Chem. Soc.* **2005**, *127*, 17192–17193.
- [9] For reviews of Pd-catalyzed allylic alkylations with enolates, see: a) M. Braun, T. Meier, *Angew. Chem.* **2006**, *118*, 7106–7109; *Angew. Chem. Int. Ed.* **2006**, *45*, 6952–6955. For selected examples, see: b) B. M. Trost, G. M. Schroeder, *J. Am. Chem. Soc.* **1999**, *121*, 6759–6760; c) M. Braun, F. Laicher, T. Meier, *Angew. Chem.* **2000**, *112*, 3637–3640; *Angew. Chem. Int. Ed.* **2000**, *39*, 3494–3497; d) D. C. Behenna, B. M. Stoltz, *J. Am. Chem. Soc.* **2004**, *126*, 15044–15045; e) J. T. Mohr, D. C. Behenna, A. M. Harned, B. M. Stoltz, *Angew. Chem.* **2005**, *117*, 7084–7087; *Angew. Chem. Int. Ed.* **2005**, *44*, 6924–6927; f) X. X. Yan, C. G. Liang, Y. Zhang, W. Hong, B. X. Cao, L. X. Dai, X. L. Hou, *Angew. Chem.* **2005**, *117*, 6702–6704; *Angew. Chem. Int. Ed.* **2005**, *44*, 6544–6546; g) B. M. Trost, J. Y. Xu, *J. Am. Chem. Soc.* **2005**, *127*, 17180–17181; h) M. Braun, T. Meier, F. Laicher, P. Meletis, M. Fidan, *Adv. Synth. Catal.* **2008**, *350*, 303–314; i) P. Meletis, M. Patil, W. Thiel, M. Braun, *Chem. Eur. J.* **2011**, *17*, 11243–11249; j) D. C. Behenna, J. T. Mohr, N. H. Sherden, S. C. Marinescu, A. M. Harned, K. Tani, M. Seto, S. Ma, Z. Novak, M. R. Kroust, R. M. McFadden, J. L. Roizen, J. A. Enquist, Jr., D. E. White, S. R. Levine, K. V. Petrova, A. Iwashita, S. C. Virgil, B. M. Stoltz, *Chem. Eur. J.* **2011**, *17*, 14199–14223.
- [10] For examples of Pd-catalyzed non-asymmetric allylic alkylation of silyl enol ethers, see: a) B. M. Trost, E. Keinan, *Tetrahedron Lett.* **1980**, *21*, 2591–2594; b) C. Carfagna, L. Mariani, A. Musco, G. Salles, R. Santi, *J. Org. Chem.* **1991**, *56*, 3924–3927.
- [11] For reviews on γ -butyrolactones, see: a) R. Bandichhor, B. Nosse, O. Reiser, *Top. Curr. Chem.* **2004**, *243*, 43–72; b) R. R. A. Kitson, A. Millemaggi, R. J. K. Taylor, *Angew. Chem.* **2009**, *121*, 9590–9615; *Angew. Chem. Int. Ed.* **2009**, *48*, 9426–9451.
- [12] For selected examples of syntheses of optically active γ -butenolides and lactones, see: a) H. van der Deen, A. D. Cuiper, R. P. Hof, A. van Oeveren, B. L. Feringa, R. M. Kellogg, *J. Am. Chem. Soc.* **1996**, *118*, 3801–3803; b) K. Geurts, S. P. Fletcher, B. L. Feringa, *J. Am. Chem. Soc.* **2006**, *128*, 15572–15573; c) B. Mao, K. Geurts, M. Fañanás-Mastral, A. W. van Zijl, S. P. Fletcher, A. J. Minnaard, B. L. Feringa, *Org. Lett.* **2011**, *13*, 948–951. For selected examples from other groups, see: d) D. A. Evans, M. C. Kozlowski, J. A. Murry, C. S. Burgey, K. R. Campos, B. T. Connell, R. J. Staples, *J. Am. Chem. Soc.* **1999**, *121*, 669–685; e) B. M. Trost, F. D. Toste, *J. Am. Chem. Soc.* **2003**, *125*, 3090–3100; f) E. L. Carswell, M. L. Snapper, A. H. Hoveyda, *Angew. Chem.* **2006**, *118*, 7388–7391; *Angew. Chem. Int. Ed.* **2006**, *45*, 7230–7233; g) H. Ube, N. Shimada, M. Terada, *Angew. Chem.* **2010**, *122*, 1902–1905; *Angew. Chem. Int. Ed.* **2010**, *49*, 1858–1861; h) J. A. Birrell, J. N. Desrosiers, E. N. Jacobsen, *J. Am. Chem. Soc.* **2011**, *133*, 13872–13875.
- [13] For selected examples of the use of TMSOF as a nucleophile to build γ -butenolides, see: a) S. P. Brown, N. C. Goodwin, D. W. C. MacMillan, *J. Am. Chem. Soc.* **2003**, *125*, 1192–1194; b) R. K. Boeckman, J. E. Pero, D. J. Boehmler, *J. Am. Chem. Soc.* **2006**, *128*, 11032–11033; c) Y. Q. Jiang, Y. L. Shi, M. Shi, *J. Am. Chem. Soc.* **2008**, *130*, 7202–7203; d) V. Liautard, V. Desvergnès, K. Itoh, H. W. Liu, O. R. Martin, *J. Org. Chem.* **2008**, *73*, 3103–3115; e) L. C. Wieland, E. M. Vieira, M. L. Snapper, A. H. Hoveyda, *J. Am. Chem. Soc.* **2009**, *131*, 570–576; f) R. P. Singh, B. M. Foxman, L. Deng, *J. Am. Chem. Soc.* **2010**, *132*, 9558–9560.
- [14] When ligand (S,S)-**L1** was used, the recovered acetate and product were obtained with opposite configurations with similar conversion and *ee* values.
- [15] a) M. A. Loreto, L. Pellacani, P. A. Tardella, *Tetrahedron Lett.* **1989**, *30*, 5025–5028; b) H. Fujioka, N. Matsunaga, H. Kitagawa, Y. Nagatomi, M. Kondo, Y. Kita, *Tetrahedron: Asymmetry* **1995**, *6*, 2117–2120.
- [16] J. Pan, J. Wang, M. M. B. Holl, J. W. Kampf, A. J. Ashe, *Organometallics* **2006**, *25*, 3463–3467.
- [17] CCDC 857213 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

- [18] The yield was low probably because of the high volatility of the recovered allylic acetate and the product.
- [19] This result was in accordance with the finding that sterically more encumbered substrates often gave lower enantioselectivity with the Pd-**L1** catalyst: G. Helmchen, U. Kazmaier, S. Forster, *Catalytic Asymmetric Synthesis*, 3rd ed. (Ed.: I. Ojima), Wiley-VCH, New York, **2010**, pp. 497–641.
- [20] a) C. T. Lee, W. T. Yang, R. G. Parr, *Phys. Rev. B* **1988**, *37*, 785–789; b) A. D. Becke, *J. Chem. Phys.* **1993**, *98*, 5648–5652; c) P. J. Stephens, F. J. Devlin, C. F. Chabalowski, M. J. Frisch, *J. Phys. Chem.* **1994**, *98*, 11623–11627.
- [21] C. P. Butts, E. Filali, G. C. Lloyd-Jones, P.-O. Norrby, D. A. Sale, Y. Schramm, *J. Am. Chem. Soc.* **2009**, *131*, 9945–9957.
- [22] For selected examples in which a ligand may direct regioselective allylic substitution by hydrogen-bond donation to the nucleophile, see: a) C. Wang, N. Pahadi, J. A. Tunge, *Tetrahedron* **2009**, *65*, 5102–5109; b) G. S. Mahadik, S. A. Knott, L. F. Szczepura, S. J. Peters, J. M. Standard, S. R. Hitchcock, *J. Org. Chem.* **2009**, *74*, 8164–8173; c) M. H. Katcher, A. Sha, A. G. Doyle, *J. Am. Chem. Soc.* **2011**, *133*, 15902–15905.
- [23] It is still not clear when the double bond migration take place. Isomerization of butenolides promoted by hydrogen bonding has been described, see: Y. Wu, R. P. Singh, L. Deng, *J. Am. Chem. Soc.* **2011**, *133*, 12458–12461.
-