

Specificity of the Reaction of 2,3-Dichloro-4,4-dimethoxy-5-(2-methylfuran-3-yl)cyclopent-2-en-1-one with Amines

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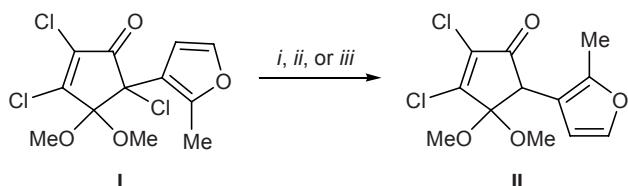
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Abstract—2,3-Dichloro-4,4-dimethoxy-5-(2-methylfuran-3-yl)cyclopent-2-en-1-one reacted with diethyl- and dipropylamines to give products of $\text{Ad}_\text{N}\text{E}$ replacement of the chlorine atom at the vinylic C^3 atom and substitutive opening of the furan ring with simultaneous deprotection of the dimethyl acetal moiety in the 2,3-dichlorocyclopentenone fragment.

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While attempting to generate anionoid intermediates from the furan fragment of trichlorocyclopentenone **I** [1] by the action of lithium diisopropylamide (tetrahydrofuran, -78°C) we revealed smooth selective reductive dechlorination at the $\text{C}^5\text{-Cl}$ bond with formation of dichlorocyclopentenone **II** (Scheme 1). The same compound was obtained later by treatment of trichlorocyclopentenone **I** with CrCl_2 [2], as well as by the action of zinc in methanol in the presence of ammonium chloride (in a poor yield).

Scheme 1.



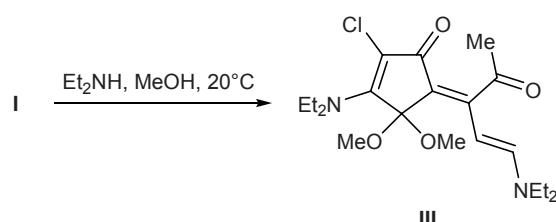
i: $(i\text{-Pr})_2\text{NLi}$, THF, -78°C (yield 60%); *ii*: CrCl_2 , Me_2CO (68%); *iii*: $\text{Zn}, \text{NH}_4\text{Cl}$, MeOH (20%).

We previously described reactions of trichlorocyclopentenone **I** with secondary amines, such as diethylamine, morpholine, pyrrolidine, dipropylamine, etc. These reactions afforded unusual products of substitutive opening of the furan ring. For example, compound **III** was formed in the reaction of **I** with diethylamine [3] (Scheme 2).

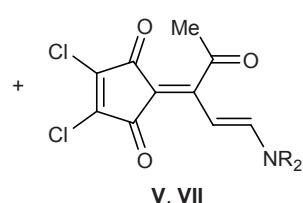
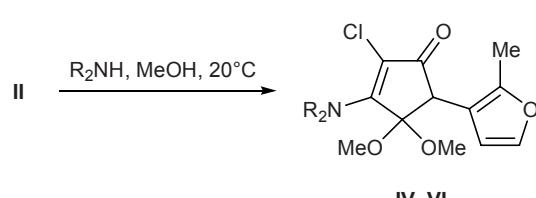
Dichloride **II** obtained from trichlorocyclopentenone **I** was also brought into reactions with amines. However, the behavior of **II** in the reaction with di-

ethylamine differed from the behavior of trichloro derivative **I**. We isolated two main products at a ratio of $\sim 2:1$. On the basis of spectral data, they were assigned structures **IV** and **V** (minor product). Likewise, the reaction of **II** with dipropylamine gave a mixture of compounds **VI** and **VII** (Scheme 3).

Scheme 2.



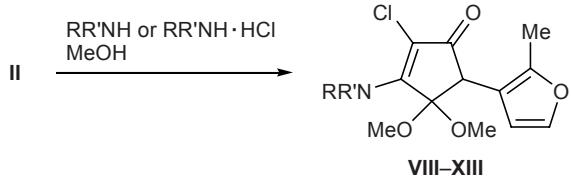
Scheme 3.



IV, V, R = Et; VI, VII, R = Pr.

On the other hand, no cross-conjugated triketones like **V** and **VII** were formed in the reactions of dichlorocyclopentenone **II** with morpholine, pyrrolidine, and dimethylamine. In these cases, we isolated in good yields the expected $\text{Ad}_\text{N}E$ replacement products [4] at the $\text{C}^3\text{-Cl}$ bond, i.e., cyclic vinylogous amides **VIII**, **IX**, and **XII** that are structurally related to compounds **IV** and **VI**. Furthermore, unlike trichlorocyclopentenone **I**, dichloro ketone **II** reacted fairly smoothly with primary amines. Its reactions with benzylamine, α -methylbenzylamine, and methylamine led to the formation of products **X**, **XI**, and **XIII** as a result of replacement of the chlorine atom on C^3 (Scheme 4).

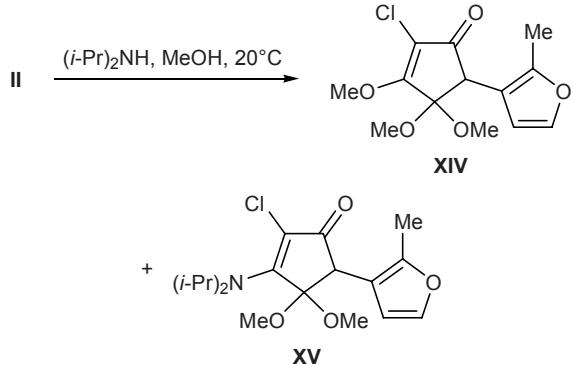
Scheme 4.



VIII, $\text{RR}'\text{N}$ = morpholino; **IX**, $\text{RR}' = (\text{CH}_2)_4$; **X**, $\text{R} = \text{H}$, $\text{R}' = \text{PhCH}_2$; **XI**, $\text{R} = \text{H}$, $\text{R}' = \text{PhCH}(\text{Me})$; **XII**, $\text{R} = \text{R}' = \text{Me}$; **XIII**, $\text{R} = \text{H}$, $\text{R}' = \text{Me}$.

Unexpectedly, the major product formed in the reaction of **II** with diisopropylamine in methanol was trimethoxy derivative **XIV** which was isolated as a mixture with expected 3-substituted ketone **XV** at a ratio of ~3:1 (according to the ^1H NMR data). Presumably, the reaction of less active dichloro derivative **II** with methanol to give compound **XIV** competes with the reaction with sterically loaded diisopropylamine (Scheme 5).

Scheme 5.



Among the examined reactions of dichloro ketone **II** with diethyl- and dipropylamines, the formation of cyclopentene triketones **V** and **VII** attracts undoubted interest. Presumably, the reaction direction leading to

triketones **V** and **VII** is determined by the following factors: first, diethyl- and dipropylamines can be regarded as somewhat stronger and less sterically hindered nucleophiles than the other amines involved; second, structural specificity of compound **II** related to the absence of chlorine on C^5 (unlike trichloro ketone **I**) makes it capable of undergoing enolization during the process.

EXPERIMENTAL

The IR spectra were recorded on Specord M-80 and UR-20 spectrophotometers from samples prepared as thin films (neat) or dispersed in Nujol. The NMR spectra were measured on a Bruker AM-300 instrument at 300 MHz for ^1H and 75.47 MHz for ^{13}C using CDCl_3 as solvent and tetramethylsilane as internal reference. The mass spectra (electron impact, 70 eV) were obtained on Shimadzu LCMS-2010 and Thermo Finnigan MAT 95XP spectrometers; ion source temperature 200°C , batch inlet probe temperature $5\text{--}270^\circ\text{C}$, temperature ramp 22 deg/min. Thin-layer chromatography was performed on Silufol and Sorbfil plates; spots were detected by treatment with iodine vapor or by spraying with a solution of *p*-methoxybenzaldehyde and sulfuric acid in ethanol and subsequent heating at $120\text{--}150^\circ\text{C}$. The products were isolated by column chromatography on silica gel L (200–280 μm , Russia) using 30–60 g of the sorbent per gram of substrate; freshly distilled solvents were used as eluents. Liquid amines were purified by drying over powdered potassium hydroxide, followed by distillation.

2,3-Dichloro-4,4-dimethoxy-5-(2-methylfuran-3-yl)cyclopent-2-en-1-one (II). Compound **I**, 0.28 g (0.86 mmol), was dissolved in 10 ml of acetone, 20 ml of a freshly prepared solution of CrCl_2 was added under stirring, and the mixture was stirred for ~1 h at room temperature. Acetone was evaporated, the aqueous phase was extracted with chloroform (3×10 ml), the extracts were combined, washed with a solution of sodium chloride, dried over MgSO_4 , and evaporated, and the residue was purified by column chromatography on silica gel using ethyl acetate–petroleum ether (1:9) as eluent. Yield 0.14 g (68%), colorless crystals, mp $96\text{--}98^\circ\text{C}$. ^1H NMR spectrum, δ , ppm: 2.24 s (3H, CH_3), 3.35 s (3H) and 3.43 s (3H, OCH_3), 3.88 s (1H, 5-H), 6.15 d (1H, 4'-H, $J = 1.9$ Hz), 7.26 d (1H, 5'-H, $J = 1.9$ Hz). ^{13}C NMR spectrum, δ_C , ppm: 11.94 (CH_3), 51.38 and 51.51 (OCH_3), 54.74 (C^5), 102.49 (C^4), 111.53 (C^2), 111.35 (C^4'), 134.81 (C^3'), 140.34 (C^5'), 150.49 (C^2'), 158.49 (C^3), 192.14 (C=O). Found, %:

1.8 Hz); 7.25 d (1H, 5'-H, $J = 1.9$ Hz). ^{13}C NMR spectrum, δ_{C} , ppm: 11.93 (CH_3); 51.32 (C^5); 51.32, 52.13, and 60.18 (OCH_3); 102.24 (C^4); 108.91 (C^2); 111.43 (C^4); 112.52 (C^3'); 140.08 ($\text{C}^{5'}$); 150.22 (C^2'); 173.56 (C^3); 193.32 (C^1).

2-Chloro-3-diisopropylamino-4,4-dimethoxy-5-(2-methylfuran-3-yl)cyclopent-2-en-1-one (XV). ^1H NMR spectrum, δ , ppm: 1.34 d (3H, CH_3 , $J = 6.4$ Hz), 2.29 s (3H, CH_3), 3.16 s and 3.31 s (3H each, OCH_3), 3.43 m and 3.63 m (1H each, CH), 3.67 s (1H, 5-H), 6.34 d (1H, 4'-H, $J = 1.8$ Hz), 7.22 d (1H, 5'-H, $J = 1.96$ Hz).

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